## Supporting information for

Engineering Hierarchical Sb<sub>2</sub>S<sub>3</sub>/N-C from Nature Minerals with Stable Phase-change towards All-climate Energy Storage

## **Figures and Captions**

Fig. S1 The SEM images of SS-OM (a, d), SS-SO (b, e) and SS-NC (c, f)

Fig. S2 The TEM of SS-OM(a), SS-SO(b) and SS-NC (c)

Fig. S3 Contact angle measurement: for SS-OM (a), SS-NC (b)

Fig. S4 The electrochemical properties of three samples at 0.5 A g<sup>-1</sup> during 150 cycles: cycling stability (a), capacity retention (b).

Fig. S5 The electrochemical properties of full cell about LiFePO4 vs. SS-NC: cycling stability (a), rate abilities(b), charge-discharge platforms at different cycles (c), differential median curves (d).

Fig. S6 The multi-cycling CV curves from 1st to 5th at 0.3 mV s<sup>-1</sup> of SS-OM (a) and SS-SO (b)

Fig. S7 Various CV curves at 2.0 mV s<sup>-1</sup> of SS-OM (a) and SS-SO (b)

Fig. S8 the log(v) versus log(i) for peak1 of the as-targeted samples (a), the ratio of capacitive contributions at 0.3 mV s<sup>-1</sup> for SS-OM (b) and SS-SO (c)

Fig. S9 the Nyquist plots of SS-OM (a) and SS-SO (b)

Fig. S10 the Nyquist Plots and relative fitting circuit at 50<sup>th</sup> of these three as-samples Fig. S11 the linear relations of Z'' vs.  $\omega^{-1/2}$  for SS-OM (a) and SS-SO (b), the relationships of Phase Angle vs. Frequency of SS-OM (c) and SS-SO (d).

Fig. S12 the  $D_{Li^+}$  values during the charge process for these three as-samples

Fig. S13 The electrochemical properties at low temperatures: their cycling properties at 0.5 A g<sup>-1</sup> (a) and 1.0 A g<sup>-1</sup> (d), the comparison of charge/discharge platforms at RT/LT for SS-OM (b) and SS-SO (c), their linear relationship of  $I_p$  versus  $v^{1/2}$  at Peak1/3 (e),

the analysis of CV curves at RT/LT (0.7 mV s<sup>-1</sup>) for SS-OM (f).

Fig. S14 SEM images avout the electrodes after 100 cycles: SS-OM (a), SS-NC (b).

Fig.S 15 Ex-situ XRD patterns of SS-NC

Table S1 The content for all elements in the SS-NC material

Table S2 The electrochemical properties of Sb2S3 samples from the previous reports.



Fig. S1 The SEM images of SS-OM (a, d), SS-SO (b, e) and SS-NC (c, f)



Fig. S2 The TEM of SS-OM(a), SS-SO(b) and SS-NC (c)



Fig. S3 Contact angle measurement: for SS-OM (a), SS-NC (b)



Fig. S4 The electrochemical properties of three samples at 0.5 A g-1 during 150 cycles : cycling stability (a), capacity retention (b).



Fig. S5 The electrochemical properties of full cell about LiFePO<sub>4</sub> vs. SS-NC: cycling stability (a), rate abilities(b), charge-discharge platforms at different cycles (c), differential median curves (d).



Fig. S6 The multi-cycling CV curves from 1st to 5th at 0.3 mV s<sup>-1</sup> of SS-OM (a) and SS-SO (b)



Fig. S7 Various CV curves at 2.0 mV s<sup>-1</sup> of SS-OM (a) and SS-SO (b)



Fig. S8 the log(v) versus log(i) for peak1 of the as-targeted samples (a), the ratio of capacitive contributions at 0.3 mV s<sup>-1</sup> for SS-OM (b) and SS-SO (c)



Fig. S9 the Nyquist plots of SS-OM (a) and SS-SO (b)



samples



Fig. S11 the linear relations of Z'' vs.  $\omega^{-1/2}$  for SS-OM (a) and SS-SO (b), the relationships of Phase Angle vs. Frequency of SS-OM (c) and SS-SO (d).



