

Supplementary Information for

Cross-linking Structure of Self-aligned P-Type SnS Nanoplates for Highly Sensitive NO₂ Detection at Room Temperature

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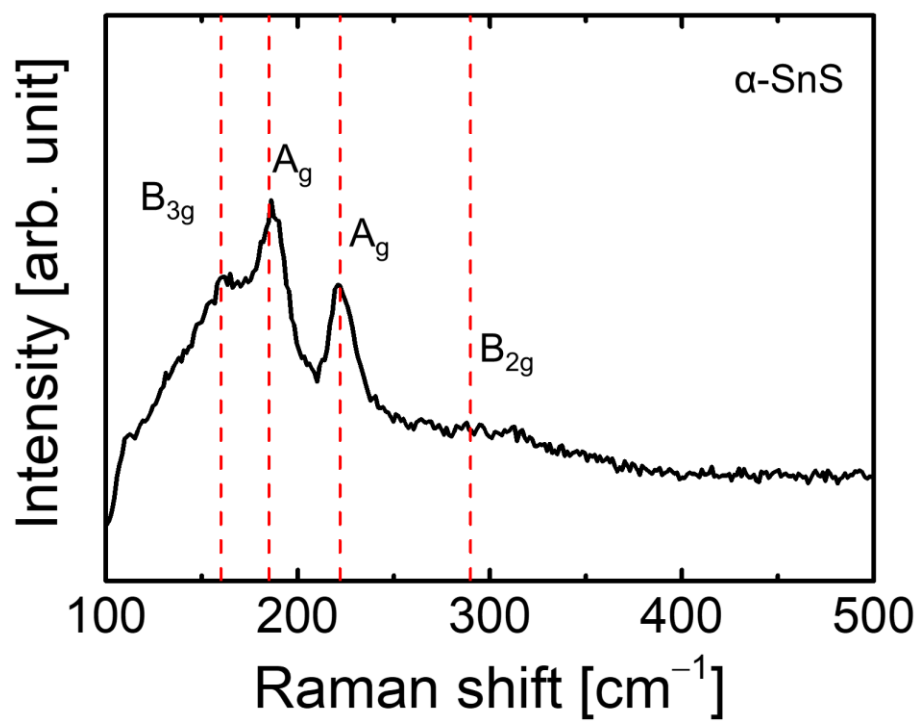


Figure S1. Raman spectrum of SnS nanoplates grown on SiO_2 nanorods.

