

## Supporting Information

*for*

### **Perylene diimide-based nanoring architecture for exogenous and endogenous ATP detection: Biochemical assay for monitoring phosphorylation of glucose**

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#### Synthesis of tyrosine-based alkyl amide (**1**)

To the solution of Boc-L-tyrosine (500 mg, 1.77 mmol) in a minimum quantity of DMF, N,N'-diisopropylethylamine (DIEA) (310  $\mu$ L, 1.77 mmol) was added. Subsequently, 1-Ethyl-3-(3'-dimethylaminopropyl) carbodiimide hydrochloride (EDC.HCl) (375 mg, 1.95 mmol) and hydroxybenzotriazole (HOBt) (264 mg, 1.95 mmol) were slowly added to the reaction mixture. After a time interval of 15 minutes, *tert*-butyl (4-aminobutyl)carbamate (334 mg, 1.77 mmol) was added to the reaction mixture and stirring continued for another 4h. The progress of the reaction was checked by *tlc*. After completion of the reaction, the DMF was removed on a rotary evaporator and residue was extracted with water and ethyl acetate. The ethyl acetate part was dried over Na<sub>2</sub>SO<sub>4</sub> and then concentrated under vacuum. The crude product was purified by column chromatography (neutralized SiO<sub>2</sub>, 3: 97, EtOAc: CHCl<sub>3</sub>, v/v) to isolate pure compound **1**, solid, R<sub>f</sub> = 0.45; yield = 56%; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, 25°C);  $\delta$  7.56 (bs, 1H, -OH), 7.05 (d, *J* = 7.5 Hz, 2H), 6.80 (d, *J* = 8.0 Hz, 2H), 5.99 (bs, 1H, -NH), 5.31 (bs, 1H, -NH), 4.79 (bs, 1H, -NH), 4.22 (bt, 1H, tyrosine -CH), 2.86–3.24 (m, tyrosine -CH<sub>2</sub> and -CH<sub>2</sub> of butyl chain, 6H), 1.47 (s, 9H), 1.44 (s, 9H), 1.34–1.21 (m, 4H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>, 25°C)  $\delta$  171.40, 156.41, 155.55, 155.48, 130.37, 128.00, 115.90, 80.20, 79.74, 56.51, 40.22, 39.00, 38.28, 28.47, 28.33, 27.63, 26.51; IR spectrum (ATR):  $\nu_{\max}$  [cm<sup>-1</sup>] = 3339.7, 2937.1, 1759, 1684.8, 1513.3, 1364.2, 1237.5, 1162.9 cm<sup>-1</sup>.

## 1. Experimental Section

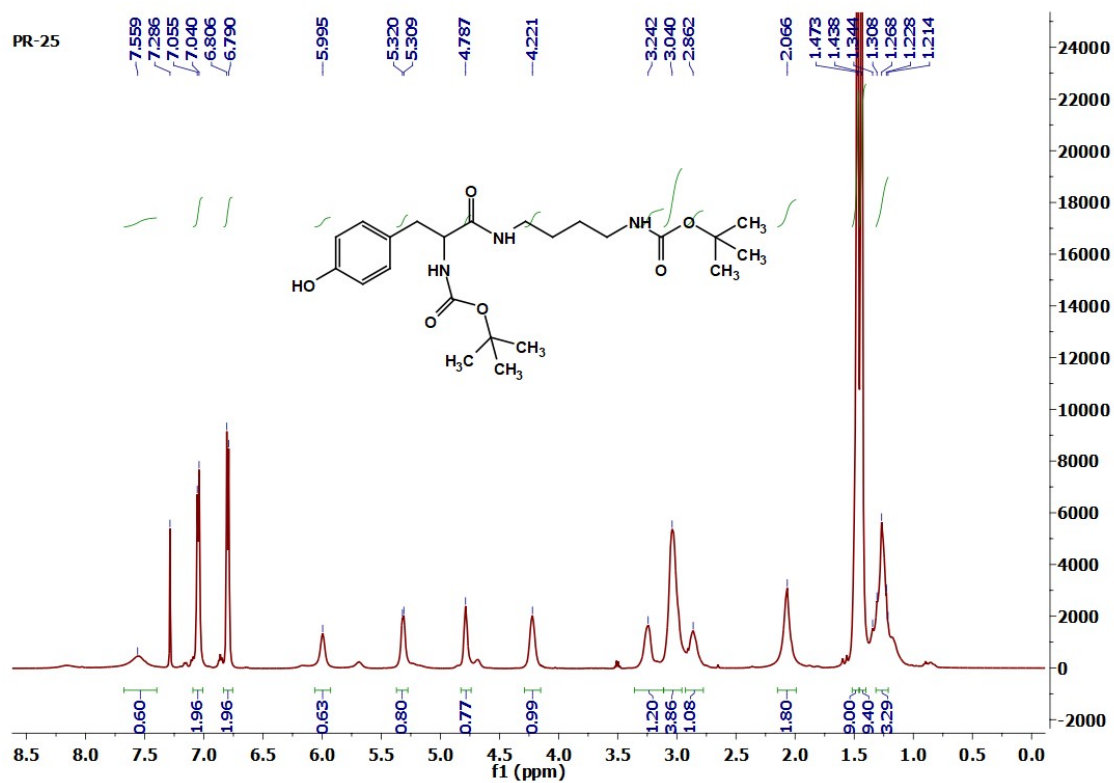


Figure S1a: <sup>1</sup>H NMR spectrum of Compound 1 in CDCl<sub>3</sub>.

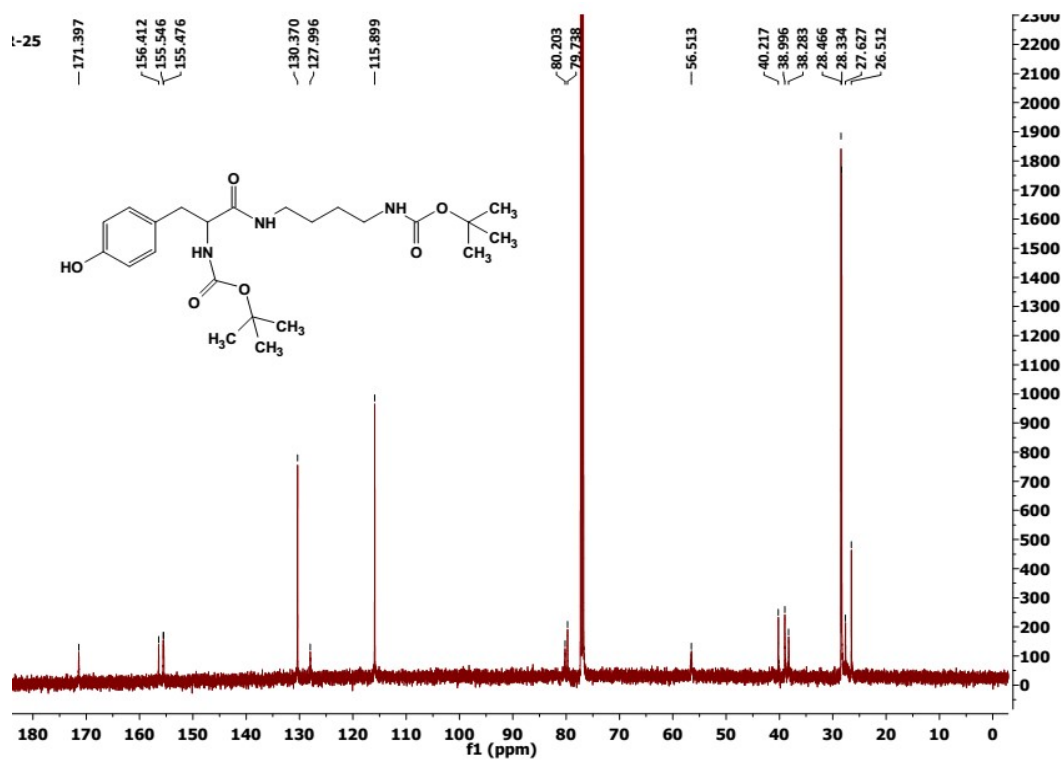


Figure S1b: <sup>13</sup>C NMR spectrum of Compound 1 in CDCl<sub>3</sub>.

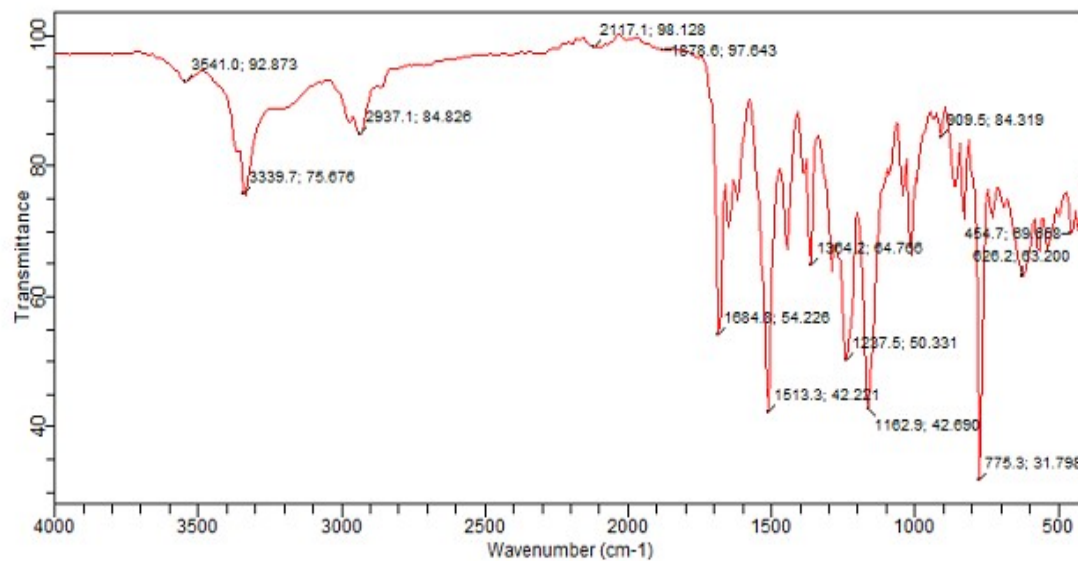


Figure S1c: FTIR (ATR) spectrum of Compound 1.

