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## Supporting Information

## Interfacial engineering of nickel/iron/ruthenium phosphides for efficient overall water splitting powered by solar energy

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Fig. S1 Single cycle CV curves of Hg/HgO electrode calibration in 1.0 M KOH.



Fig. S2 PXRD patterns of NiFeRu-LDH.



Fig. S3 (a, b) SEM images of NiFeRu-LDH/NF.



Fig. S4 PXRD patterns of  $Ni_2P$ -Fe<sub>2</sub>P-Ru<sub>2</sub>P/NF.



Fig. S5 Contact angles of NF (a) and  $Ni_2P\mbox{-}Fe_2P\mbox{-}Ru_2P$  /NF (b).



Fig. S6 (a, b) SEM images of NiFe-LDH/NF.



Fig. S7 PXRD patterns of NiFe-LDH.



Fig. S8 (a, b) SEM images of  $Ni_2P$ -Fe<sub>2</sub>P/NF.



Fig. S9 PXRD patterns of  $Ni_2P$ -Fe<sub>2</sub>P/NF.



Fig. S10 The survey XPS spectrum for  $Ni_2P$ -Fe<sub>2</sub>P-Ru<sub>2</sub>P/NF.



**Fig. S11** CV curves of  $Ni_2P$ -Fe<sub>2</sub>P/NF (a) and  $Ni_2P$ -Fe<sub>2</sub>P-Ru<sub>2</sub>P/NF (b) in the non-Faradaic potential region recorded at different scan rates.

Calculation of ECSA:

ECSA =  $\frac{C_{dl}}{C_s}$ ECSA<sub>Ni<sub>2</sub>P-Fe<sub>2</sub>P/NF</sub> =  $\frac{2.33 \text{ mF cm}^{-2}}{40 \,\mu \text{ F cm}^{-2}}$  = 58.3 cm<sup>-2</sup><sub>ECSA</sub>

 $ECSA_{Ni_{2}P-Fe_{2}P-Ru_{2}P/NF} = \frac{4.45 \text{ mF cm}^{-2}}{40 \,\mu \text{ F cm}^{-2}} = 111.2 \,\text{cm}^{-2}_{ECSA}$ 



Fig. S12 OER activity of different catalysts in 1 M KOH normalized by ECSA.



Fig. S13 Durability test of  $Ni_2P$ -Fe<sub>2</sub>P-Ru<sub>2</sub>P/NF at 100 mA cm<sup>-2</sup> for OER.



Fig. S14 (a, b) SEM images of  $Ni_2P$ -Fe<sub>2</sub>P-Ru<sub>2</sub>P/NF after OER stability.



Fig. S15 PXRD patterns of  $Ni_2P$ -Fe<sub>2</sub>P-Ru<sub>2</sub>P/NF before and after stability test for the OER.



**Fig. S16** High-resolution XPS spectra of (a) Ni 2p, (b) Fe 2p, (c) Ru 3p, and P 2p of Ni<sub>2</sub>P-Fe<sub>2</sub>P-Ru<sub>2</sub>P/NF initial and after OER stability testing for the OER.



Fig. S17 CV curves of  $Ni_2P$ -Fe<sub>2</sub>P/NF (a) and  $Ni_2P$ -Fe<sub>2</sub>P-Ru<sub>2</sub>P/NF (b) in the non-Faradaic potential region recorded at different scan rates.



Fig. S18 Durability test of  $Ni_2P$ -Fe<sub>2</sub>P-Ru<sub>2</sub>P/NF at 100 mA cm<sup>-2</sup> for OER.



Fig. S19 (a, b) SEM images of  $Ni_2P$ -Fe $_2P$ -Ru $_2P/NF$  after the HER stability.



Fig. S20 PXRD patterns of  $Ni_2P$ -Fe<sub>2</sub>P-Ru<sub>2</sub>P/NF before and after stability test for the HER.

