

Electronic Supplementary Material (ESI) for Journal of Materials Chemistry A.

This journal is © The Royal Society of Chemistry 2021

Supporting information

Ru single atoms and nanoclusters on highly porous N-doped carbon as hydrogen evolution catalyst in alkaline solutions with ultrahigh mass activity and turnover frequency

Yijun Li,^{‡a} Huijing Liu,^{‡a} Bing Li,^a Zhenzhen Yang,^a Zhenguo Guo,^a Jian-Bo He,^a

*Jianhui Xie^{*a} and Tai-Chu lau^{*b}*

^a*School of Chemistry and Chemical Engineering, Hefei University of Technology,
Hefei 230009, People's Republic of China. E-mail: jianhuixie@hfut.edu.cn*

^b*Department of Chemistry, City University of Hong Kong, Kowloon Tong, Hong
Kong, P R China. bhtclau@cityu.edu.hk*

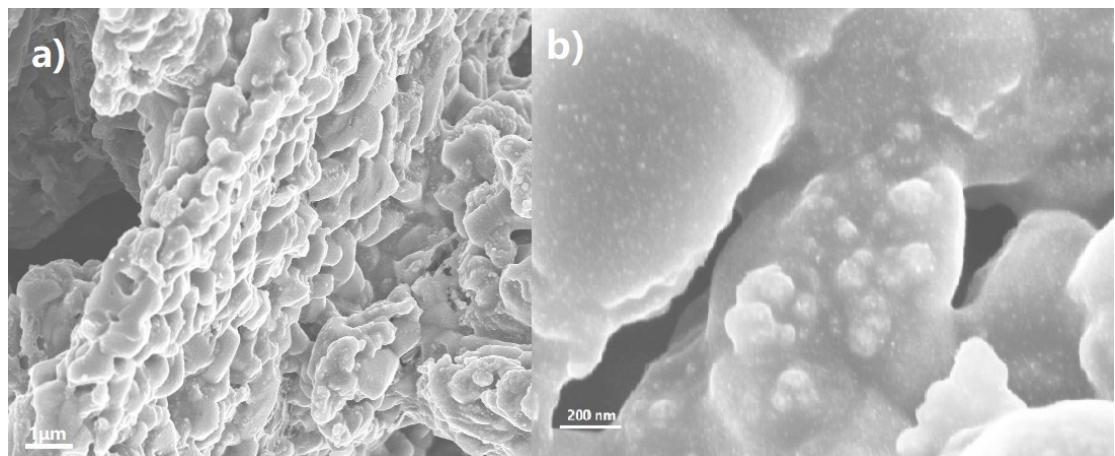


Fig. S1 SEM images of Ru/NC.

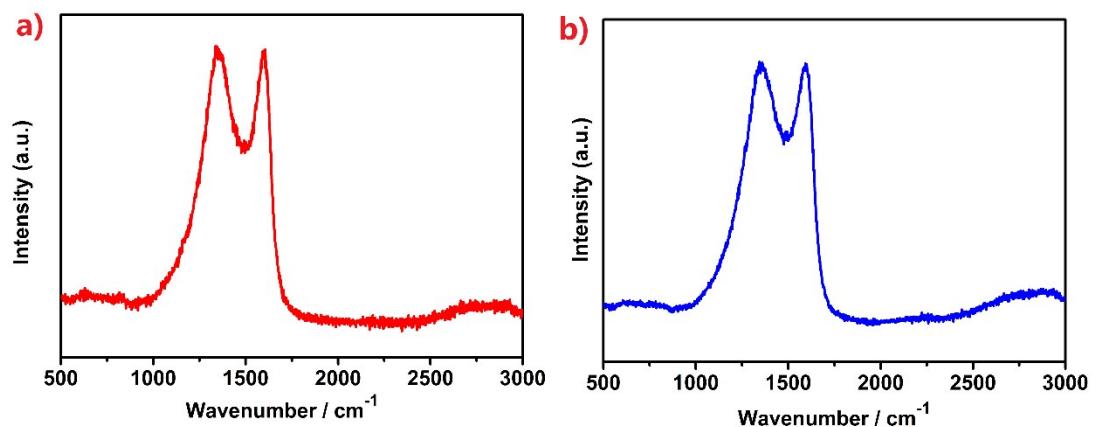


Fig. S2 Raman spectra of the catalysts. a) Ru/p-NC. b) Ru/NC.

