

## Supporting Information

### Coordination Units of Mn<sup>2+</sup> Modulation toward Tunable Emission in Zero-Dimensional Bromides for White Light-Emitting Diodes

Guojun Zhou<sup>\*a#</sup>, Jialiang Ding<sup>a#</sup>, Xingxing Jiang<sup>b</sup>, Jian Zhang<sup>a</sup>, Maxim S. Molokeev<sup>cde</sup>, Qiqiong Ren<sup>a</sup>, Jun Zhou<sup>f</sup>, Shili Li<sup>a</sup>, Xian-Ming Zhang<sup>\*ag</sup>

<sup>a</sup> Key Laboratory of Magnetic Molecules and Magnetic Information Materials (Ministry of Education), School of Chemistry and Material Science, Shanxi Normal University, Taiyuan 030006, China. Email: [zhangxm@dns.sxnu.edu.cn](mailto:zhangxm@dns.sxnu.edu.cn); Tel: +86-0357-2051402

<sup>b</sup> China Technical Institute of Physics and Chemistry, Chinese Academy of Sciences, Beijing 100190, China

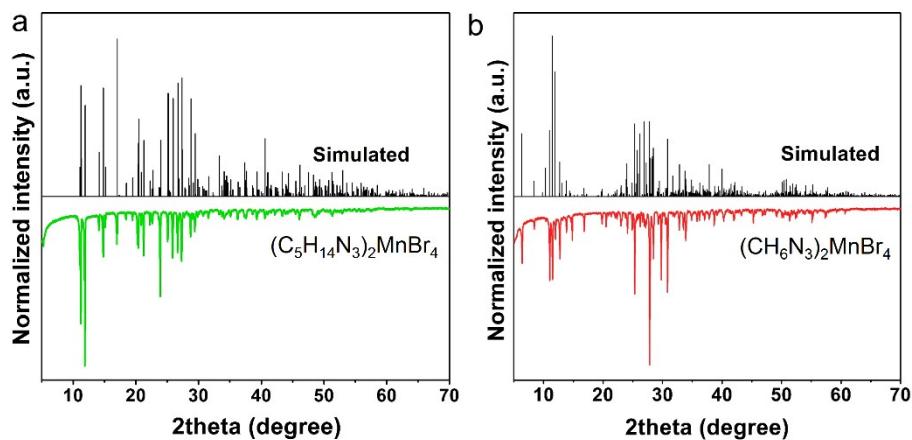
<sup>c</sup> Laboratory of Crystal Physics, Kirensky Institute of Physics, Federal Research Center KSC SB RAS, Krasnoyarsk 660036, Russia

<sup>d</sup> Department of Engineering Physics and Radioelectronics, Siberian Federal University, Krasnoyarsk 660041, Russia

<sup>e</sup> Research and Development Department, Kemerovo State University, Kemerovo, 650000, Russia

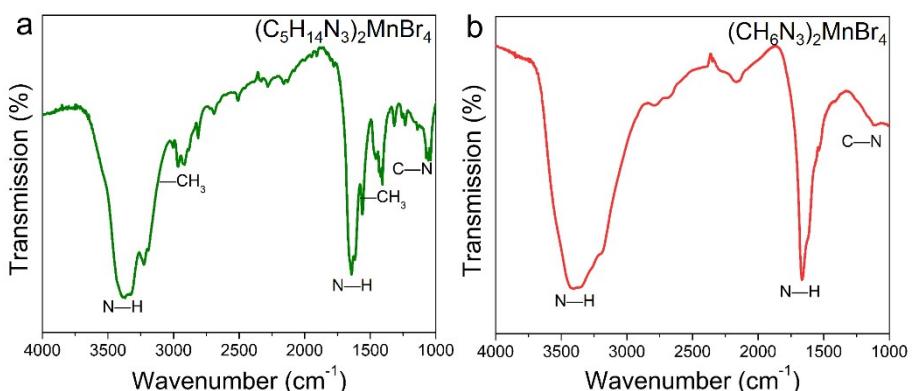
<sup>f</sup> Department of Physics, Beijing Technology and Business University, Beijing 100048, China

