

Supporting Information

Mitigating P2-O2 Transition and Na⁺/Vacancy Ordering in Na_{2/3}Ni_{1/3}Mn_{2/3}O₂ by

Anion/Cation Dual-Doping for Fast and Stable Na⁺ Insertion/Extraction

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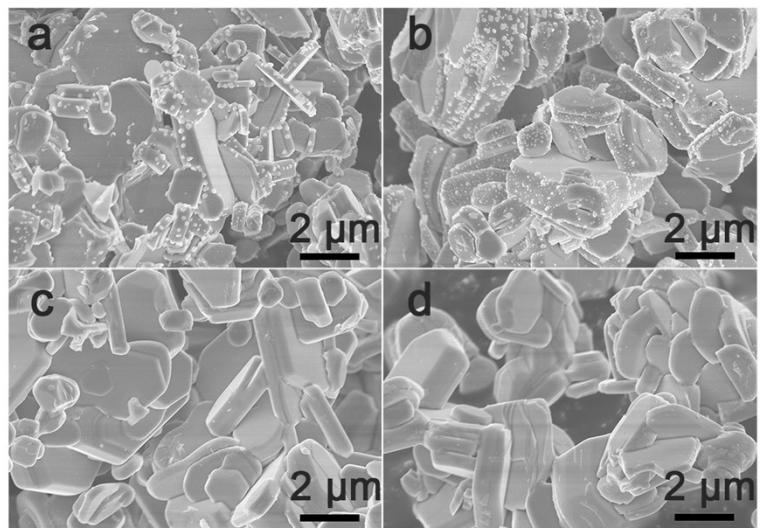


Fig. S1 SEM images of $\text{Na}_{0.67-x}\text{Ca}_x\text{Ni}_{0.33}\text{Mn}_{0.67}\text{O}_{2-2x}\text{F}_{2x}$ samples (a) $x = 0$, (b) $x = 0.01$, (c) $x = 0.03$ and (d) $x = 0.05$.

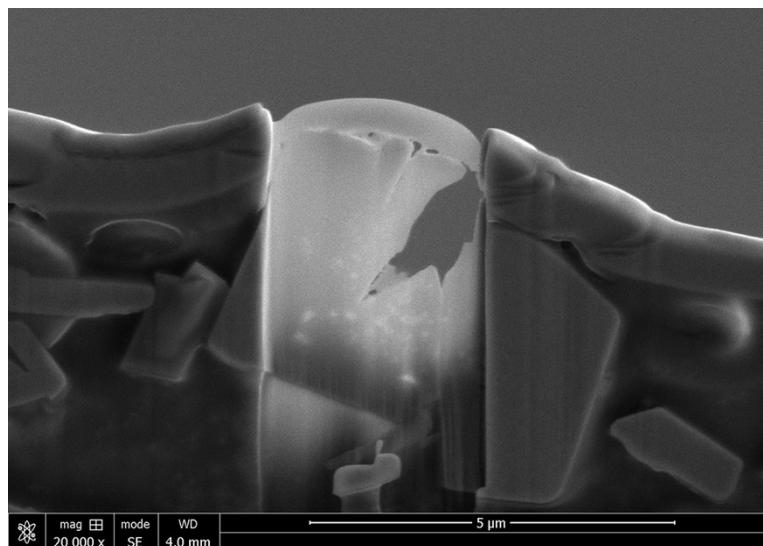


Fig. S2 FIB-cut image of TEM specimen.