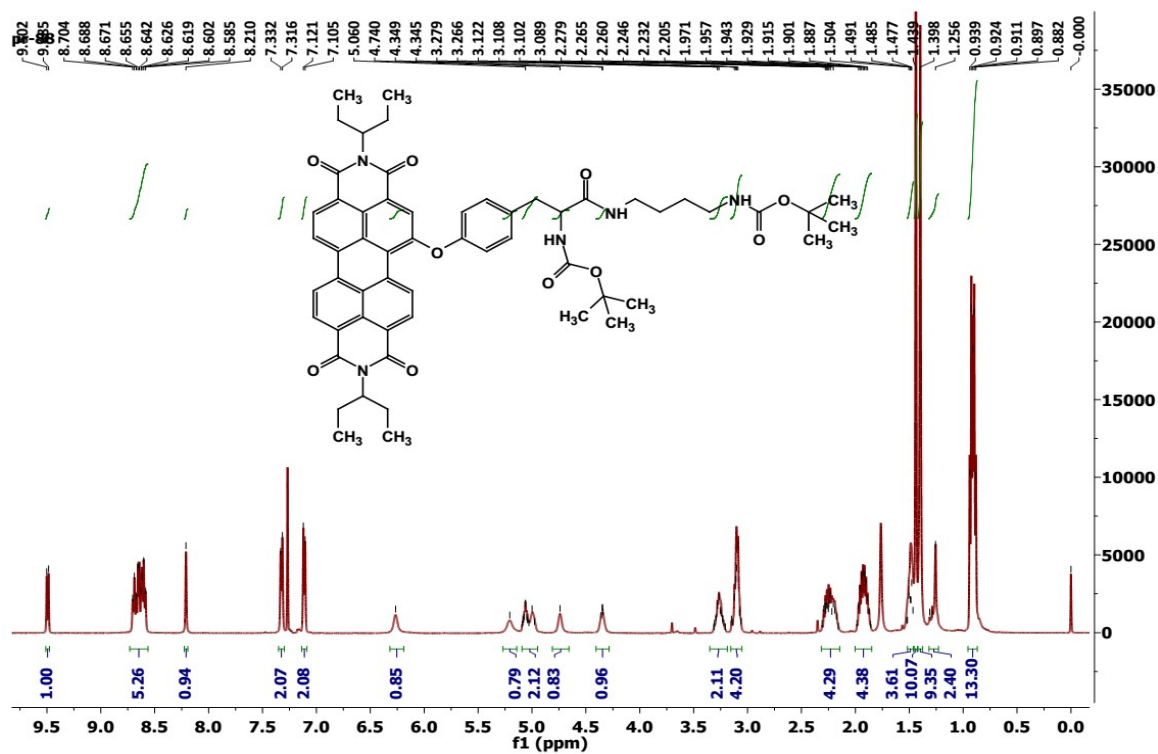


Figure S2a:  $^1\text{H}$  NMR spectrum of PDI 2 in  $\text{CDCl}_3$ .

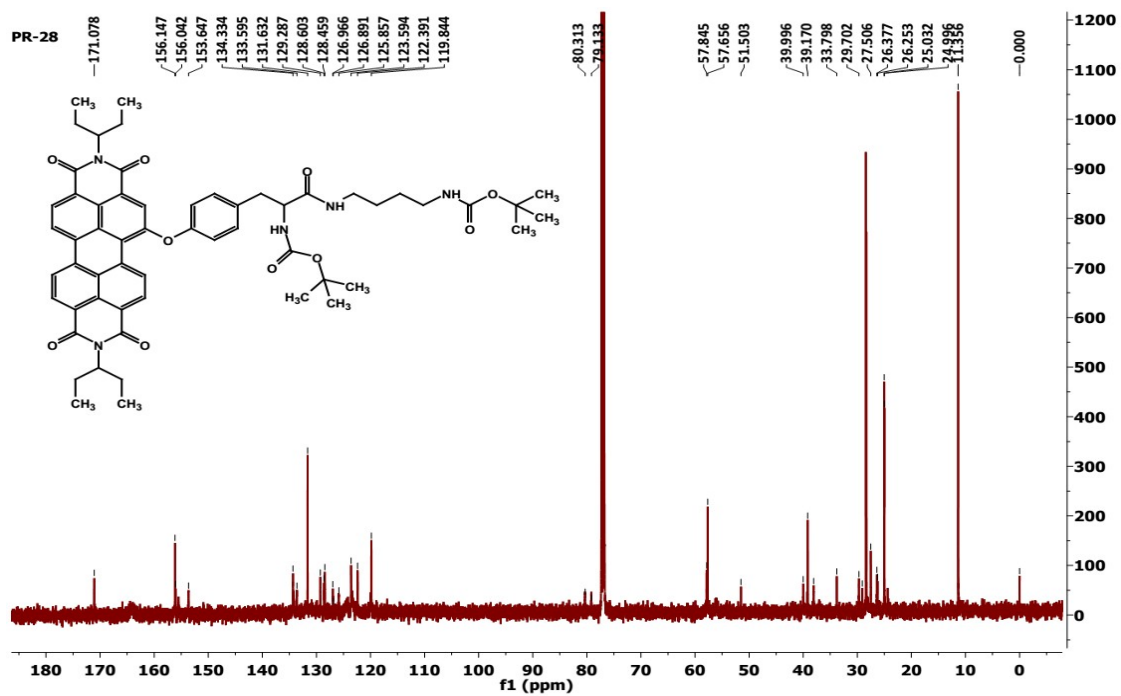


Figure S2b:  $^{13}\text{C}$  NMR spectrum of PDI **2** in  $\text{CDCl}_3$ .

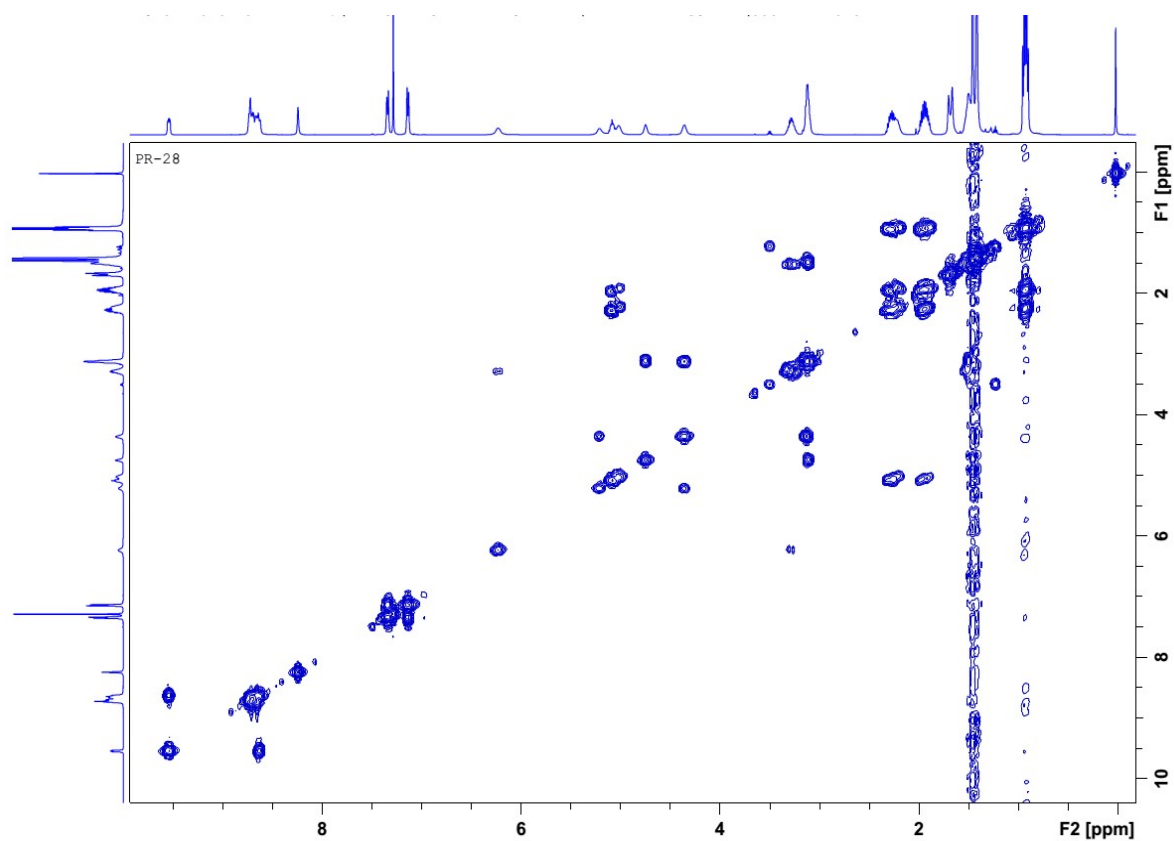


Figure S2c:  $^1\text{H}$ - $^1\text{H}$  COSEY NMR spectrum of PDI **2** in  $\text{CDCl}_3$ .

