

Hydrophilic Polymer Driven Crystallization Self-assembly: Inflammation Multi-Drug Combination Nanosystem Against Alzheimer Disease

Haodong Hu^a, Jinna Wang^a, Jian Ren^a, Xinpo Li^a, Bo Zhang^a, Zhengang Lv^{b*} and Fengying Dai^{a*}

^a State Key Laboratory of Separation Membranes and Membrane Processes/National Center for International Joint Research on Separation Membranes, School of Material Science and Engineering, Tiangong University, Tianjin 300387, China

^b State Key Laboratory of Coal Conversion, Institute of Coal Chemistry, Chinese Academy of Sciences and Synfuels China Co., Ltd., Beijing, P. R. China

*Corresponding author: Fengying Dai (email: daifengying@tjpu.edu.cn), Zhengang Lv (email: lzhg179@163.com)

SUPPORTING INFORMATION

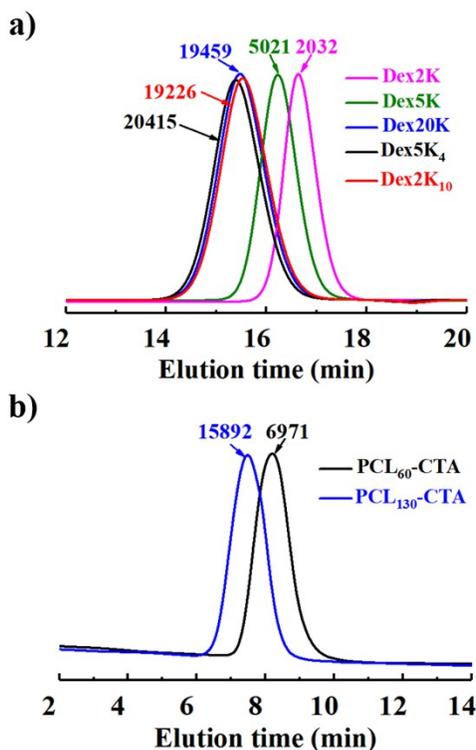


Fig. S1 GPC profiles of dextran and PCL-CTA polymers. (a) Elution time of Dex2K, Dex5K, Dex2K₁₀, Dex5K₄ and Dex20K with water as an eluent, and the flow rate was 1.0 mL/min. (b) Elution time of PCL₆₀-CTA and PCL₁₃₀-CTA with THF as an eluent, and the flow rate was 1.0 mL/min.

Table S1 Structural parameters of polymers.

sample	$m_{\text{PCL}},^{\text{a}}$	$p_{\text{PPBA}},^{\text{b}}$	$n_{\text{Dex}},^{\text{c}}$	M_n (g/mol), ^d	M_n (g/mol), ^e	$\text{PDI} (M_w/M_n),^{\text{f}}$
	NMR	NMR	NMR	NMR	GPC	GPC
Dex2K	/	/	1	2000	2032	1.21
Dex5K	/	/	1	5000	5021	1.23
Dex20K	/	/	1	20000	19459	1.31
Dex5K ₄	/	/	4	20000	20415	1.26
Dex2K ₁₀	/	/	10	20000	19226	1.28
PCL ₆₀ -CTA	60	/	/	7067	6971	1.35
PCL ₁₃₀ -CTA	130	/	/	15047	15892	1.32
PCL ₆₀ - <i>b</i> -PPBA ₁₀	60	10	/	8977	/	/
PCL ₁₃₀ - <i>b</i> -PPBA ₁₀	130	10	/	16957	/	/
PCL ₁₃₀ - <i>b</i> -PPBA ₁₂₀	130	120	/	37967	/	/
PCL ₁₃₀ - <i>g</i> -Dex20K	130	/	1	35047	/	/
PCL ₁₃₀ - <i>g</i> -Dex5K ₄	130	/	4	35047	/	/
PCL ₁₃₀ - <i>g</i> -Dex2K ₁₀	130	/	10	35047	/	/
PCL ₆₀ - <i>b</i> -PPBA ₁₀ -Dex20K	60	10	1	28977	/	/
PCL ₁₃₀ - <i>b</i> -PPBA ₁₀ -Dex20K	130	10	1	36957	/	/
PCL ₁₃₀ - <i>b</i> -PPBA ₁₀ -Dex5K ₄	130	10	4	36957	/	/
PCL ₁₃₀ - <i>b</i> -PPBA ₁₀ -Dex2K ₁₀	130	10	10	36957	/	/

^a m : DP of PCL as determined by ¹H-NMR measurement. ^b p : DP of PPBA as determined by ¹H-NMR measurement. ^c n : DP of dextran brush as determined by ¹H-NMR measurement. ^d M_n : Number-average molecular weight as calculated from DPs determined by ¹H-NMR measurement. ^e M_n : Number-average molecular weight as determined by GPC. ^f PDI: The ratio of weight-average molecular weight to number-average molecular weight as determined by GPC.

b)

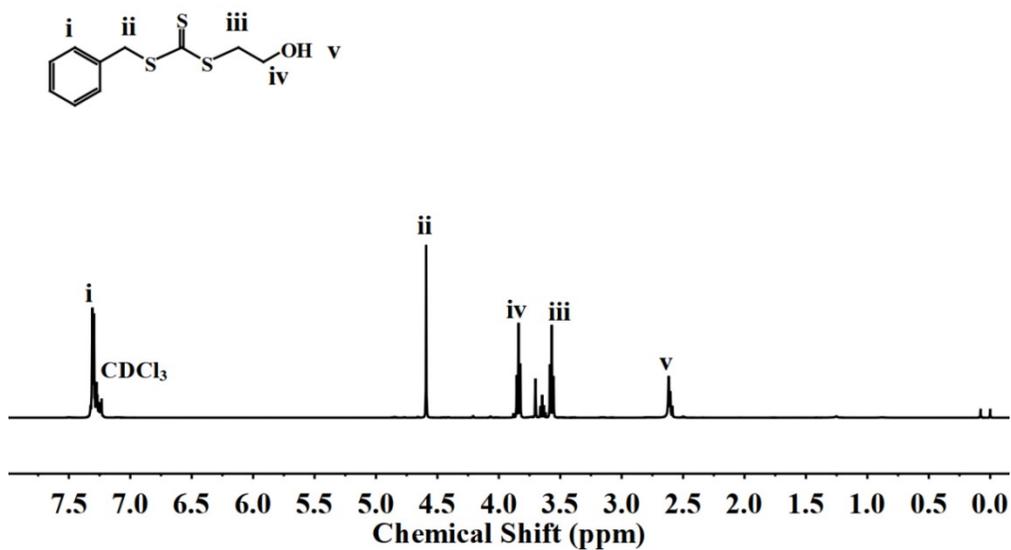


Fig. S2 Synthesis and characterization of polymers. (a) Synthesis Route of BSTSE and (b) ¹H-NMR spectra of BSTSE in CDCl₃.

