

## NiO addition effects on Ba(Zr, Ce, Y)O<sub>3-δ</sub>: sintering behavior and proton uptake

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### Supplementary Information

Table S1: Composition of pellets sintered by SSRS, and SPS with Ni on B site. SSRS samples were sintered at 1550 °C 4 h. SPS samples were sintered at 1350 °C and 1300 °C for 1 at% and 4 at% Ni and annealed at 1400 °C 4 h.

Nominal Composition	Measurement by ICP-OES / wt. %					Real composition	A: B ratio
	Ba	Zr	Ce	Y	Ni		
BZC20Y13.6-0.125Ni	48.5 ±0.5	20.7 ±0.3	9.61 ±0.1	4.12 ±0.05	0.126 ±0.009	Ba <sub>1.02</sub> Zr <sub>0.66</sub> Ce <sub>0.20</sub> Y <sub>0.13</sub> Hf <sub>0.01</sub> O <sub>3-δ</sub> + 0.16 wt. % NiO	1.02
BZC20Y13.6-0.25Ni	48.4 ±0.5	20.5 ±0.3	9.69 ±0.1	4.11 ±0.05	0.163 ±0.01	Ba <sub>1.03</sub> Zr <sub>0.66</sub> Ce <sub>0.20</sub> Y <sub>0.13</sub> Hf <sub>0.01</sub> O <sub>3-δ</sub> + 0.21 wt. % NiO	1.03
BZC20Y13.6-0.375Ni	48.2 ±0.5	21.5 ±0.3	9.02 ±0.1	4.59 ±0.06	0.344 ±0.01	Ba <sub>0.99</sub> Zr <sub>0.67</sub> Ce <sub>0.18</sub> Y <sub>0.15</sub> Hf <sub>0.01</sub> O <sub>3-δ</sub> + 0.44 wt. % NiO	0.99
BZC20Y13.6-0.5Ni	48.2 ±0.5	20.6 ±0.3	9.55 ±0.1	4.01 ±0.09	0.35 ±0.03	Ba <sub>1.03</sub> Zr <sub>0.66</sub> Ce <sub>0.20</sub> Y <sub>0.13</sub> Hf <sub>0.01</sub> O <sub>3-δ</sub> + 0.45 wt. % NiO	1.03
BZC20Y13.6-1.0Ni	48.2 ±0.5	20.7 ±0.3	9.78 ±0.1	4.01 ±0.05	0.41 ±0.03	Ba <sub>1.01</sub> Zr <sub>0.66</sub> Ce <sub>0.20</sub> Y <sub>0.13</sub> Hf <sub>0.01</sub> O <sub>3-δ</sub> + 0.52 wt. % NiO	1.02
BZC20Y13.6Ni1-SPS	48.0 ±0.5	20.6 ±0.3	9.85 ±0.1	4.56 ±0.09	0.25 ±0.01	Ba <sub>0.99</sub> Zr <sub>0.64</sub> Ce <sub>0.20</sub> Y <sub>0.14</sub> Ni <sub>0.01</sub> Hf <sub>0.01</sub> O <sub>3-δ</sub>	0.99
BZC20Y13.6Ni4-SPS	47.8 ±0.5	20.1 ±0.3	9.88 ±0.1	4.47 ±0.09	0.56 ±0.04	Ba <sub>0.99</sub> Zr <sub>0.62</sub> Ce <sub>0.20</sub> Y <sub>0.14</sub> Ni <sub>0.03</sub> Hf <sub>0.01</sub> O <sub>3-δ</sub>	0.99

Table S2: Composition of BZCY with 1.0 wt% NiO addition sintered at 1550 °C for different time.

Sintering time / h	Measurement by ICP-OES / wt. %					Real composition	A: B ratio
	Ba	Zr	Ce	Y	Ni		
1	48.2	21.6	9.68	4.19	0.487	Ba <sub>0.99</sub> Zr <sub>0.67</sub> Ce <sub>0.19</sub> Y <sub>0.13</sub> Hf <sub>0.01</sub> O <sub>3-δ</sub>	0.99
	±0.5	±0.3	±0.1	±0.06	±0.005	+ 0.62 wt. % NiO	
4	48.2	20.7	9.78	4.01	0.41	Ba <sub>1.02</sub> Zr <sub>0.66</sub> Ce <sub>0.20</sub> Y <sub>0.13</sub> Hf <sub>0.01</sub> O <sub>3-δ</sub>	1.02
	±0.5	±0.3	±0.1	±0.05	±0.03	+ 0.52 wt. % NiO	
16	48.1	20.8	9.67	4.19	0.377	Ba <sub>1.01</sub> Zr <sub>0.66</sub> Ce <sub>0.20</sub> Y <sub>0.14</sub> Hf <sub>0.01</sub> O <sub>3-δ</sub>	1.01
	±0.5	±0.3	±0.1	±0.17	±0.007	+ 0.48 wt. % NiO	