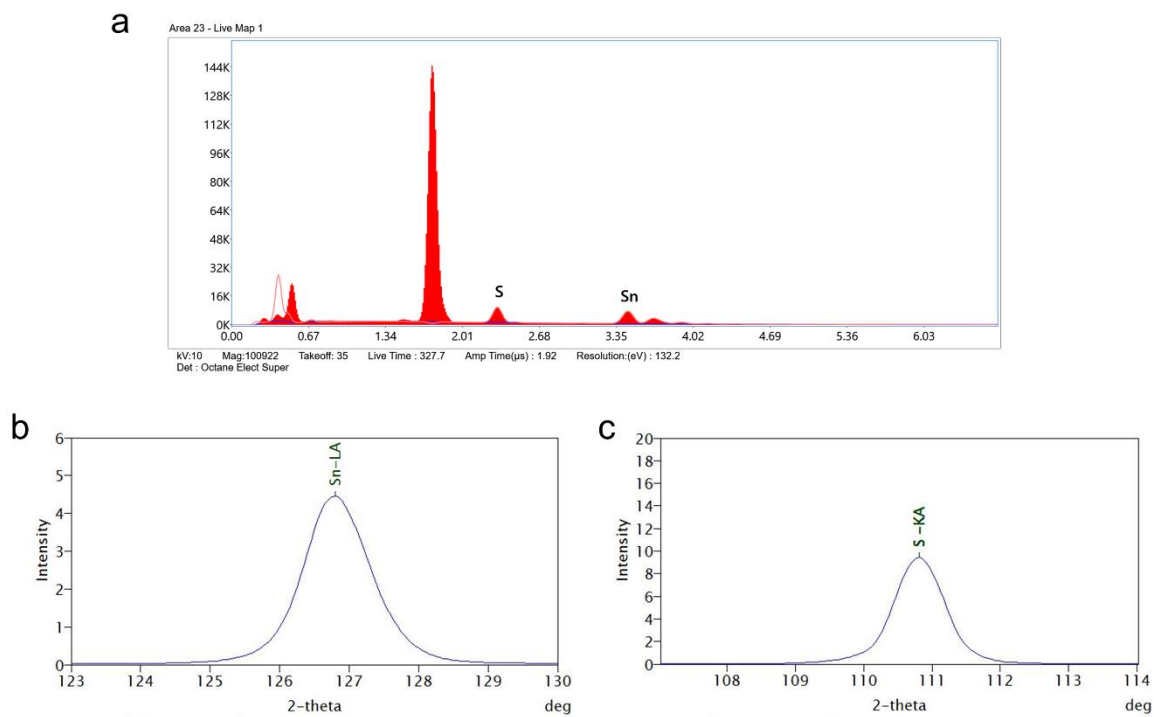


Figure S2. (a) Energy-dispersive spectra of SnS nanoplates formed on SiO₂ nanorods obtained from SEM analysis. Variation in the peak intensity of (b) Sn and (c) S elements obtained from wavelength-dispersive x-ray fluorescence.

