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SUPPLEMENTARY INFORMATION

Electro-optical π -radicals: design advances, applications and future perspectives

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Radical	$\lambda_{\rm peak}$ (nm) ^a	$(\mathrm{M}^{-1}~\mathrm{cm}^{-1})^{\mathrm{b}}$	λ_{em} (nm) ^c	$\phi_{ m em}$ (%) ^d	τ (ns) ^e	$t_{1/2}$ (s) ^f	$E_{\rm ox}$ (V) ^g	$E_{\rm red}$ (V) ^h	Ref.	Notes
TTM	542 ⁱ 540 ⁱⁱ	690 ⁱⁱ	570 ⁱ 565 ⁱⁱ	2.0 ⁱ 0.8 ⁱⁱ 76 ⁱⁱⁱ	7.0 ⁱ 6.5 ⁱⁱ	224 ⁱ 57 ⁱⁱ	$^{+1.08^{iv}}_{+0.80^{i,v}}$	$\begin{array}{c} -0.83^{iv} \\ -0.99^{i,v} \end{array}$	1-4	ⁱ CH ₂ Cl ₂ ; ⁱⁱ Cy; ⁱⁱⁱ EPA, 77 K; ^{iv} Ag/Ag ⁺ , CH ₂ Cl ₂ / DMF; ^v Fc/Fc ⁺ .
РТМ	564 ⁱ 565 ⁱⁱ	760 ⁱ 910 ⁱⁱ	605 ⁱ 604 ⁱⁱ	1.5 ⁱ 1.6 ⁱⁱ	7 ⁱ 6.1 ⁱⁱ	47 ⁱⁱ	+1.57 ⁱⁱⁱ	-0.46 ⁱⁱⁱ	2,4,5	ⁱ CCl ₄ ; ⁱⁱ Cy; ⁱⁱⁱ Ag/Ag ⁺ , CH ₂ Cl ₂ /DMF.
3I-PTM	400 ⁱ		611 ⁱ	91 ⁱ	69 ⁱ	$\begin{array}{c} 3.2\times10^{7i,ii}\\ 396^{iii} \end{array}$		-0.34 ^{iv}	6	ⁱ solid, ≤0.01% in αH-3I- PTM; ⁱⁱ extrapolated; ⁱⁱⁱ THF; ^{iv} Fc/Fc ⁺ , THF.
РуВТМ	541 ⁱ	1 010 ⁱ	585 ⁱ 674 ⁱⁱ 563 ^{ii, iii}	$\begin{array}{c} 2.2^i \\ 81^{iv} \end{array}$	$\begin{array}{c} 6.4^{i} \\ 19^{ii,\nu} \\ 24^{ii,iii,\nu} \end{array}$	$1.3\times10^{4\mathrm{i}}$	$+1.04^{i, \nu i}$	$-0.74^{i,vi}$	1,7,8	ⁱ CH ₂ Cl ₂ ; ⁱⁱ solid, 10% in αH-PyBTM, 4.2 K; ⁱⁱⁱ 14.5 T; ^{iv} EPA, 77 K; ^v at 563 nm; ^{vi} Fc/Fc ⁺ .
$\begin{array}{l} [(C_6F_5)_3BN-\\ PyBTM] \end{array}$	575 ⁱ		660 ⁱ	3 ⁱ	5.5 ⁱ			$-0.14^{\rm i, ii}$	7	i CH ₂ Cl ₂ ; ii Fc/Fc ⁺ .
bisPyTM	536 ⁱ	≤1 000 ⁱⁱ	650 ⁱ 712 ⁱⁱⁱ	0.9 ⁱ	3.6 ⁱ	6.7×10^{3i}		$-0.57^{i,iv}$	3	ⁱ CH ₂ Cl ₂ ; ⁱⁱ estimated from graph; ⁱⁱⁱ solid, 77 K; ^{iv} Fc/Fc ⁺ .
trisPyM	518 ⁱ	933 ⁱ	700 ⁱ 665 ⁱⁱ	0.9 ⁱ 3.4 ⁱⁱ	3.0 ⁱ 20.0 ⁱⁱ	2.2×10^{4i}		$-0.42^{i,iii}$	9	ⁱ CH ₂ Cl ₂ ; ⁱⁱ solid, 79 K; ⁱⁱⁱ Fc/Fc ⁺ .
trisZn			695 ⁱ	2.0 ⁱ	6.6 ⁱ 9.3 ⁱⁱ 8.9 ^{ii, iii}				9,10	ⁱ solid, 79 K; ⁱⁱ solid, 4.2 K; ⁱⁱⁱ 15 T.
bisZn			ⁱ 626 ^{ii, iii}	<0.1 ⁱ	ⁱ 0.4 ⁱⁱ 3.5 ^{ii, iii}				10	ⁱ solid, 79 K; ⁱⁱ solid, 4.2 K; ⁱⁱⁱ 15 T.