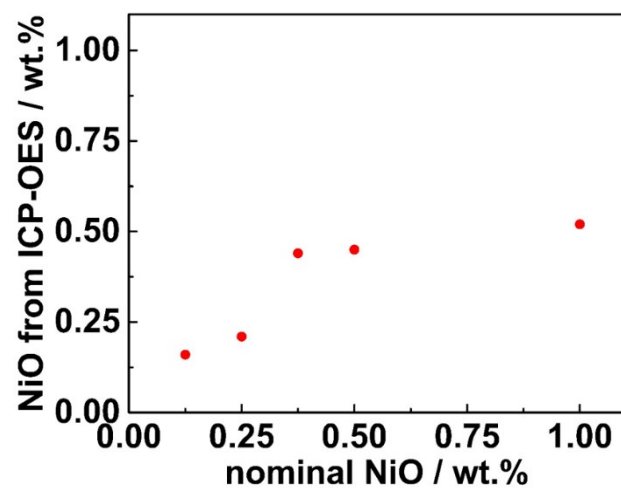


Figure S1: Actual NiO content of SSRS samples sintered at 1550 °C 4h from ICP-OES.

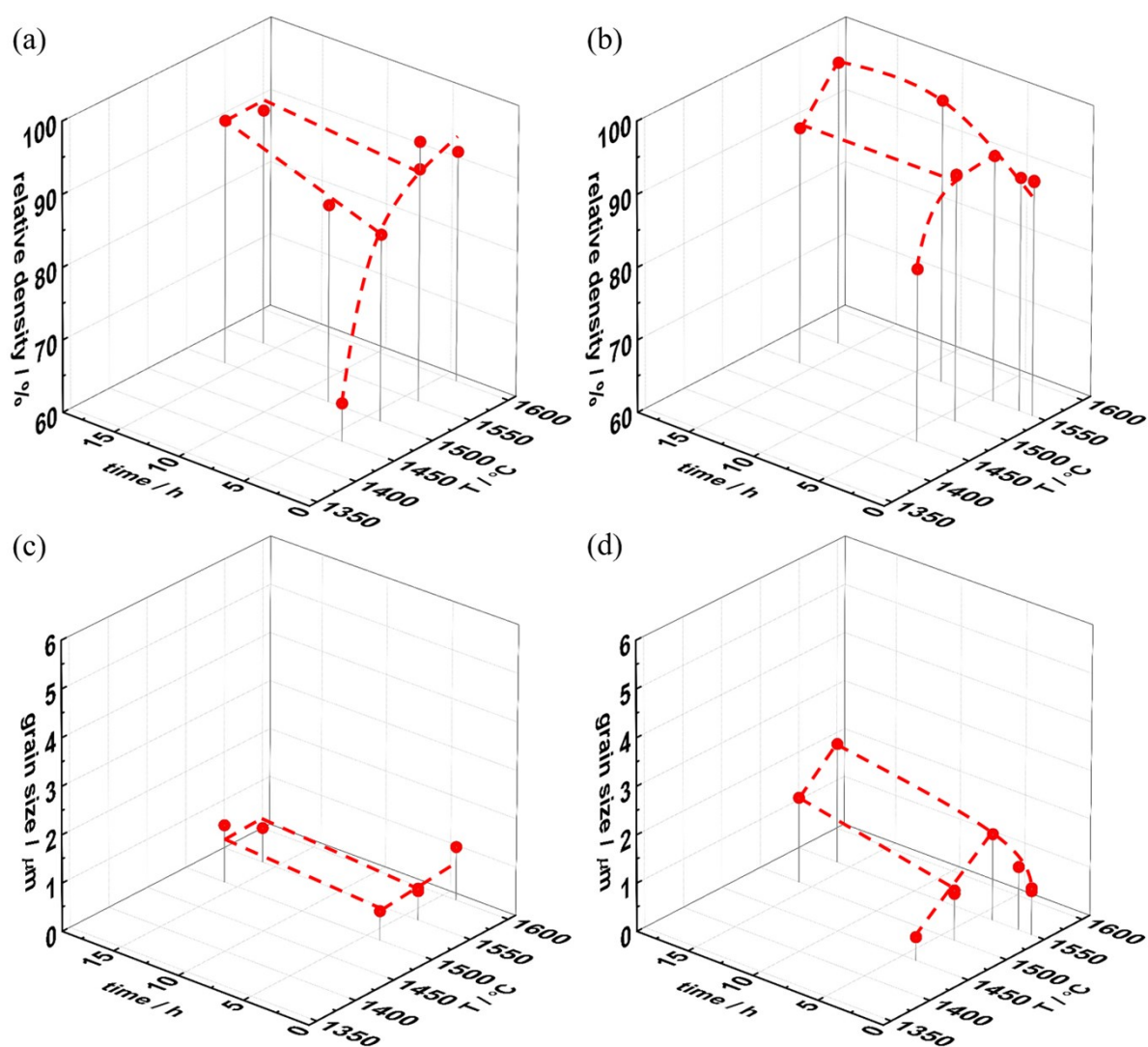


Figure S2: Relative density (top) and grain size (bottom) as function of sintering temperature and soaking time with (a,c) 0.125 wt% NiO, and (b,d) 0.375 wt% NiO. The dashed lines are only a guide to the eye.

