

## Supporting Information

### **Light-Driven Autonomous Self-Oscillation of Liquid-Crystalline Polymer Bimorph Actuator**

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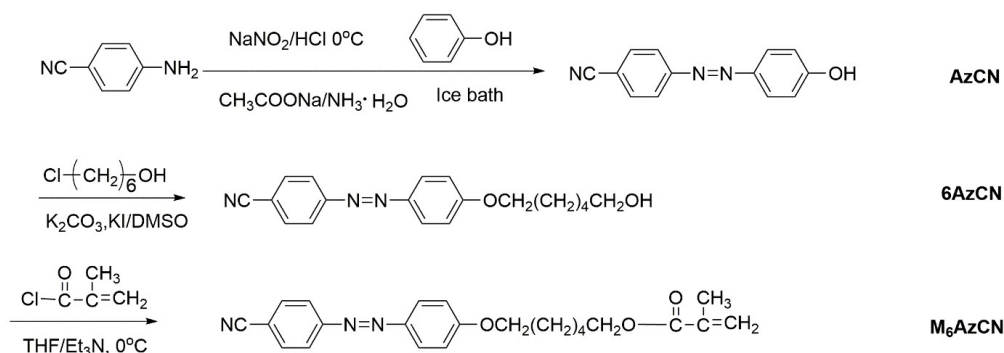
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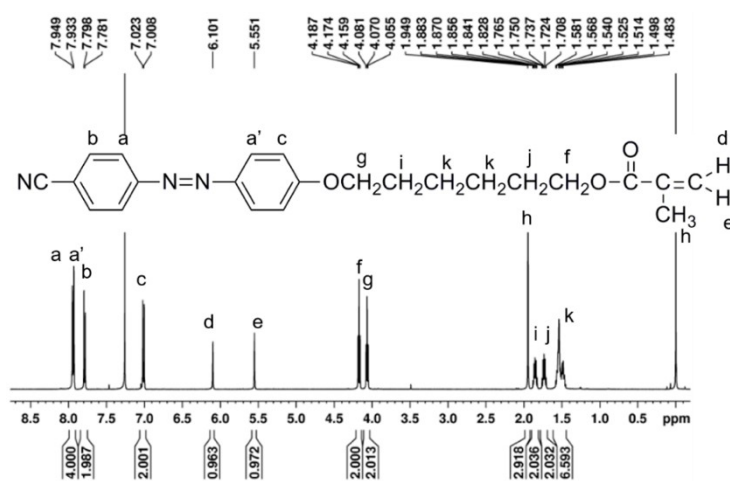
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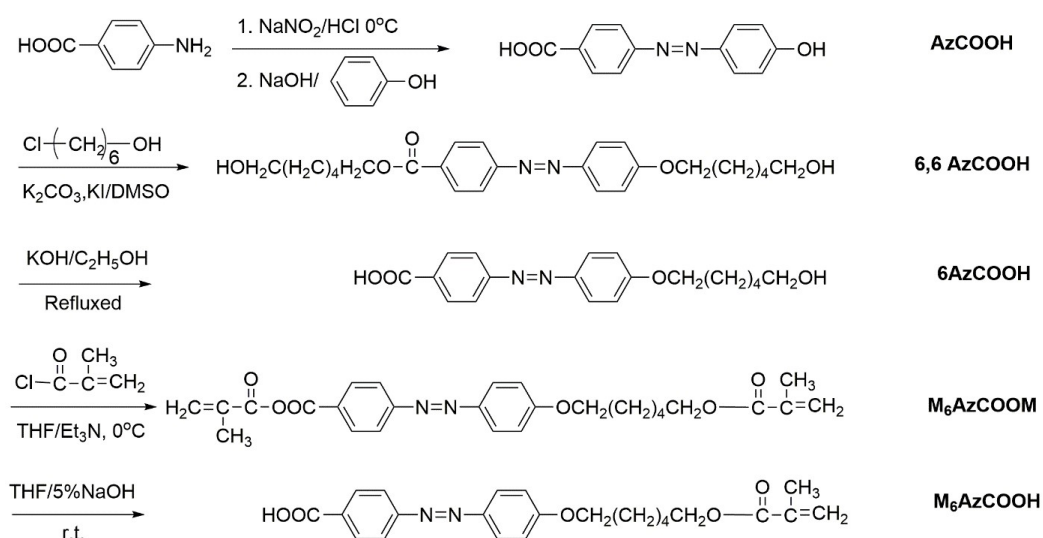


**Scheme S1.** Synthesis of the azobenzene-containing monomer M<sub>6</sub>AzCN.

The monomer M<sub>6</sub>AzCN was synthesized by the diazo-coupling reaction between 4-cyanoaniline and phenol in the presence of sodium nitrite and hydrochloric acid and the following reaction with 6-chloro-1-hexanol and methacryloyl chloride.<sup>[S1]</sup> The <sup>1</sup>H NMR spectrum of M<sub>6</sub>AzCN was recorded in CDCl<sub>3</sub> solution, as shown in Scheme S1. <sup>1</sup>H NMR (500MHz, CDCl<sub>3</sub>, 25 °C, TMS): δ 7.94-7.93 (4H, d), δ 7.78-7.80 (2H, d), δ 7.0-7.02 (2H, d), δ 6.10 (1H, s), δ 5.55 (1H, s), δ 4.15-4.18 (2H, t) δ 4.05-4.08 (2H, t), δ 1.94 (3H, s), δ 1.85 (2H, m), δ 1.76 (2H, m), δ 1.4-1.6 (4H, m).

