

Electronic Supporting Information

Ultrathin 2D/2D Ti₃C₂T_x/semiconductor dual-functional photocatalysts for simultaneous imine production and H₂ evolution

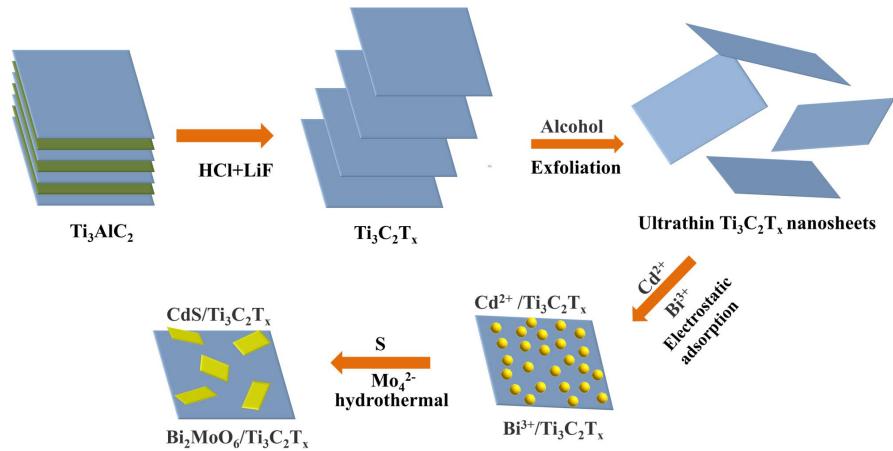
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Scheme S1. Schematic illustration of the synthesis of 2D $\text{Ti}_3\text{C}_2\text{T}_x$ /2D semiconductor hybrids.

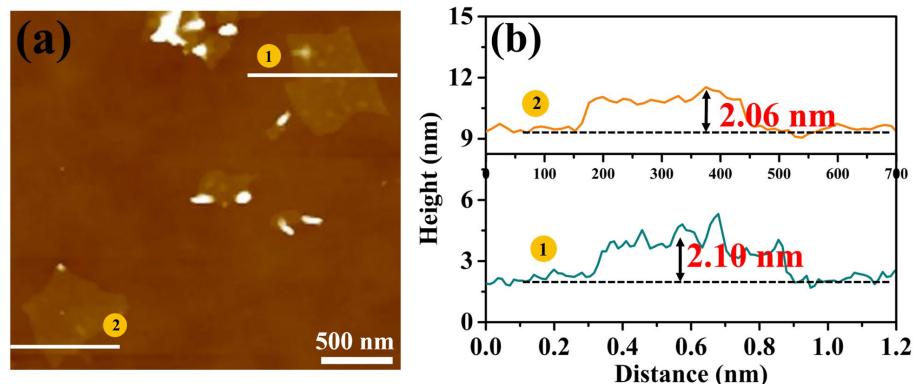


Fig. S1. (a) AFM image and (b) height cutaway view of $\text{Ti}_3\text{C}_2\text{T}_x$.

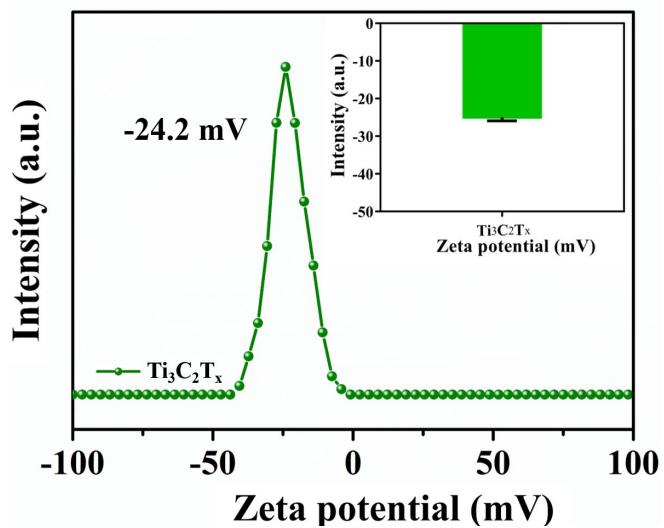


Fig. S2. Zeta potential of $\text{Ti}_3\text{C}_2\text{T}_x$.

