

**A room-temperature self-healing elastomer with ultra-high strength
and toughness fabricated via optimized hierarchical hydrogen-bonding
interactions**

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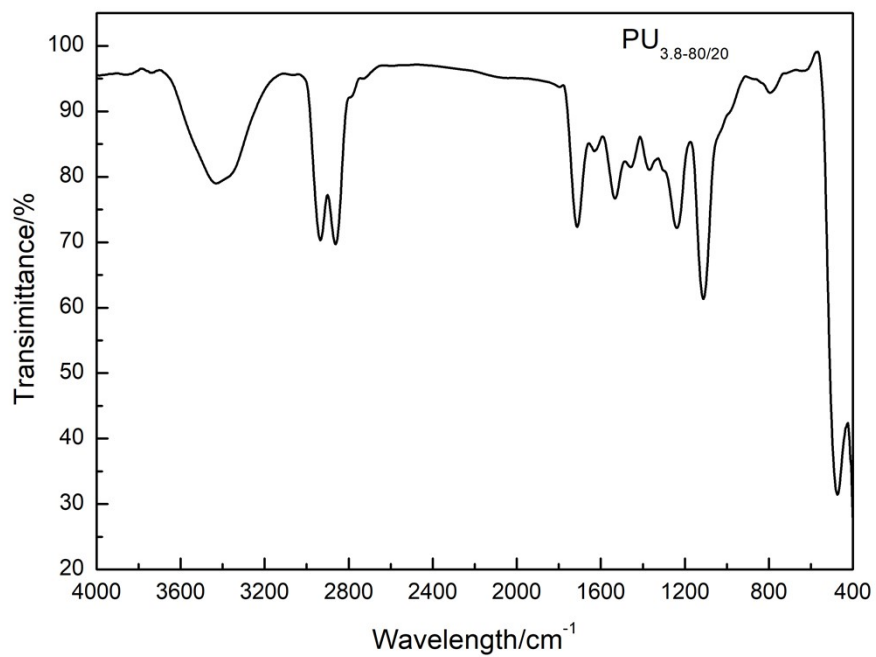


Figure S1. FTIR spectrum of PU_{3.8-80/20}.

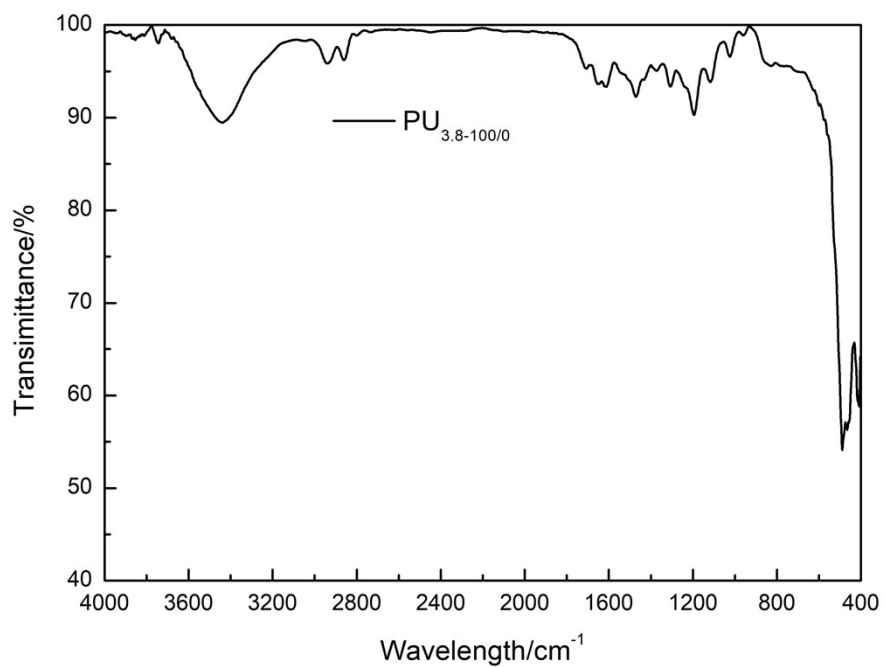


Figure S2. FTIR spectrum of PU_{3.8-100/0}.

Table S1. FT-IR characteristic peak assignments of PU_{3.8-80/20}.

Assignments	Wavenumber(cm ⁻¹)
H-bonded ν (N-H)	3428.9
va (CH ₂)	2935
vs (CH ₂)	2854
H-bonded ν (C=O) amide	1712
H-bonded ν (C-N) + δ (N-H) amide	1531
H-bonded ν (C-N) + δ (N-H) amide	1234
ν (C-O-C)	1114
δ (-C ₇ H ₅ -)	794

Table S2. FT-IR characteristic peak assignments of PU_{3.8-100/0}.

Assignments	Wavenumber(cm ⁻¹)
H-bonded ν (N-H)	3442
va (CH ₂)	2946
vs (CH ₂)	2869
H-bonded ν (C=O) amide	1716
H-bonded ν (C-N) + δ (N-H) amide	1481
H-bonded ν (C-N) + δ (N-H) amide	1201
ν (C-O-C)	1116

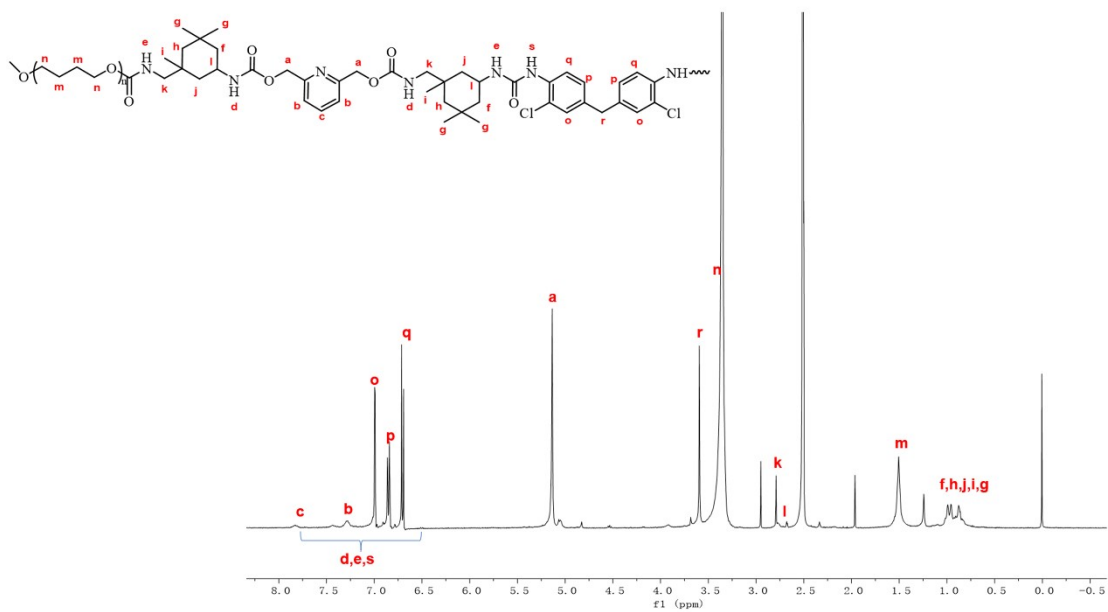


Figure S3. ¹H NMR spectrum of PU_{3.8-80/20}.

