

Supporting Information

Prediction of pressure-induced superconductivity in the novel ternary system $\text{ScCaH}_{2n}(n=1\text{-}6)$

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Figure S1. The phonon spectrum of ScCaH_{2n} ($n=1-3,5$). There is absence imaginary frequency of each structure to show its dynamic stability.

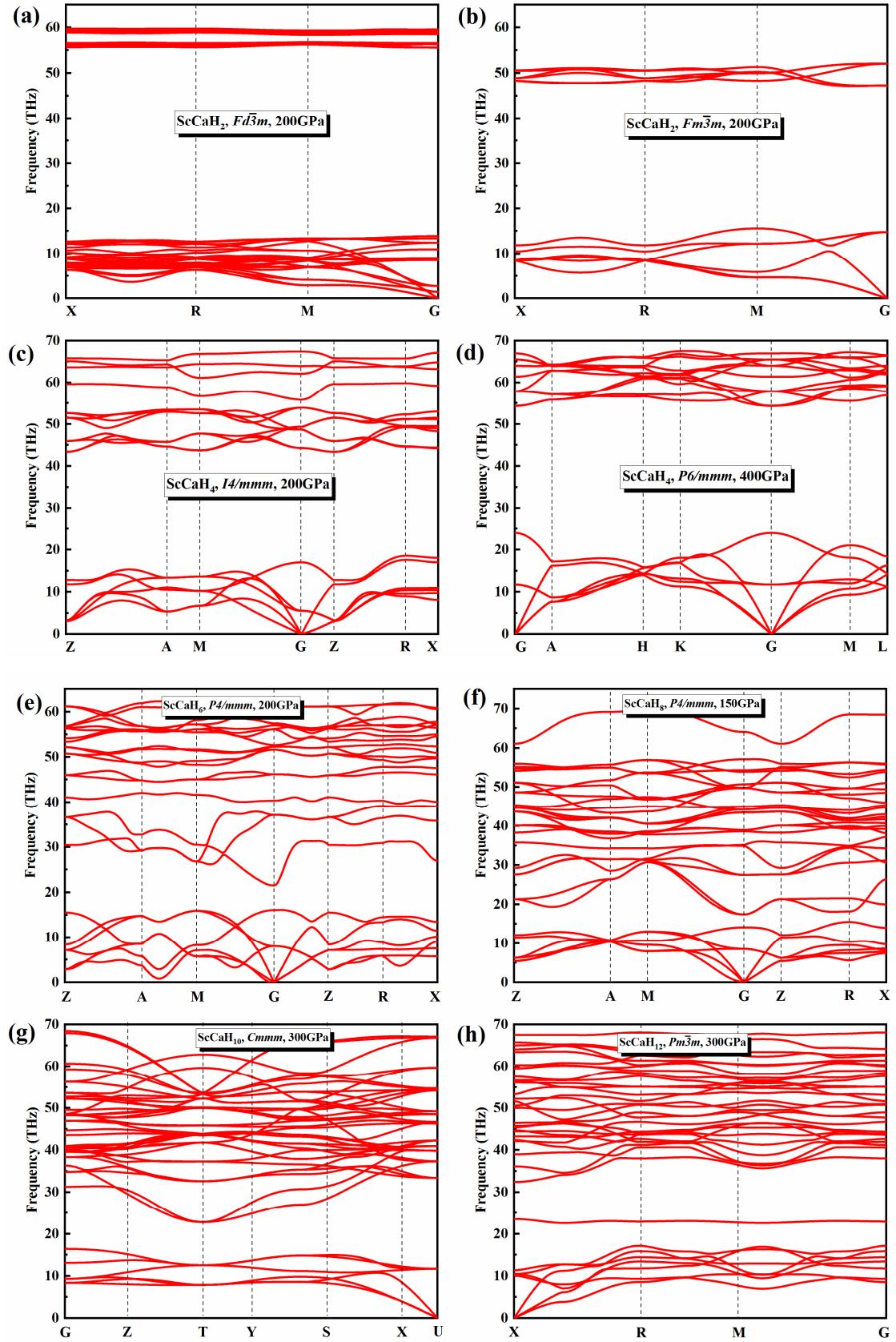


Figure S2. The enthalpies of formation with ZPE of $\text{ScCaH}_{2n}(n=2-6)$ with respect to ScCaH_2 and H_2 at 100-300 GPa.

