

Supporting Information

Hollow platinum tetrapods: using a combination of {111} facets, surface concave topology, and ultrathin walls to boost their oxygen reduction reactivity

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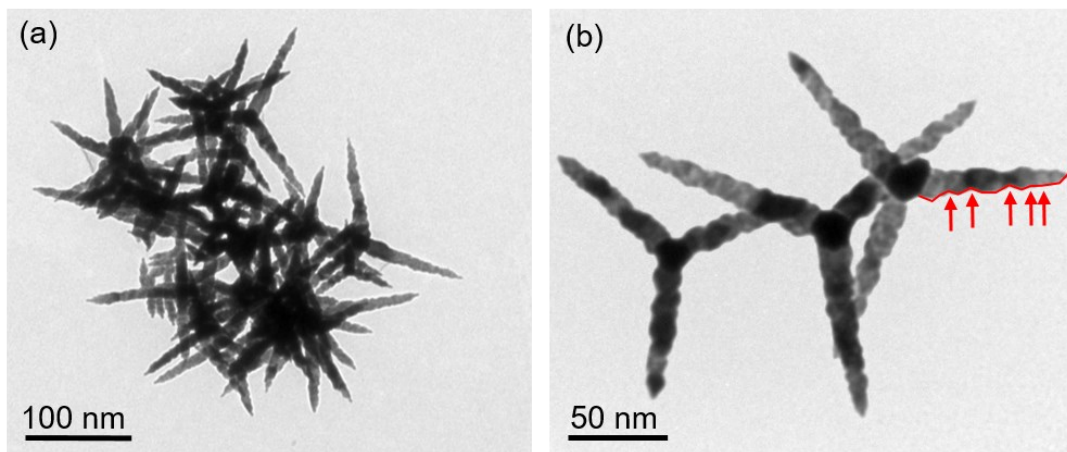


Fig. S1 TEM images of the monodispersed Pd tetrapods as seed.

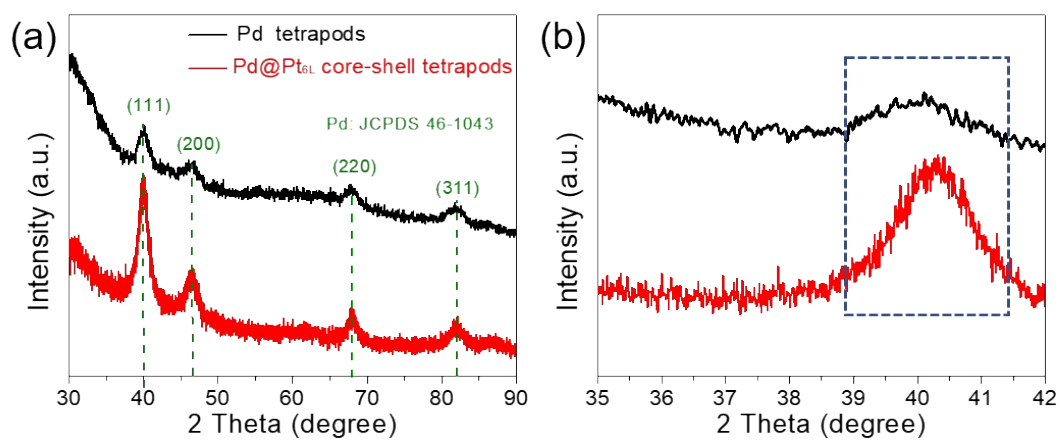


Fig. S2 XRD pattern of the Pd tetrapods and Pd@Pt_{6L} core-shell tetrapods.

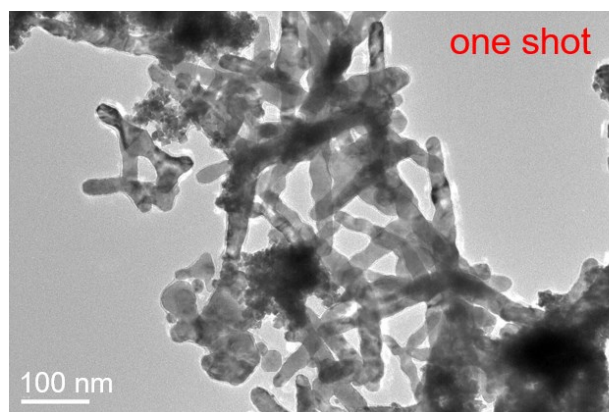


Fig. S3 TEM image of the products obtained by using the standard protocol of Pd@Pt_{6L} core-shell tetrapods except the injection rate.