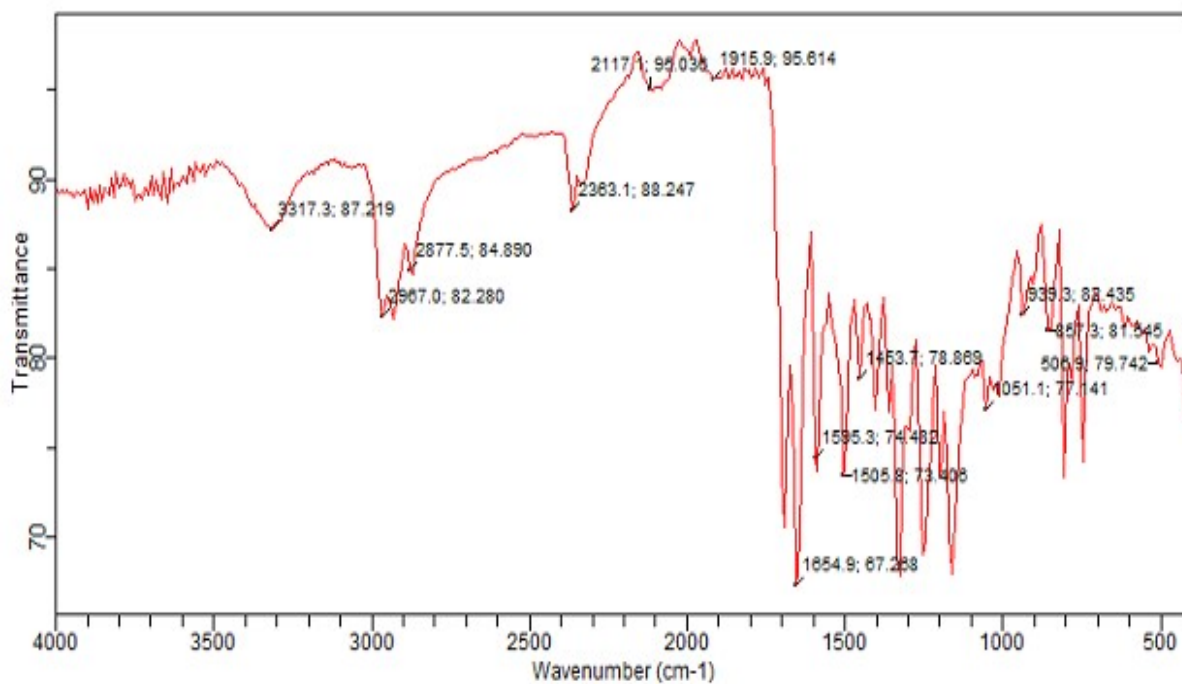


Figure S2d: FTIR spectrum of PDI 2.

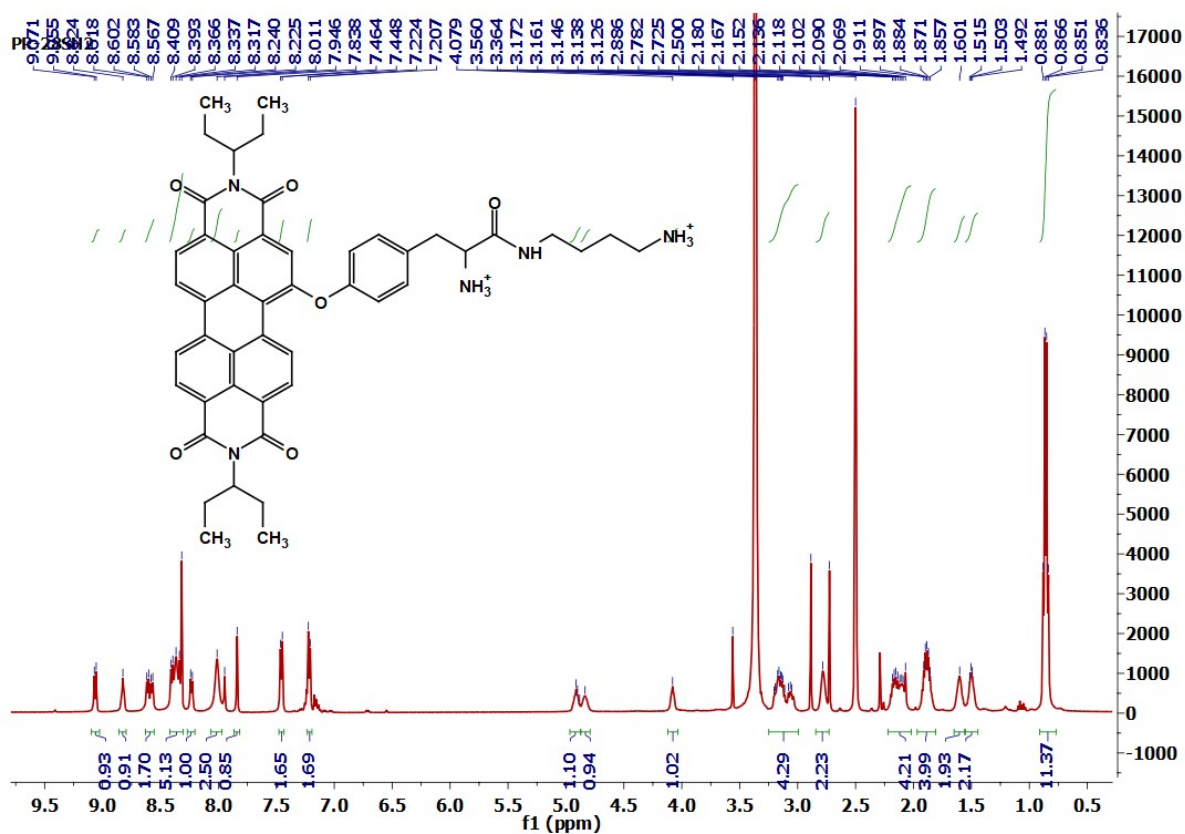


Figure S3a: <sup>1</sup>H NMR spectrum of PDI 3 in DMSO-*d*<sub>6</sub>.

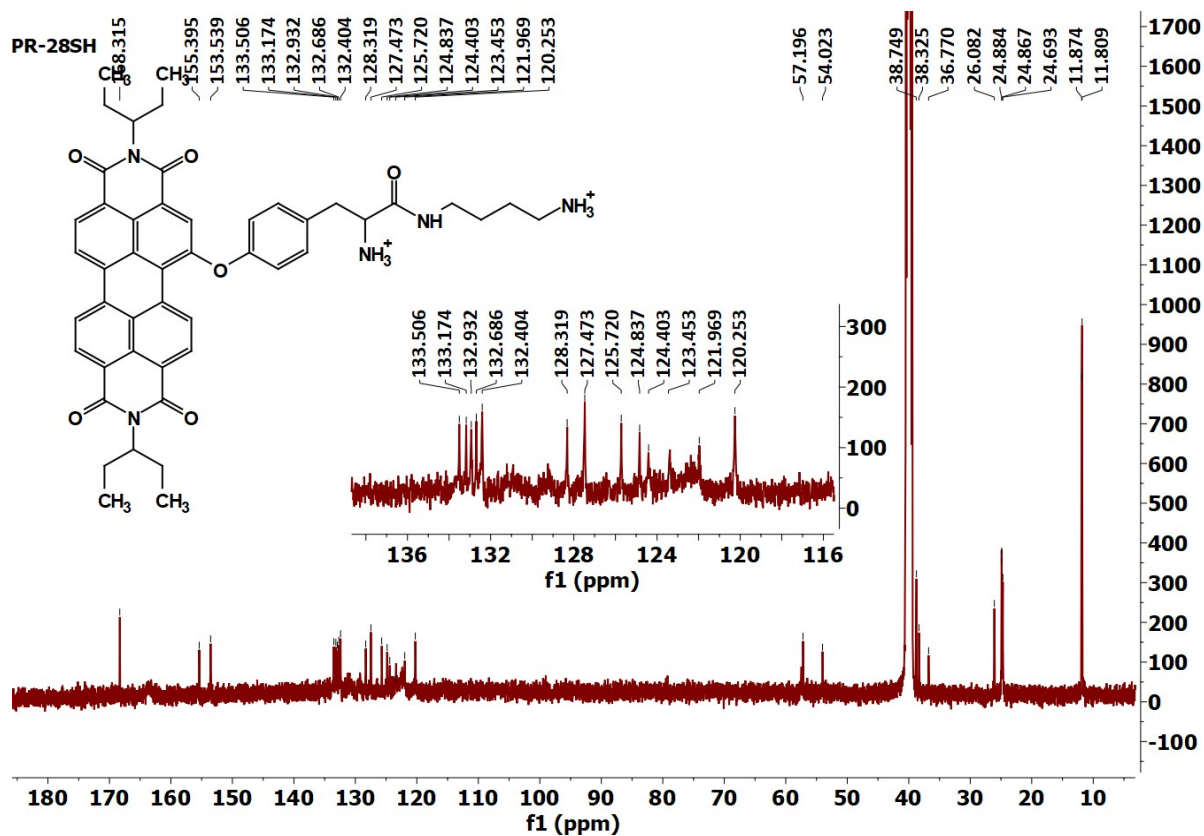


Figure S3b:  $^{13}\text{C}$  NMR spectrum of PDI 3 in  $\text{DMSO-}d_6$ .

