

Electronic Supplementary Information for

**A phosphorus modified mesoporous AuRh film as an efficient bifunctional
electrocatalyst for urea-assisted energy-saving hydrogen production**

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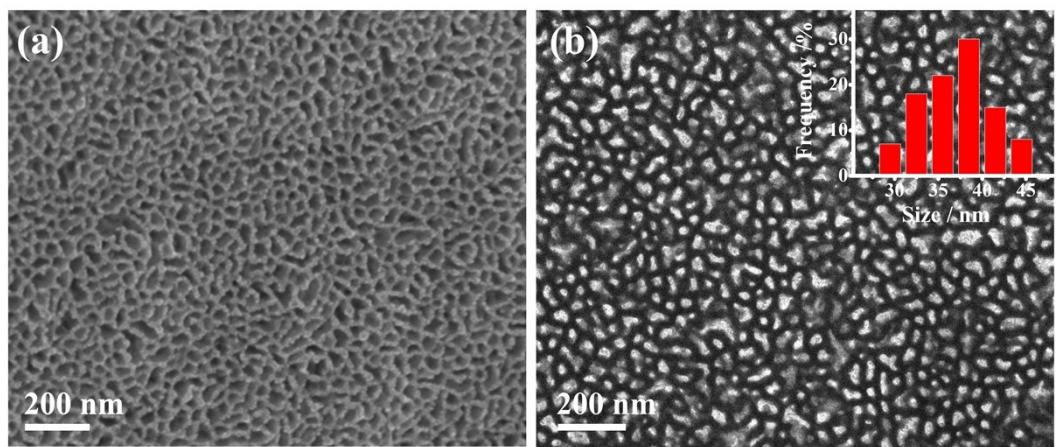


Fig. S1 (a) SEM image of the mAuRh film/NF and (b) TEM image of the mAuRh film. The inset of (b) represents the pore-size distribution histogram of mAuRh film.

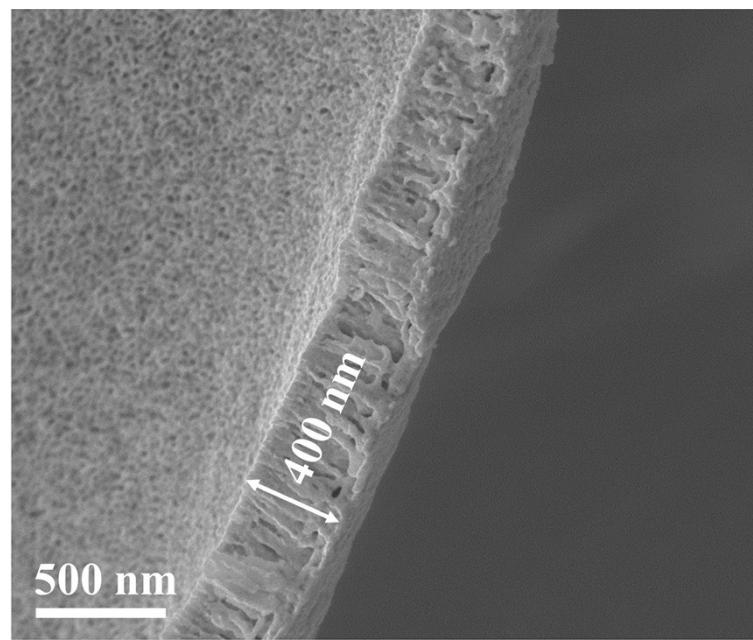


Fig. S2 Cross-sectional SEM image of P-mAuRh film/NF.