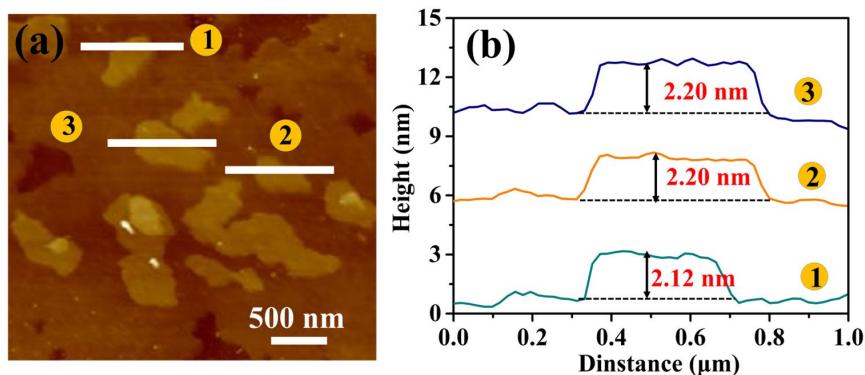


Fig. S3. (a) AFM image and (b) height cutaway view of CdS.

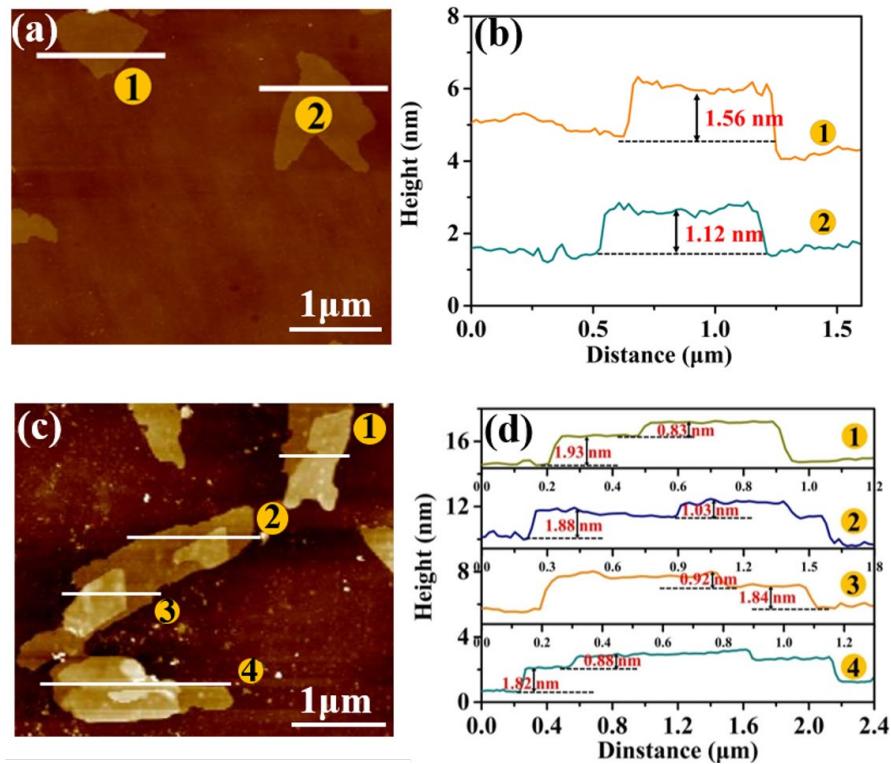


Fig. S4. AFM images and height cutaway views of (a, b) Bi_2MoO_6 and (c, d) TM1.5.

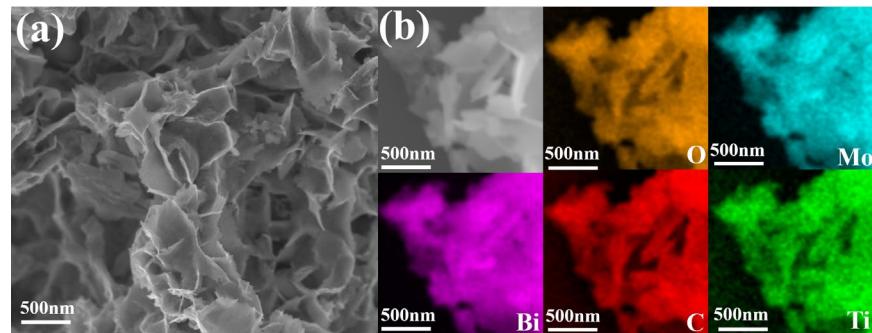


Fig. S5. SEM images of (a) Bi_2MoO_6 and (b) TM1.5 samples with the corresponding element mappings.

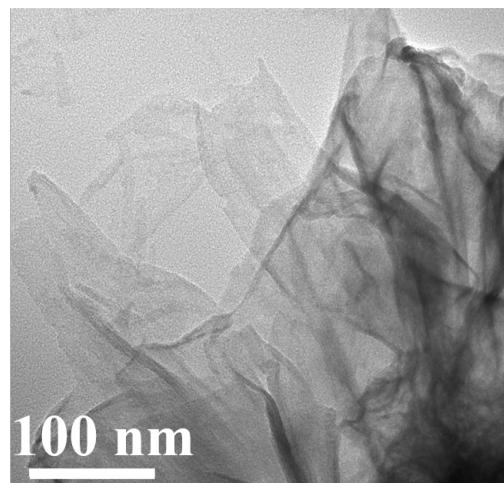


Fig. S6. Aberration-corrected STEM image of CdS.