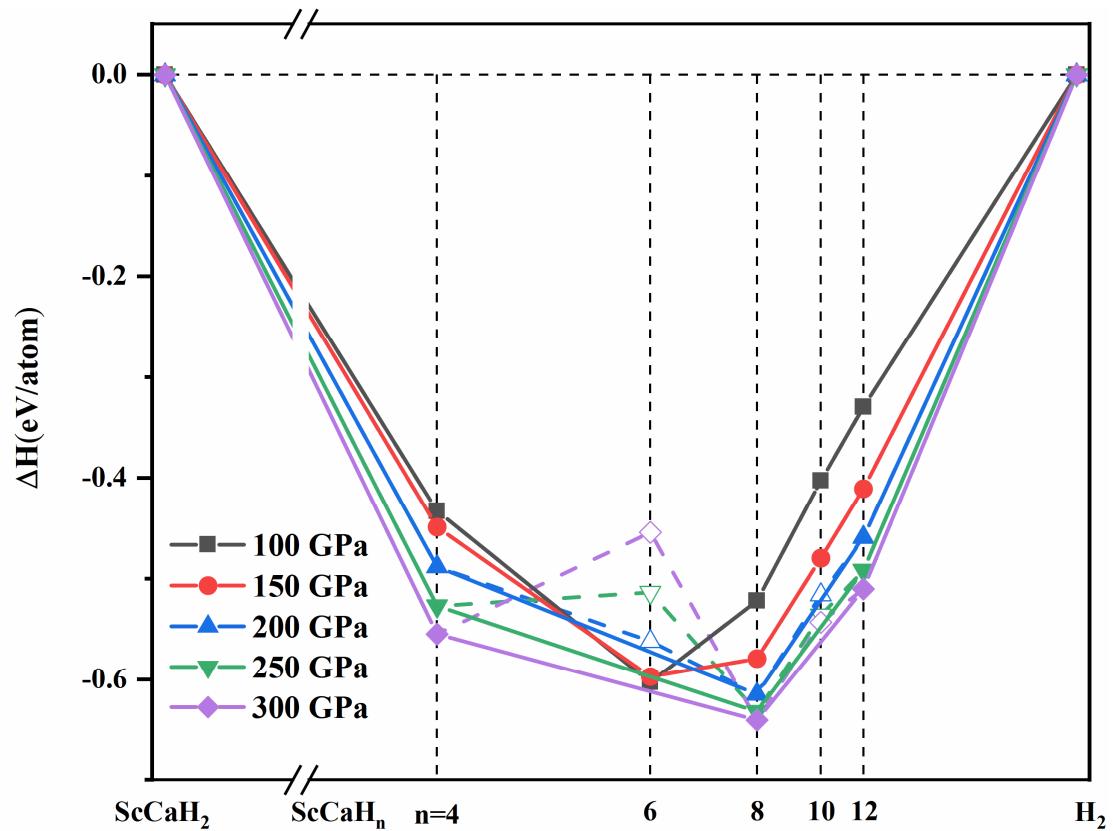
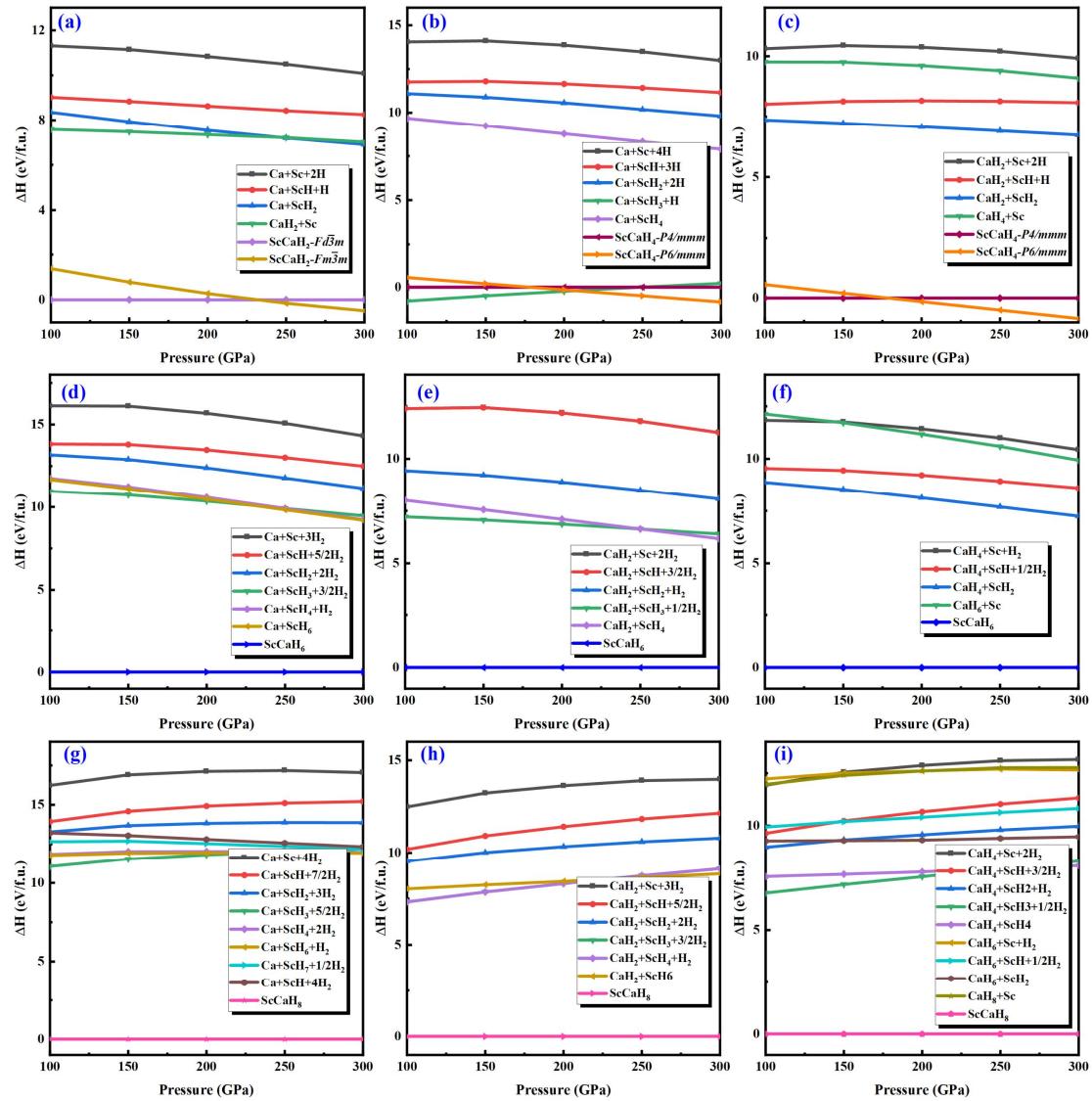