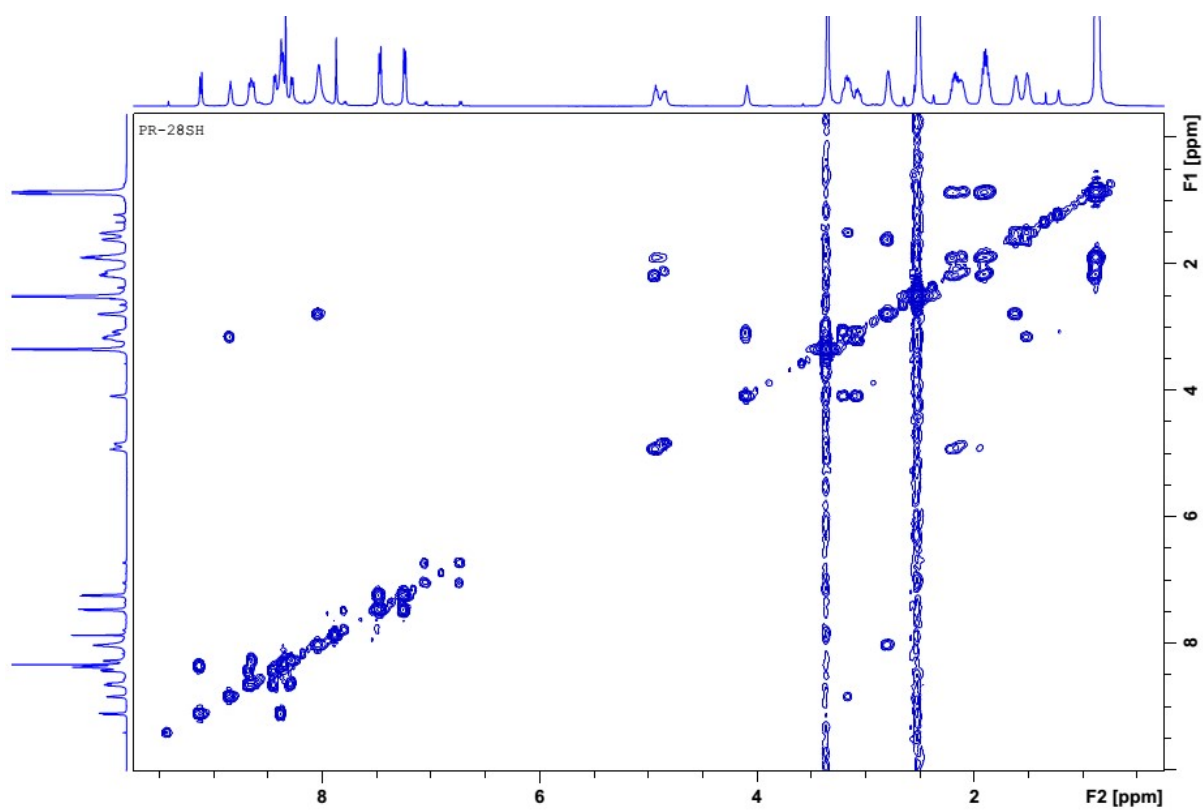
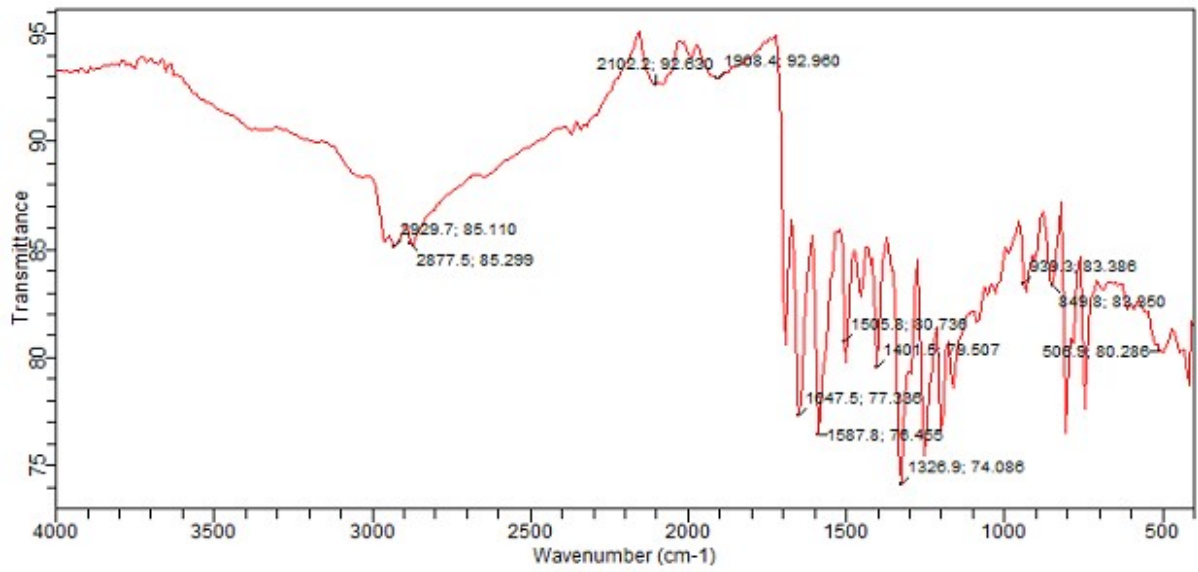
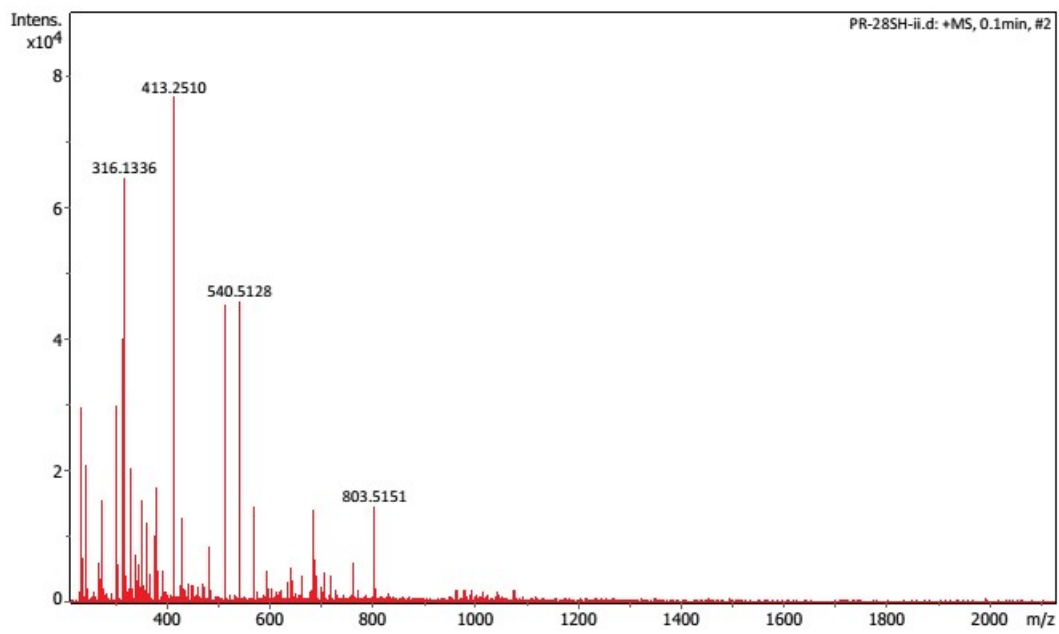


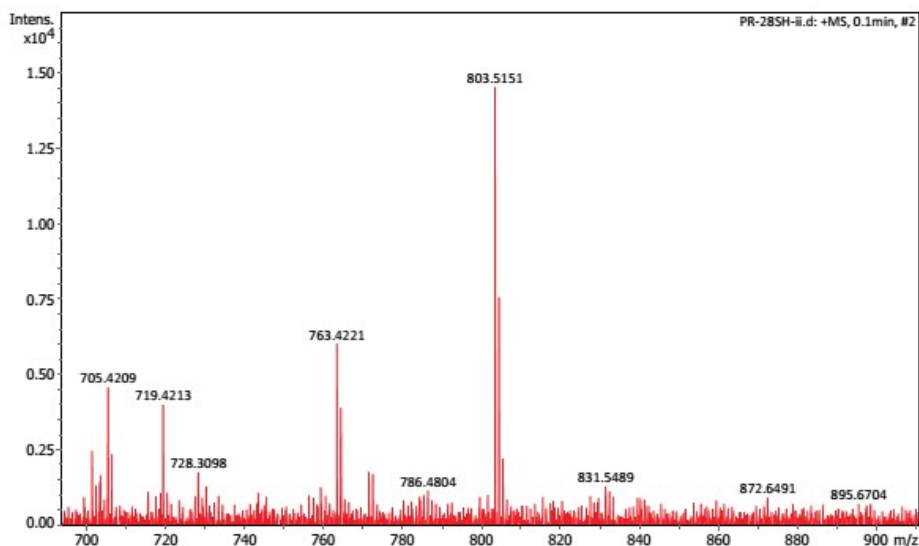
Figure S3c:  $^1\text{H-}^1\text{H}$  COY NMR spectrum of PDI 3 in  $\text{DMSO-}d_6$ .