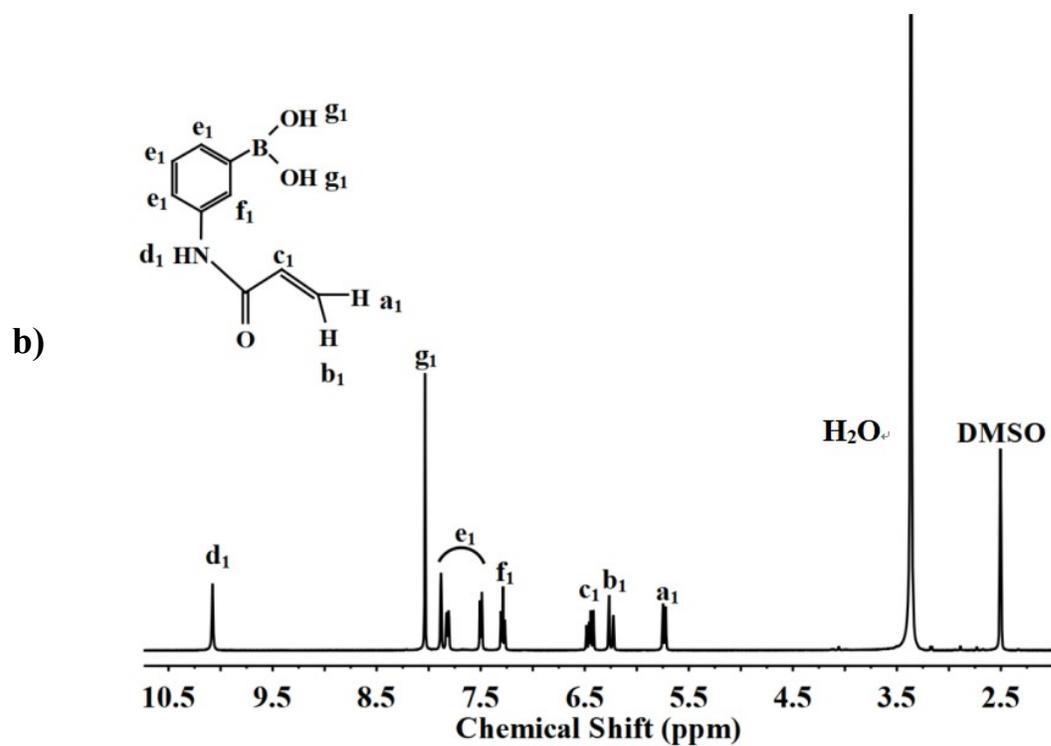


Fig. S3 Synthesis and characterization of polymers. (a) Synthesis Route of PBA and (b) $^1\text{H-NMR}$ spectra of PBA in $\text{DMSO-}d_6$.

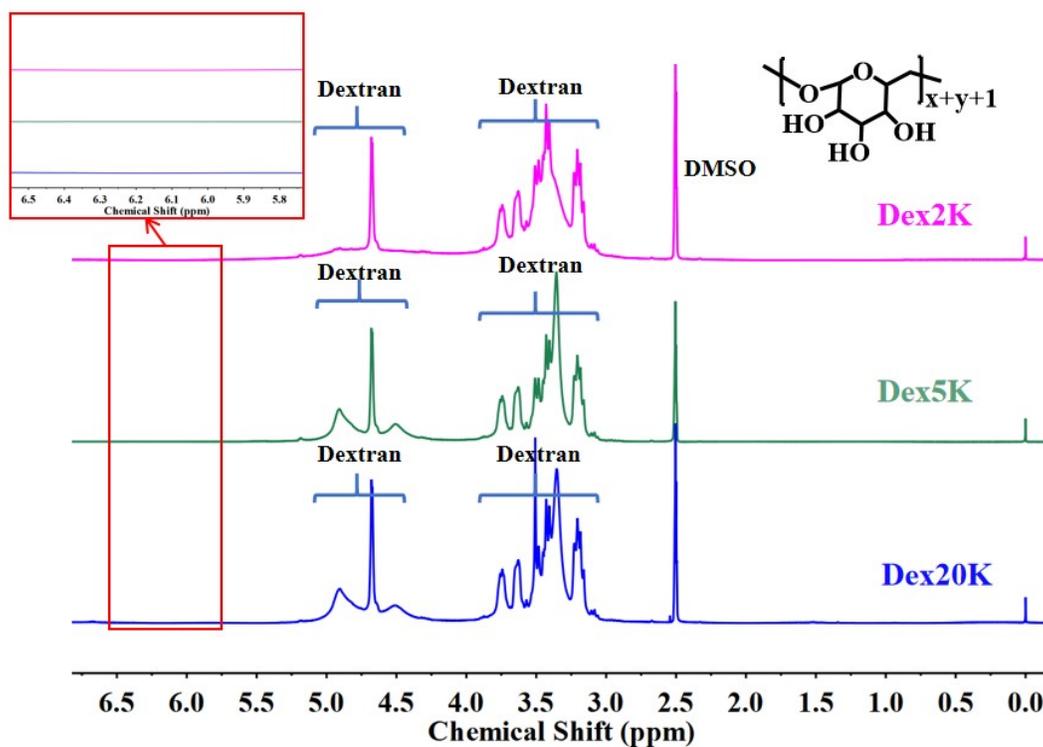


Fig. S4 $^1\text{H-NMR}$ spectra of Dex2K, Dex5K and Dex20K in $\text{DMSO-}d_6$.