<sup>g</sup> College of Chemistry & Chemical Engineering, Key Laboratory of Interface Science and Engineering in Advanced Material, Ministry of Education, Taiyuan University of Technology, Taiyuan, Shanxi 030024, P. R. China

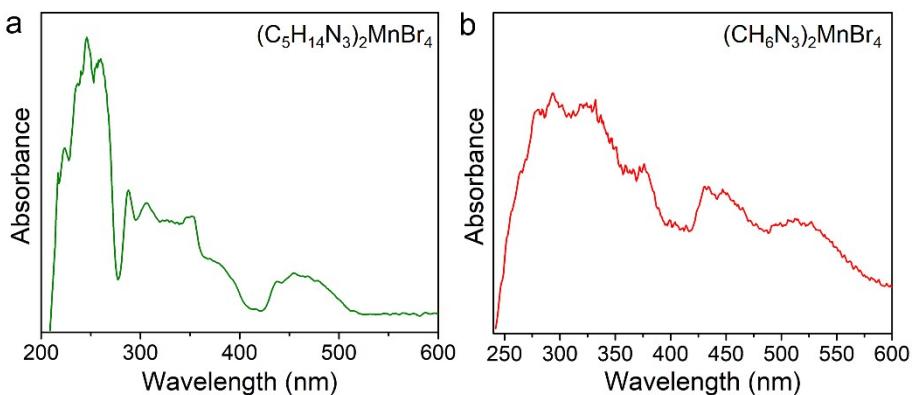


**Figure S1.** The simulated and experimental X-ray powder patterns of  $(C_5H_{14}N_3)_2MnBr_4$

(a) and  $(CH_6N_3)_2MnBr_4$  (b).