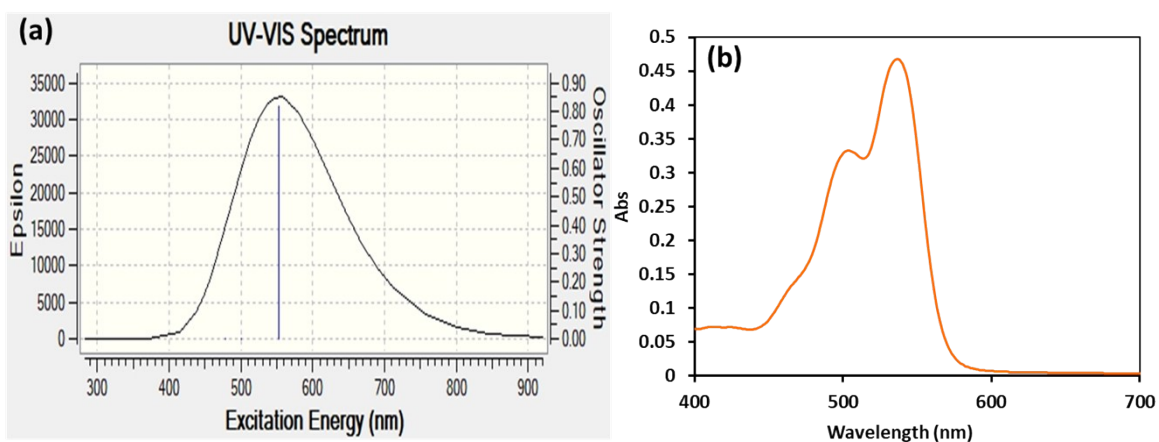


**Figure S3d:** FTIR spectrum of PDI 3.

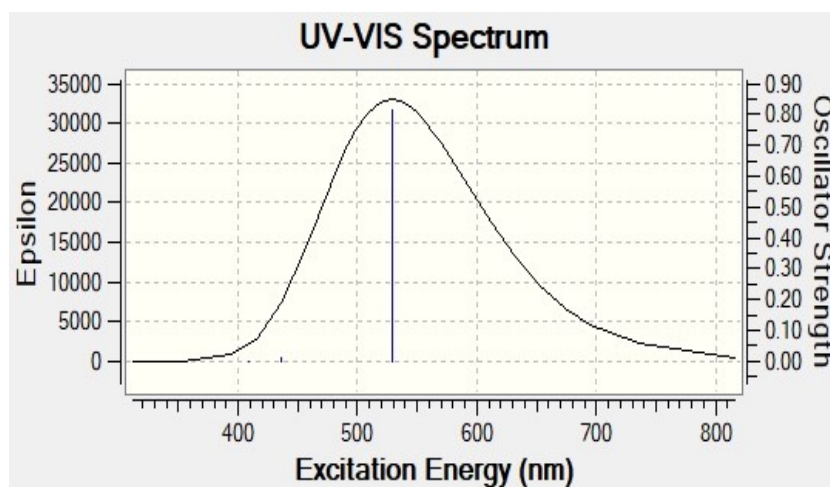




**Figure S3e:** Mass spectrum (Full and zoomed view) of PDI 3.



**Figure S4a:** (a) TD-DFT spectrum of PDI 3; (b) experimentally determined UV-Vis spectrum of PDI 3.



**Figure S4b:** TD-DFT spectrum of PDI 2.



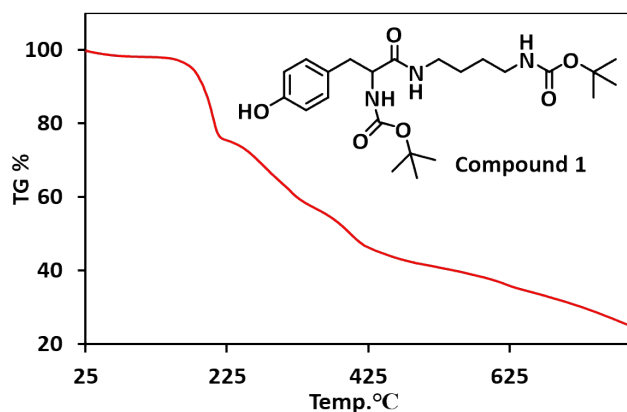


Figure S5: TGA data of compound 1.

#### Anton Paar Polarimeter - Measurement(s)

MCP 150  
Software Version: 1.50.4074.82  
Serial Number: 82104498

##### Unique Id 413

▶ Sample Name:  
▶ Date:  
▶ Username:  
▶ Sample State:

#### Compound 1 (CHCl<sub>3</sub>)

03/12/2020 - 05:00 PM  
Administrator  
Ok

▶ Measurement Mode: Specific Rotation  
▶ Measurement Result: +5.000 °  
▶ Concentration: +0.400 g/100ml  
▶ Optical Rotation: +0.002 °  
▶ Set Temperature: +25.0 °C  
▶ Temperature: +25.0 °C

#### Anton Paar Polarimeter - Measurement(s)

MCP 150  
Software Version: 1.50.4074.82  
Serial Number: 82104498

##### Unique Id 539

▶ Sample Name:  
▶ Date:  
▶ Username:  
▶ Sample State:

#### PDI 2 (CHCl<sub>3</sub>)

10/31/2020 - 11:36 AM  
Administrator  
Ok

▶ Measurement Mode: Specific Rotation  
▶ Measurement Result: +27.500 °  
▶ Concentration: +0.400 g/100ml  
▶ Optical Rotation: +0.011 °  
▶ Set Temperature: +25.0 °C  
▶ Temperature: +25.0 °C

#### Anton Paar Polarimeter - Measurement(s)

MCP 150  
Software Version: 1.50.4074.82  
Serial Number: 82104498

##### Unique Id 544

▶ Sample Name:  
▶ Date:  
▶ Username:  
▶ Sample State:

#### PDI 3 (CHCl<sub>3</sub>)

10/31/2020 - 12:04 PM  
Administrator  
Ok

▶ Measurement Mode: Specific Rotation  
▶ Measurement Result: +25.000 °  
▶ Concentration: +0.400 g/100ml  
▶ Optical Rotation: +0.010 °  
▶ Set Temperature: +25.0 °C  
▶ Temperature: +25.0 °C

Anton Paar Polarimeter - Measurement(s)    Anton Paar Polarimeter - Measurement(s)  
 MCP 150    MCP 150  
 Software Version: 1.50.4074.82    Software Version: 1.50.4074.82  
 Serial Number: 82104498    Serial Number: 82104498

**Unique Id 642**

▶ Sample Name: **PDI 3 (DMSO)**  
 ▶ Date: 07/17/2021 - 01:34 PM  
 ▶ Username: Administrator  
 ▶ Sample State: Ok

▶ Measurement Mode: Specific Rotation  
 ▶ Measurement Result: +20.000 °  
 ▶ Concentration: +0.100 g/100ml  
 ▶ Optical Rotation: +0.002 °  
 ▶ Set Temperature: +25.0 °C  
 ▶ Temperature: +25.0 °C  
 ▶ Wavelength in air: +589.28 nm  
 ▶ Wavelength in vacuum: +589.44 nm  
 ▶ Cell Length: +10.00 mm

**Unique Id 640**

▶ Sample Name: **ATP Complex(DMSO)**  
 ▶ Date: 07/17/2021 - 01:15 PM  
 ▶ Username: Administrator  
 ▶ Sample State: Ok

▶ Measurement Mode: Specific Rotation  
 ▶ Measurement Result: -80.000 °  
 ▶ Concentration: +0.100 g/100ml  
 ▶ Optical Rotation: -0.008 °  
 ▶ Set Temperature: +25.0 °C  
 ▶ Temperature: +25.0 °C  
 ▶ Wavelength in air: +589.28 nm  
 ▶ Wavelength in vacuum: +589.44 nm  
 ▶ Cell Length: +10.00 mm

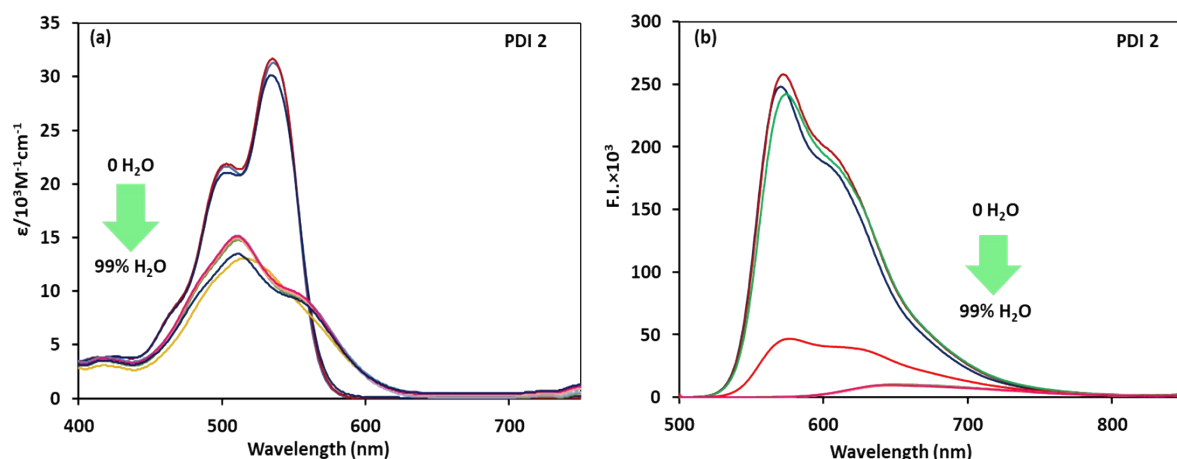
**Figure S6:** Specific optical rotations of all synthesized derivatives.

