

Electronic Supplementary Information (ESI)

Molecular simulation, characteristic and mechanism of thermal-responsive acetylated amylose V-type helical complex

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1. Supporting methods

1.1. Materials

High amylose from corn, 1-methylimidazole, propofol and hexane were purchased from Aladdin Co., Ltd. (Shanghai, China). Iodine was purchased from Tianjin Kemiou Chemical Reagent Co., Ltd. acetic anhydride was purchased from Sigma-Aldrich Trading Co., Ltd. (Shanghai, China).

1.2. The synthesis of acetylated amylose

The synthesis of acetylated amylose was according with literature early reported.^{S1} Briefly, 2.0 g amylose dissolved in 100 mL DMSO at 0 °C for 30 min, then 195 μ L of acetic anhydride and 164 μ L of 1-methylimidazole were added. The reaction continued at 80 °C for 4 h. Finally, a large volume of ethanol was added into above reaction mixture and acetylated amylose was precipitated. The precipitate was centrifuged and washed twice with ethanol, and then dried at 50 °C to obtain acetylated amylose powder.

1.3. The preparation of acetylated amylose-guest V-type helical complex

100 mg acetylated amylose was dissolved in 20 mL pure water, then about 20 mg iodine, 20 μ L propofol, 20 μ L hexane or 20 μ L 7-hydroxycoumarin solution (5 mg/mL) was added above solution, and then the mixture was stirred overnight. Finally, the complex solution was obtained by filtration using a 0.22 μ m filter.

2. Supporting data

Table S1. Structure parameters of acetylated amylose-guest V-type helix complex

Guest type	blank	iodine	propofol	hexane
Guest number	--	14	4	5
Diameter (nm)	1.477 \pm 0.064	1.248 \pm 0.034	1.211 \pm 0.064	1.338 \pm 0.077
Pitch (nm)	0.847 \pm 0.026	0.869 \pm 0.042	0.844 \pm 0.024	0.841 \pm 0.014

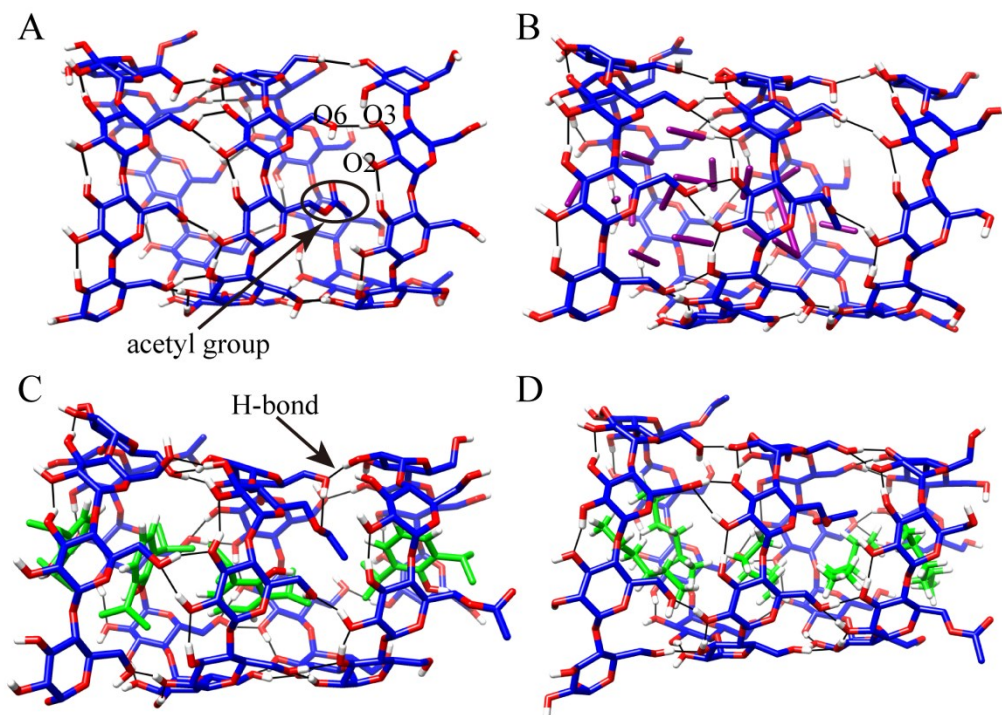


Fig. S1 Molecular structures of acetylated amylose V-type helical complexes. (A) Blank acetylated amylose helix; (B) acetylated amylose-iodine V-type helix complex; (C) acetylated amylose-propofol helix complex; (D) acetylated amylose-hexane helix complex.

