

# Sorption-enhanced mixed-gas transport in amine functionalized polymers of intrinsic microporosity (PIMs)

Katherine Mizrahi Rodriguez<sup>a,†</sup>, Francesco M. Benedetti<sup>b,†</sup>, Naksha Roy<sup>b</sup>, Albert X. Wu<sup>b</sup>, and Zachary P. Smith<sup>b</sup>

<sup>†</sup>These authors contributed equally.

<sup>a</sup>Department of Materials Science and Engineering, Massachusetts Institute of Technology,  
Cambridge, Massachusetts 02139, United States

<sup>b</sup>Department of Chemical Engineering, Massachusetts Institute of Technology, Cambridge,  
Massachusetts 02139, United States

## Table of Contents

1. Sample information, materials, and methods.....	3
2. Physical aging upper bound .....	5
3. Variable-pressure mixed-gas permeation.....	6
4. Other mixtures (CO <sub>2</sub> /N <sub>2</sub> , H <sub>2</sub> /N <sub>2</sub> , and H <sub>2</sub> /CH <sub>4</sub> ) .....	8
5. Pure-gas sorption .....	9
6. Mixed-gas sorption predictions.....	11
7. Pure and mixed-gas plasticization .....	14
8. Reference literature data.....	18
9. Sorption upper bound at infinite dilution.....	23
References.....	24

## **1. Sample information, materials, and methods**

**Gas purity and suppliers.** Pure- and mixed-gas permeation and sorption tests were performed using ultrahigh purity gases purchased from Airgas: H<sub>2</sub> (UHP300, 99.999%), CH<sub>4</sub> (UHP300, 99.99%), N<sub>2</sub> (UHP300, 99.999%), and CO<sub>2</sub> (UHP300, 99.999%).

**Pure-gas permeation.** Pure-gas permeation tests were performed at 35 °C and 1 atm using an automated constant-volume, variable-pressure system purchased from Maxwell Robotics. The temperature in the cell was maintained with a water bath using a Thermo Fisher SC150L water circulator. Polymer films of approximately 25 mm<sup>2</sup> in area were placed over a hole on a brass disk, and the edges of the film were attached to the disk using Devcon 5 min Epoxy. Samples were loaded into a stainless steel cell and tightly sealed with two concentric o-rings. At the beginning of a permeation test, the samples were de-gassed for 8 h under vacuum to remove environmental moisture or residual atmospheric gases. When switching between gases, the sample cell was flushed with helium and de-gassed under vacuum for at least 1 h. After conducting pure-gas tests at 1 atm, high-pressure CO<sub>2</sub> tests were performed up to fugacities of 29 atm. For each pressure point, samples were held for 0.5 h to ensure all samples had identical CO<sub>2</sub> exposure and sufficient time to reach steady state.

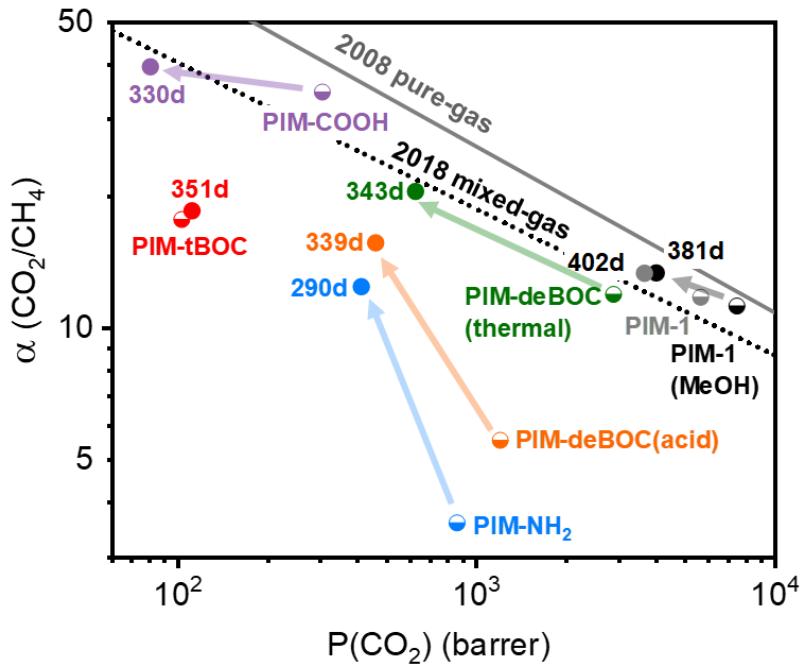
**Table S1.** Thickness ( $l$ ), aging time ( $t$ ), and treatment conditions for samples tested in mixtures.

Polymer	$l$ (μm)	$t$ (days)	treatment 1 d after casting
<b>PIM-1</b> (fresh)	58.0 ± 0.9	1	MeOH 24 h, 130 °C 12 h
<b>PIM-1</b> (381 d)	43.8 ± 0.9	381	MeOH 24 h, 130 °C 12 h
<b>PIM-NH<sub>2</sub></b> (fresh)	49 ± 2	1	MeOH 24 h, 130 °C 12 h
<b>PIM-NH<sub>2</sub></b> (290 d)	51 ± 2	290	MeOH 24 h, 130 °C 12 h
<b>PIM-NH<sub>2</sub></b> (448 d)	67.2 ± 0.9	448	MeOH 24 h, 130 °C 12 h
<b>PIM-NH<sub>2</sub></b> (cond.)	82 ± 1	433	MeOH 24 h, 130 °C 12 h followed by a pure-gas CO <sub>2</sub> conditioning test up to 31 atm
<b>PIM-t-BOC</b>	70 ± 9	351	MeOH 24 h, 130 °C 12 h
<b>PIM-deBOC(acid)</b>	65.0 ± 0.6	339	MeOH 24 h, 130 °C 12 h
<b>PIM-deBOC(thermal)</b>	83 ± 2	343	MeOH 24 h, 130 °C 12 h
<b>PIM-1</b> (402 d)	46 ± 1	402	130 °C 12 h (labeled as “untreated”)
<b>PIM-COOH</b> (330 d)	34.2 ± 0.8	330	130 °C 12 h (labeled as “untreated”)

**Table S2.** Critical temperature for gases considered in this work<sup>1</sup>.

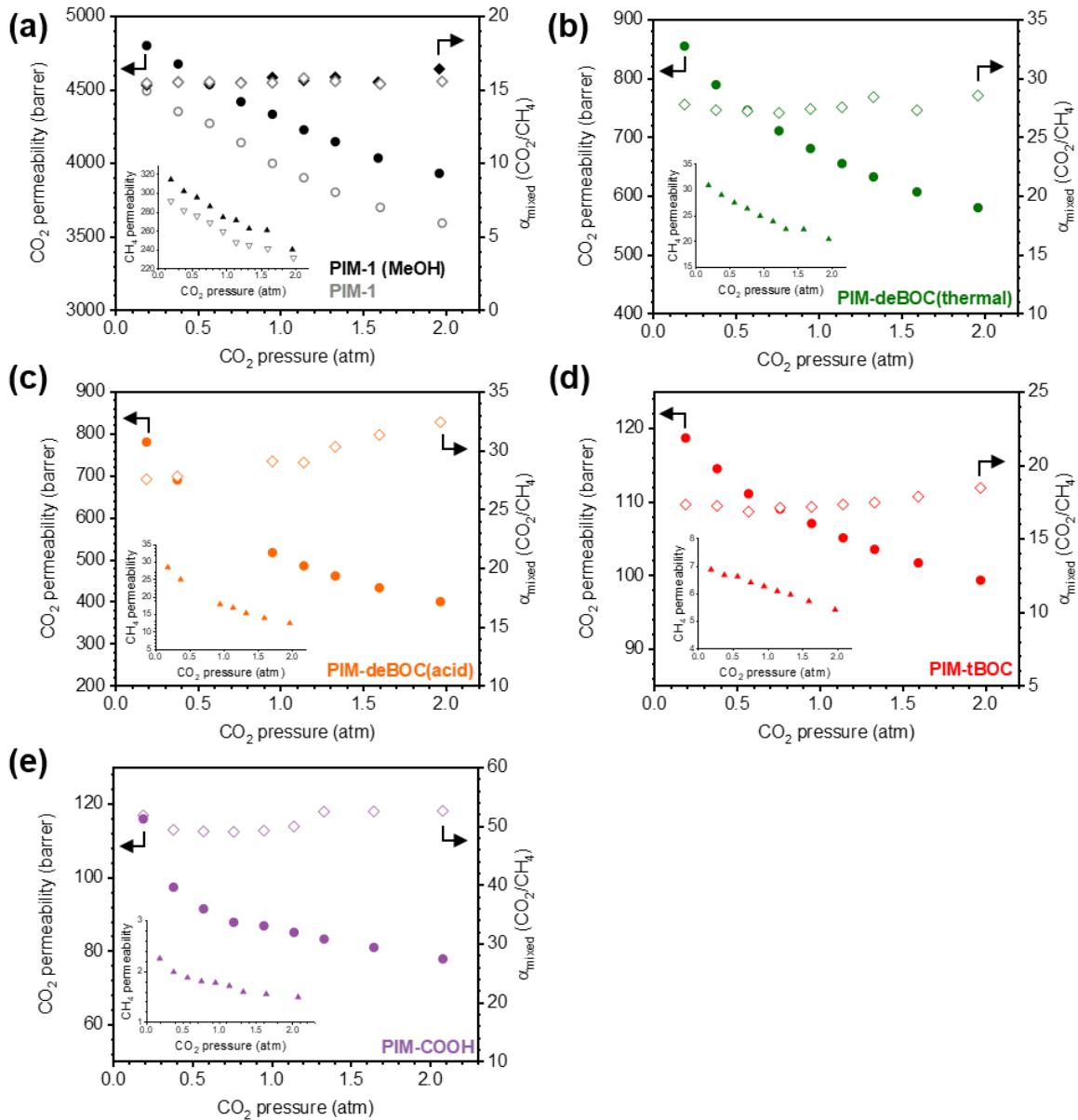
Gas	$T_c$ (K)
H <sub>2</sub>	33.2
N <sub>2</sub>	126.2
CH <sub>4</sub>	190.5
CO <sub>2</sub>	304.1