Table for Specific optical rotation values

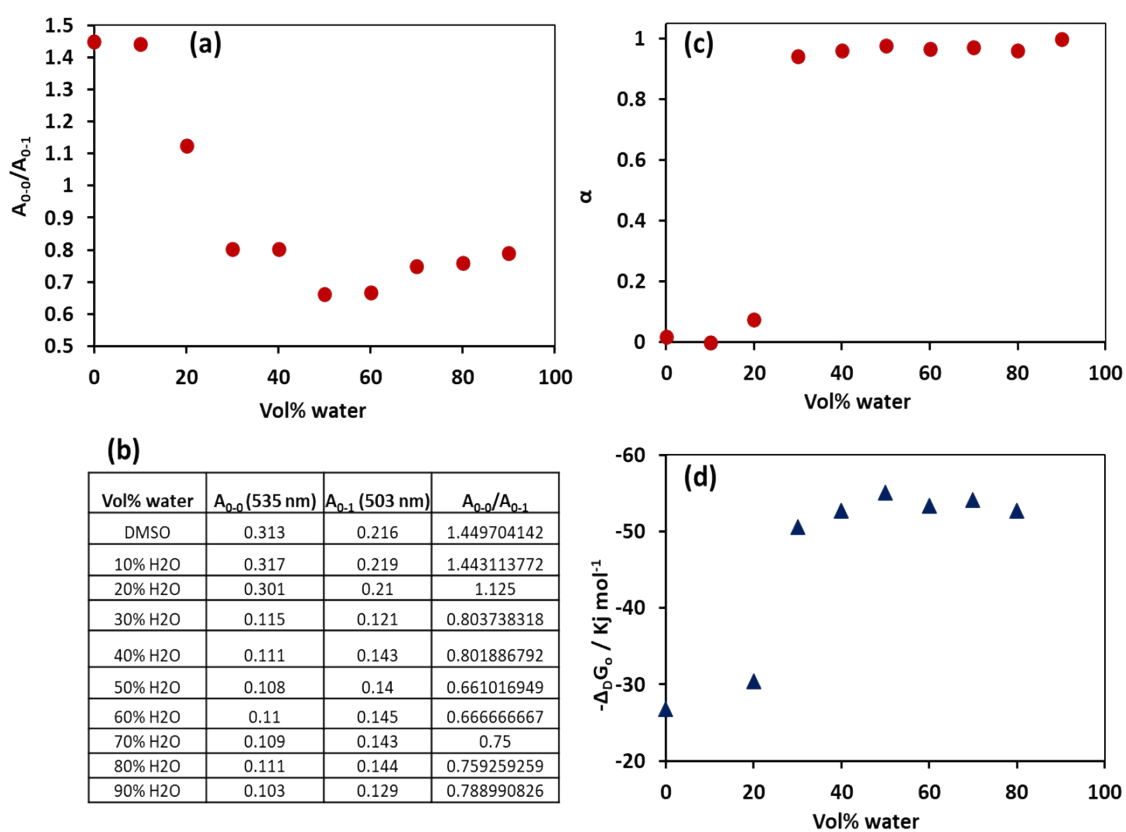
Compound	Solvent	Optical Rotation	Concentration	Specific Rotation
Compound 1	CHCl <sub>3</sub>	+0.002°	+0.400 g/100ml	+5.000°
PDI 2	CHCl <sub>3</sub>	+0.011°	+0.400 g/100ml	+27.500°
PDI 3	CHCl <sub>3</sub>	+0.010°	+0.400 g/100ml	+25.000°

Table for Specific optical rotation values

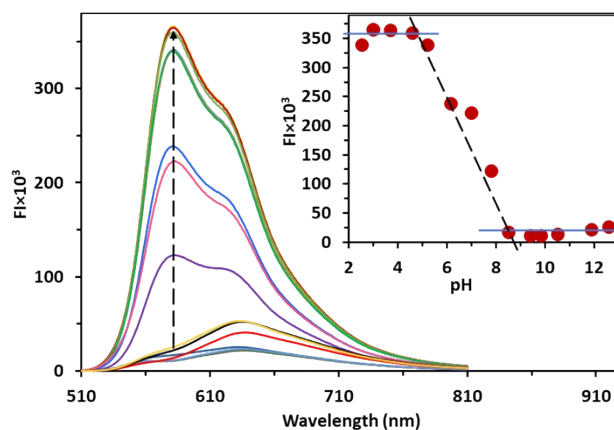
Compound	Solvent	Optical Rotation	Concentration	Specific Rotation
Compound 1	CHCl <sub>3</sub>	+0.002°	+0.400 g/100ml	+5.000°
PDI 2	CHCl <sub>3</sub>	+0.011°	+0.400 g/100ml	+27.500°
PDI 3	CHCl <sub>3</sub>	+0.010°	+0.400 g/100ml	+25.000°
PDI 3	DMSO	+0.002°	+0.100 g/100ml	+20.000°
PDI 3 + ATP	DMSO	-0.008°	+0.100 g/100ml	-80.000°



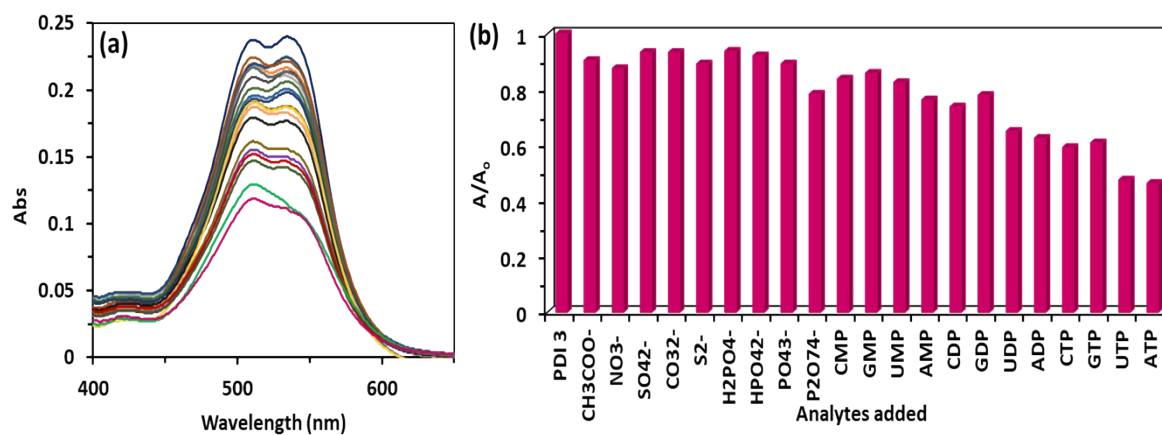
**Figure S7:** The UV–Vis (a) and fluorescence (b) spectra of **PDI 2** (10  $\mu\text{M}$ ) in different DMSO–water ratios.  $\lambda_{\text{ex}} = 490 \text{ nm}$ , slit width (ex/em) = 5/5.



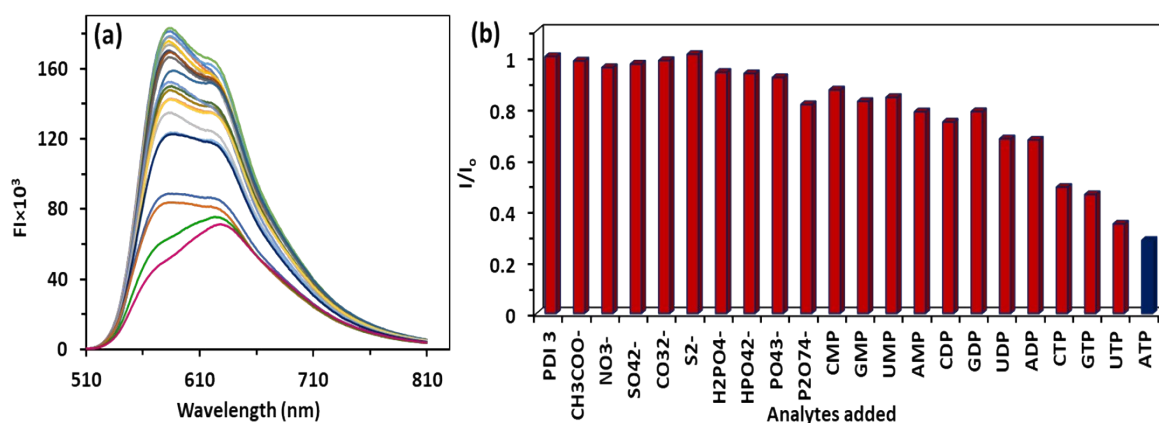
**Figure S8:** (a,b) Plot and tabulated value of Franck–Condon ( $A_{0-0}/A_{0-1}$ ) values of **PDI 2** versus H<sub>2</sub>O fraction in DMSO; (c) Plot of the degree of aggregation ( $\alpha_{\text{agg}}$ ) versus H<sub>2</sub>O fraction in DMSO and (d) plot of  $\Delta G^*$  versus H<sub>2</sub>O fraction in DMSO.



**Figure S9.** The fluorescence pH titration of PDI 3 (5  $\mu$ M) in 50% H<sub>2</sub>O-DMSO solution.



**Figure S10:** (a) UV-Vis response of PDI 3 toward various anionic analytes; (b) bar graph of  $I/I_0$  plots of absorbance and emission data showing response towards anions and NPPs.



**Figure S11:** (a) Fluorescence response of PDI 3 toward various anionic analytes; (b) bar graph of  $I/I_0$  plots of absorbance and emission data showing response towards anions and NPPs. (Not complete Image).

