

Supplementary Information for

Transforming polyethylene and polypropylene into nontraditional fluorescent polymers by thermal oxidation

Yaxin Zhao, Jiayu Long, Peifeng Zhuang, Ying, Ji, Changcheng He, and Huiliang Wang*

Beijing Key Laboratory of Energy Conversion and Storage Materials, College of Chemistry, Beijing Normal University, Beijing 100875, China

*Corresponding author: wanghl@bnu.edu.cn

Contents

Fig. S1 The fluorescence lifetimes of PE-280-120 and PP-340-30 under different excitation wavelengths.

Fig. S2 The FTIR spectra of the carbonyl group regions of the PE (a) and PP (b) samples heated in air at 280°C for different times, respectively.

Fig. S3 XPS profiles of PE (a), PP (b) and their oxidized products heated in air at different temperatures for different times.

Fig. S4 Curve-fitted XPS profiles of PE and PP and their oxidized products heated in air at different temperature for different time.

Table S1 The quantum yields (ϕ) of the PE and PP oxidized products.

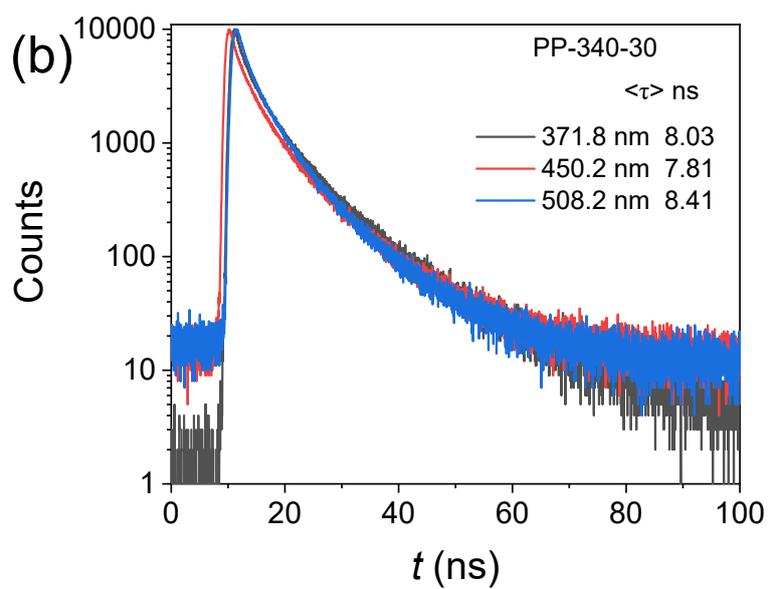
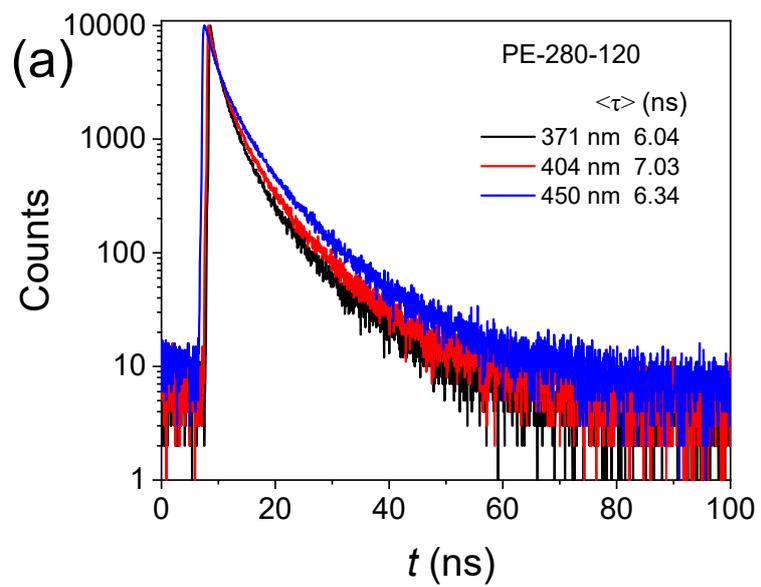


Fig. S1 The fluorescence lifetimes of PE-280-120 and PP-340-30 under different excitation wavelengths.

