

Supporting Information

Unoccupied 3d orbitals makes Li-unalloyable transition metals usable as anode materials for lithium ion batteries

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Tables S1-S2

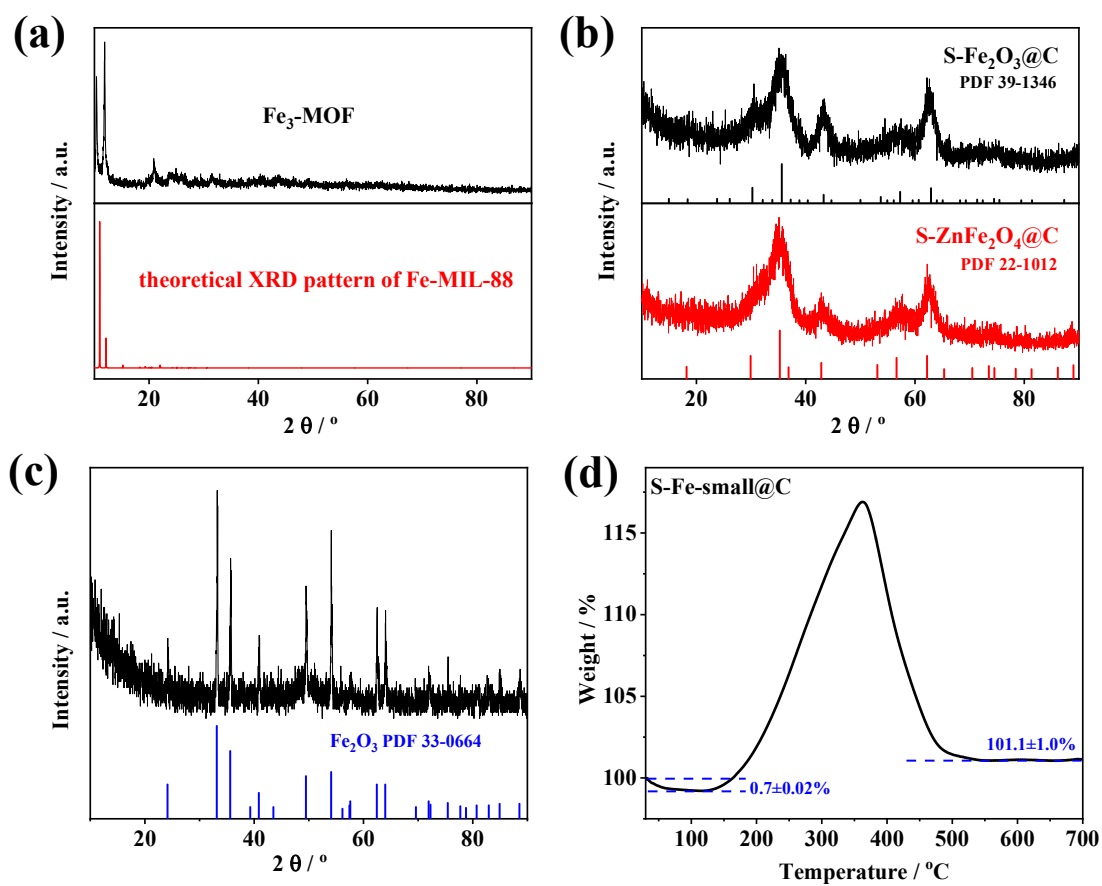


Figure S1. (a) XRD pattern of the Fe₃-MOF and standard XRD pattern of Fe-MIL-88. (b) XRD pattern of S-Fe₂O₃@C and S-ZnFe₂O₄@C. (c) XRD pattern of S-Fe@C after TG. (d) TG curve of the S-Fe-small@C.