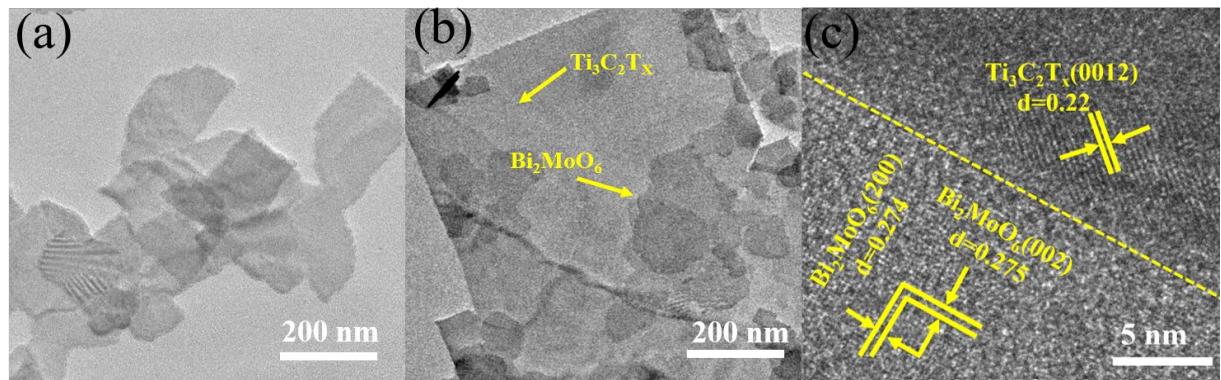


Fig. S7. Aberration-corrected STEM images of (a) Bi_2MoO_6 and (b-c) TM1.5 samples.

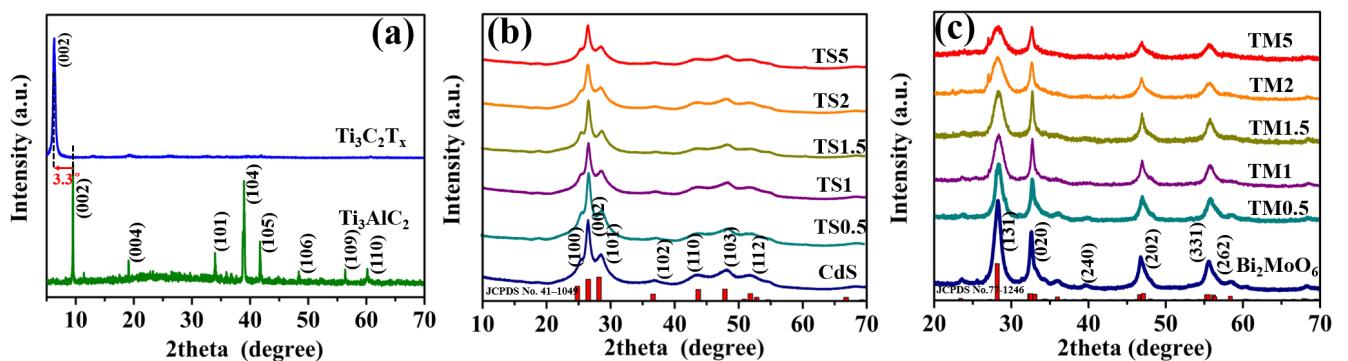


Fig. S8. (a) XRD patterns of $\text{Ti}_3\text{C}_2\text{T}_x$ and Ti_3AlC_2 samples; (b) XRD patterns of CdS and TS_x composites (where x represents the addition amount of $\text{Ti}_3\text{C}_2\text{T}_x$); (c) XRD patterns of Bi_2MoO_6 and TM_x composites (where x represents the addition amount of $\text{Ti}_3\text{C}_2\text{T}_x$).