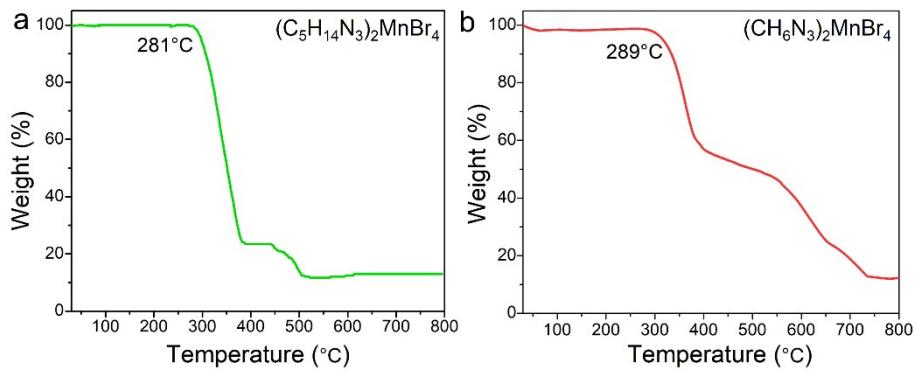


**Figure S2.** FTIR spectra of  $(C_5H_{14}N_3)_2MnBr_4$  (a) and  $(CH_6N_3)_2MnBr_4$  (b).

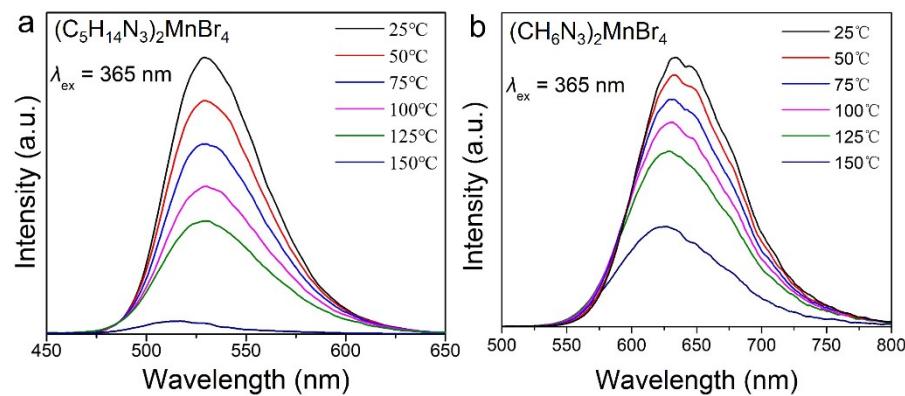


**Figure S3.** UV-visible absorption spectra of  $(C_5H_{14}N_3)_2MnBr_4$  (a) and  $(CH_6N_3)_2MnBr_4$

(b).



**Figure S4.** Thermogravimetric (TG) curves for powder sample of  $(C_5H_{14}N_3)_2MnBr_4$  (a) and  $(CH_6N_3)_2MnBr_4$  (b).



**Figure S5.** Temperature-dependent PL spectra of  $(C_5H_{14}N_3)_2MnBr_4$  (a) and  $(CH_6N_3)_2MnBr_4$  (b) under excitation at 365 nm in the temperature range of RT to 150  $^{\circ}C$ .

**Table S1.** Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ ).

Atom	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
(C <sub>5</sub> H <sub>14</sub> N <sub>3</sub> ) <sub>2</sub> MnBr <sub>4</sub>				
Mn	0.5000	0.30574 (18)	0.2500	0.0366 (5)
Br1	0.43480 (8)	0.15992 (11)	0.08398 (7)	0.0640 (4)
Br2	0.61714 (7)	0.45894 (14)	0.27426 (8)	0.0654 (5)
N1	0.7497 (6)	0.5136 (11)	0.5496 (8)	0.079 (3)
H11	0.7041	0.4831	0.4911	0.094*
H12	0.7988	0.4807	0.5701	0.094*
N2	0.6696 (5)	0.6630 (8)	0.5784 (6)	0.0491 (18)
N3	0.8158 (5)	0.6566 (9)	0.7007 (6)	0.056 (2)
C2	0.7440 (6)	0.6109 (10)	0.6080 (7)	0.0470 (19)
C3	0.5894 (6)	0.5865 (14)	0.5009 (8)	0.066 (3)
H31	0.5420	0.6405	0.4893	0.080*
H32	0.5831	0.5764	0.4345	0.080*
H33	0.5906	0.4926	0.5282	0.080*
C4	0.6517 (7)	0.8140 (11)	0.5919 (10)	0.069 (3)
H41	0.5923	0.8235	0.5635	0.083*
H42	0.6861	0.8385	0.6664	0.083*
H43	0.6650	0.8779	0.5543	0.083*
C5	0.8285 (8)	0.6862 (13)	0.8013 (8)	0.080 (4)
H51	0.8863	0.7171	0.8528	0.097*
H52	0.7899	0.7610	0.7915	0.097*
H53	0.8177	0.6003	0.8266	0.097*
C6	0.9002 (7)	0.6405 (14)	0.7193 (11)	0.077 (4)
H61	0.9439	0.6782	0.7886	0.093*
H62	0.9110	0.5402	0.7160	0.093*
H63	0.9003	0.6927	0.6654	0.093*
(CH <sub>6</sub> N <sub>3</sub> ) <sub>2</sub> MnBr <sub>4</sub>				
Mn1	0.73990 (12)	0.45012 (4)	0.27726 (7)	0.0345 (3)
Mn2	0.72539 (13)	0.56746 (5)	0.20649 (7)	0.0390 (3)
Mn3	0.74383 (13)	0.33480 (5)	0.35697 (7)	0.0408 (3)
Br1	0.75425 (8)	0.42247 (3)	0.43455 (4)	0.0357 (2)
Br2	0.72681 (8)	0.48030 (3)	0.11877 (4)	0.0367 (2)
Br3	0.94408 (8)	0.51830 (3)	0.30396 (5)	0.0369 (2)
Br4	0.53277 (8)	0.51612 (3)	0.30429 (5)	0.0359 (2)
Br5	0.53122 (8)	0.38113 (3)	0.24909 (5)	0.0383 (2)
Br6	0.94102 (8)	0.38063 (3)	0.24732 (5)	0.0378 (2)
Br7	0.73537 (10)	0.25907 (3)	0.26142 (5)	0.0470 (2)
Br8	0.73018 (9)	0.63436 (3)	0.31958 (5)	0.0469 (2)
Br9	0.96698 (10)	0.30553 (4)	0.45428 (5)	0.0528 (3)
Br10	0.92066 (10)	0.60706 (4)	0.11286 (5)	0.0556 (3)
Br11	0.53972 (11)	0.30543 (4)	0.45445 (5)	0.0571 (3)

