

Electronic Supplementary Information for Theoretical Study on the Anisotropic Photo-Induced Carrier Mobilities in Layered Double Hydroxides- Based Photocatalysts

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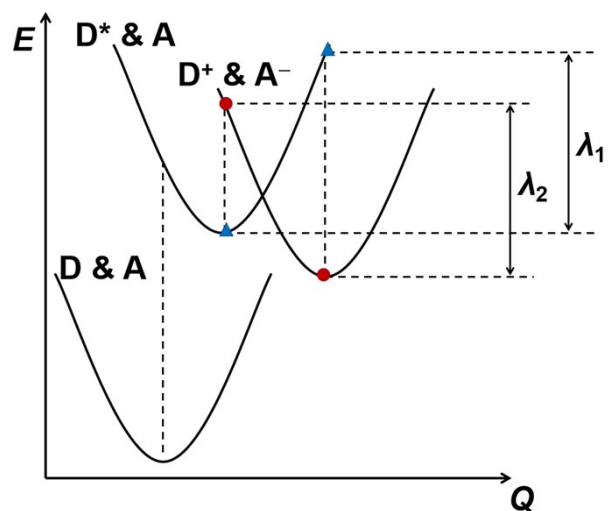


Fig. S1. Schematic illustration of the compositions of internal reorganization energy λ in Marcus theory. The curves represent the potential energy surfaces of D & A (D represents the donor in the ground state, A denotes the acceptor in the ground state), and $D^+ & A^-$ (D^+ represents the donor in the cationic state and A^- represents the acceptor in the anionic state), $D^* & A$ (D^* represents the donor in the excited state), respectively.

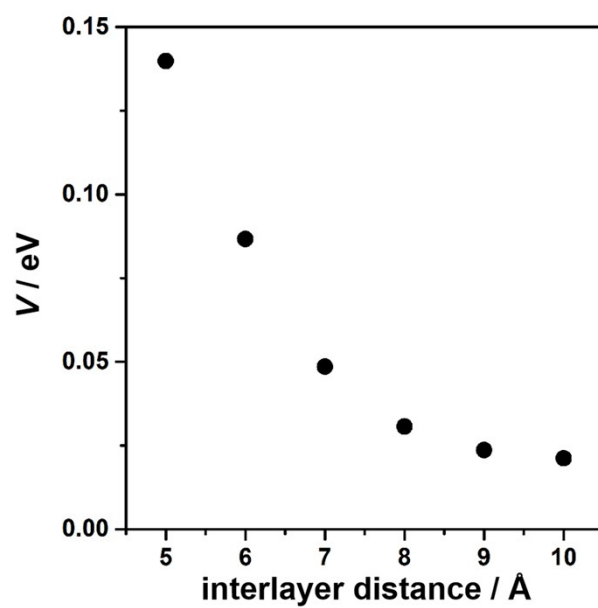


Fig. S2. Relationship between the transfer integral V and interlayer distance between two LDHs matrix in $[\text{Mg}_2\text{Al}(\text{OH})_6(\text{OH}_2)_7]\cdot\text{Cl}$.

