

In-Situ SERS Monitoring of Intracellular H₂O₂ in Single Living Cells Based on Label-Free Bifunctional Fe₃O₄@Ag Nanoparticles

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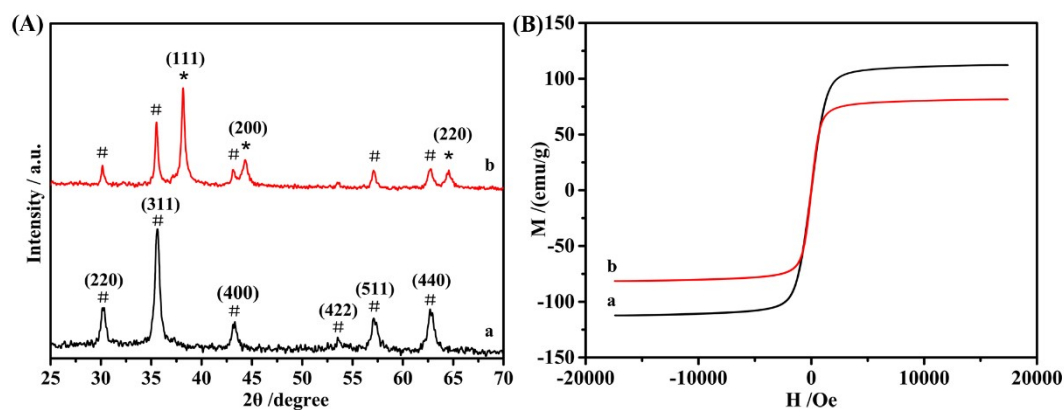


Figure S1 (A) XRD patterns and (B) magnetization curves of the synthesized Fe₃O₄ (a) and Fe₃O₄@Ag NPs (b).

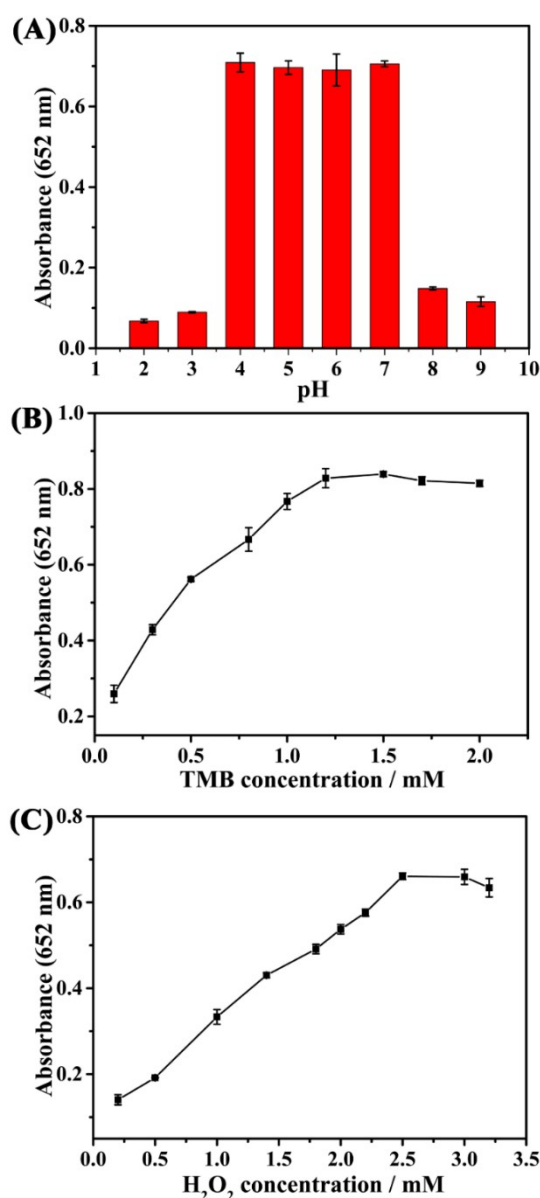


Figure S2. Effects of different conditions on the peroxidase-like activity of Fe₃O₄@Ag NPs: (A) pH; (B) concentration of TMB; (C) concentration of H₂O₂.

