Electronic supplementary information (ESI)

Layer dependence of out-of-plane electrical conductivity and Seebeck coefficient in continuous mono- to multilayer MoS₂ films

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A. Growth of the monolayer and multilayer MoS₂

The continuous and large-area ML MoS₂ films were synthesized on 300 nm-thick SiO₂/Si substrates using metal-organic chemical vapor deposition (MOCVD). A hot-wall furnace was used to synthesize the films, in which molybdenum hexacarbonyl (Mo(CO)₆, MHC), diethyl sulfide ((C₂H₅)₂S, DES), argon (Ar), and hydrogen (H₂) were used as chemical precursors to Mo, S, carrier gas, and reduction gas, respectively. The detailed process can be found in the previous report.¹ The continuous and large-area ML MoS₂ was successfully obtained at the 600°C (for 8 hr) temperature with Ar (30 cm³/min), H₂ (5 cm³/min), and DES (1 cm³/min) flows of 60 Torr. Specifically, the MHC flows freely. The 300 nm-thick SiO₂/Si substrates were horizontally situated at the center of the reaction chamber. We first deposited MoO₃ thin films with 99.9% purity (iTASCO) and approximately 5 nm thickness approximately using an e-beam evaporator (Korea Vacuum Tech Co. Ltd.) to grow large-area MTL MoS₂ films. We used a simple sulfurization procedure with H₂ and N₂ gases in a low-pressure chemical vapor deposition (CVD) system.² The temperature in the CVD furnace was first raised to 900 °C and maintained for 40 min under H₂ (100 cm³/min) and N₂ (500 cm³/min) flow with 0.8 Torr to sulfurize the MoO₃ thin film. Thereafter, sublimation of the sulfur powder (99.998%, Sigma-Aldrich) was initiated in the upstream heating zone at 200 °C. This process was repeated for 30 min to replace O₂ with sulfur in the precursor thin films successfully. The MOCVD-grown ML MoS₂ film was characterized by optical microscopy (OM), Raman, and photoluminescence (PL) spectroscopy. Furthermore, the cross-sectional TEM analysis was performed to verify the morphological features of the ML, BL, and MTL MoS₂. Moreover, the thickness of the MTL MoS₂ layers was investigated by atomic force microscopy (AFM).

B. Sample preparation for evaluating out-of-plane Seebeck coefficient of MoS₂ films

The detailed process can be found previous report.¹ The continuous and large-area ML The largearea ML MoS₂ film was transferred on the Cu thin film/300-nm-thick SiO₂/Si substrate by conventional poly (methyl methacrylate) (PMMA)-assisted wet chemical etching method to measure the out-of-plane Seebeck coefficient.^{3, 4} To be more specific, in the case of the ML MoS₂, a thin layer of PMMA (950 K A4) was spin-coated on the as-grown ML MoS₂ film grown on the SiO₂/Si substrate (Fig. 1c); the sample was placed on a hot plate at 120 °C for 2 min. The sample was floated in 5% hydrofluoric (HF) acid for 1 min, in which the Si substrate sank owing to SiO₂ etching, leaving the PMMA/MoS₂ layers to float on top of the HF solution. The PMMA-coated ML MoS₂ film was washed with de-ionized water, transferred to target Cu/SiO₂/Si substrates for measuring the out-ofplane Seebeck coefficients, and dried in ambient conditions. Subsequently, we conducted the O2 plasma treatment using a plasma surface treatment system (CUTE, Femto Science Inc.) for the Cu/SiO₂/Si substrates during 1 min, which leads to better adhesion between Cu and MoS₂ layers. The transferred ML MoS₂ film on the target substrate (Cu/SiO₂/Si substrate) was annealed at 250 °C for 2 h for better adhesion and the polymer residue removal after removing the PMMA layer using warm acetone. Additionally, Cu metal film was deposited on the transferred ML MoS₂/Cu/SiO₂/Si substrates by radio-frequency (RF) sputtering at room temperature to measure the out-of-plane Seebeck coefficient of the sample using Cu-sandwiched structure; it was then covered with a 200nm-thick Cu layer using the RF sputtering in an ultrahigh vacuum (UHV). The same method was used for the BL and MTL MoS₂ films. Particularly, the BL MoS₂ was prepared from two MOCVDgrown ML MoS₂ films using the above-mentioned transfer method. Furthermore, cross-section TEM specimens were prepared using the Cu/MoS₂/Cu/SiO₂/Si samples (ML, BL, and MTL MoS₂) with a focus ion-beam (FIB) milling technique (Fig. 1f, Fig. 3a, and Fig. 3b). The thickness of Cu thin films was ~ 200 nm. The Cu-sandwiched structures have the advantage of preserving target MoS₂ films during the FIB milling process and clearly observing their morphological features.

