

Supporting Information

Development of AIEE Active Fluorescent and Colorimetric Probe for Solid, Solution, and Vapor Phase Detection of Cyanide: Smartphone and Food Applications

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SI-1. Instruments and reagents

Proton nuclear magnetic resonance (¹H NMR) spectra of compounds **1** and **2** were recorded at Bruker Avance 400 and 500 MHz while ¹³C NMR spectra were carried out at Bruker Avance 100 and 125 MHz. Photophysical properties and sensing studies of compound **2** were scanned through fluorescence spectroscopy (FluoroMax-Plus-P-C, Horiba Jobin Yvon Technology, USA). UV-Vis spectrophotometer (SPECORD 200 PLUS-223E2003C, Analytik Jena, Germany) was used to record absorption spectra. Used reagents and chemicals were acquired from Daejung Chemicals & Metals (Korea), Alfa Aesar (UK), and Sigma Aldrich (USA). The chemicals enlist 9H-fluorenone, aniline, 9,9-Bis(4-aminophenyl)fluorene, methanesulfonic acid, ethyl acetate, *n*-hexane, DMSO-*d*₆, salicylaldehyde, methanol, DMF, acetonitrile, THF, DCM, CHCl₃, DMSO, CN⁻, I⁻, NO₃⁻, Cl⁻, F⁻, N₃⁻, Br⁻, H₂PO₄⁻, ClO₄⁻, CH₃COO⁻, and NO₂⁻.

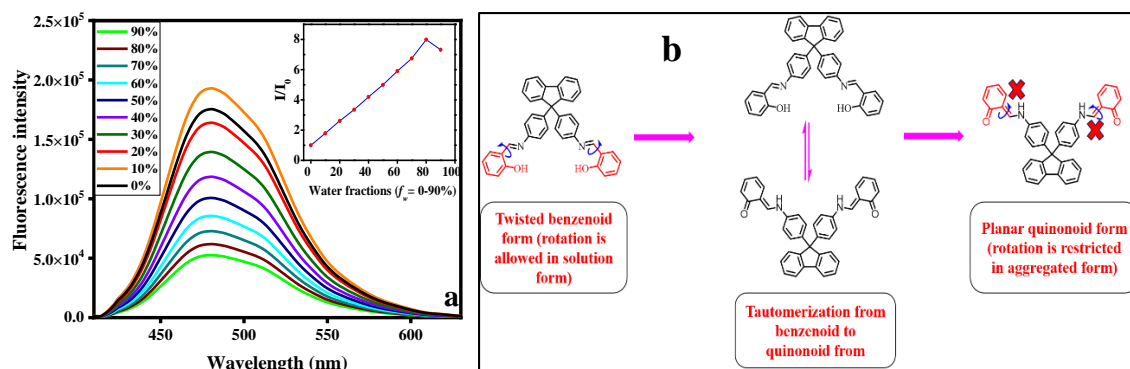


Fig. S1. The fluorescence spectral changes of probe **2** (10 μ M) at various DMF/H₂O ratios (a) and plausible AIEE mechanism (b). Inset shows the general representation of AIEE pattern.

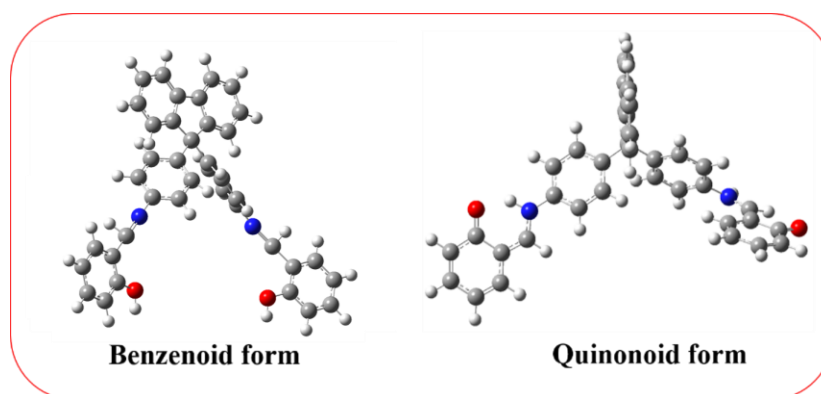


Fig. S2. Theoretical representation of the benzenoid and quinonoid form of probe **2** accomplished through Gaussian 09 package at B3LYP/6-31G (d).

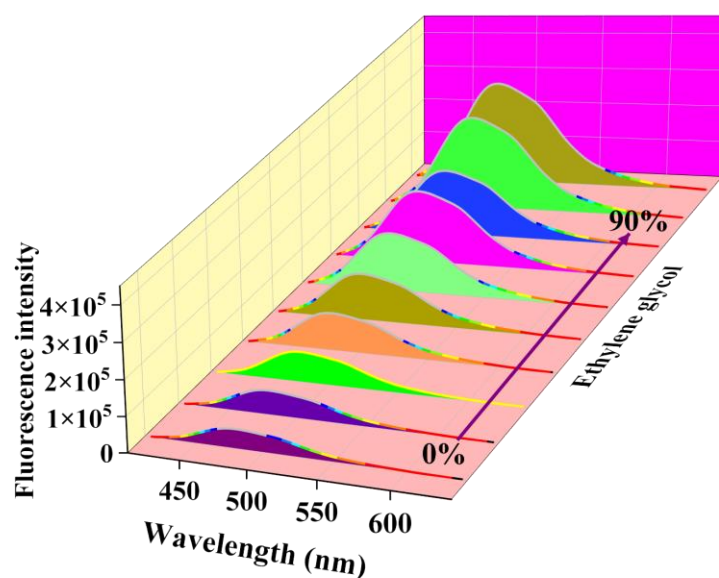


Fig. S3. The fluorescence spectral changes of probe **2** (10 μM) at various DMF/ethylene glycol ratios.

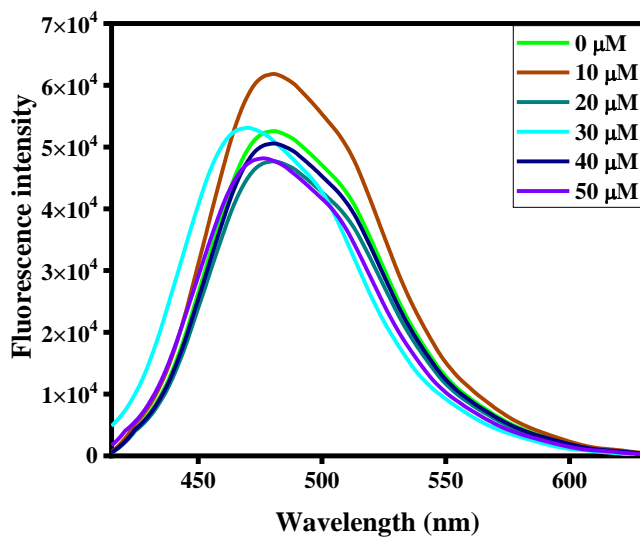


Fig. S4. The fluorescence spectral changes of probe **2** at its various concentrations (0–50 μM).