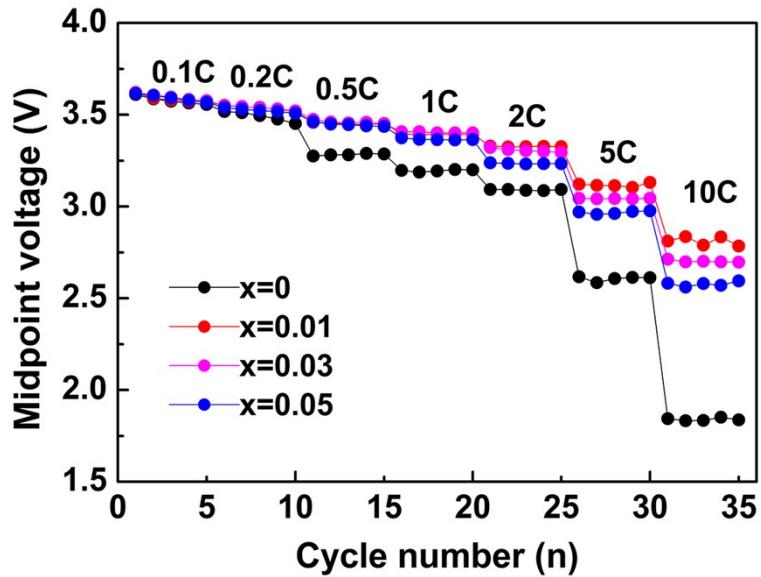


Fig. S3 Midpoint voltage of $\text{Na}_{0.67-x}\text{Ca}_x\text{Ni}_{0.33}\text{Mn}_{0.67}\text{O}_{2-2x}\text{F}_{2x}$ ($x = 0, 0.01, 0.03$ and 0.05) at various current densities.

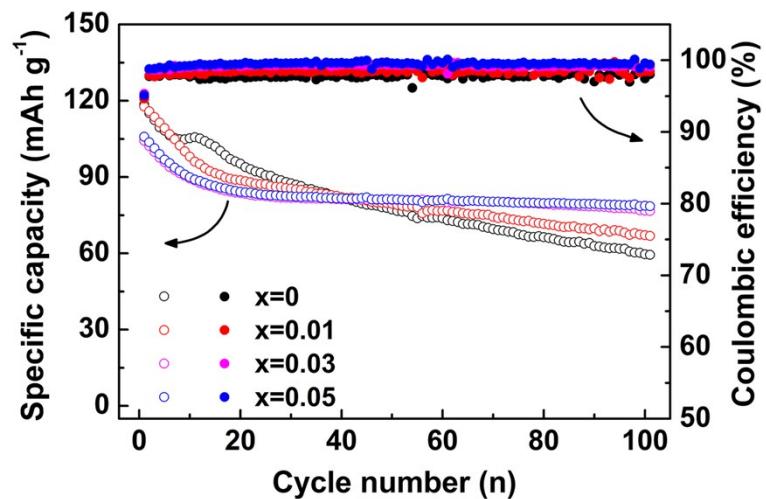


Fig. S4 Cycling performance of $\text{Na}_{0.67-x}\text{Ca}_x\text{Ni}_{0.33}\text{Mn}_{0.67}\text{O}_{2-2x}\text{F}_{2x}$ ($x = 0, 0.01, 0.03$ and 0.05) cycled at 0.2C in the voltage range of 2.0 - 4.3 V.

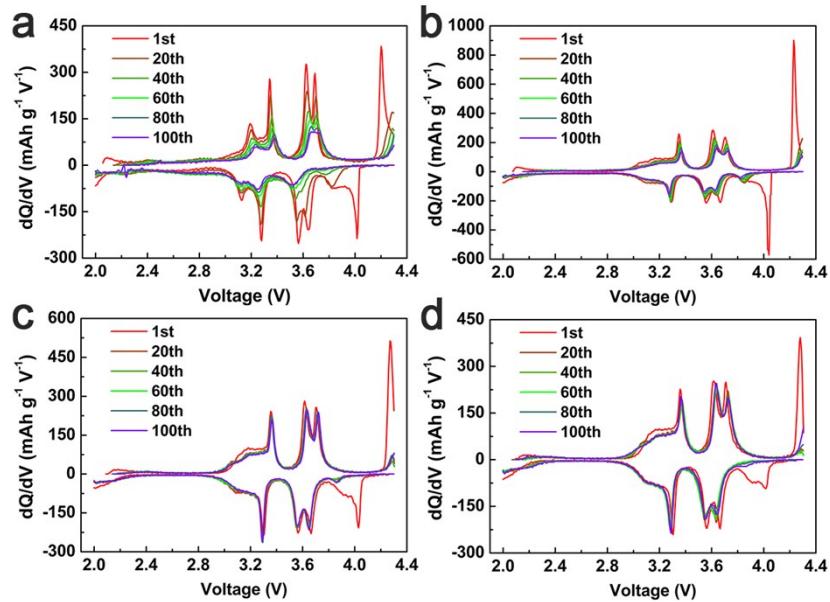


Fig. S5 The dQ/dV plots of $\text{Na}_{0.67-x}\text{Ca}_x\text{Ni}_{0.33}\text{Mn}_{0.67}\text{O}_{2-2x}\text{F}_{2x}$ (a) $x = 0$, (b) $x = 0.01$, (c) $x = 0.03$ and (d) $x = 0.05$ at 0.2C.