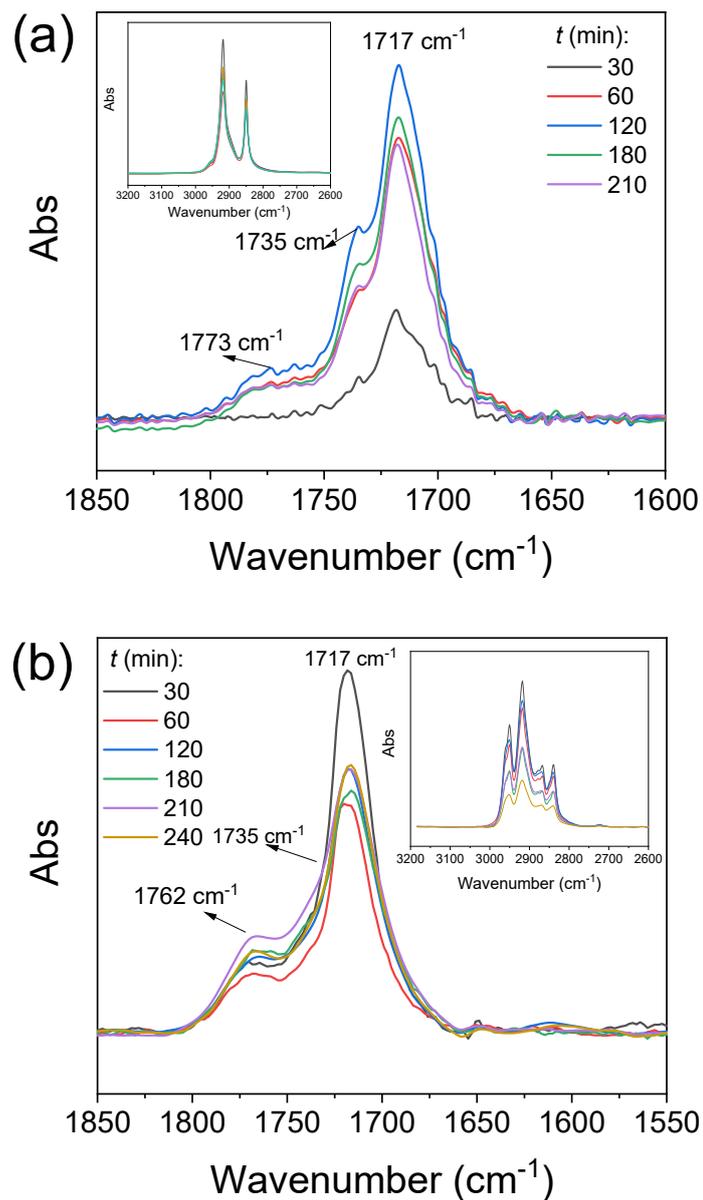


Fig. S2 The FTIR spectra of the carbonyl group regions of the PE (a) and PP (b) samples heated in air at 280°C for different times, respectively. The insets in (a) and (b) are the FTIR spectra of C-H stretching of PE and PP, respectively.