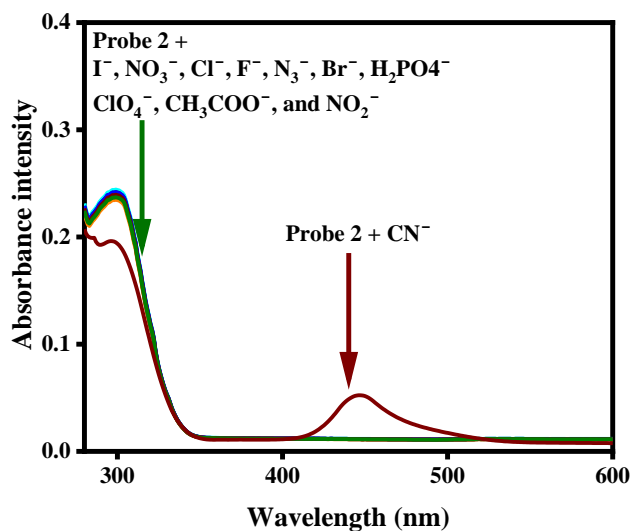


Fig. S5. The UV-vis spectral change of probe **2** (10 μM) in the presence of different anions.

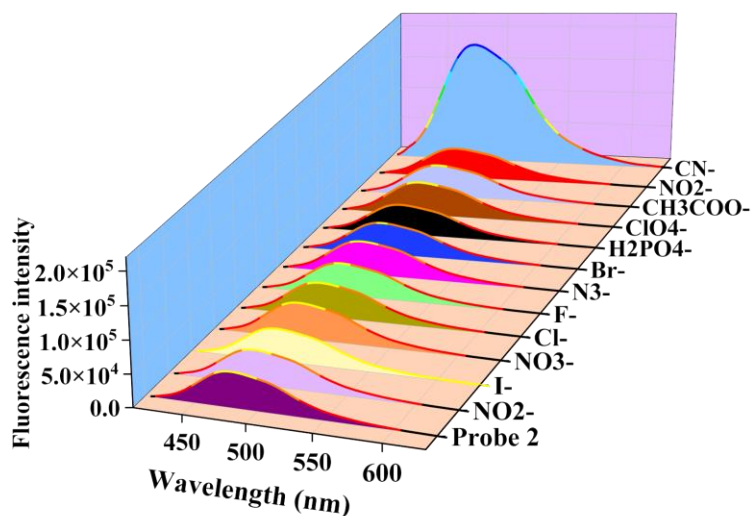


Fig. S6. The fluorescence spectral change of probe **2** (10 μ M) in the presence of different anions (Excitation wavelength = 410 nm).

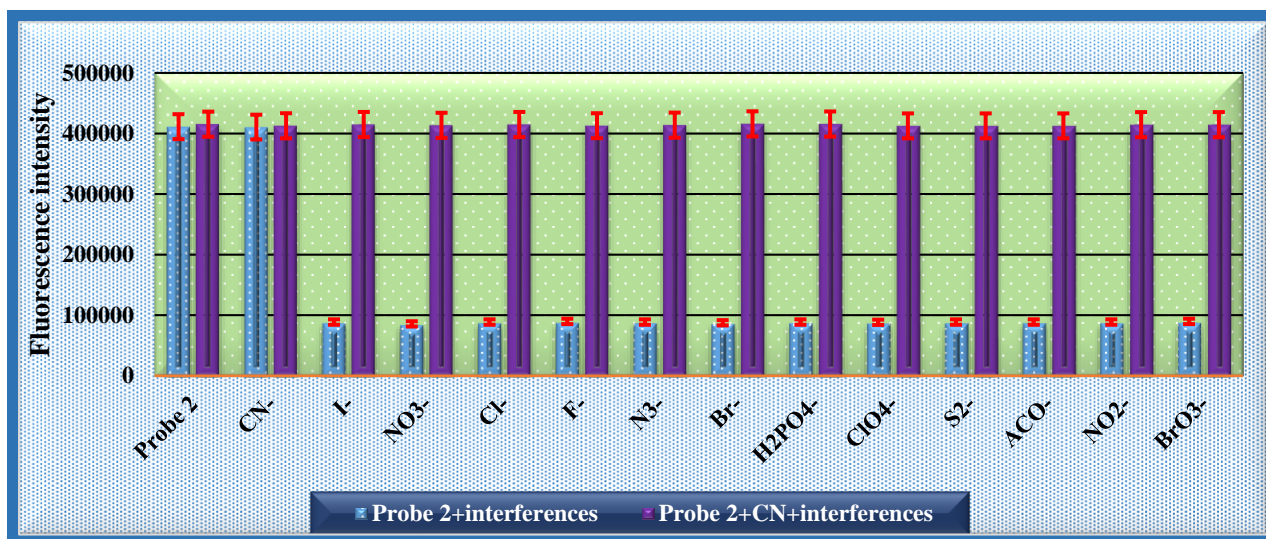


Fig. S7. The fluorescence change of probe **2** (10 μ M) containing 50 nM of CN^- ions in the presence of interferences (100 equiv).

Table S1. The comparison of LOD for CN^- and other features of probe **2** with already reported CN^- probes.

Sensors	Aggregation induced emission enhancement (AIEE)	Piezofluorochromic behavior	Vapor phase detection	Detection in actual food Samples	Smartphone based detection	LOD (nM)	Ref.

Schiff base based probe 2	Described in detail	Excellent piezofluorochromic character	Yes	Yes	Yes	6.17	This work
Spiropyran and spironaphthoxazine based sensor	Didn't display	Didn't display	Not described	Not described	Not described	91	¹
Naphthopyranbenzothiazole core sensor	Didn't display	Didn't display	Not described	Not described	Not described	330	²
aldehyde-appended salamo-type sensor	Didn't display	Didn't display	Not described	Not described	Not described	80	³
Indoleoxazine functionalized compound	Didn't display	Didn't display	Not described	Explored	Not described	100	⁴
D- π -A organic fluorophore	Didn't display	Didn't display	Not described	Not described	Not described	380	⁵
Pyridine based nanoparticles as cyanide ion sensor	Explored	Didn't display	Not described	Not described	Not described	8.2	⁶
Naphthopyran functionalized sensor	Explored	Didn't display	Not described	Not described	Not described	427	⁷
Oligothiophene-benzothiazole based sensor	Didn't display	Didn't display	Not described	Explored	Not described	15×10^3	⁸
Triphenylamine based probe	Explored	Didn't display	Not described	Explored	Not described	2×10^4	⁹
Carbazole based sensor	Explored	Didn't display	Not described	Not described	Not described	67.4	¹⁰

