## All-inorganic tin-doped Cs<sub>2</sub>BiAgCl<sub>6</sub> double perovskite with stability blue photoluminescence WLEDs

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Figure S1. PXRD patterns of the Sn-doped and pristine Cs<sub>2</sub>BiAgCl<sub>6</sub>.



Figure S2. The Rietveld refinement of the host and representative  $Cs_2BiAgCl_6:0.01Sn^{2+}$ .

x	a=b=c (Å)	V (Å <sup>3</sup> )	R <sub>wp</sub> (%)	$R_p$ (%)
standard	10.777	1251.68		
0	10.771	1249.592	4.63	3.76
0.1	10.784	1254.240	5.76	4.53

Table S1. Main parameters of processing and refinement of the  $Cs_2BiAgCl_6:xSn^{2+}$  samples.

Bond	Length (Å)	Optr. Cell	Neighbor atom coordinates			
		x = 0				
Cl-Cs	3.81403(4)	9000	0.25000 -0.25000 -0.25000			
Cl-Bi	2.57586(4)	1000	0.00000 0.00000 0.00000			
Cl-Ag	2.81533(4)	101 0-1-1	0.50000 0.00000 0.00000			
<i>x</i> = 0.1						
Cl-Cs	3.81304(7)	9000	0.25000 -0.25000 -0.25000			
Cl-Bi	2.65677(7)	1000	0.00000 0.00000 0.00000			
Cl-Ag	2.73541(8)	101 0-1-1	0.50000 0.00000 0.00000			

Table S2. Selected interatomic distances in samples.

x = 0	frac.	x	У	Z	100*U <sub>iso</sub>
$Cs^+$	1.015	0.25	0.25	0.25	3.983
Bi <sup>3+</sup>	0.983	0.00	0.00	0.00	2.107
$Ag^+$	1.016	0.50	0.50	0.50	2.413
Cl-	1.000	0.24	0.00	0.00	3.709
<i>x</i> = 0.1	frac.	x	у	Z	100*U <sub>iso</sub>
x = 0.1 Cs <sup>+</sup>	frac.	<i>x</i> 0.25	<i>y</i> 0.25	<i>z</i> 0.25	100*U <sub>iso</sub> 4.025
$x = 0.1$ $Cs^+$ $Bi^{3+}$	frac. 1.025 1.051	x 0.25 0.00	<i>y</i> 0.25 0.00	z 0.25 0.00	100*U <sub>iso</sub> 4.025 2.257
$x = 0.1$ $Cs^+$ $Bi^{3+}$ $Ag^+$	frac. 1.025 1.051 1.152	x 0.25 0.00 0.50	<i>y</i> 0.25 0.00 0.50	z 0.25 0.00 0.50	100*U <sub>iso</sub> 4.025 2.257 2.543

Table S3. Fractional atomic coordinates and isotropic displacement parameters in samples.



Figure S3. XPS spectra corresponding to (a) raw total, (b) raw Cs 3d, (c) raw Bi 4f, (d) raw Ag 3d, (e) raw Cl 2p, (f) raw Sn 3d, (g)  $O_2$  treated total, (h)  $O_2$  treated Cs 3d, (i)  $O_2$  treated Bi 4f, (j)  $O_2$  treated Ag 3d, (k)  $O_2$  treated Cl 2p, (l)  $O_2$  treated Sn 3d.



Figure S4. PL for Cs<sub>2</sub>BiAgCl<sub>6</sub>:0.01Sn<sup>2+</sup>.



Figure S5. (a) TL for  $Cs_2BiAgCl_6:0.1Sn^{2+}$ , (b) Differential scanning calorimetry (DSC) curves of pristine  $Cs_2BiAgCl_6:xSn^{2+}$  in heating progress.



Figure S6. (a) XRD of pristine and  $Cs_2BiAgCl_6:0.2Sn^{2+}$  after being exposed to air for 100 days, (b) PL intensity of pristine and  $Cs_2BiAgCl_6:0.1Sn^{2+}$  after being exposed to air for 100 days.