D–A type										
CzBTM	554 ⁱ	3 060 ⁱ	697 ⁱ	2.0 ⁱ 5.0 ⁱⁱ	4.0 ⁱ	8.6×10^{3i}	$\begin{array}{c} -0.06^{iii} \\ -0.09^{i\nu} \end{array}$	$\begin{array}{c} -0.92^{iii} \\ -1.03^{iv} \end{array}$	11,12	ⁱ Cy; ⁱⁱ solid, 5% in CBP; ⁱⁱⁱ Fc/Fc ⁺ , DMF; ^{iv} Ag/Ag ⁺ , CH ₂ Cl ₂ .
PyID-BTM	550 ⁱ	5 290 ⁱ	664 ⁱ	19.5 ⁱ 13.7 ⁱⁱ	12.8 ⁱ	4.0×10^{3i}	+0.03 ⁱⁱⁱ	-1.09 ⁱⁱⁱ	12	ⁱ Cy; ⁱⁱ solid, 2.5% in CBP; ⁱⁱⁱ Ag/Ag ⁺ , CH ₂ Cl ₂ .
TTM-1Cz	603 ⁱ 600 ⁱⁱ	3 780 ⁱ	628 ⁱ 687 ⁱⁱ	53 ⁱ 5 ⁱⁱ	41.3 ⁱ 13.2 ⁱⁱ		+0.57 ⁱⁱⁱ	-0.98 ⁱⁱⁱ	13-15	ⁱ Cy; ⁱⁱ CHCl ₃ ; ⁱⁱⁱ Fc/Fc ⁺ (-0.46 V, SCE), CH ₂ Cl ₂ .
TTM-2Cz	609 ⁱ 607 ⁱⁱ	5 130 ⁱ	680 ⁱ 651 ⁱⁱ	10 ⁱ 54 ⁱⁱ			+0.50 ⁱⁱⁱ	-0.99 ⁱⁱⁱ	16	${}^{i} CHCl_{3}; {}^{ii} Cy {}^{iii} Fc/Fc^{+} \\ (-0.46 V, SCE), CH_{2}Cl_{2}.$
TTM-3Cz	614 ⁱ 611 ⁱⁱ	7 000 ⁱ	680 ⁱ 654 ⁱⁱ	7 ⁱ 52 ⁱⁱ			+0.44 ⁱⁱⁱ	-0.98 ⁱⁱⁱ	16	ⁱ CHCl ₃ ; ⁱⁱ Cy ⁱⁱⁱ Fc/Fc ⁺ (-0.46 V, SCE), CH ₂ Cl ₂ .
TTM-1BiCz	601 ⁱ		671 ⁱ	17 ⁱ	9.9 ⁱ	6.2×10^{4i}			17	ⁱ Cy.
TTM-2BiCz	617 ⁱ		681 ⁱ	11 ⁱ	7.1 ⁱ	3.1×10^{4i}			17	ⁱ Cy.
TTM-TCz	612 ⁱ		689 ⁱ	26 ⁱ	17.2 ⁱ	$8.2\times10^{4\mathrm{i}}$			17	ⁱ Cy.
TTM-αPyID	575 ⁱ 579 ⁱⁱ		599 ⁱ 622 ⁱⁱ	63 ⁱ 91 ⁱⁱ	$\begin{array}{c} 40.9^{\mathrm{i}}\\ 38.8^{\mathrm{ii}} \end{array}$	$1.2\times 10^{4\text{ii}}$	+0.44 ⁱⁱⁱ	-0.97 ⁱⁱⁱ	15	ⁱ Cy; ⁱⁱ CHCl ₃ ; ⁱⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .
TTM-βPyID	582 ⁱ 581 ⁱⁱ		610 ⁱ 637 ⁱⁱ	98 ⁱ 89 ⁱⁱ	43.6 ⁱ 36.1 ⁱⁱ	$7.9\times10^{5\text{ii}}$	+0.47 ⁱⁱⁱ	-0.93 ⁱⁱⁱ	15	$\label{eq:constraint} \stackrel{\mathrm{i}}{}^{\mathrm{i}} \mathrm{Cy}; \stackrel{\mathrm{ii}}{}^{\mathrm{ii}} \mathrm{Fc/Fc^{+}}, \\ \mathrm{CH_2Cl_2}.$
TTM-γPyID	585 ⁱ 590 ⁱⁱ		598 ⁱ 611 ⁱⁱ	37 ⁱ 32 ⁱⁱ	36.8 ⁱ 37.6 ⁱⁱ	$9.5\times10^{2\text{ii}}$	+0.55 ⁱⁱⁱ	-0.92 ⁱⁱⁱ	15	ⁱ Cy; ⁱⁱ CHCl ₃ ; ⁱⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .
TTM-δPyID	585 ⁱ 581 ⁱⁱ		614 ⁱ 643 ⁱⁱ	89 ⁱ 99 ⁱⁱ	42.7 ⁱ 35.1 ⁱⁱ	$9.1\times10^{5\text{ii}}$	+0.50 ⁱⁱⁱ	-0.92 ⁱⁱⁱ	15	ⁱ Cy; ⁱⁱ CHCl ₃ ; ⁱⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .
