

Electronic Supporting Information

A Covalent Organic Polymer as Turn-on Fluorescence Sensor for Hydrazine Detection

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Supplementary Figures

Fig. S1-S4 show the ^1H NMR and mass spectra of TPE-(Si) $_4$ and TPE-Al, respectively.

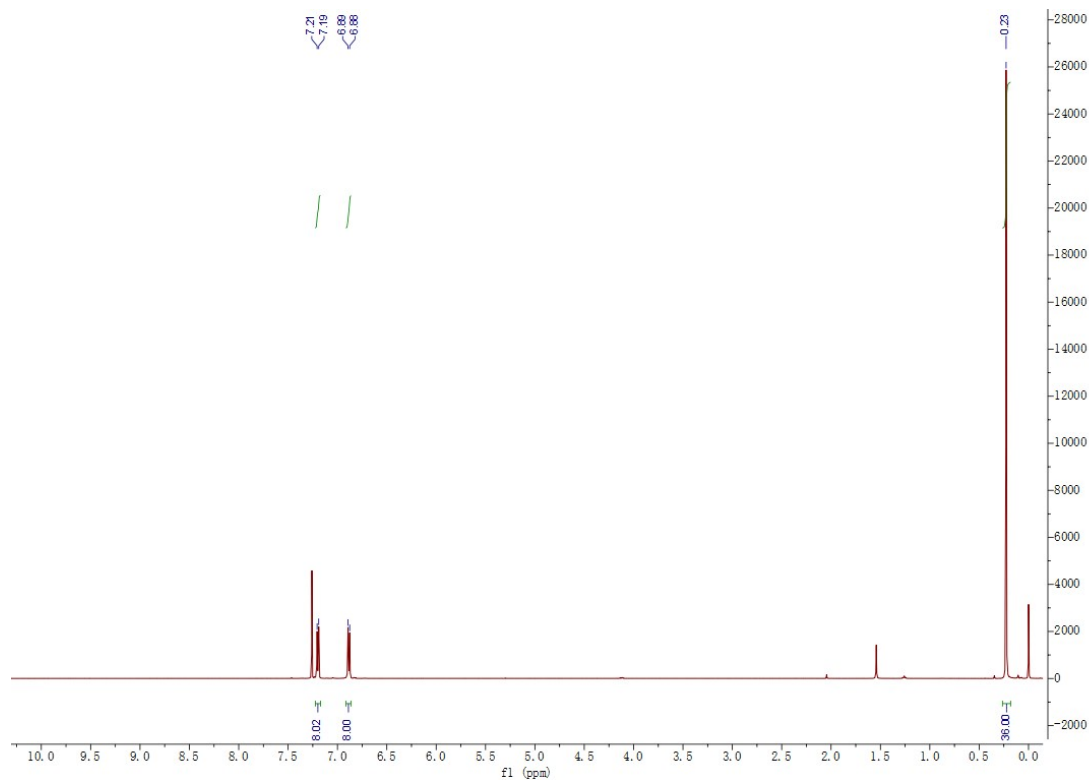


Fig. S1 ^1H NMR spectrum of TPE-(Si) $_4$.