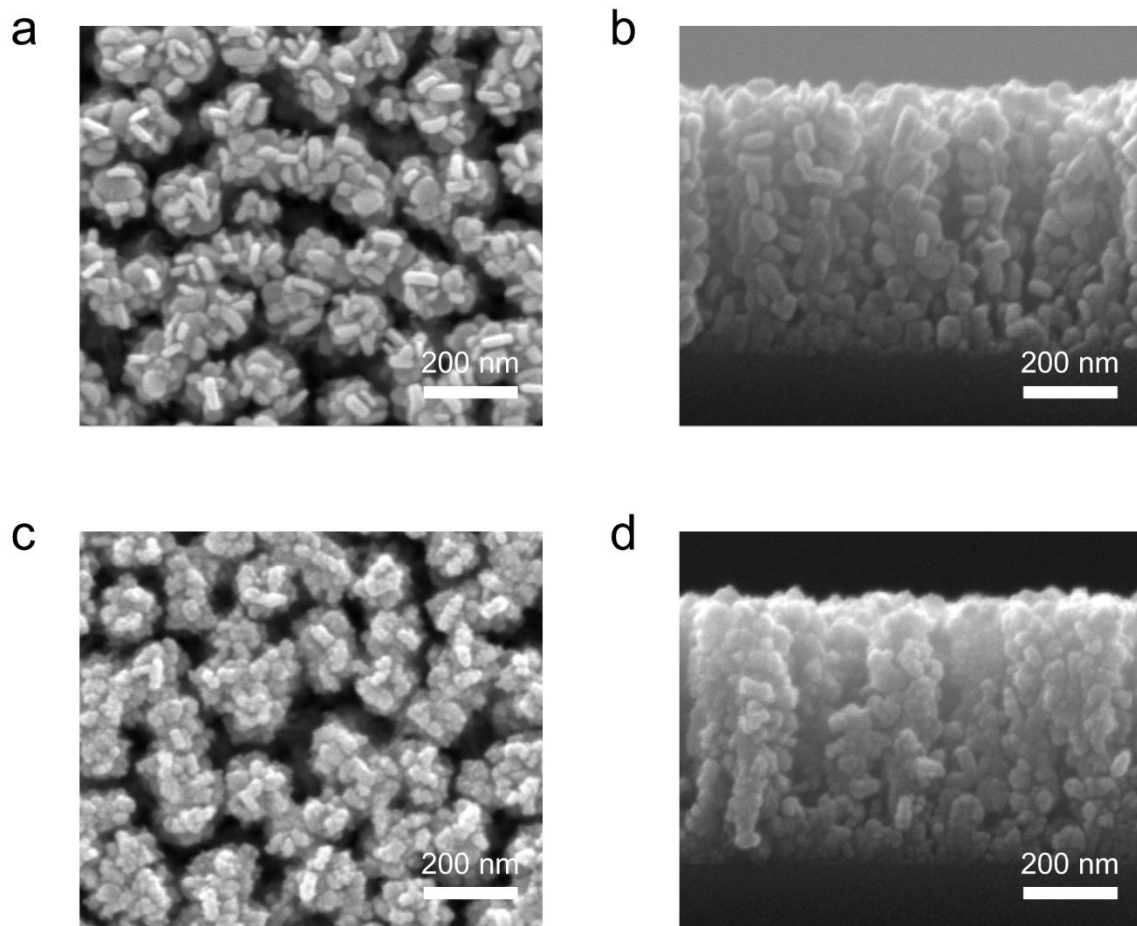


Figure S3. (a) Plan-view and (b) cross-sectional SEM images of SnS nanoplates on SiO₂ nanorods. (c) Plan-view and (d) cross-sectional SEM images of the structure after solvent cleaning using acetone, ethanol, and deionized water.

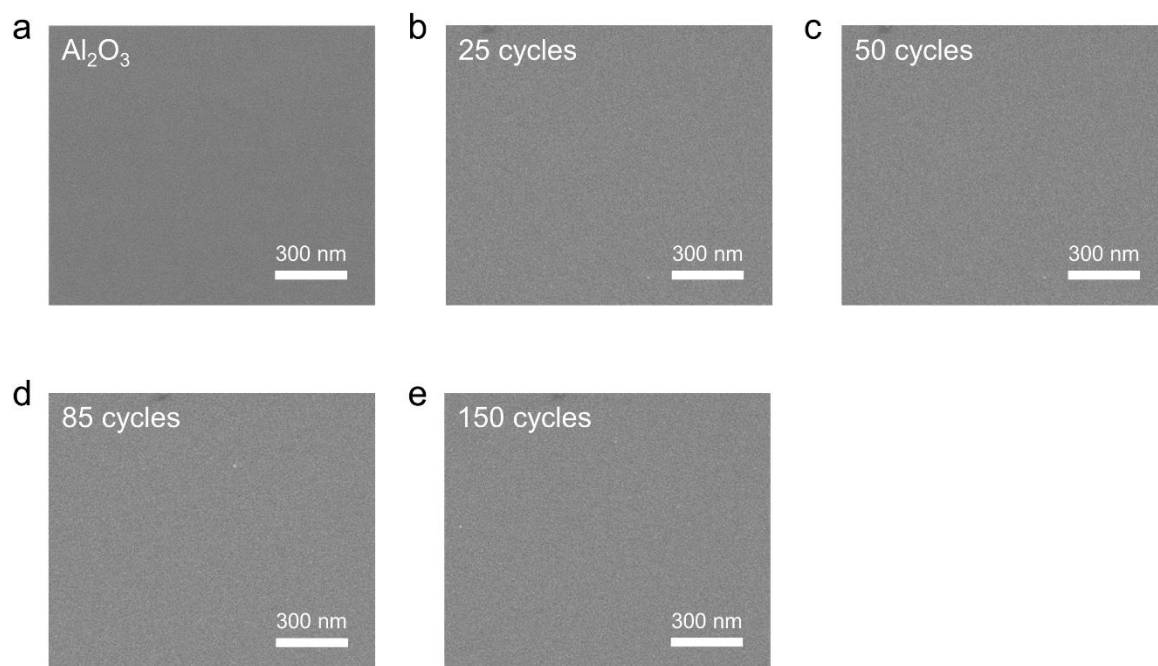
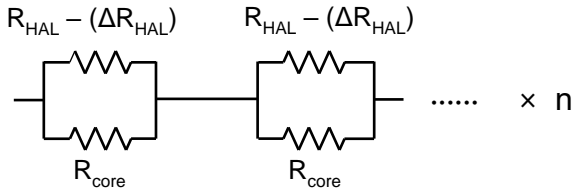


Figure S4. Plan-view SEM images of ALD-grown SnS on Al_2O_3 for (a) 0, (b) 25, (c) 50, (d) 85, and (e) 150 cycles.

Note S1



The resistance of the single SnS grain in the absence and presence of NO_2 can be expressed as $\frac{R_{core} \times R_{HAL}}{R_{core} + R_{HAL}}$ and $\frac{R_{core} \times (R_{HAL} - \Delta R_{HAL})}{R_{core} + (R_{HAL} - \Delta R_{HAL})}$, respectively, because R_{HAL} and R_{core} are connected in parallel. The R_{core} value is assumed to be a constant since the change in the R_{core} value under exposure to NO_2 is negligible. The SnS sensor was considered in the form of n grains connected in series. R_a and R_g of the SnS sensor is regarded as n times the resistance of each SnS grain.