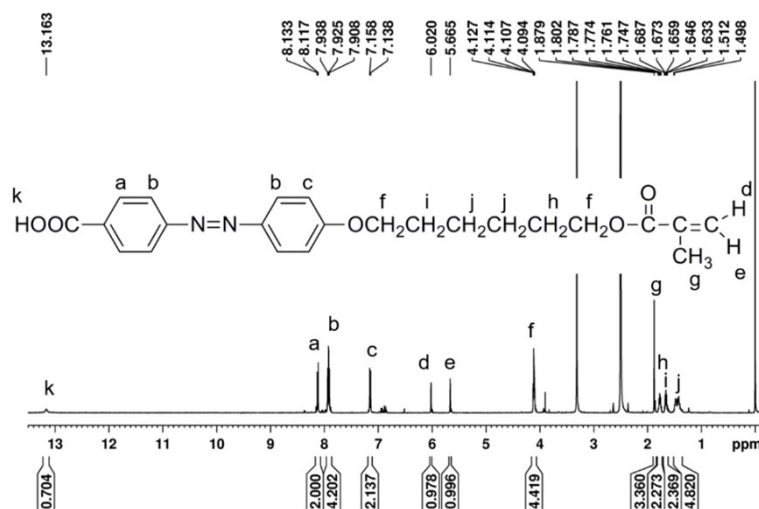


**Fig. S1.** <sup>1</sup>H NMR spectrum of the monomer M<sub>6</sub>AzCN



**Scheme S2.** Synthesis of the monomer M<sub>6</sub>AzCOOH.

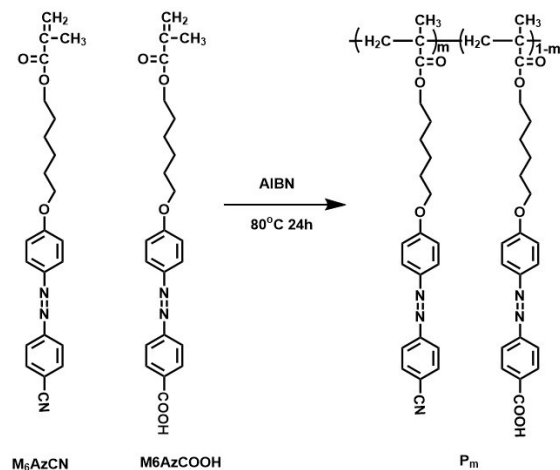
The monomer **6-(4-((4-Carboxylphenyl)diazenyl)-phenoxy) hexyl methacrylate (M<sub>6</sub>AzCOOH)** was synthesized according to the literature reported.<sup>[S2]</sup> The <sup>1</sup>H NMR spectrum of M<sub>6</sub>AzCOOH was recorded in DMSO-d<sub>6</sub> solution, as shown in Fig. S2. <sup>1</sup>H NMR (500MHz, DMSO-d<sub>6</sub>, 25 °C, TMS): δ 13.16 (1H, m), δ 8.11-8.13 (2H, d), δ 7.92-7.93 (4H, d), δ 7.13-7.15 (2H, d), δ 6.0 (1H, s), δ 5.66 (1H, s), δ 4.09-4.12 (4H, t), δ 1.87 (3H, s), δ 1.77 (2H, m), δ 1.65 (2H, m), δ 1.35-1.5 (4H, m).



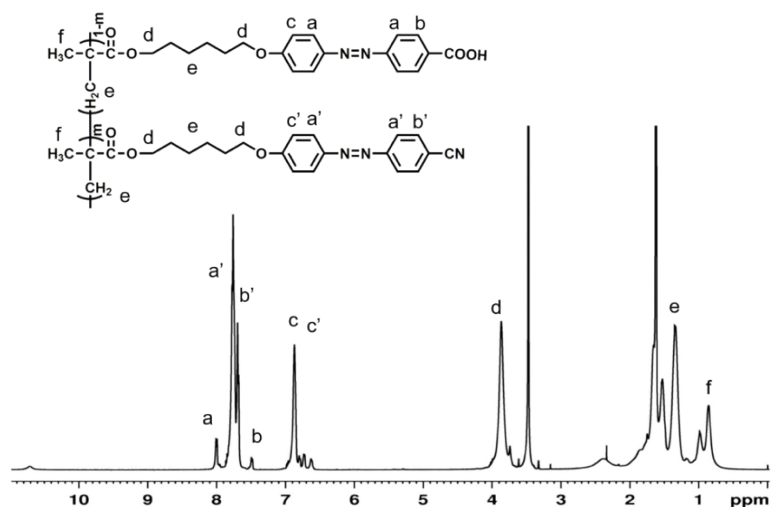
**Fig. S2.** <sup>1</sup>H NMR spectrum of the monomer M<sub>6</sub>AzCOOH.

**Copolymer P<sub>0.9</sub>.** The monomers M<sub>6</sub>AzCN (400mg, 1mmol) and M<sub>6</sub>AzCOOH (100 mg, 0.26 mmol) and AIBN (1wt%, 5 mg) were dissolved in freshly distilled DMSO (3 mL). After several freeze-pump-thaw cycles, the mixture was then placed in an oil bath preheated to 80 °C and stirred for over 24 h. The resulting solution was poured into 500 mL of cooled

methanol to obtain a crude copolymer. The copolymer was purified by reprecipitation in methanol. An orange powder was obtained after the precipitate was dried in a vacuum oven for 24 h. Yield: 60%.