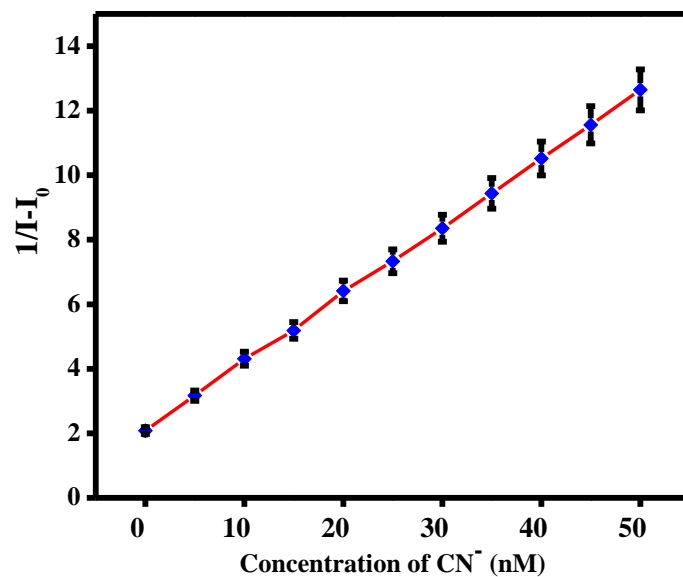


Fig. S8. The Benesi-Hildebrand plot of probe 2 (10 μM) between fluorescence response vs increasing concentrations of CN⁻.

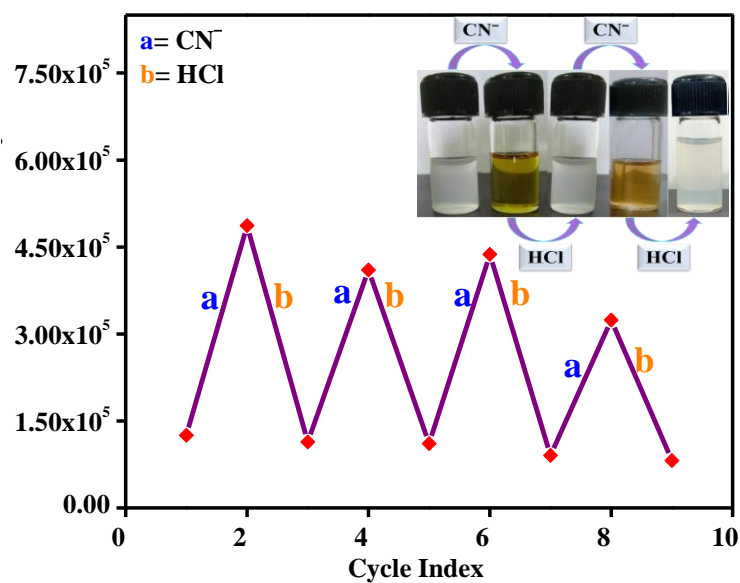


Fig. S9. The switchable fluorescence and colorimetric response of probe 2.

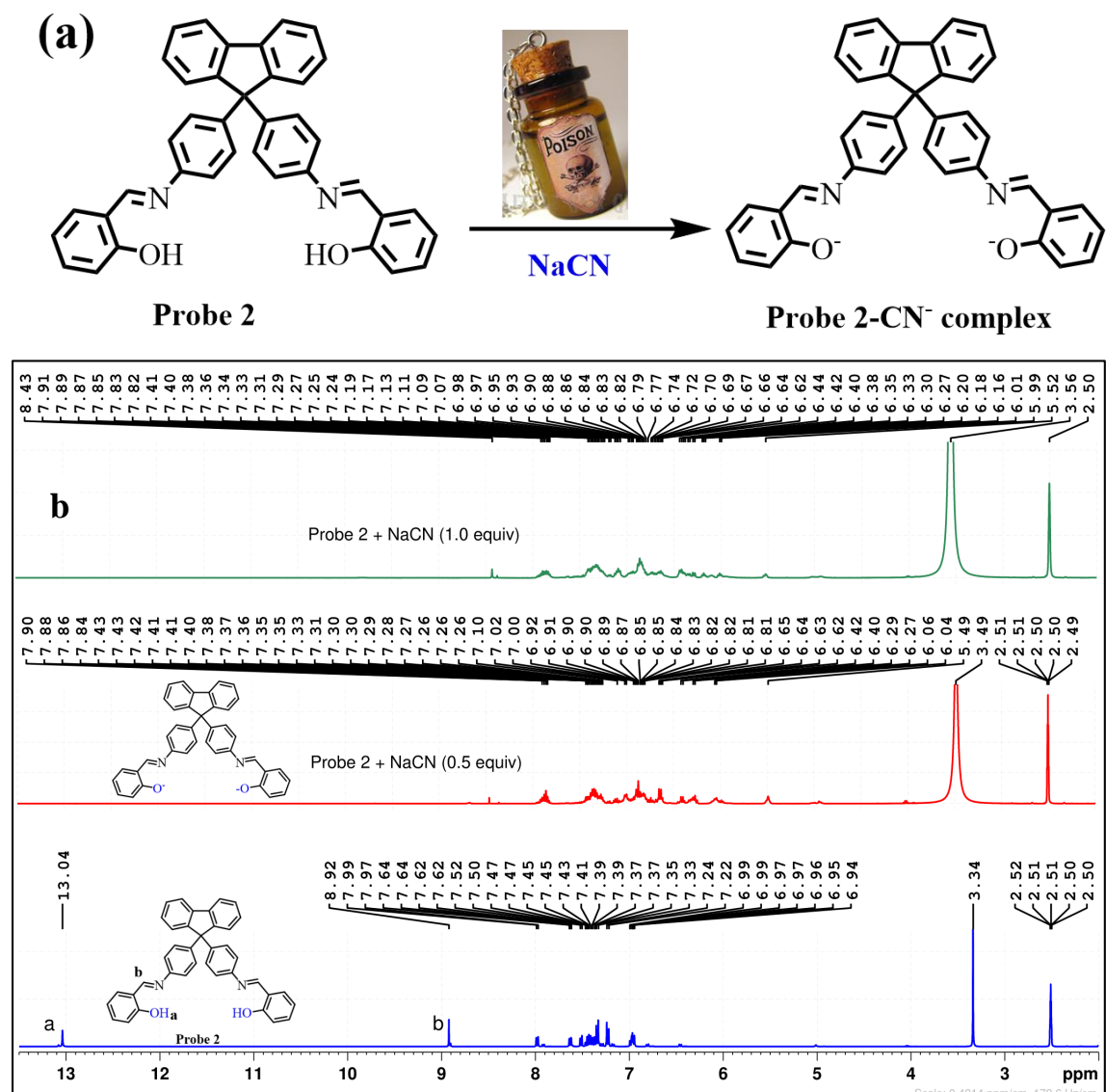


Fig. S10. The putative sensing mechanism (a) and ¹H NMR titration of probe 2 (b) upon addition of increasing amounts of CN⁻.