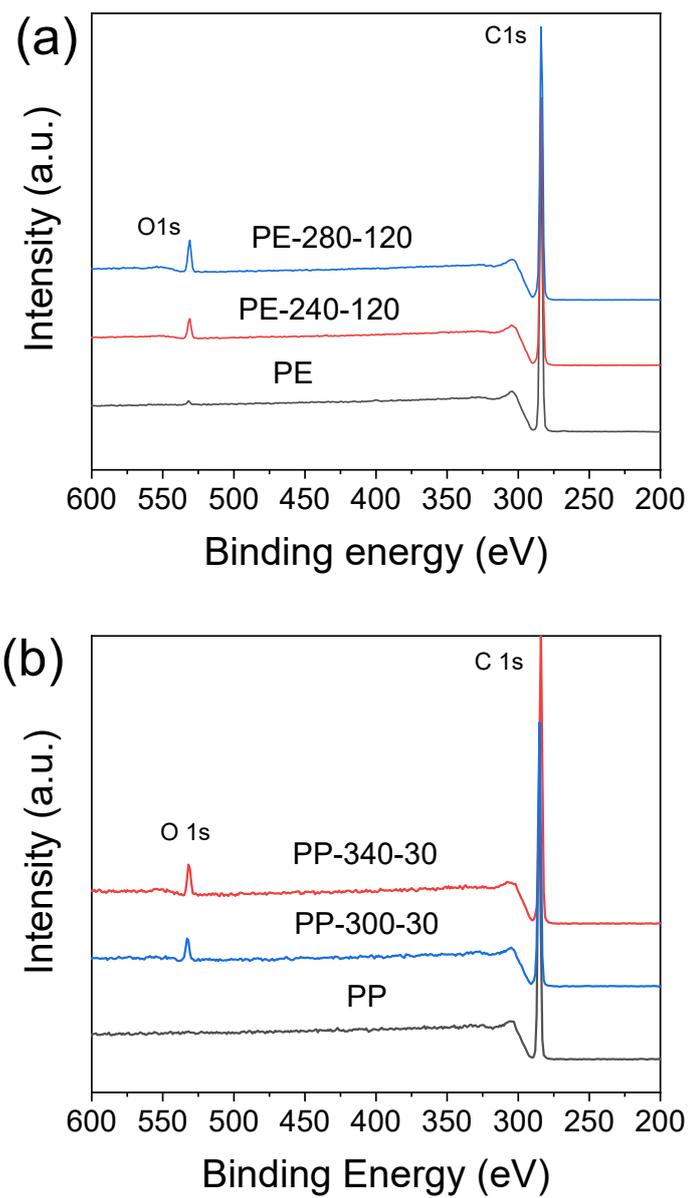


Fig. S3 XPS profiles of PE (a), PP (b) and their oxidized products heated in air at different temperatures for different times.

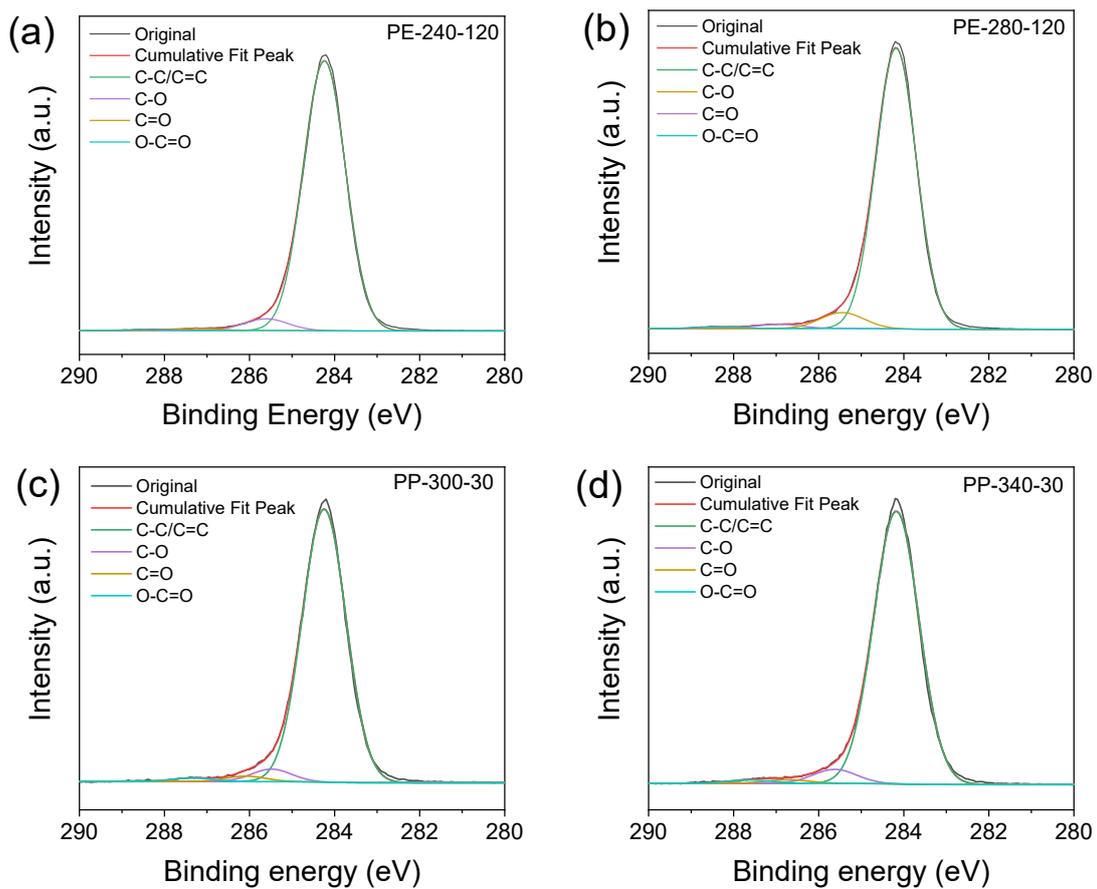


Fig. S4 Curve-fitted XPS profiles of PE and PP and their oxidized products heated in air at different temperature for different time.

Table S1 The quantum yields (ϕ) of the oxidized PE and PP products.

Sample	ϕ (%)
PE-280-30	4.9
PE-280-120	5.2
PE-280-240	2.7
PE-240-120	2.8
PE-320-120	2.4
PP-280-30	3.0
PP-300-30	2.6
PP-340-30	6.3
PP-360-30	3.6
PP-280-60	4.3
PP-280-120	4.1
PP-280-180	6.0
PP-280-210	2.5
PP-280-240	1.6