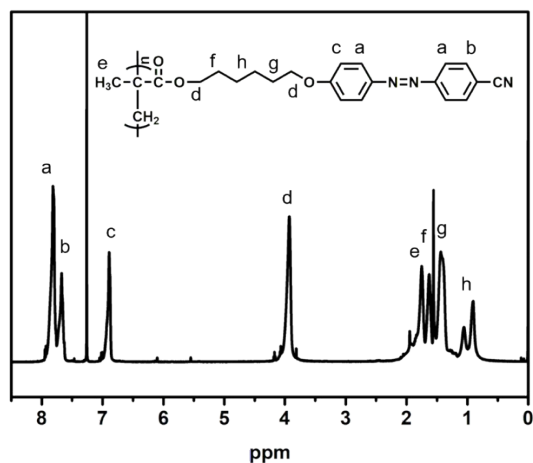


**Scheme S3.** Synthesis of the polymers.

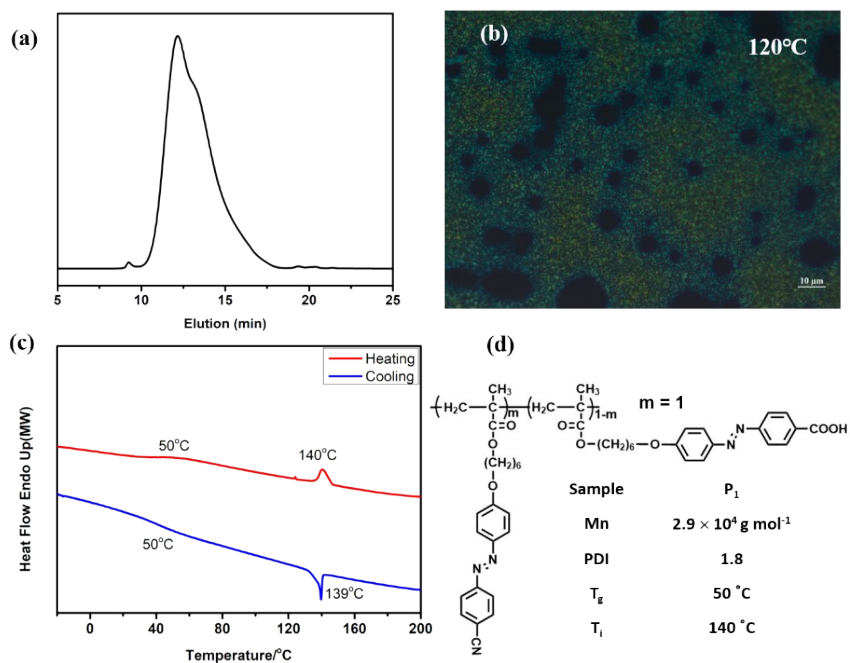


**Fig. S4.**  $^1\text{H}$  NMR spectrum of  $\text{P}_{0.9}$ .

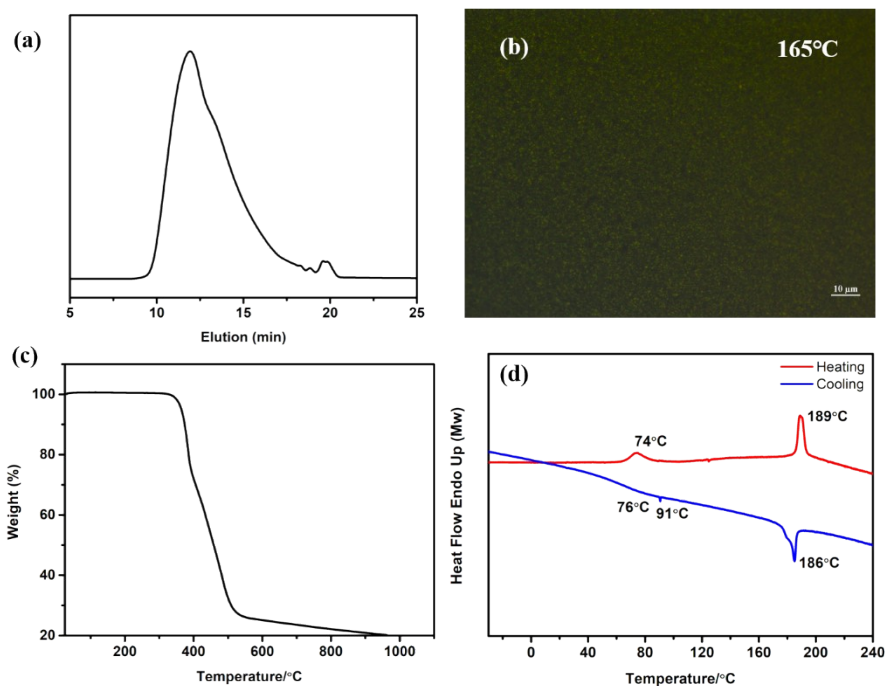
**Homopolymer  $\text{P}_1$ .** The monomers  $\text{M}_6\text{AZCN}$  (250 mg, 0.64mmol) and AIBN (1wt%, 3mg) were dissolved in freshly distilled anisole (1 mL). After several freeze-pump-thaw cycles, the mixture was then placed in an oil bath preheated to  $65^\circ\text{C}$  and stirred for over 24 h. The resulting solution was poured into 500 mL of cooled methanol to obtain a crude homopolymer. The homopolymer was purified by reprecipitation in methanol. An orange powder was obtained after the precipitate was dried in a vacuum oven for 24 h. Yield: 70%.