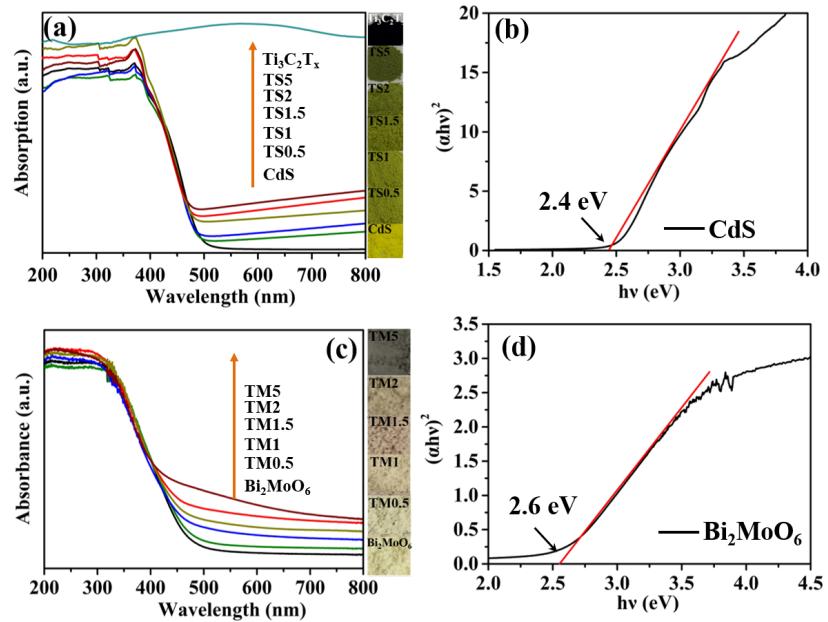


Fig. S9. (a) UV-Vis DRS spectra of $\text{Ti}_3\text{C}_2\text{T}_x$, CdS and TSx composites and photographs of different samples; (b) Tauc plot of CdS sample; (c) UV/Vis DRS spectra of Bi_2MoO_6 and TMx composites and photographs of different samples; (d) Tauc plot of Bi_2MoO_6 sample.

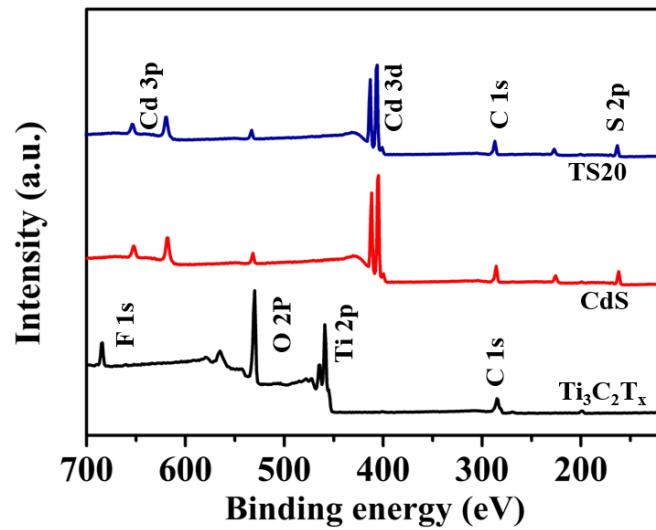


Fig. S10. XPS survey spectra of TS20, CdS and Ti₃C₂T_x

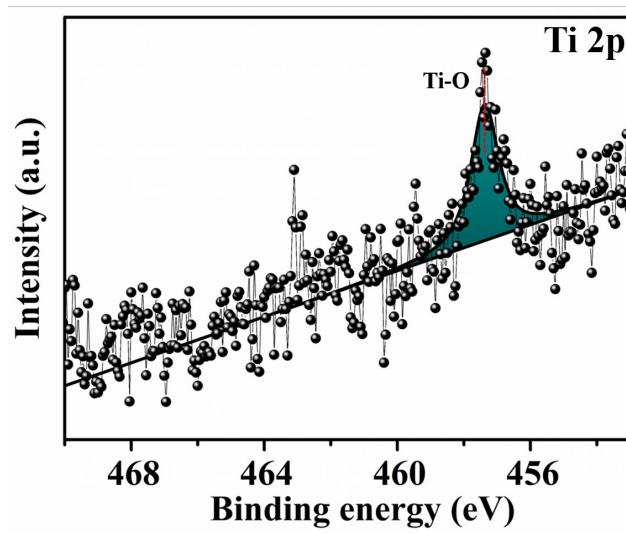


Fig. S11. High-resolution XPS spectrum of Ti 2p over TS20 sample.

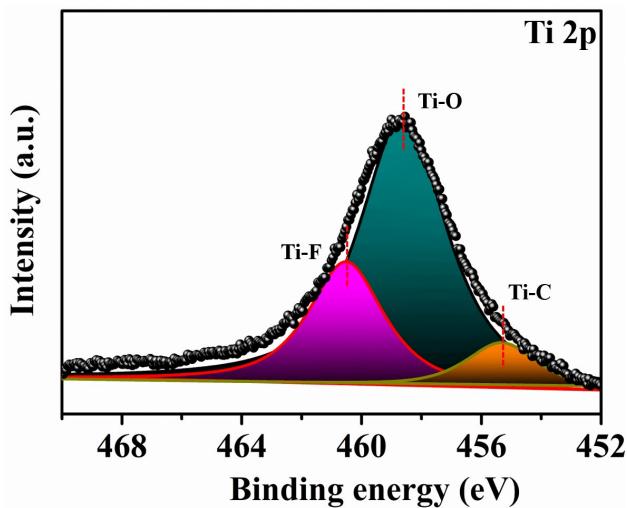


Fig. S12. High-resolution XPS spectrum of Ti 2p over TM1.5 sample.