## 2. Physical aging upper bound

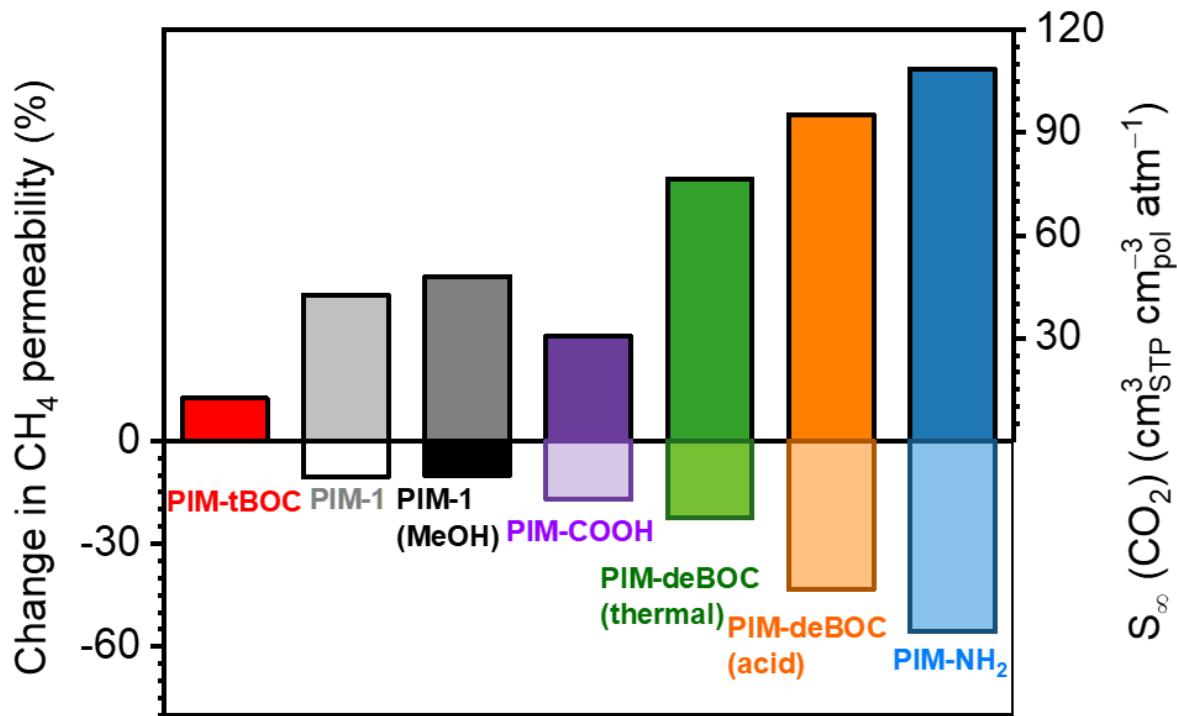


**Fig. S1.** Upper bound performance for freshly treated films (half-filled symbols) and aged films (filled symbols) tested at 1 atm and 35 °C for CO<sub>2</sub>/CH<sub>4</sub> separation. The 2008 pure-gas upper bound<sup>2</sup> and 2018 mixed-gas upper bound<sup>3</sup> are shown for reference.

### 3. Variable-pressure mixed-gas permeation

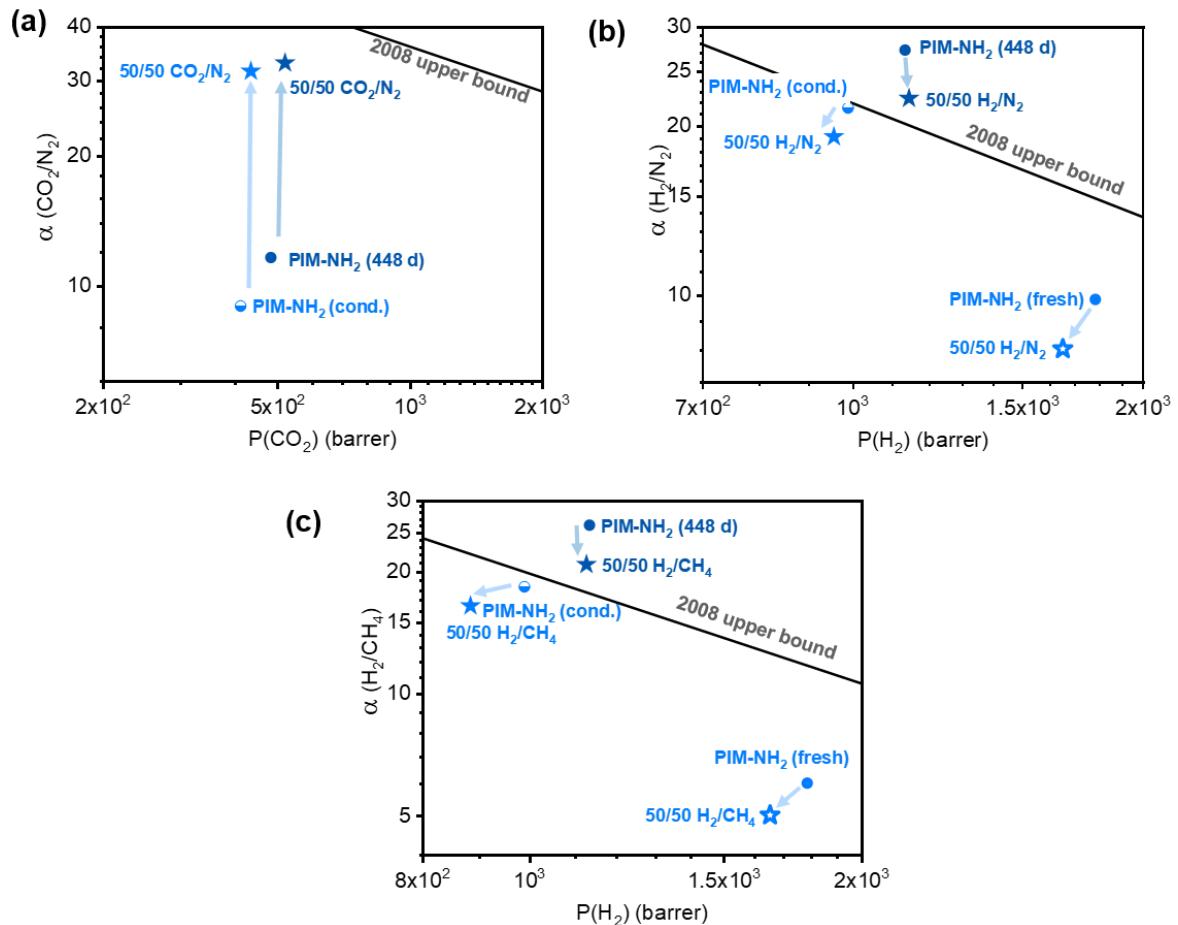


**Fig. S2.**  $\text{CO}_2$  mixed-gas permeabilities (filled circles) and  $\text{CO}_2/\text{CH}_4$  selectivities (open diamonds) for incremental  $\text{CO}_2$  compositions ranging from 10% to 90%  $\text{CO}_2$  tested at a total pressure of approximately 2 atm for (a) treated and untreated PIM-1 (381d and 330d aged, respectively), (b) PIM-deBOC(thermal), (c) PIM-deBOC(acid), (d) PIM-*t*-BOC, and (e) PIM-COOH. Insets correspond to mixed-gas  $\text{CH}_4$  mixed-gas permeabilities (filled triangles) for incremental  $\text{CO}_2$  compositions. Generally, mixed-gas tests were performed at a total pressure of  $2 \pm 0.2$  atm.



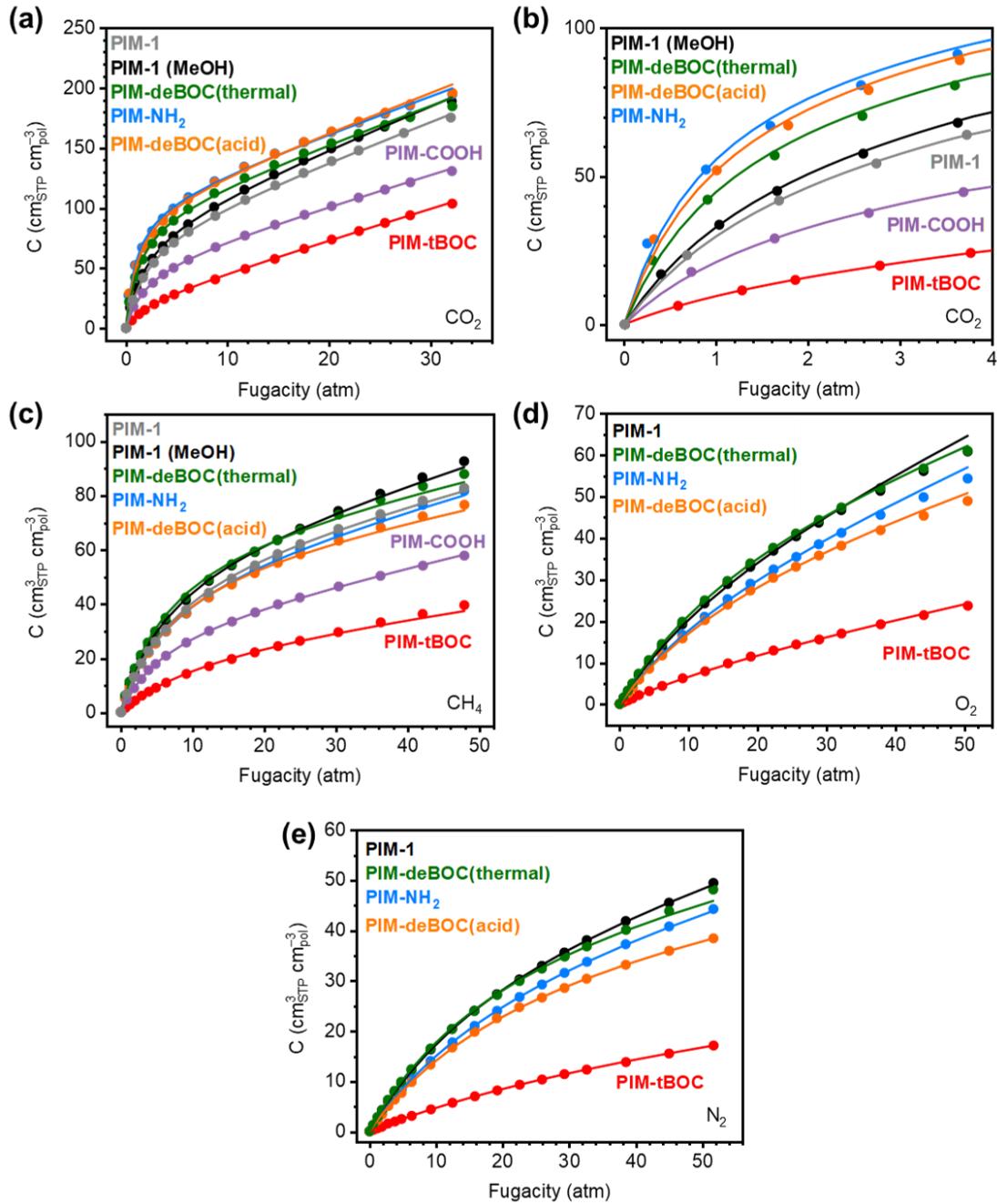
**Fig. S3** Percent decrease in mixed-gas vs pure-gas CH<sub>4</sub> permeability measured at a CO<sub>2</sub> partial pressure of 1.2 atm compared to pure-gas sorption at infinite dilution for the aged PIM-1 derivatives considered in this study. No significant change was observed for PIM-tBOC.