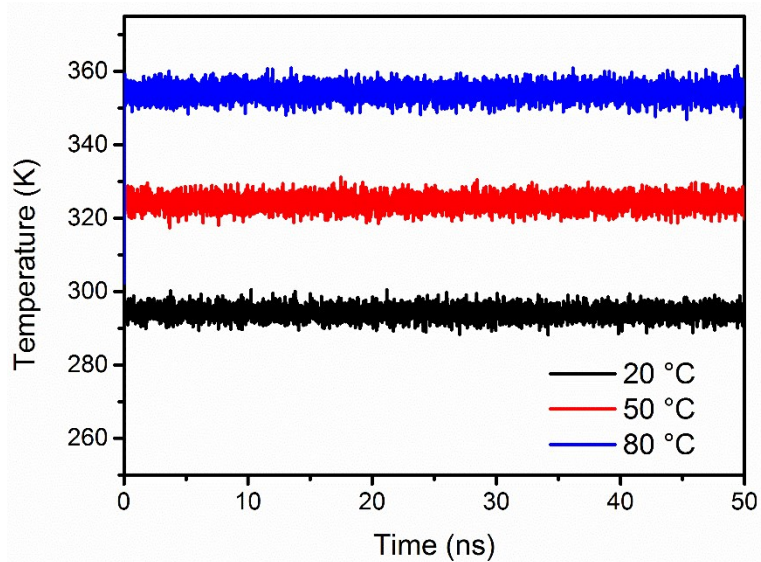


Fig. S2 The temperature fluctuation during molecular dynamic simulation.

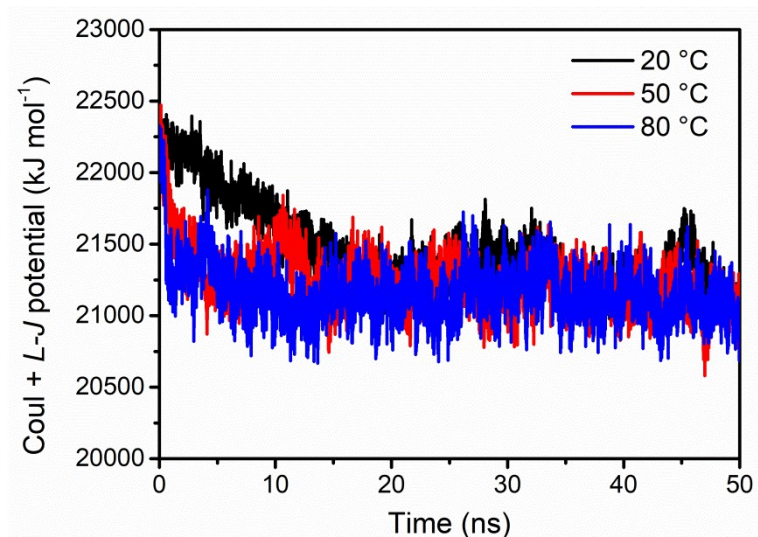


Fig. S3 The energy fluctuation (The sum of coulomb and L - J potential) of acetylated amylose during molecular dynamic simulation.

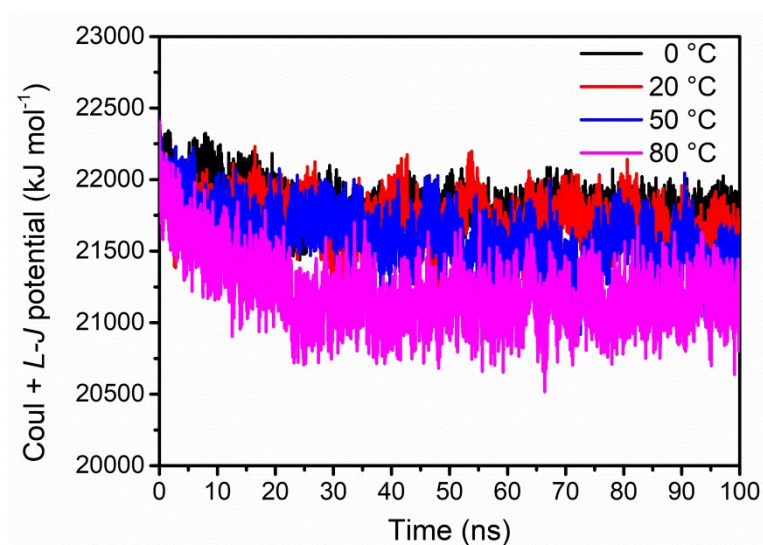


Fig. S4 Energy fluctuation of acetylated amylose in AAIHC during molecular dynamic simulation.

