

***Supplementary Information***

Structure-Property Relationship Study of Blue Thermally Activated  
Delayed Fluorescence Molecules with Different donor and position  
substitutions: Theoretical Perspective and Molecular Design

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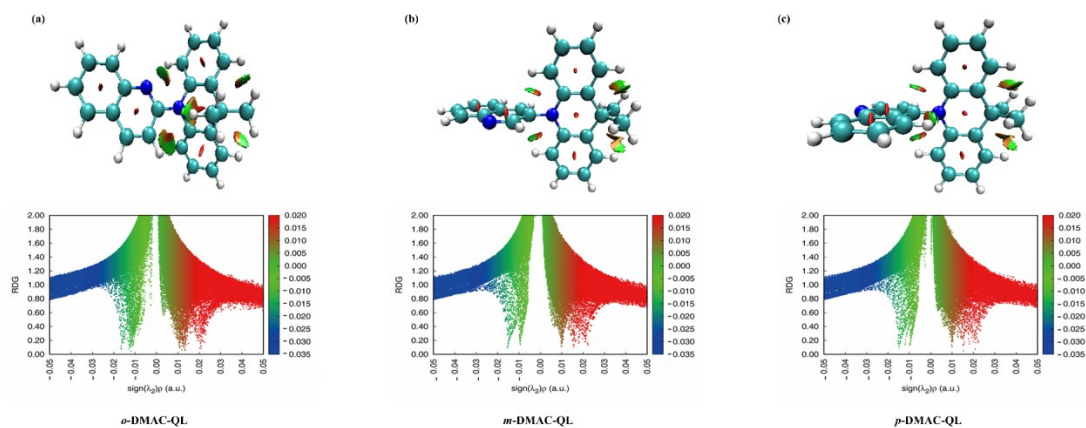
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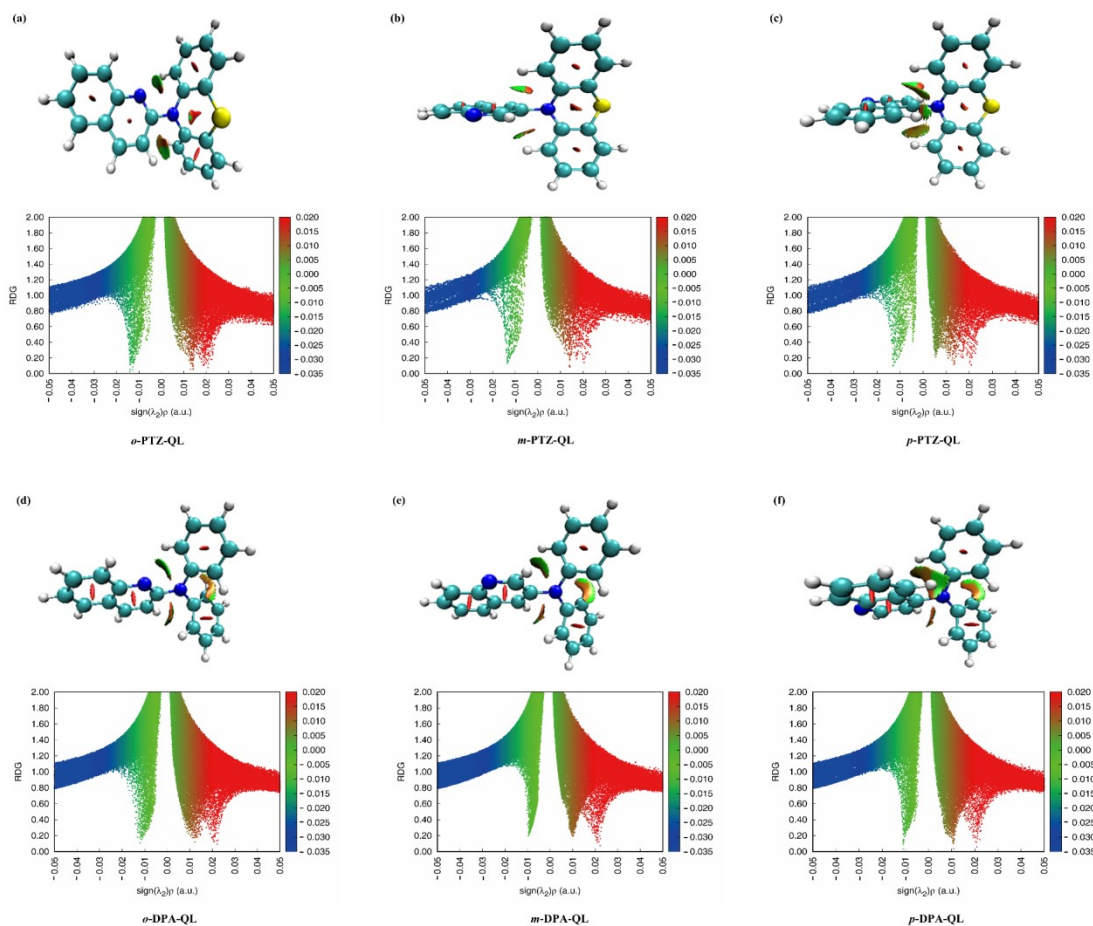
