

# Supporting Information

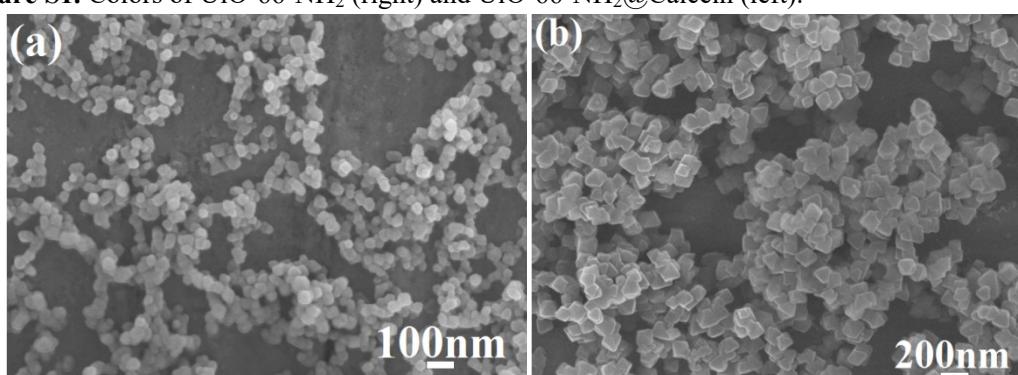
## Calcein-modified Zr(IV)-based metal-organic framework as a visualized sensor for calcium ion

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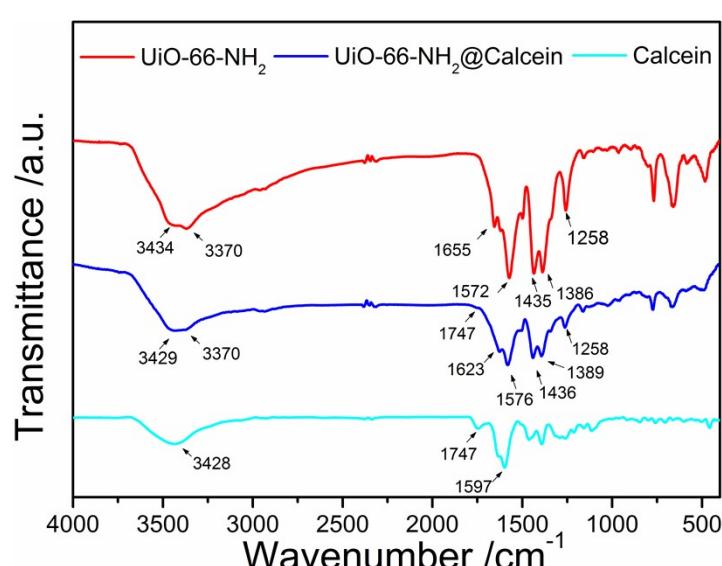
College of Chemistry and Materials Science, Key Laboratory of Functional Molecular Solids, Ministry of Education, Anhui Normal University, 189 Jiuhua Southern Road, Wuhu, 241002, PR China. Fax: +86-553-3869303. E-mail: [niyh@mail.ahnu.edu.cn](mailto:niyh@mail.ahnu.edu.cn)