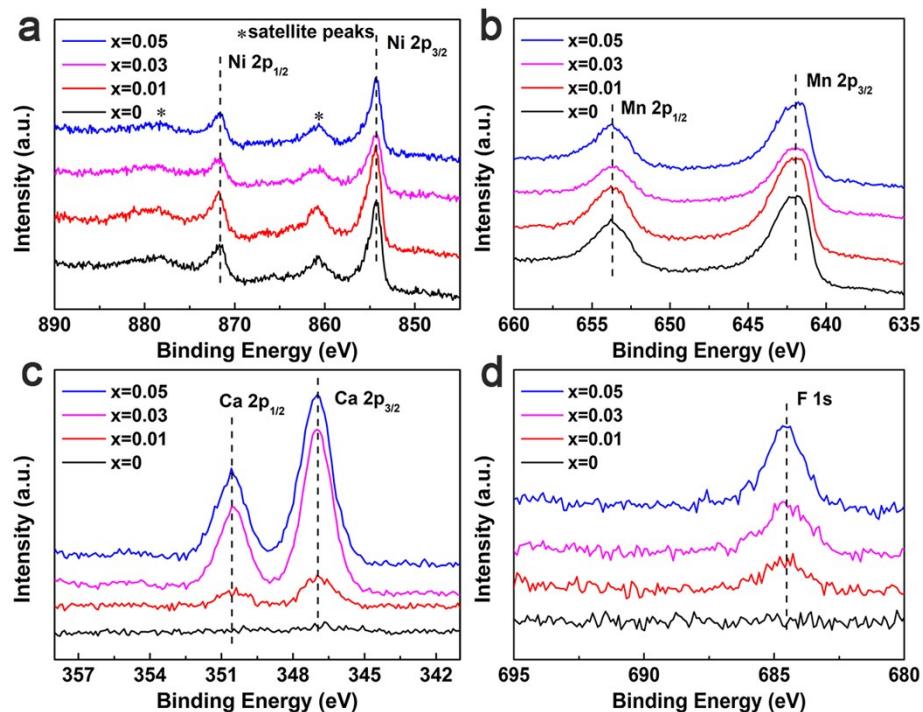


Fig. S6 XPS spectra of $\text{Na}_{0.67-x}\text{Ca}_x\text{Ni}_{0.33}\text{Mn}_{0.67}\text{O}_{2-2x}\text{F}_{2x}$ ($x = 0, 0.01, 0.03$ and 0.05) powders for (a) Ni 2p, (b) Mn 2p, (c) Ca 2p and (d) F 1s.

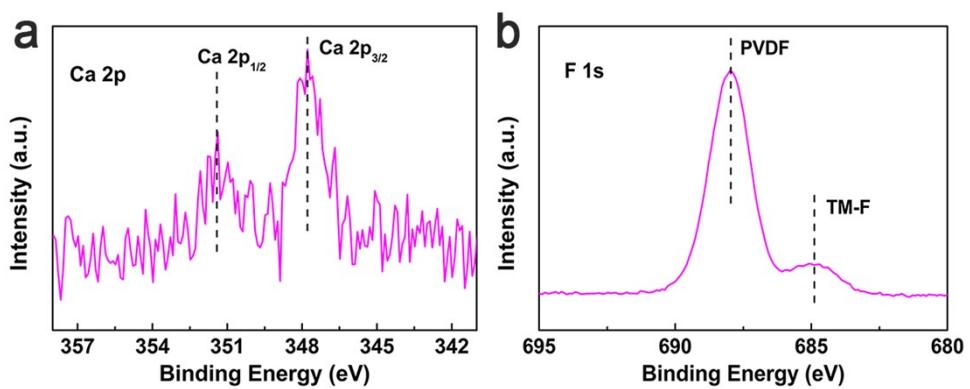


Fig. S7 XPS spectra of $\text{Na}_{0.67-x}\text{Ca}_x\text{Ni}_{0.33}\text{Mn}_{0.67}\text{O}_{2-2x}\text{F}_{2x}$ ($x = 0.03$) cathode charged to 4.3 V for (a) Ca 2p and (b) F 1s.