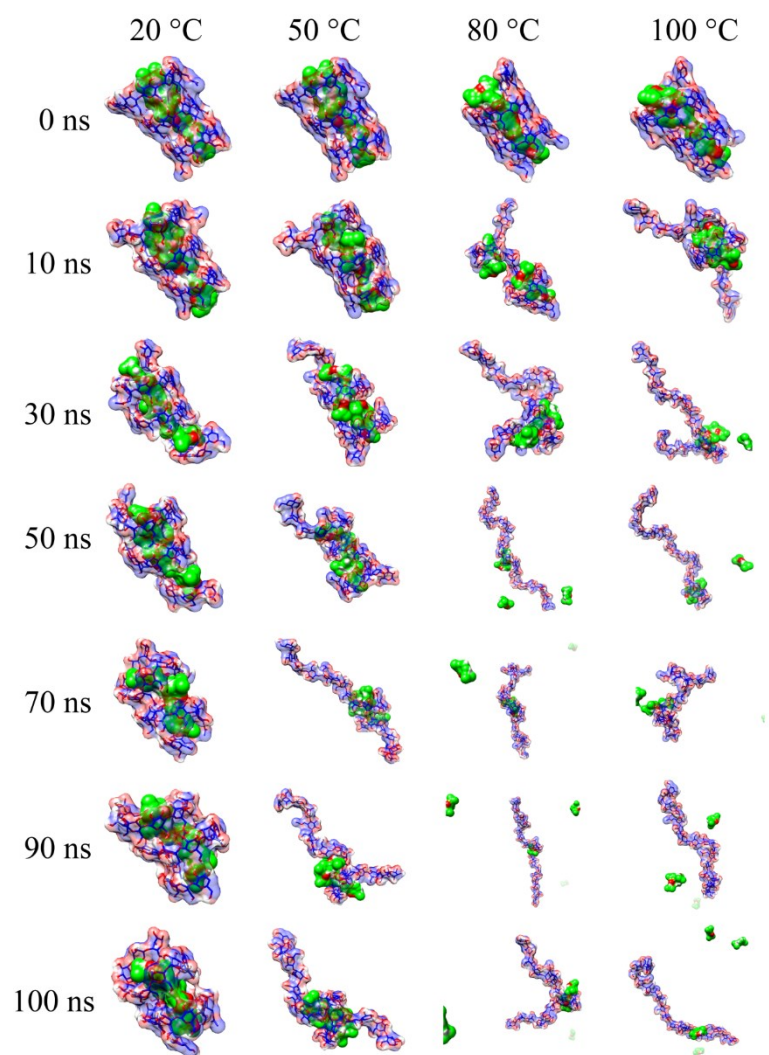


Fig. S5 Thermal-responding unfolding and releasing snapshots of AAPHc at different time-points.