C. Measurement setup for evaluating TE properties

We measured the out-of-plane Seebeck coefficients (S_{\perp}) of the large-area ML, BL, and MTL MoS₂ films using the Cu-sandwiched structure. Further details of the Seebeck coefficient measurement can be found in our previous works.⁵⁻⁷ The Cu/MoS₂/Cu/SiO₂/Si (hereafter called Cu/MoS₂/Cu) samples were first held between AlN holders, and then a temperature difference was applied across the samples using the micro-Peltier device. The temperature difference was measured using two T-type thermocouples. To minimize the substrate effect (SiO₂/Si substrate) when measuring the S_{\perp} , we measured the out-of-plane temperature difference (ΔT_{\perp}) of the Cu/MoS₂/Cu structure under the steady-state condition. The out-of-plane thermoelectric Seebeck voltages were measured by two shielded tungsten needles between the upper and lower Cu thin films. Simultaneously, the electrical conductivity of samples was measured by the two-probe method.

D. Theoretical calculations

The calculations were performed using the ab-initio total-energy and molecular-dynamics program Vienna ab-initio simulation program (VASP) developed at the Institute für Materialphysik of the Universität Wien using the projector-augmented-wave (PAW) approach.⁸ Generalized gradient approximation (GGA)⁹ was employed for the exchange-correlation energy functional. Moreover, the cutoff energy of 400 eV was used for a plane-wave basis set. The maximum remaining force on each atom was less than ~0.015 eV/Å for structural relaxation. The in-plane lattice parameters of the supercells along the MoS₂/Mg interface planes were fixed to the experimentally determined Mg lattice parameter¹⁰ of *a* = 3.2028 Å; the supercells were relaxed along the out-of-plane direction, perpendicular to the interface. The Monkhorst-Pack scheme was used to sample the Brillouin zone. The k-point mesh used for MoS₂/Mg was $22 \times 22 \times 4$, $22 \times 22 \times 3$, and $22 \times 22 \times 2$ for the structural relaxation of one, two, and five MoS₂ layers, respectively. The out-of-plane Seebeck tensor calculations of one, two, and five MoS₂ layers was $48 \times 48 \times 9$, $48 \times 48 \times 6$, and $48 \times 48 \times 6$, respectively.

Material	Density (kg/m ³)	Heat capacity at constant pressure (Cp, J/(kg.K))	Thermal conductivity (W/m·K)
Cu	8960	384.80	392.8
MoS_2	5060	397.83	34.5 ()/0.3 (⊥)

E. Material parameters for COMSOL calculation

*Interface thermal conductance (ITC) between Cu and MoS_2: ${\sim}2{\times}10^{\text{-9}}\ \text{m}^2\text{K/W}$



F. COMSOL calculation results for Cu/ML MoS₂/Cu

Figure S1. Calculated temperature distribution along the out-of-plane direction (z-axis) in Cu/ML MoS₂/Cu structure.

G. COMSOL calculation results for Cu/BL MoS₂/Cu



Figure S2. Calculated temperature distribution along the out-of-plane direction (z-axis) in Cu/BL MoS₂/Cu structure.



Figure S3. Calculated temperature distribution along the out-of-plane direction (z-axis) in Cu/MTL MoS₂/Cu structure.

I. Band structures of ML and BL MoS_2 system



Figure S4. Calculated band structures for (a) ML MoS_2 and (b) BL MoS_2

J. Thickness dependence of the out-of-plane Seebeck voltage



Figure S5. The measured out-of-plane Seebeck voltage difference of ML (~0.7 nm), BL (~1.4 nm), MTL (~4.9 nm), and ~7-nm-thick MoS_2 , respectively. The results of ~7-nm-thick MoS_2 is taken from our previous work.⁵

K. Depth-profiling analyses of the Cu/MTL MoS₂/Cu structure



Figure S6. (a)-(d) XPS spectra of the Cu $2p_{3/2}$, O 1s, Mo 3d, and S 2p core levels dependent on etching times, respectively. From these observations, the O 1s peak was found at the surface and interface Cu whereas that peak could not be observed on the interior Cu. It indicates the formation of CuO on the interface region (Cu/MoS₂) caused by wet-transfer process including oxygen (O₂) plasma treatment.

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