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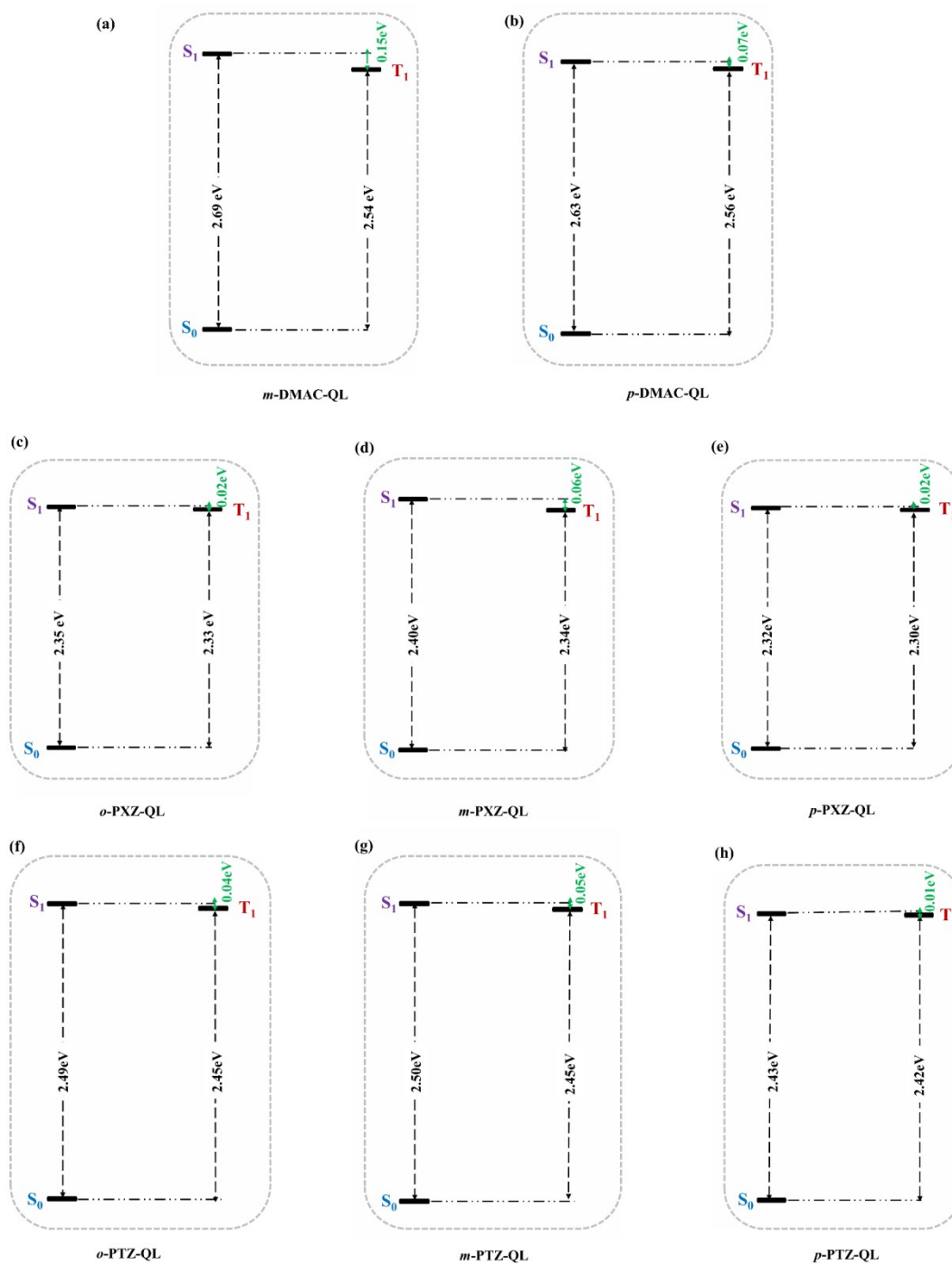
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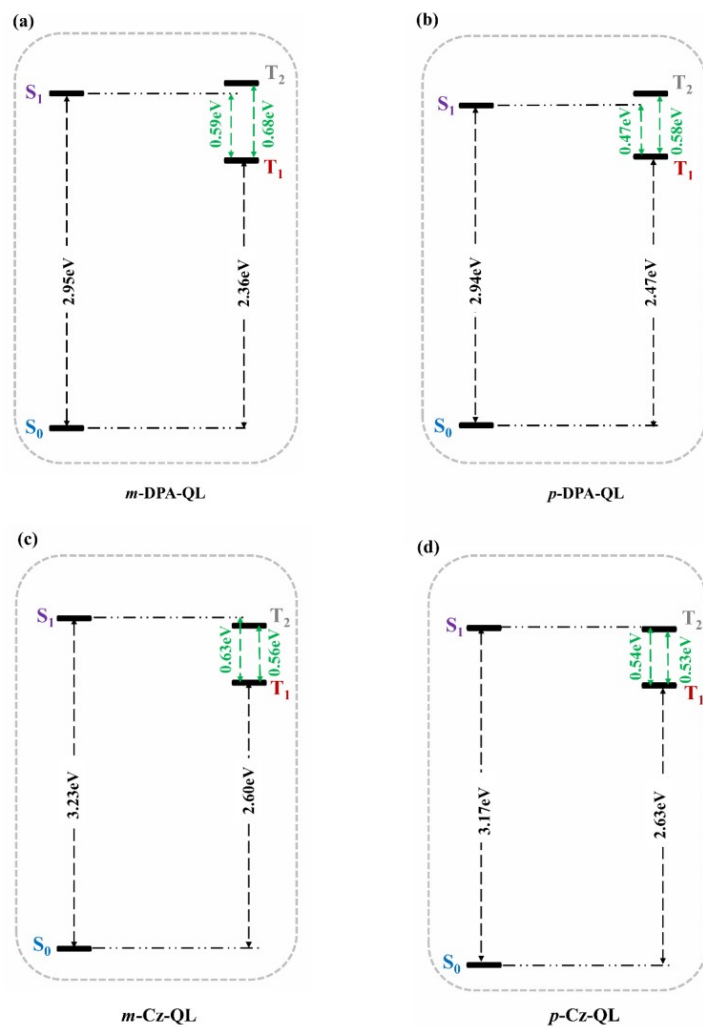
**Figure S1.** Reduced density gradient (RDG) isosurface maps (upper) and scatter graphs (lower) of RDG versus  $\text{sign}(\lambda_2)\rho$  (a.u.) of *o*-DMAC-QL, *m*-DMAC-QL and *p*-DMAC-QL.