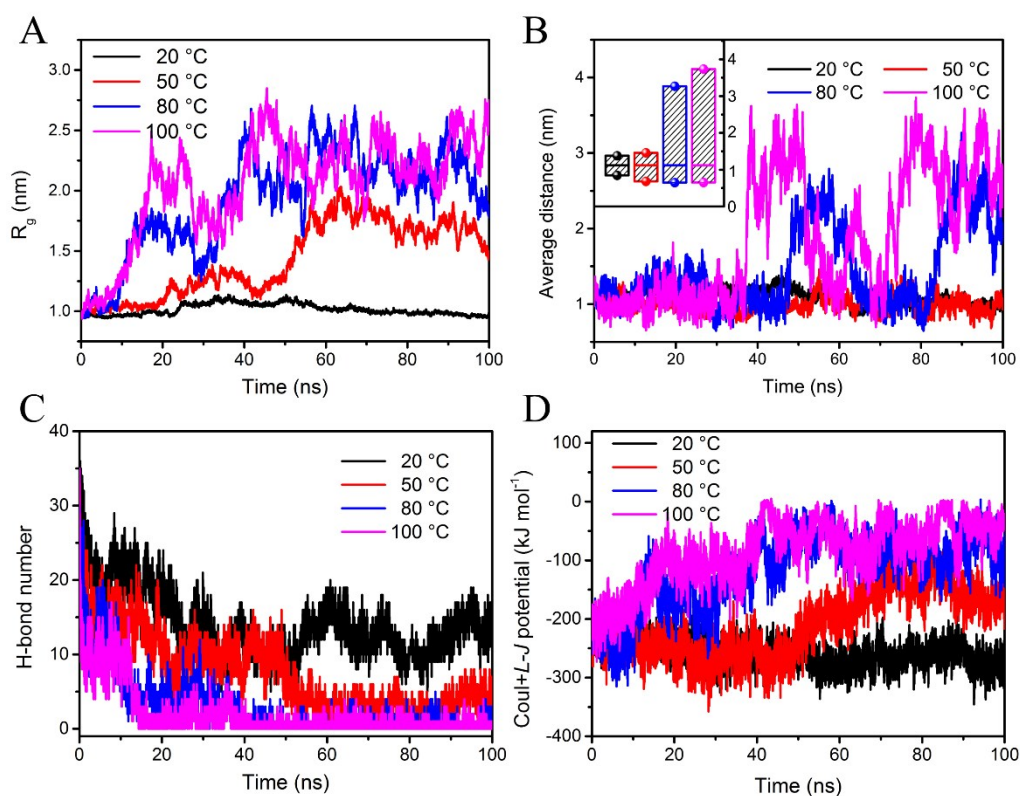


Fig. S6 Thermal-responsive helix-unfolding and guest-release characteristics in AAPHC. (A) The R_g of acetylated amylose in helix complex at different temperatures. (B) The average distance between propofol molecules and the center atom of acetylated amylose at different temperatures (inserted data is the maximum and minimum average distance at different temperatures). (C) The H-bond number of acetylated amylose-propofol V-type complex at different temperatures. (D) The binding energy between propofol molecules and acetylated amylose.

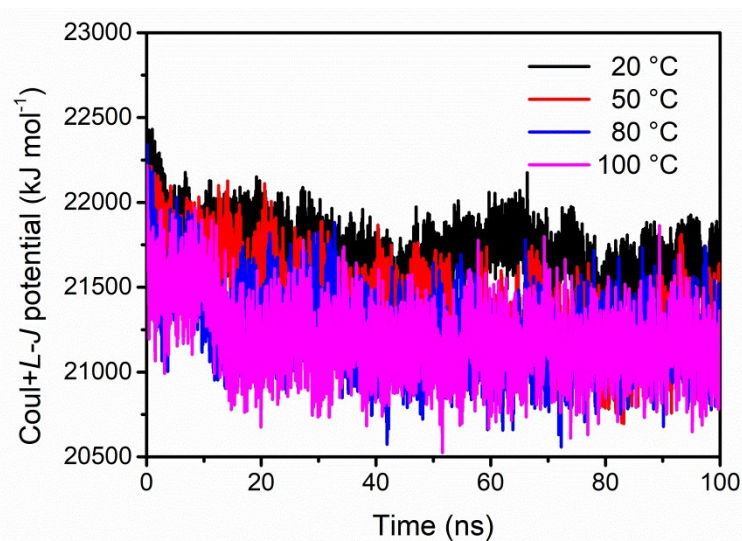


Fig. S7 Energy fluctuation of acetylated amylose in AAPHC during molecular dynamic simulation.