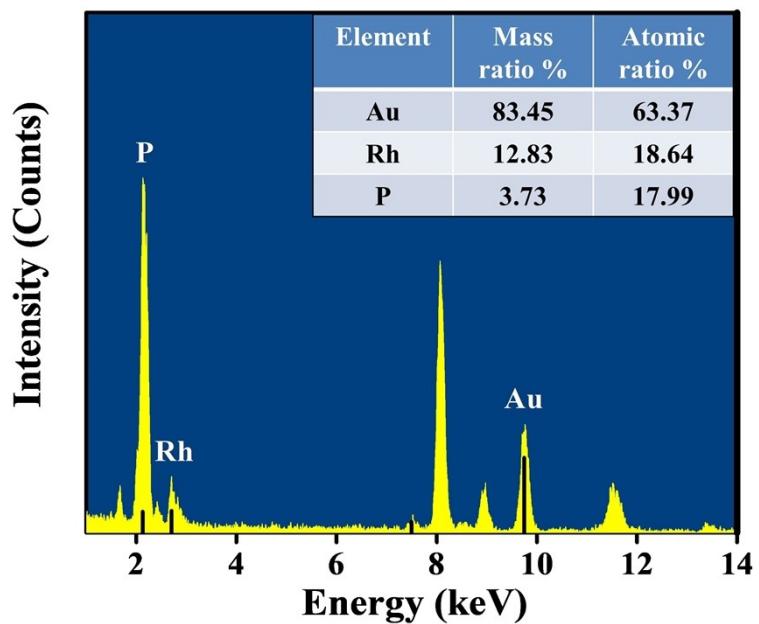


Fig. S3 EDX spectrum of P-mAuRh film.

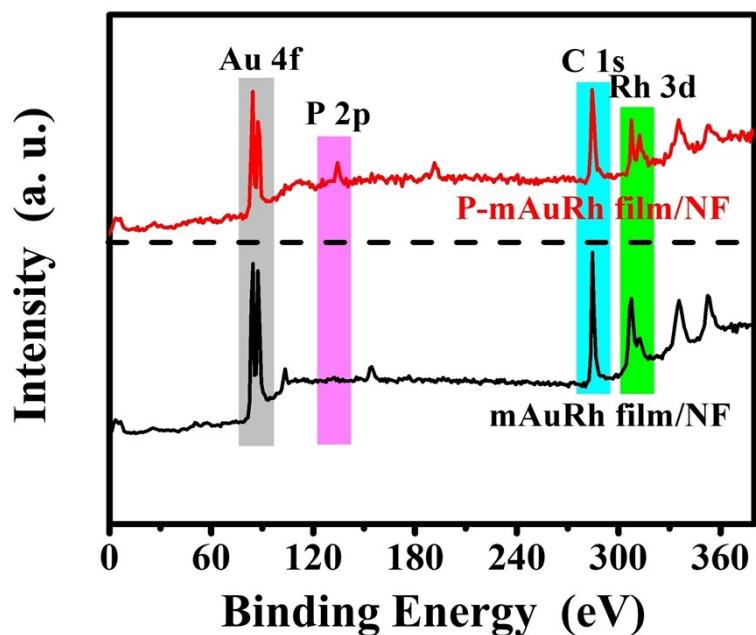


Fig. S4 XPS survey spectra of P-mAuRh film and mAuRh film.

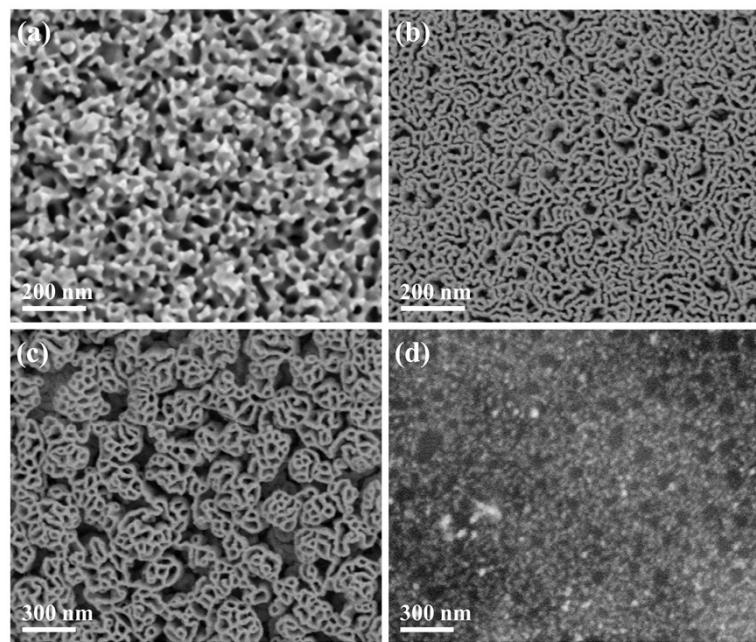


Fig. S5 SEM images of mAuRh film/NF obtained with different molar ratios of metal precursors:
(a) Au₄Rh₀, (b) Au₁Rh₁, (c) Au₁Rh₃ and (d) Au₀Rh₄.