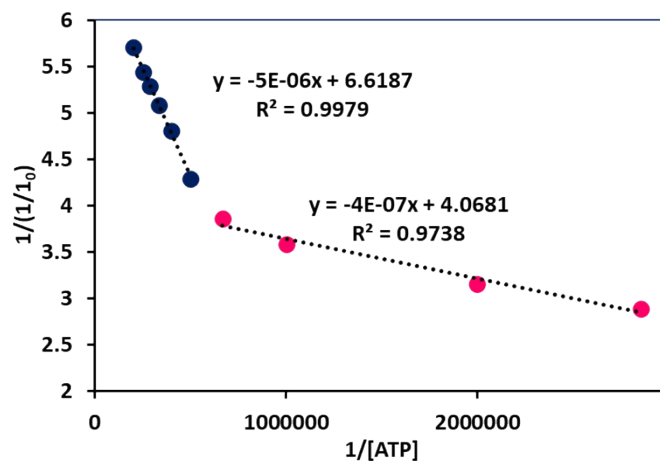


Figure S12a: B-H Plot of PDI 3 with ATP.

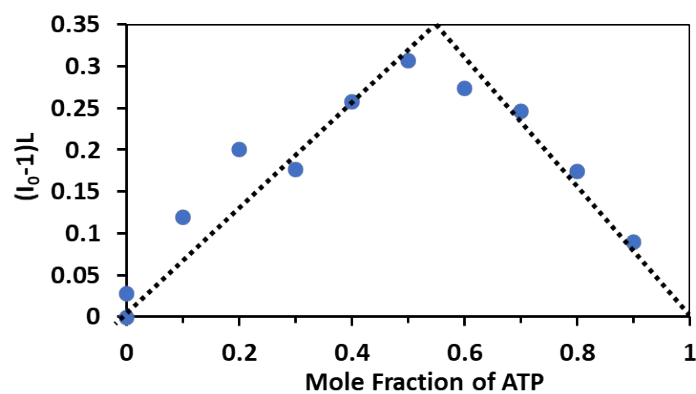
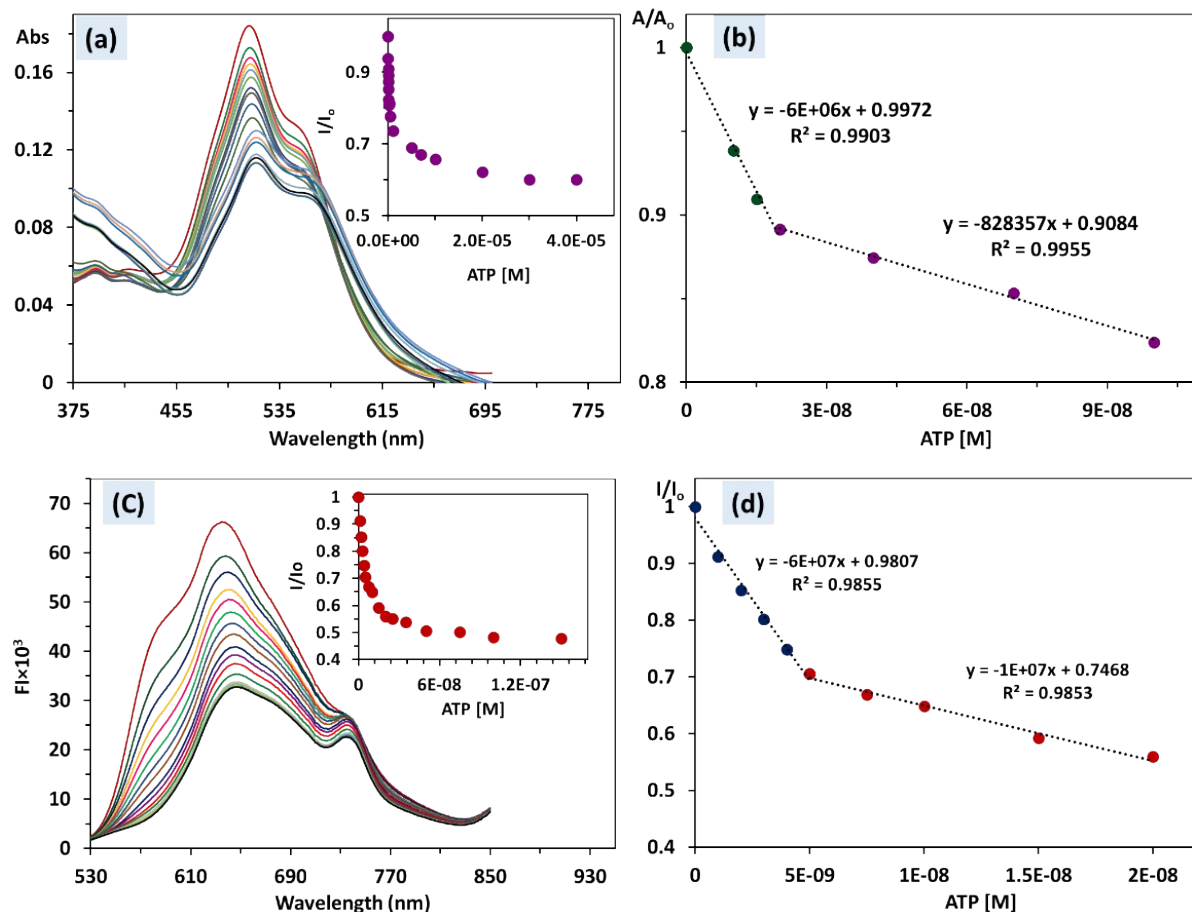


Figure S12b: Job plot of PDI 3 with ATP.

Table S2 Changes in the Franck-Codon factor ( $A_{0-0}/A_{0-1}$ ) on interaction with ATP:

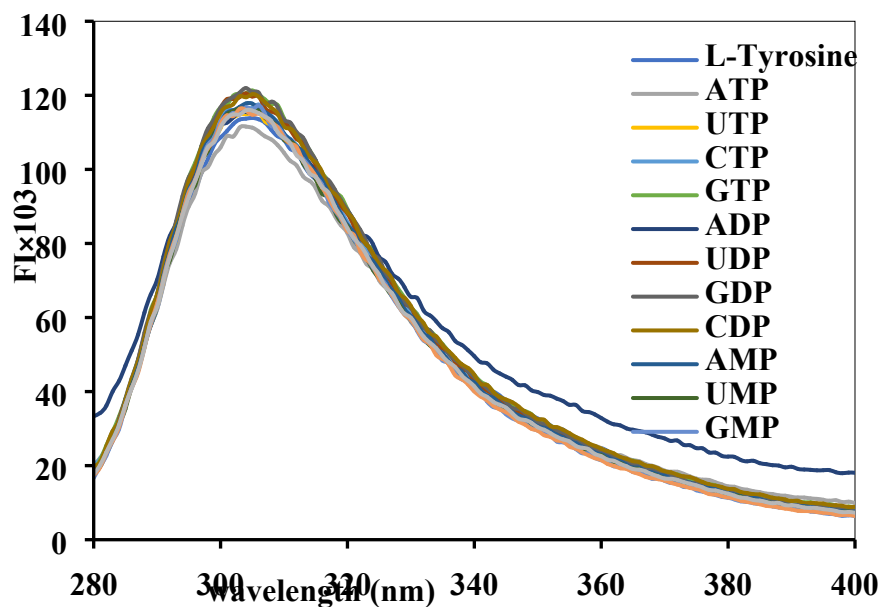
ATP Equivalents	$A_{0-1}$ (512 nm)	$A_{0-0}$ (535 nm)	$A_{0-0}/A_{0-1}$
PDI 3	0.268	0.274	1.022388
0.001 ATP	0.254	0.26	1.023622
0.005 ATP	0.23	0.238	1.034783
0.01 ATP	0.219	0.227	1.03653
0.015 ATP	0.215	0.223	1.037209
0.02 ATP	0.209	0.216	1.033493
0.07 ATP	0.192	0.197	1.026042
0.1 ATP	0.188	0.193	1.026596
0.15 ATP	0.184	0.187	1.016304
0.2 ATP	0.178	0.178	1
0.3 ATP	0.169	0.166	0.982249
0.4 ATP	0.167	0.159	0.952096
0.5 ATP	0.162	0.15	0.925926
0.8 ATP	0.156	0.142	0.910256
1 ATP	0.154	0.136	0.883117

<b>1.5 ATP</b>	0.151	0.132	0.874172
<b>2 ATP</b>	0.147	0.123	0.836735
<b>3 ATP</b>	0.145	0.12	0.827586
<b>4 ATP</b>	0.14	0.114	0.814286
<b>5 ATP</b>	0.14	0.114	0.814286
<b>6 ATP</b>	0.139	0.112	0.805755