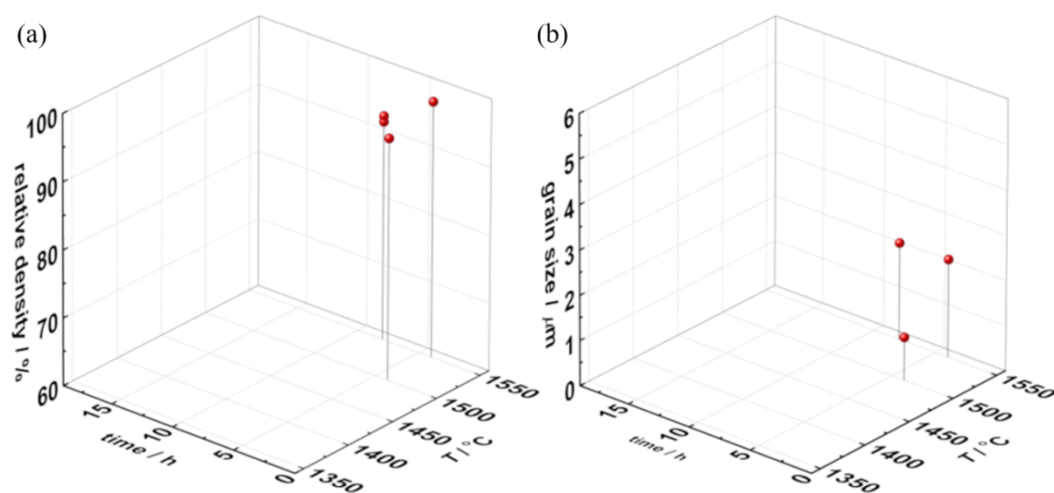


Figure S3: (a) relative density and (b) grain size as function of sintering temperature and soaking time for sample doped with 17.5 at% Y and 0.5 wt% NiO addition.

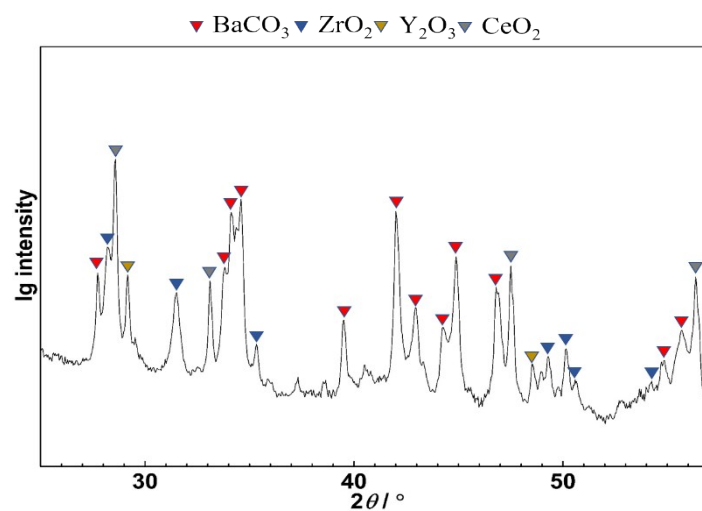


Figure S4: XRD pattern of the wet ball milled powder from the sample with 2.0 wt% NiO addition.