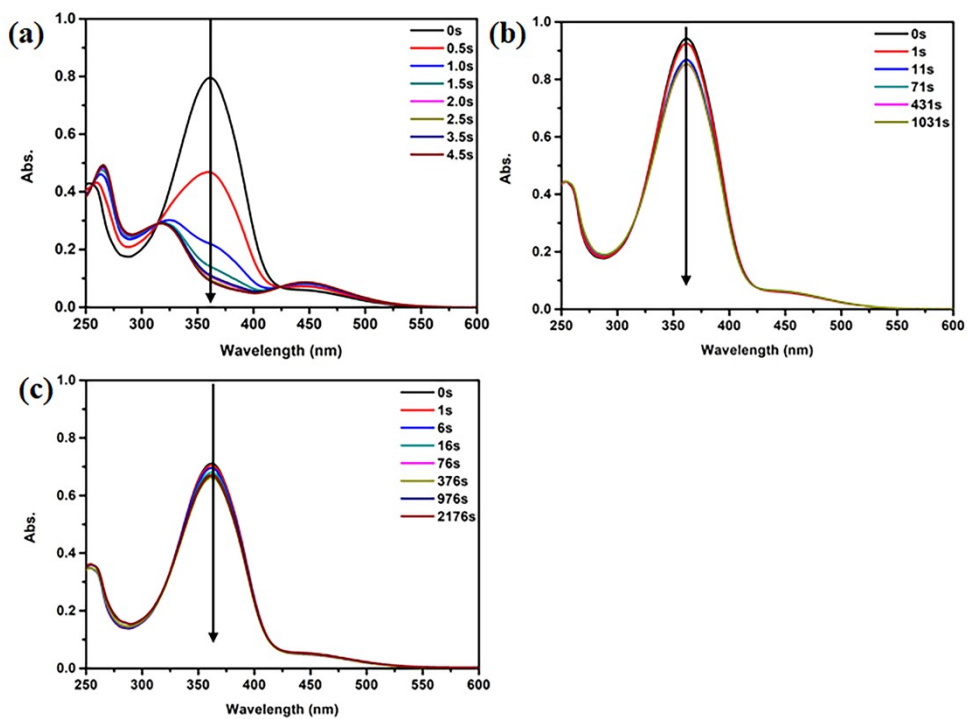
**Fig. S5.**  $^1\text{H}$  NMR spectrum of  $\text{P}_1$ .



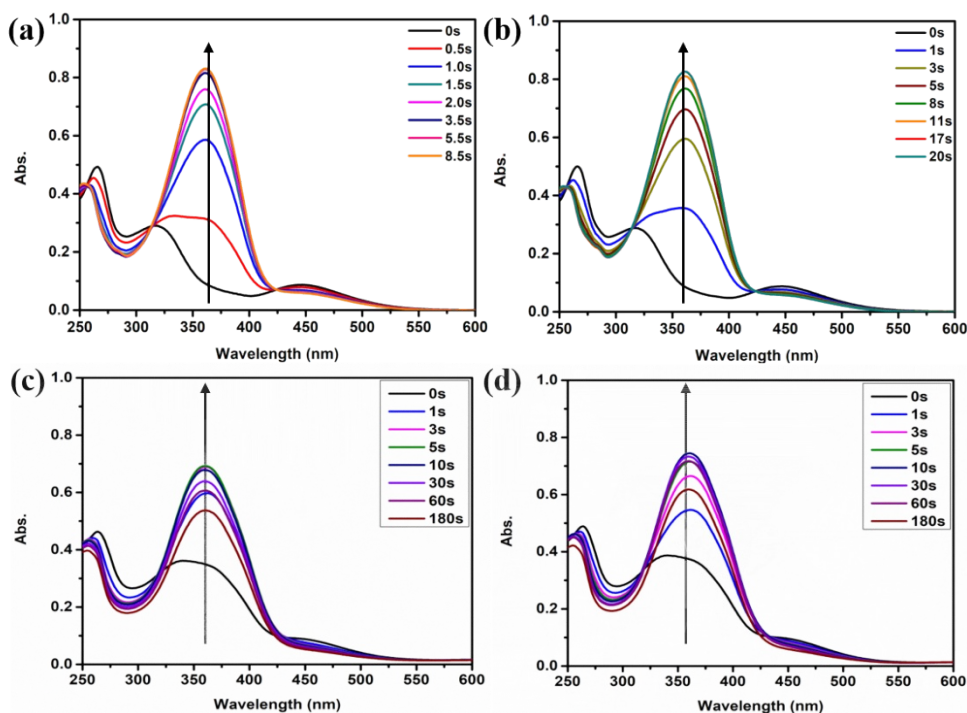
**Fig. S6.** (a) GPC curve of  $\text{P}_1$ , (b) POM picture of  $\text{P}_1$ , (c) DSC curves of  $\text{P}_1$ , (d) Chemical structure and thermal properties of  $\text{P}_1$ .