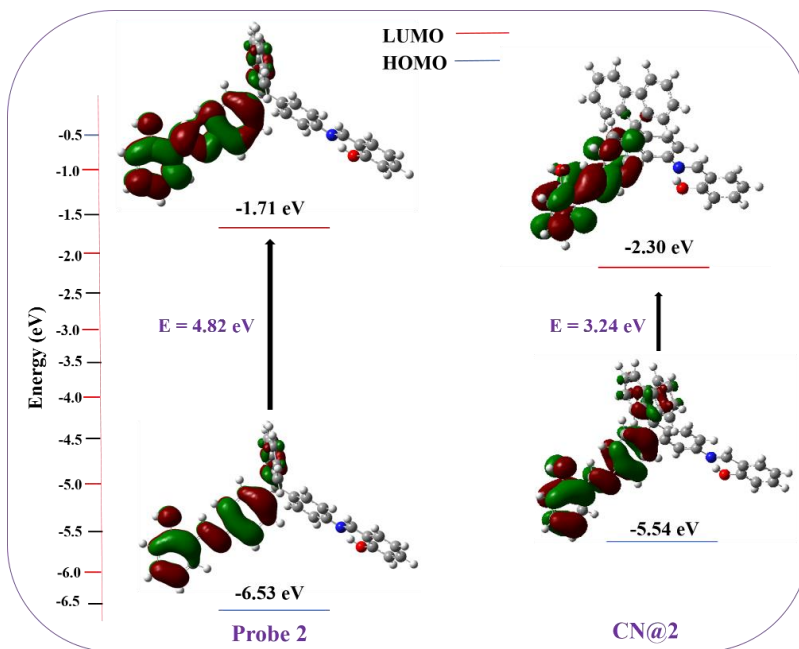


Fig. S11. Frontier molecular orbital (FMO) plots of probe **2** and its probe–CN[−] complex (achieved through Gaussian 09 program).

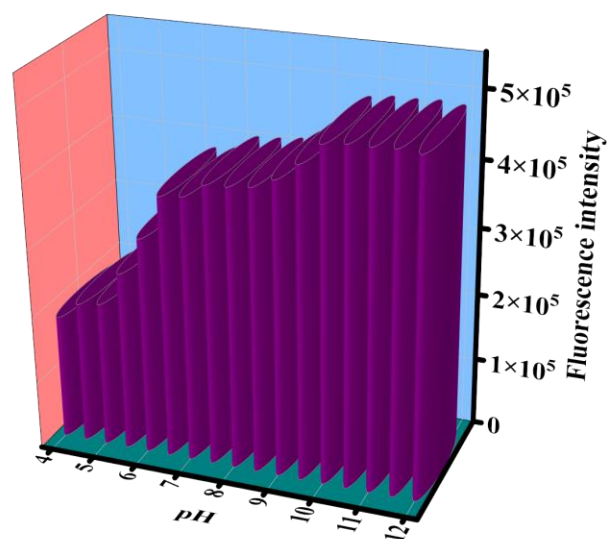


Fig. S12. Effect of pH on the enhanced fluorescence emission response of probe **2** towards CN[−].

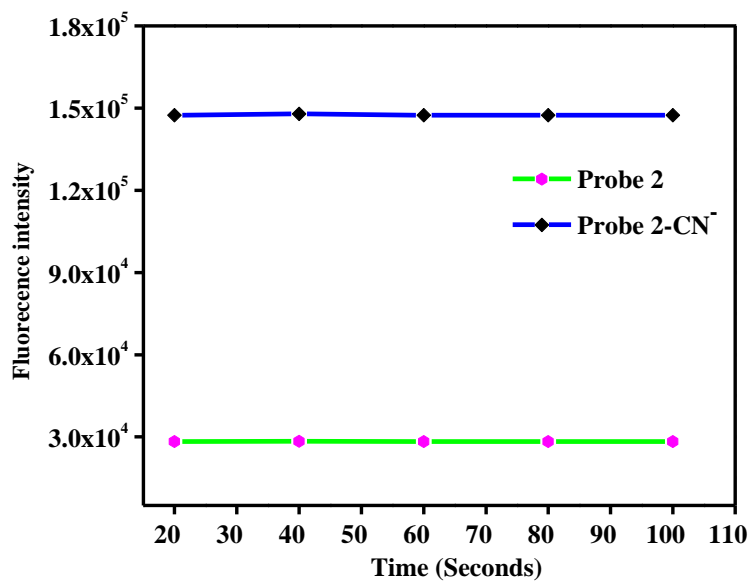
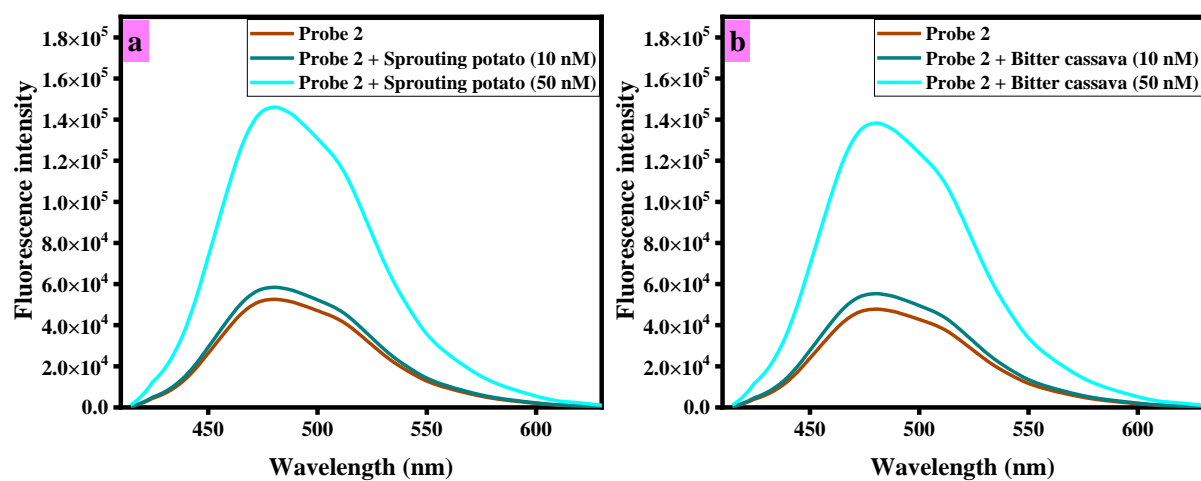


Fig. S13. Photostability analysis of probe 2 and 2-CN⁻ complex.