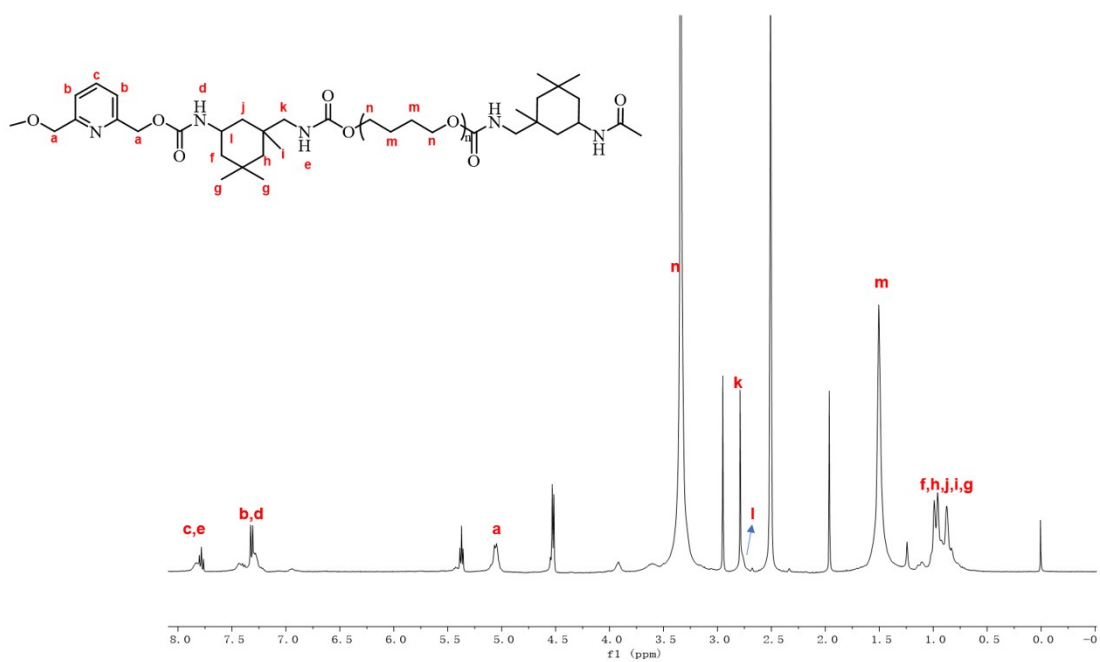
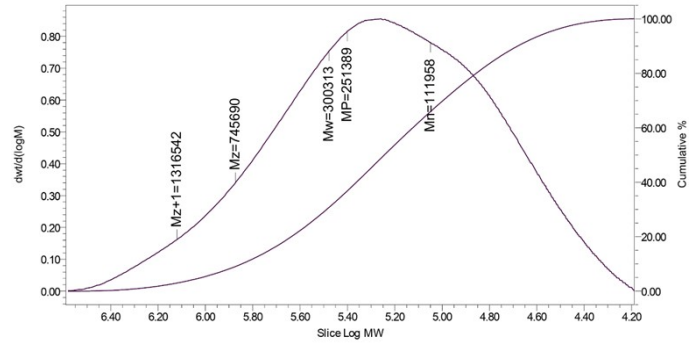
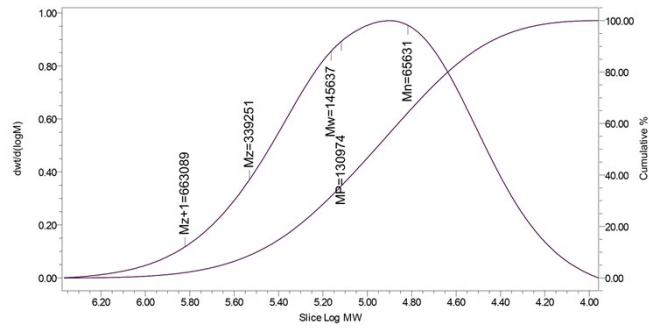


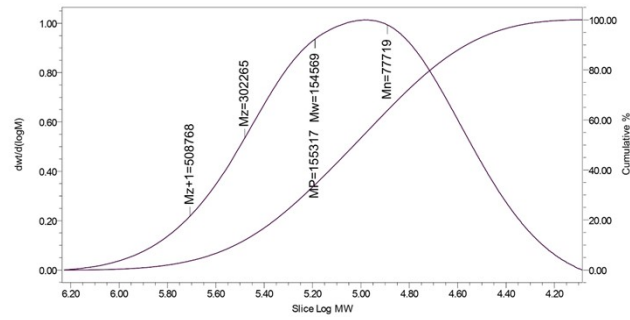
Figure S4. ¹H NMR spectrum of PU_{3.8-100/0}.