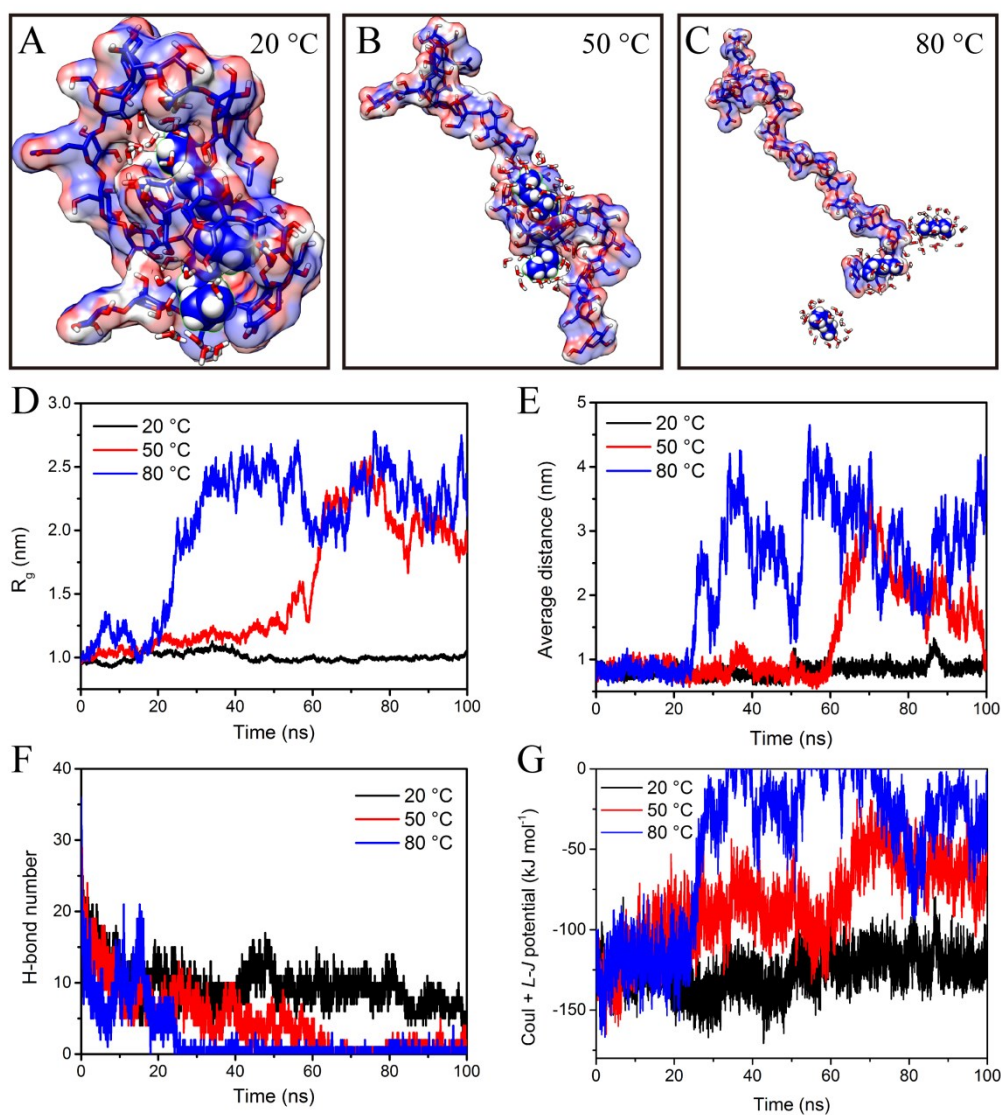


Fig. S8 Thermal-responsive helix-unfolding and guest-release characteristics in AAHHC. (A) The R_g value of AAHHC at different temperature. (B) The average distance between hexane molecules and the center atom of acetylated amylose at different temperature (inserted data is the maximum and minimum average distance at different temperature). (C) The H-bond number of AAHHC at different temperature. (D) The binding energy between hexane molecules and acetylated amylose.