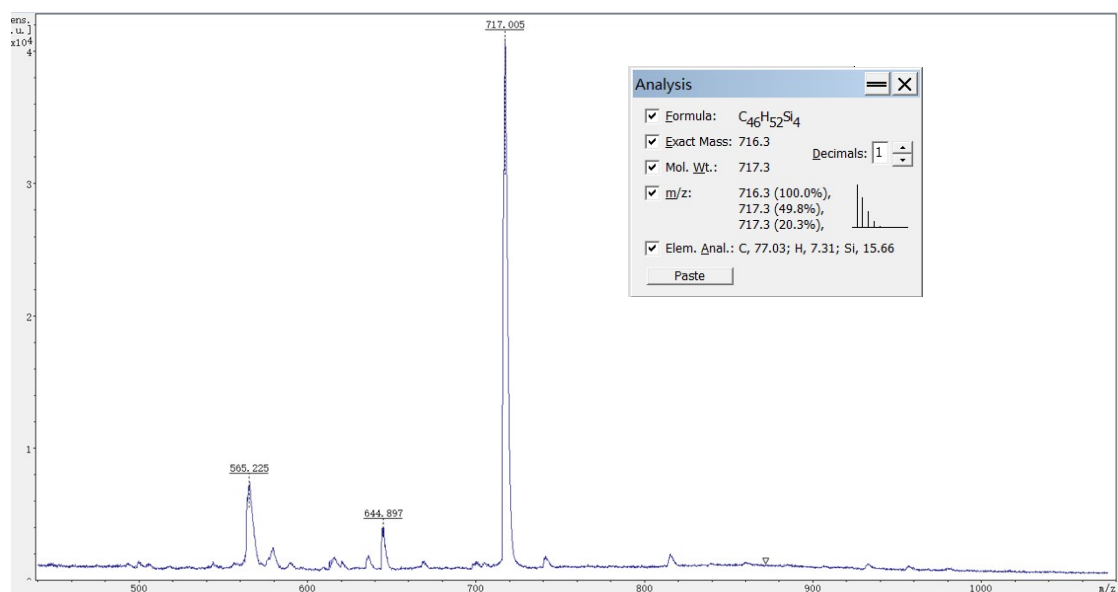


Fig. S2 Mass spectrum of TPE-(Si) $_4$.

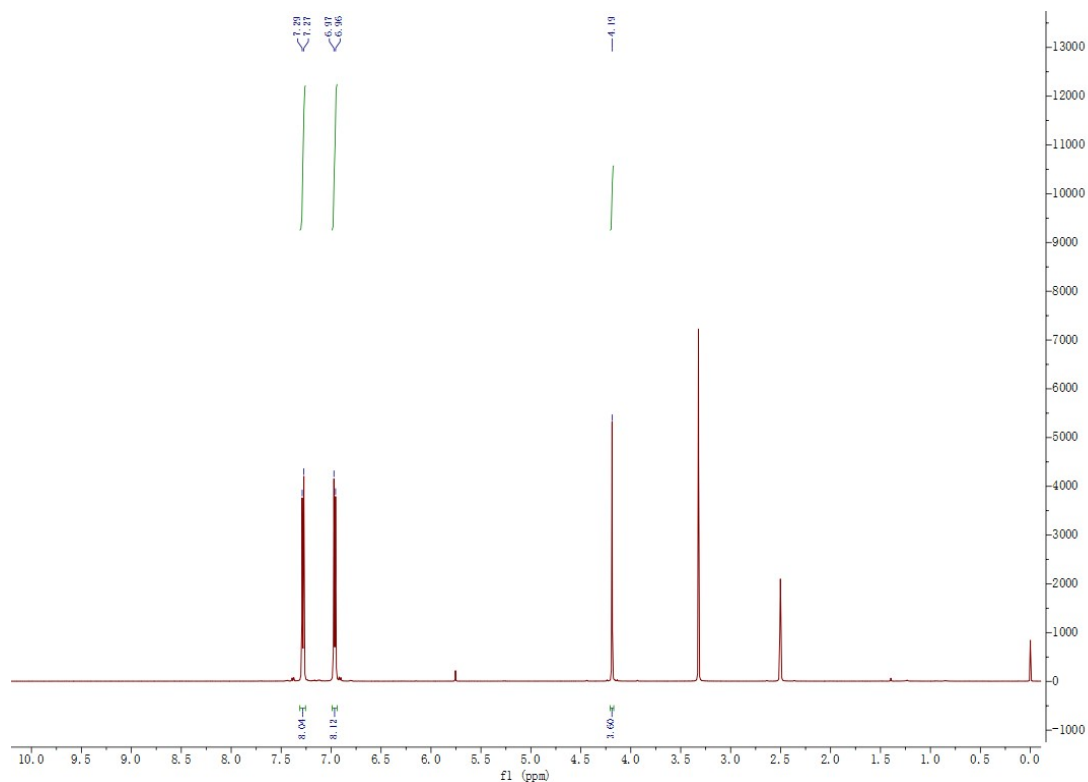


Fig. S3 ^1H NMR spectrum of TPE-Al.