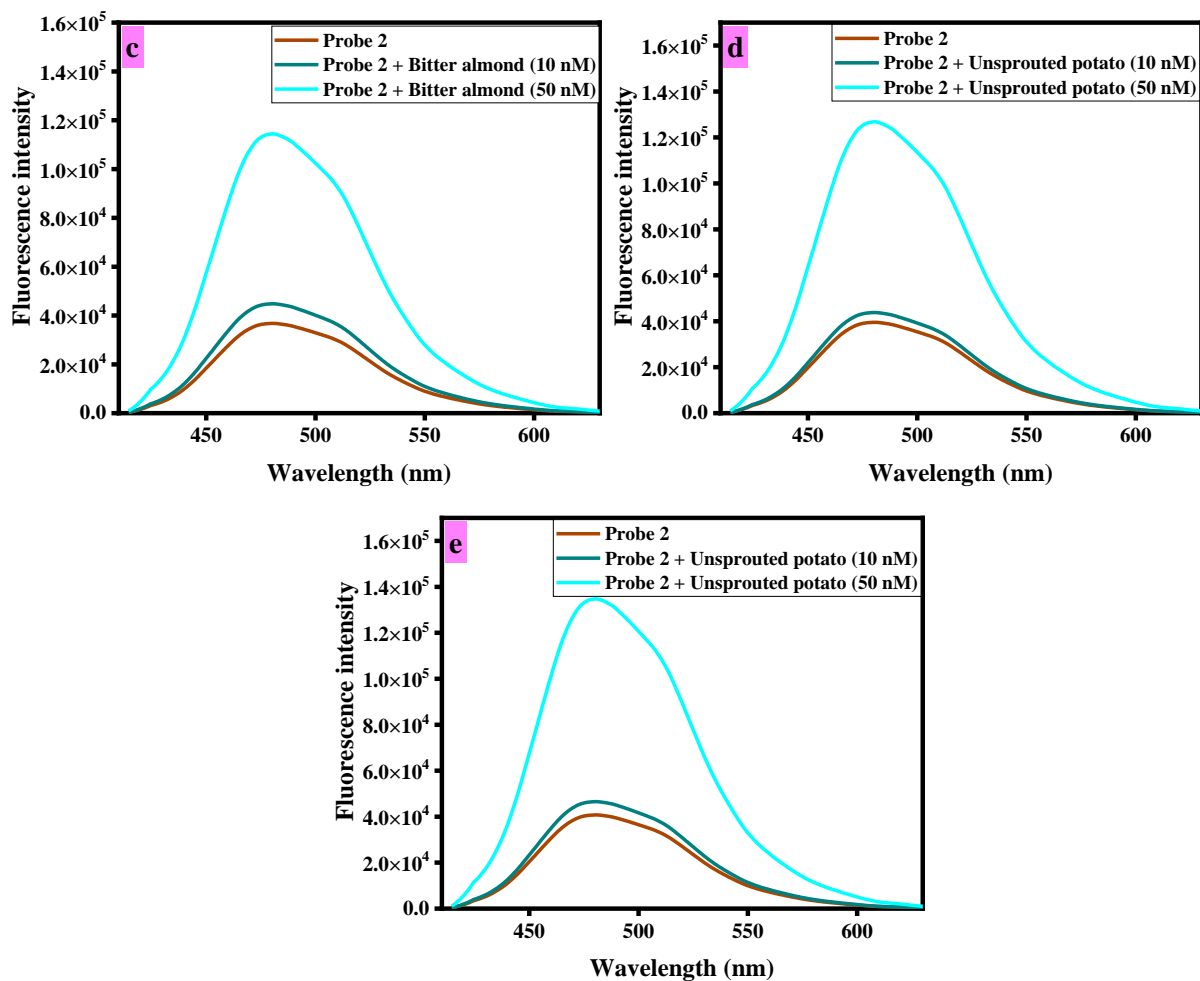


Fig. S14. The fluorescence spectral changes of probe 2 (10 μ M) towards endogenous cyanide in food samples (sprouting potato (a), bitter cassava (b), bitter almond (c), un-sprouted potatoes (d), and sweet almond (e)).

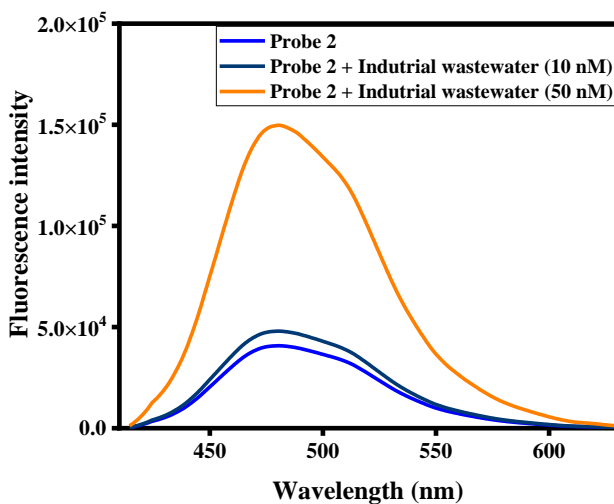
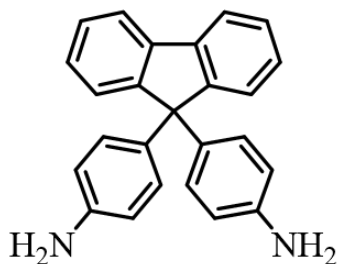
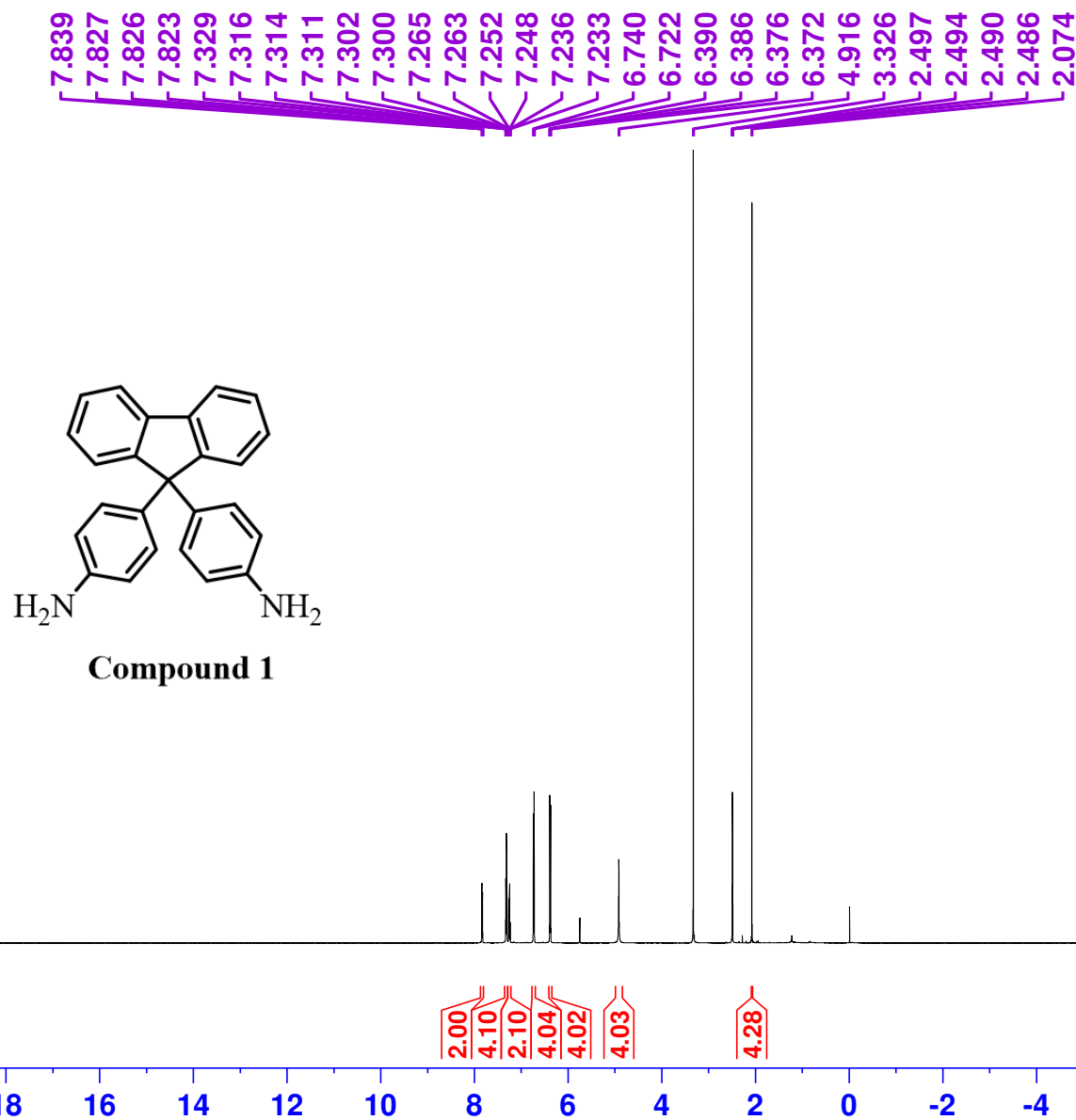