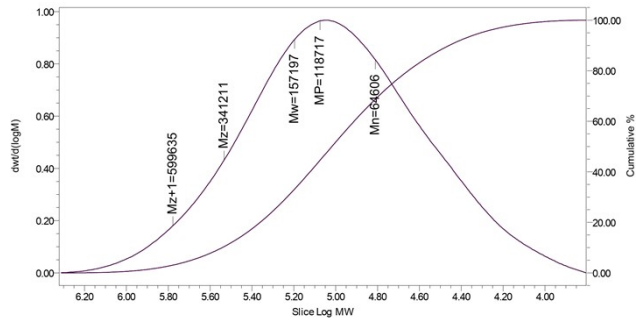
(a)



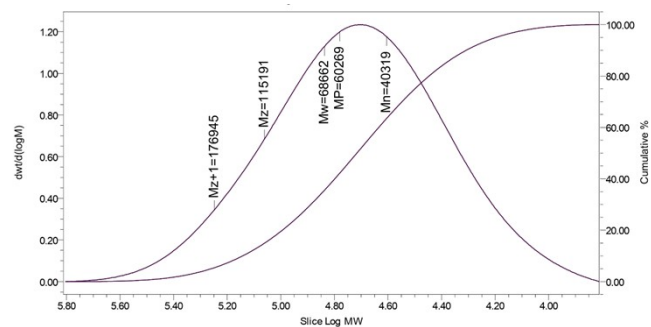
(b)



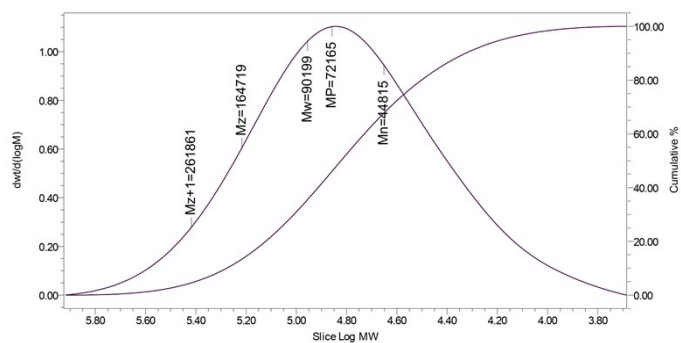
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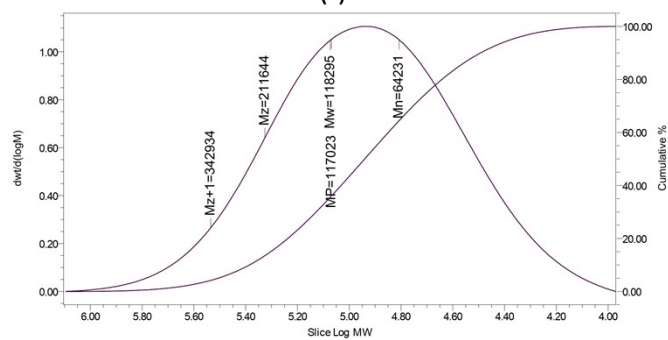
(d)



(e)



(f)



(g)

Figure S5. THF-GPC profiles of PU elastomers. (a) THF-GPC profiles of PU_{3.0-80/20}. (b) THF-GPC profiles of PU_{3.5-80/20}. (c) THF-GPC profiles of PU_{3.8-80/200}. (a) THF-GPC profiles of PU_{4.0-80/20}. (a) THF-GPC profiles of PU_{3.8-100/0}. (a) THF-GPC profiles of PU_{3.8-90/10}. (a) THF-GPC profiles of PU_{3.8-70/30}.

Table S3. The GPC results of PU elastomer.

Sample	M_n (g/mol)	M_w (g/mol)	PDI
PU _{3.0-80/20}	111958	300313	2.483
PU _{3.5-80/20}	65631	145637	2.329
PU _{3.8-80/20}	77719	154569	1.955
PU _{4.0-80/20}	64606	157197	2.171
PU _{3.8-100/0}	40319	68662	1.677
PU _{3.8-90/10}	44815	90199	1.826
PU _{3.8-70/30}	64231	118295	1.789

Table S4. Comparison of mechanical properties and self-healing efficiencies between PU_{3.8-80/20} and recently reported room-temperature self-healing materials.

Sample source	Tensile strength / MPa	Strain at break / %	Toughness / MJ.m ⁻³	Self-healing efficiency /%	Fracture energy / kJ.m ⁻²
This work	34.1	2014	127.3	83%	119.1
Ref. 5	1.1	229	6.5	92.9%	---
Ref. 7	16.08	771	75.6	94%	---
Ref. 8	5.72	1726	24.45	61.9%	---
Ref. 13	0.5	1000	---	95	---
Ref. 15	19.47	1144.3	105.2	75%	41.8
Ref. 16	4.83	2010	65.49	98%	42.65
Ref. 17	27.4	770	110	96	---
Ref. 18	1.3	2100	---	97	---
Ref. 25	6.31	710	---	91.6	---
Ref. 26	3.9	348.57	---	76	---
Ref. 28	11.91	723.57	30.07	50	---
Ref. 29	11	725.5	52.1	83	---
Ref. 30	4.2	954	---	90	---
Ref. 31	29	1800	121.8	80%	104.1
Ref. 38	30	700	---	82.27	---
Ref. 39	11.71	295	---	52.6	---
Ref. 40	19.5	675	---	83	---
Ref. 41	3.85	3000	84.02	70	---
Ref. 42	6.8	920	26.9	75	---
Ref. 43	16.1	771	---	94	---
Ref. 44	12.7	182	---	87.3	---

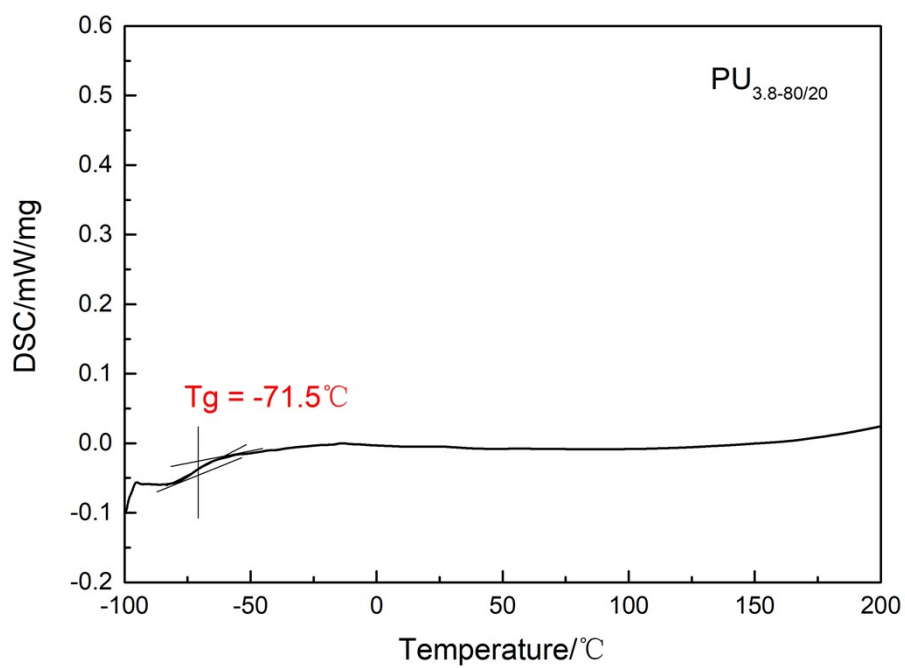


Figure S6. DSC curve of PU_{3.8-80/20}.