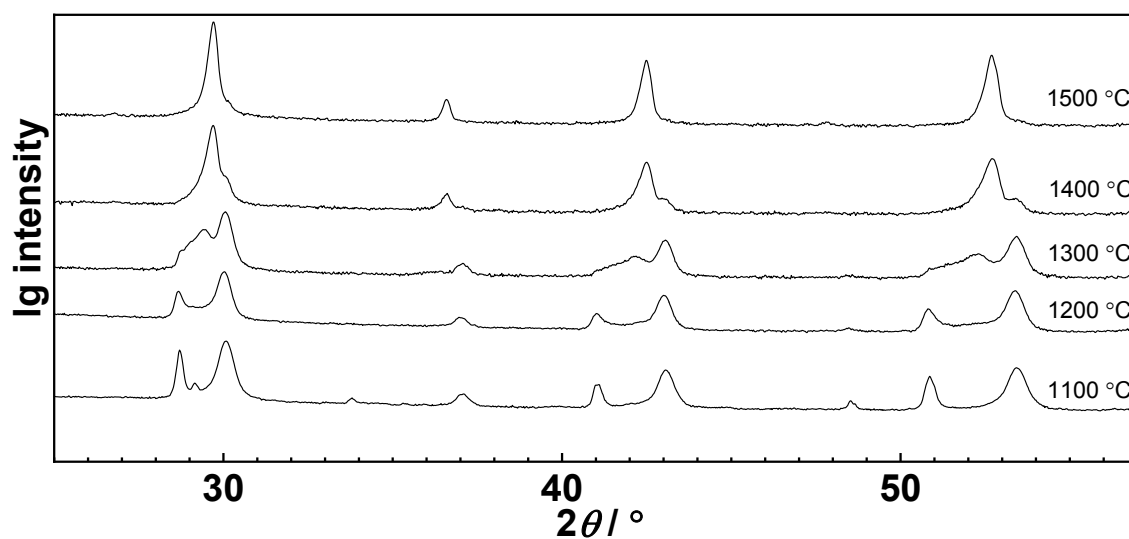


Figure S5: XRD patterns of BZCY with 0.5 wt% NiO addition sintered at different  $T$  for 10 min.

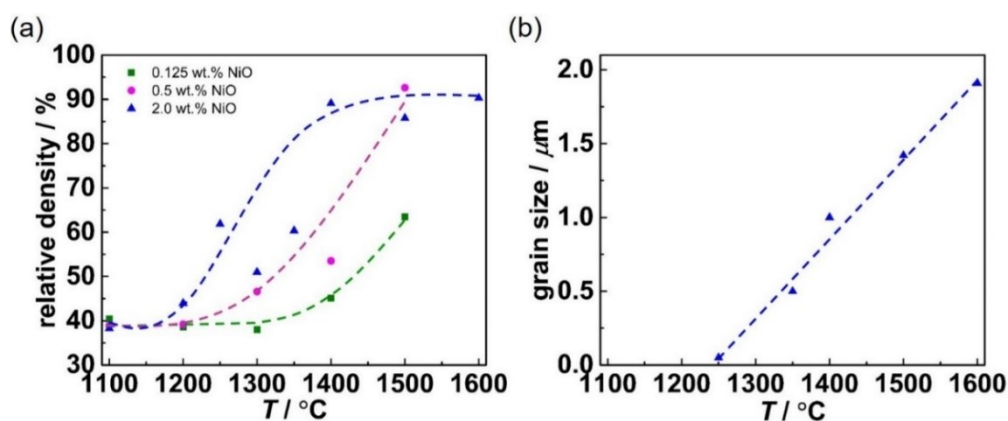


Figure S6: Evolution of grain size and density with temperature for the quenched samples of Fig. 6-7.