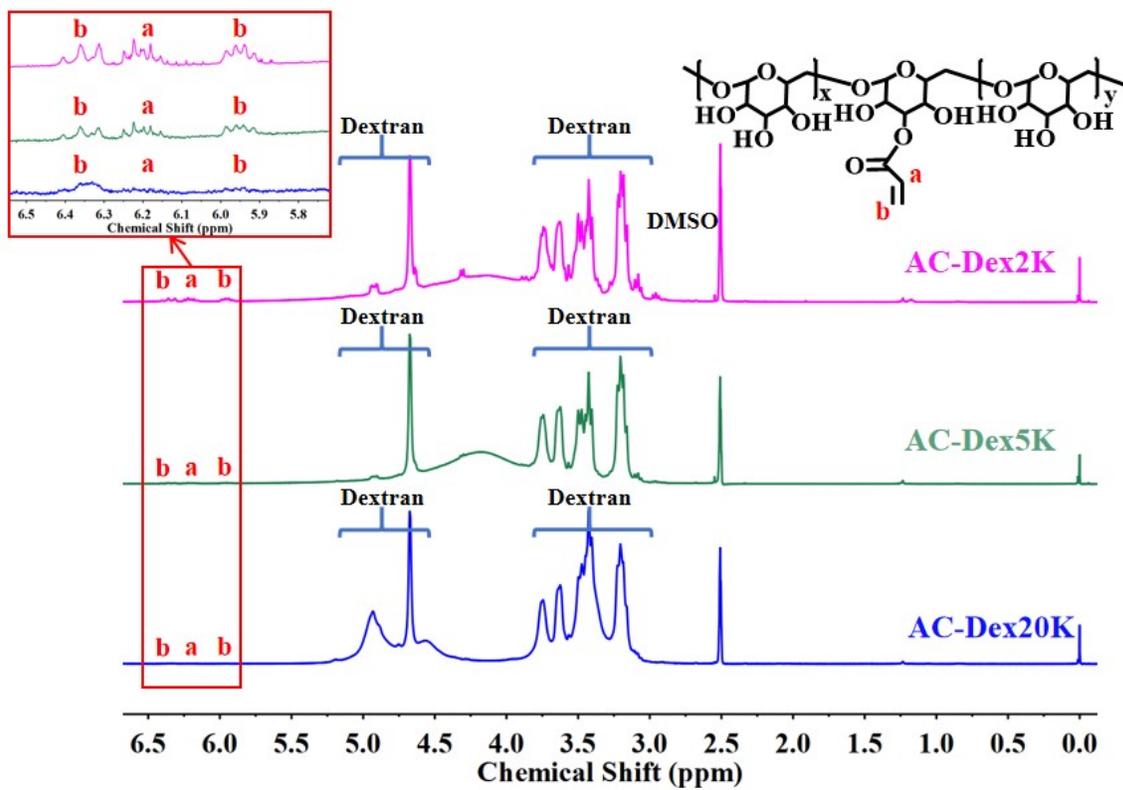


Fig. S5 $^1\text{H-NMR}$ spectra of AC-Dex2K, AC-Dex5K and AC-Dex20K in $\text{DMSO-}d_6$.

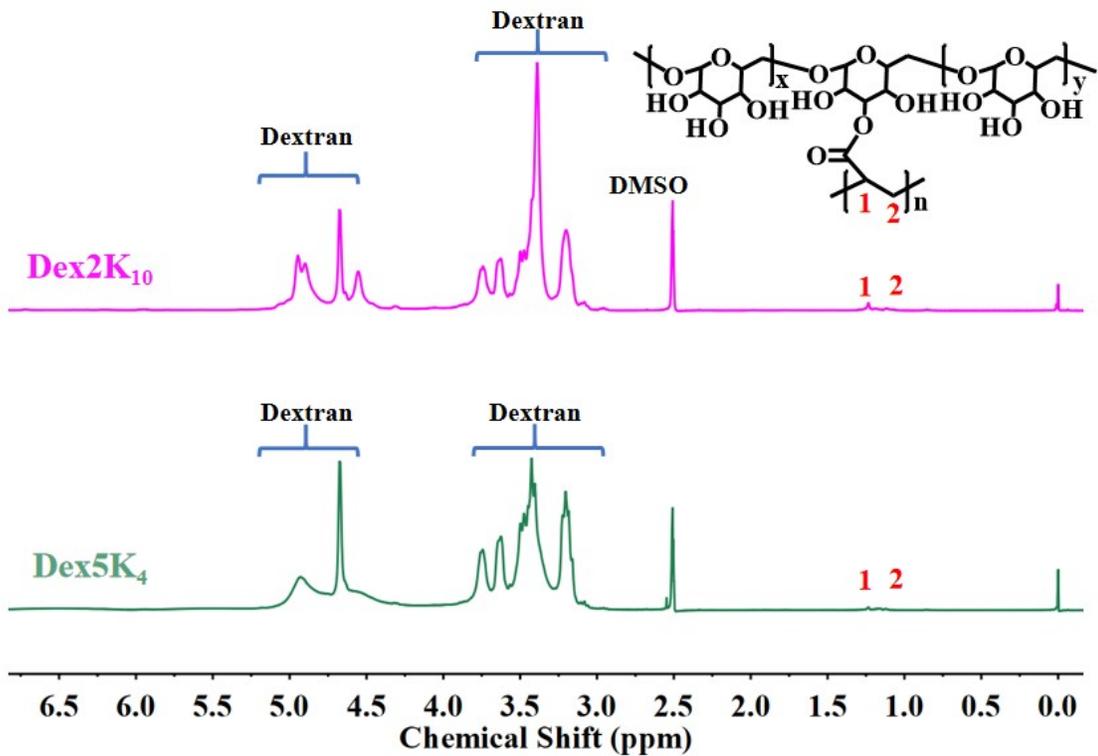


Fig. S6 ¹H-NMR spectra of Dex2K₁₀ and Dex5K₄ in DMSO-*d*₆.

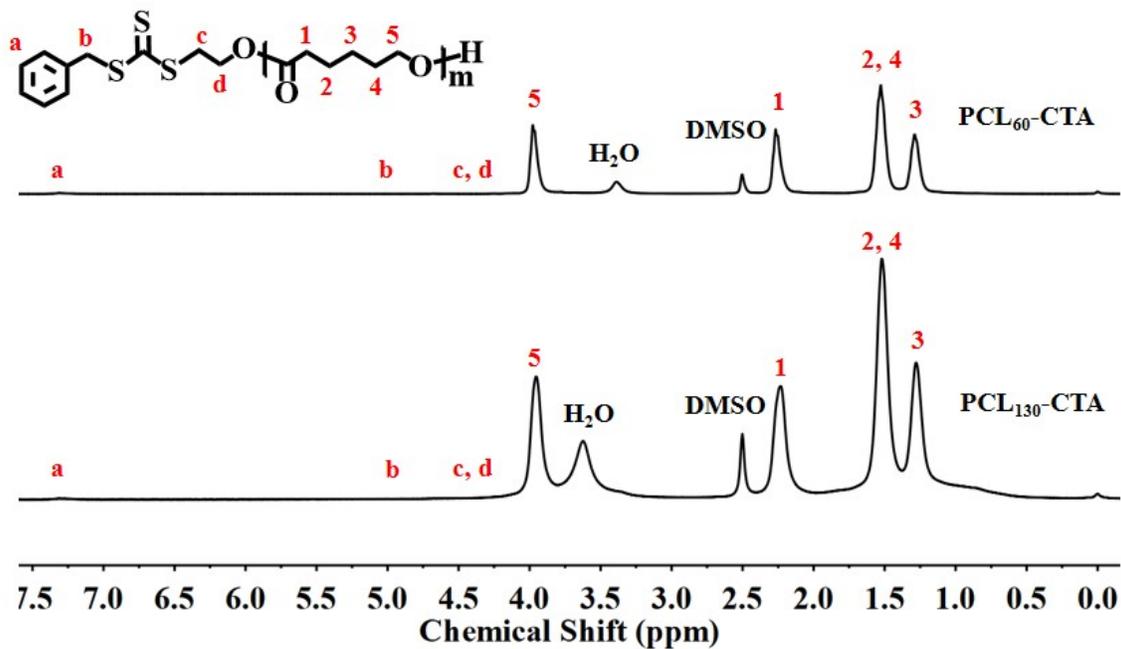


Fig. S7 ¹H-NMR spectra of PCL₆₀-CTA and PCL₁₃₀-CTA in DMSO-*d*₆.