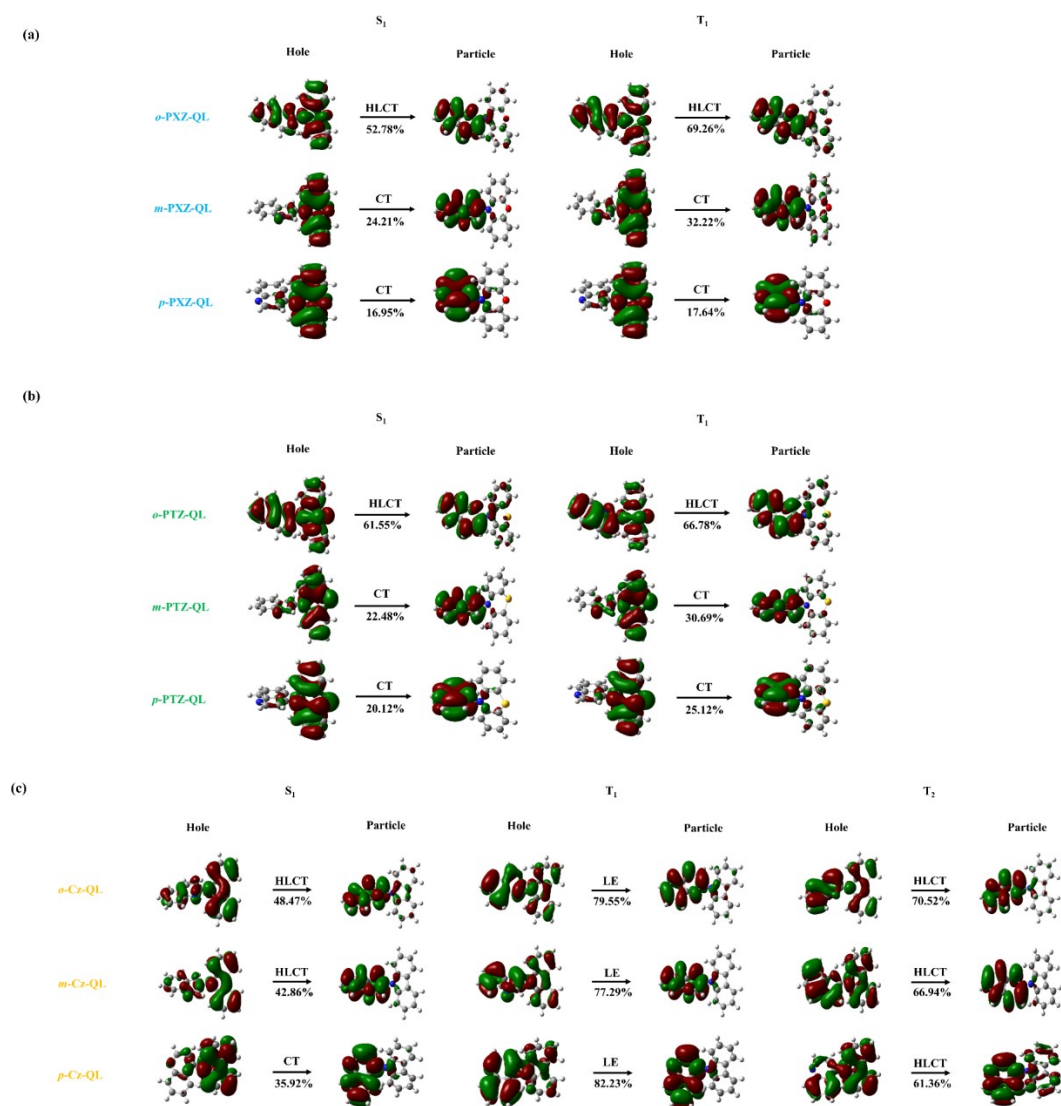
**Figure S2.** Reduced density gradient (RDG) isosurface maps (upper) and scatter graphs (lower) of RDG versus  $\text{sign}(\lambda_2)\rho$  (a.u.) of *o*-PTZ-QL, *m*-PTZ-QL, *p*-PTZ-QL, *o*-DPA-QL, *m*-DPA-QL and *p*-DPA-QL.