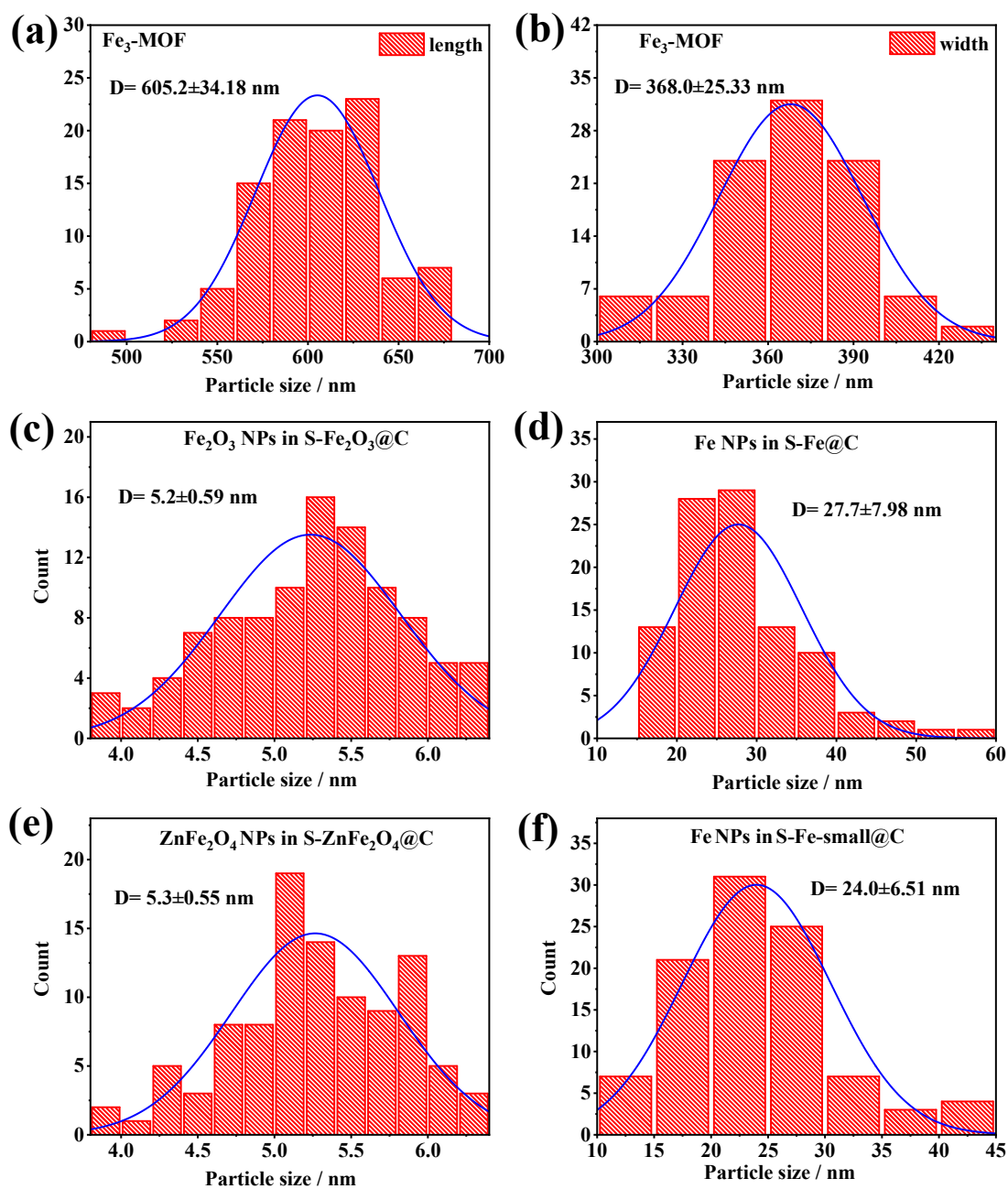


Figure S2. Particle size distribution of (a) and (b) $\text{Fe}_3\text{-MOF}$, (c) Fe_2O_3 NPs in the $\text{S-Fe}_2\text{O}_3\text{@C}$, (d) Fe NPs in the S-Fe@C , (e) ZnFe_2O_4 NPs in the $\text{S-ZnFe}_2\text{O}_4\text{@C}$ and (f) Fe NPs in S-Fe-small@C .