TTM-3PCz	593 ⁱ	3 250 ⁱ	664 ⁱ 695 ⁱⁱⁱ	29 ⁱ 46 ⁱⁱ 60 ⁱⁱⁱ	15.9 ⁱ 21.2 ⁱⁱ		+0.43 ^{iv}	-1.10 ^{iv}	2,18	ⁱ Cy; ⁱⁱ Tol; ⁱⁱⁱ solid, 3% in CBP; ^{iv} Fc/Fc ⁺ , CH ₂ Cl ₂ .

Table S1 Optical and electrochemical redox properties of π -radicals. The structures correspond to those shown in the main manuscript (additional structures are shown in Scheme S1 and S2).

TTM-3NCz	592 ⁱ 616 ⁱⁱ		667 ⁱ 707 ⁱⁱⁱ	29 ⁱ 49 ⁱⁱ 86 ⁱⁱⁱ	15.9 ⁱ 17.2 ⁱⁱ	$\geq 1.0 \times 10^{4 \text{ i, iv}}$	+0.71 ^v +0.43 ^{vi}	-0.86^{v} -1.09^{vi}	4,18	ⁱ Cy; ⁱⁱ Tol; ⁱⁱⁱ solid, 3% in CBP; ^{iv} estimated from graph; ^v Ag/Ag ⁺ , CH ₂ Cl ₂ / DMF; ^{vi} Fc/Fc ⁺ , CH ₂ Cl ₂ .
TTM-PPTA	549 ⁱ		775 ⁱ	2 ⁱ	6.8 ⁱ	$\geq 3.5 \times 10^{3i,ii}$	+0.57 ⁱⁱⁱ	-0.84 ⁱⁱⁱ	4	ⁱ Cy; ⁱⁱ estimated from graph; ⁱⁱⁱ Ag/Ag ⁺ , CH ₂ Cl ₂ / DMF.
PTM-3NCz	607 ⁱ		680 ⁱ	54 ⁱ	26.3 ⁱ	9.1×10^{6i}	+1.24 ⁱⁱ	-0.58 ⁱⁱ	4	ⁱ Cy; ⁱⁱ Ag/Ag ⁺ , CH ₂ Cl ₂ / DMF.
PTM-PCz	566 ⁱ	1 750 ⁱ	673 ⁱ	44 ⁱ	26.7 ⁱ	2.4×10^{3i}			2,4	ⁱ Cy.
PTM-3PCz	606 ⁱ	2 680 ⁱ	679 ⁱ	57 ⁱ	26.3 ⁱ	8.7×10^{3i}			2,4	ⁱ Cy.
PTM-TPA	680 ⁱ	1 960	767 ⁱ	26 ⁱ	15.3 ⁱ	3.2×10^{5i}			2	ⁱ Cy.
PTM-TPA"	787 ⁱ	1 300 ⁱ					+0.31 ⁱⁱ	-0.65^{ii}	19,20	i Cy; ii Fc/Fc ⁺ , CH ₂ Cl ₂ .
PTM-PDCz	562 ⁱ		664 ⁱ	15 ⁱ	142.0 ⁱ	$\geq\!\!3.5\times10^{3i,ii}$	+1.26 ⁱⁱⁱ	-0.50 ⁱⁱⁱ	4	ⁱ Cy; ⁱⁱ estimated from graph; ⁱⁱⁱ Ag/Ag ⁺ , CH ₂ Cl ₂ / DMF.
TPAN-PTM	973 ⁱ	3 200 ⁱ					$-0.21^{i,ii}$	$-0.73^{\mathrm{i,ii}}$	21	ⁱ CH ₂ Cl ₂ ; ⁱⁱ Fc/Fc ⁺ .
TOTA-PTM	819 ⁱ	1 600 ⁱ					+0.12 ^{i, ii}	$-0.71^{i,ii}$	21	i CH ₂ Cl ₂ ; ii Fc/Fc ⁺ .
PyPBTM	567 ⁱ	≤2 000 ^{i, ii}	650 ⁱ 644 ⁱⁱⁱ	8.6 ⁱ 22 ⁱⁱⁱ	12.0 ⁱ 21.5 ⁱⁱⁱ	$1.9\times10^{4\mathrm{iv}}$	+0.73 ^{iv, v}	$-0.84^{iv, v}$	22	ⁱ CHCl ₃ ; ⁱⁱ estimated from graph; ⁱⁱⁱ EPA, 77 K; ^{iv} CH ₂ Cl ₂ ; ^v Fc/Fc ⁺ .
PyNBTM	618 ⁱ	$\leq 4\ 000^{i,ii}$	735 ⁱ 714 ⁱⁱⁱ	6.4 ⁱ 16 ⁱⁱⁱ	3.9 ⁱ 7.8 ⁱⁱⁱ	$2.3\times10^{4\mathrm{iv}}$	+0.57 ^{iv, v}	-0.86 ^{iv, v}	22	ⁱ CHCl ₃ ; ⁱⁱ estimated from graph; ⁱⁱⁱ EPA, 77 K; ^{iv} CH ₂ Cl ₂ ; ^v Fc/Fc ⁺ .
