

When water becomes an integral part of carbon - combining theory and experiment to understand the zeolite-like water adsorption properties of porous C₂N materials

Supporting Information

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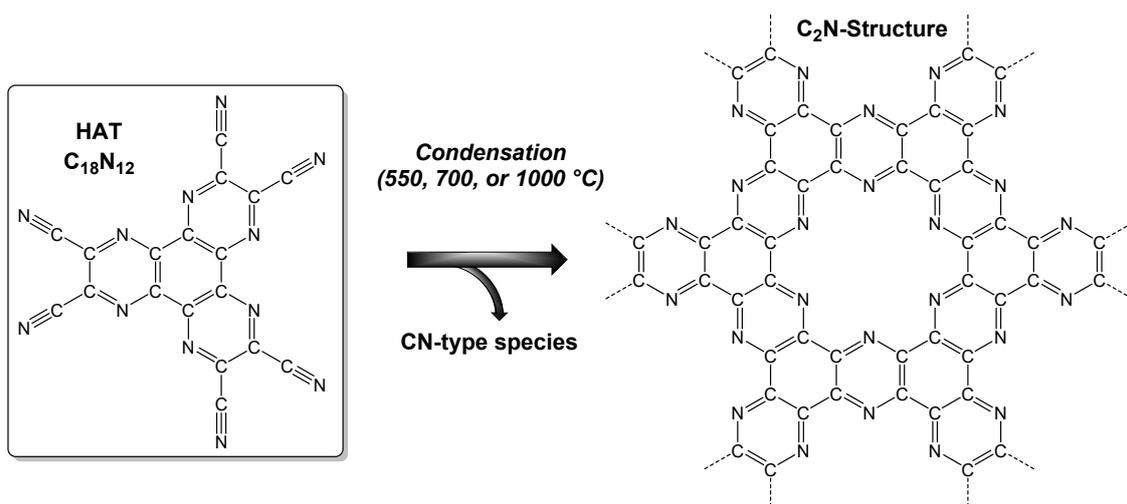


Figure S1. Thermal condensation of HAT-CN to C₂N-type products.

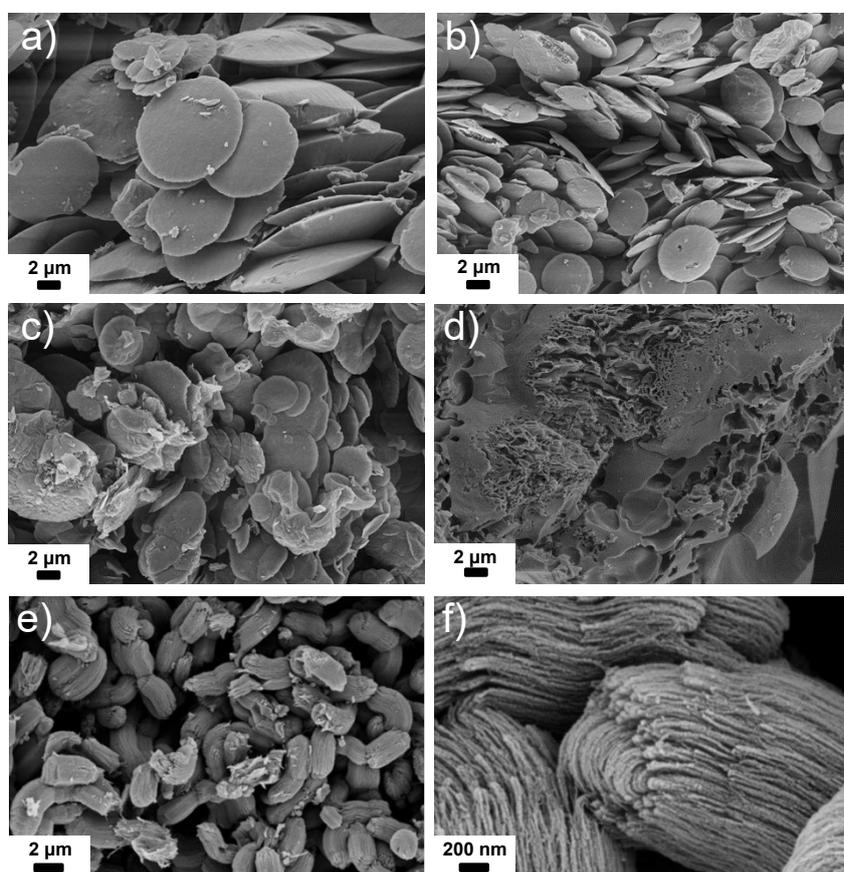


Figure S2. SEM images of C-HAT-CN-550 (a), C-HAT-CN-700 (b), C-HAT-CN-1000 (c), TiC-CDC (d), and CMK-3 (e and f).

Table S1. Adsorption energies ΔE^{ads} and incremental adsorption energies ΔE_{inc}^{ads} for different numbers of H₂O molecules adsorbed in a single pore of C₂N and C₃H, respectively.

No. of H ₂ O in single interlayer	C ₂ N			Reference C ₃ H		
	$n \cdot \Delta E^{ads}$ [kJ/mol]	ΔE^{ads} [kJ/mol]	ΔE_{inc}^{ads} [kJ/mol]	ΔE^{ads} [kJ/mol]	ΔE^{ads} [kJ/mol]	ΔE_{inc}^{ads} [kJ/mol]
1	-45.2	-45.2	-45.2	-32.8	-32.8	-32.8
2	-88.6	-44.3	-43.5	-64.2	-32.1	-31.4
3	-133.2	-44.4	-44.6	-95.6	-31.9	-31.4
4	-191.1	-47.8	-57.9	-124.7	-31.2	-29.1
5	-149.3	-29.9	+41.7	+690.8	+138.2	+815.5
4 in each interlayer	-758.0	-47.4		-504.2	-31.5	

Atomic coordinates of the final adsorption state (four adsorbed molecules in each pore) of water in C₂N as well as in the reference system are available for download at:

<https://doi.org/10.5281/zenodo.4469324>.