Br12	0.50016 (10)	0.59736 (4)	0.11062 (5)	0.0596 (3)
C1A	0.2229 (8)	0.4689 (3)	0.1057 (5)	0.040 (2)
N1A	0.0950 (7)	0.4600 (3)	0.0664 (4)	0.066 (3)
H11A	0.0924	0.4477	0.0174	0.079*
H12A	0.0136	0.4666	0.0895	0.079*
N2A	0.3454 (8)	0.4589 (3)	0.0705 (4)	0.060 (2)
H21A	0.4294	0.4632	0.0972	0.072*
H22A	0.3424	0.4481	0.0204	0.072*
N3A	0.2303 (8)	0.4851 (3)	0.1808 (5)	0.069 (3)
H31A	0.3153	0.4891	0.2067	0.083*
H32A	0.1502	0.4919	0.2051	0.083*
C1B	0.7461 (8)	0.5648 (3)	0.5451 (4)	0.0364 (18)
N1B	0.6149 (7)	0.5568 (3)	0.5074 (4)	0.059 (2)
H11B	0.6090	0.5466	0.4570	0.071*
H12B	0.5354	0.5618	0.5333	0.071*
N2B	0.8675 (8)	0.5573 (3)	0.5062 (4)	0.066 (3)
H21B	0.8622	0.5471	0.4558	0.079*
H22B	0.9527	0.5625	0.5311	0.079*
N3B	0.7544 (8)	0.5780 (3)	0.6241 (4)	0.054 (2)
H31B	0.6746	0.5813	0.6505	0.065*
H32B	0.8395	0.5832	0.6490	0.065*
C1C	0.2419 (9)	0.2557 (4)	0.2969 (7)	0.067 (3)
N1C	0.2404 (9)	0.3000 (3)	0.3139 (6)	0.074 (3)
H11C	0.3222	0.3147	0.3266	0.089*
H12C	0.1576	0.3153	0.3126	0.089*
N2C	0.3625 (10)	0.2314 (4)	0.2979 (6)	0.087 (3)
H21C	0.4461	0.2451	0.3104	0.104*
H22C	0.3598	0.2014	0.2861	0.104*
N3C	0.1150 (11)	0.2325 (4)	0.2774 (6)	0.093 (3)
H31C	0.0318	0.2476	0.2765	0.112*
H32C	0.1165	0.2025	0.2660	0.112*
C1D	0.7669 (8)	0.3730 (3)	0.6727 (5)	0.0408 (19)
N1D	0.6396 (8)	0.3647 (3)	0.6340 (5)	0.062 (2)
H11D	0.5587	0.3709	0.6578	0.074*
H12D	0.6363	0.3531	0.5845	0.074*
N2D	0.7718 (8)	0.3908 (3)	0.7485 (4)	0.059 (2)
H21D	0.6907	0.3970	0.7722	0.070*
H22D	0.8560	0.3963	0.7742	0.070*
N3D	0.8890 (8)	0.3631 (3)	0.6369 (5)	0.062 (2)
H31D	0.9735	0.3682	0.6626	0.074*
H32D	0.8852	0.3515	0.5873	0.074*
C1E	0.7842 (10)	0.1787 (4)	0.5202 (5)	0.051 (2)
N1E	0.7558 (9)	0.1886 (3)	0.4427 (4)	0.064 (2)
H11E	0.7119	0.1678	0.4106	0.076*