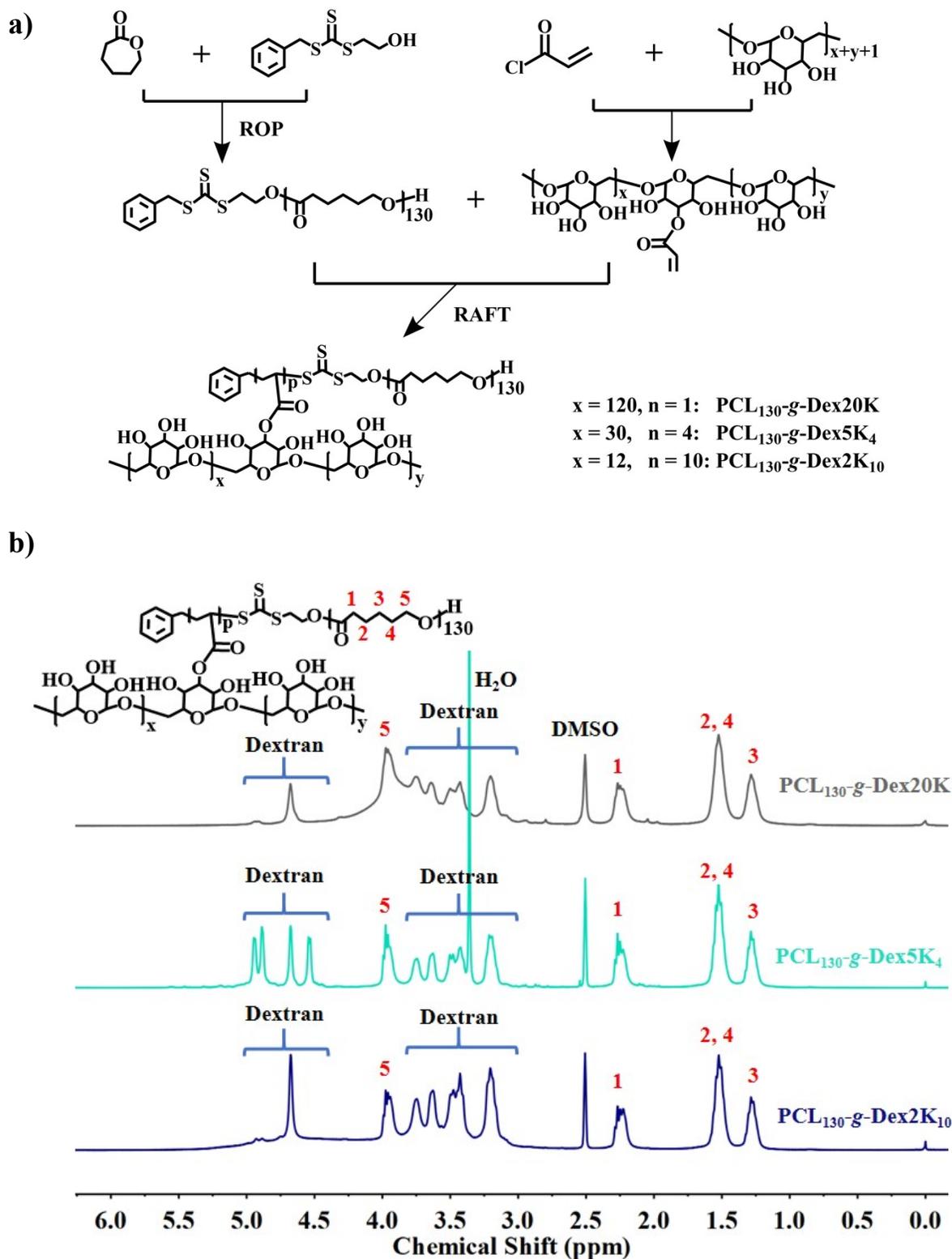


Fig. S9 Synthesis and characterization of polymers. (a) Synthesis routes and (b) ¹H-NMR spectra of PCL₁₃₀-g-Dex20K, PCL₁₃₀-g-Dex5K₄ and PCL₁₃₀-g-Dex2K₁₀ in DMSO-*d*₆.

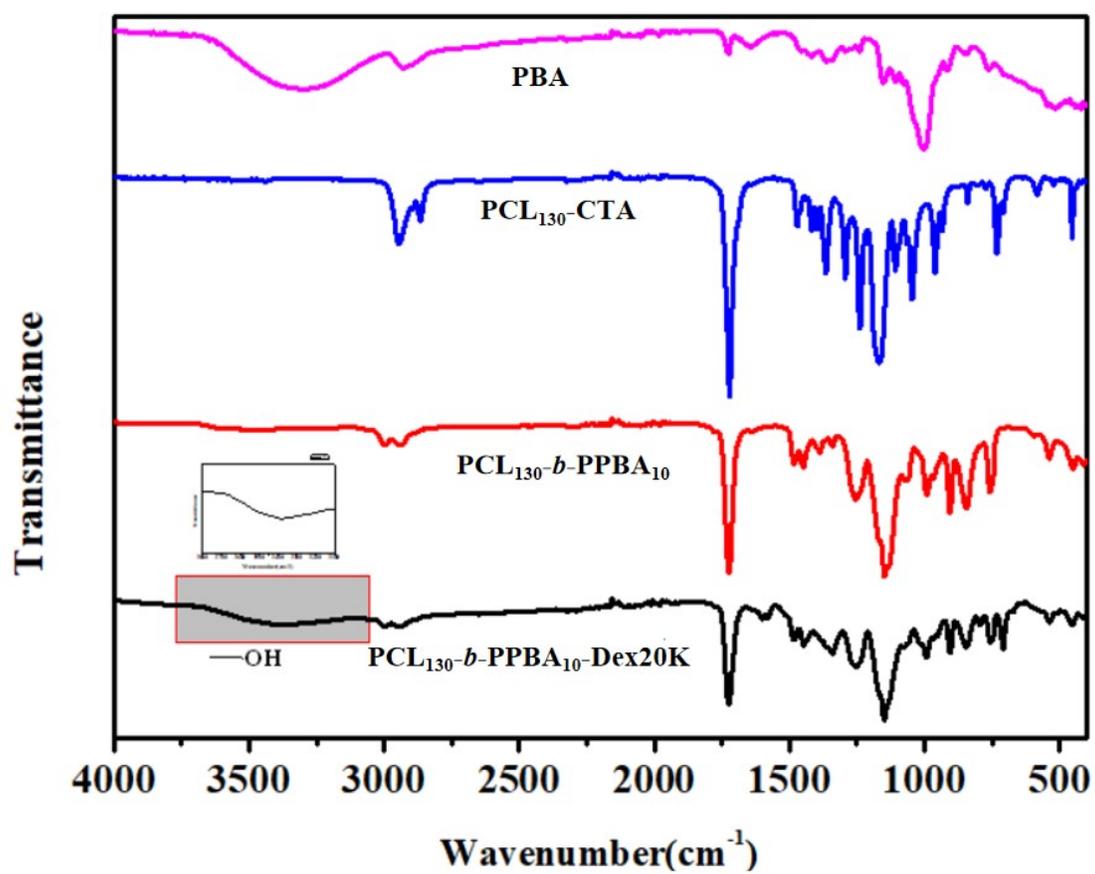


Fig. S10 FT-IR spectra of PBA, PCL₁₃₀-CTA, PCL₁₃₀-*b*-PPBA₁₀ and PCL₁₃₀-*b*-PPBA₁₀-Dex20K.