PyPhBTM	585 ⁱ	≤2 000 ^{i, ii}	729 ⁱ 688 ⁱⁱⁱ	$\begin{array}{c} 0.9^{\mathrm{i}} \\ 7^{\mathrm{iii}} \end{array}$	0.8 ⁱ 5.6 ⁱⁱⁱ	$2.6\times10^{5\mathrm{iv}}$	+0.76 ^{iv, v}	$-0.84^{iv, v}$	22	ⁱ CHCl ₃ ; ⁱⁱ estimated from graph; ⁱⁱⁱ EPA, 77 K; ^{iv} CH ₂ Cl ₂ ; ^v Fc/Fc ⁺ .
TPA-R'	710 ⁱ	1 440 ⁱ	910 ⁱ	0.1 ⁱ	0.13 ⁱ	$\geq 9.0 \times 10^{2i,ii}$	+0.38 ⁱⁱⁱ	-0.69 ⁱⁱⁱ	23	ⁱ Cy; ⁱⁱ estimated from graph; ⁱⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .
TPA-[RH]++	700–1400 ^{i, ii}						+0.44 ⁱⁱⁱ	+0.17 ⁱⁱⁱ	23	ⁱ Cy; ⁱⁱ estimated from graph; ⁱⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .
S1	577 ⁱ 567 ⁱⁱ		611 ⁱ 594 ⁱⁱ	41 ⁱ	35.7 ⁱ	1.7×10^{3i}			24	ⁱ Tol; ⁱⁱ Cy.
S2	587 ⁱ 577 ⁱⁱ		629 ⁱ 610 ⁱⁱ	87 ⁱ	33.8 ⁱ	2.9×10^{3i}			24	ⁱ Tol; ⁱⁱ Cy.
S 3	592 ⁱ 580 ⁱⁱ		632 ⁱ 616 ⁱⁱ	84 ⁱ	32.7 ⁱ	5.9×10^{3i}			24	ⁱ Tol; ⁱⁱ Cy.
S 4	575 ⁱ	1 300 ⁱ	725 ⁱ	37 ⁱ	4.9 ⁱ				20	ⁱ Cy.
85	833 ^{i, ii}	$\leq\!\!10\;000^{i,iii}$	847 ⁱ	2.8 ⁱ	1.9 ⁱ		+0.20 ^{iv}	-0.95 ^{iv}	25	ⁱ Cy; ⁱⁱ onset of absorption; ⁱⁱⁱ estimated from graph; ^{iv} Fc/Fc ⁺ , CH ₂ Cl ₂ .
S6	787 ^{i, ii}	≥10 000 ^{i, iii}	775 ⁱ	6.0 ⁱ	6.5 ⁱ		+0.30 ^{iv}	-0.94^{iv}	25	ⁱ Cy; ⁱⁱ onset of absorption; ⁱⁱⁱ estimated from graph; ^{iv} Fc/Fc ⁺ , CH ₂ Cl ₂ .
87	752 ^{i, ii}	≤20 000 ^{i, iii}	730 ⁱ	6.1 ⁱ	5.8 ⁱ		+0.41 ^{iv}	-0.89 ^{iv}	25	ⁱ Cy; ⁱⁱ onset of absorption; ⁱⁱⁱ estimated from graph; ^{iv} Fc/Fc ⁺ , CH ₂ Cl ₂ .
S 8	538 ^{i,}	682 ⁱ	571 ⁱ 785 ⁱⁱⁱ	1.7 ⁱ 41 ^{i, ii} 1.0 ⁱⁱⁱ	5.4 ⁱ 105 ⁱ 20.2 ⁱⁱⁱ	34 ⁱ		$-0.93^{\rm i,iv}$	26	ⁱ CH ₂ Cl ₂ ; ⁱⁱ 210 K; ⁱⁱⁱ solid crystal; ^{iv} Fc/Fc ⁺ .
89	400–590 ^{i, ii}	≤1 000 ^{i, ii}	≥705 ^{i, ii}	$\begin{array}{c} 0.1^{i} \\ 3^{iii} \end{array}$	0.6 ⁱ	$3.1\times10^{4\mathrm{i}}$			27	ⁱ Cy; ⁱⁱ estimated from graph; ⁱⁱⁱ EPA, 77 K.
S10	400–590 ^{i, ii}	$\leq 1\ 000^{i,ii}$	≥675 ^{i, ii}	0.1 ⁱ 5 ⁱⁱⁱ	0.25 ⁱ	6.3×10^{3i}			27	ⁱ Cy; ⁱⁱ estimated from graph; ⁱⁱⁱ EPA, 77 K.

S11	600 ⁱ	3 200 ⁱ						$-0.69^{i,ii}$	28	ⁱ CH ₂ Cl ₂ ; ⁱⁱ Fc/Fc ⁺ (-0.46 V, SCE).	
S12	573 ⁱ	3 100 ⁱ						$-0.69^{i,ii}$	28	ⁱ CH ₂ Cl ₂ ; ⁱⁱ Fc/Fc ⁺ .	
813	567 ⁱ	1 500 ⁱ						$-0.78^{i,\mathrm{ii}}$	28	i CH ₂ Cl ₂ ; ii Fc/Fc ⁺ .	