Fig. S15. The fluorescence spectra changes of probe **2** (10 μM) towards CN^- spiked industrial wastewater.

References

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Compound 1 (1H NMR, 500 MHz, DMSO-d6)



Compound 1

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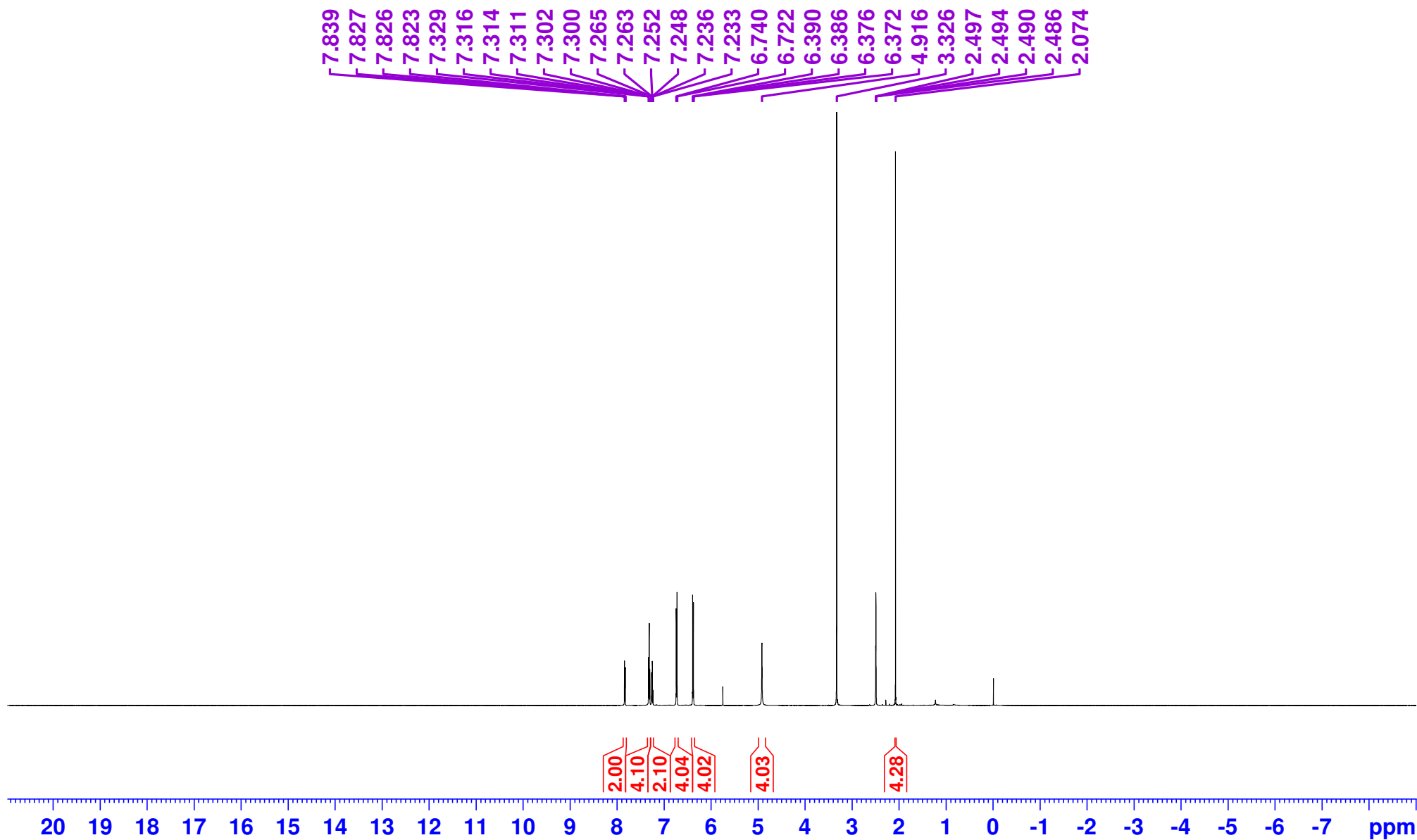