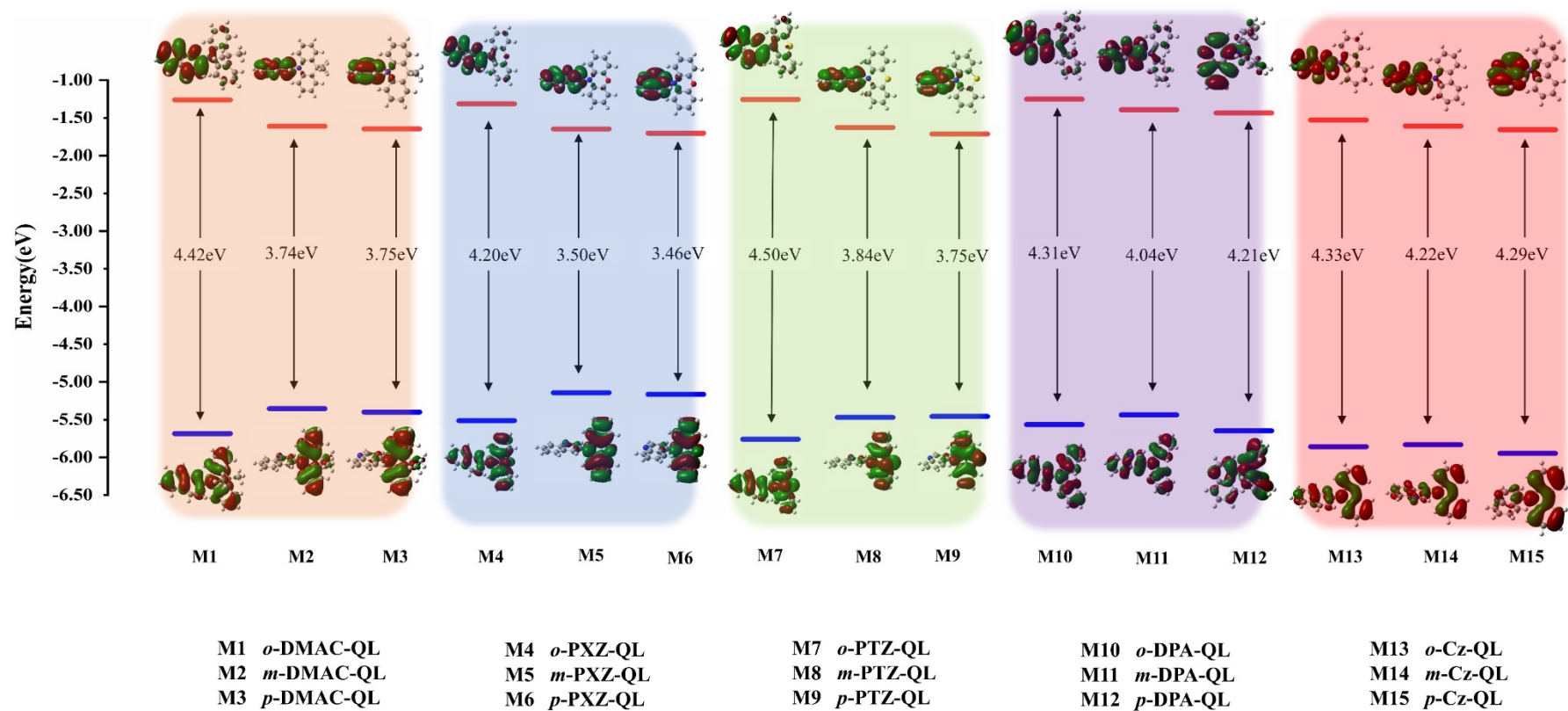
**Figure S3.** Adiabatic excitation energies for *m*-DMAC-QL, *p*-DMAC-QL, *o*-PXZ-QL, *m*-PXZ-QL, *p*-PXZ-QL, *o*-PTZ-QL, *m*-PTZ-QL and *p*-PTZ-QL in tetrahydrofuran, respectively.



