Supplementary Data

Stabilization of Lithium Metal Anodes by Conductive Metal-Organic Framework Architectures

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Table of Contents

1.	Supplementary Table
2.	Supplementary Figures

1. Supplementary Table

 Table S1. The values of adsorption energy and diffusion barrier of Li ions on different surfaces

 calculated by the DFT method.

Sample	Adsorption energy (eV Å-2)	Diffusion barrier (eV Å-2)
Cu (111)	-2.37	1.76
Ni ₃ (HITP) ₂ @Cu-foil	-3.18	0.71

2. Supplementary Figures



Fig. S1. SEM images of Ni-NDC@Cu-foil current collector: (a) low magnification, (b) high magnification. SEM

images of Ni₃(HITP)₂@Cu-foil after cycling: (c) low magnification, (d) high magnification.



Fig. S2. Full XRD patterns of different current collectors.



Fig. S3. (a) Wide-scan XPS spectra and (b) high-resolution N 1s XPS spectra of the Ni₃(HITP)₂@Cu-foil after

cycling. Li 1s XPS spectra of (c) initial Ni₃(HITP)₂@Cu-foil and (d) Li/Ni₃(HITP)₂@Cu-foil.



Fig. S4. (a) Voltage profiles of Li plating/stripping on/from planar Cu-foil at a current density of 5 mA cm⁻² with an areal capacity of 5 mAh cm⁻² after different cycles. (b) EIS plots of $Ni_3(HITP)_2@Cu-foil at different cycles.$



Fig. S5. SEM images of lithium metal deposited on the planar Cu-foil current collector under (a) low magnification and (b) high magnification.



Fig. S6. a) Geometric configuration and corresponding charge density differences of a Li atom on the Cu-foil and Ni₃(HITP)₂@Cu-foil (the yellow and cyan regions represent positive and negative densities respectively. The isovalue for the charge density was taken as ± 0.002 e Å⁻³). b) A comparison of the diffusion barriers on Cu (111) and Ni₃(HITP)₂. c) The diffusion pathway of Li atoms through the Ni₃(HITP)₂ layers.



Fig. S7. (a) Voltage profiles and (b) rate capability of the Li/Ni₃(HITP)₂@Cu-foil|LFP cells with LFP capacity of 1.5 mAh cm⁻². (c) Voltage profiles and (d) rate capability of the Li/Ni₃(HITP)₂@Cu-foil|LFP cells with LFP capacity of 2.4 mAh cm⁻².