Figure S3. (a) The enthalpy difference curves with respect to ScCaH_2 as a function of pressure; (b)-(c) The enthalpy difference curves with respect to ScCaH_4 as a function of pressure; (d)-(f) The enthalpy difference curves with respect to ScCaH_6 as a function of pressure; (g)-(i) The enthalpy difference curves with respect to ScCaH_8 as a function of pressure; (j)-(m) The enthalpy difference curves with respect to ScCaH_{10} as a function of pressure; (n)-(r) The enthalpy difference curves with respect to ScCaH_{12} as a function of pressure.



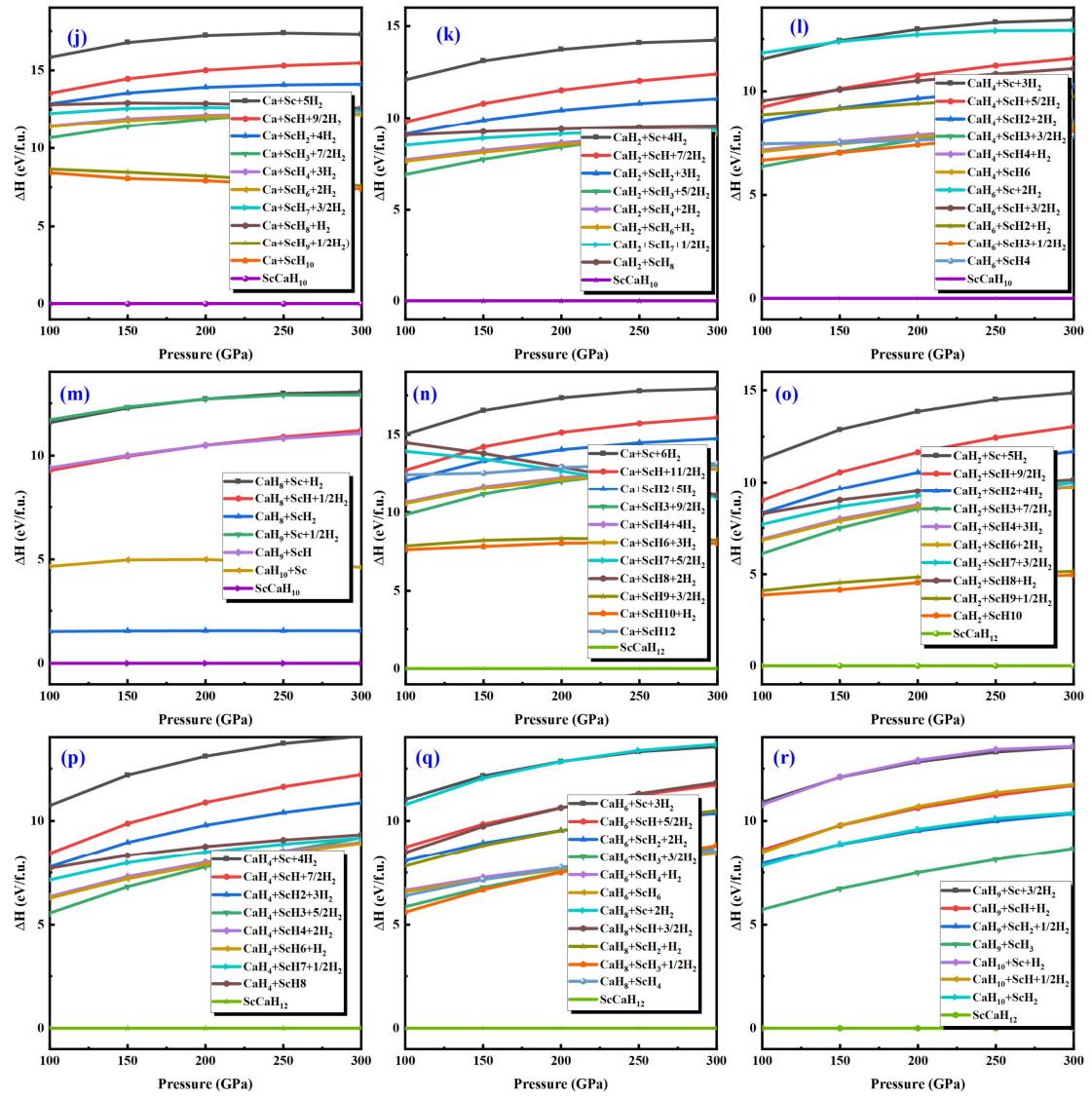


Figure S4. The predicted crystal structures for ScCaH_{2n} ($n=1, 2$). ELF in (-1471.49 -1 -1) sections for $Fd\bar{3}m$ - ScCaH_2 at 150 GPa. ELF in (-1 -1 0) sections for $Fm\bar{3}m$ - ScCaH_2 at 300 GPa. ELF in (1 1 0) sections for $P4/mmm$ - ScCaH_4 at 150 GPa. ELF in (0 0 -1) sections for $P6/mmm$ - ScCaH_4 at 300 GPa. It is obvious that in the two structures of ScCaH_2 , there is no interaction between the hydrogen atoms. In $P4/mmm$ - ScCaH_4 , there are “ H_2 ” structures and obviously interaction between these two hydrogen atoms. In $P6/mmm$ - ScCaH_4 , graphene layered structures are formed between the hydrogen atoms and the interaction between hydrogen atoms are strong.