#### 4. Other mixtures ( $\text{CO}_2/\text{N}_2$ , $\text{H}_2/\text{N}_2$ , and $\text{H}_2/\text{CH}_4$ )



**Fig. S4.** (a)  $\text{CO}_2/\text{N}_2$ , (b)  $\text{H}_2/\text{N}_2$ , and (c)  $\text{H}_2/\text{CH}_4$  upper bound plots for aged PIM-NH<sub>2</sub> (448 d) and an aged PIM-NH<sub>2</sub> sample previously conditioned at high  $\text{CO}_2$  pressure (PIM-NH<sub>2</sub> (cond.)). Circles indicate pure-gas data and stars indicate mixed-gas data for the relevant mixture measured at 35 °C and 1 atm (pure-gas) and 2 atm total pressure (mixed-gas).  $\text{H}_2/\text{CH}_4$  and  $\text{H}_2/\text{N}_2$  plots include data for a freshly methanol treated PIM-NH<sub>2</sub> film.

## 5. Pure-gas sorption



**Fig. S5.** (a and b)  $\text{CO}_2$ , (c)  $\text{CH}_4$ , (d)  $\text{O}_2$ , and (e)  $\text{N}_2$  sorption isotherms and dual-mode sorption fitting for methanol treated PIM-1, PIM-deBOC(thermal), PIM-NH<sub>2</sub>, PIM-deBOC(acid) and PIM-tBOC, measured at 35 °C. Reference  $\text{CO}_2$  and  $\text{CH}_4$  isotherms for untreated PIM-1 and PIM-COOH are also included in (a), (b), and (c)<sup>4</sup>.

**Table S3.** Dual mode sorption (DMS) model parameters and  $S_{\infty}$  values calculated for all samples.

Parameter <sup>a</sup>	Gas	PIM-1 <sup>b</sup>	PIM-NH <sub>2</sub>	PIM-t-BOC	PIM-deBOC (thermal)	PIM-deBOC (acid)	PIM-1 <sup>c, 4</sup>	PIM-COOH <sup>d, 4</sup>
<b><i>b</i></b> (atm <sup>-1</sup> )	N <sub>2</sub>	0.042	0.039	0.017	0.044	0.038	0.041	0.041
	O <sub>2</sub>	0.054	0.054	0.042	0.046	0.047	-	-
	CH <sub>4</sub>	0.142	0.147	0.076	0.146	0.137	0.130	0.157
	CO <sub>2</sub>	0.518	1.006	0.427	0.768	0.895	0.491	0.516
<b><i>k<sub>D</sub></i></b> (cm <sup>3</sup> <sub>STP</sub> cm <sup>-3</sup> <sub>pol</sub> atm <sup>-1</sup> )	N <sub>2</sub>	0.32	0.29	0.10	0.19	0.16	0.24	0.24
	O <sub>2</sub>	0.46	0.42	0.17	0.30	0.26	-	-
	CH <sub>4</sub>	0.75	0.68	0.33	0.52	0.48	0.61	0.57
	CO <sub>2</sub>	3.5	3.1	2.6	3.1	3.2	3.2	2.6
<b><i>C'<sub>H</sub></i></b> (cm <sup>3</sup> <sub>STP</sub> cm <sup>-3</sup> <sub>pol</sub> )	N <sub>2</sub>	48.1	43.4	25.6	52.2	45.8	46.0	26.5
	O <sub>2</sub>	36.4	31.0	11.0	51.3	38.1	-	-
	CH <sub>4</sub>	62.4	54.3	27.3	68.5	59.7	60.9	34.8
	CO <sub>2</sub>	86.0	104.9	23.2	95.7	102.9	80.4	54.2
<b><i>S<sup>∞</sup></i></b> (cm <sup>3</sup> <sub>STP</sub> cm <sup>-3</sup> <sub>pol</sub> atm <sup>-1</sup> )	N <sub>2</sub>	2.32	1.97	0.525	2.46	1.88	2.13	1.34
	O <sub>2</sub>	2.43	2.10	0.629	2.65	2.06	-	-
	CH <sub>4</sub>	9.64	8.65	2.41	10.5	8.64	8.54	6.05
	CO <sub>2</sub>	48.0	109	12.5	76.6	95.3	42.7	30.5

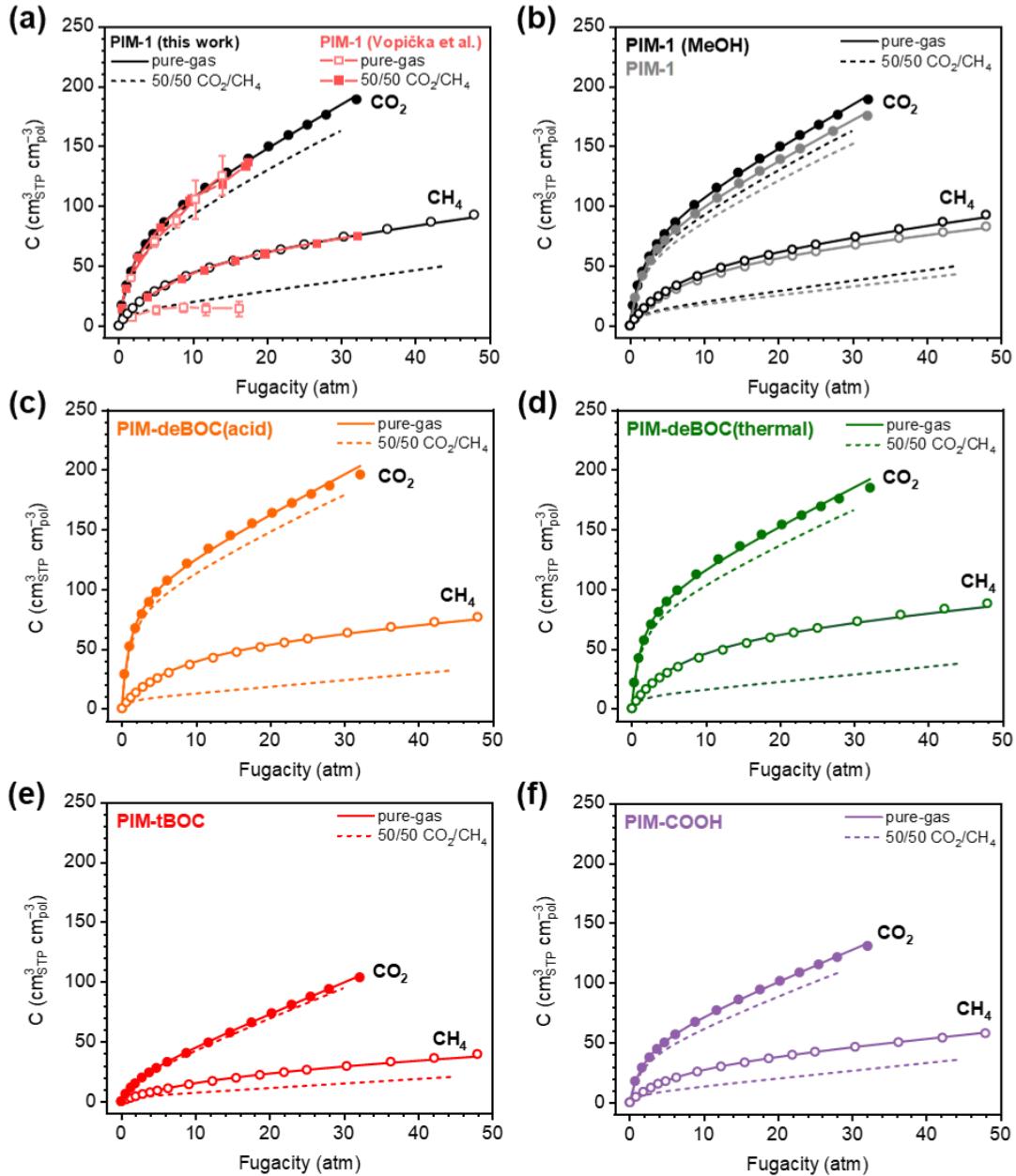
<sup>a</sup>DMS parameters were calculated by constraining the slope of the ln( $k_D$ ) versus critical temperature ( $T_c$ ) trend to equal the slope of ln( $S_{10atm}$ ) versus  $T_c$ <sup>5</sup>.

<sup>b</sup>PIM-1 film soaked in methanol for 24 h then vacuum-dried at 130 °C for 12 h.

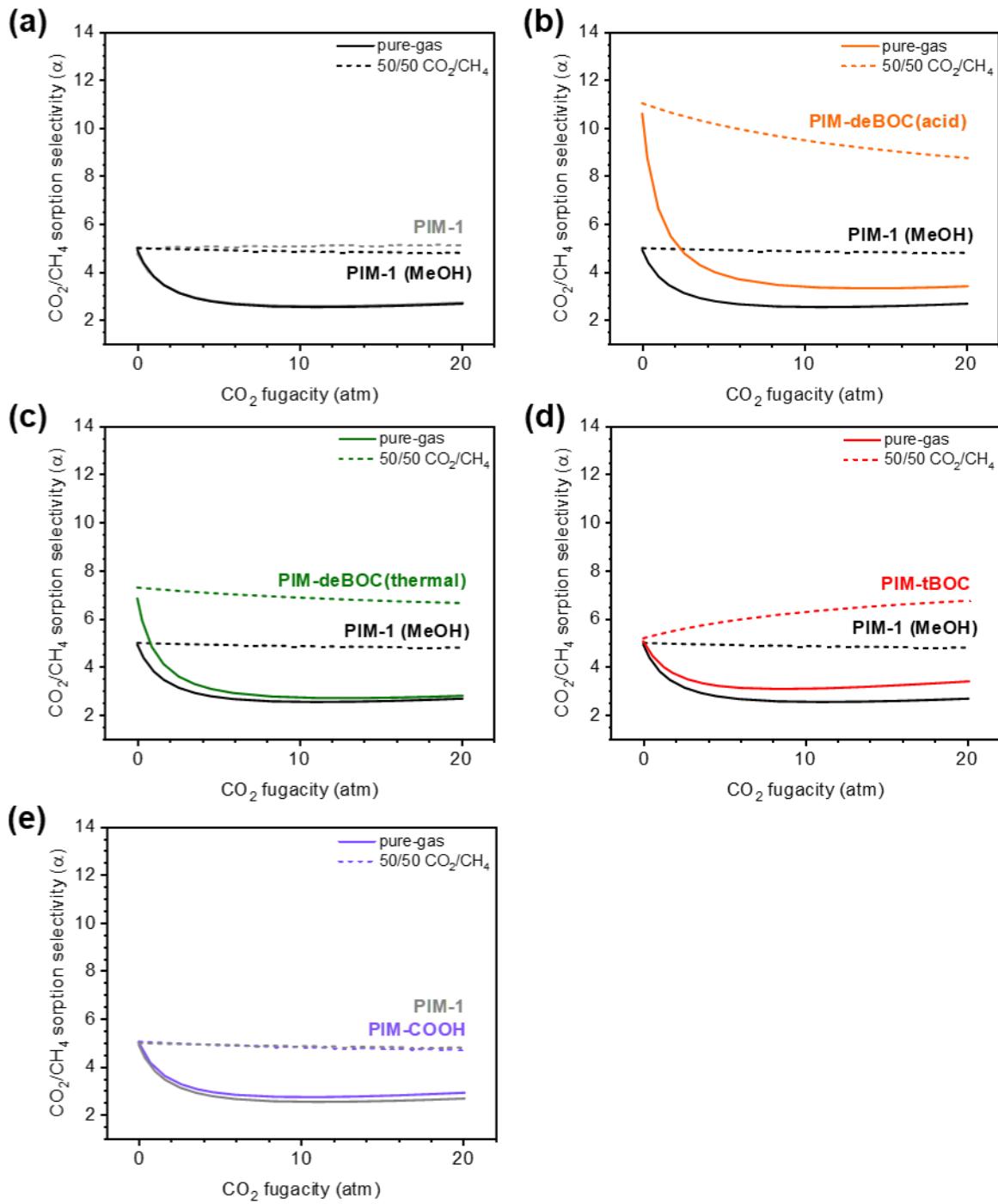
<sup>c</sup>PIM-1 film vacuum-dried at 130 °C for 12 h, as reported elsewhere<sup>4</sup>.