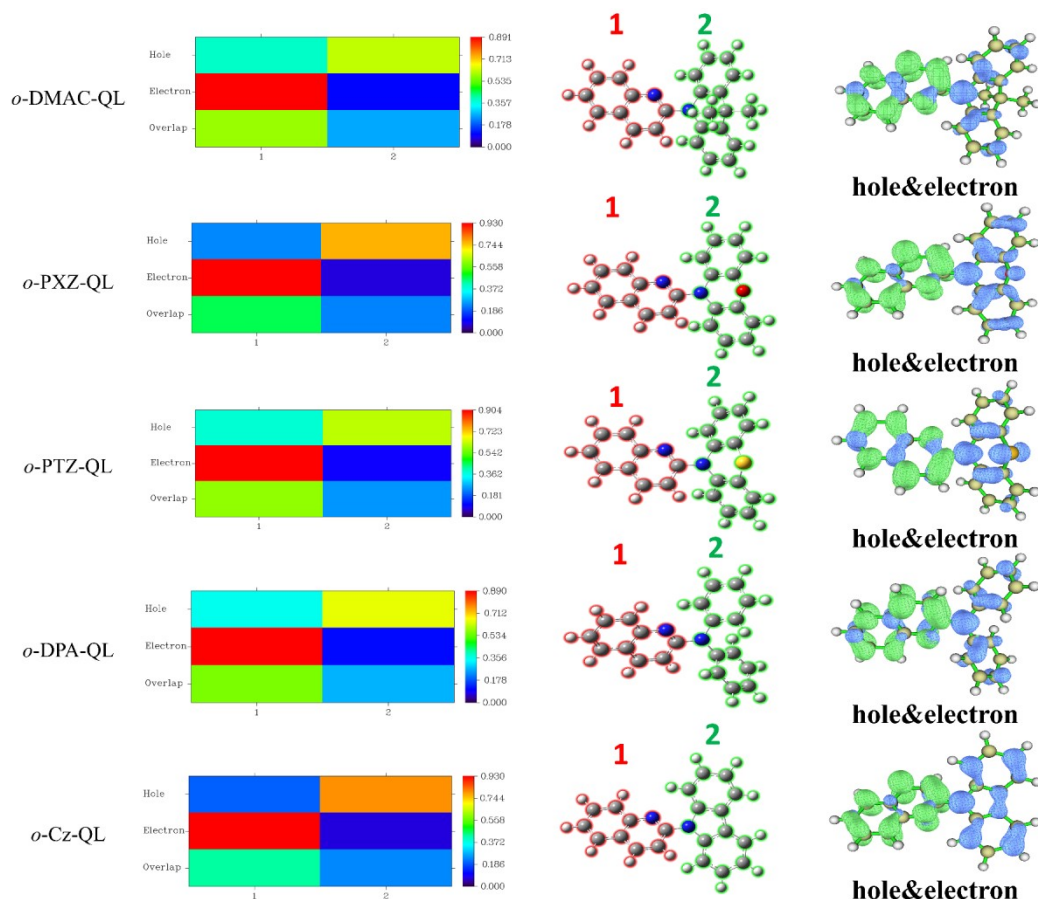
**Figure S4.** Adiabatic excitation energies for *m*-DPA-QL, *p*-DPA-QL, *m*-Cz-QL and *p*-Cz-QL, respectively.