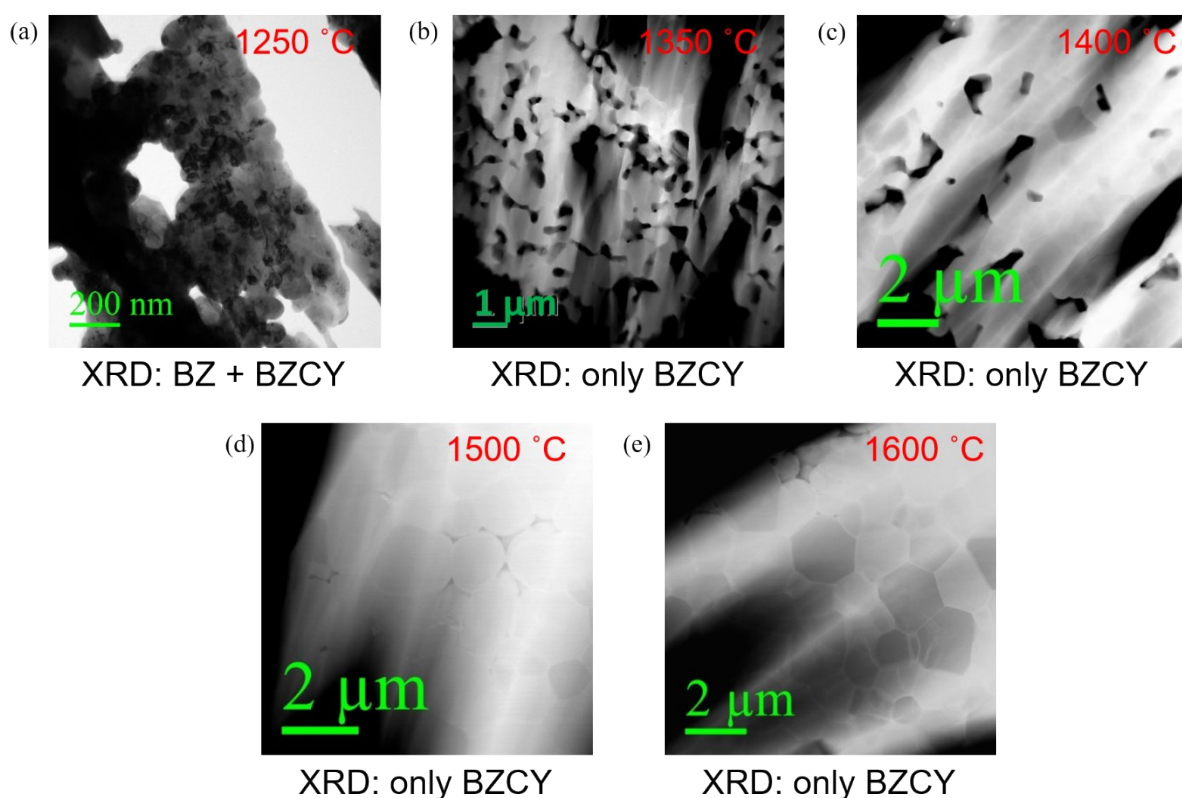


Figure S7: (a) BF image from 1250 °C and (b-e) HAADF images from quenched BZCY with 2.0 wt% NiO addition samples sintered for 10 min at different temperatures.

