

## Supporting Information for

### Synthesis and catalytic application of nanorod-like FER-type zeolite

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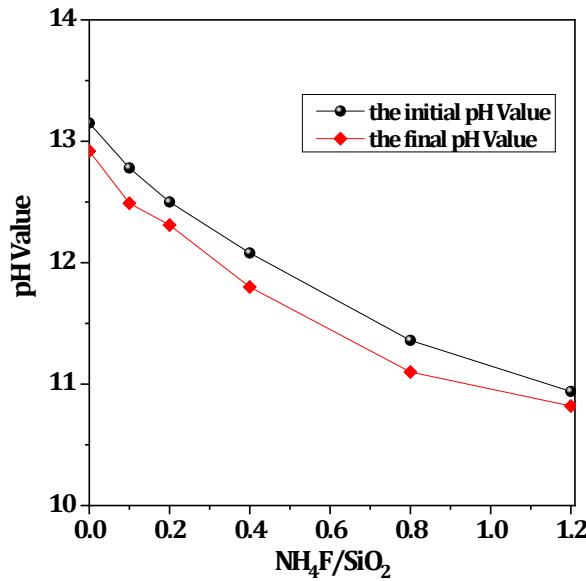
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\*Corresponding Authors

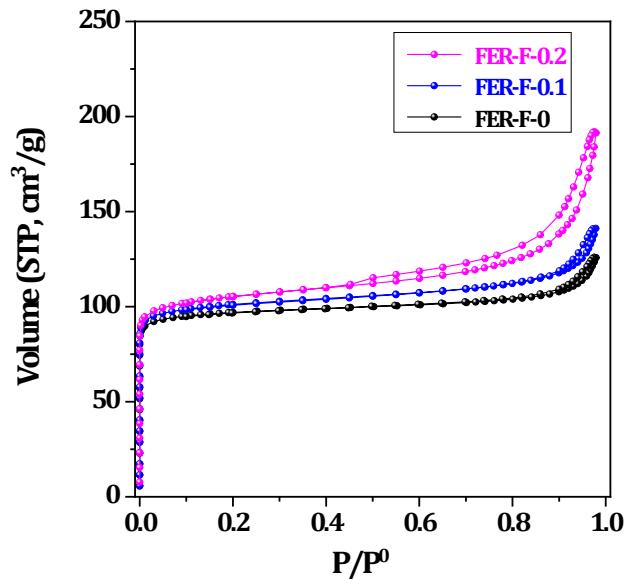
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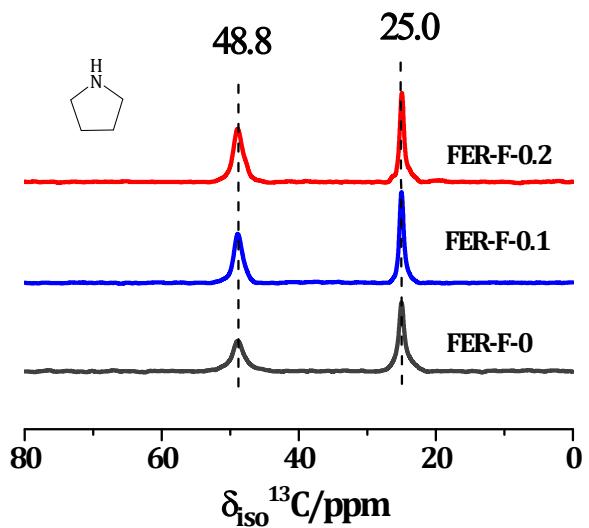
Figure S1-7; Table S1-3



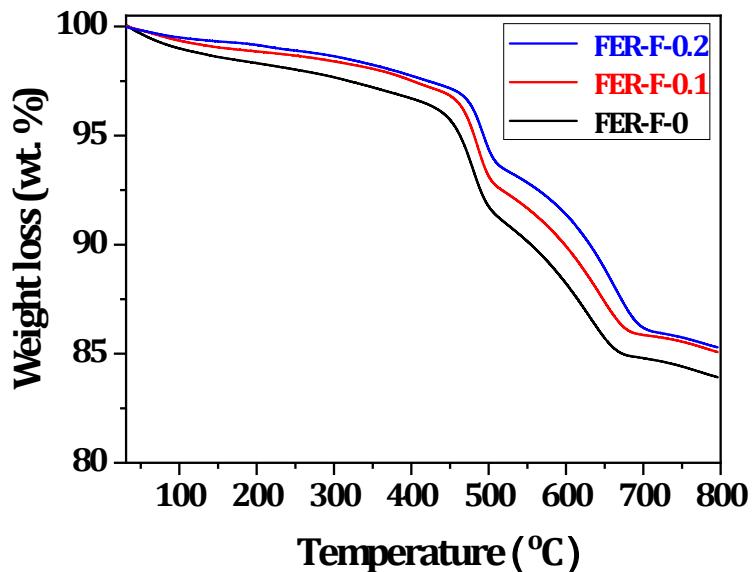
**Figure S1.** Variation of the pH value as a function of  $\text{NH}_4\text{F}$  concentration in the synthesis system.