<sup>d</sup>PIM-COOH film vacuum-dried at 130 °C for 12 h, as reported elsewhere<sup>4</sup>.

## 6. Mixed-gas sorption predictions



**Fig. S6.** Pure-gas CO<sub>2</sub> and CH<sub>4</sub> experimental sorption isotherms, represented as filled and open symbols, respectively. Pure-gas DMS model fittings (solid lines), and mixed-gas CO<sub>2</sub> and CH<sub>4</sub> sorption predictions (dashed lines) for (a) methanol treated PIM-1, (b) methanol treated and untreated PIM-1, (c) PIM-deBOC(acid), (d) PIM-deBOC(thermal), (e) PIM-*t*-BOC, and (f) PIM-COOH. Fig. S6a includes experimental pure- and 50/50 CO<sub>2</sub>/CH<sub>4</sub> mixed-gas data previously reported for PIM-1 by Vopička et al.<sup>6</sup>



**Fig. S7.** Pure-gas (solid lines) and 50/50 mixed-gas (dashed lines)  $\text{CO}_2/\text{CH}_4$  sorption selectivities for (a) methanol treated and untreated PIM-1, (b) PIM-deBOC(acid), (c) PIM-deBOC(thermal), (d) PIM-*t*-BOC, and (e) PIM-COOH.

**Table S4.** Pure- and mixed-gas diffusion and sorption selectivities for aged samples determined at CO<sub>2</sub> partial fugacities of 1 atm.

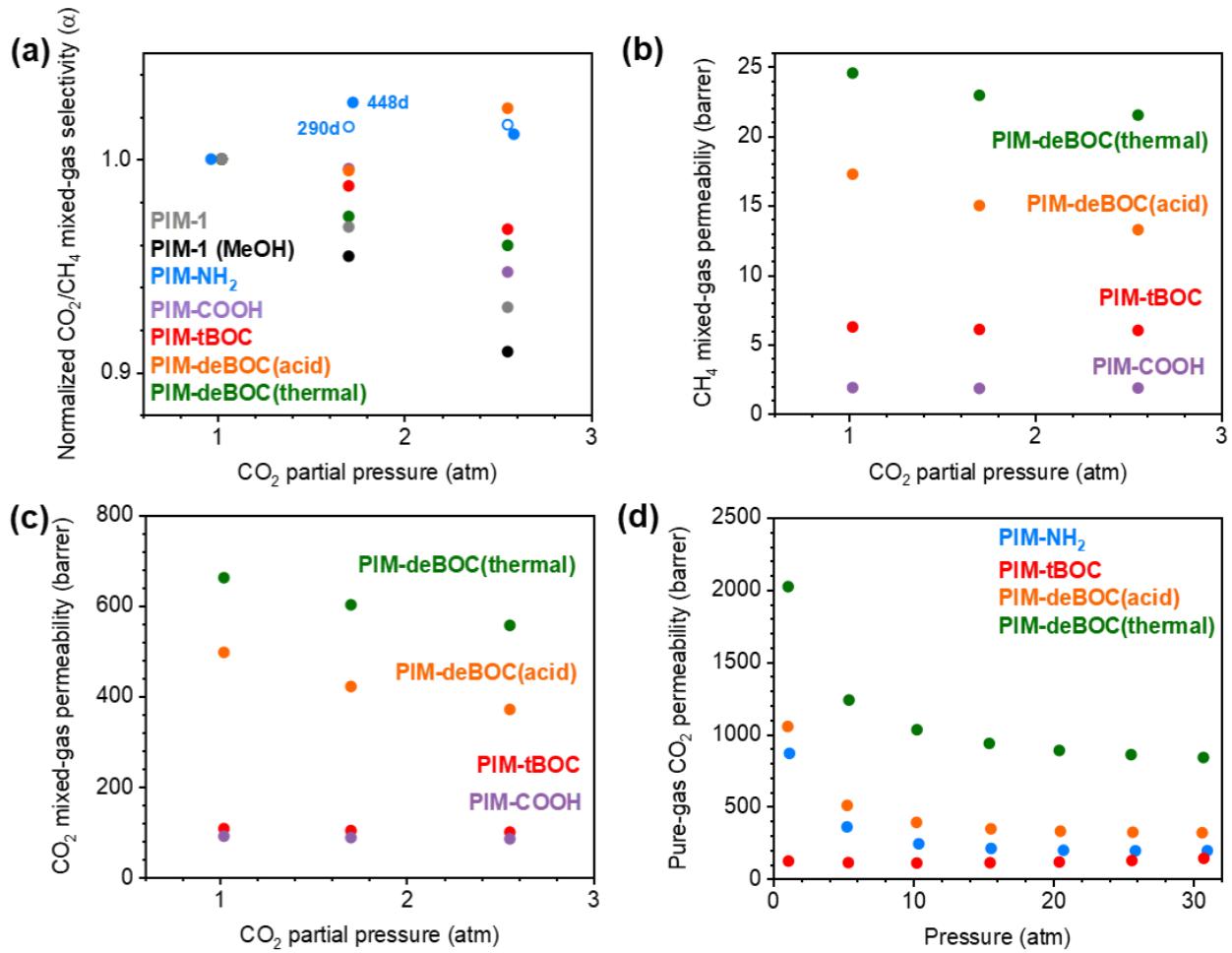
Polymer	Diffusion selectivity ( $\alpha_D$ ) <sup>a</sup>		Sorption selectivity ( $\alpha_S$ )	
	$\alpha_D^{pure}$	$\alpha_D^{mixed}$	$\alpha_S^{pure}$ <sup>b</sup>	$\alpha_S^{mixed}$ <sup>c</sup>
<b>PIM-1</b> (MeOH, 381 d)	3.5 ± 0.3	3.2 ± 0.2	3.8 ± 0.1	5.0 ± 0.2
<b>PIM-1</b> (402 d)	3.6 ± 0.2	3.1 ± 0.2	3.8 ± 0.2	5.0 ± 0.2
<b>PIM-NH<sub>2</sub></b> (290 d)	1.8 ± 0.2	2.5 ± 0.2	7.1 ± 0.3	11.9 ± 0.5
<b>PIM-deBOC(acid)</b>	2.4 ± 0.1	2.7 ± 0.2	6.6 ± 0.1	10.8 ± 0.2
<b>PIM-deBOC(thermal)</b>	4.4 ± 0.4	3.8 ± 0.3	4.6 ± 0.2	7.2 ± 0.3
<b>PIM-tBOC</b>	4.5 ± 0.9	3.2 ± 0.6	4.1 ± 0.2	5.4 ± 0.3
<b>PIM-COOH</b> (330 d)	9.8 ± 0.8	9.8 ± 0.8	4.0 ± 0.1	5.0 ± 0.2

<sup>a</sup>Calculated using the sorption–diffusion model and pure-gas sorption values or predicted mixed-gas sorption values.

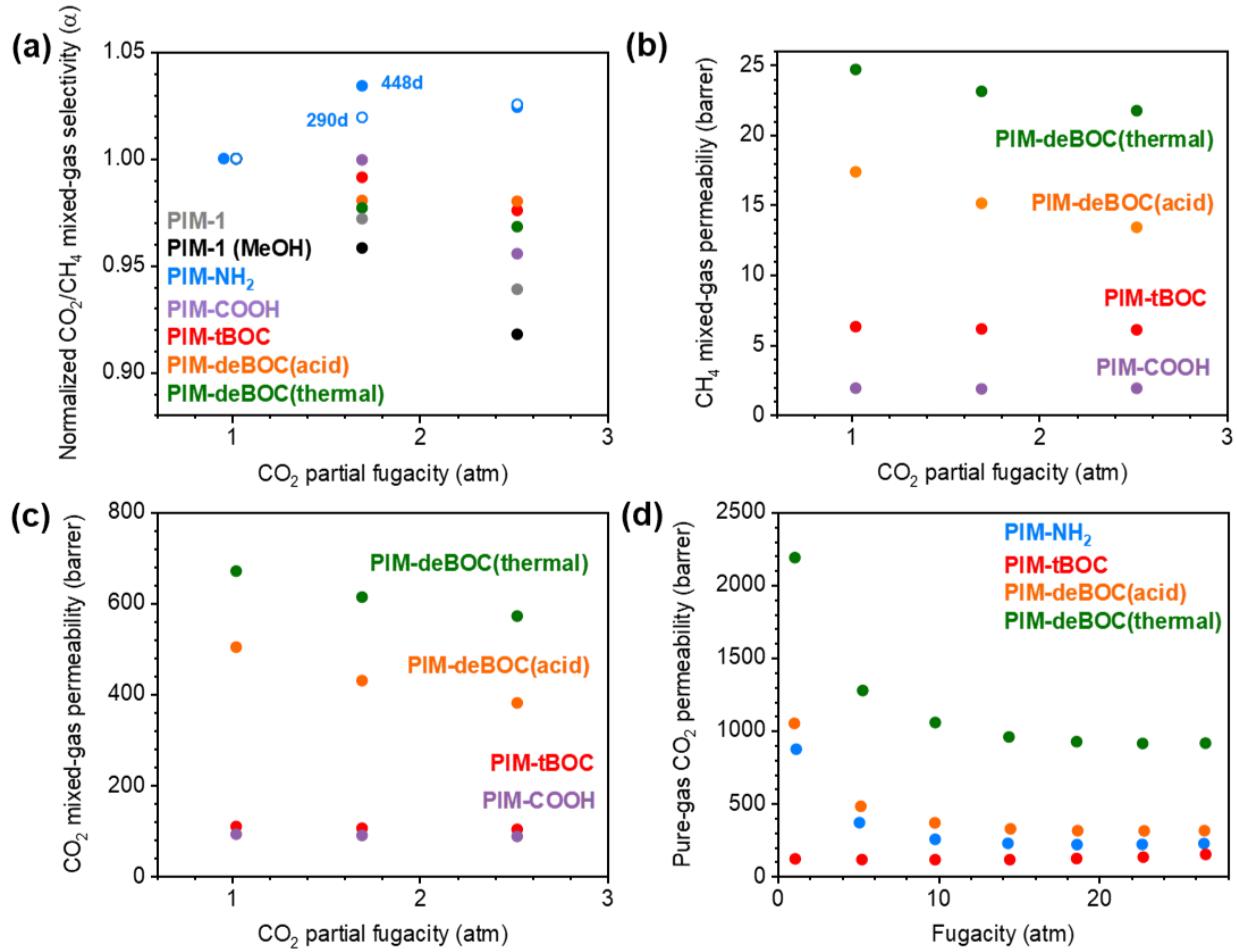
<sup>b</sup>Calculated from DMS curves calculated with best-fit (i.e., unconstrained) parameters.

