

Supplementary Information:

A New Strontium Based Reactive Carbonate Composite for Thermochemical Energy Storage

Adriana P. Vieira,* Kyran Williamson, Terry D. Humphries, Mark Paskevicius, Craig E. Buckley*

Physics and Astronomy, Fuels and Energy Technology Institute, Curtin University, GPO Box U1987, Perth, WA 6845, Australia. E-mail: apiresvieira@gmail.com; c.buckley@curtin.edu.au

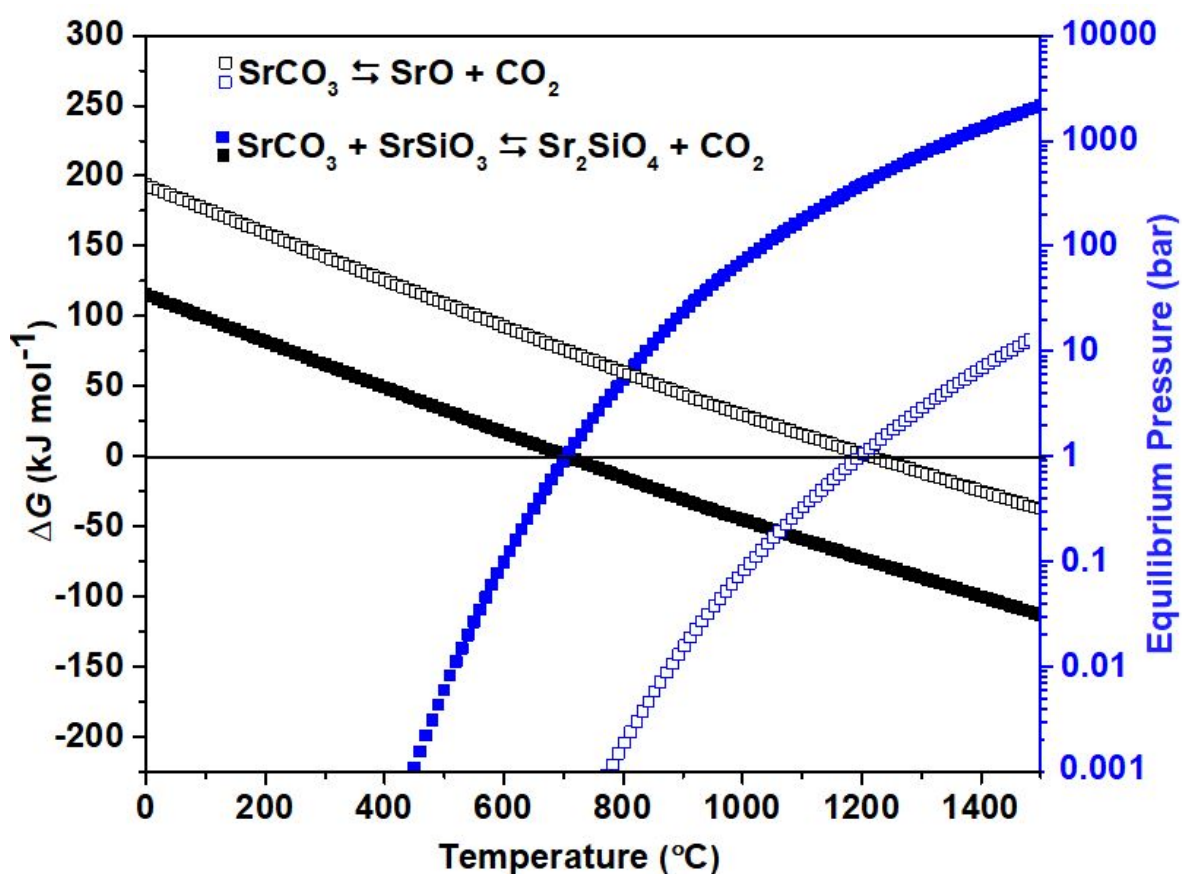


Figure SI-1: Gibbs free energy and equilibrium pressure dependence on temperature based on thermodynamics of the reactions.²⁴