Table S1: The activity of photocatalytic benzylamine oxidation integrated with H₂ production under different reaction conditions.

| Entry | Catalyst | hν | Yield _(Imines) /μmolg ⁻¹ h ⁻¹ | Yield _(H₂) /μmolg ⁻¹ h ⁻¹ |
|-------|----------|----|--|---|
| 1 | TS1.5 | - | - | - |
| 2 | TM2 | - | - | - |
| 3 | - | + | - | - |

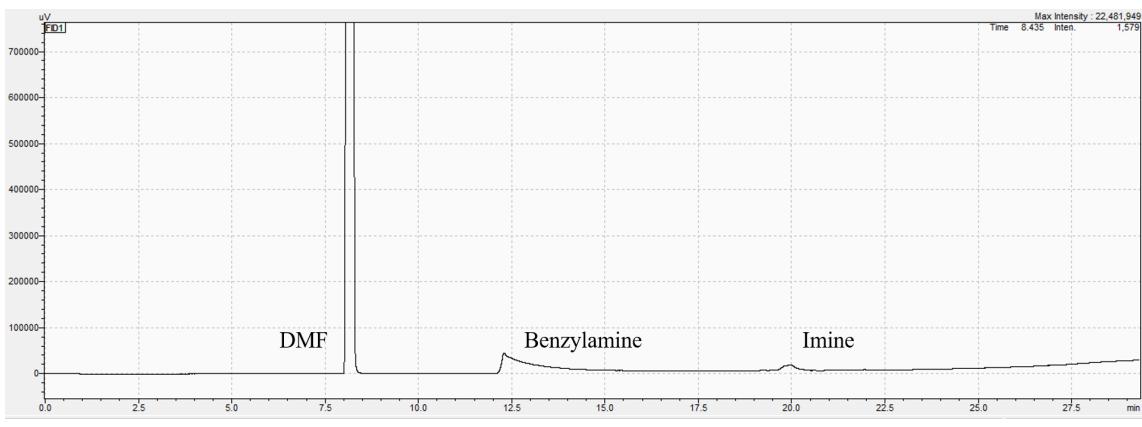


Fig. S13. GC spectrum of products after photocatalytic reaction over TS1.5 sample.

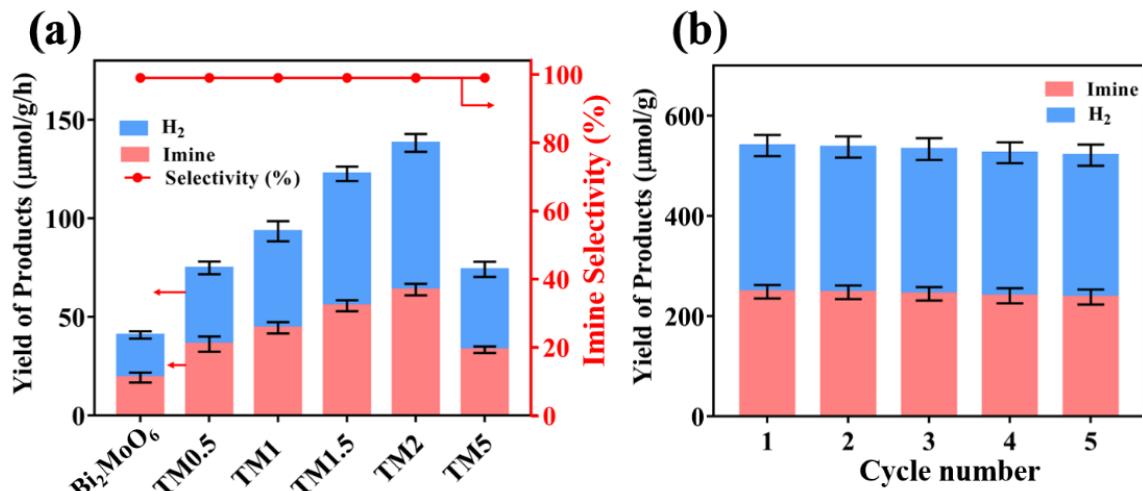


Fig. S14. (a) Photocatalytic activity for benzylamine oxidation and H_2 generation over Bi_2MoO_6 and TM_x composites under visible light irradiation ($\lambda \geq 420 \text{ nm}$); (b) Recycling tests over $\text{TM}2$ composite for photocatalytic benzylamine oxidation integrated with H_2 production (each cycle is 4 h).