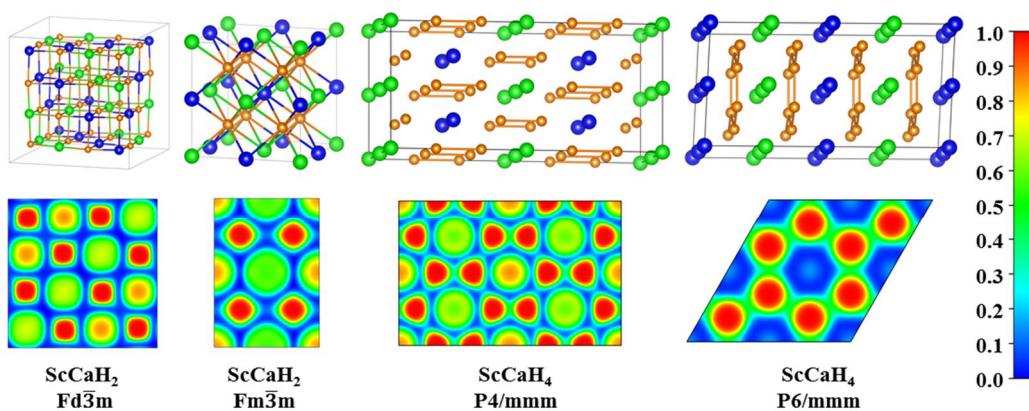


Figure S5. PCOHP for pairs of H...H in ScCaH₈ (a) Two hydrogen atoms connected by **a** bond (b) Two hydrogen atoms connected by **b** bond (c) Two hydrogen atoms connected by **c** bond at different pressures. Energies are normalized to the respective Fermi level. And the PCOHP for pairs of H...H in CaH₆.