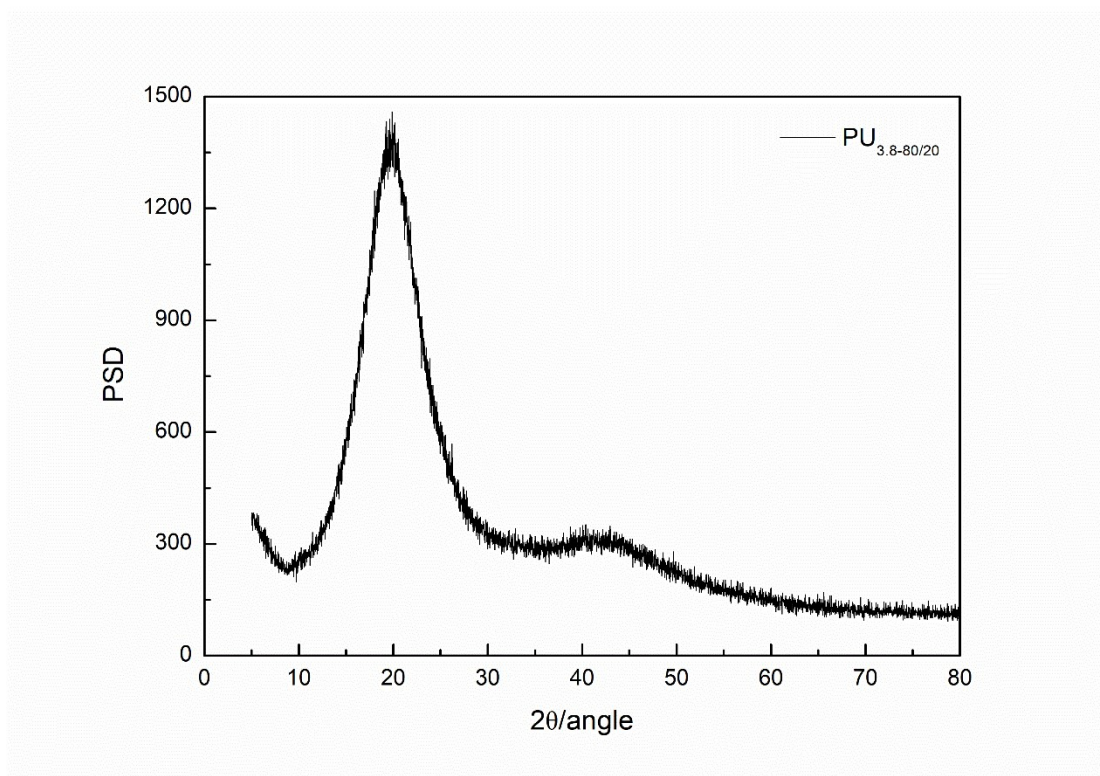


Figure S7. WXAD curve of PU_{3.8-80/20}, the XRD curve only displayed broad diffraction peaks, indicating the amorphous state at room temperature.

Table S5. The original mechanical performances of PU elastomers.

Sample	Tensile strength/MPa	Elongation/%	Toughness/MJ·m ⁻³
PU _{3.0-80/20}	8.4	1855	47.7
PU _{3.5-80/20}	21.6	2309	81.4
PU _{3.8-80/20}	34.1	2014	127.3
PU _{4.0-80/20}	22.2	2481	120.8
PU _{3.8-100/0}	0.6	2960	15.5
PU _{3.8-90/10}	2.0	3003	27.6
PU _{3.8-70/30}	27.5	1806	94.0