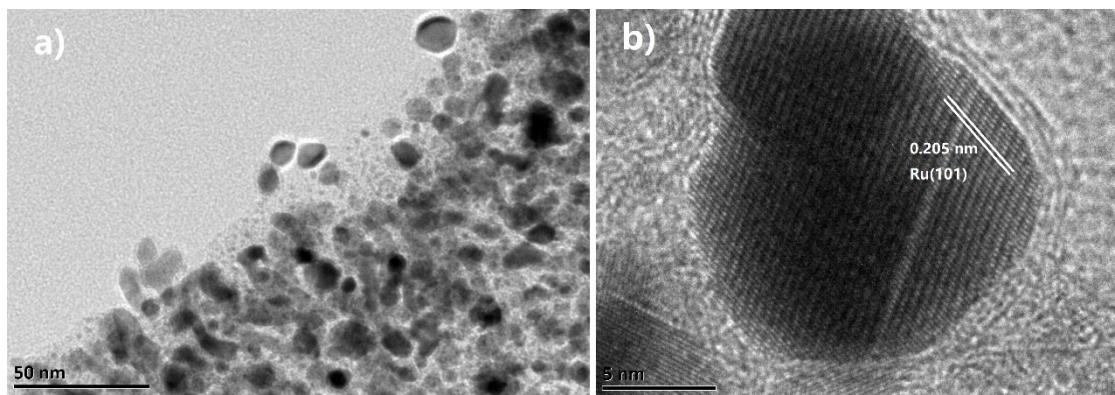


Fig. S3 a) TEM image of Ru/NC. b) HR-TEM image of Ru/NC.

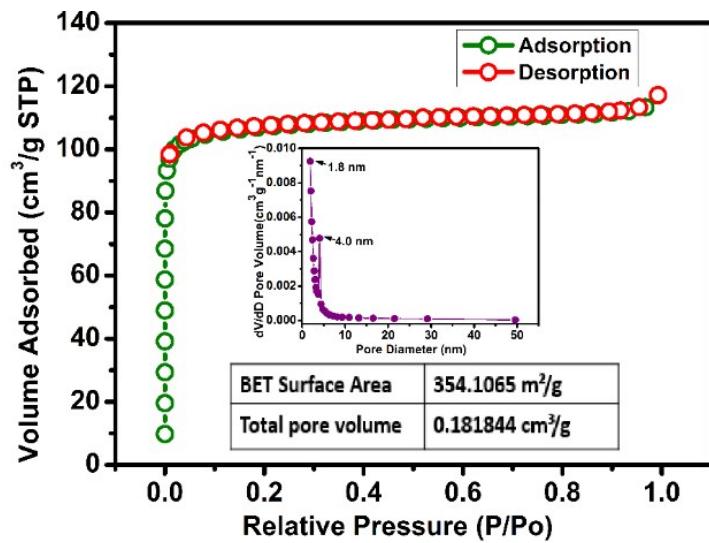


Fig. S4 N_2 adsorption/desorption isotherms of Ru/NC. Inset: corresponding pore size distribution.

