| 1 | Supporting information |
|----|---|
| 2 | A portable blood lactate sensor with non-immobilized enzyme for early sepsis |
| 3 | diagnosis |
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27 Text S1 Chemicals and materials

Carbon ink and silver/silver chloride (Ag|AgCl) ink were purchased from Gwent group 28 (Torfaen, UK). A commercial screen-printed carbon electrode was purchased from Quasense Co., Ltd 29 (Bangmod, Bangkok, Thailand). Lactate oxidase (L9795-100UN from Aerococcus viridans) 30 lyophilized powder containing 2.4 mg solid, 100 U·mL⁻¹, Sodium DL-lactate Reagent Plus \geq 99%(nt) 31 and titanium diisopropoxidebis (acetylacetonate) or TIAA (75% wt in isopropanol) were purchased 32 from the Sigma Chemical Co. (St.Louis, MO, USA). Conductive graphene dispersion was purchased 33 from Graphene Supermarket (New York, USA). Isopropanol was obtained from Univar Chemical Co. 34 (Illinois, USA). Phosphate buffer saline (PBS) tablets pH 7.4 was purchased from VWR life science 35 AMRESCO Inc. (Denver, CO, USA). 1,3-propanediol (98%wt) was obtained from ACROS organics 36 (Morris plains, NJ, USA). A stock solution of LOx enzyme was prepared in 0.10 M PB (pH 7.4) 37 aliquoted into a tube (10 µL) and stored at-20°C in a normal freezer. All aqueous solutions were 38 prepared in Milli-Q water (resistivity value 18.2 M Ω ·cm at 25 °C). 39

40

41 Text S2 Instruments

All amperometric measurements were carried out using an Emstat Pico system developed by 42 43 PalmSens BV and Analog Devices Inc. (GA Houten, Netherland). A three-electrode cell was used and the working electrode was a TiO₂ sol-G nanocomposite modified SPCE (0.3 cm in diameter). A 44 45 screen-printing block and machine were fabricated by Chaiyaboon Co., Ltd. (Bangkok, Thailand). A Hitachi/x-4800 transmission electron microscope (TEM) (Japan Electron Optics Laboratory, Japan) 46 was used for TiO₂ sol and TiO₂ sol-G nanocomposite characterization. A standard method for blood 47 lactate measurements was a blood gas analyzer (stat profile prime plus®, NOVA, USA). A 48 customized reader for a portable blood lactate detection kit was developed by Wuxi Admas 49 Technology Co., Ltd (Jiangsu, China). 50

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52 Text S3 Fabrication of a modified electrode

The SPCE was constructed by using a screen-printing machine to print a carbon ink onto a polyvinyl chloride sheet for a working electrode (WE, area = 0.07 cm^2) and counter electrode (CE). A silver/silver chloride (Ag|AgCl) ink was used as a pseudo-reference electrode (RE) and a conductive pad. The printed electrode was dried in an oven at 55 °C for 15 min. The as-prepared nanocomposite was diluted by adding isopropanol with a 1:5 dilution ratio. Finally, 0.5 µL of the nanocomposite was thoroughly dropped onto the working electrode and allowed to dry at room temperature and 59 atmospheric pressure, then the modified electrode was kept in a desiccator when not in use. As for 60 comparison, LOx modified TiO₂ sol-G/SPCE was prepared by dropping 1 μ L of LOx onto TiO₂ sol-61 G/SPCE, and incubated at - 4 °C for overnight.

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63 Text S4 Electrochemical analysis

Amperometry was conducted to optimize the experiment parameters by mixing 1 μ L of LOx (50 U·mL⁻¹) with 100 μ L of 5 mM lactate in 0.1 M PB solution (pH 7.4) for 5 min and measuring on a TiO₂ sol-G/SPCE at an optimized potential of +0.5 V for a fixed time duration following the previous literature [22]. Amperometric lactate detection on the TiO₂ sol-G/SPCE electrode was carried out at the optimized parameters and the anodic current was recorded at a steady state current of 170 s. All measurements were carried out at room temperature, 27 ± 2 °C



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Fig. S1 Effect of TiO_2 sol (a) and G (b) concentration on the current response to 5 mM lactate in 0.1 PB (pH 7.4) at +0.5 V; each data point has been obtained by measurements with 3 individual singleuse electrode; error bars indicate the standard deviations.

