

Supplementary Information

Synthesis of Hollow Rod-like Hierarchical Structures Assembled by CoFe/C Nanosheets for Enhanced Microwave Absorption

Susu Bao,^a Zhijia Song,^a Runjing Mao,^a Yue Li,^b Shuhong Zhang,^a Zhiyuan Jiang,^{*,a}

Xueai Li,^{*,b} and Zhaoxiong Xie^a

^a State Key Laboratory of Physical Chemistry of Solid Surfaces, and College of Chemistry and Chemical Engineering, Xiamen University, Xiamen 361005, China. Tel: +86-5922182432, Fax: +86-5922183047.

^b School of Environmental and Chemical Engineering, Yanshan University, Qinhuangdao, 066004, Hebei, P. R. China.

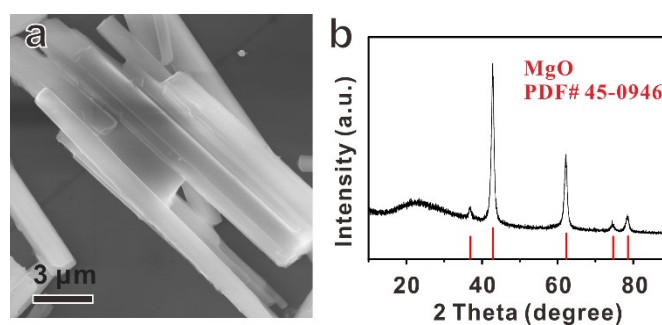


Figure S1. (a) SEM image and (b) XRD pattern of MgO micro-rods.

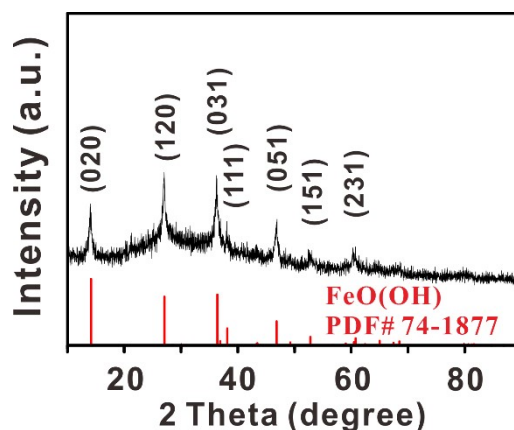


Figure S2. XRD pattern of as prepared $\text{Co}_{0.5}\text{Fe}_{0.5}\text{O}(\text{OH})$ precursors.

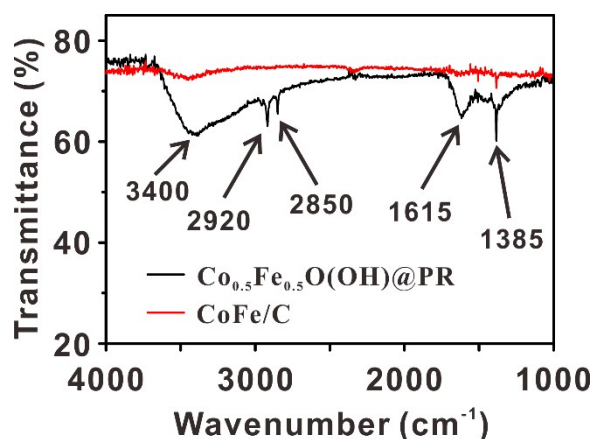


Figure S3. FT-IR spectra of $\text{Co}_{0.5}\text{Fe}_{0.5}\text{O}(\text{OH})@\text{PR}$ and CoFe/C . For CoFe/C , the peak around 3400 cm^{-1} is O–H stretching band, the peak around 1385 cm^{-1} is O–H in-plane stretching band. For $\text{Co}_{0.5}\text{Fe}_{0.5}\text{O}(\text{OH})@\text{PR}$, the peaks at 2920 , 2850 and 1615 cm^{-1} belong to phenolic resin, which can be identified as $-\text{CH}_2-$ in-phase stretching vibrations, $-\text{CH}_2-$ out-of-phase stretching vibrations, and the aromatic ring stretching band, respectively.