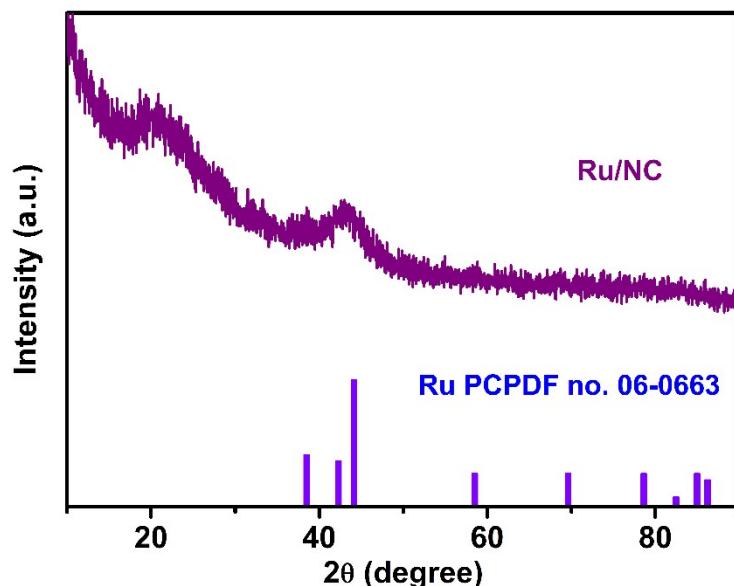


Fig. S5 XRD spectrum of Ru/NC.

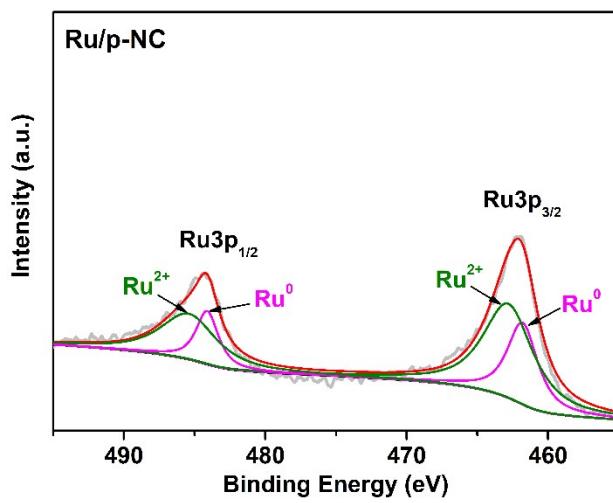


Fig. S6 High–resolution XPS spectrum of Ru 3p of Ru/p-NC.