Considering the resistance values, the response of the SnS sensor to NO_2 is developed as follows;

$$\begin{aligned}
 \text{Response} &= \frac{R_a}{R_g} - 1 = \frac{\left(\frac{R_{core} \times R_{HAL}}{R_{core} + R_{HAL}} \right) \times n}{\left(\frac{R_{core} \times (R_{HAL} - \Delta R_{HAL})}{R_{core} + (R_{HAL} - \Delta R_{HAL})} \right) \times n} - 1 \\
 &= \frac{\cancel{R_{core}} \times R_{HAL} (R_{core} + R_{HAL} - \Delta R_{HAL})}{\cancel{R_{core}} (R_{HAL} - \Delta R_{HAL}) (R_{core} + R_{HAL})} - 1 \\
 &= \frac{R_{core} \Delta R_{HAL}}{(R_{HAL} - \Delta R_{HAL}) (R_{core} + R_{HAL})} \\
 &= \frac{1}{\left(\frac{R_{HAL}}{\Delta R_{HAL}} - 1 \right) \left(1 + \frac{R_{HAL}}{R_{core}} \right)}
 \end{aligned}$$

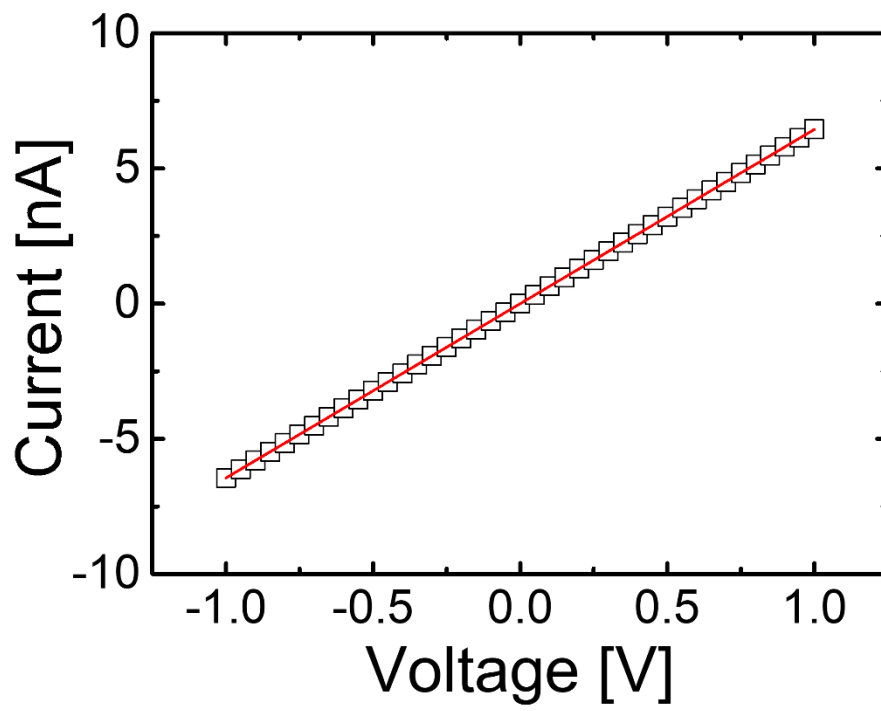


Figure S5. I-V curve of SnS nanoplates (75 ALD cycles) grown on SiO₂ nanorods.