Catalysts	Tafel slop (mV dec <sup>-</sup> <sup>1</sup> )	η10 (mV)	Reference
Ni <sub>2</sub> P-Fe <sub>2</sub> P-Ru <sub>2</sub> P/NF	30.5	195	This work
Mo-Ni <sub>3</sub> S <sub>2</sub> /Ni <sub>x</sub> P <sub>y</sub> /NF	60.6	/	Adv. Energy. Mater. 2020, 10, 1903891.
NiMoO <sub>x</sub> /NiMoS	34	186	Nat. Commun. 2020, 11, 5462.
MoS <sub>2</sub> /Co <sub>9</sub> S <sub>8</sub> /Ni <sub>3</sub> S <sub>2</sub> /Ni	58	166	J. Am. Chem. Soc. 2019, 141, 10417
CoMoS <sub>4</sub> /Ni <sub>3</sub> S <sub>2</sub>	63	200	J. Power Sources 2019, 416, 95.
Porous Ni <sub>3</sub> S <sub>4</sub>	67	257	Adv. Funct. Mater. 2019, 29, 1900315.
Co <sub>2</sub> P NCs	60	280	Adv. Mater. 2018, 30, 1705796.
Ni <sub>2</sub> P-CoP	69	320	ACS Appl. Mater. Interfaces 2017, 9, 23222.
NiCoP/C	96	330	Angew. Chem. Int. Ed. 2017, 56, 3897.
MAF-X27-OH	88	292	J. Am. Chem. Soc. 2016, 138, 8336

**Table S1.** Comparing the electrocatalytic OER performance of  $Ni_2P$ -Fe $_2P$ -Ru $_2P/NF$ with many catalysts recently reported.

Table S2.	Comparing the	electrocatalytic	HER perform	nance of Ni <sub>2</sub> P-I	$Fe_2P-Ru_2P/NF$
with many	catalysts recent	ly reported.			

Catalysts	Tafel slop (mV dec <sup>-1</sup> )	η10 (mV)	Reference
Ni <sub>2</sub> P-Fe <sub>2</sub> P-Ru <sub>2</sub> P/NF	85.1	78.6	This work
Mo-Ni <sub>3</sub> S <sub>2</sub> /Ni <sub>x</sub> P <sub>y</sub> /NF	68.4	109	Adv. Energy. Mater. 2020, 10, 1903891.
$CoMoS_4/Ni_3S_2$	169	158	J. Power Sources 2019, 416, 95.
Ni <sub>2</sub> P-Fe <sub>2</sub> P/NF	86	128	Adv. Funct. Mater. 2020, 30, 2006484.
NiFeP/NCH	125	216	J. Am. Chem. Soc. 2019, 141, 7906.
MoS <sub>2</sub> /Co <sub>9</sub> S <sub>8</sub> /Ni <sub>3</sub> S <sub>2</sub> /Ni	85	113	J. Am. Chem. Soc. 2019, 141, 10417
$(Co_{1-x}Ni_x)(S_{1-y}Py)_2/G$	85	117	Adv. Energy Mater. 2018, 8, 1802319.
Ni <sub>12</sub> P <sub>5</sub> /Ni <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub> -HS	93.1	114	<i>Appl. Catal. B Environ</i> 2017, 204, 486.

Catalysts	Cell voltage (V)	Reference
Ni <sub>2</sub> P-Fe <sub>2</sub> P-Ru <sub>2</sub> P/NF	1.49	This work
Fe-CoP/Ni(OH) <sub>2</sub>	1.52	Adv. Funct. Mater. 2021, 31, 2101578.
Mo-Ni <sub>3</sub> S <sub>2</sub> /NixPy/NF	1.46	Adv. Energy. Mater. 2020, 10, 1903891.
NiCoP@NiMn LDH/NF	1.51	Appl. Mater. Interfaces 2020, 12, 4385.
Cr-doped FeNiP/NCN	1.5	Adv. Mater. 2019, 31, 1900178.
NiFeP/SG	1.54	Nano Energy 2019, 58, 870.
NiFeP/NCH	1.59	J. Am. Chem. Soc. 2019, 141, 7906.
MoS <sub>2</sub> /Co <sub>9</sub> S <sub>8</sub> /Ni <sub>3</sub> S <sub>2</sub> /Ni	1.54	J. Am. Chem. Soc. 2019, 141, 10417
Mo-doped CoP/CC	1.56	Nano Energy 2018, 48, 73.

**Table S3.** Comparing the electrocatalytic performance of  $Ni_2P$ -Fe<sub>2</sub>P-Ru<sub>2</sub>P/NF with recently reported catalysts for overall water splitting.