

Supporting Information

FeMoO₄ Nanospheres-based Nanozymatic Colorimetry for Rapid and Sensitive Pyrophosphate Detection

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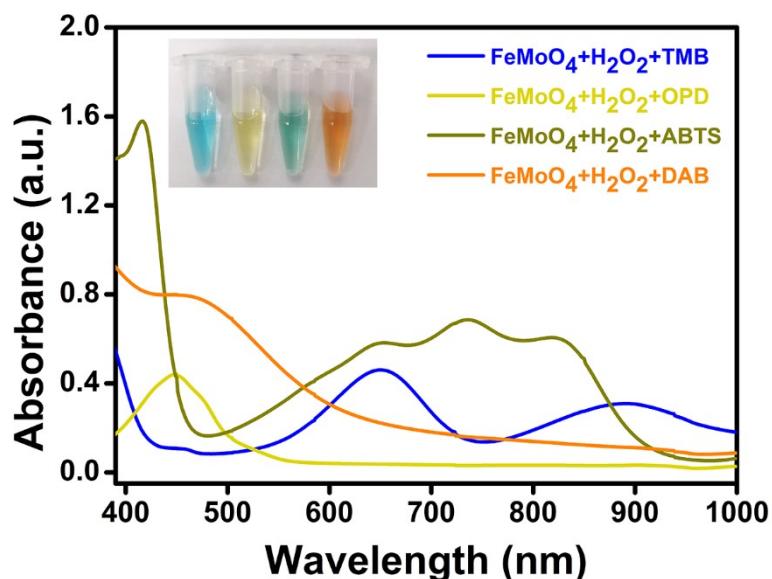


Figure S1. The characteristic UV absorption spectrum and corresponding pictures of four peroxidase substrates during the catalytic reaction.

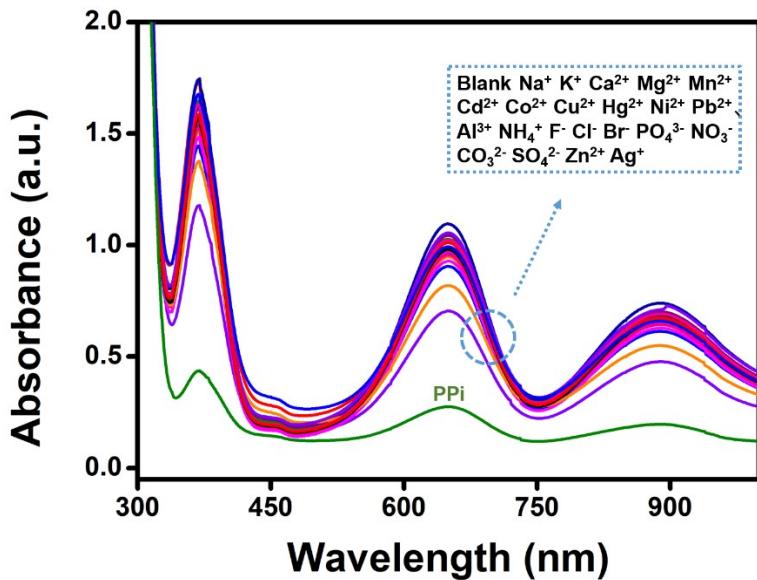


Figure S2. The selectivity of FeMoO₄-TMB-H₂O₂ reaction system toward PPi compared with other potential interference substrates. In a typical detection system (25 °C, pH 4.0), the concentrations of H₂O₂, TMB and FeMoO₄ are 0.5 mM, 0.5 mM and 40 µg mL⁻¹, respectively. Concentrations of the tested cation and anion: none, 30 µM of PPi and PO₄³⁻, 150 µM of Na⁺, K⁺, Ca²⁺, Mg²⁺, Mn²⁺, Cd²⁺, Co²⁺, Cu²⁺, Hg²⁺, Ni²⁺, Pb²⁺, Al³⁺, NH₄⁺, F⁻, Cl⁻, Br⁻, PO₄³⁻, NO₃⁻, CO₃²⁻, SO₄²⁻, Zn²⁺, Ag⁺.

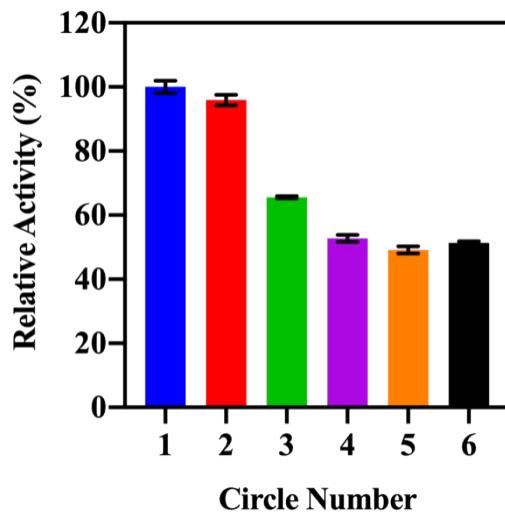


Figure S3. The recycling potential of FeMoO₄. The error bars represent means ± SD (n = 3).

Table S1 The Michaelis-Menten constants (K_m) and maximum initial reaction rates (V_{max}) of the FeMoO₄, HRP, and other iron-containing nanomaterials with peroxidase-mimic activity.

Catalyst	K_m (mM)		V_{max} (10^{-8} M s^{-1})		References
	H ₂ O ₂	TMB	H ₂ O ₂	TMB	
FeMoO ₄	0.174	0.071	28.47	7.7	This work
HRP	3.7	0.434	8.71	10	1
Fe ₃ O ₄	154	0.098	9.78	3.44	1
Fe ₃ S ₄	1.158	0.160	2.168	1.146	2
FeS ₂	0.30	0.17	5.62	3.93	3
FeVO ₄	0.0732	0.691	2.72	2.51	4

References

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