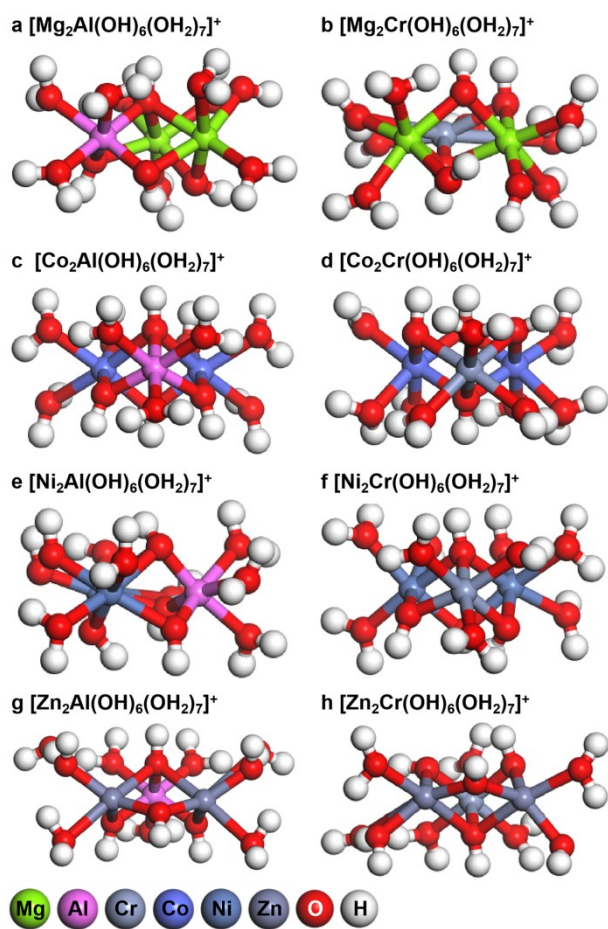


Fig. S3. Optimized geometries of (a) $[\text{Mg}_2\text{Al}(\text{OH})_6(\text{OH}_2)_7]^+$, (b) $[\text{Mg}_2\text{Cr}(\text{OH})_6(\text{OH}_2)_7]^+$, (c) $[\text{Co}_2\text{Al}(\text{OH})_6(\text{OH}_2)_7]^+$, (d) $[\text{Co}_2\text{Cr}(\text{OH})_6(\text{OH}_2)_7]^+$, (e) $[\text{Ni}_2\text{Al}(\text{OH})_6(\text{OH}_2)_7]^+$, (f) $[\text{Ni}_2\text{Cr}(\text{OH})_6(\text{OH}_2)_7]^+$, (g) $[\text{Zn}_2\text{Al}(\text{OH})_6(\text{OH}_2)_7]^+$, and (h) $[\text{Zn}_2\text{Cr}(\text{OH})_6(\text{OH}_2)_7]^+$ cluster models. The color of each element is labeled.