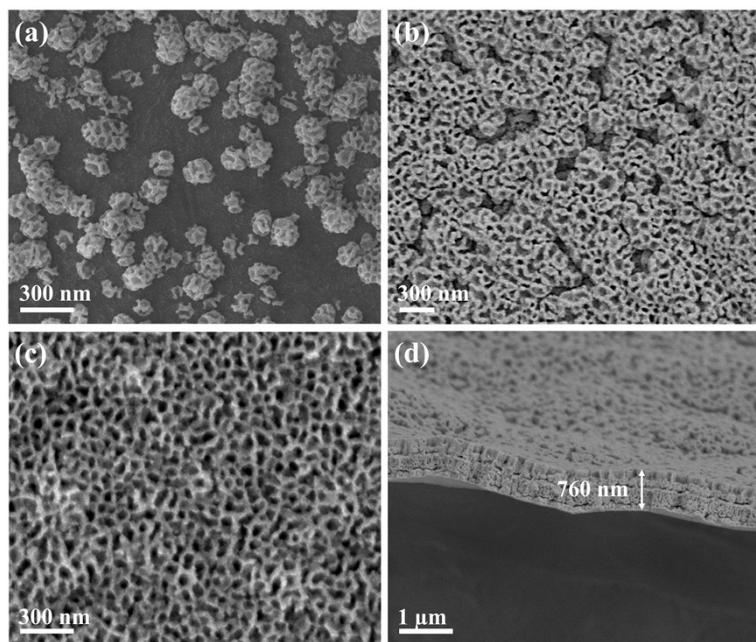


Fig. S6 SEM images of mAuRh film/NF taken from different soaking times: (a) 20 min, (b) 60 min, (c) 180 min, respectively, and (d) the cross-section image of mAuRh film/NF taken from 180 min.

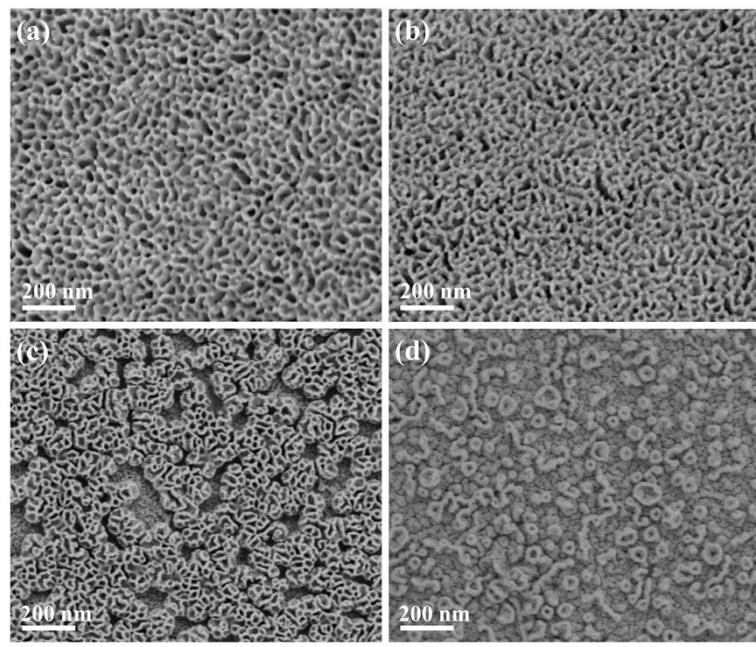


Fig. S7 SEM images of P-mAuRh film/NF under different heating treatment times: (a) 3 h, (b) 4 h, (c) 6 h and (d) 8 h.

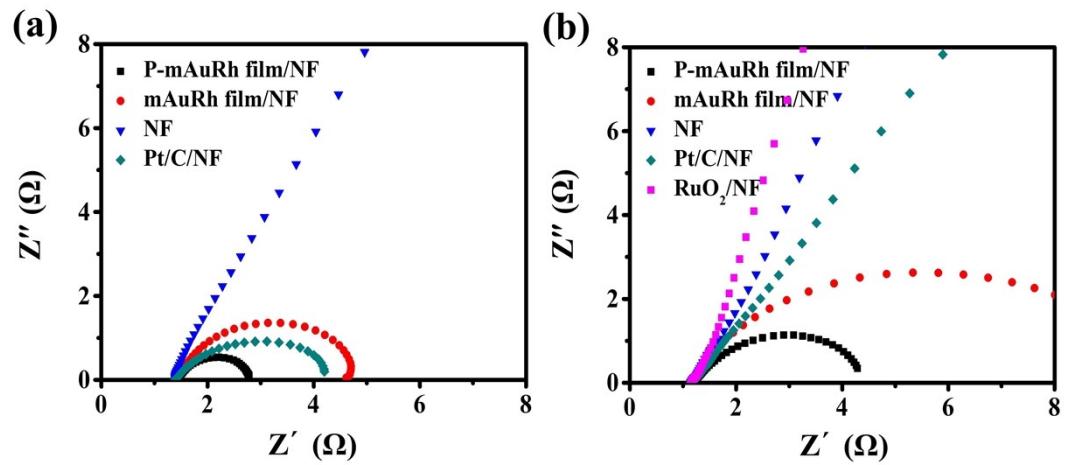


Fig. S8 EIS curves of different catalysts for (a) HER at -0.12 V (vs. RHE) and (b) UOR at 1.35 V (vs. RHE).