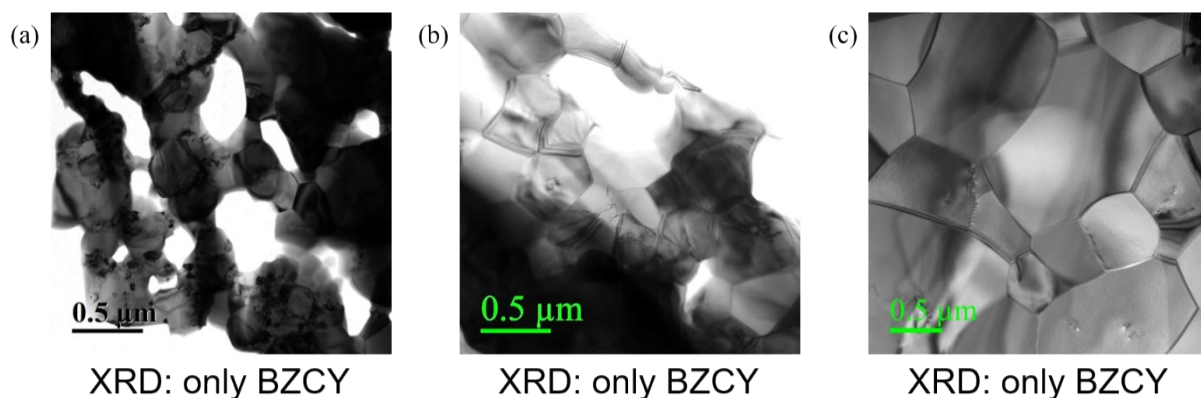


Figure S8: BF images from quenched BZCY with 2.0 wt% NiO addition samples sintered at 1350 °C for (a) 10 min, (b) 30 min and (c) 4 h.

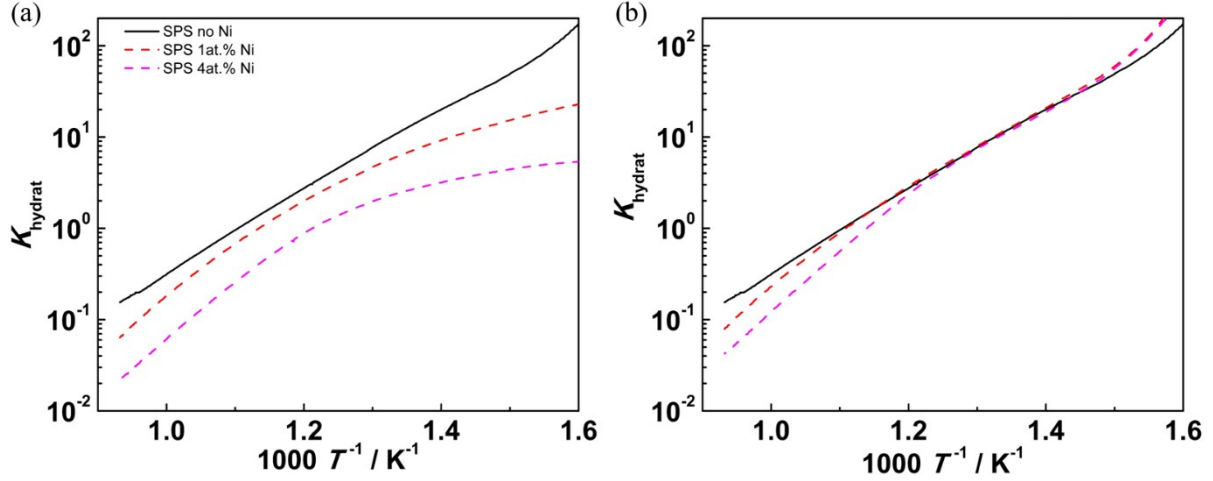


Figure S9: Van't Hoff plots for Ni-free and substitutional-Ni samples, (a)  $K_{hydrat}$  calculated with nominal acceptor concentration. (b) Calculated with effective acceptor concentration.