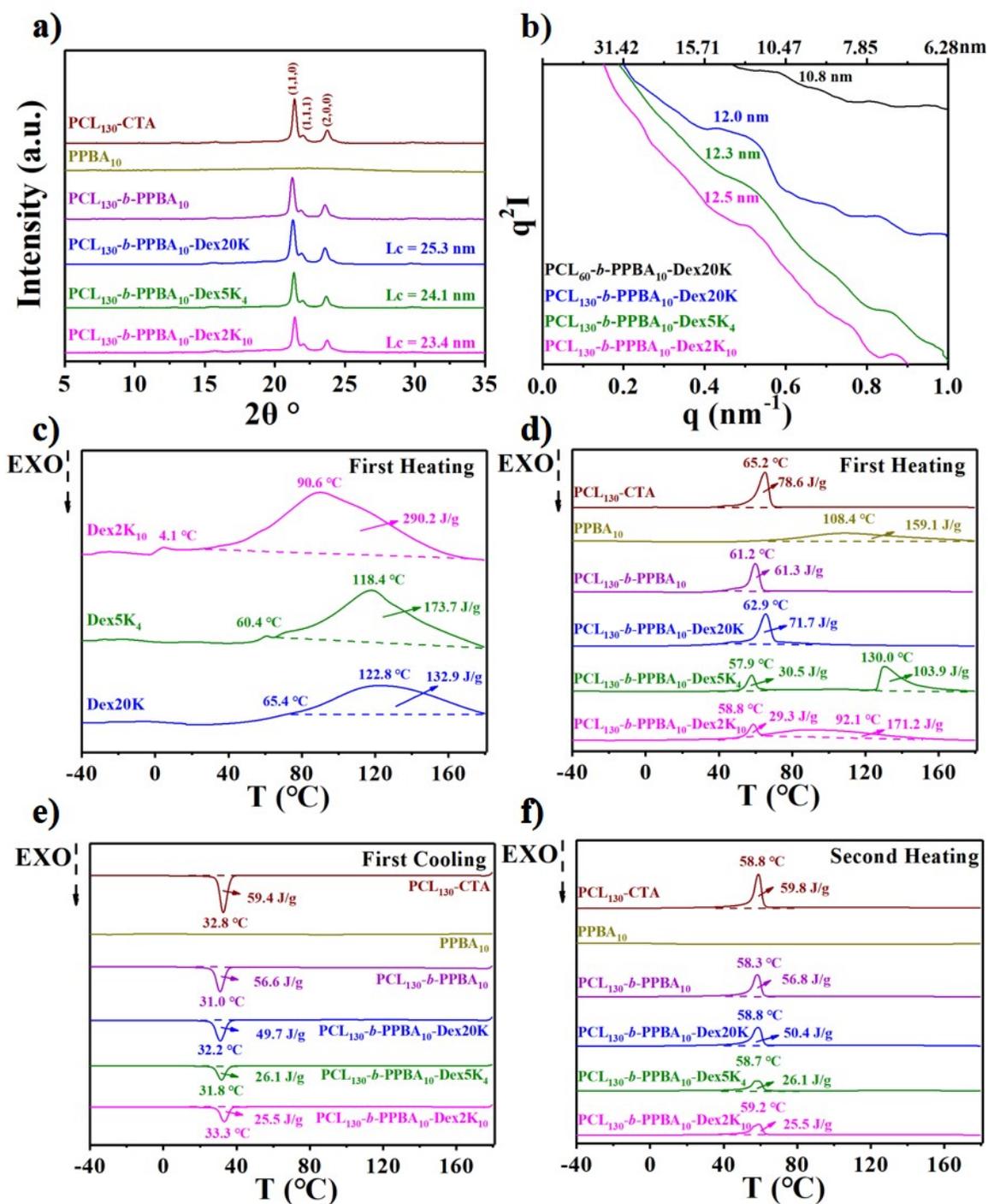


Fig. S11 Crystallization behavior and thermodynamic characterization of copolymers and nanoparticles. (a) WAXD patterns of PCL_{130} -CTA, $PPBA_{10}$, PCL_{130} -*b*- $PPBA_{10}$, PCL_{130} -*b*- $PPBA_{10}$ -Dex20K, PCL_{130} -*b*- $PPBA_{10}$ -Dex5K₄ and PCL_{130} -*b*- $PPBA_{10}$ -Dex2K₁₀. (b) SAXS patterns of PCL_{60} -*b*- $PPBA_{10}$ -Dex20K, PCL_{130} -*b*- $PPBA_{10}$ -Dex20K, PCL_{130} -*b*- $PPBA_{10}$ -Dex5K₄ and PCL_{130} -*b*- $PPBA_{10}$ -Dex2K₁₀. (c) DSC curves of Dex2K₁₀ and Dex5K₄. (d-f) DSC curves of PCL_{130} -CTA, $PPBA_{10}$, PCL_{130} -*b*- $PPBA_{10}$, PCL_{130} -*b*- $PPBA_{10}$ -Dex20K, PCL_{130} -*b*- $PPBA_{10}$ -Dex5K₄ and PCL_{130} -*b*- $PPBA_{10}$ -Dex2K₁₀.

