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Supporting information available for

End-functionalization of dithiarubicene: Modulation of optoelectronic properties by metal-catalyzed coupling reactions and device application

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S1. Theoretical calculation¹



Figure S1. HOMO surfaces of dithiarubicene derivatives having the fused five-membered ring (CN-DTR and MeO-DTR, left) and virtual reference compounds having the six-membered ring (CN-DTR' and MeO-DTR', right). The numbers indicate energy levels of HOMO and LUMO (in parenthesis) based on the DFT calculation using a B3LYP density functional and a 6-31G(d,p) basis set.

Table S1. HOMO and LUMO energy levels of five DTR derivatives. ^a						
	Ethynyl- DTR	Th-DTR	Py-DTR	CN-DTR	MeO-DTR	
LUMO (eV)	-2.89	-2.77	-2.98	-3.45	-2.57	
HOMO (eV)	-5.01	-4.87	-5.24	-5.82	-4.74	

[a] Based on the DFT calculation using a B3LYP density functional and a 6-31G(d,p) basis set.

S2. Optoelectronic properties



Figure S2. UV-Vis-NIR absorption spectra of **DTR** (green line), **CN-DTR** (blue line), **Py-DTR** (purple line), **Ethynyl-DTR** (light green line), **Th-DTR** (orange line), and **MeO-DTR** (red line) in CHCl₃ solution (10⁻⁵ M).

Compound	λ_{obsd} [nm]	λ_{calcd} [nm]	Oscillator strength
DTR	615	648	0.12
CN-DTR	587	635	0.21
Py-DTR	631	682	0.28
Ethynyl-DTR	654	720	0.35
Th-DTR	668	733	0.31
MeO-DTR	703	724	0.15

Table S2. Comparison of absorption bands between the observed value and theoretical value estimated by TD-DFT calculation (B3LYP/6-31G(d,p)).¹



Figure S3. UV-Vis-NIR absorption spectra in thin film deposited by spin-coating (dotted lines) and drop-casting (dashed lines). Absorption spectra in solution are depicted in solid lines.



Figure S4. Normalized emission spectra of **DTR** (green line), **CN-DTR** (blue line), and **Py-DTR** (purple line) in CHCl₃ solution (10⁻⁴ M).

Compound	λ_{abs} [nm]	λ_{em} [nm]	
DTR	615	702	
CN-DTR	587	671	
Py-DTR	631	728	

Table S3. Absorption maximum wavelengths andthe fluorescence maximum wavelengths.

		<u>^</u>	
Compound	$\frac{E_{\rm red(1st)}}{\rm [V]}^{1/2}$	$\frac{E_{\rm red(2nd)}^{1/2}}{\rm [V]}$	ΔE [V]
DTR	-1.56	-2.04	0.48
CN-DTR	-1.24	-1.60	0.36
Py-DTR	-1.42	-1.76	0.34
Ethynyl-DTR	-1.41	-1.81	0.40
Th-DTR	-1.45	-1.88	0.43
MeO-DTR	-1.59	-2.09	0.50

 Table S4. Reduction potentials.

Table S5. Comparison of electrochemical properties between the observed value and theoretical value estimated by DFT calculation (B3LYP/6-31G(d,p)).¹

Compound	$E_{ m HOMO}^{ m obsd}$ [eV]	$E_{ m HOMO}^{ m calcd}$ [eV]	$E_{ m LUMO}^{ m obsd}$ [eV]	$E_{ m LUMO}^{ m calcd}$ [eV]	$E_{ m g}^{ m obsd}$ [eV]	$E_{ m g}^{ m calcd}$ [eV]
DTR	(-5.07)	-5.04	-3.24	-2.66	(1.83)	2.38
CN-DTR	-5.82	-5.82	-3.56	-3.45	2.26	2.37
Py-DTR	-5.37	-5.24	-3.38	-2.98	1.99	2.26
Ethynyl-DTR	(-5.05)	-5.01	-3.39	-2.89	(1.66)	2.12
Th-DTR	-5.06	-4.87	-3.35	-2.77	1.71	2.1
MeO-DTR	-4.77	-4.74	-3.21	-2.57	1.56	2.17





Figure S5. Transfer plots of (A) **OD-MeO-DTR** and (B) **OD-CN-DTR**-based OFET device fabricated on HMDS-treated SiO₂ insulation surface in an as-cast film.



Figure S6. Output plots of (A) and (A') **OD-CN-DTR** [annealed at 100 °C], (B) **OD-MeO-DTR** [as-cast], and (C) **EH-MeO-DTR** [annealed at 100 °C]-based OFET device fabricated on HMDS-treated SiO₂ insulation surface.



Figure S7. AMF image (5 μ m × 5 μ m) of the films fabricated on HMDS-treated SiO₂ insulation surface. The inset values indicate the root-mean-square roughness (R_{RMS}).



Figure S8. Transfer (left) and output (right) curves of EH-MeO-DTR-based OFET device fabricated on HMDS-treated SiO_2 insulation surface using Au electrodes modified with 1-hexadecanethiol.

S4. OPV characteristics

Table S6. OPV characteristic with the device architecture ITO/ZnO/P3HT:OD-CN-DTR/MoO₃/Ag.

solv.	conc.	PCE	$J_{ m sc}$	V _{oc}	FF
	[mg/mL]	[%]	[mA/cm ²]	[V]	[%]
chlorobenzene	20	0.21	0.84	0.56	44
chloroform	10	0.10	0.59	0.40	43



Figure S9. *J-V* curve for ITO/ZnO/P3HT:**OD-CN-DTR**/MoO₃/Ag device. The active layer was fabricated from chlorobenzene (left) and chloroform (right) solutions.



Figure S10. EQE spectrum for ITO/ZnO/P3HT:**OD-CN-DTR**/MoO₃/Ag device, in which the active layer was fabricated from chlorobenzene solution.



Figure S11. AFM image of the active layer.



EH-Py-DTR



EH-Ethynyl-DTR



EH-Th-DTR



OD-MeO-DTR



EH-MeO-DTR



S6. Reference

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