H12E	0.7808	0.2159	0.4232	0.076*
N2E	0.7474 (10)	0.1375 (3)	0.5512 (5)	0.074 (3)
H21E	0.7034	0.1164	0.5198	0.089*
H22E	0.7671	0.1315	0.6030	0.089*
N3E	0.8583 (10)	0.2094 (4)	0.5661 (6)	0.094 (3)
H31E	0.8832	0.2026	0.6171	0.113*
H32E	0.8825	0.2365	0.5457	0.113*
C1F	0.3259 (15)	0.2024 (6)	0.5636 (8)	0.107 (4)
N1F	0.4402 (15)	0.2170 (5)	0.6017 (9)	0.144 (5)
H11F	0.4862	0.2416	0.5837	0.173*
H12F	0.4735	0.2026	0.6460	0.173*
N2F	0.2558 (15)	0.2178 (6)	0.4969 (9)	0.151 (5)
H21F	0.2883	0.2424	0.4713	0.181*
H22F	0.1766	0.2035	0.4780	0.181*
N3F	0.2815 (18)	0.1597 (7)	0.5858 (11)	0.193 (7)
H31F	0.3310	0.1442	0.6241	0.231*
H32F	0.2030	0.1473	0.5620	0.231*

**Table S2.** The main bond lengths ( $\text{\AA}$ ) of compound  $(\text{C}_5\text{H}_{14}\text{N}_3)_2\text{MnBr}_4$  and  $(\text{CH}_6\text{N}_3)_2\text{MnBr}_4$ .

$(\text{C}_5\text{H}_{14}\text{N}_3)_2\text{MnBr}_4$			
Mn—Br1	2.4876 (13)	N1—C2	1.319 (12)
Mn—Br1 <sup>i</sup>	2.4876 (13)	N2—C2	1.315 (12)
Mn—Br2	2.4986 (14)	N2—C3	1.471 (12)
Mn—Br2 <sup>i</sup>	2.4987 (14)	N2—C4	1.483 (13)
		N3—C2	1.352 (11)
		N3—C5	1.445 (13)
		N3—C6	1.481 (15)
$(\text{CH}_6\text{N}_3)_2\text{MnBr}_4$			
Mn1—Br1	2.6446 (13)	C1A—N2A	1.298 (9)
Mn1—Br3	2.6694 (14)	C1A—N3A	1.290 (10)
Mn1—Br4	2.6776 (14)	C1A—N1A	1.310 (9)
Mn1—Br2	2.6845 (13)	C1B—N1B	1.320 (9)
Mn1—Br5	2.7162 (14)	C1B—N2B	1.306 (9)
Mn1—Br6	2.7181 (14)	C1B—N3B	1.322 (9)
Mn2—Br8	2.6093 (15)	C1C—N1C	1.269 (12)
Mn2—Br10	2.6226 (14)	C1C—N2C	1.281 (12)
Mn2—Br12	2.6224 (15)	C1C—N3C	1.339 (12)
Mn2—Br4	2.8011 (14)	C1D—N2D	1.316 (10)
Mn2—Br3	2.8136 (14)	C1D—N1D	1.298 (10)
Mn2—Br2	2.8174 (15)	C1D—N3D	1.299 (10)
Mn3—Br11	2.6133 (14)	C1E—N2E	1.306 (11)
Mn3—Br9	2.6158 (15)	C1E—N1E	1.290 (10)
Mn3—Br7	2.6161 (15)	C1E—N3E	1.295 (12)
Mn3—Br1	2.7510 (15)	C1F—N1F	1.241 (16)
Mn3—Br5	2.8322 (15)	C1F—N2F	1.290 (16)
Mn3—Br6	2.8726 (14)	C1F—N3F	1.314 (19)