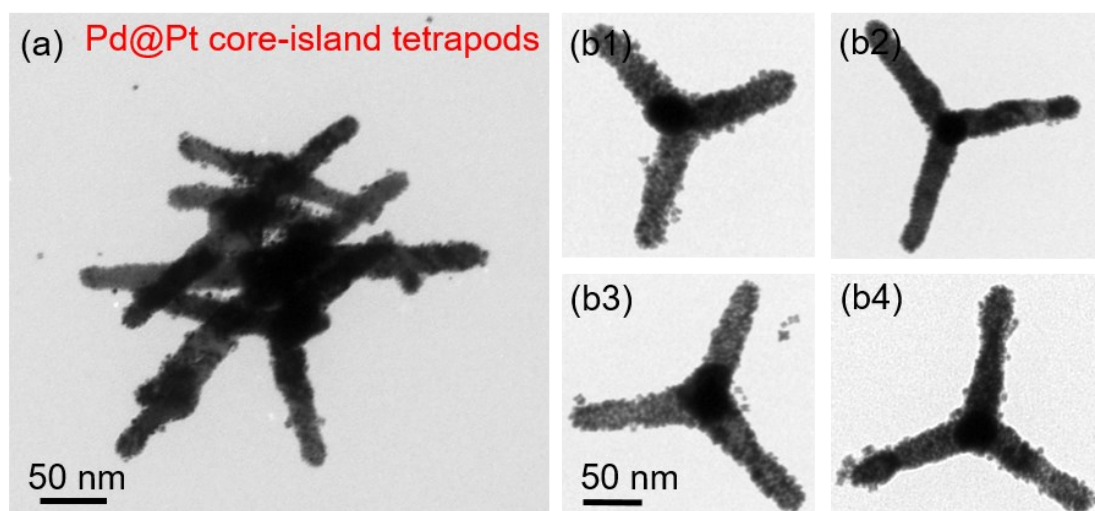


Fig. S4 TEM images of the Pd@Pt core-island tetrapods obtained by using the standard protocol of Pd@Pt_{6L} core-shell tetrapods except the reaction temperature.

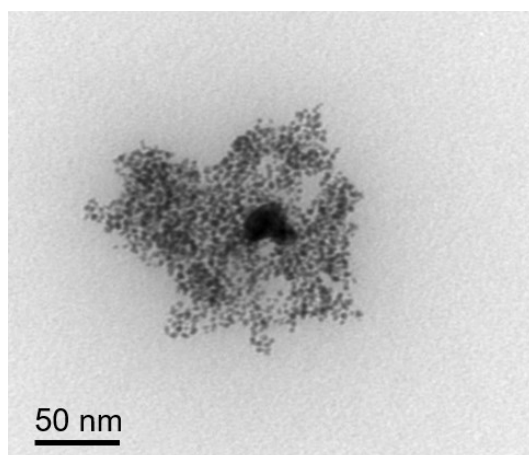


Fig. S5 TEM images of the products after etching the Pd core of Pd@Pt core-island tetrapods.