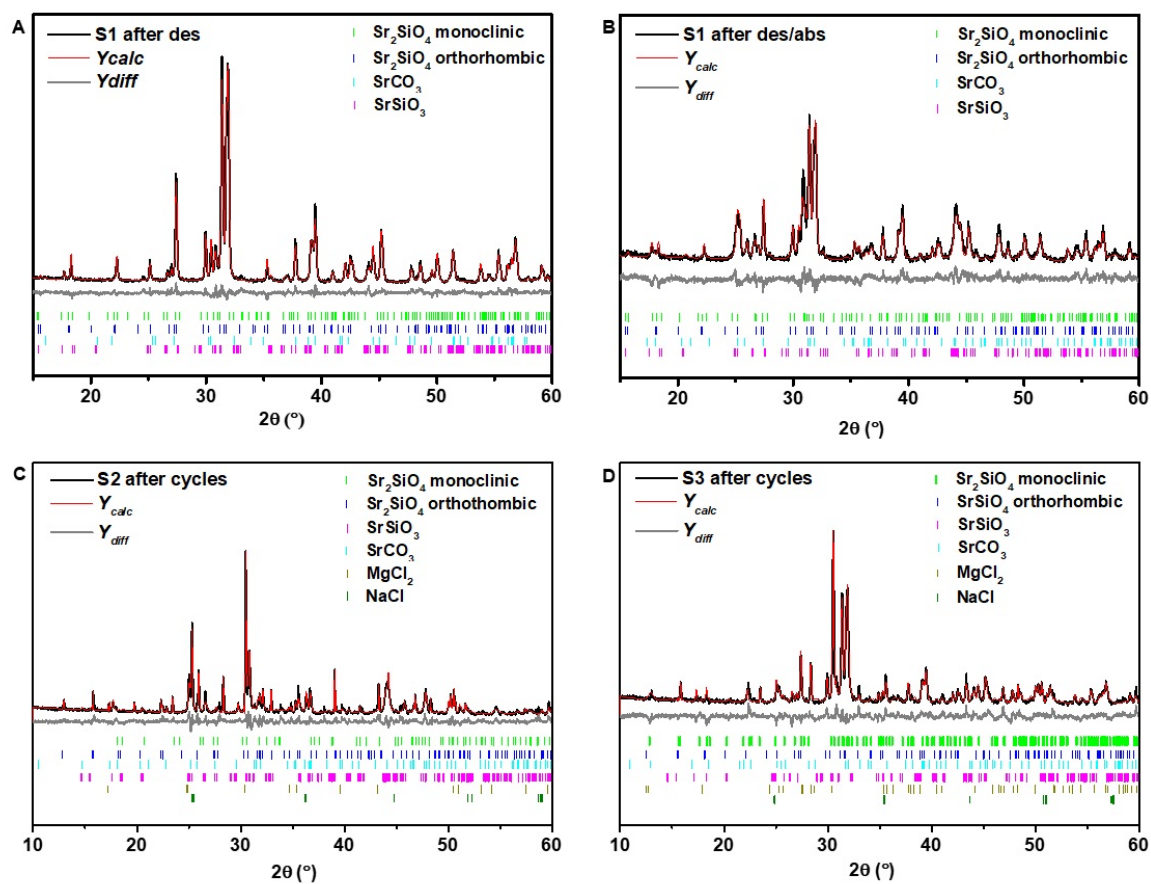


Figure SI-2: Rietveld refinement of (A) **S1** after desorption. $R_{\text{wp}} = 7.89\%$; (B) **S1** after des/abs. $R_{\text{wp}} = 8.34\%$; (C) **S2** after cycles. $R_{\text{wp}} = 7.24\%$ (C); (D) **S3** after cycles. $R_{\text{wp}} = 7.61\%$.

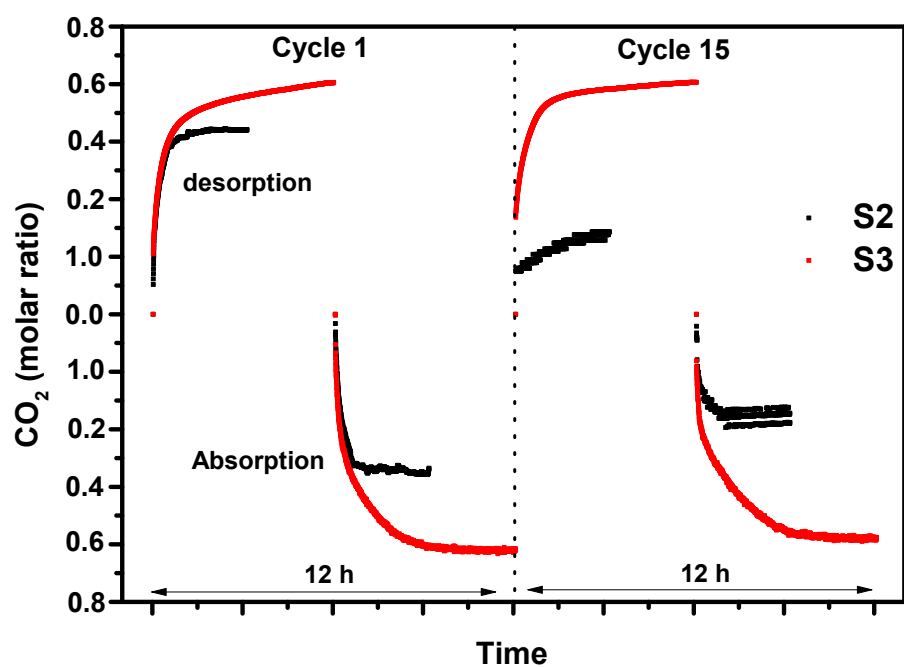


Figure SI-3. Comparison of the kinetics of CO₂ desorption/absorption between the 1st and 15th cycles for samples S2 and S3.