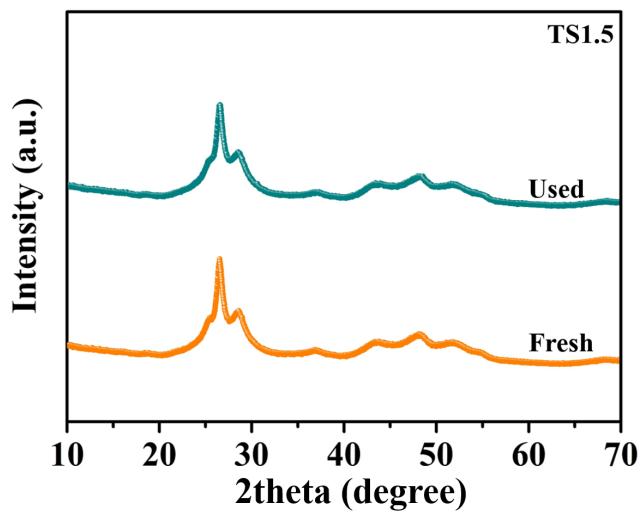


Fig. S15. XRD patterns of the TS1.5 sample before and after photocatalytic reaction.

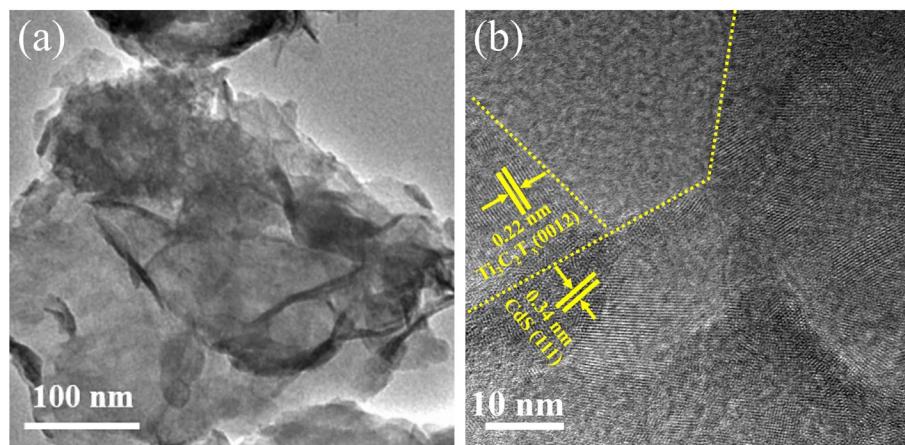


Fig. S16. Aberration-corrected STEM patterns of the TS1.5 sample after photocatalytic reaction.

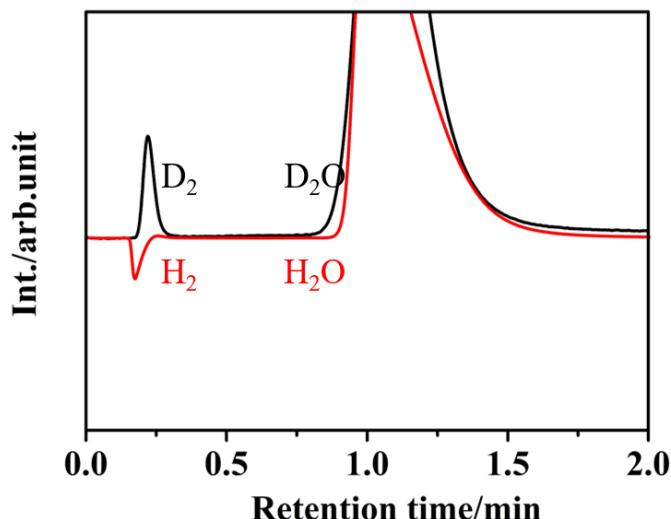


Fig. S17. GC analysis of gas phase products for photocatalytic benzylamine coupling and H_2O (or D_2O) splitting.

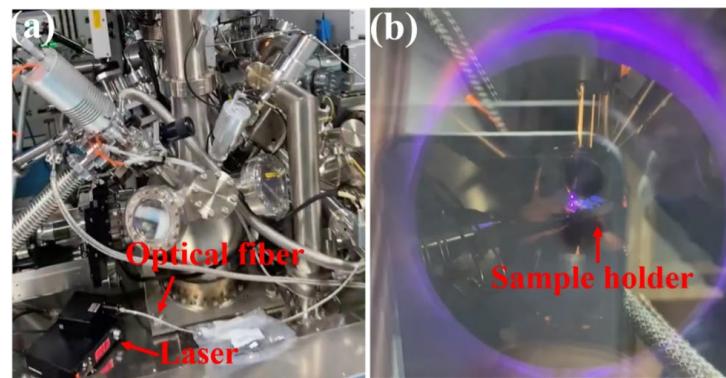


Fig. S18. In-situ irradiation XPS setup of TS20 sample.