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Fig. S2 Effect of LOx enzyme volume (a) and incubation time (b) on the current response to 5 mM
lactate in 0.1 M PB (pH 7.4) at +0.5 V; each data point has been obtained by measurements with 3
individual single-use electrode; error bars indicate the standard deviations.

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Fig. S3. Amperometric responses of TiO₂ sol-G nanocomposite/SPCEs (n=5) for various lactate concentrations
(0.001-0.020 M) in 0.1 M PB (pH 7.4) measured at +0.5 V (a). Calibration plot constructed from the signal
measurements presented in a (b).



Fig. S4. Calibration plot of of TiO₂ sol-G/SPCEs (a) and LOx/TiO₂ sol-G/SPCEs (b)



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Fig. S5. Amperometric current responses of five different LOx/TiO₂ sol-G/SPCEs







Fig. S6. Long-term stability of LOx/TiO₂ sol-G/SPCEs



97 Fig. S7. A photograph of whole blood lactate samples obtained from ICU patients collected in

- 98 lithium heparin tubes.

101 Table S1: Characteristics of sepsis patients' subject.

| Characteristic | Value, median (interquartile range) |
|---|-------------------------------------|
| Number of patients | 50 |
| Gender | |
| male | 28 |
| female | 22 |
| Age (years) mean ± SD | 60.2 ± 20.8 |
| range | 18 - 98 |
| Arterial lactate level (mmol/L) mean ± SD | 6.4 ± 5.7 |
| range | 0.8 - 20 |

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103

104 Table S2. Determination of blood lactate levels in healthy people validated with a blood gas analyzer

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| 5 | sample | Lactate concentration (mmol/L) | | Relative standard | 0/ A gourgowa |
|---|--------|--------------------------------|-----------------|-------------------|---------------|
| | | Blood gas analyzer | Electrochemical | deviation; %RSD | 70Accuracy |
| | 1 | 3.0 | 2.6 ± 0.1 | 3.8 | 86.7 |
| | 2 | 5.7 | 5.6 ± 0.1 | 1.7 | 98.2 |
| | 3 | 3.8 | 3.8 ± 0.1 | 2.6 | 100.0 |
| | 4 | 4.0 | 4.3 ± 0.2 | 4.7 | 107.5 |
| - | 5 | 2.1 | 2.2 ± 0.1 | 4.5 | 104.8 |
| | | | | | |

113 uracy (%) = Mean value of [(mean concentration observed from electrochemical)/concentration

114 observed from blood gas analyzer)] *100.

115 Two samples from healthy individuals (4.3 ± 0.2 and 5.6 ± 0.1 mM) had intensive exertion before

116 sample collection, which may have arisen from lactate accumulation in the blood.

117

118 **Table S3.** Determination of blood lactate levels in ICU patients validated with a blood gas analyzer 119 (n=3).