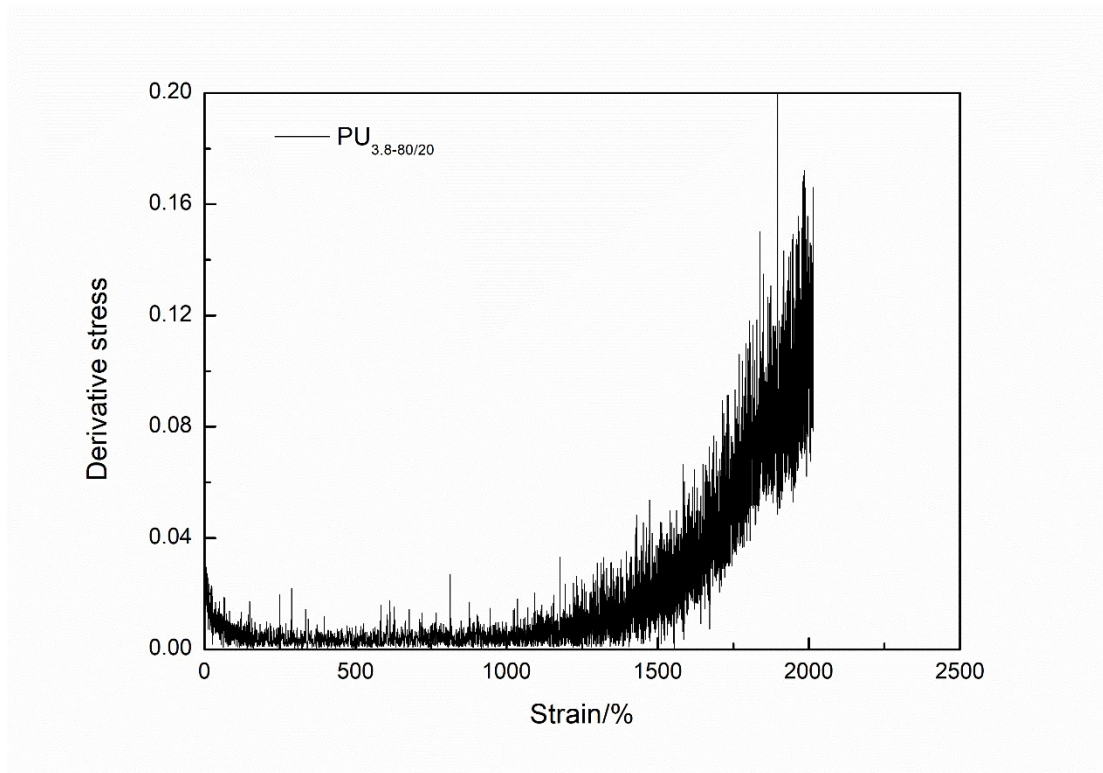
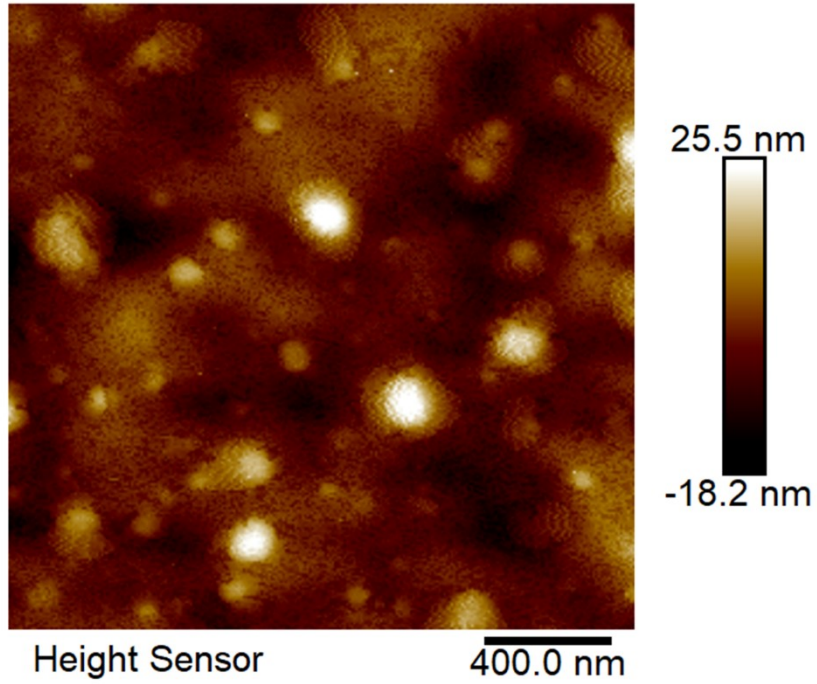
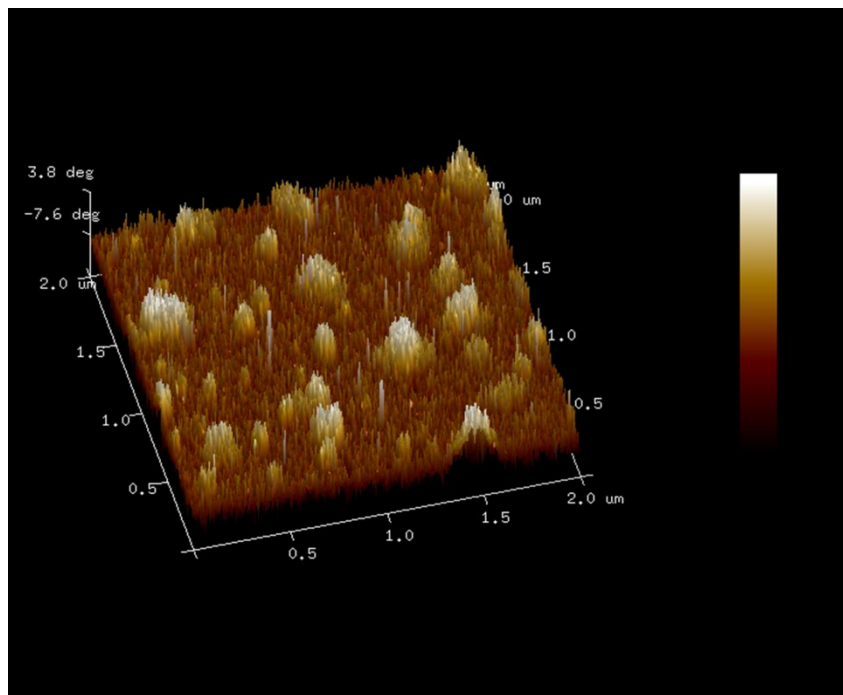


Figure S8. Derivative stress versus strain curve of PU_{3.8-80/20} for illustrating the changing rate of tensile strength.

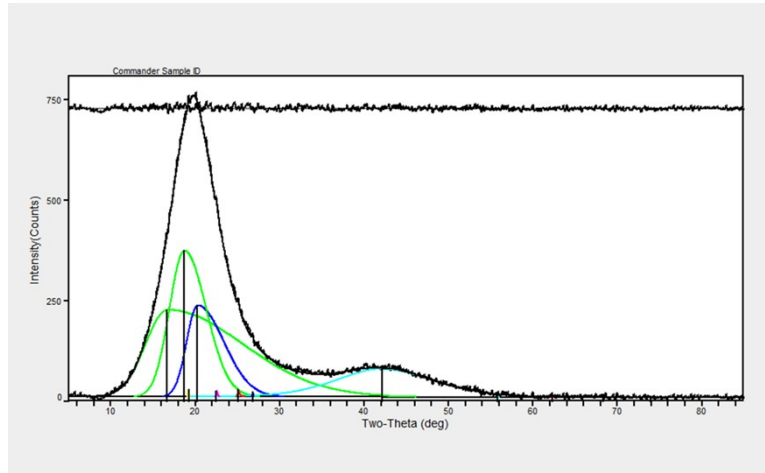


(a)

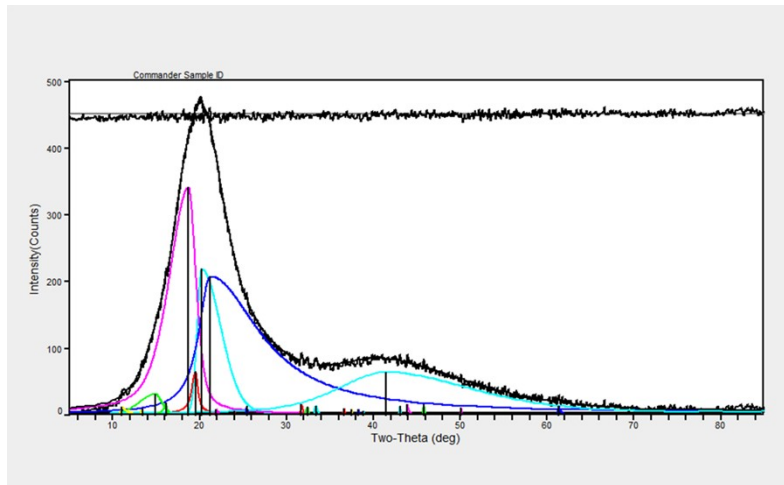


(b)