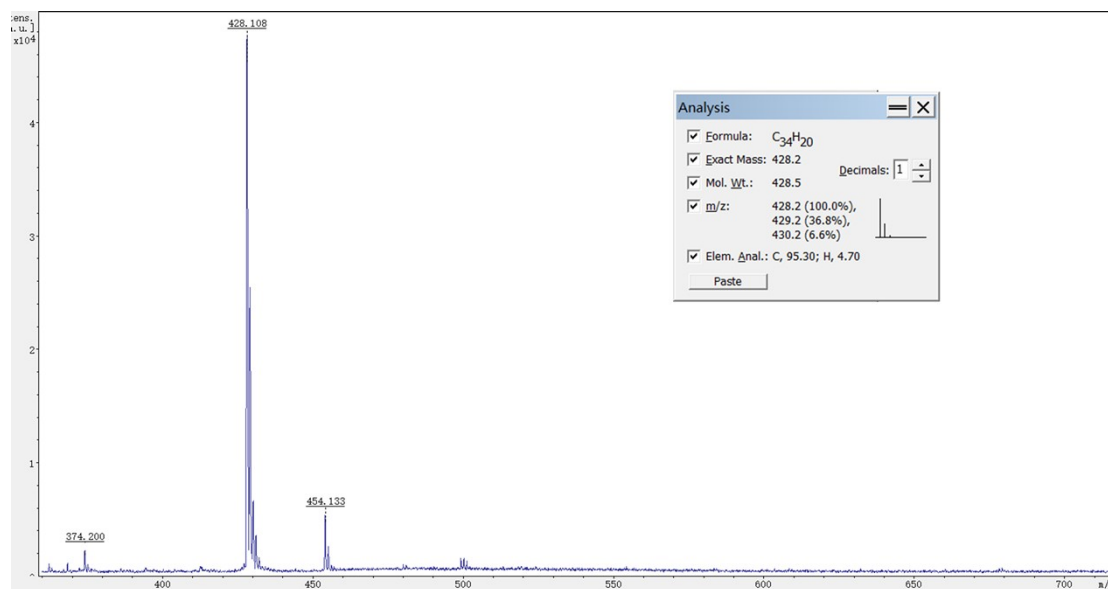


Fig. S4 Mass spectrum of TPE-Al.

Fig. S5 shows the powder X-Ray diffraction (PXRD) pattern of COP-Ta. It can be observed that there is no obvious crystallization peak, indicating that COP-Ta is an amorphous polymer.

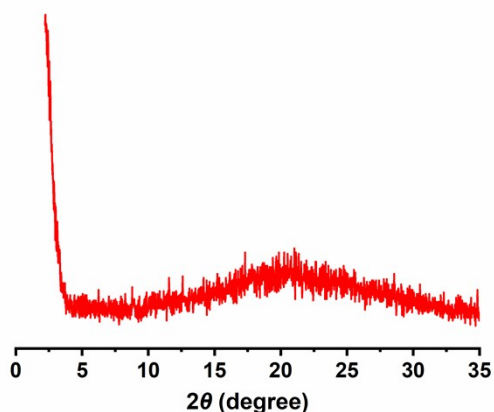


Fig. S5 PXRD pattern of COP-Ta.

The scanning electron microscopy (SEM) and transmission electron microscope (TEM) images are shown in Fig. S6. The random block morphology can be observed in the SEM image of COP-Ta (Fig. S6a). From the transmission electron microscope image (Fig. S6b), a flaky stacking structure can be observed.