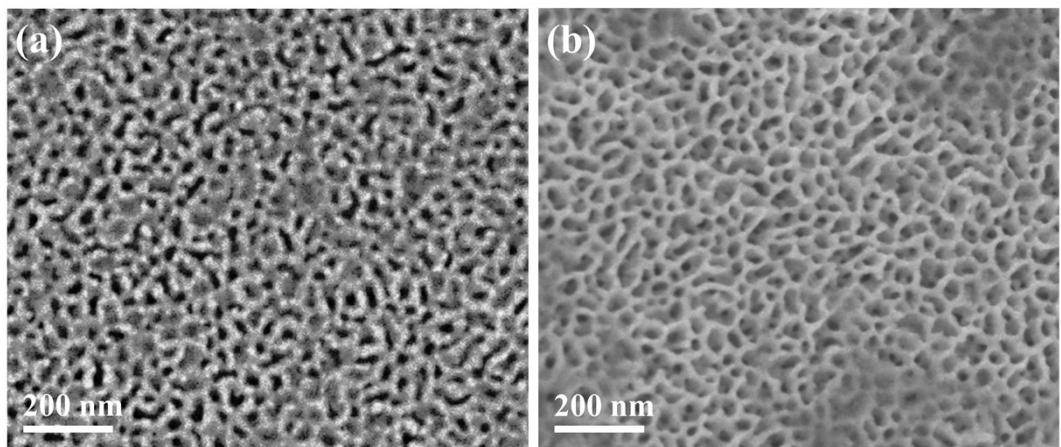


Fig. S9 SEM images of P-mAuRh film/NF after long-term durability tests for (a) HER and (b) UOR.

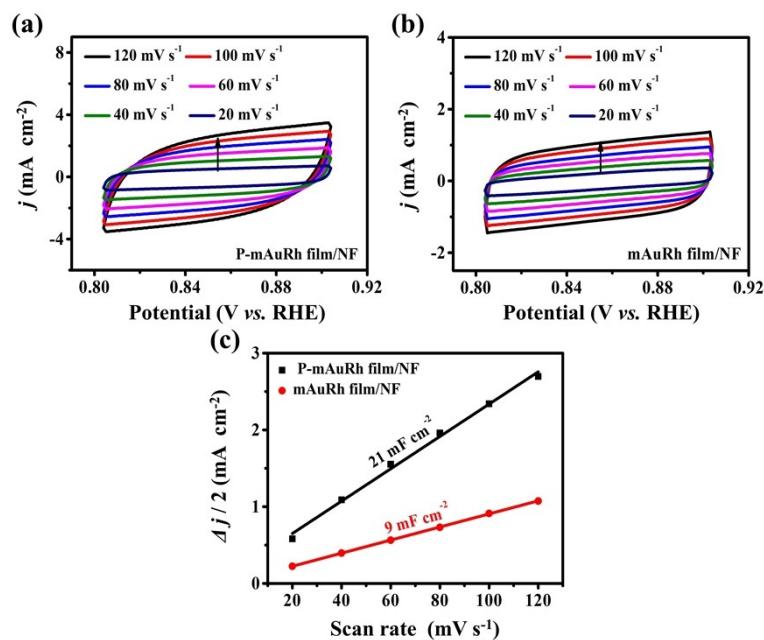


Fig. S10 CV curves of (a) P-mAuRh film/NF, (b) mAuRh film/NF from 0.804 to 0.904 V (vs. RHE). (c) C_{dl} curves of different catalysts at 0.854 V (vs. RHE) in 1.0 M KOH solution.