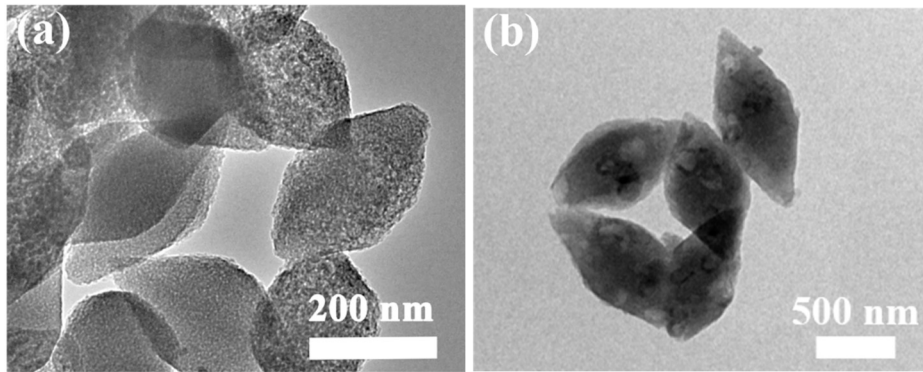
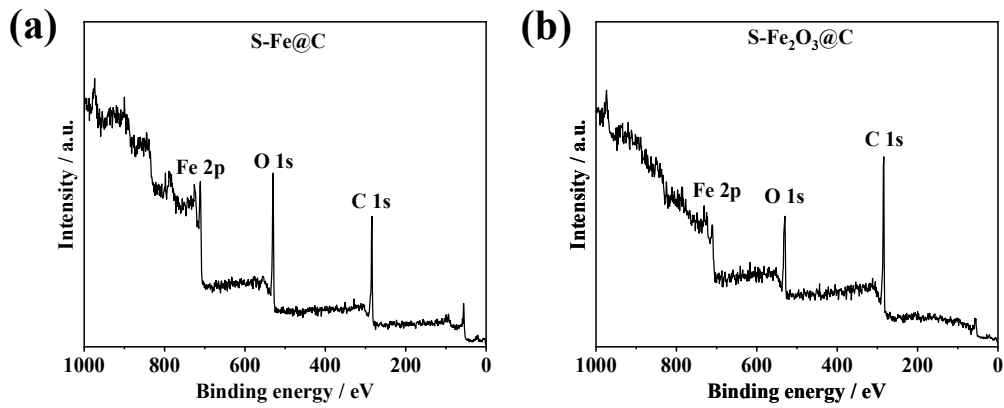


Figure S3. TEM images of (a) the porous carbon obtained by the removal of the Fe_2O_3 nanoparticles from the $\text{S-Fe}_2\text{O}_3@\text{C}$, and (b) the S-carbon synthesized by the removal of Fe from the $\text{S-Fe}@\text{C}$ through the HCl etching.