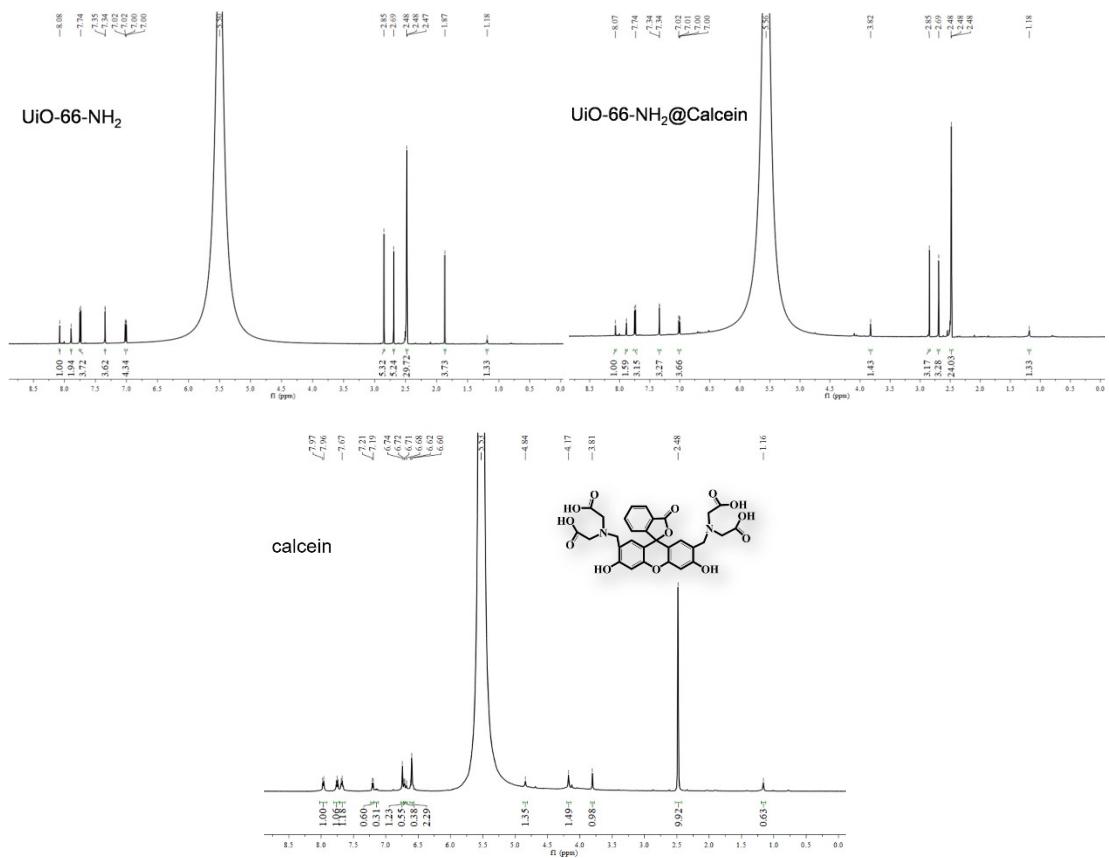
**Figure S1.** Colors of UiO-66-NH<sub>2</sub> (right) and UiO-66-NH<sub>2</sub>@Calcein (left).



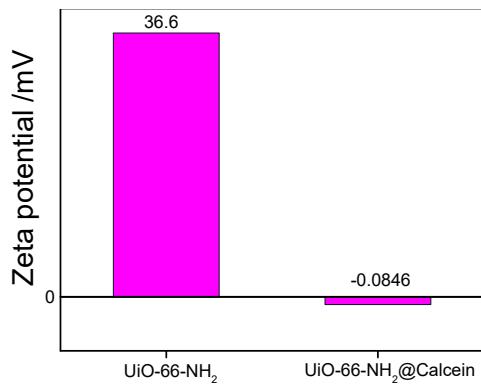
**Figure S2.** FESEM images of UiO-66-NH<sub>2</sub> synthesized in different solvents: (a) DMF and deionized water with the volume ratio of 1:1, and (b) DMF.



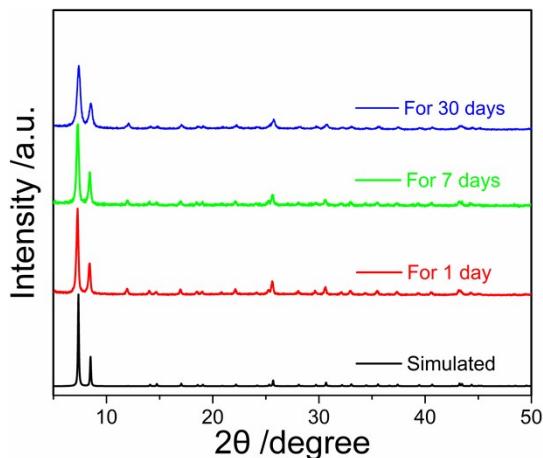
**Figure S3.** FTIR spectra of UiO-66-NH<sub>2</sub>, Calcein and UiO-66-NH<sub>2</sub>@Calcein.