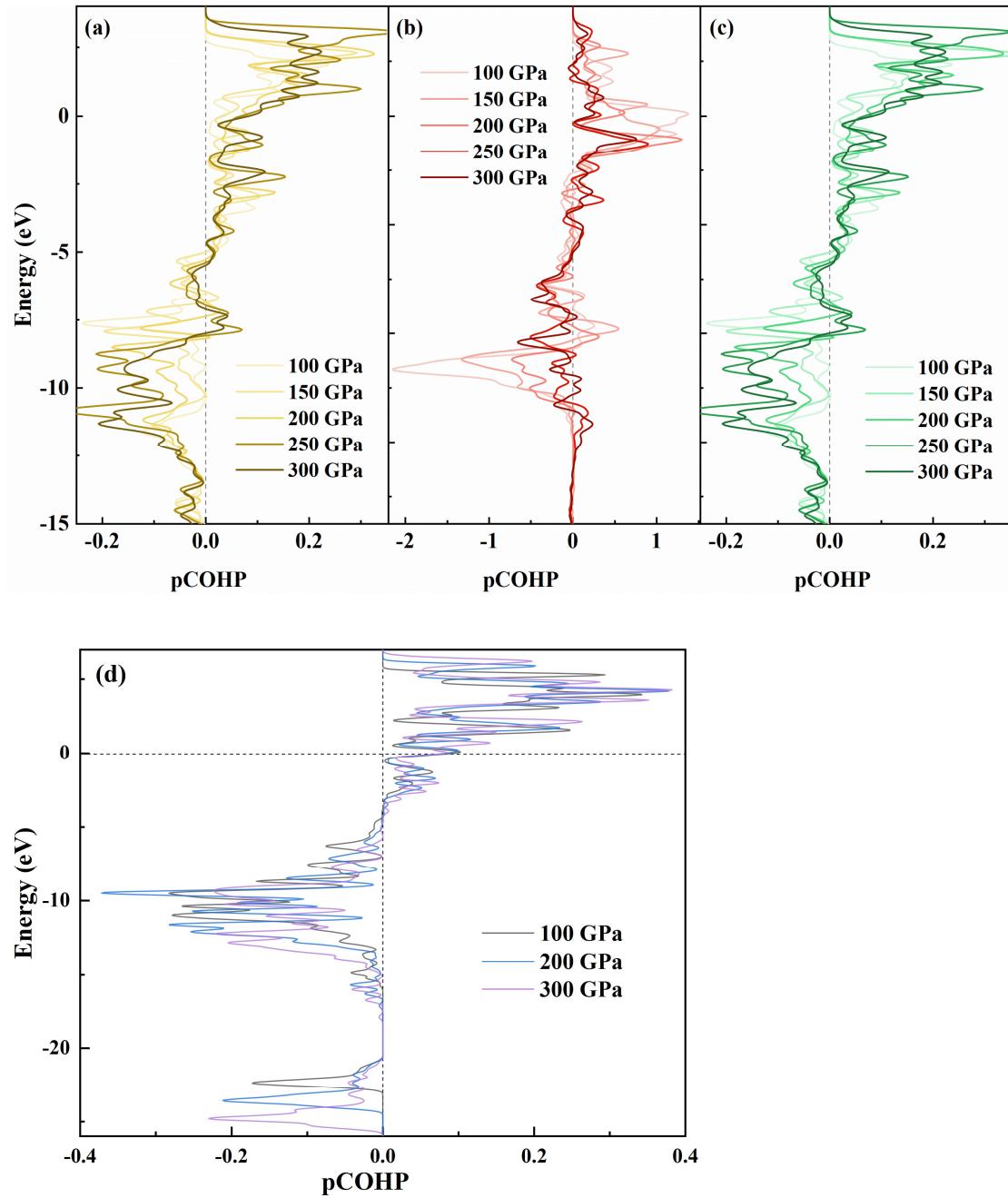


Figure S6. Electronic band structures and density of states of ScCaH_{2n} ($n=1-3,5$). The results show that these compounds are all metallic. In ScCaH_2 and ScCaH_4 , the contribution of hydrogen atom at Fermi level in density of States are less than that of Ca and Sc atoms. In ScCaH_6 and ScCaH_{10} , the contribution of hydrogen atom at Fermi level in the DOS are obviously increased.