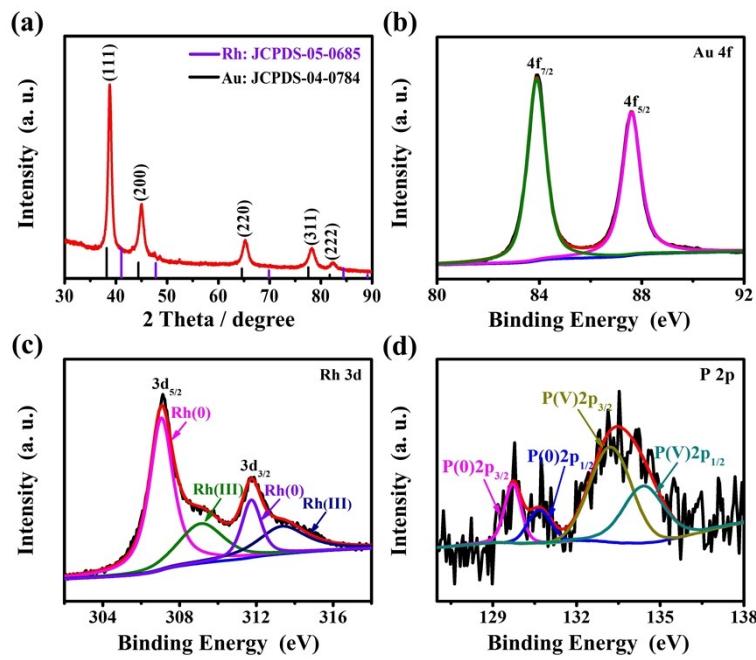


Fig. S11 (a) XRD pattern, (b) Au 4f, (c) Rh 3d and (d) P 2p XPS spectra of P-mAuRh film/NF after long-term durability test for UOR.

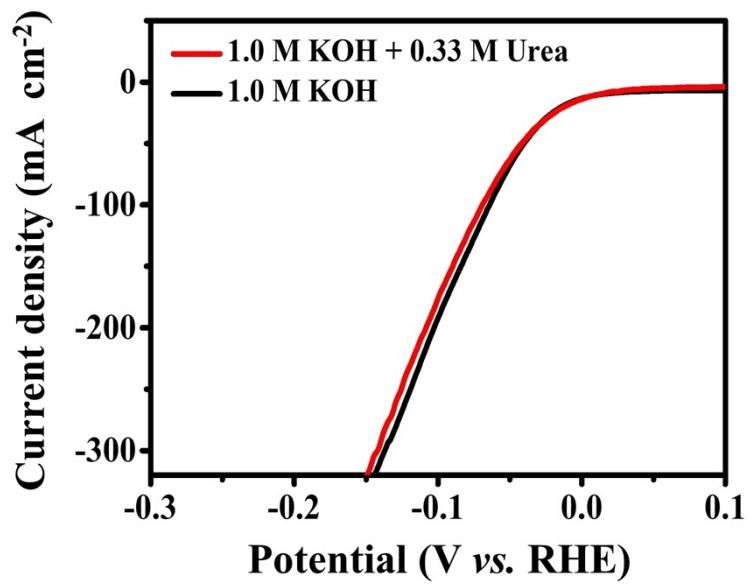


Fig. S12 LSV curves of P-mAuRh film/NF in 1.0 M KOH with and without 0.33 M urea for HER.

Table S1 HER performance comparison between the P-mAuRh film/NF and some other reported electrocatalysts.

| Electrocatalysts | Electrolyte s | substrate | Current density (mA cm ⁻²) | Overpotential (mV vs. RHE) | Ref. |
|--|------------------|-----------|--|-------------------------------|----------------------|
| P-mAuRh film/NF | 1.0 M KOH | NF | 50 | 45 | This work |
| | | | 100 | 73 | |
| Co-NCNTFs/NF | 1.0 M KOH | NF | 10 | 141 | [1] |
| Ni ₂ P-UNMs/NF | 1.0 M KOH | NF | 10 | 75 | [2] |
| Mo/Mn- Ni _x S _y /NF | 1.0 M KOH | NF | 10 | 144 | [3] |
| Ni _{1.8} Cu _{0.2} -P/NF | 1.0 M KOH | NF | 10 | 78 | [4] |
| Co-NT/NF | 1.0 M KOH | NF | 10 | 178 | [5] |
| Mn-doped NiS ₂ /NF | 1.0 M KOH | NF | 10 | 71 | [6] |
| NiSe/NF | 1.0 M KOH | NF | 10 | 96 | [7] |
| N-NiS/MoS ₂ -NF | 1.0 M KOH | NF | 10 | 71 | [8] |
| NiCoP-CoP/NF | 1.0 M KOH | NF | 10 | 73 | [9] |
| Ni _x P-400 | 1.0 M KOH | NF | 10 | 71 | [10] |