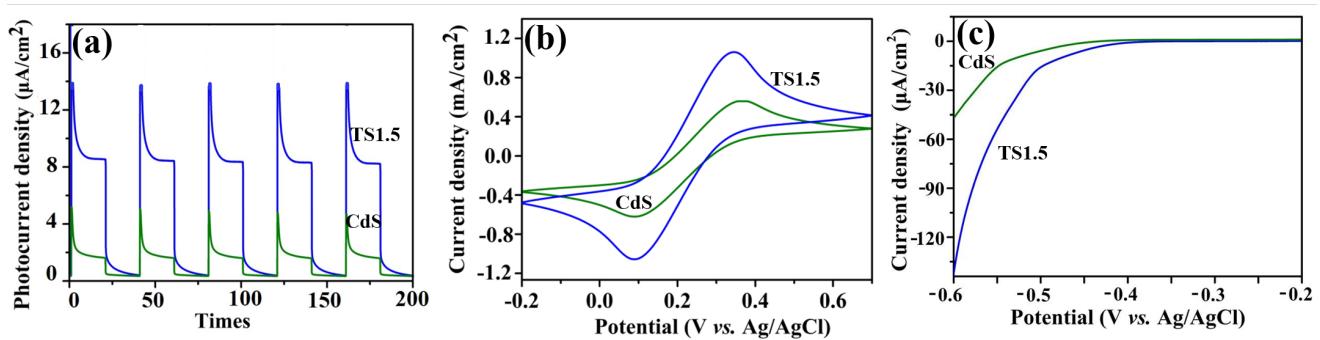


Fig. S19. (a) Transient photocurrent spectra, (b) CV and (c) LSV of TS1.5 and CdS samples.

Table S2: Photocatalytic hydrogen production uses different catalysts

| photocatalysts | Light sources | Reaction conditions | Efficiencies ($\mu\text{mol g}^{-1} \text{h}^{-1}$) | Ref. |
|--|--|---|---|-----------|
| CdS/Ti ₃ C ₂ T _x | 300 W Xe lamp $\lambda \geq 420$ nm | 10mg; DMF-based solution with H ₂ O; 5 mL | 219.7 | This work |
| Pt/F-TiO ₂ | 300 W Xe lamp 500 nm $\geq \lambda \geq 300$ nm | 15 mg; Perchloric acid aqueous solution | 160 | 1 |
| Ag/Co ₃ O ₄ /Cu ₂ O | 100W Xe lamp | 100 mg; aqueous solution; 100mL | 97.2 | 2 |
| MoS ₂ QDs@ZnIn ₂ S ₄ @RGO | 300 W Xe lamp 780 nm $\geq \lambda \geq 320$ nm | 100 mg; aqueous solution | 37.8 | 3 |
| ZnO _{0.6} S _{0.4} | 300 W Xe lamp $\lambda \geq 400$ nm | 50 mg; aqueous solution; 50 mL | 49.2 | 4 |
| Pt/TiO ₂ | 450 W Xe lamp (UV-vis) | 10 mg; aqueous solution; 215 mL | 101 | 5 |
| TiO ₂ | 300 W Xe lamp (UV-vis) | 100 mg; aqueous solution; 120 mL | 19.4 | 6 |
| CdS/CdS/GCN | 300 W Xe lamp $\lambda > 420$ nm; | 100 mg; aqueous solution; 80 mL | 5 | 7 |
| Pt/PCN-777 | 300 W Xe lamp (UV-vis) | 10 mg; DMF-based solution with H ₂ O; 5.1 mL | 335; 1.7 | 8 |