Fig. S12 the  $D_{\mathrm{Li^+}}$  values during the charge process for these three as-samples



Fig. S13 The electrochemical properties at low temperatures: their cycling properties at 0.5 A g<sup>-1</sup> (a) and 1.0 A g<sup>-1</sup> (d), the comparison of charge/discharge platforms at RT/LT for SS-OM (b) and SS-SO (c), their linear relationship of  $I_p$  versus  $v^{1/2}$  at Peak1/3 (e), the analysis of CV curves at RT/LT (0.7 mV s<sup>-1</sup>) for SS-OM (f).



Fig. S14 SEM images avout the electrodes after 100 cycles: SS-OM (a), SS-NC (b).



Fig.S 15 Ex-situ XRD patterns of SS-NC

Atoms	S2p	Cls	NIs	Sb3d	Ols
Atomic content (%)	38.9	22.05	8.99	24.71	1.96

Table S1 The content for all elements in the SS-NC material

			Canacity (avolo		
Materials	Voltage (V)	Temperature (°C)	Capacity (cycie number) at A g <sup>-1</sup>	capacity retention (%)	Ref.
Sb <sub>2</sub> S <sub>3</sub> /lotus- polle	0.01-2.50	~	591 (100) at 0.1	59.8	Ref[1]
Sb <sub>2</sub> S <sub>3</sub> /MHCS nanospheres	0.01-3.00	~	745.3 (160) at 0.1	59.6	Ref [2]
Sb <sub>2</sub> S <sub>3</sub> /MMCN	0.01-3.00	room temperature	226.0 (100) at 0.1	39.0	Ref [3]
Sb <sub>2</sub> S <sub>3</sub> /ppy	0.01-3.00	room temperature	389.0 (160) at 0.1	83.0	Ref [3]
Sb <sub>2</sub> S <sub>3</sub> /MMCN @ppy	0.01-3.00	room temperature	827.0 (100) at 0.1	92.0	Ref [3]
Sb <sub>2</sub> S <sub>3</sub> /CS	0.01-2.50	25	566 (200) at 0.2	63.2	Ref [4]
Few-layer 2D Sb <sub>2</sub> S <sub>3</sub> (2D-SS)	0.01-3.00	~	607 (200) at 0.2	60.9	Ref [5]
p-Sb <sub>2</sub> S <sub>3</sub> (ss)	0.01-3.00	room temperature	311 (50) at 0.1	41.5	Ref [6]
rGO/Sb <sub>2</sub> S <sub>3</sub> (ss)	0.01-3.00	room temperature	596 (50) at 0.1	62.7	Ref [6]
rGO/Sb <sub>2</sub> S <sub>3</sub> (ht)	0.005-3.00	room temperature	658 (50) at 0.1	48.7	Ref [6]
bundle-like Sb <sub>2</sub> S <sub>3</sub>	0.01-3.00	~	548 (100) at 0.1	79.8	Ref [7]
Sb <sub>2</sub> S <sub>3</sub> /G	0.01-3.00	room temperature	670 (200) at 0.2	89.3	Ref [8]
rGO-Sb2S3- 0.5h	0.00-2.50	~	750 (50) at 0.25	82.5	Ref [9]

Table S2 The electrochemical properties of  $Sb_2S_3$  samples from the previous reports.

rGO-Sb2S3- 5.0h	0.00-2.50	~	750 (50) at 0.25	90.0	Ref [9]
Sb <sub>2</sub> S <sub>3</sub> - Co <sub>9</sub> S <sub>8</sub> /NC	0.01-3.00	~	715 (100) at 0.2	71.5	Ref [10]
CNN@Sb <sub>2</sub> S <sub>3</sub>	0.01-3.00	ambient temperature	953 (100) at 0.2	~	Ref [11]
Sb <sub>2</sub> S <sub>3</sub> /CNT	0.01-3.00	~	443 (100) at 0.2	47.6	Ref [12]
$\begin{array}{ccc} 20\text{-}25 & \text{nm} \\ \text{colloidal} \\ \text{Sb}_2\text{S}_3 \\ \text{nanoparticles} \\ \text{(NPs)} \end{array}$	0.10-2.00	~	408.1 (1200) at 2.4	55.0	Ref [13]
Sb <sub>2</sub> S <sub>3</sub> nanorods	0.00-2.50	room temperature	298.6 (20) at 0.3375	28.6	Ref [14]
This work	0.01-3.0	-10	366.9 (100) at 1.0	90.1	

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