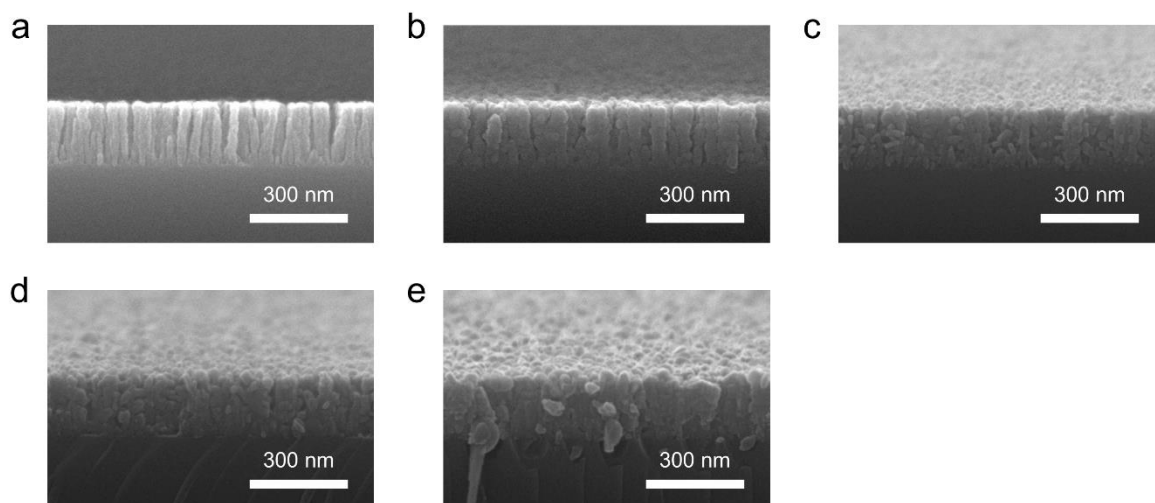


Figure S6. Cross-sectional SEM images of SnS grown on SiO₂ for (a) 0, (b) 25, (c) 50, (d) 85, and (e) 150 cycles.

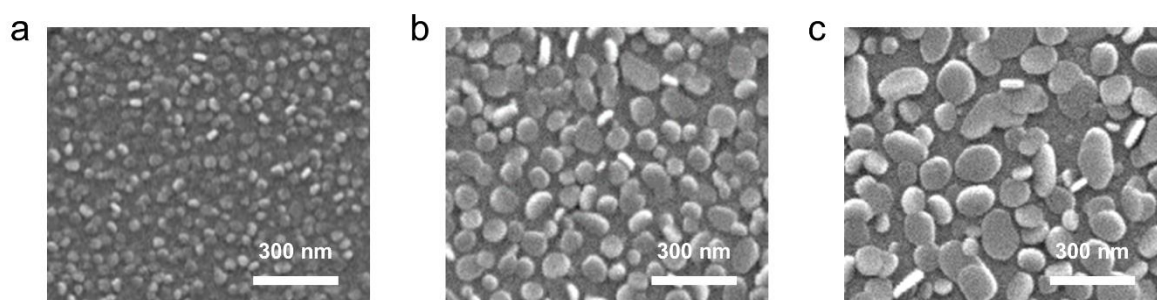


Figure S7. Plan-view SEM images of SnS grown on planar SiO₂ for (a) 25, (c) 85, and (e) 150 cycles.