Table S2 UOR performance comparison between the P-mAuRh film/NF and some other reported electrocatalysts.

| Electrocatalysts | Electrolytes | substrate | Current density (mA cm ⁻²) | Potential (V vs. RHE) | Ref. |
|---|-----------------------------------|-----------|--|-----------------------|-----------|
| P-mAuRh film/NF | 1.0 M KOH/ 0.33 M urea | NF | 50 | 1.35 | This work |
| | | | 100 | 1.358 | |
| MoS ₂ /Ni ₃ S ₂ /NF | 1.0 M KOH/ 0.33 M urea | NF | 200 | 1.35 | [11] |
| α - Ni(OH) ₂ /NF | 1.0 M KOH/ 0.33 M urea | NF | 10 | 1.44 | [12] |
| | | | 100 | 1.73 | |
| NiCo ₂ S ₄ NS/CC | 1.0 M KOH/ 0.33 M urea | CC | 10 | 1.45 | [13] |
| Ni/C-1 | 1.0 M KOH/ 0.33 M urea | - | 10 | 1.36 | [14] |
| Ni _(10%) Pd _(10%) /OMC | 1.0 M KOH/ 0.33 M urea | OMC | 30 | 1.346 | [15] |
| Ni(OH) ₂ //F-Ni ₃ S ₂ /NF (NF-20) | 1.0 M KOH/ 0.33 M urea | NF | 50 | 1.36 | [16] |
| Fe doped α - Ni(OH) ₂ /NF | 1.0 M KOH/ 0.33 M urea | NF | 100 | 1.408 | [17] |
| Ni(OH) ₂ NS@NW/NF | 1.0 M KOH/ 0.33 M urea | NF | 10 | 1.408 | [18] |
| Ni-Co ₉ S ₈ /CC | 1.0 M KOH/ 0.33 M urea | CC | 100 | 1.43 | [19] |
| Ni _{1.5} Mn _{1.5} O ₄ | 1.0 M KOH/ 0.33 M urea | - | 6.9 | 1.338 | [20] |
| Ni MOF | 1.0 M KOH/ 0.33 M urea | - | 10 | 1.36 | [21] |

Table S3. The overall urea-assistant water splitting performance comparison between the P-mAuRh film/NF and some other reported electrocatalysts.

| Electrocatalysts | | Electrolytes | Current density (mA cm ⁻²) | Voltage (V) | Ref. |
|--|--|------------------------------|--|--------------|-----------|
| Anode | Cathode | | | | |
| P-mAuRh film/NF | P-mAuRh film/NF | 1.0 M KOH/0.33 M urea | 10 | 1.33 | This work |
| | | | 50 | 1.415 | |
| | | | 100 | 1.47 | |
| MoS ₂ /Ni ₃ S ₂ /NF | MoS ₂ /Ni ₃ S ₂ /NF | 1.0 M KOH/0.33 M urea | 20 | 1.45 | [11] |
| NiCo ₂ S ₄ NS/CC | NiCo ₂ S ₄ NS/CC | 1.0 M KOH/0.33 M urea | 10 | 1.45 | [13] |
| Ni/C-1/CC | Ni/C-1/CC | 1.0 M KOH/0.33 M urea | 10 | 1.6 | [14] |
| N _{i(10%)} Pd _(10%) /OMC | N _{i(10%)} Pd _(10%) /OMC | 1.0 M KOH/0.33 M urea | 30 | 1.35 | [15] |
| Ni(OH) ₂ NS@NW/NF | Co ₂ P NW/NF | 1.0 M KOH/0.33 M urea | 5 | 1.58 | [18] |
| Ni-Co ₉ S ₈ /CC | Ni-Co ₉ S ₈ /CC | 1.0 M KOH/0.33 M urea | 10 | 1.52 | [19] |
| NiSe ₂ -NiO | commercial Pt/C | 1.0 M KOH/0.33 M urea | 10 | 1.39 | [22] |
| MoS ₂ /Ni ₃ S ₂ /NiFe-LDH | MoS ₂ /Ni ₃ S ₂ /NiFe-LDH | 1.0 M KOH/0.5 M urea | 50 | 1.343 | [23] |
| Ni-MOF-0.5 | Ni-MOF-0.5 | 1.0 M KOH/0.5 M urea | 10 | 1.52 | [24] |
| Ni ₃ N-350/NF | Ni ₃ N-350/NF | 1.0 M KOH/0.5 M urea | 100 | 1.51 | [25] |

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