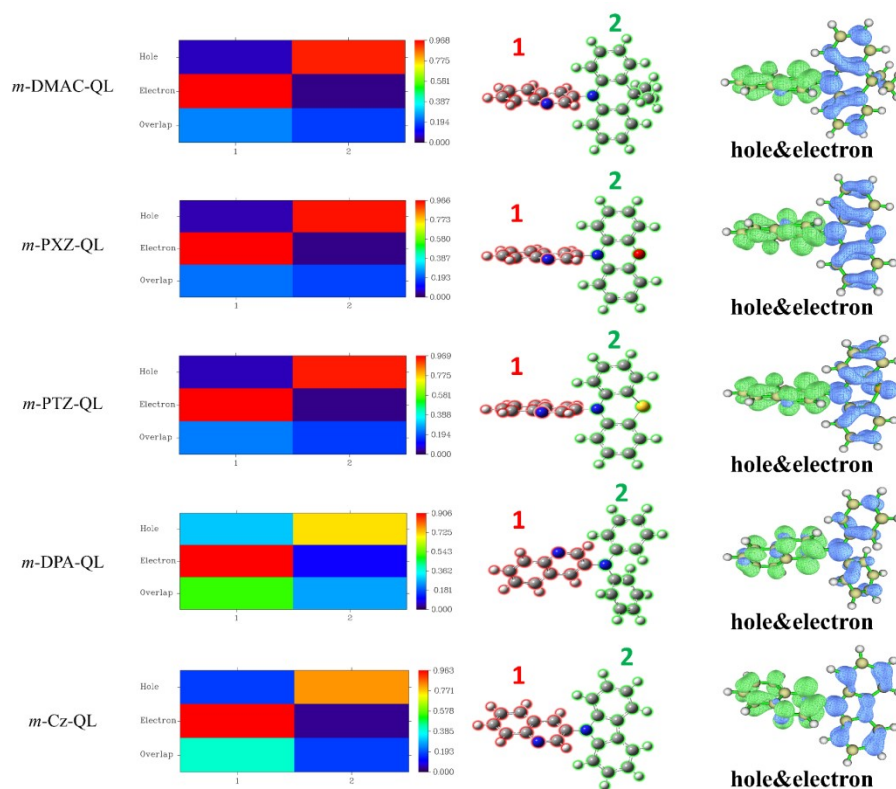
**Figure S5.** Natural transition orbitals (NTOs) of the  $S_1$  and  $T_n$  states for *o*-PXZ-QL, *m*-PXZ-QL, *p*-PXZ-QL, *o*-PTZ-QL, *m*-PTZ-QL, *p*-PTZ-QL, *o*-Cz-QL, *m*-Cz-QL and *p*-Cz-QL in tetrahydrofuran. The values below the arrows are the LE proportion in excitation.



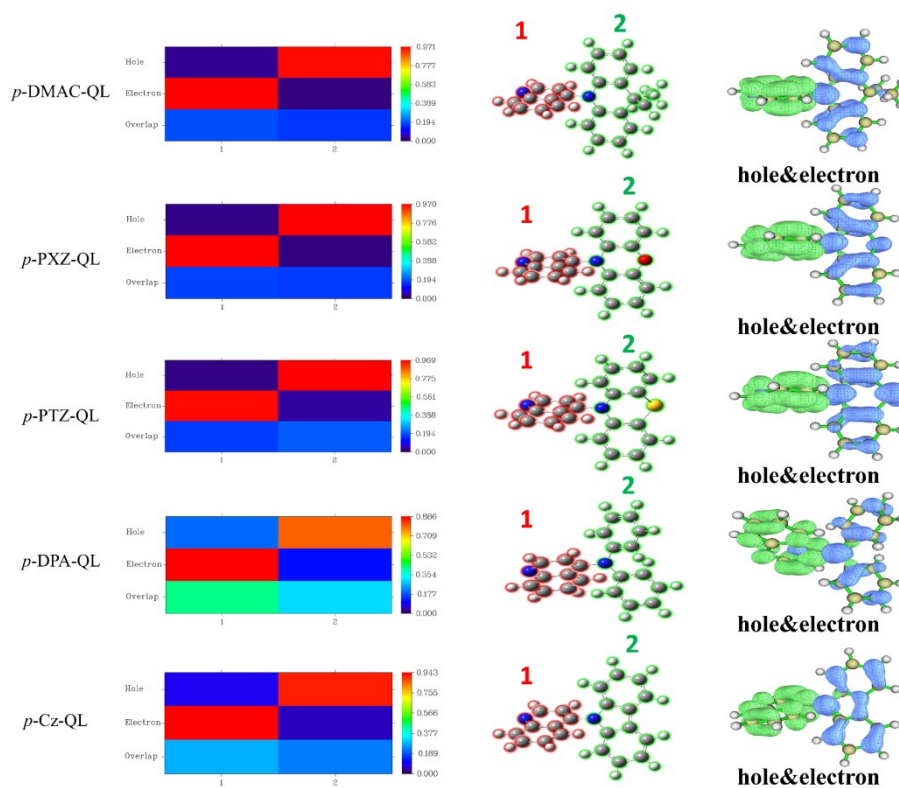
**Figure S6.** Energies and distributions of HOMO (blue lines) and LUMO (red lines) for all molecules in tetrahydrofuran.



**Figure S7.** Isosurface distributions of holes and electrons and heat map of the  $S_1$  state for *o*-DMAC-QL, *o*-PXZ-QL, *o*-PTZ-QL, *o*-DPA-QL and *o*-Cz-QL in tetrahydrofuran.



**Figure S8.** Isosurface distributions of holes and electrons and heat map of the  $S_1$  state for *m*-DMAC-QL, *m*-PXZ-QL, *m*-PTZ-QL, *m*-DPA-QL and *m*-Cz-QL in tetrahydrofuran.