Non-chlorinated											
1	1020 ⁱ	≥200 ^{i, ii}					-0.19 ^{i, iii}	$-1.48^{i,iii}$	29	ⁱ CH ₂ Cl ₂ , RT; ⁱⁱ estimated from graph; ⁱⁱⁱ Fc/Fc ⁺ .	
2	395 ⁱ	520 ⁱ					+0.40 ⁱⁱ	-1.34 ⁱⁱ	30	ⁱ Hex, vacuum; ⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ , 195 K.	
BDPA	859 ⁱ	1 580 ⁱ	^{i, ii}				+0.21 ^{i, iii}	$-1.05^{\mathrm{i,iii}}$	31	i CH ₂ Cl ₂ ; ii RT or 77 K; iii Fc/Fc ⁺ (-0.46 V, SCE).	
8	626 ⁱ	2 720 ⁱ					$+0.27^{i,ii}$	$-1.01^{i,ii}$	32	ⁱ CH ₂ Cl ₂ ; ⁱⁱ Fc/Fc ⁺ .	
9	1220 ⁱ	100 ⁱ					+0.37 ^{i, ii}	$-0.90^{\mathrm{i,ii}}$	33	ⁱ CH ₂ Cl ₂ ; ⁱⁱ Fc/Fc ⁺ .	
10	669 ⁱ	2 100 ⁱ					+0.13 ^{i, ii}	-1.23 ^{i, ii}	33	ⁱ CH ₂ Cl ₂ ; ⁱⁱ Fc/Fc ⁺ .	
11	>1000 ^{i ii} 794 ^{i, iii}	^{i ii} 10 000 ^{i, iii}					+0.29 ^{i, iv}	$-0.82^{i,iv}$	33	ⁱ CH ₂ Cl ₂ ; ⁱⁱ weak, estimated from graph; ⁱⁱⁱ second-lowest energy absorption; ^{iv} Fc/Fc ⁺ .	
12	824 ⁱ	≥30 000 ^{i, ii}					+0.46 ^{i, iii}	$\begin{array}{c} -0.01^{\rm i, iii} \\ -0.24^{\rm i, iii} \end{array}$	34	ⁱ CH ₂ Cl ₂ ; ⁱⁱ estimated from graph; ⁱⁱⁱ Fc/Fc ⁺ .	
S14	1100 ⁱ	$\leq 300^{i, ii}$					-0.15 ^{i, iii}	-1.35 ^{i, iii}	35	ⁱ CH ₂ Cl ₂ , RT; ⁱⁱ estimated from graph; ⁱⁱⁱ Fc/Fc ⁺ .	
Delocalised											
Phenalenyl	540 ^{i, ii} 595 ^{i, iii}	103 ^{i, ii} 20 000 ^{i, iii}					+0.32 ^{iv, v}	-1.28 ^{iv, v}	36,37	ⁱ CH ₂ Cl ₂ ; ⁱⁱ 278 K; ⁱⁱⁱ 191 K; ^{iv} Fc/Fc ⁺ (-0.38 V, SCE), CH ₃ CN; ^v E _{ax} and E _{red} were obtained from first and second reduction of cation of phenalenyl salt, respectively.	
18	595i 613 ⁱⁱ						+0.27 ^{iii, iv}	-1.26 ^{iii, iv}	38-40	ⁱ Hex, 200 K; ⁱⁱ solid, in KBr; ⁱⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ ; ^{iv} E _{ax} and E _{red} were obtained from first and second reduction of cation of 18 salt, respectively.	
19	642 ⁱ 650 ⁱⁱ							-0.30 ^{iii, iv}	41-43	ⁱ Hex; ⁱⁱ solid, in KBr; ⁱⁱⁱ Fc/Fc ⁺ , CH ₃ CN; ^{iv} obtained from oxidation of anion of 19 salt.	
21	852 ⁱ	≥200 ^{i, ii}					+0.56 ^{iii, iv}	+0.09 ^{iii, iv}	44	ⁱ CH ₂ Cl ₂ ; ⁱⁱ estimated from graph; ⁱⁱⁱ Fc/Fc ⁺ , DMF; ^{iv} E_{red} and E_{ox} were obtained from first and second oxidation of anion of 21 salt, respectively.	
ТОТ	834 ⁱ 1134 ⁱⁱ	≥2 000 ^{i, iii} ≥9 000 ^{i, iv}					+0.79 ^{v, vi}	-0.35 ^{v, vi}	45,46	ⁱ CHCl ₃ ; ⁱⁱ solid, in KBr; ⁱⁱⁱ estimated from graph, 300 K; ^{iv} estimated from graph, 215 K; ^v Fc/Fc ⁺ , CH ₂ Cl ₂ ; ^{vi} E _{red} and E _{ox} were obtained from first and second oxidation of anion of TOT salt, respectively.	
22							-0.39 ^{i, ii}	-1.42 ^{i, ii}	47	ⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ ; ⁱⁱ <i>E_{ax}</i> and <i>E_{red}</i> were obtained from first and second reduction of cation of 22 salt, respectively.	