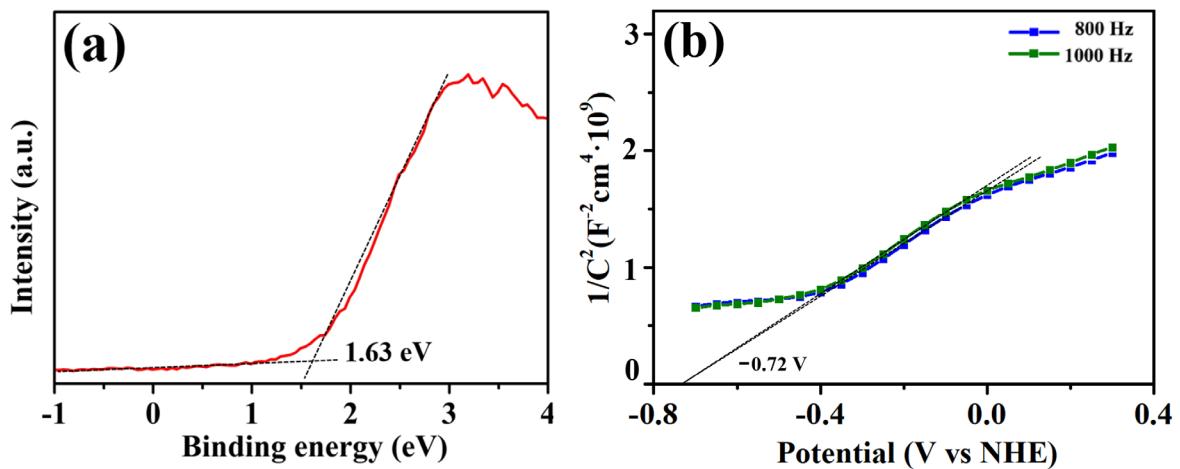


Fig. S20. (a) Ultraviolet photoelectron spectroscopy (UPS) spectrum and (b) Mott-Schottky plots of CdS sample.

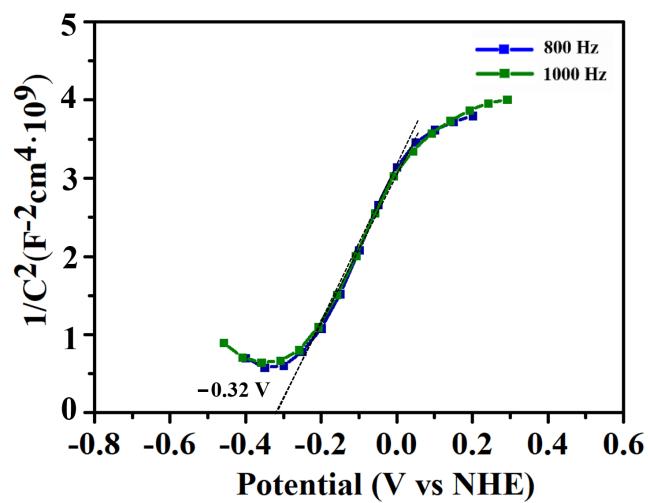


Fig. S21. (a) Mott-Schottky plots of Bi_2MoO_6 sample.

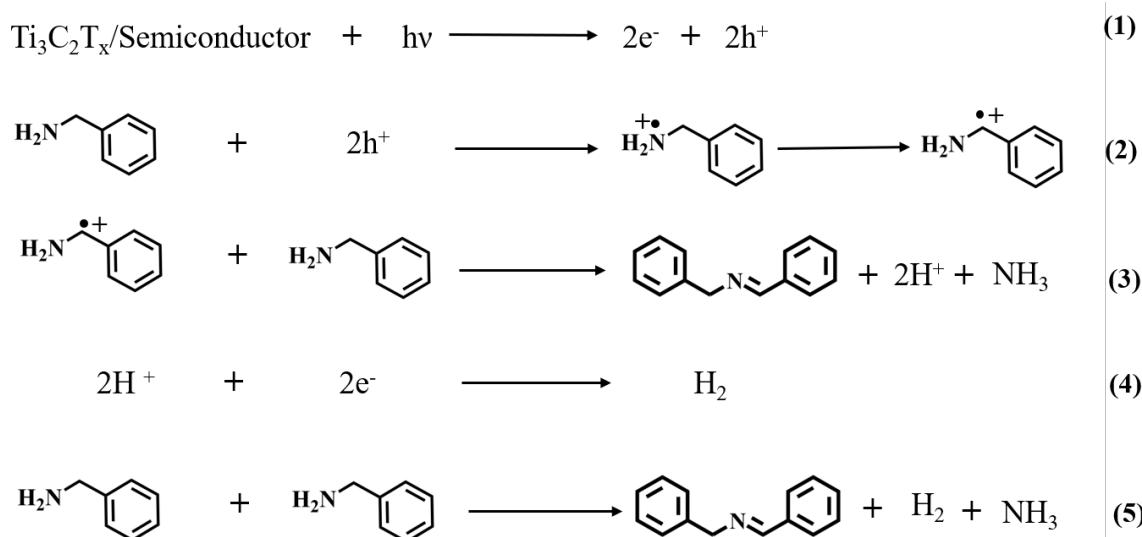


Fig. S22. Proposed reaction pathway for photocatalytic imine production and H₂ evolution over Ti₃C₂T_x/semiconductors.

References:

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