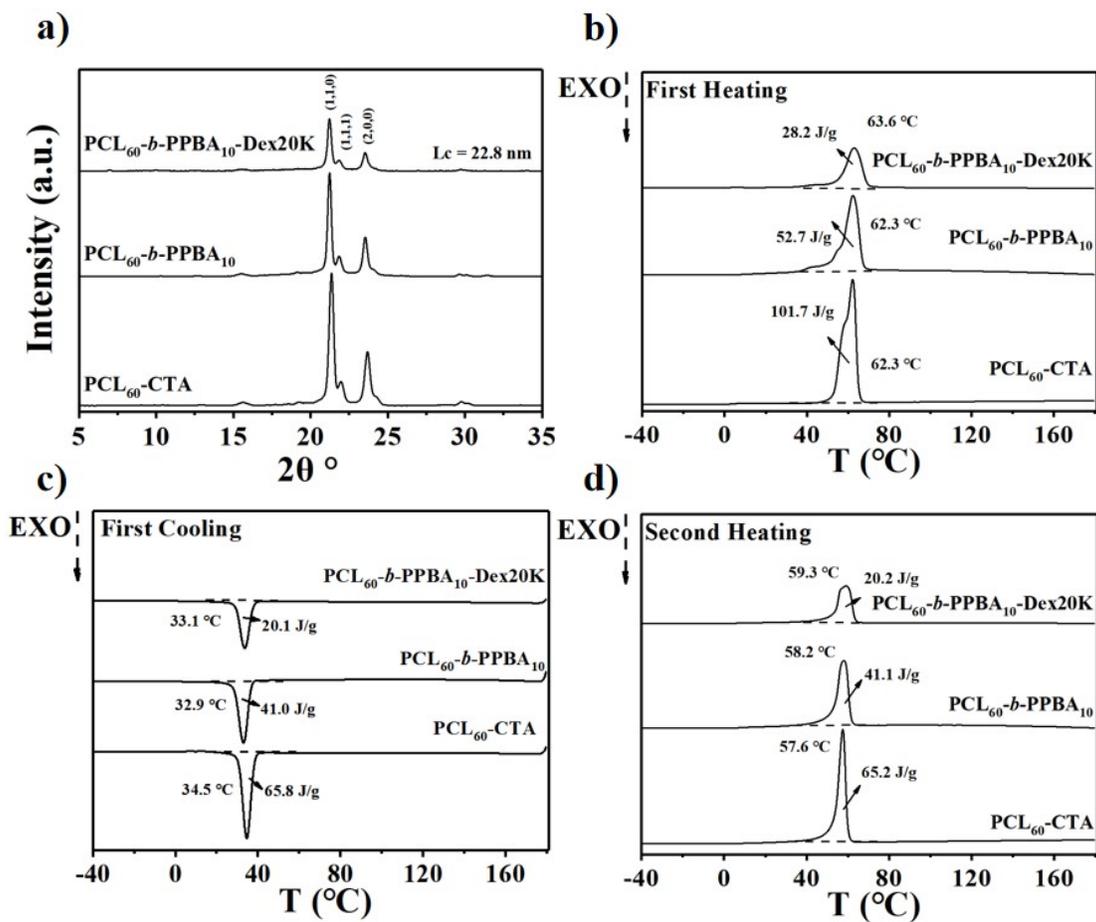


Fig. S12 Crystallization behavior and thermodynamic characterization of copolymers. (a) WAXD spectrum and (b-d) DSC curves of $\text{PCL}_{60}\text{-CTA}$, $\text{PCL}_{60}\text{-}b\text{-PPBA}_{10}$ and $\text{PCL}_{60}\text{-}b\text{-PPBA}_{10}\text{-Dex20K}$.

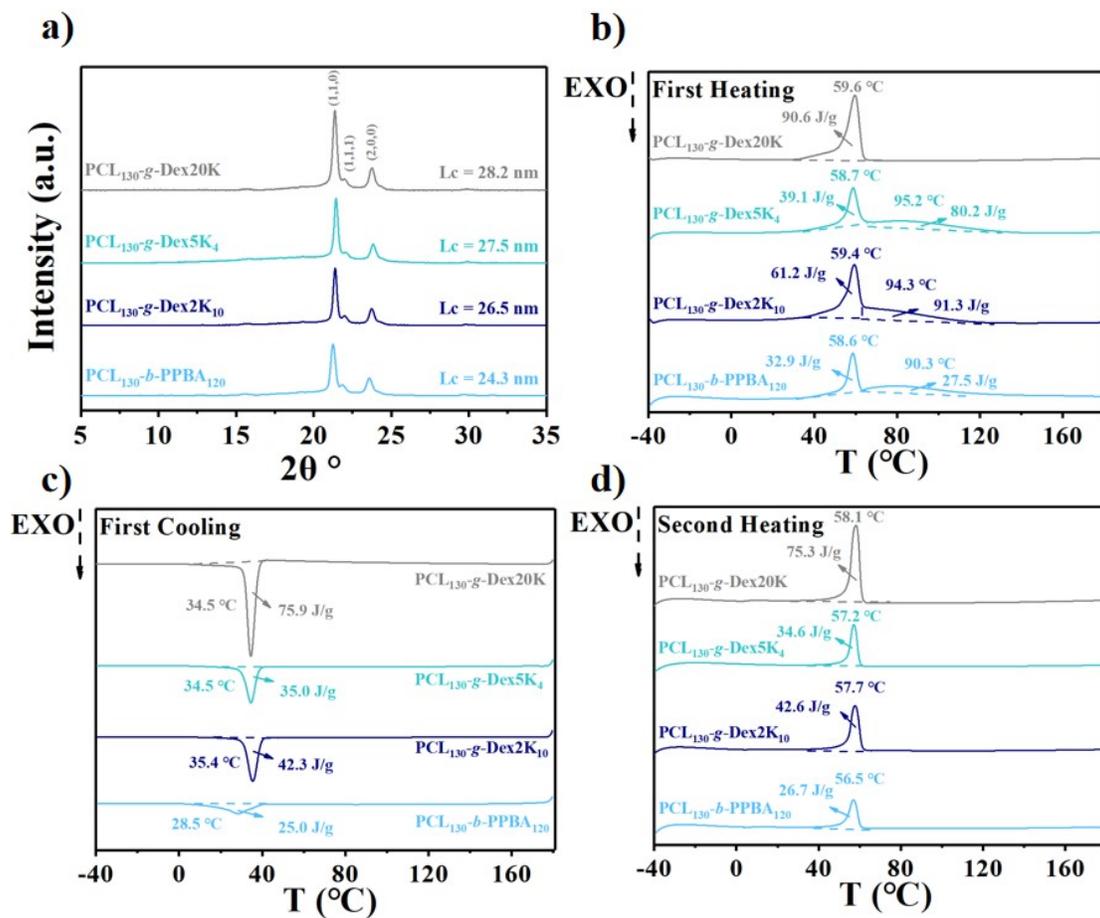


Fig. S13 Crystallization behavior and thermodynamic characterization of copolymers. (a) WAXD spectrum and (b-d) DSC curves of PCL₁₃₀-g-Dex20K, PCL₁₃₀-g-Dex5K₄, PCL₁₃₀-g-Dex2K₁₀, PCL₁₃₀-b-PPBA₁₂₀.

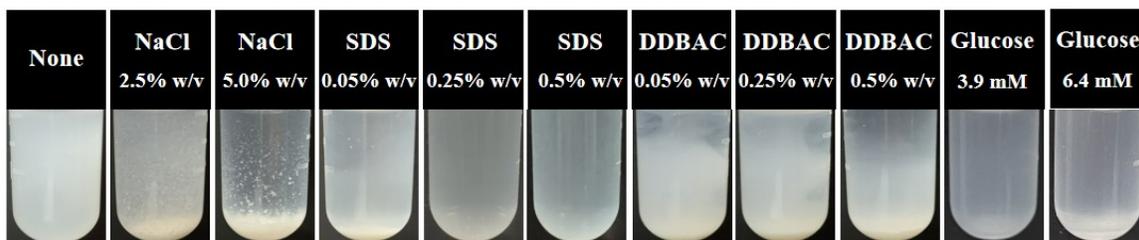


Fig. S14 Stability test of PCL₁₃₀-*b*-PPBA₁₂₀ vesicle dispersions. Photographs before and after NaCl (the final concentration: 2.5% w/v, 5% w/v), SDS (the final concentration: 0.05% w/v, 0.25% w/v, 0.5% w/v), DDBAC (the final concentration: 0.05% w/v, 0.25% w/v, 0.5% w/v) and glucose (the final concentration: 3.9 mM, 6.4 mM) treatments (the final concentration of PCL₁₃₀-*b*-PPBA₁₂₀ vesicle is 0.5% w/v).