23	1355 ⁱ	7 910 ⁱ							47	ⁱ Tol.		
S15	848 ⁱ 876 ⁱⁱ	i, iii 9 700 ^{i, iv}					+0.75 ^{i, v, vi}	-0.33 ^{i, v, vi}	46	ⁱ CH ₂ Cl ₂ ; ⁱⁱ solid, in KBr; ⁱⁱⁱ 300 K, no peak; ^{iv} 180 K; ^v Fc/Fc ⁺ ; ^{vi} E _{red} and E _{ox} were obtained from first and second oxidation of anion of S12 salt, respectively.		
S16	1416 ⁱ						+0.74 ^{ii, iii}	+0.03 ^{ii, iii}	46	ⁱ solid, in KBr; ⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ ; ⁱⁱⁱ E_{red} and $E_{\alpha\alpha}$ were obtained from first and second oxidation of anion of S13 salt, respectively.		
S17	685 ⁱ	≥4 000 ⁱⁱ					-0.47 ^{i, iii}	-1.46 ^{i, iii}	48	ⁱ CH ₂ Cl ₂ ; ⁱⁱ estimated from graph; ⁱⁱⁱ Fc/Fc ⁺ .		
S18							-0.35	-1.74	40	1 Fc/Fc $^{+}$, CH ₂ Cl ₂ .		
Diradicals												
26	690 ^{i,} ≤900 ^{i, ii}	≥40 000 ^{i, iii} ^{i, ii}					$^{+0.17^{i,iv}}_{+0.70^{i,iv}}$	$\begin{array}{c} -1.28^{i,iv} \\ -1.67^{i,iv} \end{array}$	49	ⁱ CH ₂ Cl ₂ ; ⁱⁱ weak shoulder; ⁱⁱⁱ estimated from graph; ^{iv} Fc/Fc ⁺ .		
27	537 ⁱ 730 ^{i, ii}	15 200 ⁱ 790 ^{i, ii}					+0.59 ^{i, iii}	$-1.51^{i,iii}$	50	ⁱ CH ₂ Cl ₂ ; ⁱⁱ shoulder; ⁱⁱⁱ Fc/Fc ⁺ .		
28	697 ⁱ ≤1050 ^{i, ii}	12 800 ⁱ					$^{+0.38^{i,iii}}_{+0.77^{i,iii}}$	$\begin{array}{c} -1.22^{i,iii} \\ -1.61^{i,iii} \end{array}$	51	ⁱ CH ₂ Cl ₂ ; ⁱⁱ weak shoulder; ⁱⁱⁱ Fc/Fc ⁺ .		
29	638 ⁱ 1700 ⁱ	$\substack{\geq 35\ 000^{i,ii} \\ \geq 700^{i,ii}}$					$^{+0.13^{i,iii}}_{+0.90^{i,iii}}$	$\begin{array}{c} -1.13^{\rm i, iii} \\ -2.03^{\rm i, iii} \end{array}$	52	ⁱ CH ₂ Cl ₂ , RT; ⁱⁱ estimated from graph; ⁱⁱⁱ Fc/Fc ⁺ .		
30	950 ^{i, ii}						$^{+0.33^{i,iii}}_{+0.77^{i,iii}}$	$\begin{array}{c} -0.60^{\rm i,iii} \\ -1.20^{\rm i,iii} \end{array}$	53	ⁱ CH ₂ Cl ₂ ; ⁱⁱ estimated from graph; ⁱⁱⁱ Fc/Fc ⁺ .		
31	822 ⁱ 654 ⁱⁱ	25 860 ⁱ 15 030 ⁱⁱ					$+0.47^{i,iii}$ +0.86 ^{i,iii} +0.87 ^{ii,iii}	-1.36 ^{i, iii} -1.46 ^{i, iii} -1.50 ^{ii, iii}	54	ⁱ CH ₂ Cl ₂ ; ⁱⁱ DMSO; ⁱⁱⁱ Fc/Fc ⁺ .		
36	1130 ⁱ	^{i, ii} ≥7 000 ^{i, iii}					$+0.38^{i,iv}$ +0.52 ^{i, iv}	-0.98 ^{i, iv}	55	ⁱ CH ₂ Cl ₂ ; ⁱⁱ weak, diradical; ⁱⁱⁱ FeCl ₃ (2 equiv.), dication of 36 ; ^{iv} Fc/Fc ⁺ .		
37	600 ⁱ ≤1050 ^{i, ii}	41 000 ⁱ					$^{+0.26^{i,iii}}_{+0.66^{i,iii}}$	$\begin{array}{c} -1.13^{i,iii} \\ -1.42^{i,iii} \end{array}$	56	ⁱ CH ₂ Cl ₂ ; ⁱⁱ weak shoulder; ⁱⁱⁱ Fc/Fc ⁺ .		