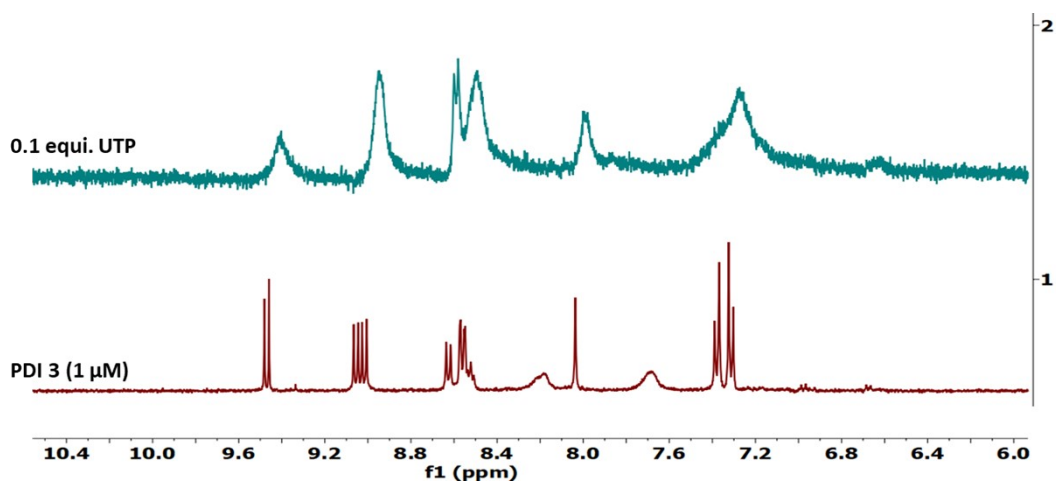


**Fig. S13** Changes in (a) absorbance (10  $\mu$ M) and (c) emission (5  $\mu$ M) spectra of PDI 3 with gradual increase of ATP concentrations in 9:1 HEPES buffer: DMSO medium; ( $\lambda_{ex}$  = 490 nm; slit width ( $E_x/E_m$ ) = 10/10 nm; [Inset of (a,c)] plot of absorbance ( $A/A_0$ ) and emission ( $I/I_0$ ) changes with change in ATP concentration, respectively; Linear correlation plot of (b)  $A/A_0$  and (d)  $I/I_0$  versus the concentration of ATP to determine the lowest limit of detection.





**Fig. S13a** Changes in emission (5 μM) spectra of tyrosine with gradual increase of nucleoside polyphosphates (NPPs) concentrations in 9:1 HEPES buffer: DMSO medium; ( $\lambda_{\text{ex}} = 260 \text{ nm}$ ; slit width ( $E_x/E_m$ ) = 10/10 nm.



**Fig. S14** Partial <sup>1</sup>H NMR spectra (stacked) of PDI 3 (1 mM) after incremental addition of ATP in DMSO-*d*<sub>6</sub> solution.

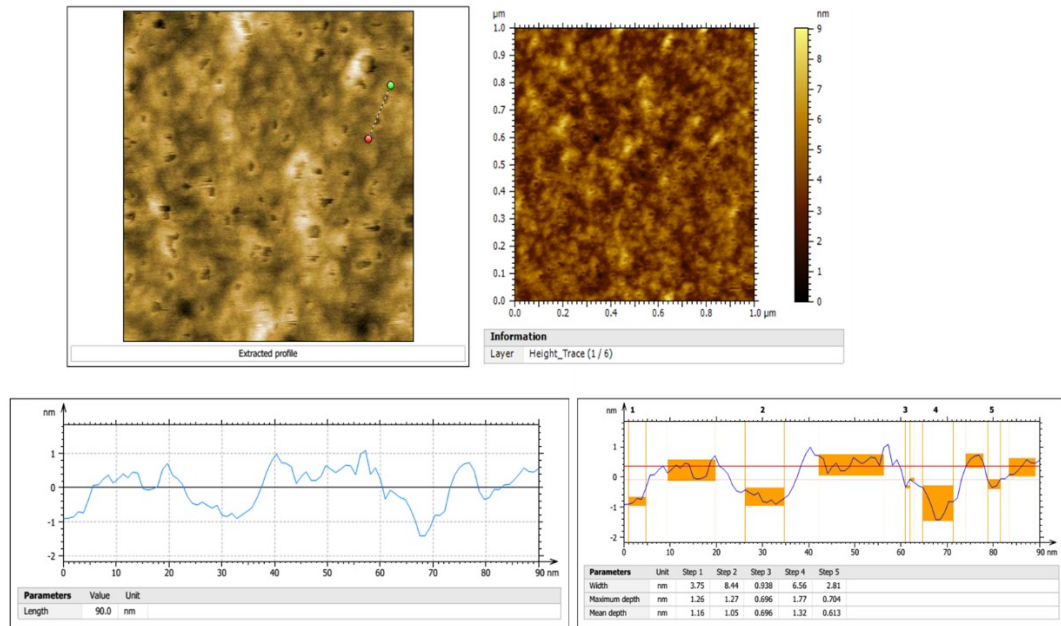


Fig. S15. AFM images of PDI 3 in 50% H<sub>2</sub>O-DMSO solution.

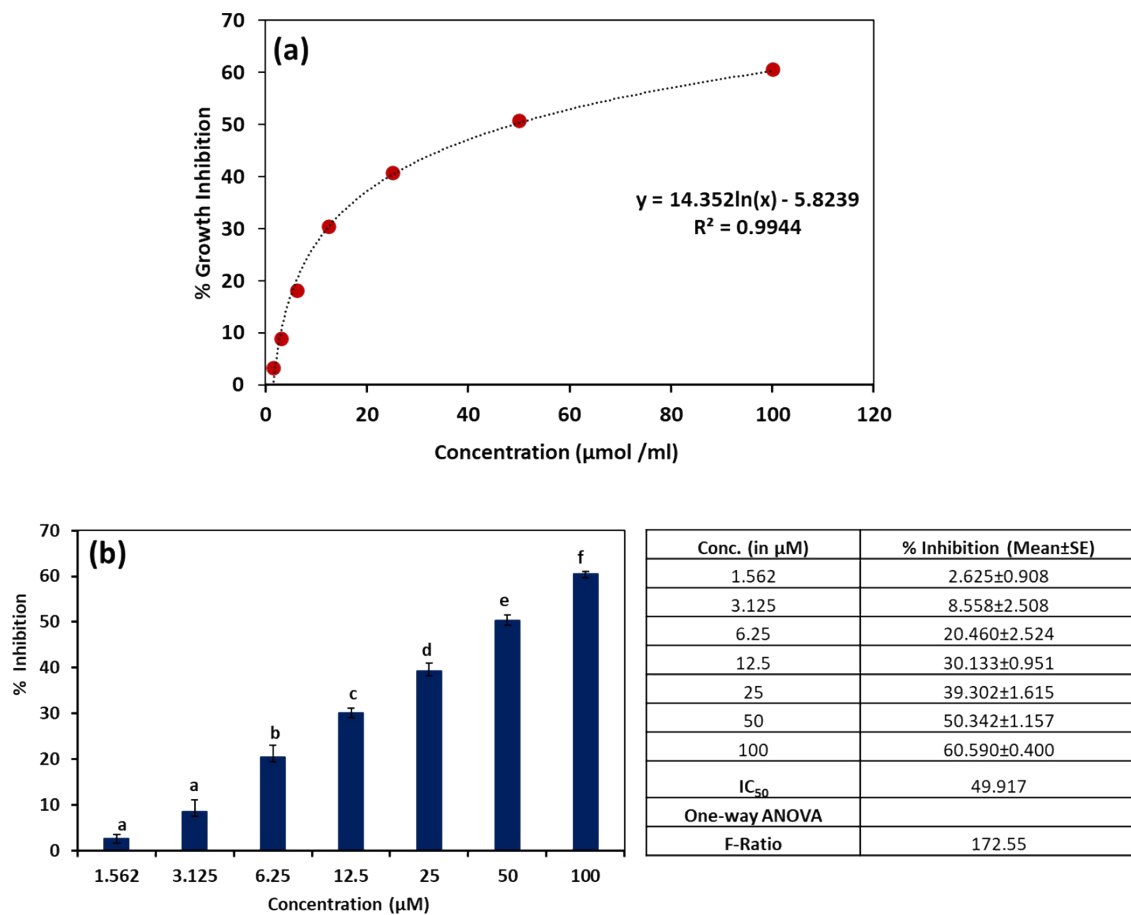
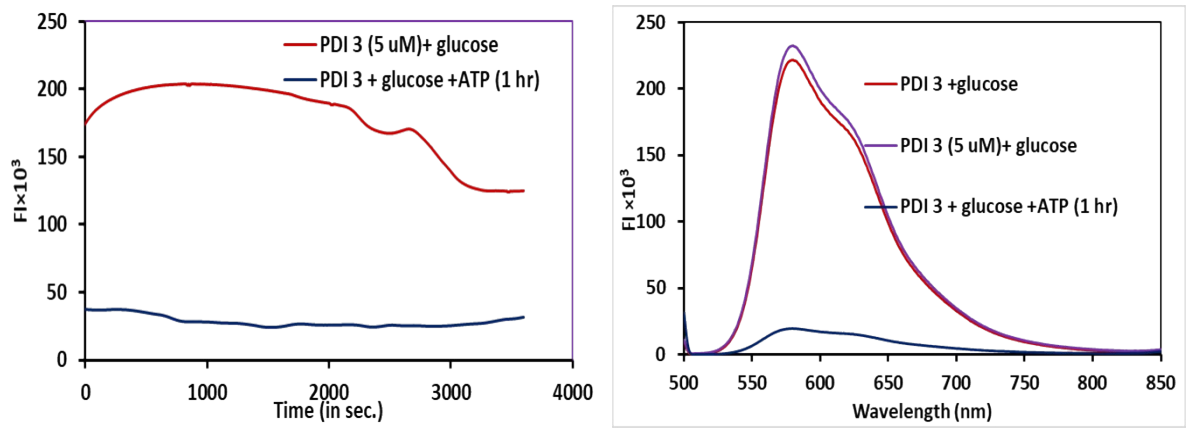


Fig. S16 MTT assay to check the cytotoxicity of PDI 3 with statistical distribution.



**Fig. S17:** Fluorescence spectrum of PDI 3+ATP after regular interval of time.