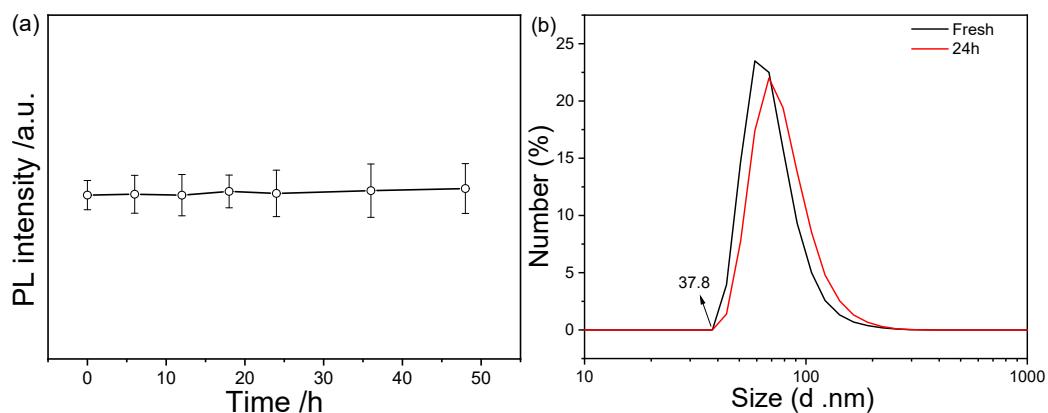
**Figure S4.** <sup>1</sup>H NMR spectra of as-obtained UiO-66-NH<sub>2</sub>, UiO-66-NH<sub>2</sub>@Calcein and calcein.



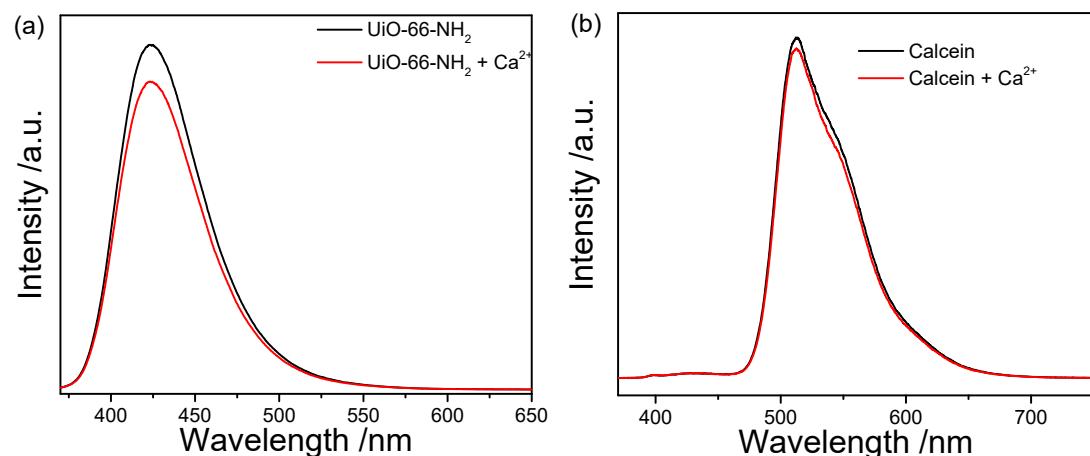
**Figure S5.** Zeta potentials of UiO-66-NH<sub>2</sub> and UiO-66-NH<sub>2</sub>@Calcein.



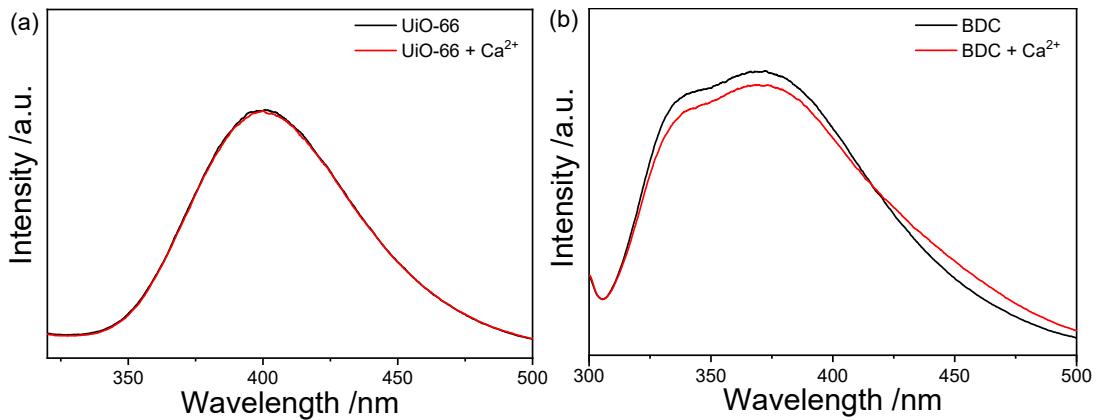
**Figure S6.** PXRD patterns of  $\text{UiO-66-NH}_2@\text{Calcein}$  immersed in water for 1, 7 and 30 days, respectively.