38	444 ^{i, ii} 757 ^{i, iii}						$\begin{array}{c} +0.27^{i,iii,iv} \\ +0.38^{i,iii,iv} \\ +0.94^{i,ii,iv} \end{array}$	$\begin{array}{c} -0.97^{i,iii,iv} \\ -1.08^{i,iii,iv} \\ -1.74^{i,ii,iv} \end{array}$	57	ⁱ CH ₂ Cl ₂ ; ⁱⁱ folded form; ⁱⁱⁱ twisted form; ^{iv} Fc/Fc ⁺ .		
S19	629 ⁱ	58 900 ⁱ					$^{+0.17^{i,ii}}_{+0.65^{i,ii}}$	$\begin{array}{c} -1.50^{i,ii} \\ -1.87^{i,ii} \end{array}$	58	ⁱ CH ₂ Cl ₂ ; ⁱⁱ Fc/Fc ⁺ .		
S20	688 ⁱ	35 600 ⁱ					$^{+0.26^{i,ii}}_{+0.72^{i,ii}}$	$\begin{array}{c} -1.28^{i,ii} \\ -1.79^{i,ii} \end{array}$	58	ⁱ CH ₂ Cl ₂ ; ⁱⁱ Fc/Fc ⁺ .		
S21	697 ⁱ	50 000 ⁱ					$^{+0.18^{i,ii}}_{+0.65^{i,ii}}$	$\begin{array}{c} -1.30^{i,ii} \\ -1.83^{i,ii} \end{array}$	58	ⁱ CH ₂ Cl ₂ ; ⁱⁱ Fc/Fc ⁺ .		
S22	738 ⁱ	40 300 ⁱ					+0.20 ^{i, ii}	-0.75 ^{i, ii}	58	ⁱ CH ₂ Cl ₂ ; ⁱⁱ Fc/Fc ⁺ .		
823	600–1000 ⁱ						+0.28 ⁱⁱ	-1.13 ⁱⁱ	59	ⁱ CH ₂ Cl ₂ , estimated from graph; ⁱⁱ Fc/Fc ⁺ (-0.46 V, SCE), CH ₂ Cl ₂ .		
S24	746 ⁱ	178 000 ⁱ					$+0.08^{i,ii}$	-1.07 ^{i, ii}	60	ⁱ CH ₂ Cl ₂ ; ⁱⁱ Fc/Fc ⁺ .		
825	756 ⁱ	115 000 ⁱ					$^{+0.11^{i,ii}}_{+0.51^{i,ii}}$	$\begin{array}{c} -1.09^{i,ii} \\ -1.62^{i,ii} \end{array}$	61	ⁱ CH ₂ Cl ₂ ; ⁱⁱ Fc/Fc ⁺ .		
S26	865 ⁱ	78 430 ⁱ					$-0.13^{ii} +0.26^{ii}$	$\begin{array}{c} -1.17^{ii} \\ -1.55^{ii} \end{array}$	62	ⁱ Hex/CH ₂ Cl ₂ 49:1; ⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .		
S27	642 ^{i, ii} 637 ^{i, iii}	$\begin{array}{c} 398\;000^{i,ii} \\ \leq \!\! 10\;000^{I,iii,iv} \end{array}$	699 ^v	<1 ^v			+0.55 ^{v, vi}	$\begin{array}{c} -0.93^{v,vi} \\ -1.92^{v,vi} \end{array}$	63	ⁱ DMF; ⁱⁱ 298 K; ⁱⁱⁱ 383 K; ^{iv} estimated from graph; ^v CH ₂ Cl ₂ ; ^{vi} Fc/Fc ⁺ .		

829	660–1100 ^{i, ii}	ⁱⁱⁱ					$^{+0.11^{i,iv}}_{+0.78^{i,iv}}$	-1.13 ^{i, iv}	64	ⁱ CH ₂ Cl ₂ ; ⁱⁱ estimated from graph; ⁱⁱⁱ weak; ^{iv} Fc/Fc ⁺ .
S30	660–1100 ^{i, ii}	ⁱⁱⁱ					$^{+0.09^{i,iv}}_{+0.78^{i,iv}}$	-1.15 ^{i, iv}	64	ⁱ CH ₂ Cl ₂ ; ⁱⁱ estimated from graph; ⁱⁱⁱ weak; ^{iv} Fc/Fc ⁺ .
Polyradical	5									
42	820 ⁱ 794 ⁱⁱ 848 ⁱⁱⁱ	6 500 ⁱ 5 600 ⁱⁱ					+0.34 ^{ii, iv}	$-0.69^{ii,iv}$	65	ⁱ Tol; ⁱⁱ CH ₂ Cl ₂ ; ⁱⁱⁱ solid film; ^{iv} Fc/Fc ⁺ .
PS-CzTTM	622 ⁱ	5 430 ⁱ	660 ⁱ 694 ⁱⁱ	38 ⁱ 25 ⁱⁱ	19.2 ⁱ 10.1 ⁱⁱ	$1.6\times10^{4\mathrm{i}}$	+0.41 ⁱⁱⁱ	-0.97 ⁱⁱⁱ	66	ⁱ Cy; ⁱⁱ solid film, RT; ⁱⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .
FR-1	768 ⁱ	800 ⁱ					+0.12 ⁱⁱ	-1.10 ⁱⁱ	67	ⁱ Tol; ⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .