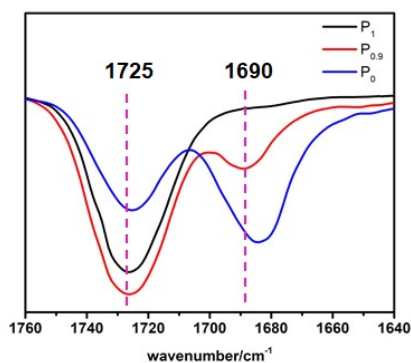
**Fig. S7.** (a) GPC curve of  $P_{0.9}$ , (b) POM picture of  $P_{0.9}$ , (c) TGA curve of  $P_{0.9}$ , (d) DSC curves of  $P_{0.9}$ .



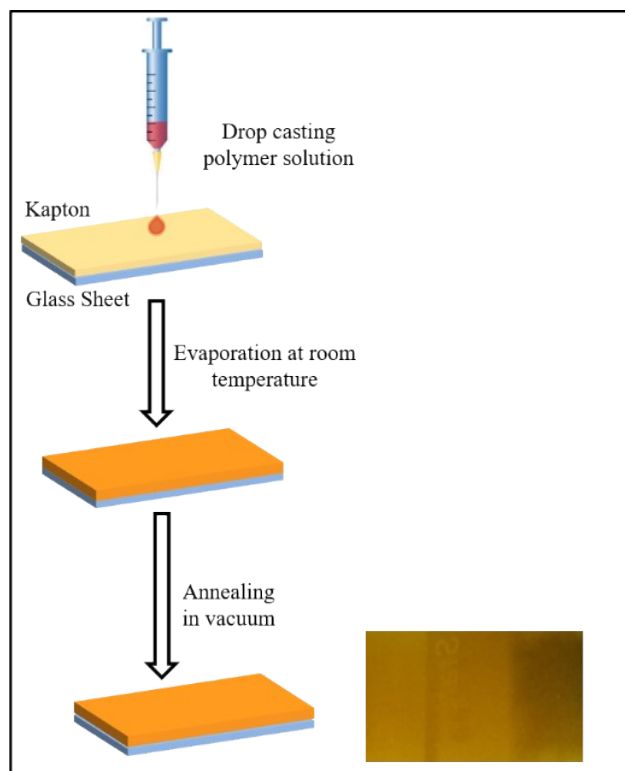
**Fig. S8.** UV-vis spectrum of  $P_{0.9}$  in tetrahydrofuran (THF) irradiated by 365 nm (a), 460 nm (b), 530 nm (c) light at  $100 \text{ mW cm}^{-2}$ .



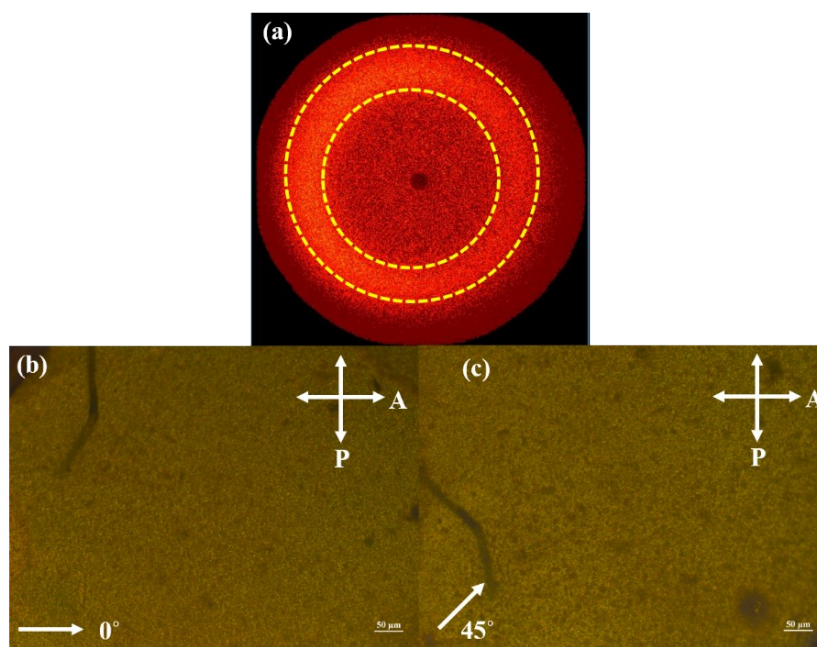
**Fig. S9.** UV-vis spectrum for reversion process of  $P_{0.9}$  irradiated by 460 nm (a), 530 nm (b) light in tetrahydrofuran (THF) and 460 nm (c), 530 nm (d) light in film state at  $100 \text{ mW cm}^{-2}$  after 365 nm light irradiation.



**Fig. S10.** FTIR spectrum of the homopolymer  $P_1$  (polymerized by monomer  $M_6\text{AzCN}$ ), copolymer  $P_{0.9}$ , homopolymer  $P_0$  (polymerized by monomer  $M_6\text{AzCOOH}$ ).