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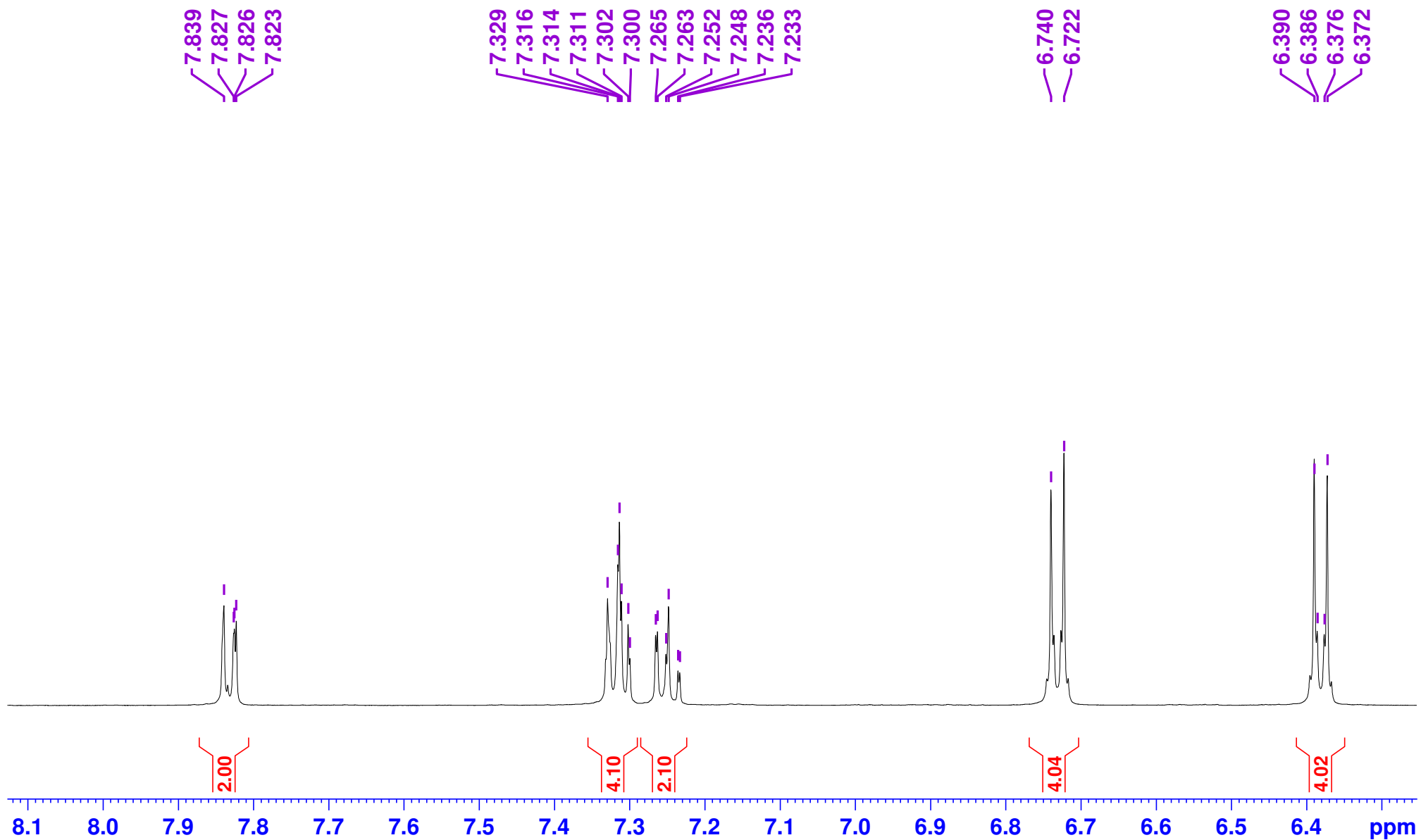
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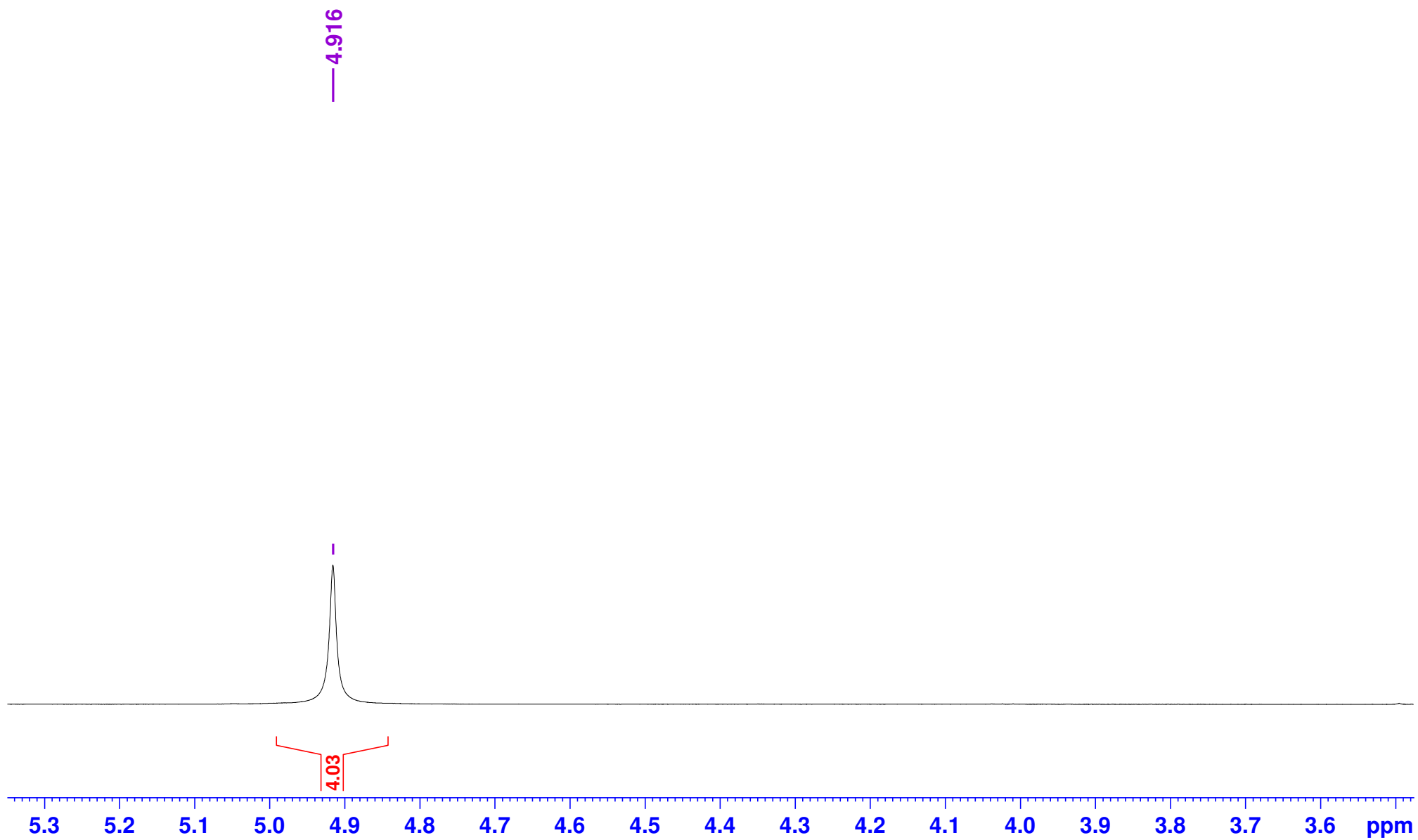
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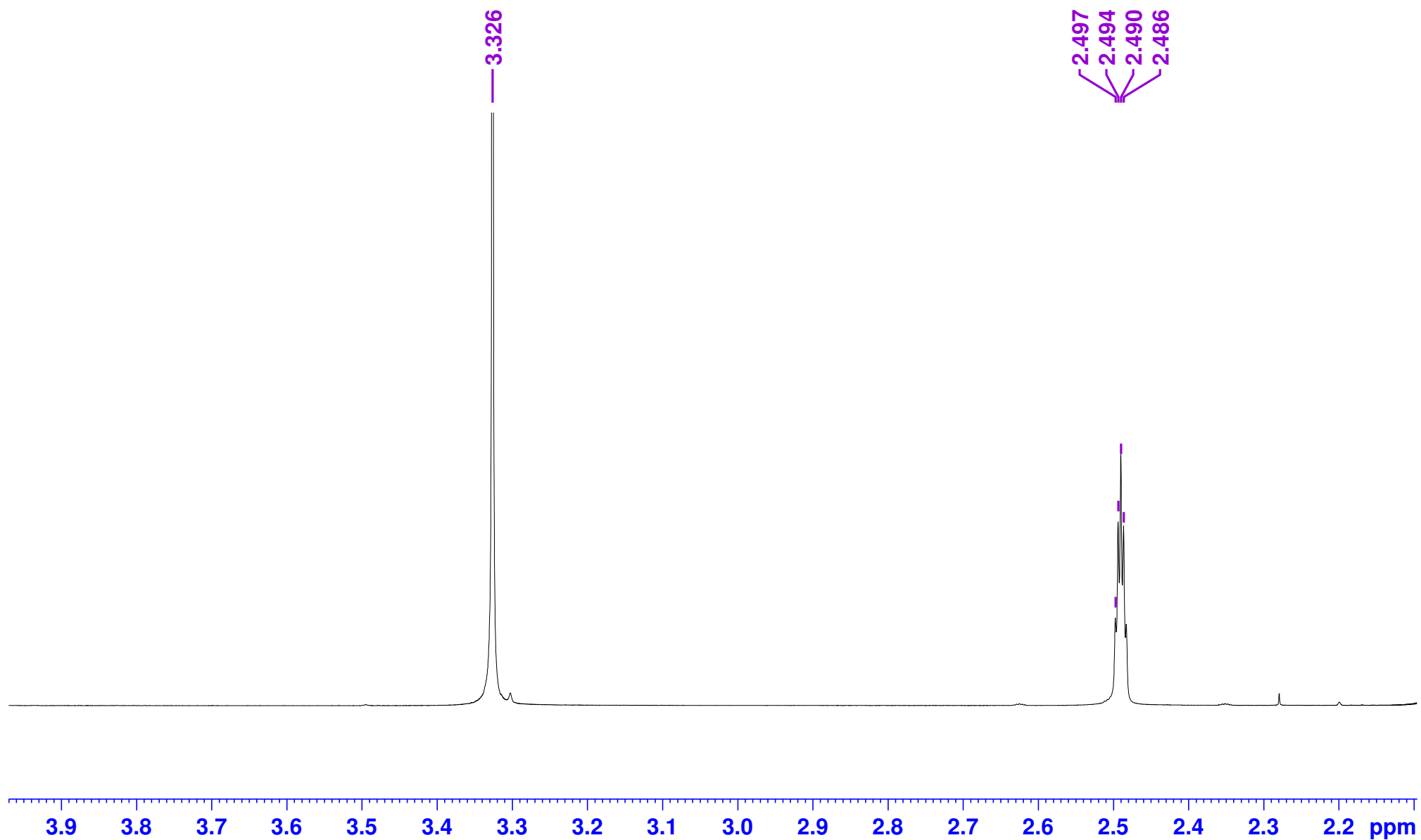
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Compound 1 (1H NMR, 500 MHz, DMSO-d6)