**Table S3.** Hydrogen-bond geometry in the structures (Å, °).

D—H	d(D—H)	d(H···A)	□	D—H···A	D···A	A	Transformation for A atom
$(C_5H_{14}N_3)_2MnBr_4$							
N(1)—H(11)	0.86	2.78	146	3.482(10)	Br(2)	x, y, z	
N(1)—H(12)	0.86	2.81	165	3.651(13)	Br(1)	1/2+x, 1/2-y, 1/2+z	
$(CH_6N_3)_2MnBr_4$							
N(1A)—H(11A)	0.86	2.59	167	3.439(7)	Br(10)	x, y, z	
N(1B)—H(11B)	0.86	2.66	169	3.505(7)	Br(4)	x, y, z	
N(1C)—H(11C)	0.86	2.78	134	3.433(9)	Br(11)	x, y, z	
N(1D)—H(11D)	0.86	2.65	156	3.455(7)	Br(8)	x, y, z	
N(1E)—H(11E)	0.86	2.75	148	3.516(8)	Br(12)	x, y, z	
N(1F)—H(11F)	0.86	2.80	150	3.573(14)	Br(11)	x, y, z	
N(1A)—H(12A)	0.86	2.68	164	3.514(6)	Br(2)	x, y, z	
N(1B)—H(12B)	0.86	2.72	162	3.552(6)	Br(1)	x, y, z	
N(1C)—H(12C)	0.86	2.84	155	3.635(8)	Br(6)	x, y, z	
N(1D)—H(12D)	0.86	2.59	159	3.412(8)	Br(11)	x, y, z	
N(1E)—H(12E)	0.86	2.88	132	3.519(7)	Br(7)	x, y, z	
N(1F)—H(12F)	0.86	2.90	154	3.692(14)	Br(5)	x, y, z	
N(2A)—H(21A)	0.86	2.73	157	3.537(7)	Br(2)	x, y, z	
N(2B)—H(21B)	0.86	2.71	161	3.537(7)	Br(3)	x, y, z	
N(2C)—H(21C)	0.86	2.79	145	3.529(9)	Br(7)	x, y, z	
N(2D)—H(21D)	0.86	2.62	159	3.443(7)	Br(12)	x, y, z	
N(2E)—H(21E)	0.86	2.77	148	3.524(8)	Br(12)	x, y, z	
N(2F)—H(21F)	0.86	2.90	145	3.635(15)	Br(11)	x, y, z	
N(2A)—H(22A)	0.86	2.90	148	3.653(7)	Br(12)	x, y, z	
N(2B)—H(22B)	0.86	2.70	164	3.539(7)	Br(1)	x, y, z	
N(2C)—H(22C)	0.86	2.63	147	3.387(11)	Br(8)	x, y, z	
N(2D)—H(22D)	0.86	2.64	161	3.468(7)	Br(10)	x, y, z	
N(2E)—H(22E)	0.86	2.76	157	3.563(8)	Br(6)	x, y, z	
N(3A)—H(31A)	0.86	2.56	166	3.402(8)	Br(4)	x, y, z	
N(3B)—H(31B)	0.86	2.73	161	3.558(7)	Br(5)	x, y, z	
N(3C)—H(31C)	0.86	2.69	157	3.499(10)	Br(7)	x, y, z	
N(3D)—H(31D)	0.86	2.67	155	3.469(7)	Br(8)	x, y, z	
N(3F)—H(31F)	0.86	2.73	164	3.559(17)	Br(5)	x, y, z	
N(3A)—H(32A)	0.86	2.62	168	3.463(8)	Br(3)	x, y, z	
N(3B)—H(32B)	0.86	2.72	162	3.545(7)	Br(6)	x, y, z	
N(3C)—H(32C)	0.86	2.76	143	3.485(11)	Br(8)	x, y, z	
N(3D)—H(32D)	0.86	2.64	160	3.457(8)	Br(9)	x, y, z	
N(3E)—H(32E)	0.86	2.57	166	3.408(11)	Br(9)	x, y, z	