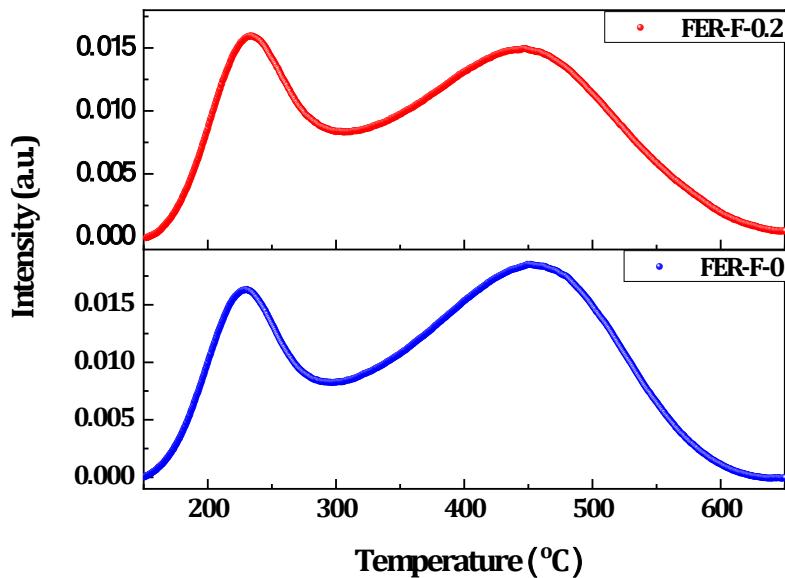
**Figure S2.** Nitrogen adsorption/desorption isotherms of as-synthesized FER samples.



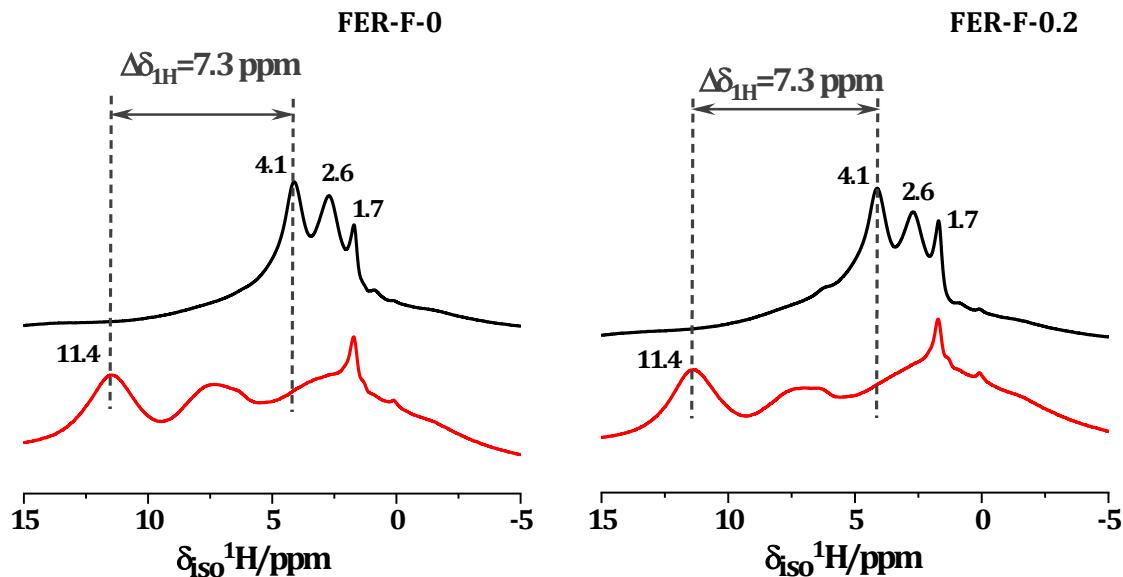
**Figure S3.**  $^{13}\text{C}$  CP MAS NMR spectra of as-synthesized three different FER zeolites.