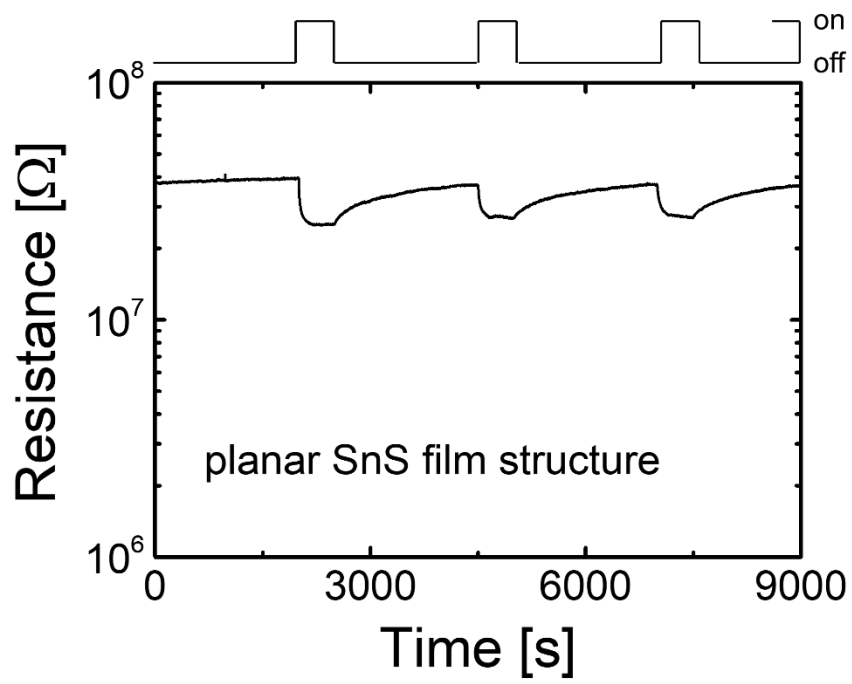


Figure S8. Variation in the resistance of the SnS planar film structured sensor grown for 900 cycles to 5 ppm NO₂.

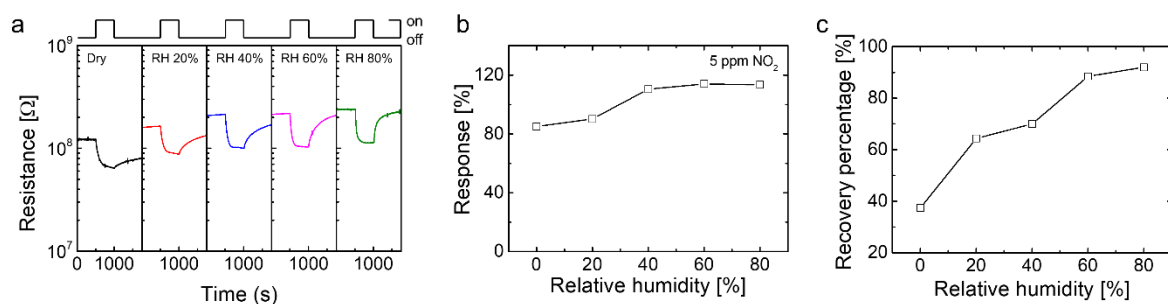


Figure S9. Variations in (a) the resistance and (b) the response of the SnS nanoplates on SiO₂ nanorods to 5 ppm NO₂ in the RH range from 0 to 80%. (c) Variation in the recovery percentage of the resistance after 500 s from the start of recovery.

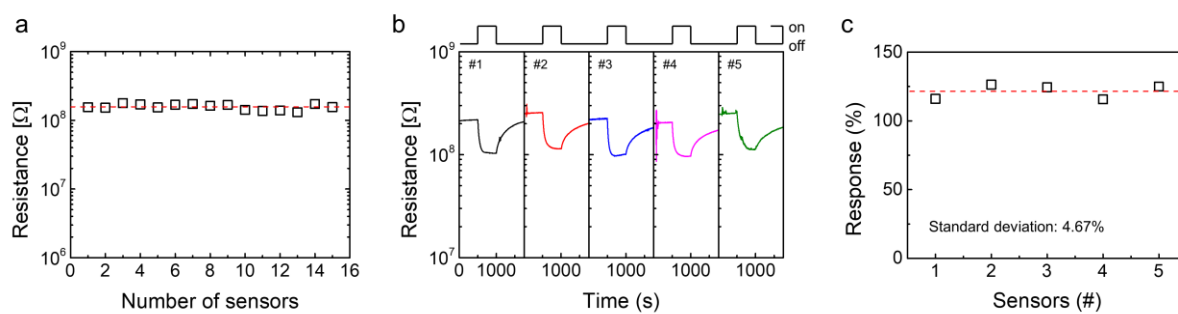


Figure S10. (a) Variation in the base resistance of 15 different SnS sensors grown on SiO₂ nanorods for 75 ALD cycles. (b) Variations in the resistance and (c) the response of five different SnS sensors under ambient condition of 60% RH at room temperature upon exposure to 5 ppm NO₂.