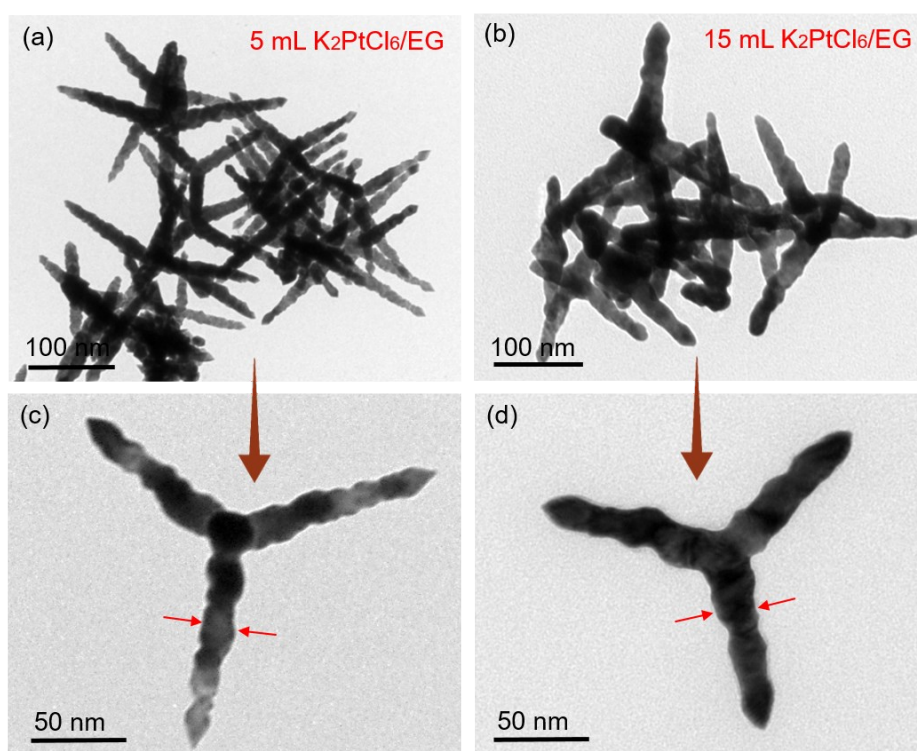


Fig. S6 TEM images of the products obtained by using the standard protocol of Pd@Pt_{6L} core-shell tetrapods except the volume of precursor. (a), (c) 5 mL of K₂PtCl₆/EG solution. (b), (d) 15 mL of K₂PtCl₆/EG solution.

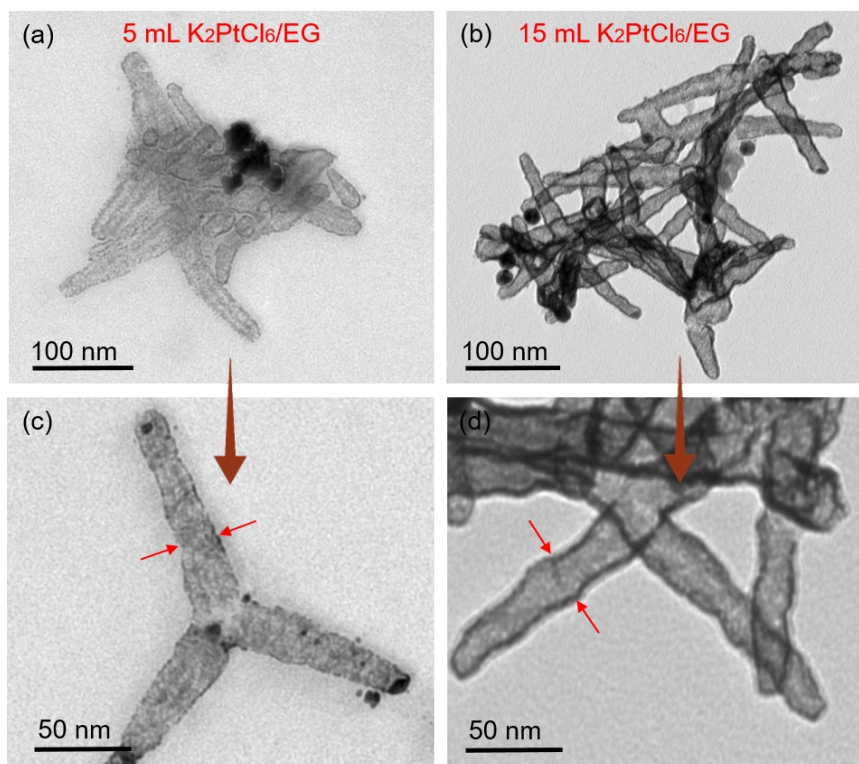


Fig. S7 TEM images of the products obtained by etching the sample from Fig. S6.

