Electronic Supplementary Information

Ultra-Stable CsPbBr₃ Nanocrystals with Lead-Carboxylate/SiO₂ Encapsulation for LED Application

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Fig. S1 PL (a) and ATR-FTIR (b) spectra of C-PNCs and C-PNCs-SiO₂, in which C-PNCs-SiO₂ was prepared by prolonging the reaction time to 8 h without adding ethanol.



Fig. S2 TEM image of C-PNCs-SiO₂ prepared without adding ethanol.



Fig. S3 TEM images of pristine I-PNCs (a) and C-PNCs (b).



Fig. S4 TEM images of C-PNCs-SiO₂ prepared at different reaction time.



Fig. S5 UV-Vis and PL spectra of C-PNCs-SiO₂ prepared at different reaction time.



Fig. S6 (a) FTIR spectra of C-PNCs-SiO₂ synthesized with 0 μ L, 50 μ L and 100 μ L ethanol; (b) PLQY of C-PNCs-SiO₂ synthesized with 0 μ L, 25 μ L, 50 μ L, 75 μ L and 100 μ L ethanol.



Fig. S7 The UV-Vis spectra of C-PNCs-SiO₂ prepared with different concentrations of TMOS.



Fig. S8 TEM images of SiO₂ modified I-PNCs in the presence of ethanol.



Fig. S9 Local enlarged view of silicone resin film under optical microscope and the insert shows photographs of I-PNCs-SiO₂.



Fig. S10 Transmittance of the composite films of C-PNCs-SiO₂ and I-PNCs-SiO₂ in silicone resin.



Fig. S11 Photographs of SiO₂ modified I-PNCs in silicone resin.



Fig. S12 Photostability of C-PNCs-SiO₂ and I-PNCs-SiO₂ powders under continuous excitation.



Fig. S13 Photostability of C-PNCs-SiO₂ prepared without using ethanol.



Fig. S14 PL peak shift of C-PNCs-SiO₂ and I-PNCs-SiO₂ in silicone resin during ultrasonic treatment in water.



Fig. S15 Photographs of C-PNCs-SiO₂ in silicone resin stored in air at 20-40% RH for 45 days.



Fig. S16 Three times thermal recovery process of I-PNCs-SiO₂ (a) and C-PNCs-SiO₂ (b) at different temperatures. (c) The change of PL intensity after different cycles of thermal treatment.



Fig. S17 Color coordinate shift of C-PNCs-SiO₂ converted LEDs under 5-50mA current.



Fig. S18 The integrated intensity of red, green and blue light in LED at different operation power.