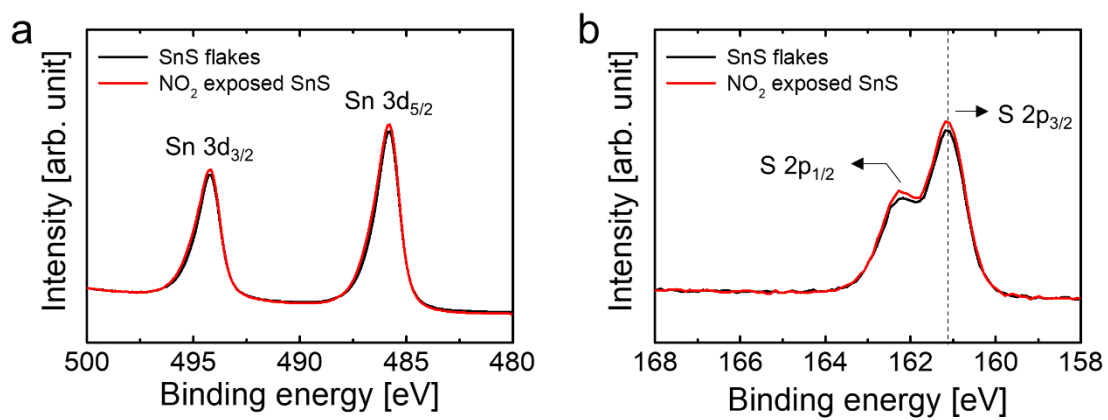


Figure S11. XPS spectra of (a) Sn 3d and (b) S 2p core levels in SnS nanoplates grown on SiO₂ nanorods before and after exposure to 100 ppm NO₂ for 5 min.

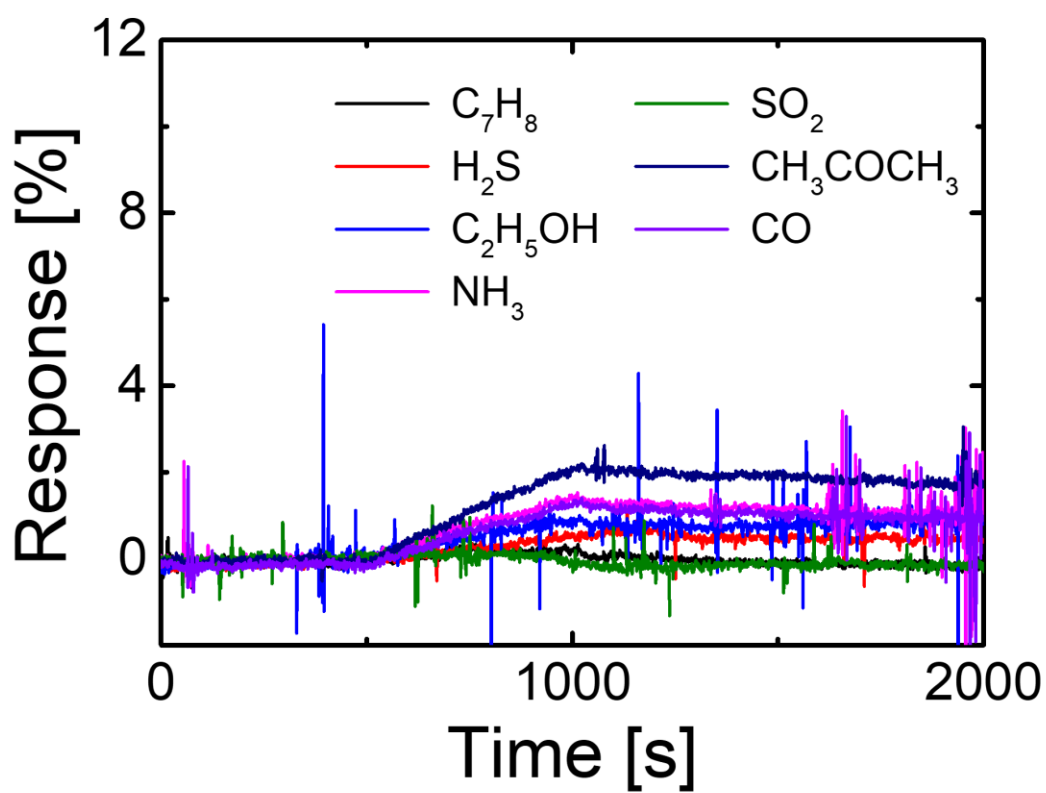


Figure S12. Variation in the response of the SnS sensor upon exposure to various gases (NO₂, C₇H₈, H₂S, C₂H₅OH, NH₃, SO₂, CH₃COOCH₃, and CO) with a concentration of 5 ppm.