FR-2	745 ⁱ	19 100 ⁱ					+0.17 ⁱⁱ +0.31 ⁱⁱ	-1.13 ⁱⁱ -1.32 ⁱⁱ	67	ⁱ Tol; ⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .
FR-3	955 ⁱ	11 300 ⁱ					+0.05 ⁱⁱ +0.32 ⁱⁱ	$-1.01^{ii} \\ -1.20^{ii}$	67	ⁱ Tol; ⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .
FR-4	735 ⁱ	22 100 ⁱ					+0.13 ⁱⁱ +0.45 ⁱⁱ	$-1.09^{ii} \\ -1.30^{ii}$	67	ⁱ Tol; ⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .
FR-5	968 ⁱ	14 200 ⁱ					$+0.00^{ii}$ $+0.16^{ii}$	$-1.01^{ii} \\ -1.10^{ii}$	67	ⁱ Tol; ⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .
FR-6	739 ⁱ	26 400 ⁱ					$^{+0.02^{ii}}_{+0.15^{ii}}$	$-1.07^{ m ii}\ -1.18^{ m ii}$	67	ⁱ Tol; ⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .
FR-MC4	710 ⁱ 1074 ^{i, ii}	≤60 000 ^{i, iii} ^{i, ii}					$^{+0.01^{iv}}_{+0.55^{iv}}$	$^{-1.07^{iv}}_{-1.50^{iv}}$	68	ⁱ Tol; ⁱⁱ weak shoulder; ⁱⁱⁱ estimated from graph; ^{iv} Fc/Fc ⁺ , CH ₂ Cl ₂ .
FR-MC5	750 ⁱ 966 ^{i, ii}	$\leq 30 \underset{_i, ii}{000^{i, iii}}$					$\begin{array}{c} -0.04^{iv} \\ +0.04^{iv} \end{array}$	$\begin{array}{c} -1.01^{iv} \\ -1.19^{iv} \end{array}$	68	ⁱ Tol; ⁱⁱ weak shoulder; ⁱⁱⁱ estimated from graph; ^{iv} Fc/Fc ⁺ , CH ₂ Cl ₂ .
FR-MC6	772 ⁱ 957 ^{i, ii}	$\geq\!\!20000^{i,iii}_{-_^{i,ii}}$					$^{+0.03^{iv}}_{+0.12^{iv}}$	${}^{-1.07^{iv}}_{-1.23^{iv}}$	68	ⁱ Tol; ⁱⁱ weak shoulder; ⁱⁱⁱ estimated from graph; ^{iv} Fc/Fc ⁺ , CH ₂ Cl ₂ .
45	1650 ⁱ						$\begin{array}{c} -0.16^{ii} \\ -0.05^{ii} \end{array}$	$-0.99^{ii} \\ -1.05^{ii}$	69	ⁱ CDCl ₃ , weak shoulder; ⁱⁱ Fc/Fc ⁺ , CH ₂ Cl ₂ .
RP-T	865 ⁱ	≥2 000 ^{i, ii}					+0.70 ⁱⁱⁱ	-0.82 ⁱⁱⁱ	70	ⁱ THF; ⁱⁱ estimated from graph; ⁱⁱⁱ solid film, Fc/Fc ⁺ , CH ₃ CN.
RP-BT	864 ⁱ	≥250 ^{i, ii}							70	ⁱ THF; ⁱⁱ estimated from graph.
RP-TT	857 ⁱ	$\geq 1~000^{i,ii}$							70	ⁱ THF; ⁱⁱ estimated from graph.

^a Lowest-energy absorption peak wavelength. ^b Molar extinction coefficient at the peak wavelength. ^c Photoluminescence peak wavelength. ^d Photoluminescence quantum yield. ^e Luminescence lifetime. ^f Photostability (half-life of emission intensity). ^g Electrochemical oxidation potential. ^h Electrochemical reduction potential. Redox potentials are reported *vs*. ferrocene/ferrocenium (Fc/Fc⁺) or Ag/Ag⁺ redox couple as obtained by cyclic voltammetry and noted in the table. Up to four redox reactions are reported for diradicals and polyradicals. If potential values were referenced to standard calomel electrode (SCE), conversion values of -0.46 (V *vs*. Fc/Fc⁺ in CH₂Cl₂)⁷¹ and -0.38 (V *vs*. Fc/Fc⁺ in CH₃CN)⁷² were used, as noted in the table. All data have been obtained at room temperature, unless noted otherwise. Key: --, not reported; Cy, cyclohexane; Hex, hexane; Tol, toluene; EPA, diethyl ether/isopentane/ethanol 5:5:2 v/v; DMSO, dimethyl sulfoxide; DMF, *N*,*N*-dimethylformamide; THF, tetrahydrofuran; CBP, 4,4'di(9*H*-carbazol-9-yl)-1,1'-biphenyl.



Scheme S1 Chemical structures of additional π -radicals listed in Table S1.



Scheme S2 Chemical structures of additional π -radicals, diradicals and polyradicals listed in Table S1 (R groups are omitted from some resonance structures for clarity).

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