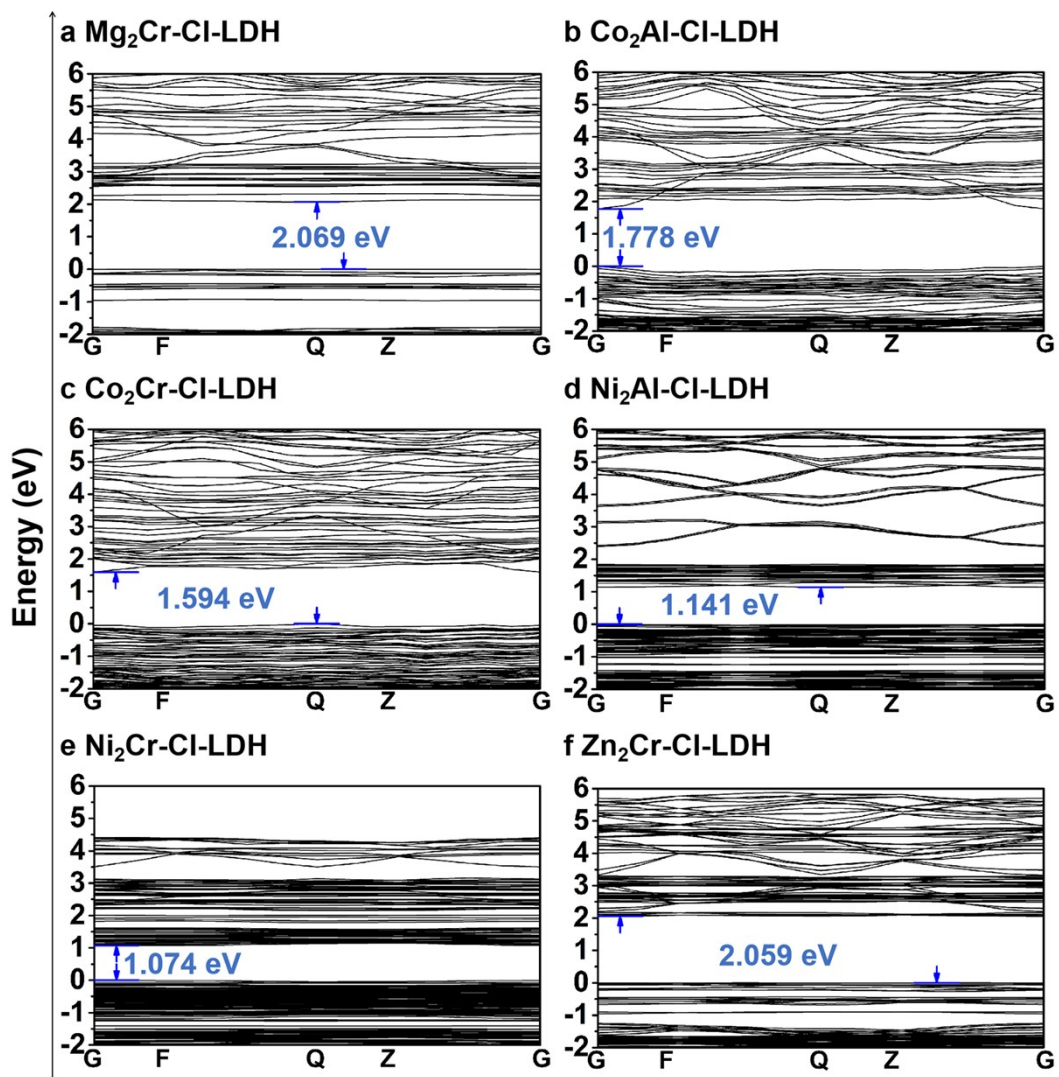


Fig. S4. Band Structures of (a) $\text{Mg}_2\text{Cr-CI-LDH}$, (b) $\text{Co}_2\text{Al-CI-LDH}$, (c) $\text{Co}_2\text{Cr-CI-LDH}$, (d) $\text{Ni}_2\text{Al-CI-LDH}$, (e) $\text{Ni}_2\text{Cr-CI-LDH}$, and (f) $\text{Zn}_2\text{Cr-CI-LDH}$ at the level of PBE. The valence band maximums, conduction band minimums, and band gap energies are labeled.

Table S1. Experimental and Calculated (with and without van der Waals Correction) Lattice Parameters for Periodic Models of M_2M' -Cl-LDHs ($M = \text{Mg, Co, Ni, and Zn}$; $M' = \text{Al, and Cr}$)

model	lattice parameter (\AA)					
	experimental		calculated			
	$a = b$	c	without van der Waals correction		with van der Waals correction	
			$a = b$	c	$a = b$	c
Mg ₂ Al-Cl-LDH	3.04 ^[S1]	7.54 ^[S1]	3.08	7.49	3.07	7.38
Mg ₂ Cr-Cl-LDH	3.01 ^[S2]	7.72 ^[S3]	3.10	7.58	3.11	7.50
Co ₂ Al-Cl-LDH	3.07 ^[S4]	7.59 ^[S4]	3.07	7.59	3.05	7.52
Co ₂ Cr-Cl-LDH	3.08 ^[S5]	7.86 ^[S6]	3.08	7.73	3.08	7.70
Ni ₂ Al-Cl-LDH	3.01 ^[S7]	7.64 ^[S7]	3.03	8.05	3.03	7.78
Ni ₂ Cr-Cl-LDH	3.08 ^[S8]	7.74 ^[S8]	3.07	8.66	3.07	8.28
Zn ₂ Al-Cl-LDH	3.08 ^[S9]	7.80 ^[S9]	3.11	7.81	3.10	7.81
Zn ₂ Cr-Cl-LDH	3.11 ^[S10]	7.71 ^[S10]	3.15	7.73	3.15	7.72

Table S2. Structural Information for the Optimized Geometries of
 $[\text{Mg}_2\text{Al}(\text{OH})_6(\text{OH}_2)_7]\cdot\text{Cl}$, $[\text{Mg}_4\text{Al}_2(\text{OH})_{12}(\text{OH}_2)_{10}]\cdot 2\text{Cl}$, and
 $[\text{Mg}_6\text{Al}_3(\text{OH})_{18}(\text{OH}_2)_{12}]\cdot 3\text{Cl}$ Cluster Models

model	distance (Å)		bond length (Å)	
	Mg-Mg	Mg-Al	Mg-O	Al-O
$[\text{Mg}_2\text{Al}(\text{OH})_6(\text{OH}_2)_7]\cdot\text{Cl}$	3.073	3.015	2.294	1.920
$[\text{Mg}_4\text{Al}_2(\text{OH})_{12}(\text{OH}_2)_{10}]\cdot 2\text{Cl}$	2.981	3.031	2.261	1.932
$[\text{Mg}_6\text{Al}_3(\text{OH})_{18}(\text{OH}_2)_{12}]\cdot 3\text{Cl}$	3.152	3.092	2.157	1.932

model	bond angle (°)			dihedral angle (°)
	Mg-O-Mg	O-Mg-O	O-Al-O	O-Al-O-Mg
$[\text{Mg}_2\text{Al}(\text{OH})_6(\text{OH}_2)_7]\cdot\text{Cl}$	97.122	89.105	87.092	176.715
$[\text{Mg}_4\text{Al}_2(\text{OH})_{12}(\text{OH}_2)_{10}]\cdot 2\text{Cl}$	92.938	76.663	87.286	173.455
$[\text{Mg}_6\text{Al}_3(\text{OH})_{18}(\text{OH}_2)_{12}]\cdot 3\text{Cl}$	95.401	81.328	86.255	174.054