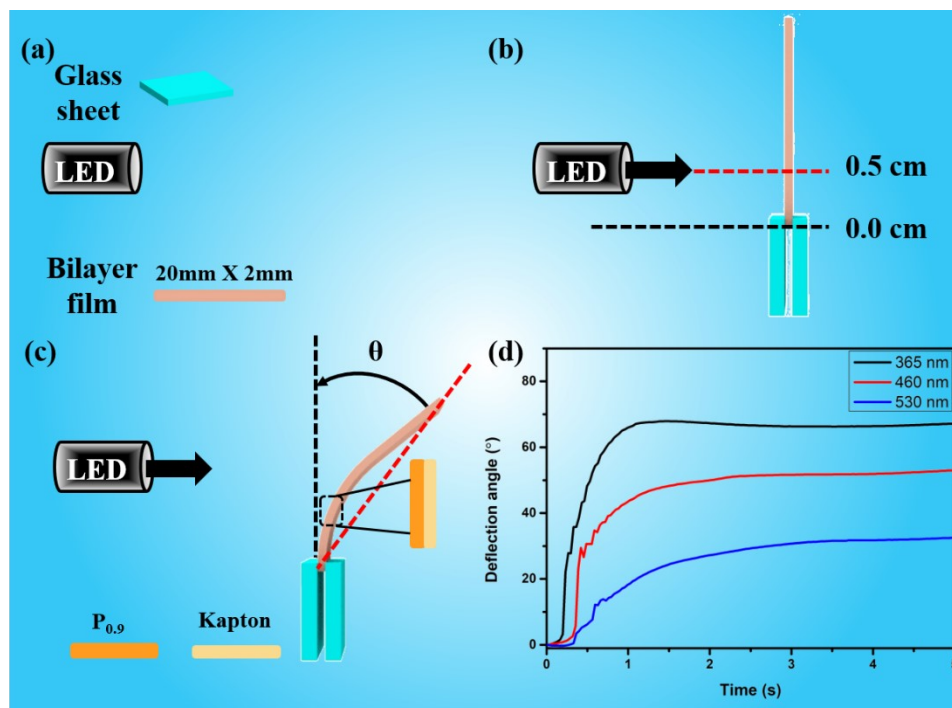


**Scheme S4.** Fabrication process of the bilayer film.

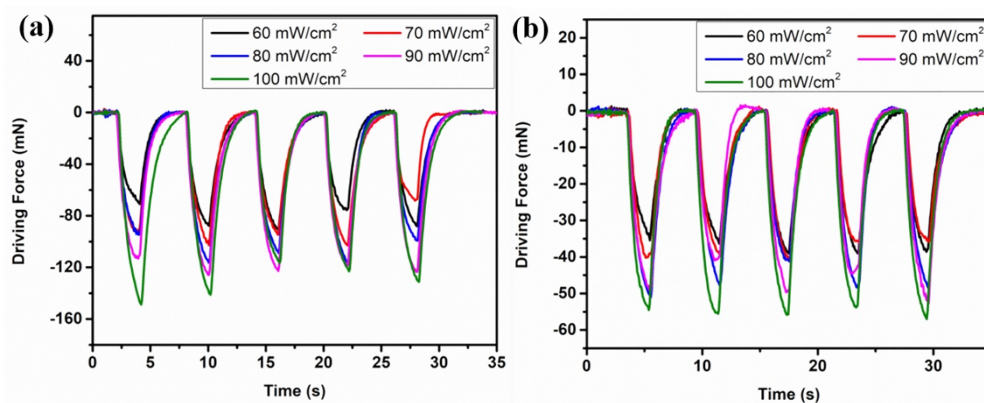


**Fig. S11.** (a) 2D XRD results for the bilayer film containing  $P_{0.9}$ . POM pictures (b) and (c) for the bilayer films containing  $P_{0.9}$  after annealing. White solid arrows represent the transmission axes of the polarizer (P) and analyzer (A), the sample angle to the analyzer:  $\theta = 0^\circ$ ,  $\theta = 45^\circ$ , respectively.