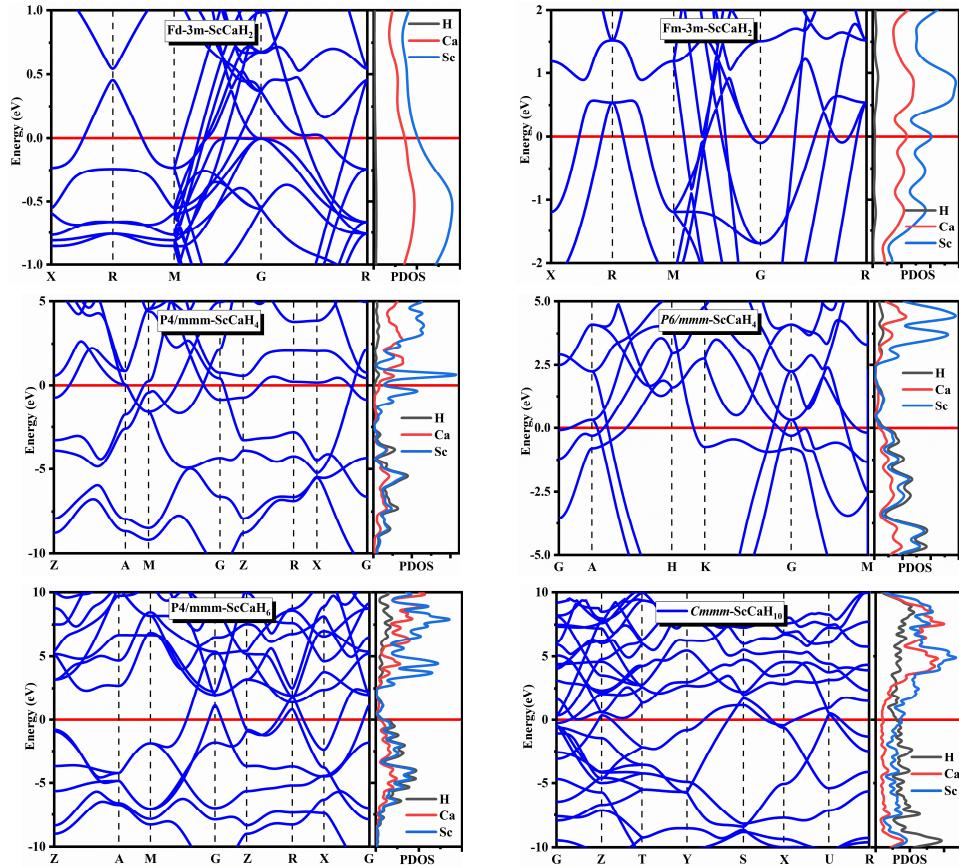


Figure S7. Critical temperature T_c as a function of the empirical parameter μ^* .