(*n*=3

| sampla | Lactate concentration (mmol/L) | | Relative standard | 0/ A courseva | |
|--------|--------------------------------|-----------------|--------------------------|---------------|--|
| sample | Blood gas analyzer | Electrochemical | deviation; %RSD | 76Accuracy. | |
| 1 | 1.6 | 1.5 ± 0.2 | 13.3 | 93.8 | |
| 2 | 2.2 | 2.5 ± 0.2 | 8.0 | 113.6 | |
| 3 | 3.0 | 2.6 ± 0.1 | 3.8 | 86.7 | |
| 4 | 5.7 | 5.7 ± 0.1 | 1.7 | 100.0 | |
| 5 | 17.0 | 16.8 ± 0.2 | 1.2 | 98.8 | |
| 6 | 20.0 | 15.7 ± 0.4 | 2.5 | 78.5 | |
| 7 | 5.7 | 4.5 ± 0.3 | 6.7 | 78.9 | |
| 8 | 1.1 | 1.0 ± 0.1 | 10.0 | 90.9 | |
| 9 | 19.5 | 15.0 ± 0.6 | 4.0 | 76.9 | |
| 10 | 12.9 | 11.2 ± 0.2 | 1.8 | 86.8 | |
| 11 | 11.5 | 12.9 ± 0.2 | 1.6 | 112.2 | |
| 12 | 2.3 | 2.35 ± 0.3 | 12.8 | 102.2 | |
| 13 | 1.7 | 1.8 ± 0.2 | 11.1 | 105.9 | |
| 14 | 13.8 | 14.8 ± 0.2 | 1.4 | 107.2 | |
| 15 | 16.1 | 12.1 ± 1.5 | 12.4 | 75.2 | |
| 16 | 2.5 | 2.9 ± 0.4 | 13.8 | 116.0 | |
| 17 | 2.6 | 2.5 ± 0.2 | 8.0 | 96.2 | |
| 18 | 4.2 | 5.1±0.2 | 3.9 | 121.4 | |
| 19 | 2.2 | 2.3 ± 0.1 | 4.3 | 104.5 | |
| 20 | 5.2 | 5.0 ± 0.2 | 4.0 | 96.2 | |
| 21 | 10.8 | 11.5 ± 0.8 | 7.0 | 106.5 | |
| 22 | 8.6 | 8.3 ± 0.2 | 2.4 | 96.5 | |
| 23 | 5.9 | 6.5 ± 0.1 | 1.5 | 110.2 | |
| 24 | 7.2 | 8.5 ± 0.1 | 1.2 | 118.1 | |
| 25 | 6.9 | 6.1 ± 0.4 | 6.6 | 88.4 | |
| 26 | 4.9 | 5.7 ± 0.6 | 10.5 | 116.3 | |
| 27 | 0.8 | 0.9 ± 0.1 | 11.1 | 112.5 | |
| 28 | 1.7 | 1.9 ± 0.3 | 15.8 | 111.8 | |
| 29 | 7.9 | 6.0 ± 0.3 | 4.3 | 75.9 | |
| 30 | 1.0 | 1.0 ± 0.1 | 10.0 | 100.0 | |
| 31 | 19.7 | 16.1 ± 1.0 | 6.2 | 81.7 | |
| 32 | 17.7 | 16.8 ± 1.3 | 7.7 | 94.9 | |
| 33 | 1.9 | 1.9 ± 0.1 | 5.3 | 100 | |
| 34 | 1.4 | 1.4 ± 0.1 | 7.1 | 100 | |

| 120 | 35 | 3.9 | 40 ± 01 | 25 | 102.6 | aAcc |
|-------|----|------|----------------|------|-------|---------|
| | 55 | 5.7 | 1.0 ± 0.1 | 2.5 | 102.0 | |
| 121 | 36 | 3.8 | 4.2 ± 0.1 | 2.4 | 110.5 | urac |
| 100 | 37 | 4.2 | 4.6 ± 0.3 | 6.5 | 109.5 | |
| 122 | 38 | 7.6 | 8.9 ± 0.8 | 9.0 | 117.1 | у |
| 123 | 39 | 3.8 | 3.6 ± 0.2 | 5.6 | 94.7 | (%) |
| 124 | 40 | 11.5 | 12.4 ± 0.4 | 3.2 | 107.8 | = |
| 125 | 41 | 5.8 | 6.4 ± 0.3 | 4.7 | 110.3 | Maa |
| 123 | 42 | 0.9 | 0.9 ± 0.1 | 11.1 | 100 | |
| 126 | 43 | 1.2 | 1.3 ± 0.2 | 15.4 | 108.3 | n |
| 127 | 44 | 1.0 | 1.0 ± 0.1 | 10.0 | 100.0 | valu |
| 128 | 45 | 14.8 | 14.1 ± 0.5 | 3.5 | 95.3 | eof |
| 120 | 46 | 0.9 | 0.9 ± 0.1 | 11.1 | 100.0 | |
| 129 | 47 | 2.8 | 3.1 ± 0.1 | 3.2 | 110.7 | [(me |
| 130 | 48 | 5.0 | 4.8 ± 0.6 | 12.5 | 96.0 | an |
| 131 | 49 | 2.3 | 2.3 ± 0.3 | 13.0 | 100.0 | conc |
| | 50 | 1.5 | 1.6 ± 0.2 | 12.5 | 106.7 | |
| 132 - | | | | | 1 | — entra |

133 tion observed from electrochemical)/concentration observed from blood gas analyzer)] *100

134 The blood samples were collected from patients who were admitted to all sections of the intensive

135 care units (ICUs) at King Chulalongkorn Memorial Hospital at the specified time as described in the

136 experimental section. Due to the samples being collected without the duration of treatment, some

137 patients might recover from treatment, resulting in a lower lactate level than the cut-off.

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