Table S3. Structural Information for the Optimized Geometries of[M₂M'(OH)₆(OH₂)₇]⁺·Cl Cluster Models (M = Mg, Co, Ni, and Zn; M' = Al, and Cr)

model	distance (Å)		bond length (Å)	
	M-M	M-M'	M-O	M'-O
[Mg ₂ Al(OH) ₆ (OH ₂) ₇] ⁺	3.035	3.003	2.295	1.921
[Mg ₂ Cr(OH) ₆ (OH ₂) ₇] ⁺	3.030	3.092	2.278	1.996
[Co ₂ Al(OH) ₆ (OH ₂) ₇] ⁺	3.207	2.951	2.218	1.920
[Co ₂ Cr(OH) ₆ (OH ₂) ₇] ⁺	3.156	3.140	2.300	1.988
[Ni ₂ Al(OH) ₆ (OH ₂) ₇] ⁺	2.857	3.255	2.410	1.936
[Ni ₂ Cr(OH) ₆ (OH ₂) ₇] ⁺	3.027	3.022	2.255	1.996
[Zn ₂ Al(OH) ₆ (OH ₂) ₇] ⁺	3.126	3.051	2.325	1.921
[Zn ₂ Cr(OH) ₆ (OH ₂) ₇] ⁺	3.081	3.098	2.283	1.994
model	angle (°)		dihedral angle (°)	
	M-O-M	O-M-O	O-M'-O	O-M'-O-M
[Mg ₂ Al(OH) ₆ (OH ₂) ₇] ⁺	95.227	85.852	87.265	175.627
[Mg ₂ Cr(OH) ₆ (OH ₂) ₇] ⁺	95.741	75.209	85.209	176.489
[Co ₂ Al(OH) ₆ (OH ₂) ₇] ⁺	99.012	83.343	88.172	172.560
[Co ₂ Cr(OH) ₆ (OH ₂) ₇] ⁺	99.116	80.819	86.238	176.947
[Ni ₂ Al(OH) ₆ (OH ₂) ₇] ⁺	72.844	67.444	90.543	147.219
[Ni ₂ Cr(OH) ₆ (OH ₂) ₇] ⁺	97.661	82.916	86.724	170.754
[Zn ₂ Al(OH) ₆ (OH ₂) ₇] ⁺	96.993	74.797	86.561	175.534
[Zn ₂ Cr(OH) ₆ (OH ₂) ₇] ⁺	97.460	76.706	87.480	176.029

Table S4. Band Gap Energies (E_g), Work Functions (W), Valence Band Maximum (E_{VBM}), and Conduction Band Minimum (E_{CBM}) for M_2M' -Cl-LDHs ($M = \text{Mg, Co, Ni, and Zn}$; $M' = \text{Al, and Cr}$) under Lattice Dilation and Compression in x and y