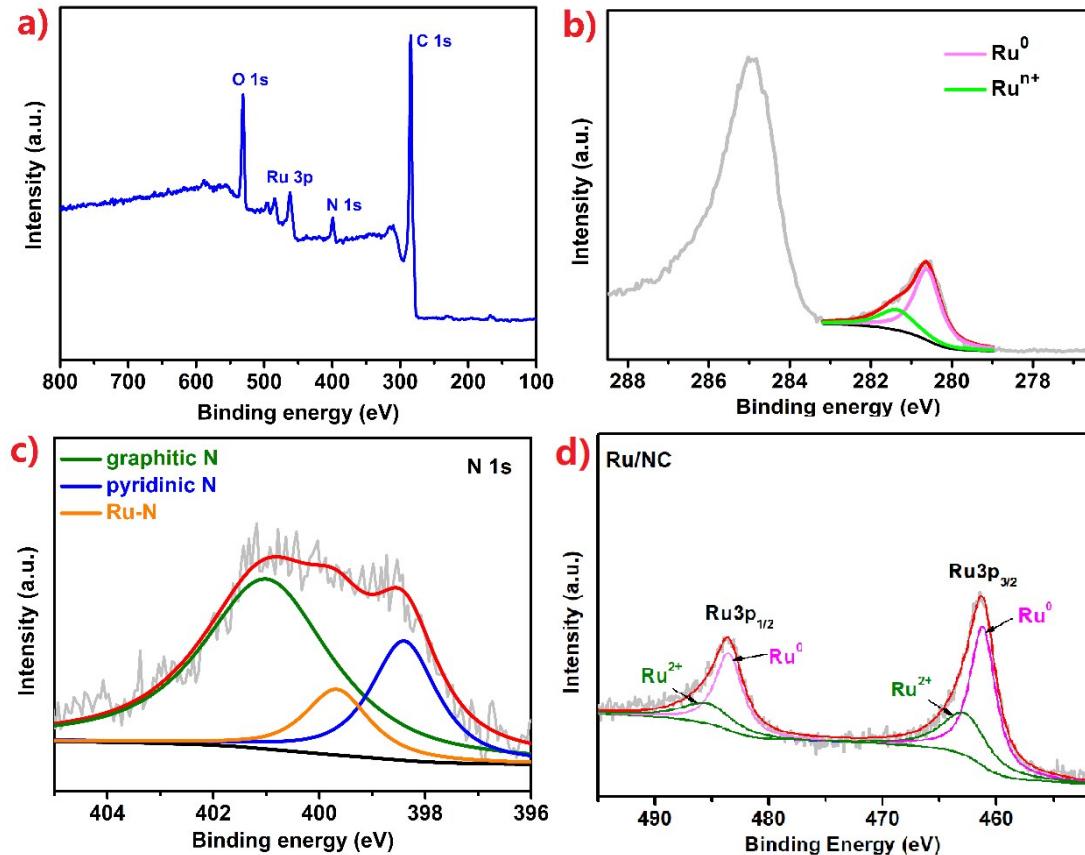


Fig. S7 a) XPS spectrum of Ru/NC. b) High–resolution XPS spectrum of C1s and Ru 3d_{5/2}. c) High–resolution XPS spectrum of N 1s. d) High–resolution XPS spectrum of Ru 3p.

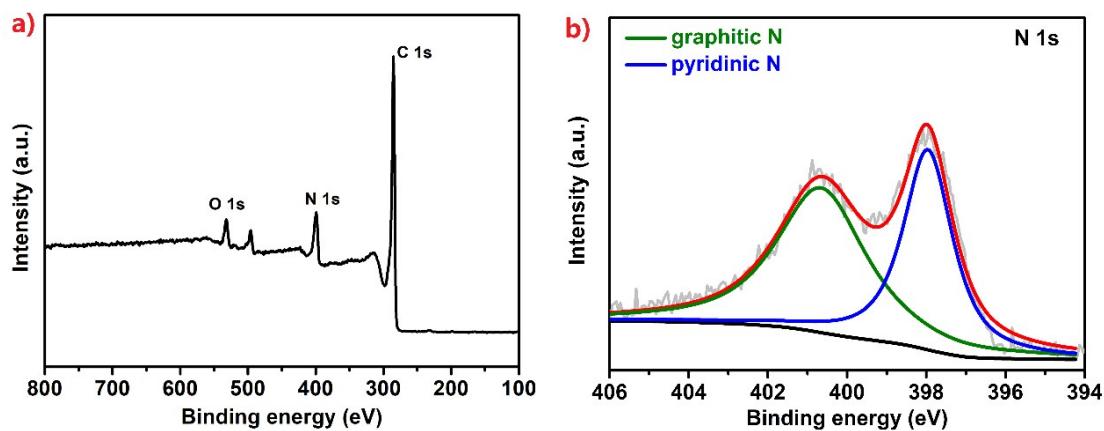


Fig. S8 a) XPS spectrum of NC. d) High-resolution XPS spectrum of N 1s.