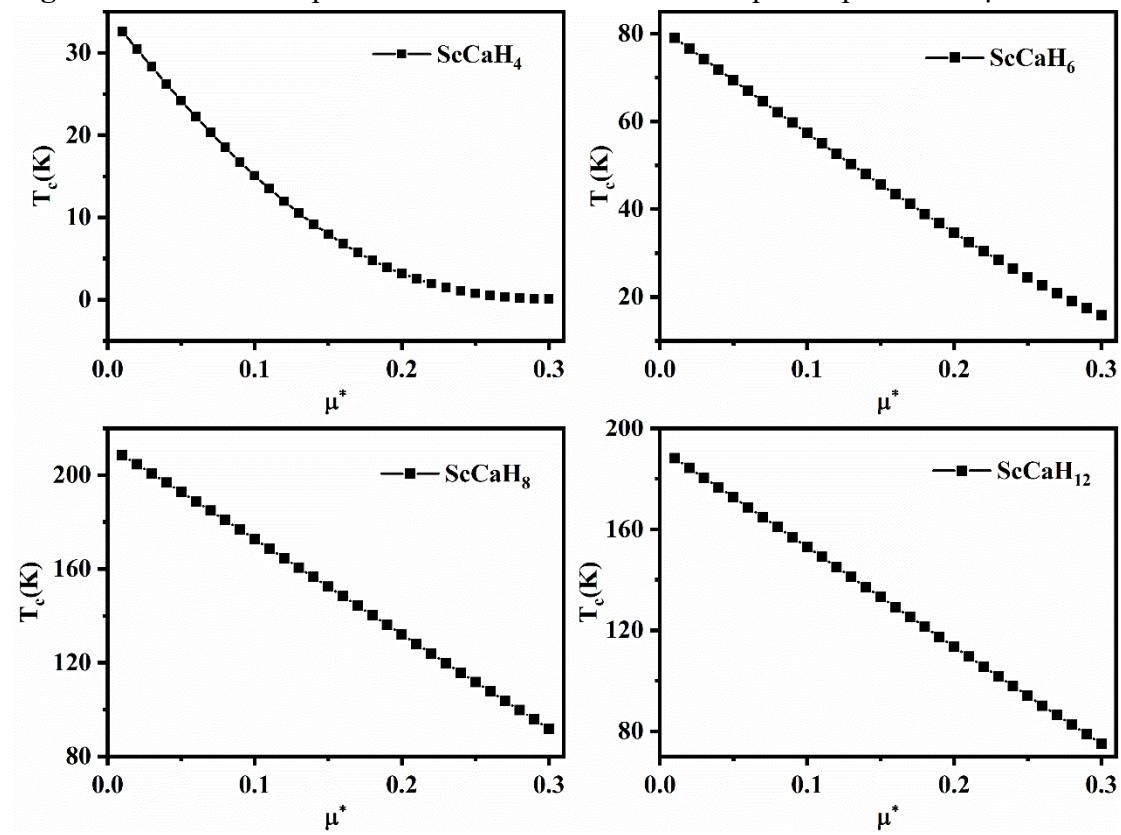


Table S1. Calculated structural parameters of the predicted phase for ScCaH₂

Compound	Space group	Lattice Parameters (Å)	Atomic coordinates (fractional)			
			Atoms	X	Y	Z
ScCaH₂	<i>Fd</i>3<i>m</i>	a=b=c=7.4515 $\alpha=\beta=\gamma=90^\circ$	H(32e) Ca(16d) Sc(16d)	0.11746 0.87500 0.62500	0.88254 0.87500 0.87500	0.11746 -0.37500 0.37500
			H(8c)	0.75000	0.25000	0.25000
			Ca(4b) Sc(4a)	0.00000 0.00000	0.00000 0.00000	0.50000 0.00000
	<i>Fm</i>3<i>m</i>	a=b=c4.2574 $\alpha=\beta=\gamma=90^\circ$		H(2h)	0.50000	0.50000
				H(2g) Ca(1d)	0.00000 0.50000	0.00000 0.50000
				Sc(1a)	0.00000	0.00000
ScCaH₄	<i>P</i>4/<i>mmm</i>	a=b=c=2.38750 $\alpha=\beta=\gamma=90^\circ$		H(4h)	0.33333	0.66667
				Ca(1a) Sc(1b)	0.00000 0.00000	0.00000 0.50000
				H(4i)	0.50000	0.0000
		a=b=2.43200, c=4.41970 $\alpha=\beta=90^\circ, \gamma=120^\circ$		H(1c) H(1b) Ca(1d)	0.50000 0.0000 0.50000	0.0000 0.50000 0.50000
				Sc(1a)	0.0000	0.0000
				H(2g)	0.00000	0.00000
				H(4i) H(2h)	0.50000 0.50000	0.23478 0.36819
				Ca(1b) Sc(1c)	0.00000 0.50000	0.50000 0.00000
				H(8m)	0.75000	0.75000
				H(8o) H(4g)	0.86107 0.37705	0.61550 0.00000
ScCaH₈	<i>P</i>4/<i>mmm</i>	a=b=2.59780, c=4.86800 $\alpha=\beta=\gamma=90^\circ$		Ca(2c) Sc(2a)	0.50000 0.0000	0.50000 0.00000
