## **Supporting Information**

CdAg Alloy@Polymer dots of Biginelli Polyamide for Highly Sensitive and Selective Recognition of Nerve Agent mimic in an Aqueous and Vapor phase

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Table S1: Zeta potential values of Pdots.

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**Figure S3:** (A) UV- visible absorbance spectral profiles CdAg Alloy@Pdots in comparison with Pdots only, Pdots+Cd(II) comlex and Pdots+Cd(II)+Ag(I) complex. (B) EDX analysis of CdAgAlloy@Pdots showing presence of all elements, the peak corresponding to Cu mark with a star comes from the carbon coated copper grid used for STEM-EDX mapping. (C) STEM image of CdAgAlloy@Pdots and the elemental mapping of Oxygen, Nitrogen, Silver, and Cadmium respectively.

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Figure S5: (A) Stacked FTIR spectra of complex CdAg Alloy@Pdots+DCP and CdAg Alloy@Pdots+DCNP in comparison with CdAg Alloy@Pdots, Pdots only, DCP only and DCNP only. (B) <sup>31</sup>P NMR of CdAgAlloy@Pdots in presence of increasing equvalents of DCP recorded in  $D_2O$  solvent.

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**Table S2**: Comparison of LOD and studies with reported other sensing materials of DCP and DCNP.



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<b>Table SI:</b> Leta potential values of Pdot
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	Ze	Average	Zeta		
Pdots	Reading 1	Reading 2	Reading 3	Potential	
	-27.2 mV	-27.4 mV	-28.0 mV	-27.5 n	nV



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Table S2	: Comparison	of LOD	and	studies	with	reported	other	sensing	materials	of	DCP	and
DCNP.												

Entry	Probe	Analyte	Solvent	Response	LO	Response	Applicati	Reference
				Туре	D	Time	on	
1.	4- diphenylami no-2- hydroxy benzaldehyd e oxime (TPOD)	DCP	CH <sub>3</sub> CN- H <sub>2</sub> O (4:6, v/v)	Fluorescence Ratiometric	0.14 μΜ	30 s	Vapor Phase	[1]
2	Naphthalimi de-based- fluorescent probe	DCP	DMF	Fluorescence Enhancement	5.5 nM	300 s	Vapor Phase	[2]
3	Rhodamine- deoxylactum -based Fluorescent probe	DCP	DMF (3% Et <sub>3</sub> N)	Fluorescence Enhancement	9.66 nM	30 s	Vapor Phase	[3]
4	Rhodamine based Probe	DCP	EtOH– H <sub>2</sub> O (1 : 1 v/v)	UV/Fluoresc ence Enhancement	-	-	-	[4]
5	Polymer fibers	DFP/D CP	MeOH/ H <sub>2</sub> O	UV/ Fluorescence , Ratiometric	0.18 μM/ 0.16 μM	20 s	Vapor Phase	[5]

6	Binaphthol- Si Complex	DCP	DMF	Fluorescence Quenching	0.00 97m mol/ L	4 s	Vapor Phase	[6]
7	Azine based fluorescent probe	DCP	Acetonit rile/wat er (2:8)	Fluorescence Enhancement	9.86 nM	2 min.	Vapor Phase	[7]
8	PTS based Covalent assambly	DCP	Acetonit rile	Fluorescence Enhancement	10.4 nM	100 s	Vapor Phase	[8]
9	Heteroleptic Eu(III) Luminescent probes	DCP	MeCN	Luminescenc e Quenching	10 ppb	10 min.	Vapor Phase	[9]
10	Triazole- based Fluorescent probe	DNT/H <sub>2</sub> O <sub>2</sub> /DCP	THF	Fluorescence Quenching/ enhancement /ratiometric	-	2 s	Vapor Phase	[10]
11	1D photonic crystal (1D PC) films	DCP	-	Colorimetric	8 ppm	-	Vapor Phase	[11]
12	BTCP Fluorescent probe	DCP	MeCN/ H2O (1/1, v/v,)	UV/ Fluorescence Enhancement	15.8 nM	10 s	Vapor Phase and Bioimagi ng	[12]
13	Benzothiazol e-based Fluorescent Probe	DCP	MeCN	UV/ Fluorescence Enhancement	0.18 6 µM	6 s	Vapor Phase	[13]
14	4H-1, 2, 4- triazole (TAZ)-based Polymeric Probe	DCP	THF	Fluorescence Quenching	2.3 nM	5 s	Vapor Phase	[14]
15	Polymeric Film	DCP/A mmonia	Water	Colorimetric and Absorbance shift	18.4 μM/ 133μ Μ	< 1 min.	Vapor Phase	[15]
16	CdAg Alloy@Pdots	DCP/D CNP	Water	Fluorescence Enhancement	0.85 nM and 1.2 nM	20 s	Vapor Phase	Present Work

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