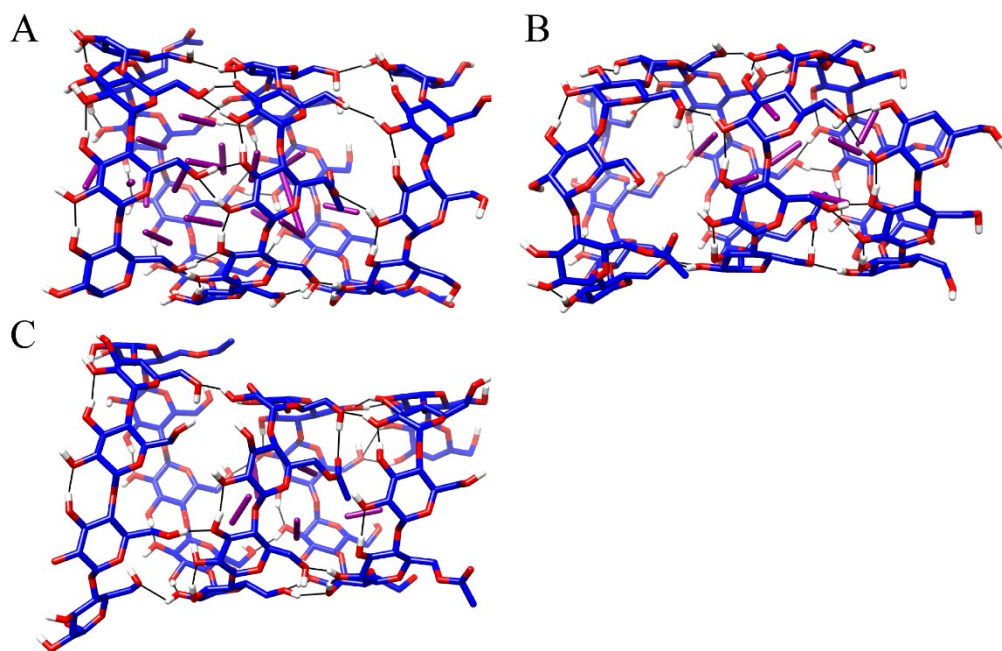


Fig. S9 Molecular structures of AAIHC with different iodine loaded. (A) 14; (B) 7; (C) 5.

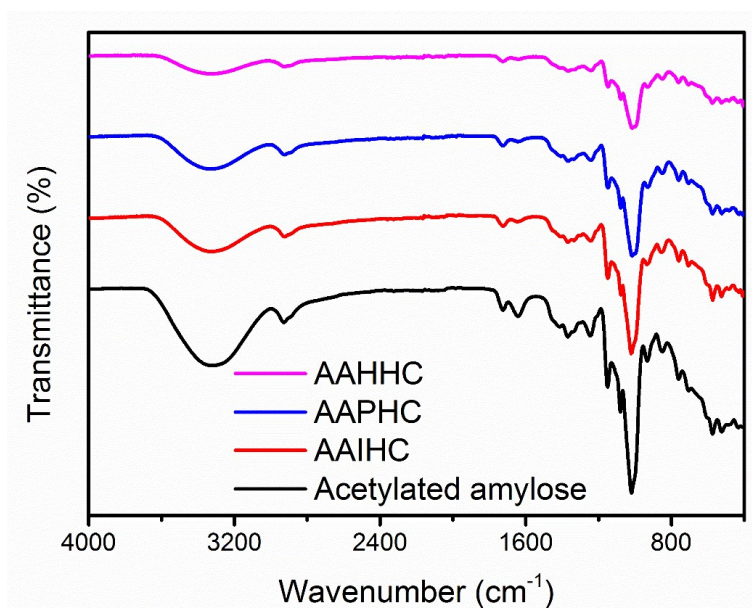


Fig. S10 The FT-IR spectra of AAGHCs.

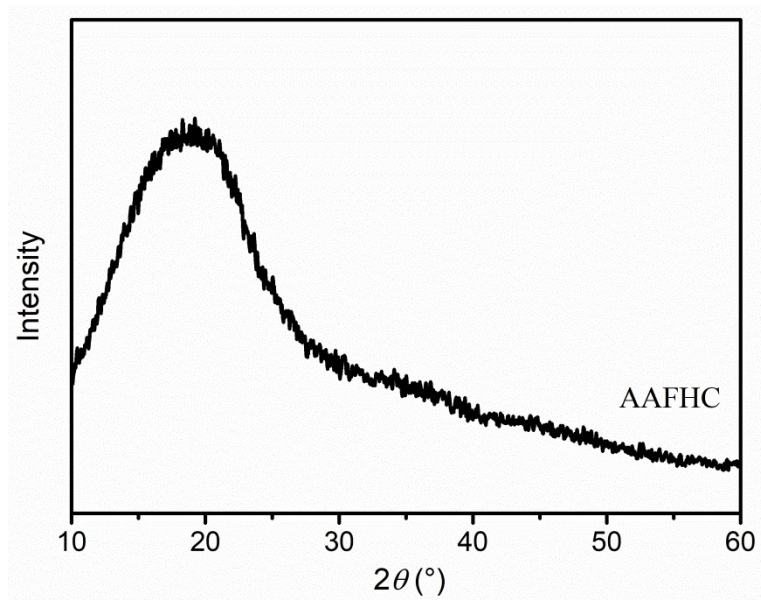


Fig. S11 The XRD pattern of AAFHC.