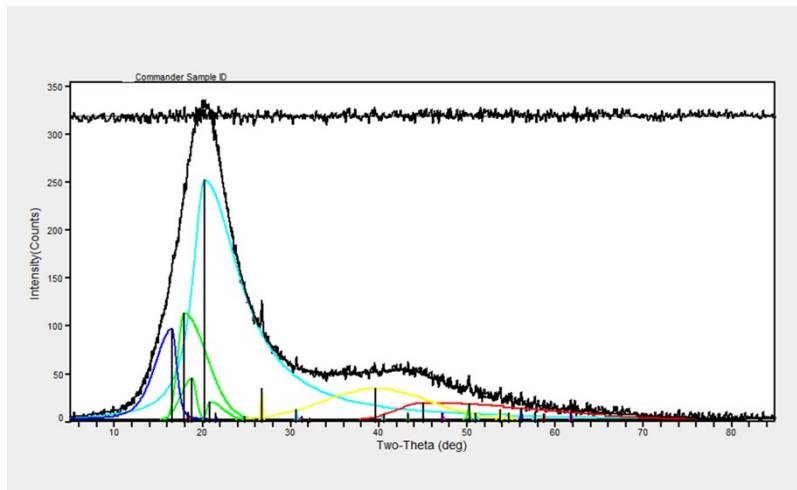
Figure S9. AFM images of $\text{PU}_{3.8-80/20}$. (a) AFM height sensor image of $\text{PU}_{3.8-80/20}$. (b) AFM 3D phase image of $\text{PU}_{3.8-80/20}$.



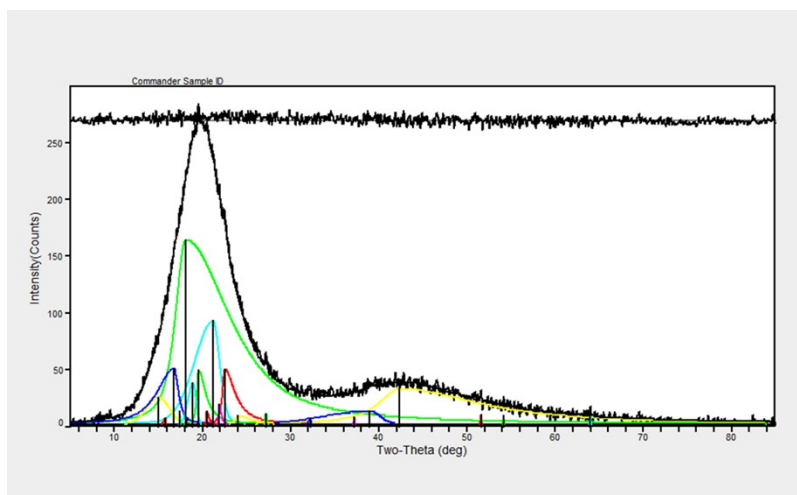
(a)



(b)



(c)



(d)

Figure S10. Fitting calculation (MDI Jade 6 software) of the crystallinity of elastomer PU_{3.8-80/20} in different strain state. (a) Fitting calculation (MDI Jade 6 software) of the crystallinity of elastomer PU_{3.8-80/20} in strain state 0%. (b) Fitting calculation (MDI Jade 6 software) of the crystallinity of elastomer PU_{3.8-80/20} in strain state 500%. (c) Fitting calculation (MDI Jade 6 software) of the crystallinity of elastomer PU_{3.8-80/20} in strain state 1000%. (d) Fitting calculation (MDI Jade 6 software) of the crystallinity of elastomer PU_{3.8-80/20} in strain state 1500%.

Table S6. The room-temperature self-healing result of PU elastomer at 25 °C for 48h.

Sample	Tensile strength (SE)/%	Elongation (SE)/%	Toughness (SE)/%
PU _{3.0-80/20}	94.7	91.3	94.6
PU _{3.5-80/20}	88.2	108.8	98.2
PU _{3.8-80/20}	83.3	100.1	88.4
PU _{4.0-80/20}	76.4	101.1	82.0
PU _{3.8-100/0}	98.7	101.2	99.2
PU _{3.8-90/10}	97.0	93.1	96.2
PU _{3.8-70/30}	82.8	99.3	90.7

Table S7. The mechanical performances of PU_{3.8-80/20} with self-healing time from 12h to 48h.

PU _{3.8-80/20}	Tensile strength (SE)/%	Elongation (SE)/%	Toughness (SE)/%
12h	43.6	78.8	45.9
24h	68.1	90.4	66.7
36h	77.1	94.3	80.1
48h	83.3	100.1	88.4

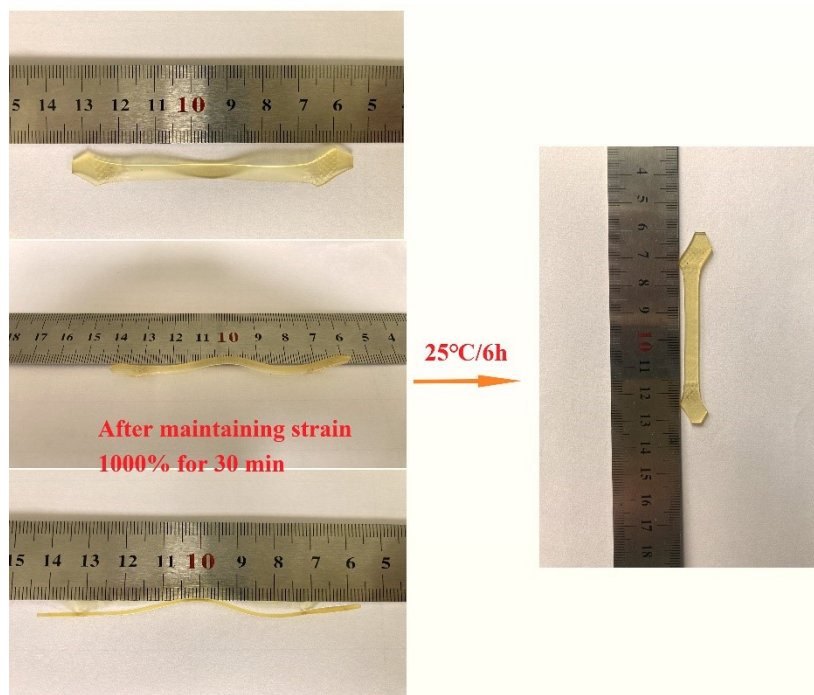


Figure S11. Optical images for after stretching and recovery of PU_{3.8-80/20} film.