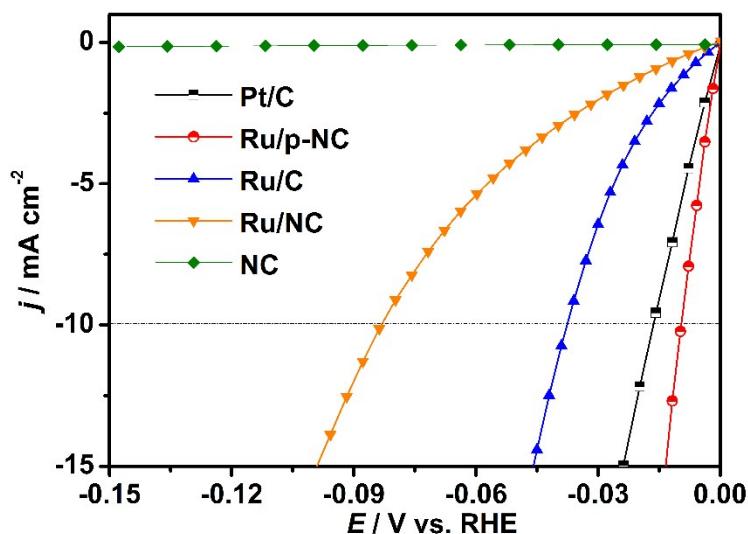


Fig. S9 A enlarged view of the HER polarization curves of Ru/p-NC, Ru/NC, Pt/C (20 wt %) and Ru/C (5 wt%) in 1.0 M KOH at a scan rate of 5 mV s⁻¹.

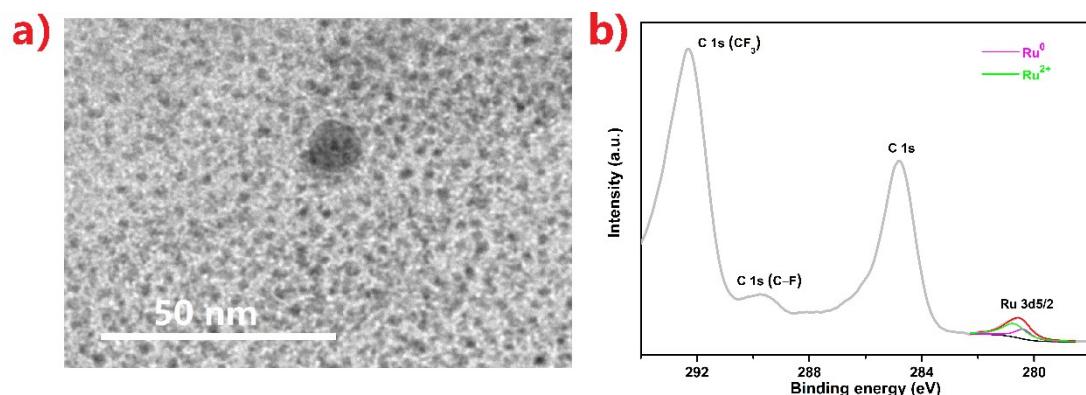


Fig. S10 a) TEM image and b) High-resolution XPS spectrum of Ru/p-NC after the test of long-term stability.

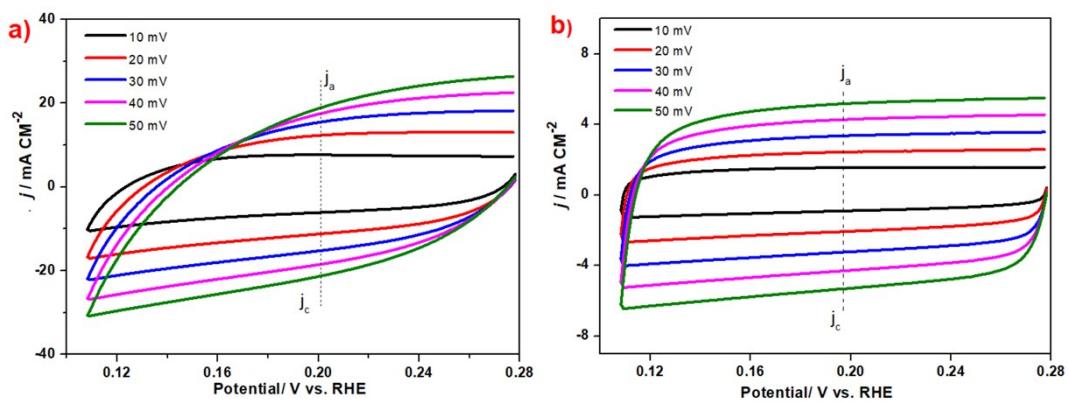


Fig. S11 Cyclic voltammetry curves of a) Ru/p-NC and b) Ru/NC.