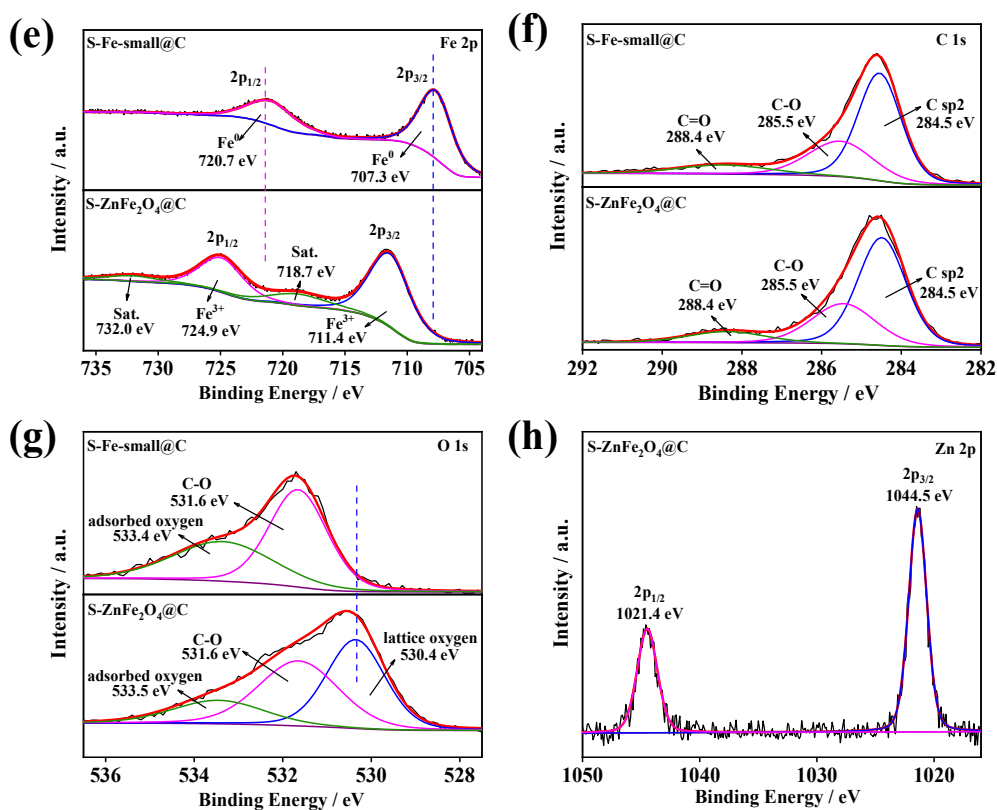


Figure S4. XPS survey spectra of (a) S-Fe@C, (b) S-Fe₂O₃@C, (c) S-ZnFe₂O₄@C and (d) S-Fe-small@C. High-resolution XPS spectra of (e) Fe 2p, (f) C 1s, and (g) O 1s for S-ZnFe₂O₄@C and S-Fe-small@C. (h) High-resolution XPS spectra of Zn 2p for S-ZnFe₂O₄@C.

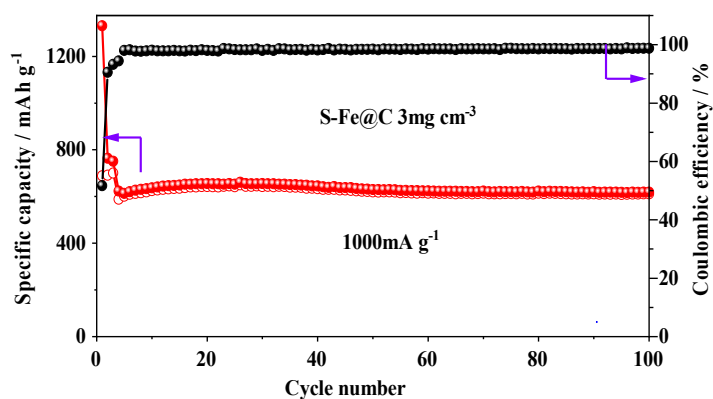


Figure S5. Cycling performance and coulombic efficiencies of S-Fe@C with a mass loading $\sim 3.0 \text{ mg cm}^{-2}$ at 1.0 A g^{-1} .

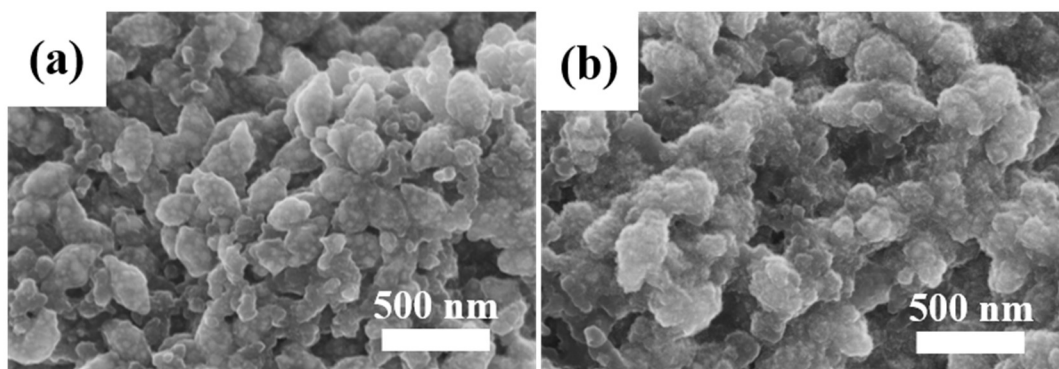


Figure S6. SEM image of S-Fe@C in the presence of carbon black and PVDF binders before and after 50 cycles at 1.0 A g⁻¹. The presence of carbon black and PVDF binders makes the profile of the particles less distinct.

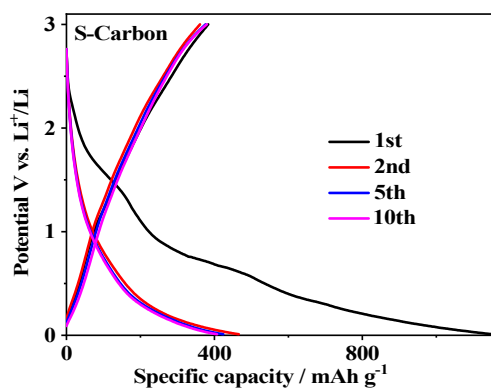


Figure S7. Charge/discharge voltage profiles of S-carbon at 100 mA g⁻¹.