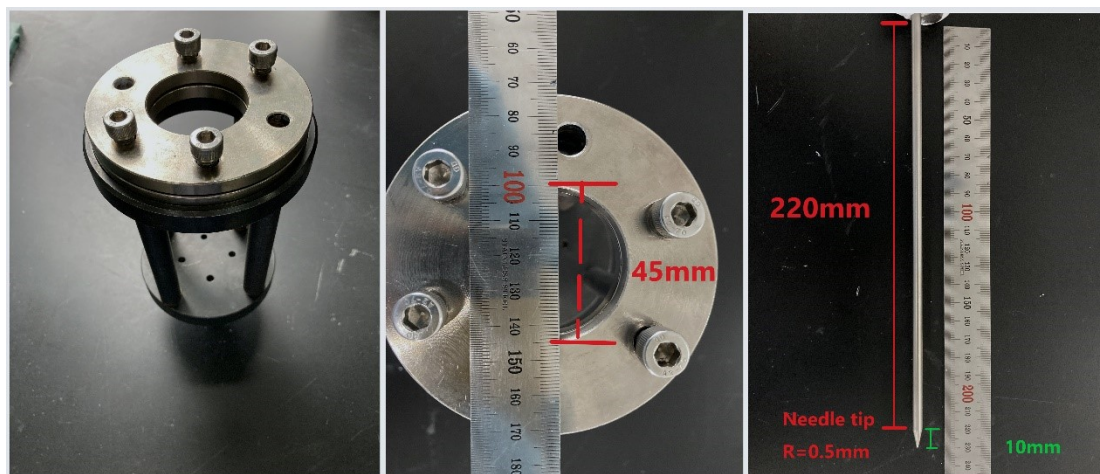
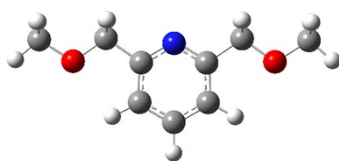
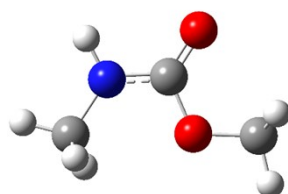


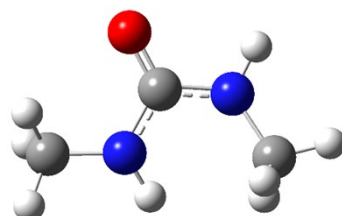
Figure S12. The dimension of steel needle and fixture in puncture test.



(a)

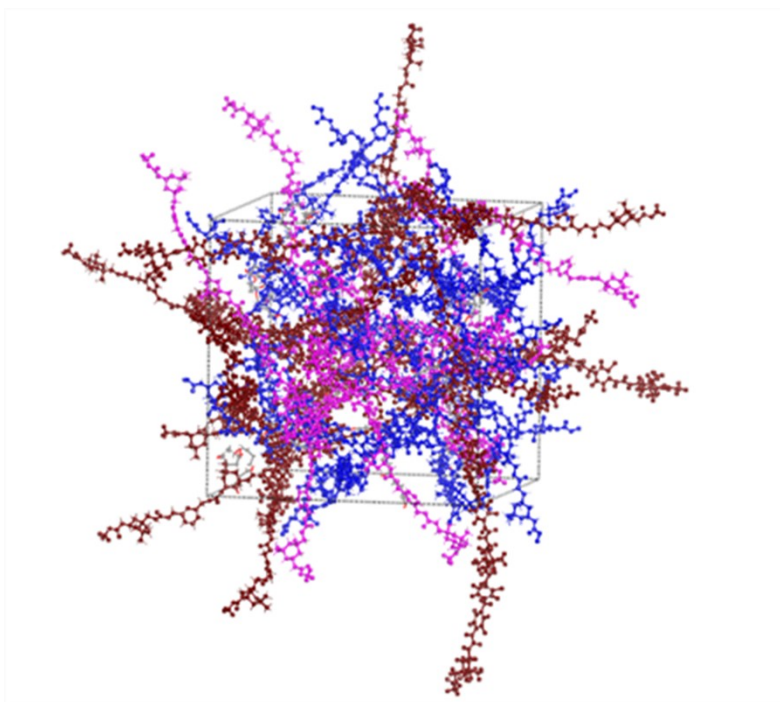


(b)

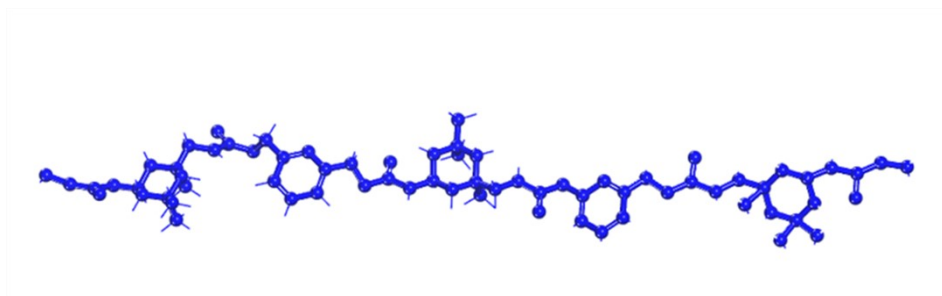
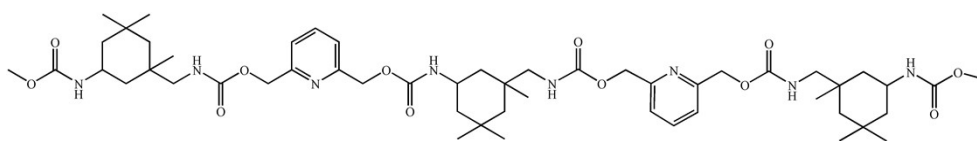


(c)