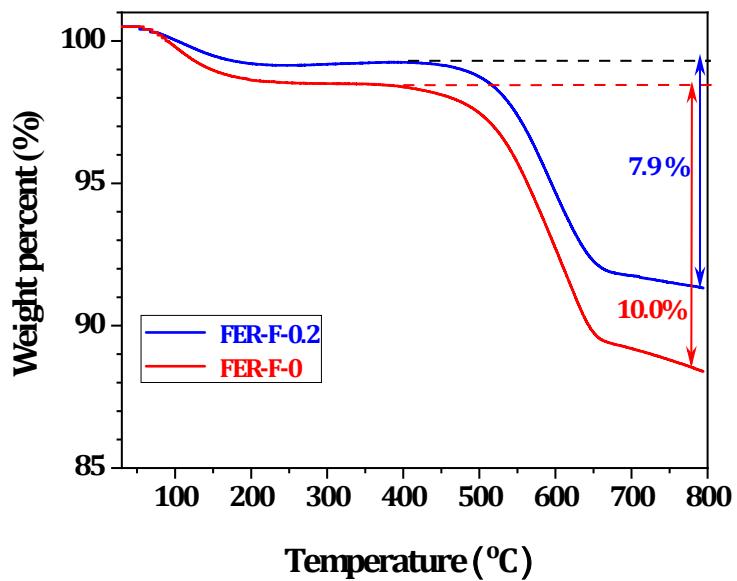
**Figure S4.** TG curves of as-synthesized FER zeolite samples: FER-F-0, FER-F-0.1 and FER-F-0.2.



**Figure S5.** NH<sub>3</sub>-TPD profiles of as-synthesized samples FER-F-0 and FER-F-0.2.



**Figure S6.** <sup>1</sup>H MAS NMR spectra of the dehydrated FER samples recorded before (black curves) and after (red curves) loading with D<sub>3</sub>-acetonitrile.



**Figure S7.** TGA curves of spent FER-F-0 and FER-F-0.2 catalysts used in the 1-butene skeletal isomerization.

**Table S1.** Molar composition of the initial mixtures and crystallization conditions of synthesized FER-type zeolites.

Sample	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	SDA	Na <sub>2</sub> O	H <sub>2</sub> O	NH <sub>4</sub> F	T (°C)	Phase
FER-F-0	1	0.05	0.6	0.08	20	0	180	FER
FER-F-0.1	1	0.05	0.6	0.08	20	0.1	180	FER
FER-F-0.2	1	0.05	0.6	0.08	20	0.2	180	FER
FER-F-0.4	1	0.05	0.6	0.08	20	0.4	180	FER
FER-F-0.8	1	0.05	0.6	0.08	20	0.8	180	FER+ MTN
FER-F-1.2	1	0.05	0.6	0.08	20	1.2	180	FER+ MTN

**Table S2.** Textural properties of as-synthesized FER samples.

Samples	S <sub>BET</sub> <sup>a</sup> (m <sup>2</sup> /g)	S <sub>ext</sub> <sup>b</sup> (m <sup>2</sup> /g)	V <sub>micro</sub> <sup>b</sup> (cm <sup>3</sup> /g)	V <sub>total</sub> <sup>c</sup> (cm <sup>3</sup> /g)	Si/Al <sup>d</sup>
FER-F-0	383	56	0.13	0.19	8.7
FER-F-0.1	397	67	0.13	0.22	9.3
FER-F-0.2	409	86	0.13	0.30	9.5

<sup>a</sup> determined by the multi-point BET method.

<sup>b</sup> calculated by the t-plot method.

<sup>c</sup> the volume adsorbed at P/P<sup>0</sup> = 0.97.

<sup>d</sup> determined by the ICP.

**Table S3.** Coke amount, average coke accumulation rate (R<sub>coke</sub>) during the the 1-butene skeletal isomerization over FER zeolite catalysts under study, determined for TOS = 1440 min.

samples	Coke(wt%)	R <sub>coke</sub> (mg/h)
FER-F-0	10.0	0.463
FER-F-0.2	7.9	0.357