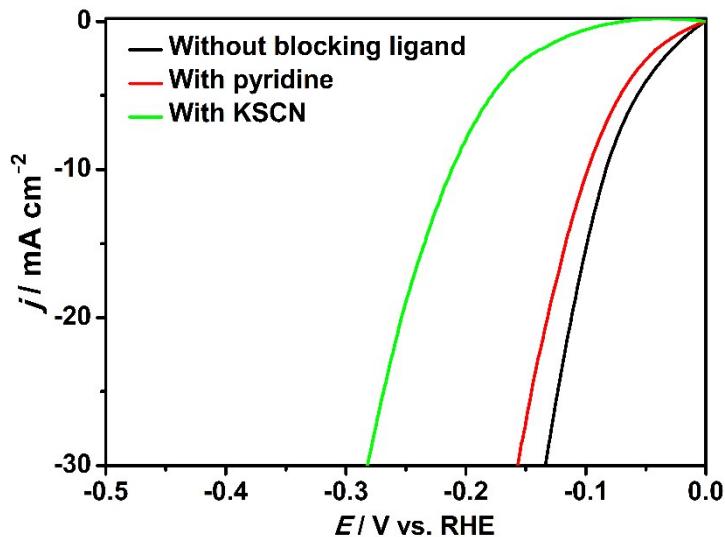


Fig. S12 HER polarization curves of the Ru/NC electrocatalyst with and without blocking ligands in 1 M KOH solution.

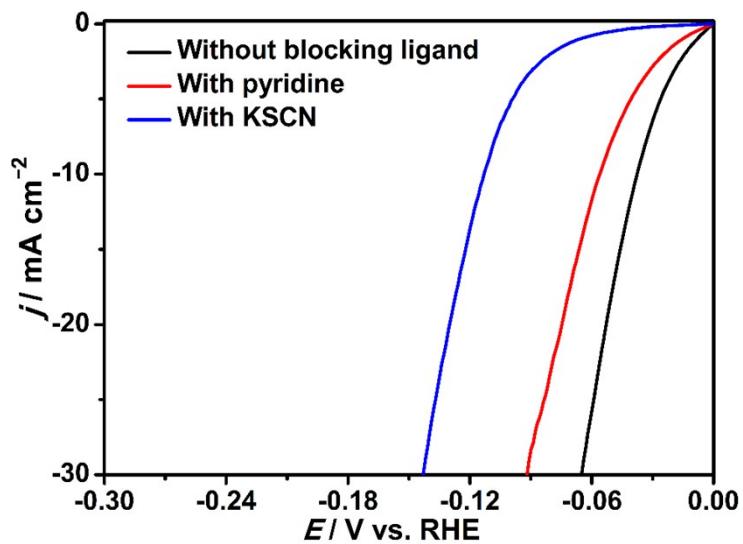


Fig. S13 HER polarization curves of the Ru/C electrocatalyst with and without blocking ligands in 1 M KOH solution.