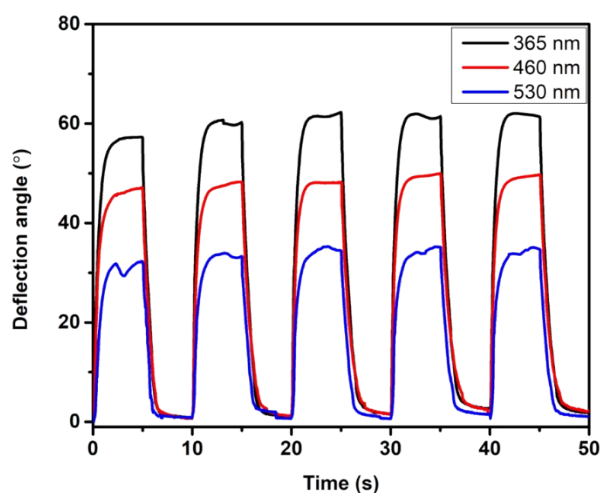




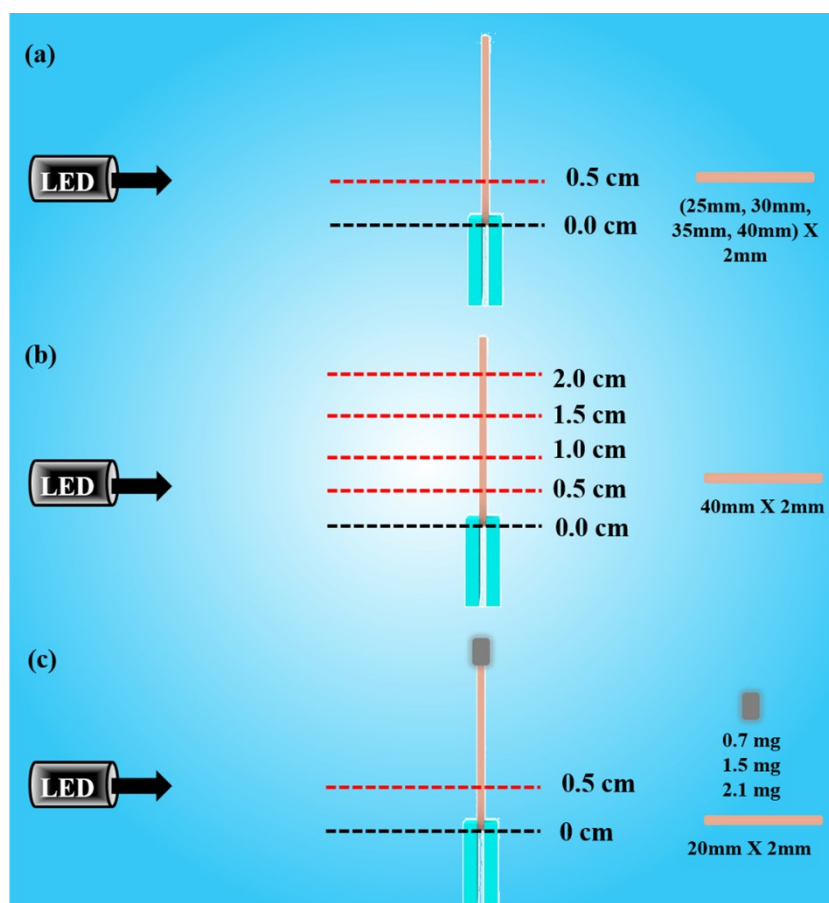
**Fig. S12.** Materials (a) and scheme (b) of optical setup for photomechanical experiments. (c) Light shined on the photoactive polymer side. (d) Photomechanical behavior for the bilayer strip containing the copolymer P<sub>0.9</sub> illuminated respectively by 365 nm, 460 nm and 530 nm light at 100 mW cm<sup>-2</sup> from the AZ-containing polymer side.



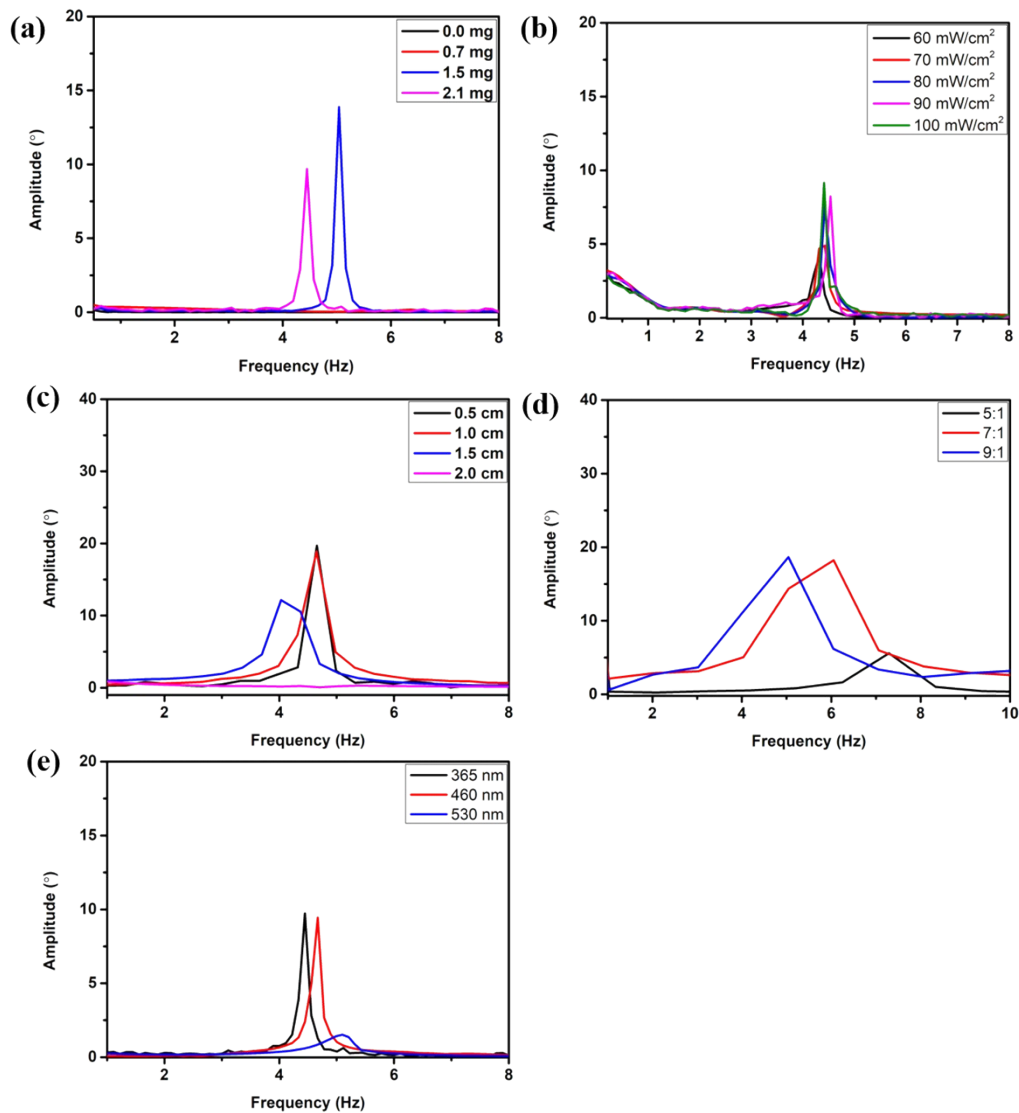
**Fig. S13.** Driven force of the bilayer films containing P<sub>0.9</sub> (a) and P<sub>1</sub> (b) under 365 nm light irradiation, which was measured in-situ with a test machine.