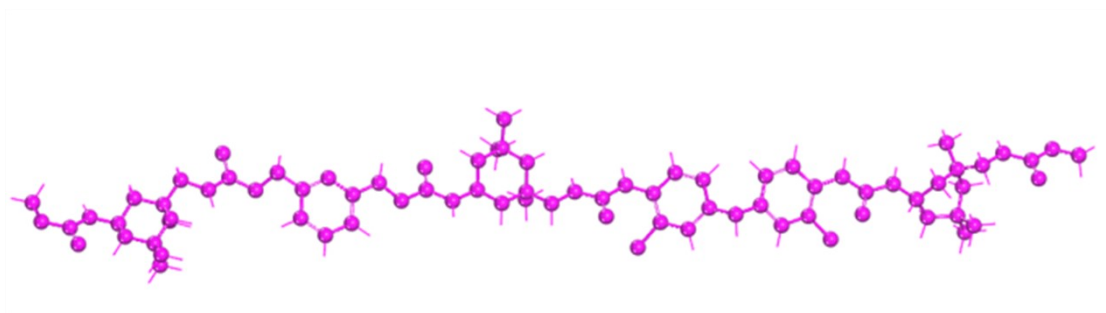
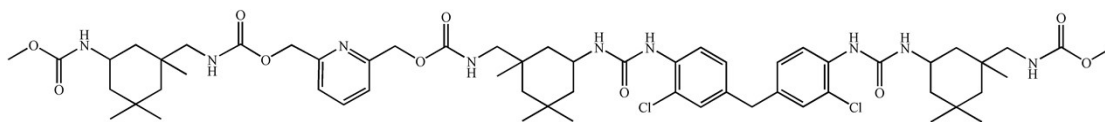
Figure S13. Optimized molecular structure for simulation. (a) Monomer A. (b) Monomer B. (c) Monomer C.



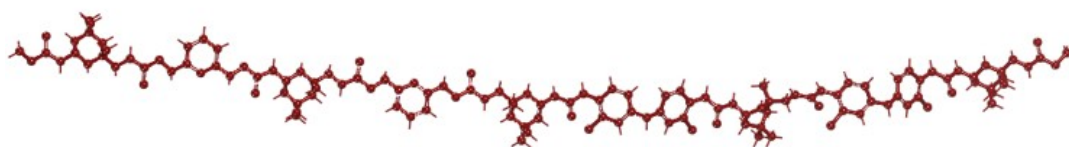
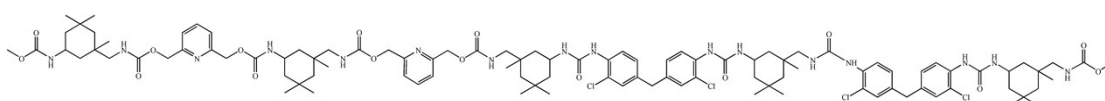
(a)



(b)



(c)



(d)

Figure S14. Schematic representation of a simulated molecular cell of PU_{3.8-80/20} in the equilibrium state. (a) hard domains are emphasized on the ball-and-stick models. (b) Quantum chemical simulation of hard domains. Optimized hard segments a for simulations. (c) Quantum chemical simulation of hard domains. Optimized hard segments b for simulations. (d) Quantum chemical simulation of hard domains. Optimized hard segments c for simulations.

Quantum chemistry calculation

All calculations in this work were performed using Gaussian 09 program package¹. Full geometry optimizations were performed to locate all the stationary points, using B3LYP-D3²/6-311+G (d, p)³⁻⁴. Single point energies were also obtained for all optimized structures using dispersion-corrected density functional (DFT-D3(BJ)) with the 6-311+G (d, p) basis set. Quantum theory of atoms-in-molecules (QTAIM) descriptors and other H-Bondings were calculated based on high-quality density functional theory wave functions using the Multiwfn software⁵. The monomer structures A, B, C were illustrated in Figure S13.

Molecular simulation method

Molecular dynamic simulations were performed to study the polymer behavior. Our simulation calculation was conducted on four structures with an integration time-step of 1 fs. Periodic boundary conditions were applied in the *x*- and *y*-dimensions. The box size of the samples was 50 × 50 × 50 nm³ (Figure S14 a). First, the conjugate gradient algorithm and energy minimization were performed to obtain a stable structure. Condensed-phased Optimized Molecular Potential for Atomistic Simulation Studies force field was also used to optimize these structures. Each sample was then equilibrated under the NPT ensemble at a constant temperature of 300 K to achieve an equilibrium state with zero pressure for 30 ns. The equilibration molecular systems of the pure separation membrane could be obtained after geometrically optimizing. The system is composed of 30 a, 10 b, 10 c (Figures S14 b, c, d), and 10 soft segments. Furthermore, a potential cutoff radius of 2.25 nm is applied in the calculation of the non-bonded interaction. And the PPPM has been used to describe the electrostatic. The Andersen feedback thermostat and Berendsen barostat algorithm are applied in the system with temperature and pressure conversion. Finally, the properties of our structures are obtained in the last 3000 ps. The binding energy can be used to measure the intensity of the interaction between a, b and c.

Reference

1. M.J. Frisch, G.W. Trucks, H.B. Schlegel, G.E. Scuseria, M.A. Robb, J.R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G.A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H.P. Hratchian, A.F. Izmaylov, J. Bloino, G. Zheng, J.L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J.A. Montgomery, J.J.E. Peralta, F. Ogliaro, M. Bearpark, J.J. Heyd, E. Brothers, K.N. Kudin, V.N. Taroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J.C. Burant, S.S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J.M. Millam, M. Klene, J.E. Knox, J.B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R.E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R.L. Martin, K. Morokuma, V.G. Zakrzewski, G.A. Voth, P. Salvador, J.J. Dannenberg, S. Dapprich, A.D. Daniels, O. Farkas, J.B. Foresman, J.V. Ortiz, J. Cioslowski, D.J. Fox, Gaussian 09 (Revision D.01), I. Gaussian, Wallingford, CT, **2013**.
2. Grimme, S.; Antony, J.; Ehrlich, S.; Krieg, H., A consistent and accurate ab initio parametrization of density functional dispersion correction (DFT-D) for the 94 elements H-Pu. *J. Chem. Phys.* **2010**, *132* (15), 154104.
3. Xu S., He T., Li J., Huang Z., & Hu C., Enantioselective synthesis of D-lactic acid via chemocatalysis using MgO: Experimental and molecular-based rationalization of the triose's reactivity and preliminary insights with raw biomass. *Appl. Catal. B: Environ.* **2021**, 292:120145.
4. McLean, A. D.; Chandler, G. S. Contracted Gaussian basis sets for molecular calculations. I. Second row atoms, $Z = 11-18$. *J. Chem. Phys.* **1980**, *72*, 5639-5648.
- 5 Lu, T.; Chen, F., Multiwfn: a multifunctional wavefunction analyzer. *J. Comput. Chem.* **2012**, *33* (5), 580-92.