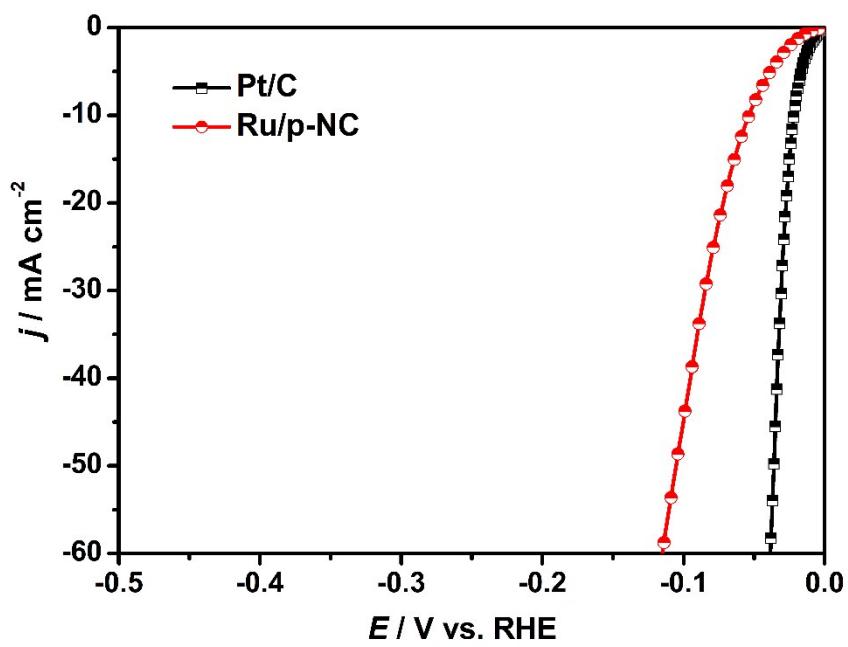


Fig. S14 HER polarization curves of the Ru/p-NC and Pt/C in 0.5 M H_2SO_4 .

Table S1 Charge transfer resistances (R_{ct}) obtained by Nyquist plots in 1.0 M KOH at an applied overpotential of 10 mV.

Catalyst	R_{ct} / Ω
Ru/p-NC	4.7
Ru/NC	35.5
Ru/C	20.0
Pt/C	9.5

Table S2 Summary of HER performance of Ru-based catalysts in alkaline condition.

Catalyst	η_{10} / mV	Tafel slope / mV dec^{-1}	Mass activity/ $\text{A mg}_{\text{Ru}}^{-1}$	TOF/ $\text{H}_2 \text{ s}^{-1}$	R_{ct}	Reference
Ru/p-NC	10	17	17 A $\text{mg}_{\text{Ru}}^{-1}$ at 25 mV	8.9 at 25 mV	4.7 Ω	This work
Ru@C ₂ N	17	38	—	0.76 at 25mV, 1.66 at 50mV	43.7 Ω	1
Ru@GnP	22	28	0.23A $\text{mg}_{\text{Ru}}^{-1}$ at 25mV	—	—	2
Ru@NC	26	36	17A $\text{mg}_{\text{Ru}}^{-1}$ at 100 mV	10.8 at 100 mV	—	3
Ru@MW CNT	17	27	0.186 A $\text{mg}_{\text{Ru}}^{-1}$ at 20 mV	0.4 at 25 mV	2.38 Ω cm ² at 45mV	4
Ru/NC/G F	21	31	8A $\text{mg}_{\text{Ru}}^{-1}$ at 100 mV	4.55 at 100 mV	5 Ω	5
Ru@C ₄ N	7	18	—	0.65 at 25 mV	0.9478 Ω	6
Ru-HPC	—	29	—	1.79 at 25 mV 9.2 at 100 mV	—	7
Ru/C-300	14	32	—	—	—	8
Ru/CN-800	14	30	—	—	13.7 Ω	9
Ru@NCH NSs	28	32	0.1A $\text{mg}_{\text{Ru}}^{-1}$ at 38.2mV	—	—	10

Ru-NC-700	12	14	—	—	20.7Ω	11
Ru-MoO ₂	31	29	—	—	—	12
Ru@CN-0.16	32	53	—	—	—	13
RuCo@N C	28	31	—	—	8.58Ω	14
RuP ₂ @NP C	52	69	—	—	—	15
RuNi/CQ Ds-600	13	40	$1.68\text{A mg}_{\text{Ru}}^{-1}$ at 13mV	—	6.45Ω	16
2DPC-RuMo	18	25	—	3.57 at 50 mV	0.9688 Ω	17