<sup>c</sup>Calculated from mixed-gas DMS curves predicted with constrained DMS parameters.

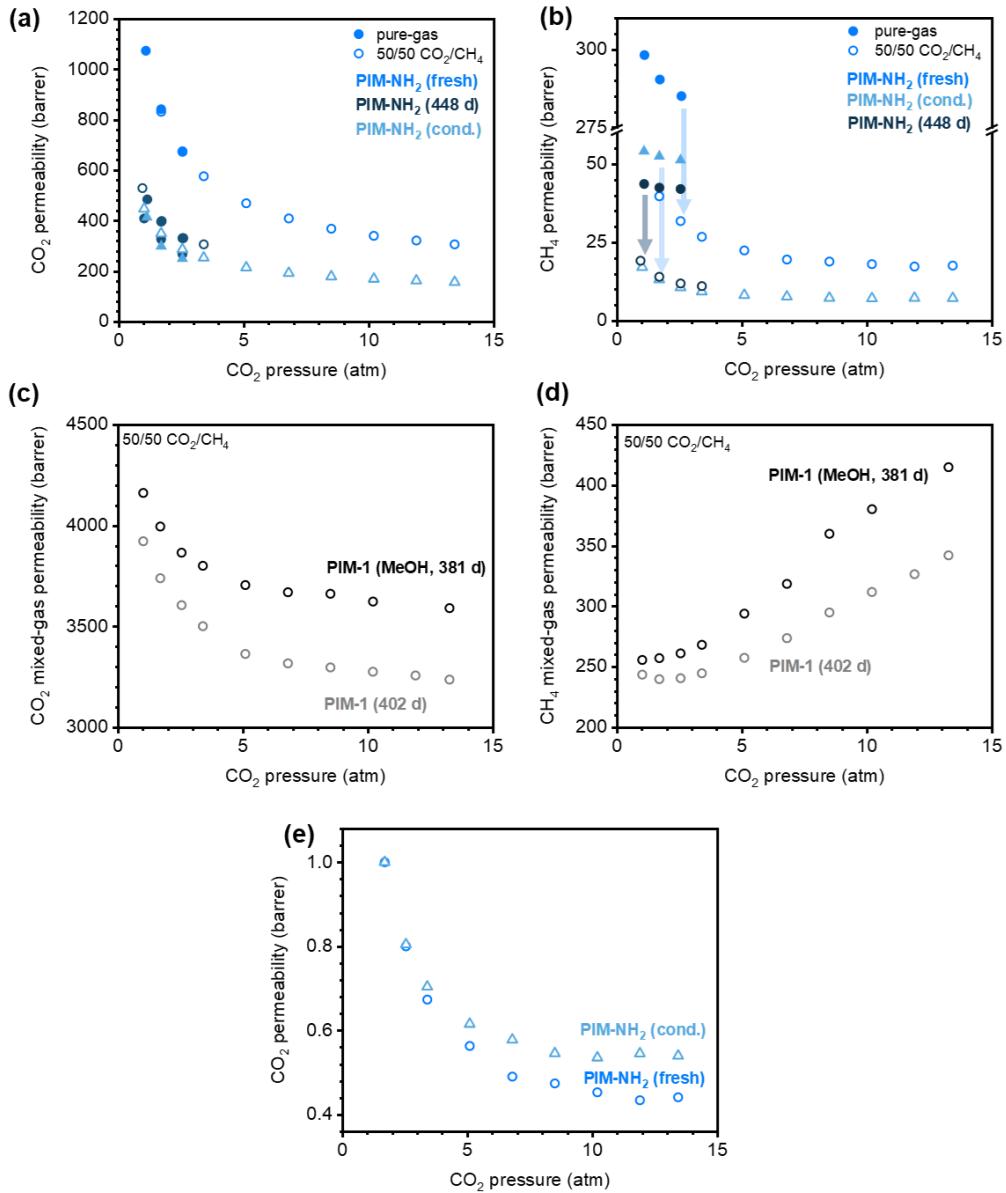
## 7. Pure and mixed-gas plasticization



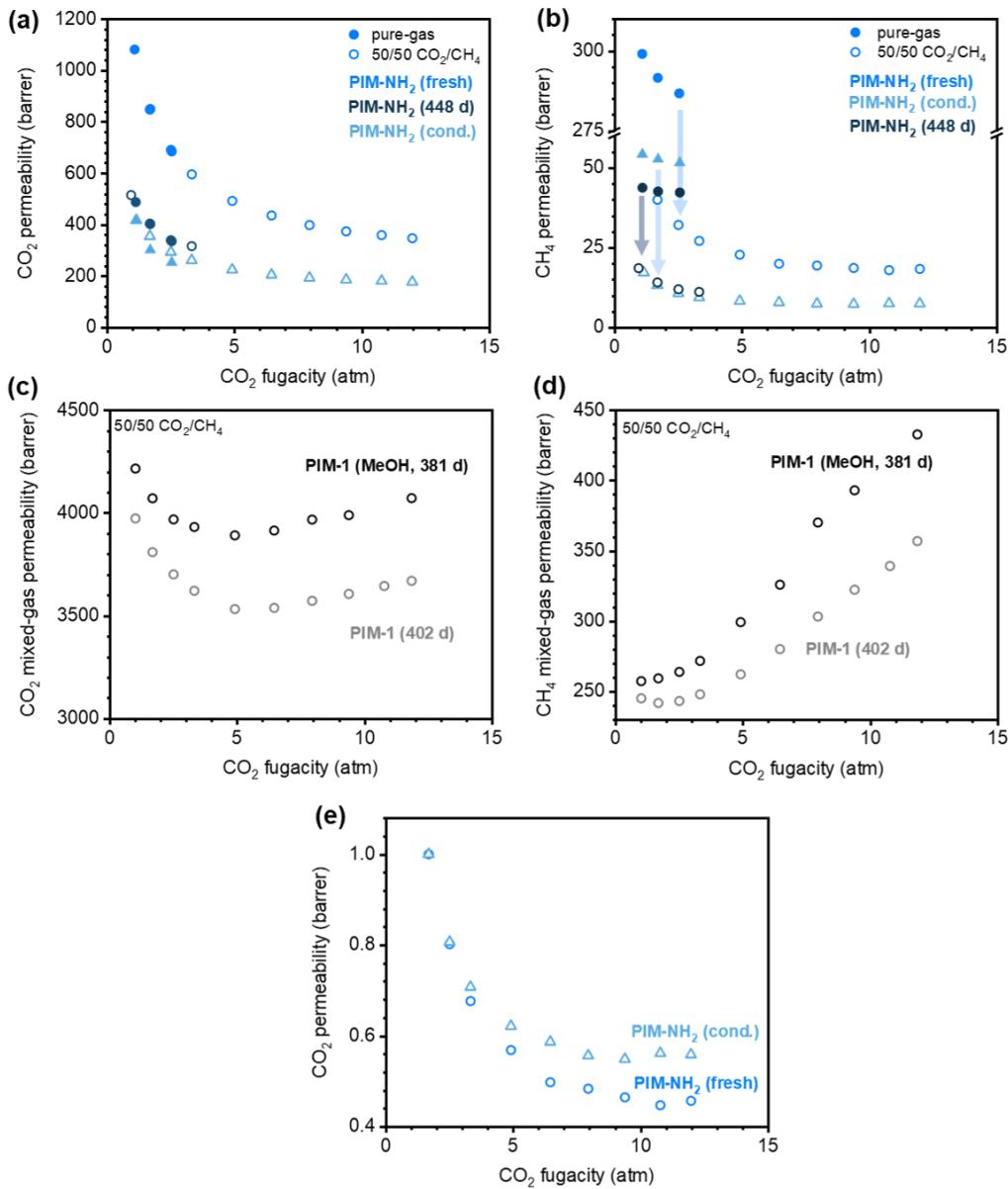
**Fig. S8** (a) Normalized  $\text{CO}_2/\text{CH}_4$  mixed-gas selectivities. Filled and open blue circles indicate data for PIM-NH<sub>2</sub> (448d) and PIM-NH<sub>2</sub> (290d), respectively. (b)  $\text{CH}_4$  mixed-gas permeabilities, and (c)  $\text{CO}_2$  mixed-gas permeabilities for PIMs considered in this work. Mixed-gas measurements were performed with 50/50  $\text{CO}_2/\text{CH}_4$  compositions at 35 °C. (d) Pure-gas  $\text{CO}_2$  permeability for fresh PIM-NH<sub>2</sub>, PIM-t-BOC, PIM-deBOC(acid), and PIM-deBOC(thermal) tested at 35 °C.



**Fig. S9.** *Fugacity-based:* (a) Normalized  $\text{CO}_2/\text{CH}_4$  mixed-gas selectivities. Filled and open blue circles indicate data for PIM-NH<sub>2</sub> (448d) and PIM-NH<sub>2</sub> (290d), respectively. (b)  $\text{CH}_4$  mixed-gas permeabilities, and (c)  $\text{CO}_2$  mixed-gas permeabilities for PIMs considered in this work. Mixed-gas measurements were performed with 50/50  $\text{CO}_2/\text{CH}_4$  compositions at 35 °C. (d) Pure-gas  $\text{CO}_2$  permeability for fresh PIM-NH<sub>2</sub>, PIM-t-BOC, PIM-deBOC(acid), and PIM-deBOC(thermal) tested at 35 °C.