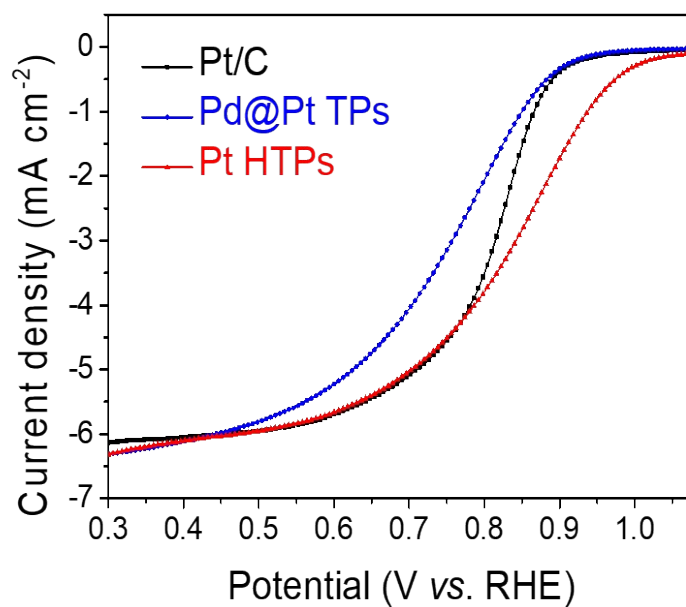


Fig. S8 ORR polarization curves recorded in O_2 -saturated 0.1 M $HClO_4$ solution.

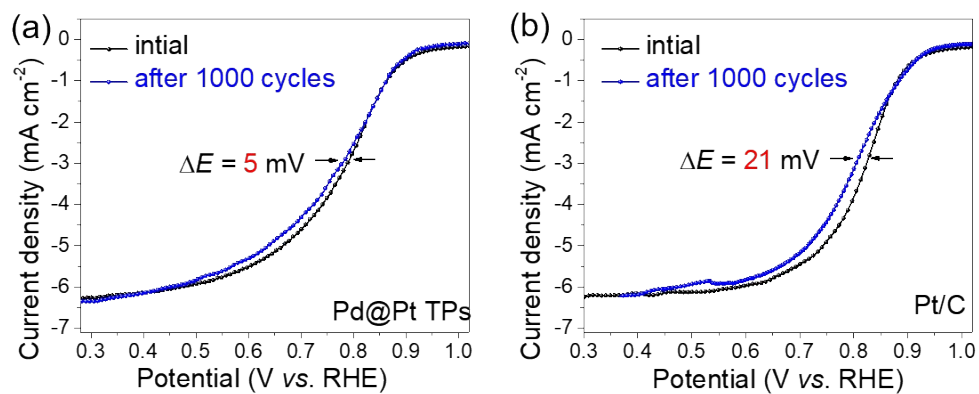


Fig. S9 ORR polarization curves of the (a) Pd@Pt TPs, and (b) Pt/C before and after 1000 cycles at a scan rate 100 mV s⁻¹.

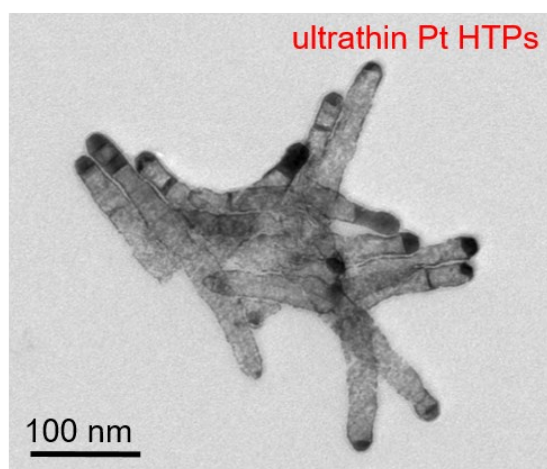


Fig. S10 TEM image of the ultrathin Pt HTPs after ADTs.