3. Appendix

1. The pdb and itp file of iodine

Pdb file:

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ATOM      2  I2  IODI      1      -2.194   0.729   0.000   1.00   0.00   I
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Itp file:

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[ atoms ]
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2 I 1 IODI I2 2 0.000 126.9040
```

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[ bonds ]
```

```
; ai aj fu b0 kb, b0 kb
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```
[ exclusions ]
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1 2
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2. The pdb and itp file of pro

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22-May-19
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AUTHOR 2 http://compbio.biosci.uq.edu.au/atb
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HETATM 2 C9 Pro 0 2.566 -0.525 0.237 1.00 0.00
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C
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C
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C
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C
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O
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[ atoms ]

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4	CH3	1	Pro	C9	1	0.000	15.0350	; 0.000
5	C	1	Pro	C3	2	-0.140	12.0110	
6	HC	1	Pro	H3	2	0.140	1.0080	; -0.000
7	C	1	Pro	C4	3	-0.140	12.0110	
8	HC	1	Pro	H4	3	0.140	1.0080	; -0.000
9	C	1	Pro	C5	4	-0.140	12.0110	
10	HC	1	Pro	H5	4	0.140	1.0080	; 0.000
11	C	1	Pro	C6	5	0.000	12.0110	
12	CH1	1	Pro	C10	5	0.00	13.0190	
13	CH3	1	Pro	C11	5	0.000	15.0350	; 0.000
14	C	1	Pro	C1	6	0.153	12.0110	
15	OA	1	Pro	O1	6	-0.511	15.9994	
16	H	1	Pro	H13	6	0.358	1.0080	; 0.000
17	CH3	1	Pro	C12	7	0.000	15.0350	; 0.000

; total charge of the molecule: 0.000

[bonds]

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5	7	2	0.1390	8.6600e+06
7	8	2	0.1090	1.2300e+07
7	9	2	0.1390	8.6600e+06
9	10	2	0.1090	1.2300e+07
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14	15	2	0.1360	1.0200e+07
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[pairs]

; ai aj funct ; all 1-4 pairs but the ones excluded in GROMOS itp

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14	17	1

[angles]

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	5	3	14	2	120.00	560.00
	3	5	6	2	120.00	505.00
	3	5	7	2	120.00	560.00
	6	5	7	2	120.00	505.00
	5	7	8	2	120.00	505.00
	5	7	9	2	120.00	560.00
	8	7	9	2	120.00	505.00
	7	9	10	2	120.00	505.00
	7	9	11	2	120.00	560.00
	10	9	11	2	120.00	505.00
	9	11	12	2	120.00	560.00
	9	11	14	2	120.00	560.00
	12	11	14	2	120.00	560.00
	11	12	13	2	111.00	530.00
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	13	12	17	2	111.00	530.00
	3	14	11	2	126.00	640.00
	3	14	15	2	121.00	685.00
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[dihedrals]

; GROMOS improper dihedrals

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	9	11	14	3	2	0.00	167.36
	11	14	3	5	2	0.00	167.36
	14	3	5	7	2	0.00	167.36
	3	2	5	14	2	0.00	167.36
	14	3	11	15	2	0.00	167.36

5	3	6	7	2	0.00	167.36
7	5	8	9	2	0.00	167.36
9	7	10	11	2	0.00	167.36
11	9	12	14	2	0.00	167.36
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12	11	17	13	2	35.26	334.72

[dihedrals]

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	6	8			
	6	9			
	6	14			
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3. The pdb and itp file of hexane

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CONECT	6	5						

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; Name nrexcl

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20	HC	1	Hex	H12	20	0.069	1.0080

; total charge of the molecule: 0.000

[bonds]

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5	6	2	0.1090	1.2300e+07
5	7	2	0.1090	1.2300e+07
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8	9	2	0.1100	1.2100e+07
8	10	2	0.1100	1.2100e+07
8	11	2	0.1530	7.1500e+06
11	12	2	0.1100	1.2100e+07
11	13	2	0.1100	1.2100e+07
11	14	2	0.1530	7.1500e+06
14	15	2	0.1090	1.2300e+07
14	16	2	0.1090	1.2300e+07
14	17	2	0.1530	7.1500e+06
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17	19	2	0.1090	1.2300e+07
17	20	2	0.1090	1.2300e+07

[pairs]

; ai aj funct ; all 1-4 pairs but the ones excluded in GROMOS itp

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16	19	1
16	20	1
15	18	1
15	19	1
15	20	1

[angles]

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	4	2	5	2	111.40	532.00
	2	5	6	2	109.50	448.00
	2	5	7	2	109.50	448.00
	2	5	8	2	111.00	530.00
	6	5	7	2	106.75	503.00
	6	5	8	2	109.50	448.00
	7	5	8	2	109.50	448.00
	5	8	9	2	109.50	448.00
	5	8	10	2	109.50	448.00
	5	8	11	2	111.00	530.00
	9	8	10	2	106.75	503.00
	9	8	11	2	109.50	448.00
	10	8	11	2	109.50	448.00
	8	11	12	2	109.50	448.00
	8	11	13	2	109.50	448.00

8	11	14	2	111.00	530.00
12	11	13	2	106.75	503.00
12	11	14	2	109.50	448.00
13	11	14	2	109.50	448.00
11	14	15	2	109.50	448.00
11	14	16	2	109.50	448.00
11	14	17	2	111.00	530.00
15	14	16	2	106.75	503.00
15	14	17	2	109.50	448.00
16	14	17	2	109.50	448.00
14	17	18	2	111.40	532.00
14	17	19	2	111.40	532.00
14	17	20	2	111.40	532.00
18	17	19	2	107.57	484.00
18	17	20	2	107.57	484.00
19	17	20	2	107.57	484.00

[dihedrals]

; GROMOS improper dihedrals

; ai	aj	ak	al	funct	angle	fc
------	----	----	----	-------	-------	----

[dihedrals]

; ai	aj	ak	al	funct	ph0	cp	mult
1	2	5	8	1	0.00	5.92	3
2	5	8	11	1	0.00	5.92	3
5	8	11	14	1	0.00	5.92	3
8	11	14	17	1	0.00	5.92	3
11	14	17	20	1	0.00	5.92	3

[exclusions]

; ai aj funct ; GROMOS 1-4 exclusions

4. The mdp file for molecular dynamic simulation (take the protocol guest is iodine and temperature is 50 °C as example)

