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Electronic Supplementary Information for

Boronic Acid-Functionalized Spherical Polymer Brushes for Efficient and Selective Enrichment of Glycoproteins

Chen Hua ^a, Kaimin Chen *b and Xuhong Guo*a,c

^a State Key Laboratory of Chemical Engineering, East China University of Science and Technology, Shanghai 200237,

People's Republic of China

^b College of Chemistry and Chemical Engineering, Shanghai University of Engineering Science, Shanghai 201620,

People's Republic of China

^c Engineering Research Center of Materials Chemical Engineering of Xinjiang Bingtuan, Shihezi University, Xinjiang 832000, People's Republic of China

^{*} corresponding author e-mail: kmchen@sues.edu.cn (K. Chen); guoxuhong@ecust.edu.cn (X. Guo)

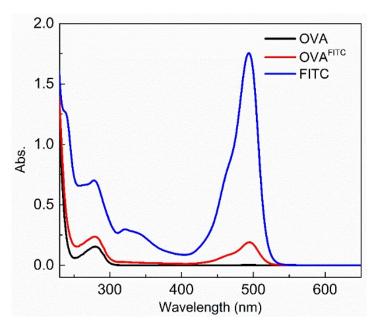


Fig. S1 UV-vis spectra of OVA, FITC and OVAFITC.

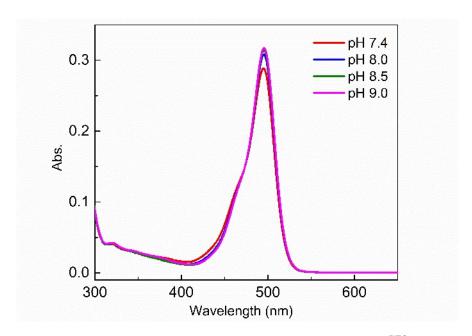


Fig. S2 The effect of pH values on the UV-vis absorbance of OVAFITC.

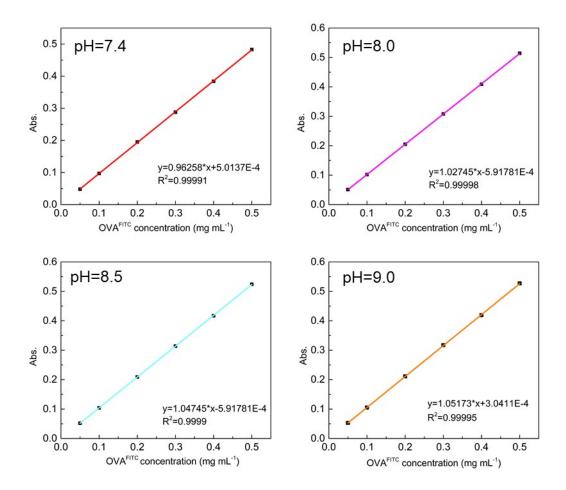
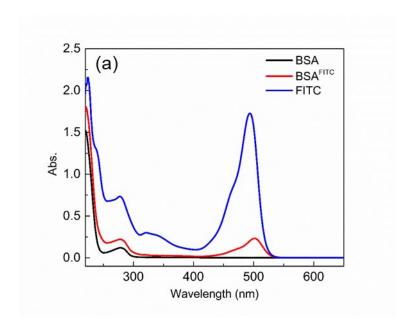


Fig. S3 Standard curves of OVAFITC at different pH values.



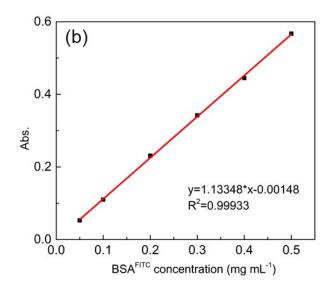


Fig. S4 UV-vis spectra of BSA, FITC and BSA^{FITC} (a); Standard curve of BSA^{FITC} at pH 7.4.

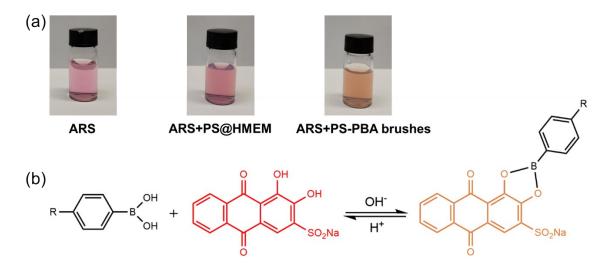


Fig. S5 (a) Optical photographs of the color reaction of boronic acid with ARS; (b) reaction equation of phenylboronic acid and ARS. Experimental condition: 0.05 mM ARS, 0.05 mg/mL PS@HMEM and PS-PBA brushes, pH 7.4, 20 mM PBS.

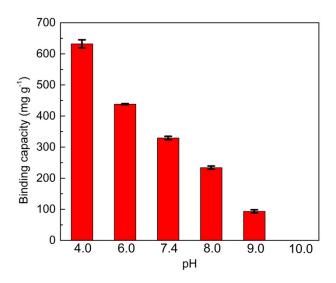


Fig. S6 Influence of pH value on the binding capacity of PS-PBA brushes for OVA in 50 mM MOPS buffer. PS-PBA brushes and OVA concentrations were 0.4 mg mL^{-1} and 0.3 mg mL^{-1} , respectively.

The results showed that the binding capacity was 631.9 mg g^{-1} at pH 4.0 and 438.1 mg g^{-1} at pH 6.0, respectively. The high binding capacities were ascribed to the lower electrostatic repulsion at pH 4.0 and pH 6.0 and the non-specific hydrophobic interaction between PS-PBA brushes and OVA.

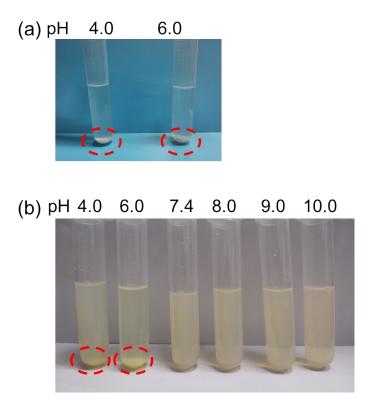


Fig. S7 (a) Photographs of PS-PBA brushes after shaking for 5 h at ionic strength of 50 mM. (b) Photographs of OVA/PS-PBA brushes of different pH values after shaking for 5 h at ionic strength of 50 mM. PS-PBA brushes and OVA concentrations were 0.4 mg mL⁻¹ and 0.3 mg mL⁻¹, respectively.