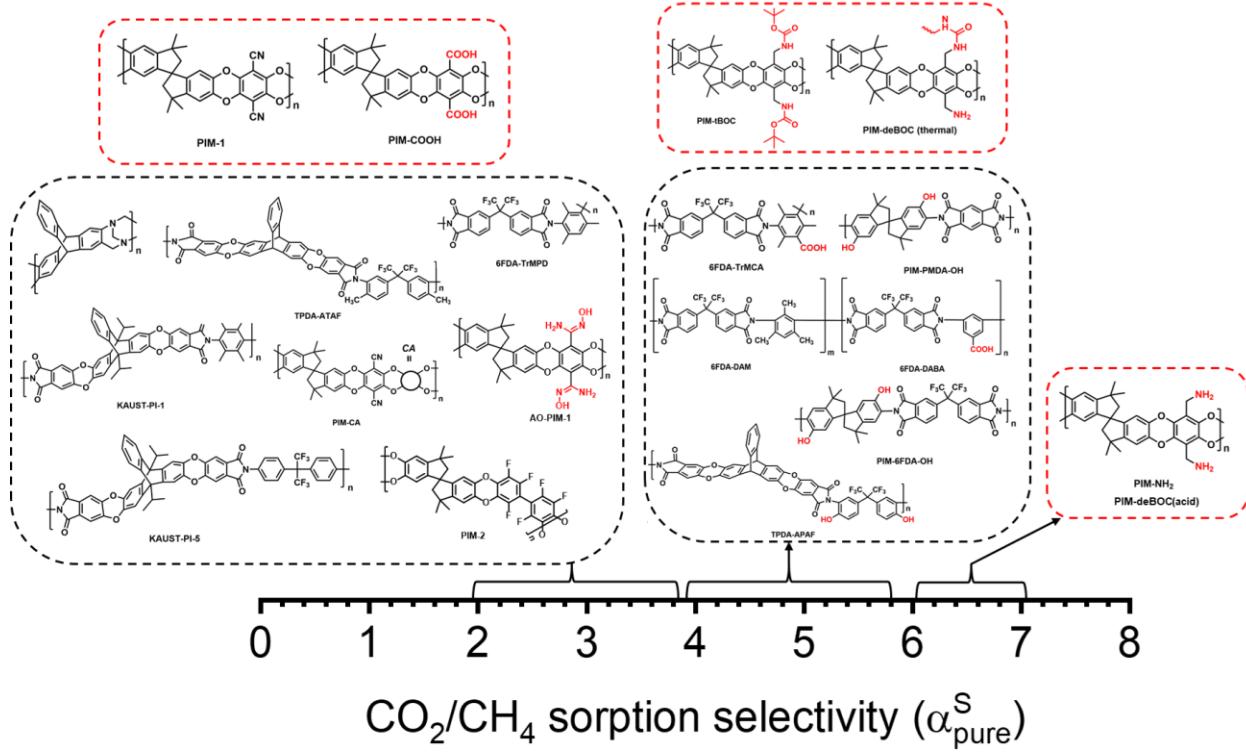


**Fig. S10.** Pure- (filled circles) and mixed-gas (open circles) (a) CO<sub>2</sub> and (b) CH<sub>4</sub> permeabilities for a PIM-NH<sub>2</sub> (fresh), PIM-NH<sub>2</sub> (448 d), and PIM-NH<sub>2</sub> (cond.). Arrows indicate the reduction in CO<sub>2</sub> or CH<sub>4</sub> permeabilities from pure-gas to mixed-gas tests. Mixed-gas (c) CO<sub>2</sub> and (d) CH<sub>4</sub> permeabilities for methanol treated (black) and untreated (gray) PIM-1 aged for 381 days and 402 days, respectively. (e) Normalized mixed-gas CH<sub>4</sub> permeabilities for PIM-NH<sub>2</sub> (fresh) and PIM-NH<sub>2</sub> (cond.) starting from a CO<sub>2</sub> partial pressure of 1.75 atm. Permeabilities were calculated with respect to pressure.



**Fig. S11. Fugacity-based:** Pure- (filled circles) and mixed-gas (open circles) (a) CO<sub>2</sub> and (b) CH<sub>4</sub> permeabilities for a PIM-NH<sub>2</sub> (fresh), PIM-NH<sub>2</sub> (448 d), and PIM-NH<sub>2</sub> (cond.). Arrows indicate the reduction in CO<sub>2</sub> or CH<sub>4</sub> permeabilities from pure-gas to mixed-gas tests. Mixed-gas (c) CO<sub>2</sub> and (d) CH<sub>4</sub> permeabilities for methanol treated (black) and untreated (gray) PIM-1 aged for 381 days and 402 days, respectively. (e) Normalized mixed-gas CH<sub>4</sub> permeabilities for PIM-NH<sub>2</sub> (fresh) and PIM-NH<sub>2</sub> (cond.) starting from a CO<sub>2</sub> partial fugacity of 1.75 atm. Permeabilities were calculated with respect to fugacity.

## 8. Reference literature data



**Fig. S12.** Pure-gas  $\text{CO}_2/\text{CH}_4$  sorption selectivity and relevant chemical structures for the samples considered in **Fig. 5a**. Chemical functionality with potential for hydrogen bonding are denoted in red. Polymers from this work are enclosed in red dashed boxes. Structures with sorption and mixed-gas data taken from literature are enclosed in black dotted boxes, including PIM-2<sup>7</sup>, PIM-6FDA-OH and PIM-PMDA-OH<sup>8</sup>, PIM-Trip-TB<sup>9</sup>, 6FDA-DAM:DABA (3:2, 1:1, 1:2)<sup>10</sup>, KAUST-PI-1 and KAUST-PI-5<sup>11</sup>, TPDA-APAF and TPDA-ATAF<sup>12</sup>, 6FDA-TrMPD and 6FDA-TrMCA<sup>13</sup>, AO-PIM-1<sup>14,15</sup>, PIM-CA (0.5, 1, 2, 3, and 10%)<sup>16</sup>, and PIM-1<sup>16–19</sup>. Post-synthetically treated PIM-1 films were also plotted in **Fig. 5**, including PIM-UV (30 min)<sup>20</sup>, TOX-PIM-1<sup>19</sup>.

**Table S5.** Reference pure- and mixed-gas permeation data for  $\text{CO}_2/\text{CH}_4$ .

Polymer	T (°C)	P <sub>Total</sub> (psia)	$\text{CO}_2/\text{CH}_4$ Comp.	$P_{\text{CH}_4}$ (barrier)	$P_{\text{CO}_2}$ (barrier)	$\alpha_{\text{CO}_2/\text{CH}_4}^{\text{pure}}$	$\alpha_{\text{CO}_2/\text{CH}_4}^{\text{mixed}}$	Ref.
$P_{\text{Total}} = 29 \text{ psia}$								
PIM-2	25	14	pure	203.9	2960.9	14.5	-	7
PIM-2	-	29	35/65	457.2	6229.7	-	13.6	
PIM-Trip-TB	-	15	pure	310.5	4109.2	13.2	-	9

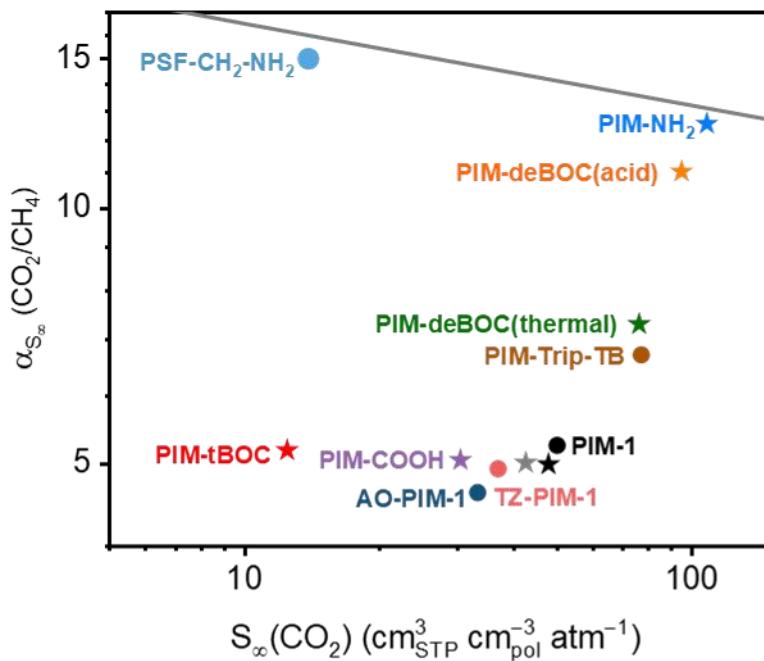
PIM-Trip-TB	-	29	50/50	299.4	3913.7	-	13.1	
PIM-1	30	29	pure	363	4533	12.5	-	21
PIM-1	30	29	30/70	378.5	4391	-	11.6	
PIM-1 (18.8% ZIF-8)	30	29	pure	157	2663	16.9	-	
PIM-1 (18.8% ZIF-8)	30	29	30/70	143	2832	-	19.8	
PIM-Trip-TB	25	14.7	pure	664	8616	12.9	-	22
PIM-Trip-TB	25	29.4	50/50	979	7267	-	7.4	
CoPIM-TB-1	25	14.7	pure	575	7835	13.6	-	
CoPIM-TB-1	25	29.4	50/50	654	6271	-	9.6	
CoPIM-TB-2	25	14.7	pure	448	6767	15.1	-	
CoPIM-TB-2	25	29.4	50/50	536	5818	-	10.9	
C-CoPIM-TB-1	25	14.7	pure	233	5437	23.3	-	
C-CoPIM-TB-1	25	29.4	50/50	264	5241	-	19.9	
C-CoPIM-TB-2	25	14.7	pure	169	4251	25.2	-	
C-CoPIM-TB-2	25	29.4	50/50	184	3931	-	21.4	
CF <sub>3</sub> -ROMP	35	14.5	pure	644	6377	9.9	-	23
CF <sub>3</sub> -ROMP	35	29	50/50	779	7063	-	9.1	
CF <sub>3</sub> -ROMP	35	14.5	pure	1086	8867	8.2	-	
CF <sub>3</sub> -ROMP	35	29	50/50	1151	9266	-	8.1	
CF <sub>3</sub> -ROMP	35	14.5	pure	2368	13418	5.7	-	
CF <sub>3</sub> -ROMP	35	29	50/50	2183	15036	-	6.89	
PIM-MP-TB	35	14.5	pure	26	633	24.4	-	24
PIM-MP-TB	25	29	52.1/47.9	36	766	-	21.3	
PIM-6FDA-OH	35	14.5	pure	9.1	263	28.9	-	8
PIM-6FDA-OH	22	29	50/50	5.3	223	-	42.1	
PIM-PMDA-OH	35	14	pure	7.7	198	25.7	-	
PIM-PMDA-OH	22	29	50/50	3	101	-	33.7	
PIM-TMN-Trip	25	14.6	pure	722	10910	15.1	-	25
PIM-TMN-Trip	25	29.1	60/40	489	11300	-	23.1	
<b>P<sub>Total</sub> = 43.5 psia</b>								
PIM-2	25	14.5	pure	203.9	2960.9	14.5	-	7
PIM-2	-	43.5	35/65	457.2	5898.6	-	12.9	
PIM-TMN-Trip	25	29.1	pure	735	9773	13.3	-	25
PIM-TMN-Trip	25	43.5	60/40	566	11390	-	20.1	
PIM-EA(H <sub>2</sub> )-TB	-	14.5	pure	62.6	1391	22.2	-	26
PIM-EA(H <sub>2</sub> )-TB	-	43.5	35/65	70.9	1373.5	-	19.4	
PIM-SBF-1	25	14.5	pure	102	2410	23.6	-	27
PIM-SBF-1	-	43.5	35/65	89.9	2780	-	30.9	
PIM-SBF-5	25	14.5	pure	925	10000	10.8	-	
PIM-SBF-5	-	43.4	35/65	932	11100	-	11.9	
<b>P<sub>Total</sub> = 58 psia</b>								
6FA-DAM:DABA 3:2	35	24.6	pure	4.2	185.9	44.3	-	10
6FA-DAM:DABA 3:2	35	61.5	50/50	3.3	178.6	-	54.1	