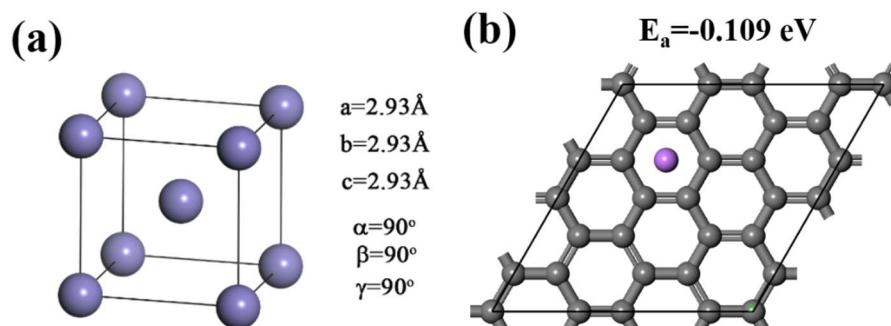


Figure S8. (a) Lattice constants of the cubic structure Fe. (b) Adsorption of the Li atom on a single layer graphene.

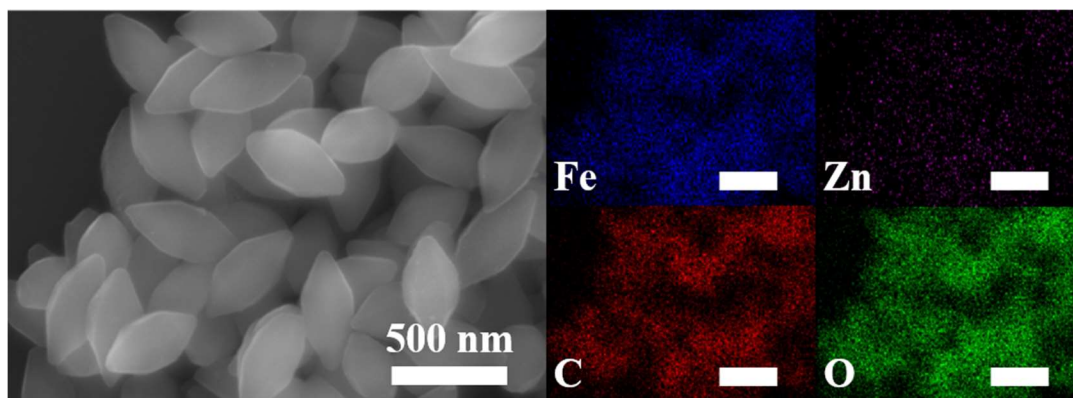


Figure S9. Elemental mapping images of S-ZnFe₂O₄@C.

