

Supporting Information:

Structural Stability and Defect-Tolerance of Ionic Spinel Semiconductor for High-Efficiency Solar Cells

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Part 1. Chemical potential conditions of HgX₂S₄ phase

Since the formation energy of defects is the function of chemical potential, we need to obtain the phase diagram which limits the region of chemical potential before calculating defects' formation energies. HgX₂S₄ phase diagram is thermodynamically limited by the following conditions. To avoid the pure phase of elements, the following conditions must be satisfied

$$\mu_{\text{Hg}} \leq 0, \mu_{\text{X}} \leq 0, (\text{X} = \text{In}, \text{Sc} \text{ and } \text{Y}), \mu_{\text{S}} \leq 0. \quad (\text{S1})$$

Similarly, to exclude the formation of secondary phase of compounds, their formation enthalpies must satisfy eq. S2 for HgIn₂S₄, eq. S3 for HgSc₂S₄ and eq. S4 for HgY₂S₄.

$$\left\{ \begin{array}{ll} \mu_{\text{Hg}} + \mu_{\text{S}} \leq \Delta H_{\text{HgS}} (-0.42 \text{ eV}) & 2\mu_{\text{In}} + 3\mu_{\text{S}} \leq \Delta H_{\text{In}_2\text{S}_3} (-4.03 \text{ eV}) \\ \mu_{\text{Hg}} + 4\mu_{\text{S}} \leq \Delta H_{\text{HgS}_4} (-1.37 \text{ eV}) & 3\mu_{\text{In}} + \mu_{\text{S}} \leq \Delta H_{\text{In}_3\text{S}} (-1.27 \text{ eV}) \\ \mu_{\text{In}} + \mu_{\text{Hg}} \leq \Delta H_{\text{InHg}} (-0.42 \text{ eV}) & 3\mu_{\text{In}} + 4\mu_{\text{S}} \leq \Delta H_{\text{In}_3\text{S}_4} (-5.08 \text{ eV}) \\ \mu_{\text{In}} + 3\mu_{\text{Hg}} \leq \Delta H_{\text{InHg}_3} (-0.46 \text{ eV}) & 5\mu_{\text{In}} + 4\mu_{\text{S}} \leq \Delta H_{\text{In}_5\text{S}_4} (-6.13 \text{ eV}) \\ 3\mu_{\text{In}} + \mu_{\text{Hg}} \leq \Delta H_{\text{In}_3\text{Hg}} (-1.29 \text{ eV}) & 5\mu_{\text{In}} + 8\mu_{\text{S}} \leq \Delta H_{\text{In}_5\text{S}_8} (-8.48 \text{ eV}) \\ \mu_{\text{In}} + \mu_{\text{S}} \leq \Delta H_{\text{InS}} (-1.61 \text{ eV}) & 6\mu_{\text{In}} + 7\mu_{\text{S}} \leq \Delta H_{\text{In}_6\text{S}_7} (-10.23 \text{ eV}) \\ \mu_{\text{In}} + 2\mu_{\text{S}} \leq \Delta H_{\text{InS}_2} (-0.64 \text{ eV}) & 11\mu_{\text{In}} + 16\mu_{\text{S}} \leq \Delta H_{\text{In}_{11}\text{S}_{16}} (-21.31 \text{ eV}) \end{array} \right. \quad (\text{S2})$$

$$\left\{ \begin{array}{ll} \mu_{\text{Hg}} + \mu_{\text{S}} \leq \Delta H_{\text{HgS}} (-0.42 \text{ eV}) & 3\mu_{\text{Sc}} + \mu_{\text{Hg}} \leq \Delta H_{\text{Sc}_3\text{Hg}} (-1.14 \text{ eV}) \\ \mu_{\text{Hg}} + 4\mu_{\text{S}} \leq \Delta H_{\text{HgS}_4} (-1.37 \text{ eV}) & \mu_{\text{Sc}} + \mu_{\text{S}} \leq \Delta H_{\text{ScS}} (-4.02 \text{ eV}) \\ \mu_{\text{Sc}} + \mu_{\text{Hg}} \leq \Delta H_{\text{ScHg}} (-1.06 \text{ eV}) & \mu_{\text{Sc}} + 2\mu_{\text{S}} \leq \Delta H_{\text{ScS}_2} (-3.68 \text{ eV}) \\ \mu_{\text{Sc}} + 3\mu_{\text{Hg}} \leq \Delta H_{\text{ScHg}_3} (-1.18 \text{ eV}) & 2\mu_{\text{Sc}} + 3\mu_{\text{S}} \leq \Delta H_{\text{Sc}_2\text{S}_3} (-10.60 \text{ eV}) \end{array} \right. \quad (\text{S3})$$