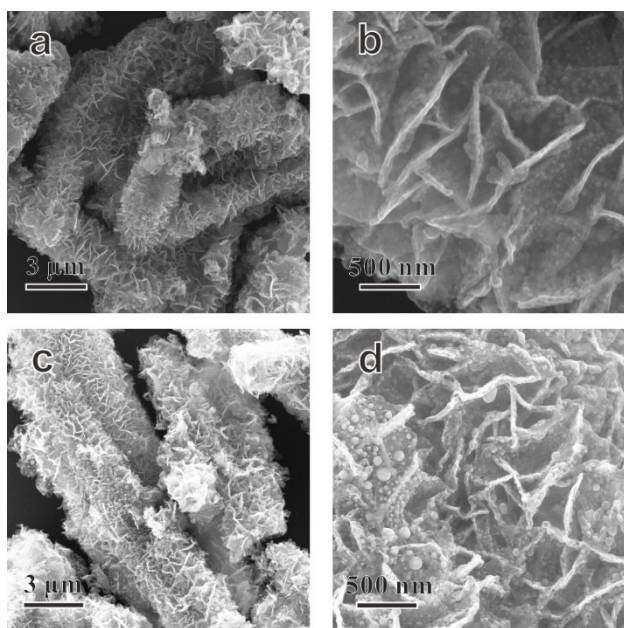


Figure S4. (a), (b) SEM images of $\text{CoFe}/\text{C}-\text{M}$ and (c), (d) SEM images of $\text{CoFe}/\text{C}-\text{L}$.

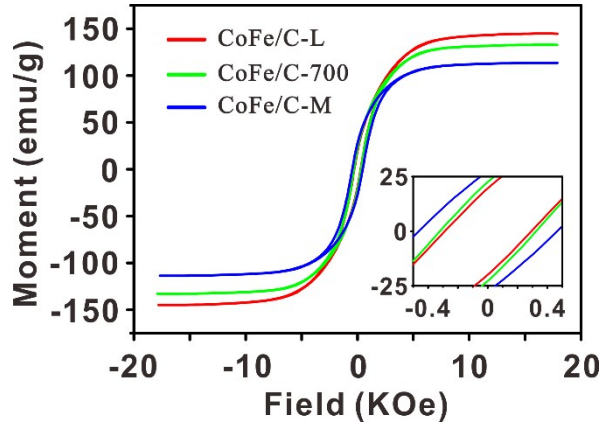


Figure S5. Room-temperature magnetization curve of CoFe/C hierarchical structures. For CoFe/C-L, the M_s value is 144.8 emu/g, H_c is 293.3 Oe. For CoFe/C-700, the M_s value is 132.8 emu/g, H_c is 324.5 Oe. For CoFe/C-M, the M_s value is 113.6 emu/g, H_c is 464.5 Oe.

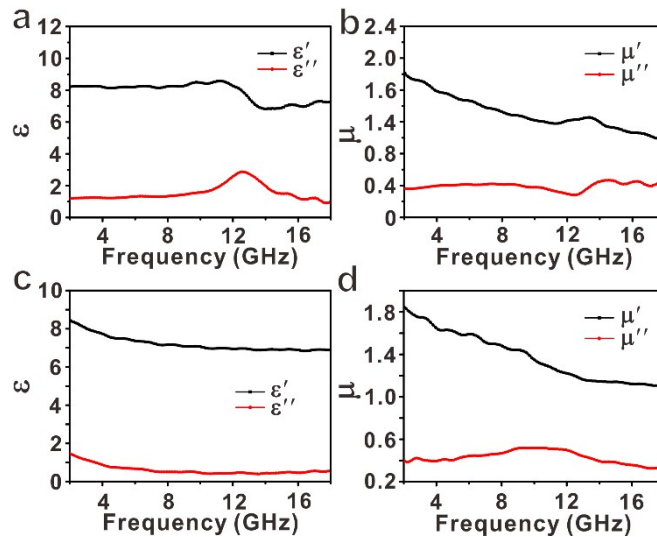


Figure S6. (a) Relative complex permittivity, (b) Relative complex permeability of the sample obtained by calcining $\text{Co}_{0.5}\text{Fe}_{0.5}\text{O}(\text{OH})@\text{PR}$ precursors at 800 °C for 2 h; (c) Relative complex permittivity, (d) Relative complex permeability of the sample obtained by calcining $\text{Co}_{0.5}\text{Fe}_{0.5}\text{O}(\text{OH})@\text{PR}$ precursors at 800 °C for 4 h.

Table S1. Iron cobalt ratio of iron cobalt oxide hydroxide hierarchical composites measured by EDX and ICP.

Sample	Initial feed molar ratio $(\frac{n_{Fe^{2+}}}{n_{Co^{2+}}})$	Molar ratio of final product $(\frac{n_{Fe}}{n_{Co}})$ by EDX	Molar ratio of final product $(\frac{n_{Fe}}{n_{Co}})$ by ICP
hydrated iron cobalt oxide	1	0.93	0.94

Table S2. Carbon contents and magnetic component contents of CoFe/C hierarchical composites measured by elementary analysis method.

Samples	The mass percent of carbon (wt %)	The mass percent of magnetic component (wt %)	Calcination temperature (°C)
CoFe/C-700	21.7	78.3	700
CoFe/C-M	27.5	72.5	700
CoFe/C-L	16.7	83.3	700