Supporting Information for: Metallization and superconductivity of NBH₁₂ compounds stabilized by dihydrogen bonds

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Figure S1. Phonon dispersions of $NH_3BH_3(H_2)_3$ at 20, 100, 240, 250 GPa.



Figure S2. The enthalpy difference curves with respect to AB(H₂)₃ from divergent ideal routes as a function of pressure.



Figure S3. The lattice constants of $P2_1$ -NH₃BH₃(H₂)₃ as a function of pressure.



Figure S4. The tilt angle of the NH₃BH₃ molecular layers in *P*2₁-NH₃BH₃(H₂)₃ and pure NH₃BH₃ as a function of pressure.



Figure S5. The dihydrogen bond length as a function of bond angle between AB and H_2 molecules.



Figure S6. The dihydrogen bonding between AB and H_2 molecules



Figure S7. The dihydrogen bonding distance as a function of pressure. (a) The dihydrogen bond between AB in $AB(H_2)_3$. The dihydrogen bond between AB and H_2 in $AB(H_2)_3$.



Figure S8. The atom-projected electron band structure and density of states (PDOS) of $P2_1$ -NH₃BH₃(H₂)₃ at 150 GPa, 175 GPa, 200 GPa and 225 GPa are spectively shown in (a), (b), (c) and (d). The Fermi level is set to zero and depicted as the gray line. The red, blue, and green dots respectively represent the contributions of H, B, and N atoms to the electron band structure.

Compound	Space group	Lattice	Atomic coordinates (fractional)			
		parameters (Å)	atoms	X	Y	Z
NH3BH3(H2)3	<i>P</i> 2 ₁		H1(2a)	1.07438	0.58226	-0.37043
			H2(2a)	0.57949	0.58556	-0.30810
			H3(2a)	0.42703	0.08214	-0.24234
			H4(2a)	1.11350	0.15920	-0.31033
		a =3.7998	H5(2a)	0.97738	0.39077	-0.50924
		b =6.618	H6(2a)	0.53444	0.26260	-0.49728
		c =3.4681	H7(2a)	1.02686	0.91098	-0.03396
			H8(2a)	0.74368	0.35328	-0.11938
		<i>α</i> =γ=90°	H9(2a)	0.89938	0.18044	-0.20690
		$\beta = 62.866$	H10(2a)	0.51233	0.86844	-0.49233
			H11(2a)	1.23237	0.22976	-0.89500
			H12(2a)	0.18445	0.59830	-0.24966
			B(2a)	0.47333	0.98709	0.01060
			N(2a)	0.84543	0.87670	-0.16776

Table S1 Calculated structural parameters of the predicted phase of $NH_3BH_3(H_2)_3$ at 100 GPa.