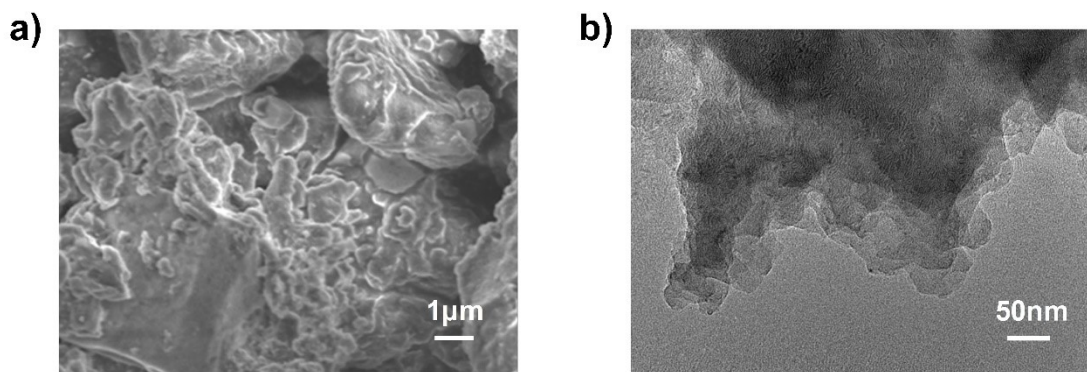


Fig. S6 a) SEM and b) TEM images of COP-Ta.

Fig. S7 shows the TGA curve of COP-Ta from room temperature to 800 $^\circ\text{C}$ in N_2 atmosphere. COP-Ta has almost no weight loss before 300 $^\circ\text{C}$; After 300 $^\circ\text{C}$, COP-Ta begins to lose weight gradually. When the temperature rises to 450 $^\circ\text{C}$, the weight loss of COP-Ta is only 10%. Finally, when the temperature rises to 800 $^\circ\text{C}$, COP-Ta

decomposes only a small part. All these phenomena show that the polymer has an excellent thermal stability.

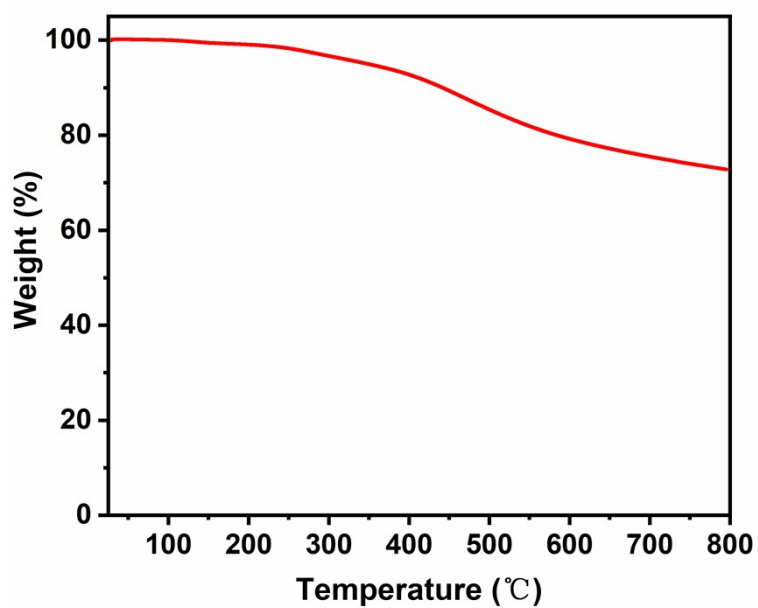


Fig. S7 TGA curve of COP-Ta.

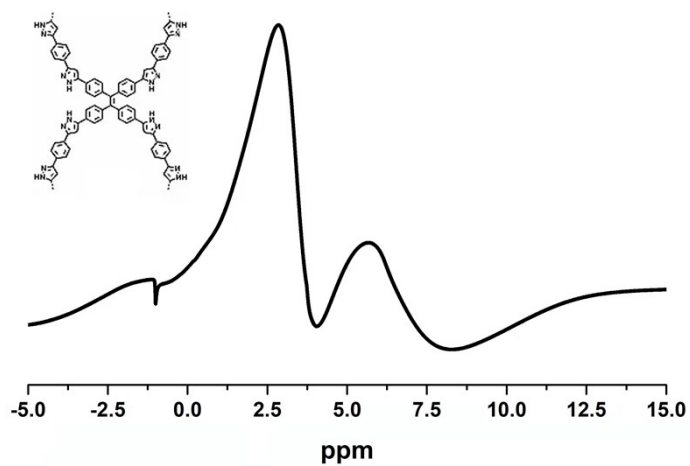


Fig. S8 Solid-state ^1H NMR spectrum of COP-Ta after hydrazine addition.