$$\left\{ \begin{array}{ll} \mu_{\text{Hg}} + \mu_{\text{S}} \leq \Delta H_{\text{HgS}} (-0.42 \text{ eV}) & \mu_{\text{Y}} + \mu_{\text{S}} \leq \Delta H_{\text{YS}} (-4.39 \text{ eV}) \\ \mu_{\text{Hg}} + 4\mu_{\text{S}} \leq \Delta H_{\text{HgS}_4} (-1.37 \text{ eV}) & \mu_{\text{Y}} + 2\mu_{\text{S}} \leq \Delta H_{\text{YS}_2} (-0.30 \text{ eV}) \\ \mu_{\text{Y}} + \mu_{\text{Hg}} \leq \Delta H_{\text{YHg}} (-1.26 \text{ eV}) & 2\mu_{\text{Y}} + 3\mu_{\text{S}} \leq \Delta H_{\text{Y}_2\text{S}_3} (-11.64 \text{ eV}) \\ \mu_{\text{Y}} + 2\mu_{\text{Hg}} \leq \Delta H_{\text{YHg}_2} (-1.57 \text{ eV}) & 5\mu_{\text{Y}} + 7\mu_{\text{S}} \leq \Delta H_{\text{Y}_5\text{S}_7} (-27.94 \text{ eV}) \\ \mu_{\text{Y}} + 3\mu_{\text{Hg}} \leq \Delta H_{\text{YHg}_3} (-1.83 \text{ eV}) & \end{array} \right. \quad (\text{S4})$$

Part 2. Degeneracy factor of intrinsic defects in HgX₂S₄ spinel system

The degeneracy factor g_q of a charged defect level and the arrangement of electron on defect energy levels of different defects are listed in the following Table S1.

Table S1. Degeneracy factor of intrinsic defects in HgX₂S₄ spinel system combing with the consideration of energy level splitting in crystal field.

Defects	q	g_q		Crystal fields		Configurations (take $q = +1/q = -1$ as an example)	
		X=In	X=Sc/Y	X=In	X=Sc/Y	X=In	X=Sc/Y
Hg _i	0	$C_2^2 = 1$				∥ Hg∥	
	1+		$C_2^1 = 2$		D_{3d}	6s ↑	
	2+		$C_2^0 = 1$				
X _i	0	$C_2^1 = 2$	$C_2^1 C_2^1 = 4$			∥ In∥	∥ X∥ E'_g
	1+	$C_2^2 = 1$	$C_2^2 = 1$		D_{3d}	5p → $5p_x / 5p_y$ → 3d → A_{1g}	
	2+	$C_2^1 = 2$	$C_2^1 = 2$			5p → $5p_z$ → 5s → 4s → E_g	
	3+	$C_2^0 = 1$	$C_2^0 = 1$			5s → 4s → E_g	
X _{Hg}	0	$C_2^1 = 2$	$C_2^1 = 2$		T_d	∥ In∥	∥ X∥ T_2
	1+	$C_2^0 = 1$	$C_2^0 = 1$			5p → 5s → 3d → E	
V _S	0	$C_2^1 (C_2^1)^3 = 16$				∥ Hg	+ 3Sc
	1+	$C_2^1 (C_2^1)^3 = 16$				6s ↑↓	3d → 4s → ↑↓
	2+	$C_2^0 (C_2^0)^3 = 1$					(take X=Sc as an example)
V _X	0	$C_2^1 C_3^2 C_2^1 C_3^1 = 36$					(S)
	1-	$C_2^1 C_3^1 = 6$				3p ↑↓ ↑ ↑	
	2-	$C_2^1 C_3^2 = 6$				3s ↑↓ ↑↓	
	3-	$C_3^0 = 1$					
V _{Hg}	0	$C_2^1 C_3^1 = 6$					(S)
	1-	$C_2^1 C_3^2 = 6$				3p ↑↓ ↑↓ ↑	
	2-	$C_3^0 = 1$				3s ↑↓ ↑↓	
Hg _X	0	$C_2^1 C_3^2 = 6$					(S)
	1-	$C_3^0 = 1$			O_h	3p ↑↓ ↑↓ ↑↓	

Part 3. Optical absorption coefficient and partial charge density of CBM

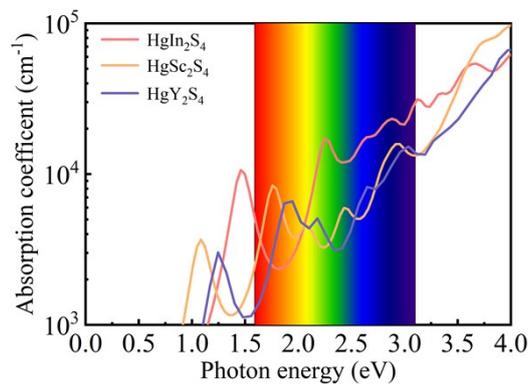


Fig. S1. Optical absorption coefficient of HgX_2S_4 spinel semiconductors ($X=\text{In, Sc}$ and Y).

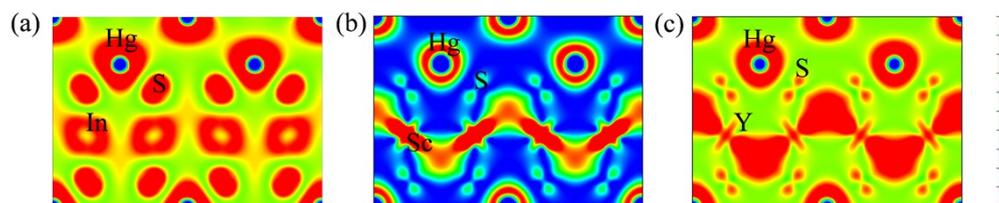


Fig. S2. Partial charge density of CBM for (a) HgIn_2S_4 , (b) HgSc_2S_4 and (c) HgY_2S_4 .