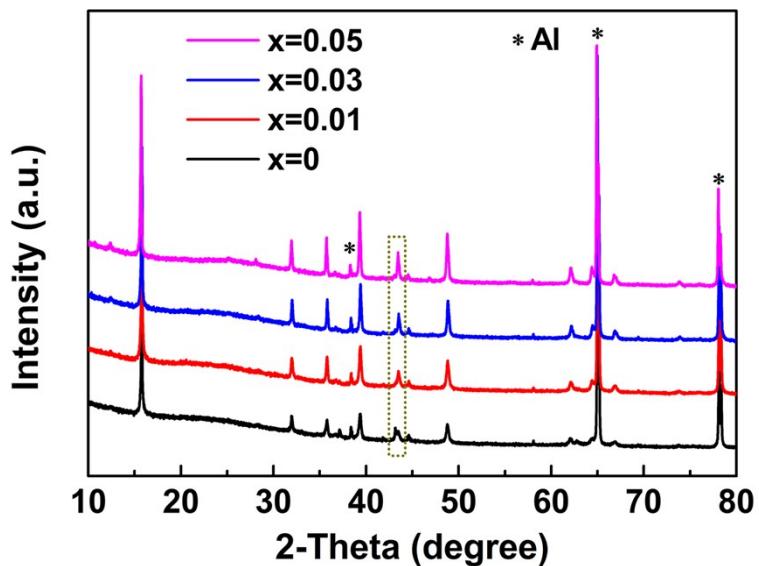


Fig. S8 Comparison of XRD patterns of $\text{Na}_{0.67-x}\text{Ca}_x\text{Ni}_{0.33}\text{Mn}_{0.67}\text{O}_{2-2x}\text{F}_{2x}$ ($x = 0, 0.01, 0.03$ and 0.05) after 100 cycles at 0.2C.

Table S1. Crystallographic parameters of $\text{Na}_{0.67-x}\text{Ca}_x\text{Ni}_{0.33}\text{Mn}_{0.67}\text{O}_{2-2x}\text{F}_{2x}$ ($x = 0$) refined by the Rietveld method.

Site	x	y	z	occ.
Na_f	0	0	0.25	0.3016
Na_e	0.3333	0.6667	0.75	0.3684
Ni	0	0	0	0.3260
Mn	0	0	0	0.6740
O	0.3333	0.6667	0.0890	1.0000
$a = 2.886(5) \text{ \AA}$ $c = 11.151(3) \text{ \AA}$ $V = 80.45(3) \text{ \AA}^3$ $R_p = 4.96\%$ $R_{wp} = 6.55\%$				

Table S2. Crystallographic parameters of $\text{Na}_{0.67-x}\text{Ca}_x\text{Ni}_{0.33}\text{Mn}_{0.67}\text{O}_{2-2x}\text{F}_{2x}$ ($x = 0.03$) refined by the Rietveld method.

Site	x	y	z	occ.
Na_f	0	0	0.25	0.2427
Na_e	0.3333	0.6667	0.75	0.3981
Ca_e	0.3333	0.6667	0.75	0.0292
Ni	0	0	0	0.3204
Mn	0	0	0	0.6796
O	0.3333	0.6667	0.0886	0.9400
F	0.3333	0.6667	0.0886	0.0600
$a = 2.885(3) \text{ \AA}$ $c = 11.137(5) \text{ \AA}$ $V = 80.31(3) \text{ \AA}^3$ $R_p = 4.56\%$ $R_{wp} = 5.74\%$				

Table S3. Comparison of the electrochemical performance of $\text{Na}_{0.67-x}\text{Ca}_x\text{Ni}_{0.33}\text{Mn}_{0.67}\text{O}_{2-2x}\text{F}_{2x}$ and other P2-Na-Ni-Mn-O type materials.