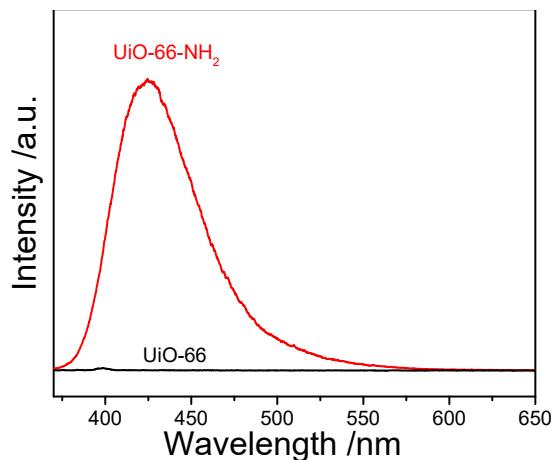
**Figure S7.** (a) The fluorescence intensity changes of  $\text{UiO-66-NH}_2@\text{Calcein}$  dispersed in water for 0 ~ 48 h; and (b) The DLS results of the  $\text{UiO-66-NH}_2@\text{Calcein}$  suspension before and after placed for 24 h.



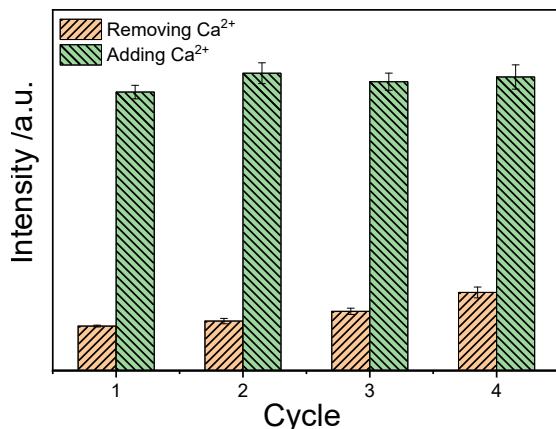
**Figure S8.** Fluorescence emission spectra of  $\text{UiO-66-NH}_2$  (a) and calcein (b) before and after adding  $\text{Ca}^{2+}$  ions.



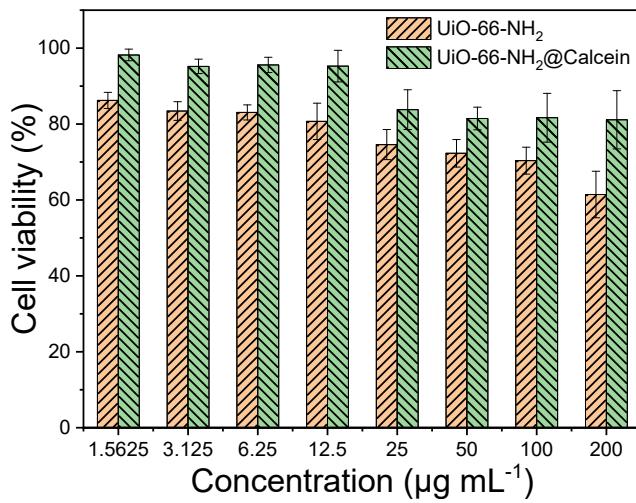
**Figure S9.** Fluorescence emission spectra of UiO-66 (a) and BDC (b) before and after adding Ca<sup>2+</sup> ions under the excitation of 280 nm light.



**Figure S10.** Fluorescence emission spectra of UiO-66-NH<sub>2</sub> and UiO-66 under the excitation of 344 nm light.



**Figure S11.** Cycle experiments of the as-prepared probe. Error bars: standard deviation (SD), experimental times: n = 3.



**Figure S12.** The viabilities of HeLa cell incubated in the suspensions containing various concentrations of Uio-66-NH<sub>2</sub> and Uio-66-NH<sub>2</sub>@Calcein.

**Table S1.** Detection limit comparison of several Ca<sup>2+</sup> ion sensors

Ca <sup>2+</sup> sensor	Detection limit	Ref.
NiCo <sub>2</sub> O <sub>4</sub> /three-dimensional graphene foam	4.45 μM	1
triazolyl coumarin molecule	2.61 μM	2
fluorescent dye oregon-green bapta-1 (OG-dextran)	0.1–100 mM	3
Co <sub>3</sub> O <sub>4</sub>	4 μM	4
Co <sub>3</sub> O <sub>4</sub> conical nanotube on F-doped tin oxide	3.7 μM	5
PC-PDA	0.97 μM	6
<b>UiO-66-NH<sub>2</sub>@Calcein</b>	<b>0.6 μM</b>	<b>this work</b>

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