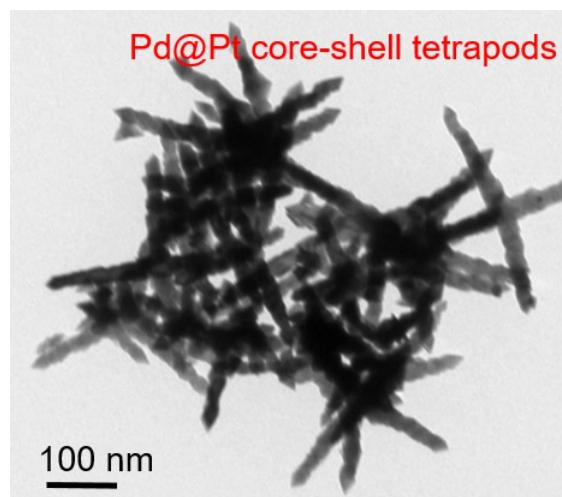


Fig. S11 TEM image of the Pd@Pt core-shell tetrapods after ADTs.

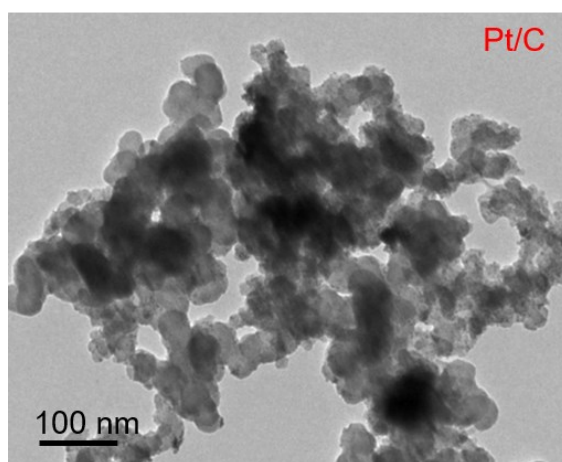


Fig. S12 TEM image of the Pt/C after ADTs.

Table S1 Comparison of the ORR performance of the ultrathin Pt HTPs with some previously reported noble metal-based catalysts in 0.5 M H₂SO₄ solution.

Number	Catalysts	E_{onset} (V vs. RHE)	$E_{1/2}$ (V vs. RHE)	Reference
1	Pd@Pt TPs	0.995	0.789	This work

2	Pt HTPs	1.014	0.836	This work
3	Pt/C	1.009	0.825	This work
4	Pt-WP-CL/AEG-3	0.720	0.610	[1]
5	Pt/TiO ₂ -2/C	0.880	0.727	[2]
6	Pt/OMC	~ 0.930	~ 0.750	[3]
7	5 wt % Pt-CeO _x NW/C	0.890	0.750	[4]
8	tensile strained 5 nm Pt	~ 0.890	0.672	[5]
9	Pt/C-(NH ₄) ₂ PtCl ₆	~ 0.950	~ 0.810	[6]

References

1. C. Zhang, Y. Dai, H. Chen, Y. Ma, B. Jing, Z. Cai, Y. Duan, B. Tang, J. Zou, *J. Mater. Chem. A*, 2018, **6** (45), 22636-22644.
2. E. C. M. Barbosa, L. S. Parreira, I. C. de Freitas, L. R. Aveiro, D. C. de Oliveira, M. C. dos Santos, P. H. C. Camargo, *ACS Appl. Energy Mater.*, 2019, **2**, 5759-5768.
3. M. Sakthivel, J.-F. Drillet, *Appl. Catal. B: Environ.*, 2018, **231**, 62-72.
4. S. Chauhan, T. Mori, T. Masuda, S. Ueda, G. J. Richards, J. P. Hill, K. Ariga, N. Isaka, G. Auchterlonie, J. Drennan, *ACS Appl. Mater. Interfaces*, 2016, **8**, 9059-9070.
5. M. Du, L. Cui, Y. Cao, A. J. Bard, *J. Am. Chem. Soc.*, 2015, **137**, 7397-7403.
6. R. Sharma, Y. Wang, F. Li, J. Chamier, S. M. Andersen, *ACS Appl. Energy Mater.*, 2019, **2**, 6875-6882.