				H(12h)	-0.00000	0.76123
				Ca(1a) Sc(1b)	0.00000 0.50000	0.00000 0.50000
ScCaH₁₀	<i>C</i>mmm	a=c=4.71340, b=3.36210 $\alpha=\beta=\gamma=90^\circ$				
ScCaH₁₂	<i>P</i>m3<i>m</i>	a=b=c=4.00260 $\alpha=\beta=\gamma=90^\circ$				

Table S2. Integrated COHP (ICOHP) up to the Fermi level for H...H in ScCaH₈.

Pressure	ICOHP (eV/atom pair)			
	a	b	c	CaH₆
100	-0.41772	-1.29912	-0.41699	-1.02540
150	-0.49917	-1.17591	-0.50345	-1.16565
200	-0.61116	-0.89808	-0.62243	-1.25351
250	-0.80201	-0.68686	-0.79061	-1.30817
300	-0.75937	-0.62798	-0.75859	-1.35751

Table S3. The calculated EPC parameter λ , logarithmic average phonon frequency ω_{log} (K), and superconducting transition temperatures T_c at high pressures.

Structure	Pressure (GPa)	λ	ω_{log} (K)	T_c (K) ($\mu^* = 0.1$)
<i>Cmcm</i> -CaH ₄	200	0.46	1398.0	11.6 ^a
<i>I4/mmm</i> -ScH ₄	250	0.81	1891.8	78.0 ^b
<i>P4/mmm</i> -ScCaH ₈	200	1.89	1252.4	212.1 ^c
<i>Im\bar{3}m</i> -CaH ₆	200	1.98	1237.7	211.3 ^d
<i>Im\bar{3}m</i> -ScH ₆	350	1.25	1433.4	169.0 ^b
<i>Pm\bar{3}m</i> -ScCaH ₁₂	200	1.70	1201.1	182.1 ^c

^a Inorg. Chem. 2019, 58, 2558

^b J. Phys. Chem. C 2018, 122, 6298

^c This work

^d J. Phys.: Condens. Matter 2019, 31, 245404