Material	Voltage window	Capacity retention	Reference
$\text{Na}_{2/3}\text{Cu}_{1/6}\text{Ni}_{1/6}\text{Mn}_{2/3}\text{O}_2$	2.5-4.4 V	97.0%, 30 cycles, 1C	1
$\text{Na}_{2/3}\text{Ni}_{1/4}\text{Mn}_{2/3}\text{Cu}_{1/12}\text{O}_2$	2.0-4.5 V	89.0%, 50 cycles, 0.05C	2
$\text{Na}_{0.67}\text{Ni}_{0.25}\text{Mg}_{0.05}\text{Mn}_{0.7}\text{O}_2$	2.0-4.5 V	91.5%, 50 cycles, 200 mA g ⁻¹	3
$\text{Na}_{0.67}\text{Mn}_{0.65}\text{Ni}_{0.2}\text{Co}_{0.15}\text{O}_2$	2.0-4.3 V	62.0%, 100 cycles, 0.1C	4
$\text{Na}_{2/3}\text{Ni}_{1/3}\text{Mn}_{1/3}\text{Ti}_{1/3}\text{O}_2$	2.5-4.35 V	83.9%, 500 cycles, 1C	5
$\text{Al}_2\text{O}_3\text{-Na}_{2/3}[\text{Ni}_{1/3}\text{Mn}_{2/3}]\text{O}_2$	2.5-4.3 V	73.2%, 300 cycles, 0.5C	6
$\text{Na}_{0.78}\text{Al}_{0.05}\text{Ni}_{0.33}\text{Mn}_{0.60}\text{O}_2$	2.0-4.5 V	83.9%, 50 cycles, 0.1C	7
$\text{Na}_{0.67}\text{Fe}_{0.20}\text{Ni}_{0.15}\text{Mn}_{0.65}\text{O}_2$	2.0-3.8 V	70.0%, 900 cycles, 1C	8
$\text{Na}_{2/3}[\text{Ni}_{0.3}\text{Co}_{0.1}\text{Mn}_{0.6}]\text{O}_2$	2.0-4.3 V	79.2%, 50 cycles, 0.1C	9
$\text{Na}_{2/3}\text{Ni}_{1/3}\text{Mn}_{7/12}\text{Fe}_{1/12}\text{O}_2$	2.0-4.4 V	85.0%, 300 cycles, 5C	10
$\text{Na}_{0.67-x}\text{Ca}_x\text{Ni}_{0.33}\text{Mn}_{0.67}\text{O}_{2-2x}\text{F}_{2x}$	2.0-4.3 V	87.2%, 500 cycles, 1C	This work

References

1. L. Zheng, J. Li and M. N. Obrovac, *Chem. Mater.*, 2017, **29**, 1623-1631.
2. K. Kubota, Y. Yoda and S. Komaba, *J. Electrochem. Soc.*, 2017, **164**, A2368-A2373.
3. G. Singh, N. Tapia-Ruiz, J. M. Lopez del Amo, U. Maitra, J. W. Somerville, A. R. Armstrong, J. Martinez de Ilarduya, T. Rojo and P. G. Bruce, *Chem. Mater.*, 2016, **28**, 5087-5094.
4. Y. Wen, J. Fan, C. Shi, P. Dai, Y. Hong, R. Wang, L. Wu, Z. Zhou, J. Li, L. Huang and S.-G. Sun, *Nano Energy*, 2019, **60**, 162-170.

5. P.-F. Wang, H.-R. Yao, X.-Y. Liu, Y.-X. Yin, J.-N. Zhang, Y. Wen, X. Yu, L. Gu and Y.-G. Guo, *Sci. Adv.*, 2018, **4**, eaar6018.
6. Y. Liu, X. Fang, A. Zhang, C. Shen, Q. Liu, H. A. Enaya and C. Zhou, *Nano Energy*, 2016, **27**, 27-34.
7. Y. Shi, S. Li, A. Gao, J. Zheng, Q. Zhang, X. Lu, L. Gu and D. Cao, *ACS Appl. Mater. Interfaces*, 2019, **11**, 24122-24131.
8. C. Luo, A. Langrock, X. Fan, Y. Liang and C. Wang, *J. Mater. Chem. A*, 2017, **5**, 18214-18220.
9. P. Hou, Y. Sun, F. Li, Y. Sun, X. Deng, H. Zhang, X. Xu and L. Zhang, *Nanoscale*, 2019, **11**, 2787-2794.
10. Q. Yang, P.-F. Wang, J.-Z. Guo, Z.-M. Chen, W.-L. Pang, K.-C. Huang, Y.-G. Guo, X.-L. Wu and J.-P. Zhang, *ACS Appl. Mater. Interfaces*, 2018, **10**, 34272-34282.