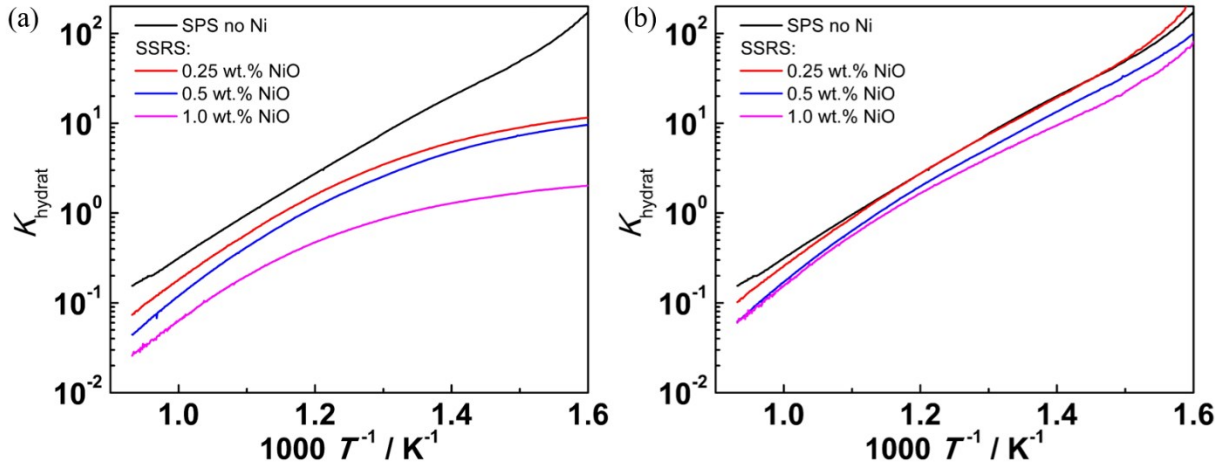


Figure S10: Van't Hoff plots with effective dopant concentration for different excess NiO addition. (a)  $K_{hydrat}$  calculated with nominal acceptor concentration. (b) Calculated with effective acceptor concentration; the linear fitting temperature range for  $\Delta H^\circ$  and  $\Delta S^\circ$  is 350 – 800 °C.

Note that for the Ni-excess samples in Fig. S10a the bending starts already at much higher temperatures (below  $\approx 570$  °C) compared to the samples with substitutional Ni in Fig. S9a (only below  $\approx 420$  °C). The stronger bending corresponds to lower effective acceptor concentrations, cf. Fig. 10c in the main paper.

**Procedure for  $[A'_{eff}]$  determination:** The plateau which the proton uptake approaches at 300°C (Fig. 10a,b) gives already a rough estimate of  $[A'_{eff}]$  (when the proton concentration approaches saturation,  $[OH_O^\bullet] = [A'_{eff}]$  holds). The actual  $[A'_{eff}]$  values of Fig. 10c were determined from this first estimate and the condition that the van 't Hoff plot should become as straight as possible. The use of a modified electroneutrality condition with  $[A'_{eff}]$

$$2[V_O^{\oplus}] + [OH_O^\bullet] = [A'_{eff}]$$

yields a modified expression for the mass action constant:

$$K_{hydrat} = \frac{2[\text{OH}_\text{O}^\text{g}]^2}{p\text{H}_2\text{O}([\text{A}'_{\text{eff}}][\text{OH}_\text{O}^\text{g}](3-[\text{A}'_{\text{eff}}]/2-[\text{OH}_\text{O}^\text{g}])}$$

The  $\Delta H^0$  and  $\Delta S^0$  values of Fig. 11 in the main paper were determined by fitting the van 't Hoff plots (employing  $[\text{A}'_{\text{eff}}]$ ) in the range of 350-800 °C. Since the van 't Hoff plots (Fig. S9,10) have some slight deviations from perfect linearity, choosing a different  $T$  range for the fitting slightly changes also the  $\Delta H^0$  and  $\Delta S^0$  values. Table S3 compares the results for the 350-800 °C and 500-800 °C fitting ranges. While the high- $T$  range leads to systematically more negative  $\Delta H^0$  and  $\Delta S^0$  values, the trends displayed in Fig. 11 of the main paper remain unaffected. The choice of the fitting range is always a balance; at high  $T$   $K_{hydrat}$  is potentially more affected by uncertainties in the determination of the absolute proton concentration from the dry/humid switching, while at lower  $T$  (high  $K_{hydrat}$ ) the uncertainty in  $[\text{A}'_{\text{eff}}]$  may have a stronger effect. Thus it is most important to treat all samples on a equal basis as done here.

Table S3: Comparison of  $\Delta H^0$  and  $\Delta S^0$  values obtained by different fitting ranges of the van 't Hoff plots

sample	$\Delta H^0$ / kJ/mol	$\Delta S^0$ / J/molK	$\Delta H^0$ / kJ/mol	$\Delta S^0$ / J/molK
	fitted in 350-800 °C range Fig. 11		fitted in 500-800 °C range	
SPS no Ni	-85	-94	-89	-99
SSRS 0.25wt% NiO	-88	-98	-98	-109
SSRS 0.5wt% NiO	-91	-105	-101	-116
SSRS 1wt% NiO	-88	-99	-97	-112
SPS Ni on B 0.25wt%	-94	-106	-108	-119
SPS Ni on B site 1wt%	-105	-121	-121	-138