Electronic Supplementary Information (ESI)

Reversible stimuli-responsive luminescence of bimetallic cuprous complexes based on NH-deprotonated 3-(2'-pyridyl)pyrazole

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Fig. S1 ¹H NMR spectra of 1 in CD₂Cl₂.



Fig. S2 ¹H NMR spectra of **2** in CD_2Cl_2 .



Fig. S4 31 P NMR spectra of **2** in CD₂Cl₂.



Fig. S6 13 C NMR spectrum of 2 in CD₂Cl₂.



Fig. S7 Packing diagram of 1 along the *a* axis (a) with and (b) without CH₂Cl₂ solvate molecules;(c) the zoom of the window of the pore-channel structure of 1.



Fig. S8 Packing diagram of 2 along the *b* axis (a) with and (b) without CH₂Cl₂ solvate molecules;(c) the zoom of the window of the pore-channel structure of 2.



Fig. S9 ¹H NMR spectra of *desolvated* $\mathbf{1}$ (a) and *crystalline* $\mathbf{1}$ (b) in CDCl₃.



Fig. S10 PXRD patterns of the *crystalline*, *ground*, and *reverted* samples of 2 and that simulated from single-crystal data of 2.



Fig. S11 FT-IR spectra of dppm and the *crystalline*, *ground*, and *reverted* samples of **2**. Inset: the zoom of the Cl–O and methylene C–H stretching vibration absorptions.



Fig. S12 TGA curves of *desolvated* **1**, *reverted* **1**, and *crystalline* **2**.

compound	1	2
formula	$C_{59}H_{52}Cl_{3}Cu_{2}N_{3}O_{4}P_{4}$	$C_{60}H_{51}Cl_3Cu_2F_3N_3O_4P_4\\$
fw	1224.34	1292.34
<i>T</i> (K)	293(2)	299(2)
crystal system	triclinic	Monoclinic
space group	$P \overline{1}$	$P2_{1}/c$
<i>a</i> (Å)	12.9124(7)	11.7627(6)
<i>b</i> (Å)	14.3129(8)	18.5166(10)
<i>c</i> (Å)	17.8368(11)	27.2950(15)
α (deg)	71.683(2)	90
β (deg)	79.859(2)	98.1430(10)
γ (deg)	64.650(2)	90
$V(\text{\AA}^3)$	2824.8(3)	5885.1(5)
Ζ	2	4
$\rho_{\rm calcd} ({\rm g \ cm}^{-3})$	1.439	1.459
$\mu (\mathrm{mm}^{-1})$	1.057	1.026
no. reflections collected	43308	101096
no. unique reflections	12857	13170
$R_{\rm int}$	0.0327	0.0672
no. observed reflections	12857	13170
no. parameters	676	758
GOF on F^2	1.031	1.066
<i>R</i> 1 [$I > 2\sigma(I)$]	0.0403	0.0553
wR2	0.0920	0.1032

Table S1. Crystal data and structure refinement parameters of $1 \mbox{ and } 2$

compound	1	2
Cu1-N1	2.169(2)	2.189(3)
Cu1–N2	1.986(2)	2.017(3)
Cu1-P1	2.2514(7)	2.2613(10)
Cu1–P2	2.2462(7)	2.2454(10)
Cu2-N3	1.997(2)	2.032(3)
Cu2–P3	2.2310(7)	2.2426(10)
Cu2-P4	2.2426(7)	2.2432(10)
N1-Cu1-N2	78.56(9)	77.38(12)
N1-Cu1-P1	113.01(6)	111.86(8)
N1-Cu1-P2	114.68(6)	113.15(8)
N2-Cu1-P1	113.09(6)	113.59(8)
N2-Cu1-P2	118.84(7)	116.25(8)
P1–Cu1–P2	113.98(3)	117.86(4)
N3-Cu2-P3	116.26(7)	118.11(9)
N3-Cu2-P4	113.78(7)	117.47(9)
P3-Cu2-P4	127.50(3)	122.69(4)

Table S2. Selected Bond Lengths (Å) and Angles (deg) of $1 \mbox{ and } 2$

 Table S3. Photophysical Data of 1 and 2

compound	medium	$\lambda_{abs} [nm] (\varepsilon [M^{-1} cm^{-1}])$	$\lambda_{\rm em}$ [nm]	τ [μs]	$\Phi_{ m em}$ [%]
1	CH_2Cl_2	276 (31467)	526	3	0.4
	solid		461, ^{<i>a</i>} 508, ^{<i>c</i>} 472 ^{<i>d</i>}	210, ^{<i>a</i>} 48, ^{<i>c</i>} 185 ^{<i>d</i>}	38, ^{<i>a</i>} 21, ^{<i>c</i>} 55 ^{<i>d</i>}
2	CH_2Cl_2	275 (39030), 313 (14248)	527	6	1.1
	solid		492, ^b 506, ^c 493 ^d	$47,^{b} 32,^{c} 40^{d}$	88, ^b 28, ^c 81 ^d

^{*a*} *Desolvated* sample. ^{*b*} *Crystalline* sample. ^{*c*} *Ground* samples. ^{*d*} *Reverted* samples.