Supporting information for

Trihexyltetradecylphosphonium chloride based ratiometric fluorescent nanosensors for multiplex anion discrimination

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Figure S1. Chemical structures of trihexyltetradecylphosphonium chloride ([THTP][Cl]), chromoionophore III (ETH5350), ETH9033 and ETH9009.



Figure S2. Fluorescence spectra of the [THTP][C1]-based nanosensors without ionophore fordifferent anions including of NaCl (A), NaBr (B), NaI (C) and Na2SO4 (D) with concentrationsfrom 10^{-8} to1M.



Figure S3. Fluorescence spectra of the [THTP][Cl]-based nanosensors with ionophore ETH9033 for different anions including of NaClO₄ (A), NaBr (B), NaCl (C), NaI (D), NaNO₂ (E) and Na₂SO₄ (F) with concentrations from 10^{-6} to 1 M.



Figure S4. (A) Absorbance spectra for NaNO₃ and (B) absorbance ratios of A_{650}/A_{500} versus the logarithm values of anion including NaClO₄ (grey column), NaI (purple column), NaNO₃ (green column), NaBr (yellow column), NaCl (red column) and Na₂SO₄ (blue column) in the range from 10⁻⁶ to 1 M for the [THTP][Cl]-based nanosensor with ETH9033.



Figure S5. Fluorescence spectra of the [THTP][Cl]-based nanosensors in the presence of ionophore ETH9009 for different anions including of NaClO₄ (A), NaNO₃ (B), NaBr (C), NaI (D), Na₂SO₄ (E) and NaNO₂ (F) with concentrations from 10^{-8} to 1 M.



Figure S6. (A) Absorbance spectra for NaCl and (B) absorbance ratios of A_{650}/A_{500} versus the logarithm values of anion including NaClO₄ (grey column), NaI (purple column), NaNO₃ (green column), NaBr (yellow column), NaCl (red column), and Na₂SO₄ (blue column) in the range from 10⁻⁶ to 1 M for the [THTP][Cl]-based nanosensor with ETH9009.