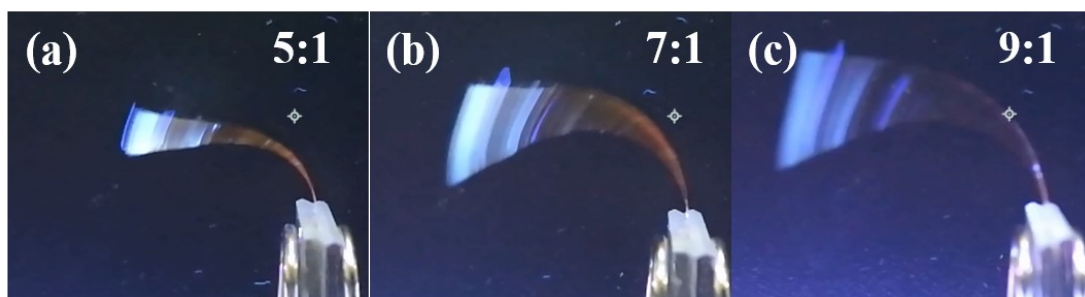
**Fig. S14.** The photomechanical behaviour of the bilayer strip as turning on and off the 365 nm, 460 nm and 530 nm light at  $100 \text{ mW cm}^{-2}$ , respectively.



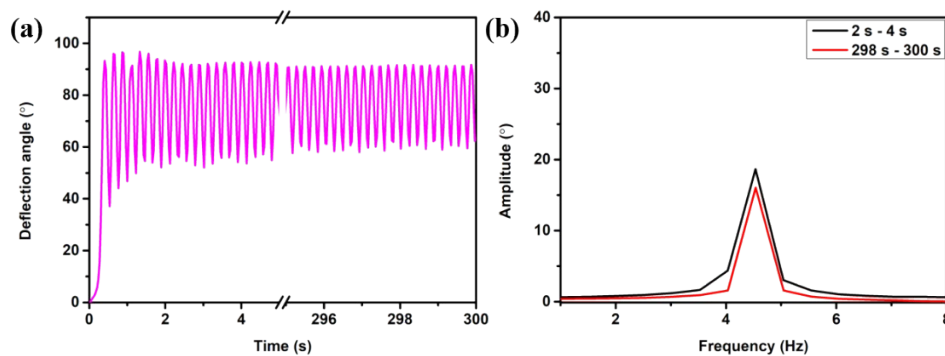
**Scheme S5.** The photomechanical behaviour of the bilayer strip was modulated by changing the length of the strip (a), the light-irradiated position (b), and the attached different amounts of scotch tape on the end of the actuator (c), respectively, when other conditions are fixed.



**Fig. S15.** The frequency and amplitude of the self-oscillator with (a) different amounts of attached scotch tape, (b) different light intensities of 365 nm light, (c) different light-irradiated position under 365 nm light at 100 mW cm<sup>-2</sup>, (d) different aspect ratios of film, (e) actinic light at 365 nm, 460 nm and 530 nm light (100 mW cm<sup>-2</sup>).



**Fig. S16.** Snapshots of displacement of self-oscillators with different aspect ratios.



**Fig. S17.** (a) Self-sustained oscillation of the oscillator under 365 nm light irradiation at 100 mW cm<sup>-2</sup>. (b) The oscillating frequency and amplitude for the strip during 2s-4s and 298s-300s.

### Supporting Movies

**Movie S1.** Photomechanical behavior of the P<sub>0,9</sub>/Kapton bilayer strip with 365 nm, 460 nm or 530 nm light illumination at 100 mW cm<sup>-2</sup> from both sides (MP4)

**Movie S2.** Photomechanical behavior of the P<sub>1</sub>/Kapton bilayer strip with 365 nm light illumination at 100 mW cm<sup>-2</sup> from both sides (MP4).

**Movie S3.** The P<sub>0,9</sub>/Kapton bilayer strip floated upward during 365 nm, 460 nm or 530 nm light irradiation and sunk after removal of light (MP4).

**Movie S4.** The self-oscillating behavior of the P<sub>0,9</sub>/Kapton bilayer strip with 365 nm light illumination at 100 mW cm<sup>-2</sup> (MP4).

**Movie S5.** The self-oscillating behavior of the P<sub>0,9</sub>/Kapton bilayer strip with 460 nm light illumination at 100 mW cm<sup>-2</sup> (MP4).

**Movie S6.** The self-oscillating behavior of the P<sub>0,9</sub>/Kapton bilayer strip with 530 nm light illumination at 100 mW cm<sup>-2</sup> (MP4).

**Movie S7.** The self-oscillating behavior of the P<sub>0,9</sub>/Kapton bilayer strip with 365 nm light illumination at 100 mW cm<sup>-2</sup>, under the condition that both of the capsule and the actuator uncharged, just the capsule charged or both of the capsule and the actuator charged (MP4).

**Movie S8.** The self-oscillator of the P<sub>0,9</sub>/Kapton bilayer strip for signal transmission under 365 nm light illumination at 100 mW cm<sup>-2</sup> (MP4).

### Supporting References

- S1 H. Yu, T. Iyoda, T. Ikeda, *J. Am. Chem. Soc.*, 2006, **128**, 11010-11011.  
 S2 H. Ren, D. Chen, Y. Shi, H. Yu, Z. Fu, *Polym. Chem.*, 2015, **6**, 270-277.