## Ca-ion Modified Vanadium Oxide Nanoribbons with Enhanced Zn-ion Storage Capability

Dezhou Zheng,<sup>a†</sup> Xiaokang Pei,<sup>a†</sup> Hai Lin,<sup>a</sup> Hongwei Tang,<sup>a</sup> Yin Song,<sup>a</sup> Qi Feng,<sup>a</sup> Guangxia Wang,<sup>a</sup> Wei Xu,<sup>a</sup> Fuxin Wang,<sup>\*a</sup> and Xihong Lu<sup>\*ab</sup>

<sup>a</sup>School of Applied Physics and Materials, Wuyi University, Jiangmen 529020, PR
China. E-mail: wangfux91@126.com; luxh6@mail.sysu.edu.cn
<sup>b</sup> The Key Lab of Low-carbon Chem & Energy Conservation of Guangdong Province,
MOE of the Key Laboratory of Bioinorganic and Synthetic Chemistry, School of
Chemistry, Sun Yat-Sen University, Guangzhou 510275, PR China



Figure S1. The EDS mapping of the VO sample.



Figure S2. (a, c) TEM and HRTEM images of VO. (b) SAED image.



Figure S3 V  $2p_{3/2}$  spectra of the synthesized VO and CVO samples.



Figure S4. O 1s spectra of the VO and CVO sample.



Figure S5 CV and GCD curves of all batteries at a scan rate of 0.1 mV s<sup>-1</sup> and current density of 2 A  $g^{-1}$ .



Figure S6. The first CV curve of VO and CVO samples at a scan rate of 0.1 mV s<sup>-1</sup>.



Figure S7. Capacity retention of VO and CVO.



Figure S8. Capacitive contribution (inset) and diffusion contribution of CVO and VO electrode at 0.1 mV s<sup>-1</sup>.



Figure S9 (a) The GCD profile of CVO at 2 A g<sup>-1</sup>. (b) Ex-situ XRD patterns of CVO



Figure S10 The GCD profile of CVO at 2 A g<sup>-1</sup>, the insets show the SEM for different state during the charging and discharging process.