Directions						
model	direction	$\Delta l / l_0$	E_g (eV)	W (eV)	E_{VBM} / eV	E_{CBM} / eV
Mg ₂ Al-Cl-LDH	x	-1%	4.357	4.847	-5.305	-0.973
		-0.5%	4.344	4.853	-5.248	-0.904
		0%	4.333	4.860	-5.100	-0.812
		0.5%	4.332	4.500	-5.020	-0.738
		1%	4.319	4.871	-4.970	-0.651
	y	-1%	4.353	4.662	-5.255	-0.641
		-0.5%	4.338	4.610	-5.238	-0.686
		0%	4.311	4.699	-5.207	-0.697
		0.5%	4.293	4.661	-5.389	-0.715
		1%	4.274	4.634	-5.371	-0.744
Mg ₂ Cr-Cl-LDH	x	-1%	2.867	4.237	-6.383	-3.504
		-0.5%	2.836	4.365	-6.351	-3.547
		0%	2.824	4.259	-6.270	-3.581
		0.5%	2.768	4.232	-6.246	-3.618
		1%	2.742	4.229	-6.200	-3.647
	y	-1%	2.819	4.230	-6.285	-3.521
		-0.5%	2.806	4.282	-6.260	-3.542
		0%	2.803	4.259	-6.241	-3.549
		0.5%	2.791	4.245	-6.239	-3.558
		1%	2.775	4.230	-6.218	-3.579
Co ₂ Al-Cl-LDH	x	-1%	2.626	5.547	-6.860	-4.234
		-0.5%	2.581	5.563	-6.854	-4.272
		0%	2.651	5.580	-6.906	-4.254
		0.5%	2.584	5.608	-6.900	-4.316
		1%	2.650	5.623	-6.948	-4.298
	y	-1%	2.615	5.449	-6.757	-4.141
		-0.5%	2.632	5.494	-6.810	-4.178
		0%	2.651	5.580	-6.906	-4.254
		0.5%	2.644	5.634	-6.956	-4.312
		1%	2.651	5.690	-7.015	-4.365
Co ₂ Cr-Cl-LDH	x	-1%	2.239	4.369	-7.937	-5.581
		-0.5%	2.254	4.312	-7.969	-5.549
		0%	2.269	4.416	-7.989	-5.502
		0.5%	2.279	4.297	-8.002	-5.485
		1%	2.301	4.352	-8.051	-5.458

		-1%	2.237	4.619	-7.938	-5.501
		-0.5%	2.242	4.687	-7.954	-5.512
	<i>y</i>	0%	2.251	4.628	-7.986	-5.530
		0.5%	2.297	4.658	-8.008	-5.544
		1%	2.292	4.680	-8.026	-5.566
		-1%	2.565	4.671	-5.953	-3.389
		-0.5%	2.567	4.647	-5.937	-3.364
	<i>x</i>	0%	2.588	4.629	-5.930	-3.343
		0.5%	2.594	4.640	-5.923	-3.335
		1%	2.597	4.602	-5.901	-3.303
Ni ₂ Al-Cl-LDH		-1%	2.294	4.672	-5.819	-3.525
		-0.5%	2.322	4.609	-5.806	-3.468
	<i>y</i>	0%	2.338	4.637	-5.800	-3.448
		0.5%	2.355	4.622	-5.770	-3.404
		1%	2.371	4.562	-5.748	-3.377
		-1%	2.079	4.843	-5.883	-3.803
		-0.5%	2.094	4.793	-5.840	-3.746
	<i>x</i>	0%	2.163	4.720	-5.802	-3.638
		0.5%	2.146	4.696	-5.769	-3.623
		1%	2.092	4.669	-5.715	-3.623
Ni ₂ Cr-Cl-LDH		-1%	2.115	4.550	-5.608	-3.492
		-0.5%	2.059	4.633	-5.663	-3.603
	<i>y</i>	0%	2.163	4.720	-5.802	-3.638
		0.5%	2.085	4.774	-5.816	-3.732
		1%	2.110	4.824	-5.879	-3.769
		-1%	3.401	4.854	-5.961	-2.453
		-0.5%	3.405	4.848	-5.955	-2.445
	<i>x</i>	0%	3.417	4.852	-5.951	-2.444
		0.5%	3.408	4.838	-5.942	-2.434
		1%	3.411	4.829	-5.935	-2.424
Zn ₂ Al-Cl-LDH		-1%	3.382	4.848	-5.942	-2.458
		-0.5%	3.384	4.850	-5.939	-2.457
	<i>y</i>	0%	3.389	4.841	-5.936	-2.446
		0.5%	3.391	4.835	-5.930	-2.440
		1%	3.388	4.833	-5.927	-2.439
		-1%	2.466	5.031	-6.264	-3.798
		-0.5%	2.459	5.055	-6.285	-3.825
	<i>x</i>	0%	2.478	5.065	-6.304	-3.826
		0.5%	2.424	5.074	-6.286	-3.862
		1%	2.409	5.077	-6.282	-3.872
Zn ₂ Cr-Cl-LDH		-1%	2.495	4.868	-6.116	-3.620
		-0.5%	2.497	4.957	-6.205	-3.709
	<i>y</i>	0%	2.478	5.065	-6.304	-3.826
		0.5%	2.436	5.152	-6.370	-3.934

1%	2.390	5.215	-6.410	-4.020
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