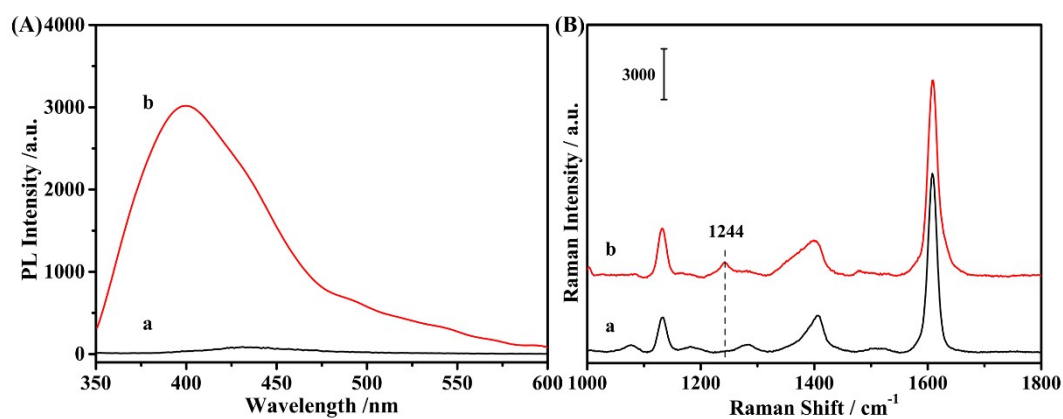


Figure S3. Fluorescence emission spectra (PL = photoluminescence) (A) and Raman spectra (B) of TA in the presence of (a) H₂O₂, (b) Fe₃O₄@Ag NPs and H₂O₂.

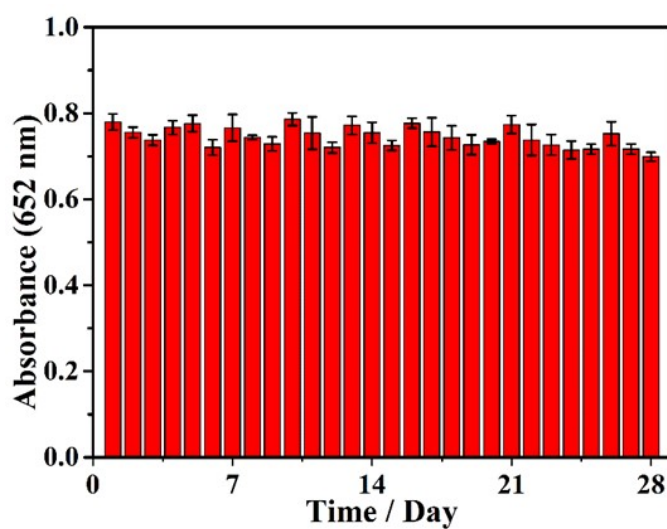


Figure S4. Long-term catalytic stability of Fe₃O₄@Ag NPs stored at room temperature using TMB as the substrate in the presence of H₂O₂.

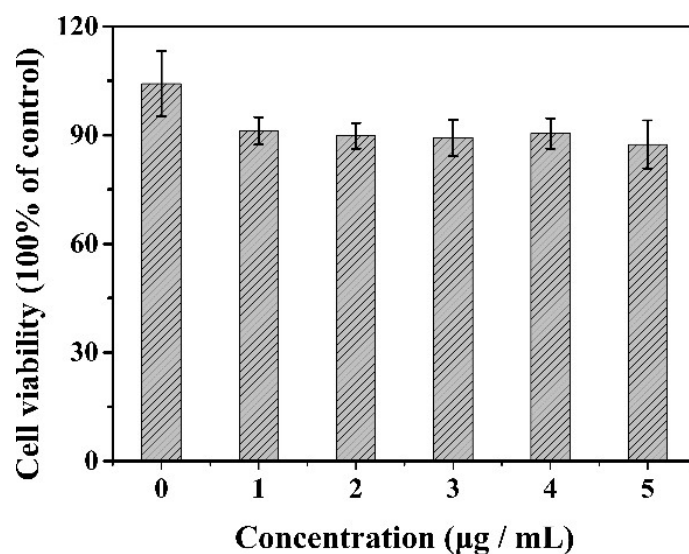


Figure S5. The MTT-based cytotoxicity assay of Fe₃O₄@Ag NPs toward B16 cells.

Table S1. Comparison of Michaelis-Menten constants (K_m) and maximum reaction rates (V_{max}) of the oxidation reaction catalyzed by Fe₃O₄@Ag NPs and reported HRP.

catalyst	substrate	K_m (mM)	V_{max} (10^{-8} M s ⁻¹)
Fe ₃ O ₄ @Ag	TMB	0.61	11.11
Fe ₃ O ₄ @Ag	H ₂ O ₂	24.36	6.82
HRP ¹	TMB	0.434	10.0
HRP ¹	H ₂ O ₂	3.70	8.71

Notes and references

1. Gao, L.; Zhuang, J.; Nie, L.; Zhang, J.; Zhang, Y.; Gu, N.; Wang, T.; Feng, J.; Yang, D.; Perrett, S.; Yan, X. *Nature Nanotech.*, 2007, **2**, 577-583.