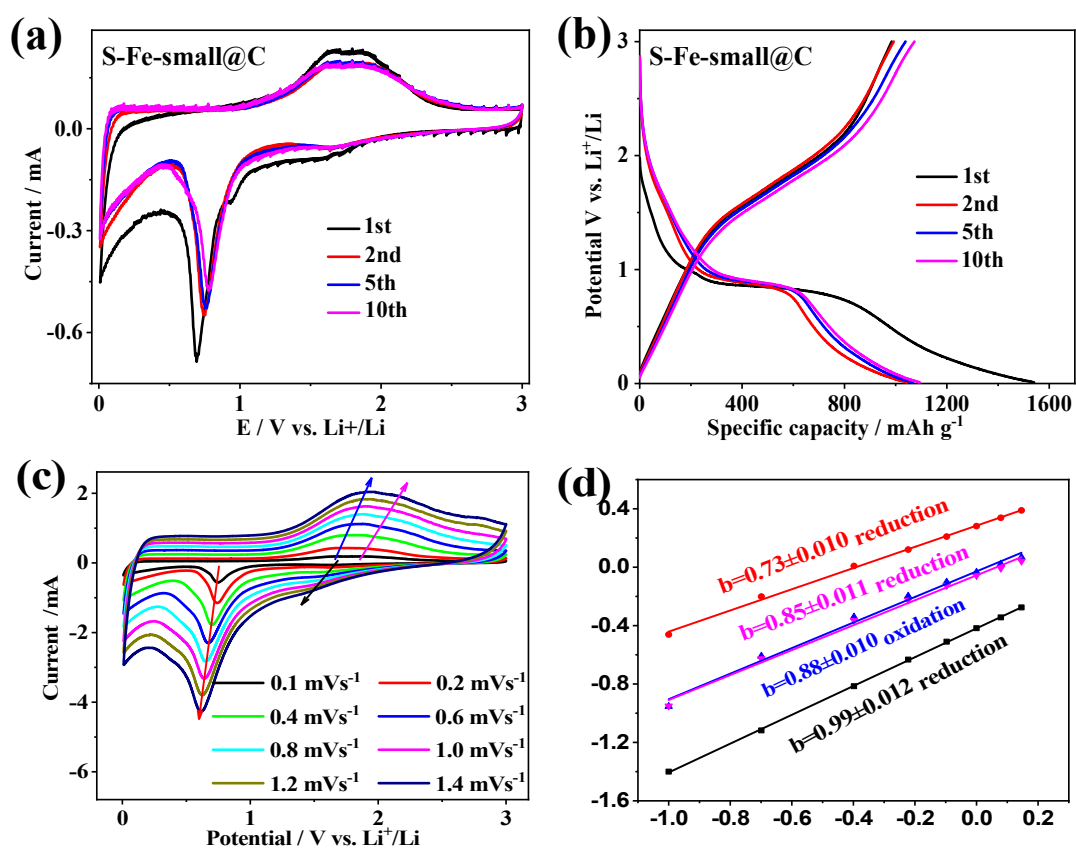


Figure S10. (a) CVs of S-Fe-small@C at 0.1 mV s⁻¹. (b) Charge/discharge voltage profiles of (b) S-Fe-small@C. (c) CVs of S-Fe-small@C at different scan rates. (d) b-value of S-Fe-small@C.

Table S1. Stable reversible capacity comparison of the S-Fe@C and S-Fe-small@C with other anode materials for LIBs.

Electrode material	Mass loading mg cm ⁻²	Current density (mA g ⁻¹)	Cycles	Capacity (mAh g ⁻¹)	Reference
S-Fe@C	~1.0	100	100	967.3±16.2	This work
		1000	1000	700.4±10.5	
S-Fe-small@C	~3.0	1000	100	611.0±10.5	This work
		100	100	1091.9±17.6	
Sb@C	~1.0	1000	1000	925.1±13.7	1
Sn@aCT		100	350	870	
TiO ₂ -Sn@CNFs	1.5	200	1100	643	2
Sn-Sb micro/nano-structures	1.0	100	100	751	3
Sb/Sb ₂ O ₃ /CNT/GNR	0.9-1.1	50	100	619	4
SnSe/SnO _x @CNFs	1.73	200	70	740.7	5
Sb ₂ Se ₃	1.1	1000	1000	260.8	6
SnSe _{0.5} S _{0.5} /C	1.0	200	150	785	7
Bi ₂ S ₃ @C NW	1.0-1.5	1000	700	501	8
Bi ₂ S ₃ @Co ₉ S ₈	0.9-1.2	100	100	798	9
Fe-MIL-88 B	1	60	400	744.5	10
Fe-BTC	2	1000	400	408.8	11
MIL-53(Fe) @RGO	0.75	100	100	550	12
Mg-MOF-74/Cu	1.0-1.5	500	300	531.7	13
Fe ₂ O ₃ @PAN	0.6-0.8	1000	500	506.6	14
Fe ₃ O ₄ /C@VO _x	1.2	1000	500	845	15
C@Fe ₃ C/Fe	1.5	2000	1000	392	16
Fe ₃ N@C	0.6-1.2	100	500	358	17
Ge ₃ N ₄ @C	3.5	550	200	702	18
N-Ti ₃ C ₂ /Fe ₂ O ₃	1.5	2000	400	549	19
Co ₃ O ₄ @Co ₃ V ₂ O ₈	0.8	100	100	948	20
ZnMn ₂ O ₄	1.0	200	100	867	21
Zn ₂ SiO ₄ @NC	2.0	1000	400	540	22
NiS _x @C	1.0	1000	2000	460	23
CoS ₂ /CNTs/graphene	2.0	100	100	368.2	24
NHMCFs/MoSe ₂		1000	400	582.5	25
ZnS/NC	1.5-2.0	200	100	757	26
Ni ₂ P/Ni	1.5	1000	1000	597	27
Ni ₂ Pp⊂GN		100	250	511	28
FeS ₂ /Fe ₇ S ₈ -rGO	1.0	200	250	650	29
GeP ₃ /C	1.3	1000	1000	312	30
CoP@C		1000	800	500	31
MoS ₂ /Mo ₂ TiC ₂ Tx		100	100	509	32
(FeCoNiCrMn) ₃ O ₄	~3.0	500	300	402	33
Fe-BTC	2.0	1000	400	408.8	34
Zn ₂ SiO ₄ @NC	2.0	1000	400	540	23
CoS ₂ /CNTs/graphene	2.0	100	100	368.2	25
VN nanosheet	2.0	100	100	520	35