Table S2 Summary of PCL₁₃₀-*b*-PPBA₁₀-Dex20K micelle's stability test.

type of added surfactant or glucose	concentration (w/v %)	D _h ^a (nm)	PDI ^a	T ^b (%)
none (i.e., pristine PCL ₁₃₀ - <i>b</i> -PPBA ₁₀ -Dex20K micelle)	0.0	132.4	0.053	12.2
NaCl	2.5	133.6	0.062	12.5
	5.0	132.5	0.065	12.3
	0.05	133.7	0.081	12.7
SDS	0.25	134.6	0.076	11.9
	0.5	137.2	0.091	12.6
	0.05	136.5	0.065	12.4
DDBAC	0.25	138.2	0.063	12.7
	0.5	137.3	0.057	12.5
	0.07	136.5	0.056	12.3
glucose	0.12	137.4	0.058	12.6

a: Hydrodynamic diameter (D_h) and polydispersity (PDI) measured by DLS. b: Transmittance measured at 600 nm by UV-vis spectroscopy.

Table S3 Summary of PCL₁₃₀-*b*-PPBA₁₀-Dex5K₄ vesicle's stability test.

type of added surfactant or glucose	concentration (w/v %)	D _h ^a (nm)	PDI ^a	T ^b (%)
none (i.e., pristine PCL ₁₃₀ - <i>b</i> - PPBA ₁₀ -Dex5K ₄ vesicle)	0.0	173.8	0.062	13.1
NaCl	2.5	174.7	0.064	13.4
	5.0	173.5	0.086	13.6
	0.05	176.2	0.053	12.9
SDS	0.25	177.5	0.066	13.5
	0.5	176.8	0.062	13.2
	0.05	179.2	0.095	13.7
DDBAC	0.25	175.3	0.062	13.5
	0.5	169.5	0.057	12.9
glucose	0.07	173.2	0.057	13.2
	0.12	174.6	0.064	13.1

a: Hydrodynamic diameter (D_h) and polydispersity (PDI) measured by DLS. b: Transmittance measured at 600 nm by UV–vis spectroscopy.

Table S4 Summary of PCL₁₃₀-*b*-PPBA₁₀-Dex2K₁₀ vesicle's stability test.

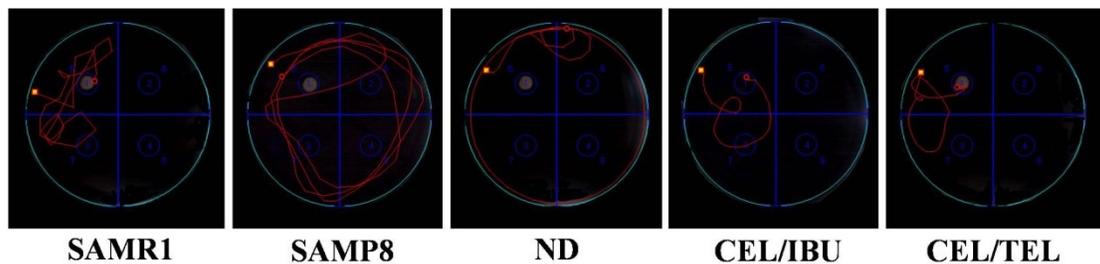
type of added surfactant or glucose	concentration (w/v %)	D _h ^a (nm)	PDI ^a	T ^b (%)
none (i.e., pristine PCL ₁₃₀ - <i>b</i> - PPBA ₁₀ -Dex2K ₁₀ vesicle)	0.0	156.8	0.071	12.6
NaCl	2.5	157.3	0.065	12.8
	5.0	158.3	0.061	13.2
	0.05	154.6	0.057	12.5
SDS	0.25	155.8	0.064	12.6
	0.5	157.3	0.092	12.8
	0.05	156.3	0.110	12.5
DDBAC	0.25	158.3	0.066	12.7
	0.5	157.6	0.085	13.2
glucose	0.07	157.3	0.065	12.3
	0.12	157.8	0.071	12.2

a: Hydrodynamic diameter (D_h) and polydispersity (PDI) measured by DLS. b: Transmittance measured at 600 nm by UV-vis spectroscopy.

Table S5 Summary of PCL₁₃₀-*b*-PPBA₁₂₀ vesicle's stability test.

type of added surfactant or glucose	concentration (w/v %)	D _h ^a (nm)	PDI ^a	T ^b (%)
none (i.e., pristine PCL ₁₃₀ - <i>b</i> -PPBA ₁₂₀ vesicle)	0.0	121.3	0.053	12.6
NaCl	2.5	814.3	0.252	35.2
	5.0	1151.2	0.731	82.3
	0.05	752.3	0.364	34.2
SDS	0.25	892.1	0.536	46.2
	0.5	1021.6	0.643	56.3
	0.05	101.3	0.151	10.2
DDBAC	0.25	75.3	0.216	8.3
	0.5	42.1	0.263	6.4
glucose	0.07	834.2	0.342	42.1
	0.12	1022.5	0.642	65.2

a: Hydrodynamic diameter (D_h) and polydispersity (PDI) measured by DLS. b: Transmittance measured at 600 nm by UV-vis spectroscopy.

**Fig. S15** The typical escape routes of each group mice in the morris water maze on the fifth day.

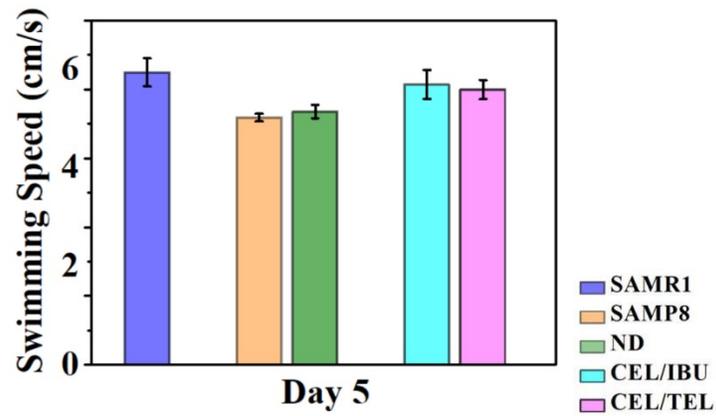


Fig. S16 The swimming speed of each group on the fifth day.

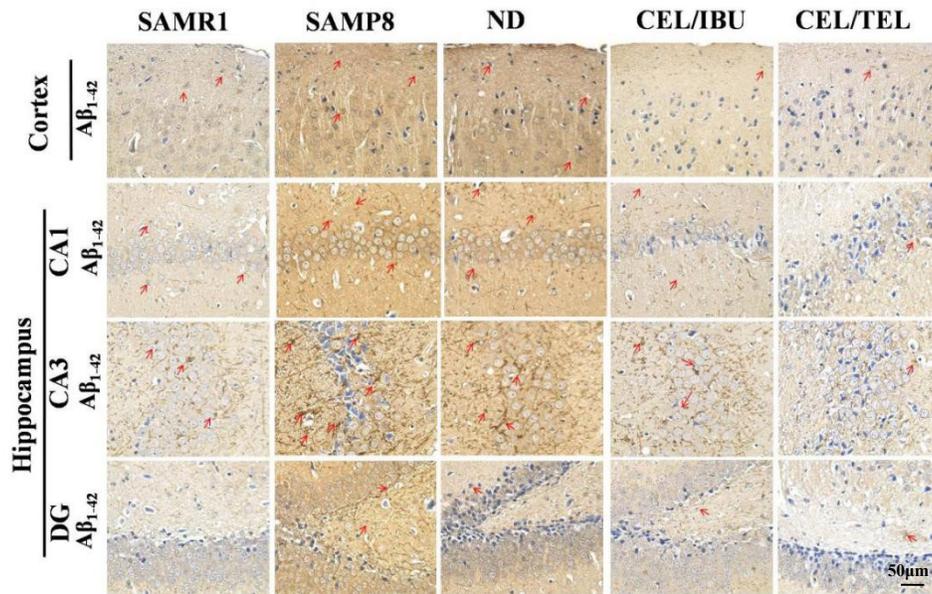


Fig. S17 The immunohistochemical analysis of $A\beta_{1-42}$ deposition in the hippocampus and cortex of mice. Scale bars = 50 μm .