```

title                = Yo
define               =
integrator           = md
dt                   = 0.002      ;2 * 50000000 = 100000000 ps (100 ns)
nsteps               = 50000000
nstxout              = 0           ; suppress .trr output
nstvout              = 0           ; suppress .trr output
nstenergy            = 5000        ; save energies every 2 ps
nstlog               = 5000        ; update log file every 2 ps
nstxtcout            = 5000        ; write .xtc trajectory every 2 ps
energygrps           = GLCA_GLCY IODI SOL
; Bond parameters
continuation         = yes         ; first dynamics run

```

```

constraint_algorithm = lincs      ; holonomic constraints
constraints          = all-bonds  ; all bonds (even heavy atom-H bonds) constrained
nstlist              = 10
cutoff-scheme        = Verlet
ns-type              = grid
rlist                = 0.9
rlistlong            = 1.4
nstcalcr             = 1
coulombtype          = Reaction-Field
rcoulomb              = 0.9
epsilon-rf            = 2.2
fourierspacing       = 0.16
vdw-type              = Cut-off
rvdw                 = 0.9
tcoupl               = V-rescale
tc-grps              = GLCA_GLCY IODI SOL
tau-t                = 0.1  0.1  0.1
ref-t                = 353  353  353
Pcoupl               = Parrinello-Rahman
pcoupltype           = isotropic
tau-p                = 1.0
ref_p                = 1.0
compressibility       = 4.5e-7
pbc                  = xyz          ; 3-D PBC
; Dispersion correction
DispCorr              = EnerPres    ; account for cut-off vdW scheme
; Velocity generation
gen_vel               = no          ; assign velocities from Maxwell distribution

```

noted: GLCA_GLCY means acetylated amylose; IODI means iodine; SOL means water molecule.

References

S1 Y. Liu, W. Gao, C. Zhang, P. Tang, Y. Zhao and D. Wu, *Chemical Communications*, 2017, **53**, 10680-10683.