Table S2. Rate performance comparison of the S-Fe@C and S-Fe-small@C with other anode materials for LIBs.

Electrode material	Mass loading mg cm ⁻²	0.1 A g ⁻¹	0.2 A g ⁻¹	0.5 A g ⁻¹	1.0 A g ⁻¹	2.0 A g ⁻¹	5.0 A g ⁻¹	10.0 A g ⁻¹	Reference
S-Fe@C	~1.0	851.0 ±12.9	825.9 ±12.5	809.0 ±12.1	762.6 ±11.4	666.8 ±10.3	534.4 ±8.3	410.1 ±6.6	This work
S-Fe-small@C	~1.0	1092.9 ±17.9	1046.2 ±16.5	971.9 ±16.1	900.2 ±14.3	815.7 ±12.2	659.5 ±9.8	490.6 ±7.7	This work
Sb@C		623	558	496	439	385			1
TiO ₂ -Sn@CNFs			570		280				3
Sn-Sb micro/nano structures	1.0	798.8	717.8	636.5	545.5	455.0			4
Sb/Sb ₂ O ₃ /CNT/GNR	0.9-1.1	642	514	428	369	327			5
SnO ₂ @SnS ₂ @NG		898	820	715	612	497	343		36
SnSe/SnOx@CNFs	1.73	803	709	579	391	301			6
Sb ₂ Se ₃	1.1	638.2	611.5	543.6	472.3	389.5			7
SnSe _{0.5} S _{0.5} /C	1.0	989	830	729	646	553	389		8
Bi ₂ S ₃ @Co ₉ S ₈	0.9-1.2	785	672	612	573	512			10
MoS ₂ @SnS-QDs/CNN	1.0-1.2	1130	1078	947	765	591			37
C@Fe ₃ C/Fe	1.5	1177	1012	954	845	682	521		17
Co ₃ O ₄ @Co ₃ V ₂ O ₈	0.8	1068	916	782	678	578	550		21
PCN-600(iron porphyrin-based MOF)	0.5				625	470			38
Fe-MIL-88 B	1	692.0	582.6	350.4	232.7	143.5	133.4		11
Fe-BTC	2	873.8 /915.7	990.1 /996.3		523.2 /525.0	302.8 /304.8			12
MIL-53(Fe)@RGO	0.75	510		400	360	300			13
Mg-MOF-74/Cu	1.0-1.5	796.2		560.5	372.6	259.1			14
NiO@N-C		1065	1031		689	634			39
Fe ₃ O ₄ /C@VO _x	1.2		810	718	605	483	340		16
Fe ₂ N@C	1.4-2.1	567	526	500	474	450	404	356	40
Fe ₃ O ₄ nanoparticle		1084	883	809	648	545	410		41
ZnMn ₂ O ₄	1.0	857	784	636	481	355	94	36	22
Zn ₂ SiO ₄ @NC	2.0	640	620	560	450	370	280		23
NiS _x @C	1.0	790	720	605	540	446	350	280	24
CoS ₂ /CNTs/graphene	2.0		381			251	212		25
NHMCFs/MoSe ₂		666.4	635.3	352.9				244.3	26
ZnS/NC	1.5-2.0	771	725		600	511			27
MoS ₂ /Mo ₂ TiC ₂ Tx		523	484	407	315	182			33
NiCoPS ₃ /NC	0.9-1.2	976	781	723	665	625	570		42
Ni ₂ P/Ni	1.5	611	583	521	483	449			28
Ni ₂ Pp@GN		520	449	397	291	283	246		29
CoP@C		770	690	560	490	420	340		32

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