References

- (1) J. Mahmood, F. Li, S.-M. Jung, M. S. Okyay, I. Ahmad, S.-J. Kim, N. Park, H. Y. Jeong and J.-B. Baek, *Nat. Nanotechnol.* 2017, **12**, 441-446.
- (2) F. Li, G.-F. Han, H.-J. Noh, I. Ahmad, I.-Y. Jeon and J.-B. Baek, *Adv. Mater.*, 2018, **30**, 1801741.
- (3) Z.-L. Wang, K. Sun, J. Henzie, X. Hao, C. Li, T. Takei, Y.-M. Kang and Y. Yamauchi, *Angew. Chem., Int. Ed.*, 2018, **57**, 5848-5852.
- (4) D. H. Kweon, M. S. Okyay, S.-J. Kim, J.-P. Jeon, H.-J. Noh, N. Park, J. Mahmood and J.-B. Baek, *Nat. Commun.*, 2020, **11**, 1278.
- (5) J. Zhang, P. Liu, G. Wang, P. P. Zhang, X. D. Zhuang, M. W. Chen, I. M. Weidinger and X. L. Feng, *J. Mater. Chem. A*, 2017, **5**, 25314-25318.
- (6) S.-W. Sun, G.-F. Wang, Y. Zhou, F.-B. Wang and X.-H. Xia, *Appl. Mater. Interfaces*, 2019, **11**, 19176-19182.
- (7) T. Qiu, Z. Liang, W. Guo, S. Gao, C. Qu, H. Tabassum, H. Zhang, B. Zhu, R. Zou and Y. Shao-Horn, *Nano Energy*, 2019, **58**, 1-10.
- (8) C. Xu, M. Ming, Q. Wang, C. Yang, G. Fan, Y. Wang, D. Gao, J. Bi and Y. J. Zhang, *J. Mater. Chem. A*, 2018, **6**, 14380-14386.
- (9) X. Cheng, H. Wang, M. Ming, W. Luo, Y. Wang, Y. Yang, Y. Zhang, D. Gao, J. Bi

- and G. Fan, *ACS Sustain. Chem. Eng.*, 2018, **6**, 11487-11492.
- (10) J.-S. Li, M.-J. Huang, X.-N. Chen, L.-X. Kong, Y.-W. Zhou, M.-Y. Wang, J.-L. Li, Z.-X. Wu and X.-F. Xu, *Chem. Comm.*, 2020, **56**, 6802-6805.
- (11) B. Lu, L. Guo, F. Wu, Y. Peng, J. E. Lu, T. J. Smart, N. Wang, Y. Z. Finfrock, D. Morris, P. Zhang, N. Li, P. Gao, Y. Ping and S. Chen, *Nat. Commun.*, 2019, **10**, 631.
- (12) P. Jiang, Y. Yang, R. Shi, G. Xia, J. Chen, J. Su and Q. Chen, *J. Mater. Chem. A*, 2017, **5**, 5475-5485.
- (13) J. Wang, Z. Wei, S. Mao, H. Li and Y. Wang, *Energy Environ. Sci.*, 2018, **11**, 800-806.
- (14) J. Su, Y. Yang, G. Xia, J. Chen, P. Jiang and Q. Chen, *Nat. Commun.*, 2017, **8**, 14969.
- (15) Z. Pu, I. S. Amiinu, Z. Kou, W. Li and S. Mu, *Angew. Chem., Int. Ed.*, 2017, **56**, 11559-11564.
- (16) Y. Liu, X. Li, Q. Zhang, W. Li, Y. Xie, H. Liu, L. Shang, Z. Liu, Z. Chen, L. Gu, Z. Tang, T. Zhang, and S. Lu, *Angew. Chem. Int. Ed.*, 2020, **59**, 1718 – 1726.
- (17) K. Tu, D. Tranca, F. Rodríguez-Hernández, K. Jiang, S. Huang, Q. Zheng, M. Chen, C. Lu, Y. Su, Z. Chen, H. Mao, C. Yang, J. Jiang, H. Liang and X. Zhuang, *Adv. Mater.*, 2020, **32**, 2005433.