6FDA-DAM:DABA 1:1	35	22.9	pure	2.2	110.9	51.5	-	
6FDA-DAM:DABA 1:1	35	50.5	50/50	1.9	113.3	-	59.6	
6FDA-DAM:DABA 1:2	35	21.3	pure	0.8	41.7	54.2	-	
6FDA-DAM:DABA 1:2	35	52.7	50/50	0.5	43.1	-	86.2	
PIM-2	25	14.5	pure	203.9	2960.9	14.5	-	7
PIM-2	-	58	35/65	465.6	5636.2	-	12.1	
6FDA-mPDA	35	29.4	pure	0.2	14	70	-	
6FDA-mPDA	35	58	50/50	0.2	14.7	-	67.9	
6FDA-DAP	35	29.4	pure	0.1	11	91.7	-	
6FDA-DAP	35	58	50/50	0.1	10.5	-	93.2	
6FDA-DAR	35	29.4	pure	0.1	8	94.1	-	
6FDA-DAR	35	58	50/50	0.1	7.5	-	94.8	
TDA1-APAF	35	29	pure	0.7	40	54.8	-	29
TDA1-APAF	35	58	50/50	0.55	37	-	67.3	
TDA1-APAF	35	29	pure	0.4	30	75	-	
TDA1-APAF	35	58	50/50	0.3	28	-	87.6	
PIM-1	22	58	pure	393	5147	13.1	-	30
PIM-1	22	53.6	50/50	324	3897	-	12.0	
TOX/PIM-1	22	58	pure	145	3892	26.8	-	
TOX-PIM-1	22	56.6	50/50	28.3	1101	-	38.9	
TPDA-mPDA	35	29.4	pure	11	349	31.7	-	31
TPDA-mPDA	35	59	50/50	9.7	293.5	-	30.3	
TPDA-DAR	35	29.4	pure	4.7	215	46.2	-	
TPDA-DAR	35	58	50/50	3.9	180.4	-	46.7	
AO-PIM-1	35	29	pure	34	1153	33.9	-	14
AO-PIM-1	35	58	50/50	35.5	847	-	23.9	
PIM-1	35	29	pure	362	5919	16.4	-	
PIM-1	35	58	50/50	427	5630	-	13.2	
KAUST-PI-1	35	29	pure	97.5	2329	23.9	-	11
KAUST-PI-1	35	58	50/50	97.5	2409	-	24.7	
KAUST-PI-5	35	29	pure	79.2	1560	19.7	-	
KAUST-PI-5	35	58	50/50	67.3	1431	-	21.2	
TR PIM-6FDA-OH (440°C)	35	29	pure	54	816	15.1	-	32
TR PIM-6FDA-OH (440°C)	35	58	50/50	44	784	-	17.8	
TPDA-APAF 250	35	29	pure	0.9	46	52.9	-	
TPDA-APAF 250	35	58	50/50	0.6	39	-	60.9	
TPDA-APAF 120	35	29	pure	2.6	99	38.1	-	12
TPDA-APAF 120	35	58	50/50	2.1	89	-	42.4	
TPDA-ATAF 250	35	29	pure	3.8	125	32.8	-	
TPDA-ATAF 250	35	58	50/50	3.6	121	-	33.61	
TPDA-ATAF 120	35	29	pure	11	325	29.6	-	
TPDA-ATAF 120	35	58	50/50	10	299	-	29.9	
6FDA-TrMPD (6FDA-DAM)	35	29	pure	23	498	21.6	-	
								13

6FDA-TrMPD (6FDA-DAM)	35	58	50/50	19.1	448	-	23.4	
6FDA-TrMCA	35	29	pure	3.2	144	45	-	
6FDA-TrMCA	35	60	50/50	2.9	141	-	47.9	
PIM-1	25	29	pure	536	6576	12.3	-	33
PIM-1	25	58	50/50	409	5840	-	14.3	
6FDA-DATRI	35	15	pure	6.2	189	30.5	-	34
6FDA-DATRI	35	58	50/50	-	-	-	28.7	
6FDA-DAT1-OH	35	29	pure	1.4	70	50.0	-	35
6FDA-DAT1-OH	35	58	50/50	1.51	69.5	-	46.0	
6FDA-DAT1	35	29	pure	2.82	102.25	36.3	-	36
6FDA-DAT1	35	58	50/50	2.52	96	-	38.1	
6FDA-DAT2	35	29	pure	5.22	155.75	29.8	-	
6FDA-DAT2	35	58	50/50	4.64	138.75	-	29.9	
PIM-TMN-Trip	25	44	pure	730	10880	14.9	-	25
PIM-TMN-Trip	25	58	60/40	594	10730	-	18.1	
<b>P<sub>Total</sub> = 58 psia</b>								
cPIM-1/Matrimid (10:90)	35	51	pure	0.6	17.9	32.5	-	37
cPIM-1/Matrimid (10:90)	35	103	50/50	0.5	13.9	-	28.4	
cPIM-1/Matrimid (30:70)	35	51	pure	1.7	48.7	27.9	-	
cPIM-1/Matrimid (30:70)	35	103	50/50	1.8	43.2	-	24.3	
cPIM-1/Matrimid (50:50)	35	51	pure	6.1	145	23.8	-	
cPIM-1/Matrimid (50:50)	35	103	50/50	6.7	131	-	19.6	
cPIM-1/Matrimid (70:30)	35	51	pure	25	486	19.4	-	
cPIM-1/Matrimid (70:30)	35	103	50/50	22.8	417	-	18.3	
cPIM-1/Matrimid (90:10)	35	51	pure	57	982	17.2	-	
cPIM-1/Matrimid (90:10)	35	103	50/50	53.5	905	-	16.9	
PIM-300-2.0d	35	51	pure	73	4000	54.8	-	38
PIM-300-2.0d	35	100	50/50	42.9	2317	-	54.0	
PIM-UV 20min	35	50	pure	62.1	1869	30.1	-	
PIM-UV 20min	35	100	50/50	61.3	1554	-	25.4	
PIM-UV 30min	35	50	pure	23.1	724	31.3	-	20
PIM-UV 30min	35	100	50/50	20.4	724	-	35.5	
PIM-1	22	58.0	pure	397	5135	12.9	-	
PIM-1	22	116	50/50	298	3020	-	10.1	
TOX-PIM-1	22	58	pure	15.9	1100	69.2	-	19
TOX-PIM-1	22	116	50/50	20.4	890	-	43.6	
TOX-PIM-1	22	58	pure	58	1956	33.7	-	
TOX-PIM-1	22	116	50/50	38.2	1390	-	36.4	
Ultem/PIM-1 (90/10)	35	50	pure	0.1	4	32.9	-	39
Ultem/PIM-1 (90/10)	35	100	50/50	0.1	3.4	-	37.4	
cPIM-1/Torlon (10/90)	35	51	pure	-	1.2	39.8	-	
cPIM-1/Torlon (10/90)	35	102	50/50	-	1.1	-	39.3	40
cPIM-1/Torlon (30/70)	35	51	pure	0.1	4.8	34.4	-	

cPIM-1/Torlon (30/70)	35	102	50/50	0.1	4.5	-	34.8	
cPIM-1/Torlon (50/50)	35	51	pure	0.7	21.4	30.6	-	
cPIM-1/Torlon (50/50)	35	102	50/50	0.7	19.5	-	29.9	
AO-PIM-1	35	14	pure	29	980.7	33.8	-	41
AO-PIM-1	35	87	50/50	35	700	-	20	
AO-PIM-1 (9% adamantane)	35	14	pure	82.5	2483.6	30.1	-	
AO-PIM-1 (9% adamantane)	35	87	50/50	73.3	1320	-	18	
PIM-1	30	29	pure	361.7	4521	12.5	-	
PIM-1	30	96	30/70	359.8	4318	-	12	
PIM-1	35	51.4	pure	198	3127	15.8	-	18
PIM-1	35	102	50/50	228	2708	-	11.9	
PIM-CA-0.5%	35	51	pure	267	4388	16.4	-	
PIM-CA-0.5%	35	102	50/50	304	3993	-	13.1	
PIM-CA-1%	35	51	pure	282	4878	17.3	-	
PIM-CA-1%	35	102	50/50	331	4602	-	13.9	
PIM-CA-2%	35	51	pure	232	4469	19.3	-	
PIM-CA-2%	35	102	50/50	283	4226	-	14.9	
PIM-CA-3%	35	51	pure	204	4139	20.3	-	
PIM-CA-3%	35	102	50/50	257	4031	-	15.7	
PIM-CA-10%	35	51	pure	97	2738	28.2	-	16
PIM-CA-10%	35	102	50/50	113	2284	-	20.2	
PIM-1	30	29	pure	363	4521	12.5	-	
PIM-1	30	96	30/70	377	4336	-	11.5	
PIM-EA( $\text{ME}_2$ )-TB	25	29	pure	50	1320	26.4	-	
PIM-EA( $\text{ME}_2$ )-TB	35	88	50/50	118	1626	-	13.8	
PIM-EA( $\text{ME}_2$ )-TB (10% PAF-1)	35	44	pure	833	6329	7.6	-	
PIM-EA( $\text{ME}_2$ )-TB (10% PAF-1)	35	88	50/50	1085	7237	-	6.7	
PIM-EA( $\text{H}_2$ )-TB (10% PAF-1)	35	44	pure	286	4429	15.5	-	
PIM-EA( $\text{H}_2$ )-TB (10% PAF-1)	35	88	50/50	511	4842	-	9.5	
6FDA-DAM:DABA 3:2	35	55	pure	3.98	166	41.7	-	10
6FDA-DAM:DABA 3:2	35	109	50/50	3.3	163	-	49.4	
6FDA-DAM:DABA 1:1	35	52	pure	1.94	102	52.6	-	
6FDA-DAM:DABA 1:1	35	106	50/50	1.9	108	-	56.8	
6FDA-DAM:DABA 1:2	35	50	pure	0.65	37	56.9	-	
6FDA-DAM:DABA 1:2	35	100	50/50	0.45	39	-	86.7	