**Figure S9.** Isosurface distributions of holes and electrons and heat map of the  $S_1$  state for *p*-DMAC-QL, *p*-PXZ-QL, *p*-PTZ-QL, *p*-DPA-QL and *p*-Cz-QL in tetrahydrofuran.



**Table S1.** Orbital energies of HOMO and LUMO, energy gaps ( $\epsilon_{\text{gap}}$ ) and overlaps (S).

	HOMO (eV)	LUMO (eV)	$\epsilon_{\text{gap}}$ (eV)	S
<i>o</i> -DMAC-QL	-5.68	-1.26	4.42	0.61
<i>m</i> -DMAC-QL	-5.35	-1.61	3.74	0.21
<i>p</i> -DMAC-QL	-5.40	-1.64	3.75	0.15
<i>o</i> -PXZ-QL	-5.51	-1.31	4.20	0.52
<i>m</i> -PXZ-QL	-5.14	-1.65	3.50	0.21
<i>p</i> -PXZ-QL	-5.16	-1.70	3.46	0.15
<i>o</i> -PTZ-QL	-5.76	-1.25	4.50	0.60
<i>m</i> -PTZ-QL	-5.47	-1.63	3.84	0.20
<i>p</i> -PTZ-QL	-5.46	-1.71	3.75	0.17
<i>o</i> -DPA-QL	-5.56	-1.25	4.31	0.60
<i>m</i> -DPA-QL	-5.44	-1.39	4.04	0.61
<i>p</i> -DPA-QL	-5.65	-1.43	4.21	0.58
<i>o</i> -Cz-QL	-5.86	-1.53	4.33	0.47
<i>m</i> -Cz-QL	-5.83	-1.61	4.22	0.42
<i>p</i> -Cz-QL	-5.95	-1.66	4.29	0.37

The  $\epsilon_{\text{gap}}$  and S are the energy gap and the orbital overlap integral between the HOMO and LUMO, respectively.

**Table S2.** The LE proportion in excitation and the difference of LE components between  $S_1$  and  $T_n$  states for total 15 molecules.

	$S_1$	$T_1$	$T_2$	Difference between $T_1-S_1$	Difference between $T_2-S_1$
<i>o</i> -DMAC-QL	60.71%	75.47%	\	14.75%	\
<i>m</i> -DMAC-QL	24.22%	29.97%	\	5.74%	\
<i>p</i> -DMAC-QL	17.54%	19.02%	\	1.48%	\
<i>o</i> -PXZ-QL	52.78%	69.26%	\	16.48%	\
<i>m</i> -PXZ-QL	24.21%	32.22%	\	8.01%	\
<i>p</i> -PXZ-QL	16.95%	17.64%	\	0.69%	\
<i>o</i> -PTZ-QL	61.55%	66.78%	\	5.23%	\
<i>m</i> -PTZ-QL	22.48%	30.69%	\	8.21%	\
<i>p</i> -PTZ-QL	20.12%	25.12%	\	4.99%	\
<i>o</i> -DPA-QL	60.47%	74.26%	82.99%	13.79%	22.52%
<i>m</i> -DPA-QL	59.28%	76.07%	77.28%	16.79%	18.00%
<i>p</i> -DPA-QL	55.41%	77.62%	75.51%	22.21%	20.09%
<i>o</i> -Cz-QL	48.47%	79.55%	70.52%	31.08%	22.05%
<i>m</i> -Cz-QL	42.86%	77.29%	66.94%	34.43%	24.08%
<i>p</i> -Cz-QL	35.92%	82.23%	61.36%	46.31%	25.44%

**Table S3.** Calculated intersystem crossing rates ( $K_{ISC}$ ) and reverse intersystem crossing rates ( $K_{RISC}$ ) between the singlet and triplet excited states of all molecules.