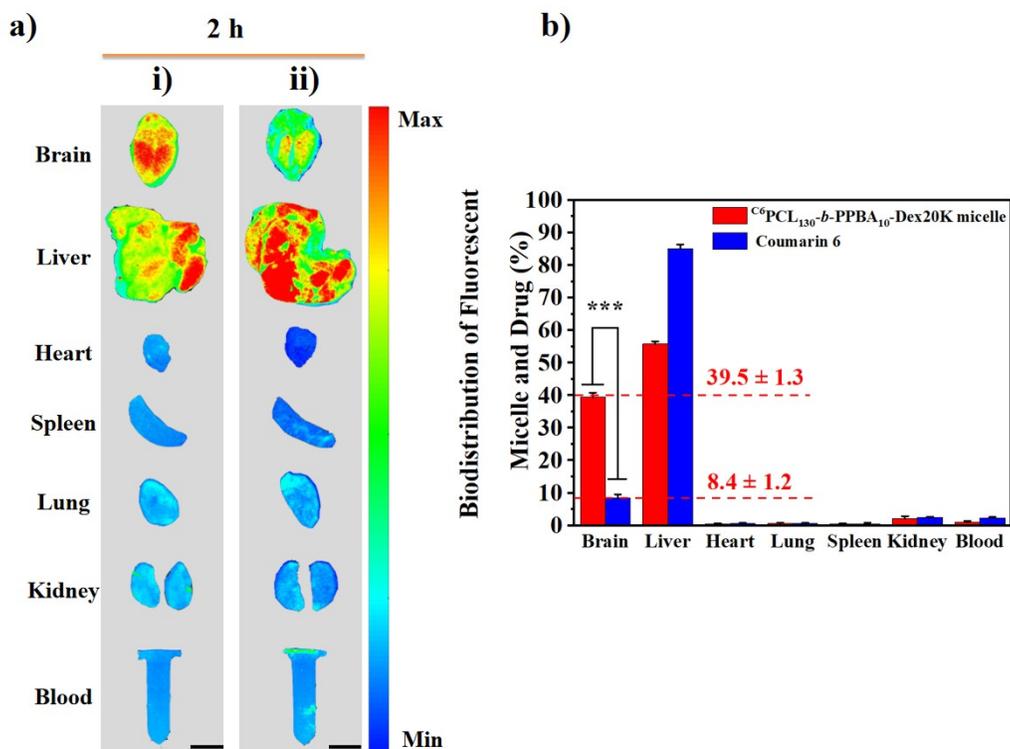


Fig. S18 The delivery efficiency and biodistribution data of micelle. (a) Ex vivo fluorescence tissue images of SAMP8 mice after treated with (i) ${}^{\text{C6}}\text{PCL}_{130}\text{-}b\text{-PPBA}_{10}\text{-Dex20K}$ micelle or (ii) coumarin 6 (C6). (b) The biodistribution of ${}^{\text{C6}}\text{PCL}_{130}\text{-}b\text{-PPBA}_{10}\text{-Dex20K}$ micelle and coumarin 6 2 hours after nasal administration. Scale bars = 5 mm.

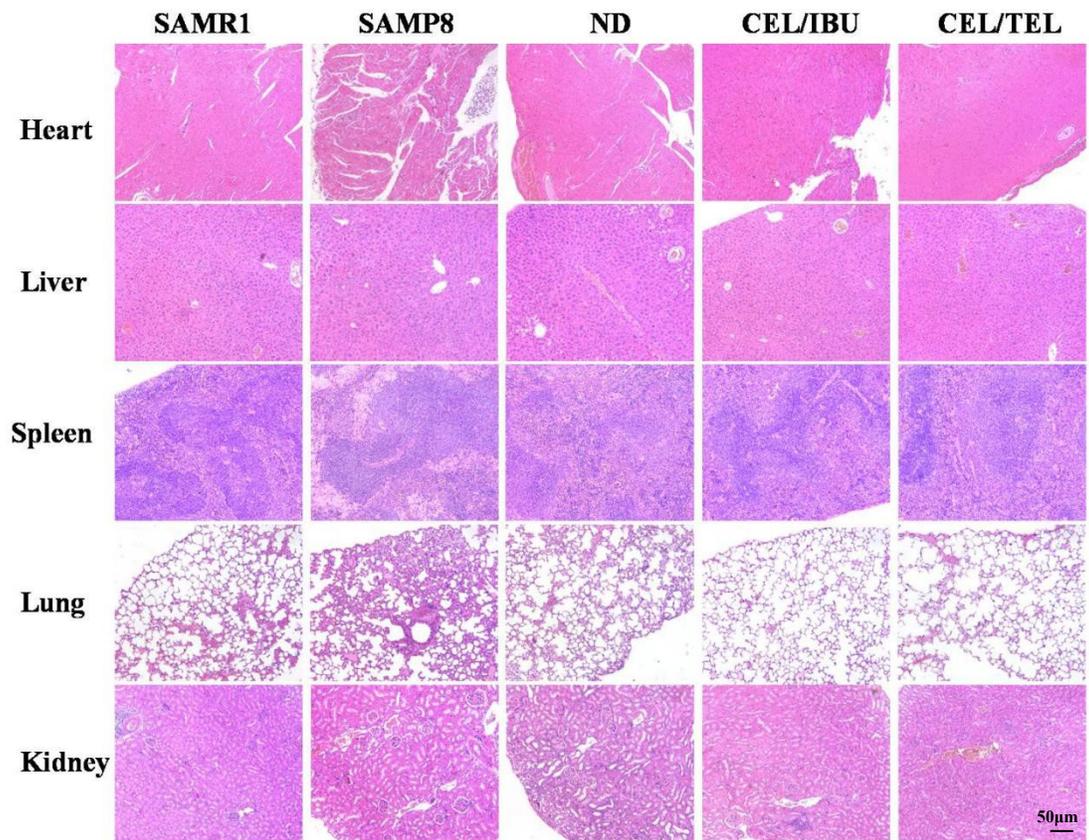


Fig. S19 H&E staining pictures of heart, liver, spleen and lung in each group of mice (100X). Scale bars = 50 µm.