## 9. Sorption upper bound at infinite dilution



**Fig. S13.** CO<sub>2</sub>/CH<sub>4</sub> 2014 pure-gas sorption upper bound<sup>43</sup> at infinite dilution. Plotted data are AO-PIM-1<sup>15</sup> (dark blue circle), TZ-PIM-1<sup>15</sup> (pink circle), PIM-1<sup>6</sup> (black circle), PIM-Trip-TB<sup>9</sup> (brown circle), amine-functionalized polysulfone<sup>44</sup> (light blue circle), methanol treated PIM-1 (gray star), untreated PIM-1 (black star), PIM-NH<sub>2</sub> (light blue star), PIM-deBOC(acid) (orange star), PIM-deBOC(thermal) (green star), PIM-t-BOC (red star), and PIM-COOH (purple star). Sorption and CO<sub>2</sub>/CH<sub>4</sub> sorption selectivities at infinite dilution were calculated from best-fit DMS modeling of pure-gas sorption data for the six PIM samples included in this work. Literature values for microporous polymers are discussed in work by Genduso et al.<sup>9,45</sup>

## References

- 1 S. Matteucci, Y. Yampolskii, B. D. Freeman and I. Pinna, *Mater. Sci. Membr. Gas Vap. Sep.*, 2006, 1–47.
- 2 L. M. Robeson, *J. Membr. Sci.*, 2008, **320**, 390–400.
- 3 Y. Wang, X. Ma, B. S. Ghanem, F. Alghunaimi, I. Pinna and Y. Han, *Mater. Today Nano*, 2018, **3**, 69–95.
- 4 K. Mizrahi Rodriguez, A. X. Wu, Q. Qian, G. Han, S. Lin, F. M. Benedetti, H. Lee, W. S. Chi, C. M. Doherty and Z. P. Smith, *Macromolecules*, 2020, **53**, 6220–6234.
- 5 Z. P. Smith, D. F. Sanders, C. P. Ribeiro, R. Guo, B. D. Freeman, D. R. Paul, J. E. McGrath and S. Swinnea, *J. Membr. Sci.*, 2012, **415–416**, 558–567.
- 6 O. Vopička, M. G. De Angelis, N. Du, N. Li, M. D. Guiver and G. C. Sarti, *J. Membr. Sci.*, 2014, **459**, 264–276.
- 7 A. Fuoco, B. Satilmis, T. Uyar, M. Monteleone, E. Esposito, C. Muzzi, E. Tocci, M. Longo, M. P. De Santo, M. Lanč, K. Friess, O. Vopička, P. Izák and J. C. Jansen, *J. Membr. Sci.*, 2020, **594**, 117460.
- 8 X. Ma, R. Swaidan, Y. Belmabkhout, Y. Zhu, E. Litwiller, M. Jouiad, I. Pinna and Y. Han, *Macromolecules*, 2012, **45**, 3841–3849.
- 9 G. Genduso, Y. Wang, B. S. Ghanem and I. Pinna, *J. Membr. Sci.*, 2019, **584**, 100–109.
- 10 Z. Liu, Y. Liu, W. Qiu and W. J. Koros, *Angew. Chemie.*, 2020, **59**, 14877–14883.
- 11 R. Swaidan, B. Ghanem, M. Al-Saeedi, E. Litwiller and I. Pinna, *Macromolecules*, 2014, **47**, 7453–7462.
- 12 R. Swaidan, B. Ghanem, E. Litwiller and I. Pinna, *J. Membr. Sci.*, 2015, **475**, 571–581.
- 13 M. A. Abdulhamid, G. Genduso, Y. Wang, X. Ma and I. Pinna, *Ind. Eng. Chem. Res.*, 2020, **59**, 5247–5256.
- 14 R. Swaidan, B. S. Ghanem, E. Litwiller and I. Pinna, *J. Membr. Sci.*, 2014, **457**, 95–102.
- 15 A. E. Gemedà, Solubility, Diffusivity and Permeability of Gases in Glassy Polymers. Ph.D. Thesis, Università di Bologna, Bologna, Italy, 2015.
- 16 J. Wu, J. Liu and T. S. Chung, *Adv. Sustain. Syst.*, 2018, **2**, 1800044.
- 17 X. Wu, W. Liu, H. Wu, X. Zong, L. Yang, Y. Wu, Y. Ren, C. Shi, S. Wang and Z. Jiang, *J. Membr. Sci.*, 2018, **548**, 309–318.
- 18 X. Wu, Y. Ren, G. Sui, G. Wang, G. Xu, L. Yang, Y. Wu, G. He, N. Nasir, H. Wu and Z. Jiang, *AICHE J.*, 2020, **66**, e16800.
- 19 Q. Song, S. Cao, R. H. Pritchard, B. Ghalei, S. A. Al-Muhtaseb, E. M. Terentjev, A. K. Cheetham and E. Sivaniah, *Nat. Commun.*, 2014, **5**, 4813.
- 20 F. Y. Li, Y. Xiao, Y. K. Ong and T. S. Chung, *Adv. Energy Mater.*, 2012, **2**, 1456–1466.

- 21 Y. Wang, Y. Ren, H. Wu, X. Wu, H. Yang, L. Yang, X. Wang, Y. Wu, Y. Liu and Z. Jiang, *J. Membr. Sci.*, 2020, **602**, 117970.
- 22 C. Zhang, L. Fu, Z. Tian, B. Cao and P. Li, *J. Membr. Sci.*, 2018, **556**, 277–284.
- 23 Y. He, F. M. Benedetti, S. Lin, C. Liu, Y. Zhao, H. Z. Ye, T. Van Voorhis, M. G. De Angelis, T. M. Swager and Z. P. Smith, *Adv. Mater.*, 2019, **31**, 1807871.
- 24 R. Williams, L. A. Burt, E. Esposito, J. C. Jansen, E. Tocci, C. Rizzuto, M. Lanč, M. Carta and N. B. McKeown, *J. Mater. Chem. A*, 2018, **6**, 5661–5667.
- 25 P. Stanovsky, M. Karaszova, Z. Petrusova, M. Monteleone, J. C. Jansen, B. Comesaña-Gándara, N. B. McKeown and P. Izak, *J. Membr. Sci.*, 2021, **618**, 118694.
- 26 E. Esposito, I. Mazzei, M. Monteleone, A. Fuoco, M. Carta, N. B. McKeown, R. Malpass-Evans and J. C. Jansen, *Polymer*, 2019, **11**, 46.
- 27 C. G. Bezzu, M. Carta, M. C. Ferrari, J. C. Jansen, M. Monteleone, E. Esposito, A. Fuoco, K. Hart, T. P. Liyana-Arachchi, C. M. Colina and N. B. McKeown, *J. Mater. Chem. A*, 2018, **6**, 10507–10514.
- 28 N. Alaslai, B. Ghanem, F. Alghunaimi, E. Litwiller and I. Pinna, *J. Membr. Sci.*, 2016, **505**, 100–107.
- 29 F. Alghunaimi, B. Ghanem, N. Alaslai, M. Mukaddam and I. Pinna, *J. Membr. Sci.*, 2016, **520**, 240–246.
- 30 Q. Song, S. Cao, R. H. Pritchard, H. Qiblawey, E. M. Terentjev, A. K. Cheetham and E. Sivaniah, *J. Mater. Chem. A*, 2015, **4**, 270–279.
- 31 N. Alaslai, B. Ghanem, F. Alghunaimi and I. Pinna, *Polymer*, 2016, **91**, 128–135.
- 32 R. Swaidan, X. Ma, E. Litwiller and I. Pinna, *J. Membr. Sci.*, 2013, **447**, 387–394.
- 33 N. Tien-Binh, D. Rodrigue and S. Kaliaguine, *J. Membr. Sci.*, 2018, **548**, 429–438.
- 34 Y. J. Cho and H. B. Park, *Macromol. Rapid Commun.*, 2011, **32**, 579–586.
- 35 N. Alaslai, X. Ma, B. Ghanem, Y. Wang, F. Alghunaimi and I. Pinna, *Macromol. Rapid Commun.*, 2017, **38**, 1700303.
- 36 F. Alghunaimi, B. Ghanem, N. Alaslai, R. Swaidan, E. Litwiller and I. Pinna, *J. Membr. Sci.*, 2015, **490**, 321–327.
- 37 W. F. Yong and T. S. Chung, *Polymer*, 2015, **59**, 290–297.
- 38 F. Y. Li, Y. Xiao, T. S. Chung and S. Kawi, *Macromolecules*, 2012, **45**, 1427–1437.
- 39 L. Hao, P. Li and T. S. Chung, *J. Membr. Sci.*, 2014, **453**, 614–623.
- 40 W. F. Yong, F. Y. Li, T. S. Chung and Y. W. Tong, *J. Membr. Sci.*, 2014, **462**, 119–130.
- 41 Z. Wang, Q. Shen, J. Liang, Y. Zhang and J. Jin, *Sep. Purif. Technol.*, 2013, **233**, 116008.
- 42 C. H. Lau, K. Konstas, C. M. Doherty, S. J. D. Smith, R. Hou, H. Wang, M. Carta, H.

Yoon, J. Park, B. D. Freeman, R. Malpass-Evans, E. Lasseguette, M. C. Ferrari, N. B. McKeown and M. R. Hill, *Nanoscale*, 2020, **12**, 17405–17410.

- 43 Y. Lou, P. Hao and G. Lipscomb, *J. Membr. Sci.*, 2014, **455**, 247–253.
- 44 K. Ghosal, R. T. Chern, B. D. Freeman, W. H. Daly and I. I. Negulescu, *Macromolecules*, 1996, **29**, 4360–4369.
- 45 G. Genduso, B. Ghanem and I. Pinna, *Membranes*, 2019, **9**, 10.