	$K_{ISC}$ ( $S_1 \rightarrow T_1$ ) ( $s^{-1}$ )	$K_{ISC}$ ( $S_1 \rightarrow T_2$ ) ( $s^{-1}$ )	$K_{RISC}$ ( $T_1 \rightarrow S_1$ ) ( $s^{-1}$ )	$K_{RISC}$ ( $T_2 \rightarrow S_1$ ) ( $s^{-1}$ )
<i>o</i> -DMAC-QL	$1.75 \times 10^4$		$8.84 \times 10^6$	
<i>m</i> -DMAC-QL	$3.34 \times 10^6$		$6.09 \times 10^5$	
<i>p</i> -DMAC-QL	$2.88 \times 10^5$		$2.18 \times 10^7$	
<i>o</i> -PXZ-QL	$6.08 \times 10^4$		$3.30 \times 10^8$	
<i>m</i> -PXZ-QL	$2.44 \times 10^6$		$2.47 \times 10^7$	
<i>p</i> -PXZ-QL	$5.38 \times 10^5$		$1.23 \times 10^8$	
<i>o</i> -PTZ-QL	$8.52 \times 10^4$		$1.04 \times 10^5$	
<i>m</i> -PTZ-QL	$5.45 \times 10^6$		$2.54 \times 10^6$	
<i>p</i> -PTZ-QL	$7.85 \times 10^5$		$4.48 \times 10^6$	
<i>o</i> -DPA-QL	$3.13 \times 10^7$	$1.04 \times 10^7$	$1.41 \times 10^1$	$1.07 \times 10^7$
<i>m</i> -DPA-QL	$1.72 \times 10^7$	$7.07 \times 10^5$	$2.16 \times 10^{-2}$	$3.18 \times 10^8$
<i>p</i> -DPA-QL	$1.52 \times 10^8$	$6.85 \times 10^4$	$2.41 \times 10^0$	$1.64 \times 10^7$
<i>o</i> -Cz-QL	$3.77 \times 10^6$	$2.85 \times 10^8$	$2.04 \times 10^0$	$8.14 \times 10^5$
<i>m</i> -Cz-QL	$9.83 \times 10^6$	$2.59 \times 10^8$	$7.84 \times 10^{-4}$	$1.89 \times 10^7$
<i>p</i> -Cz-QL	$3.69 \times 10^7$	$4.33 \times 10^5$	$7.27 \times 10^{-3}$	$2.11 \times 10^5$

**Table S4.** Total reorganization energies between  $S_1$  and  $S_0$  of 15 molecules in tetrahydrofuran.

	Reorganization energy ( $cm^{-1}$ )
<i>o</i> -DMAC-QL	16093.94

<i>m</i> -DMAC-QL	1492.42
<i>p</i> -DMAC-QL	1490.64
<i>o</i> -PXZ-QL	10171.06
<i>m</i> -PXZ-QL	2071.93
<i>p</i> -PXZ-QL	1827.46
<i>o</i> -PTZ-QL	17080.32
<i>m</i> -PTZ-QL	4407.28
<i>p</i> -PTZ-QL	3019.29
<i>o</i> -DPA-QL	5838.92
<i>m</i> -DPA-QL	2818.24
<i>p</i> -DPA-QL	4660.44
<i>o</i> -Cz-QL	1994.44
<i>m</i> -Cz-QL	1457.75
<i>p</i> -Cz-QL	1978.16

**Table S5.** Calculated spin-orbit coupling constants (in  $\text{cm}^{-1}$ ) between the  $S_0$  and  $T_1$  for all molecules in tetrahydrofuran (THF).

	$\langle S_0   \hat{H}_{so}   T_1 \rangle$			
	X-component ( $\text{cm}^{-1}$ )	Y-component ( $\text{cm}^{-1}$ )	Z-component ( $\text{cm}^{-1}$ )	Total ( $\text{cm}^{-1}$ )
<i>o</i> -DMAC-QL	2.78	-1.36	1.79	3.57
<i>m</i> -DMAC-QL	1.54	-1.36	-2.27	3.06
<i>p</i> -DMAC-QL	1.46	-1.03	-1.14	1.92
<i>o</i> -PXZ-QL	1.87	1.06	-2.48	3.28
<i>m</i> -PXZ-QL	-1.61	1.59	-2.43	3.32
<i>p</i> -PXZ-QL	-1.93	-3.00	0.48	2.02
<i>o</i> -PTZ-QL	-2.27	$-9.62 \times 10^{-4}$	2.28	3.21
<i>m</i> -PTZ-QL	1.19	$5.61 \times 10^{-4}$	-2.85	3.09
<i>p</i> -PTZ-QL	-1.03	$-1.98 \times 10^{-4}$	5.99	1.12
<i>o</i> -DPA-QL	2.31	0.84	0.80	2.59
<i>m</i> -DPA-QL	-1.14	-1.03	-1.14	1.92
<i>p</i> -DPA-QL	-1.15	0.26	0.16	1.19
<i>o</i> -Cz-QL	-0.67	-0.92	0.86	1.43
<i>m</i> -Cz-QL	-1.23	1.05	1.19	2.00
<i>p</i> -Cz-QL	-0.53	0.33	-0.17	0.65

**Table S6.** Electric emission transition dipole moments of total 15 molecules.

$\mu$ (Debye)	
<i>o</i> -DMAC-QL	0.28

<i>m</i> -DMAC-QL	0.29
<i>p</i> -DMAC-QL	0.07
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<i>o</i> -PXZ-QL	0.04
<i>m</i> -PXZ-QL	0.06
<i>p</i> -PXZ-QL	0.05
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<i>o</i> -PTZ-QL	0.13
<i>m</i> -PTZ-QL	0.20
<i>p</i> -PTZ-QL	0.05
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<i>o</i> -DPA-QL	2.45
<i>m</i> -DPA-QL	2.99
<i>p</i> -DPA-QL	2.20
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<i>o</i> -Cz-QL	2.77
<i>m</i> -Cz-QL	2.37
<i>p</i> -Cz-QL	0.04
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