Compound 1 (1H NMR, 500 MHz, DMSO-d6)



Compound 1 (13C NMR, 125 MHz, DMSO-d6)



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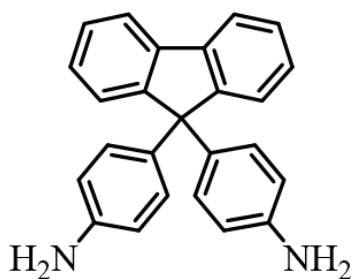
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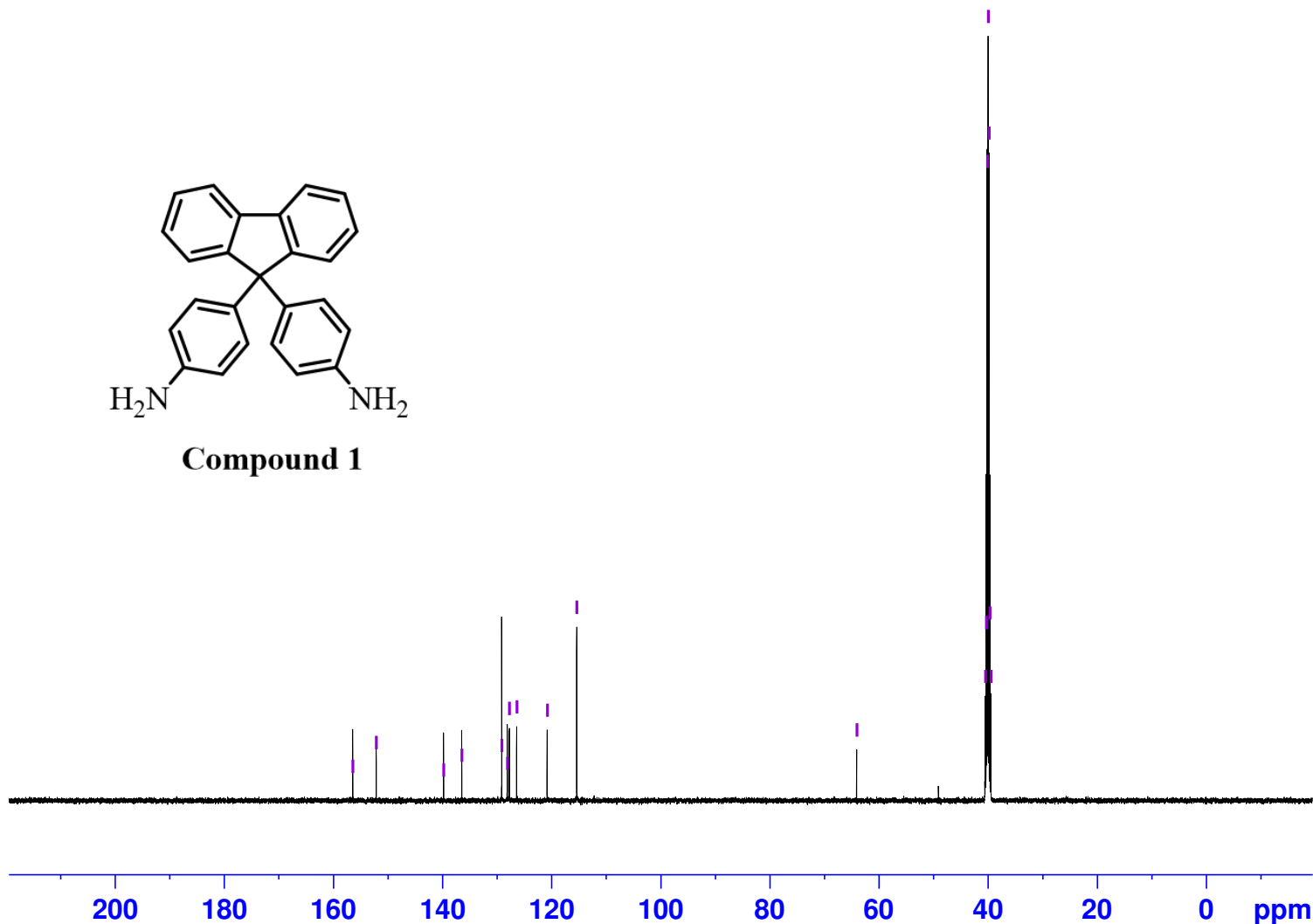
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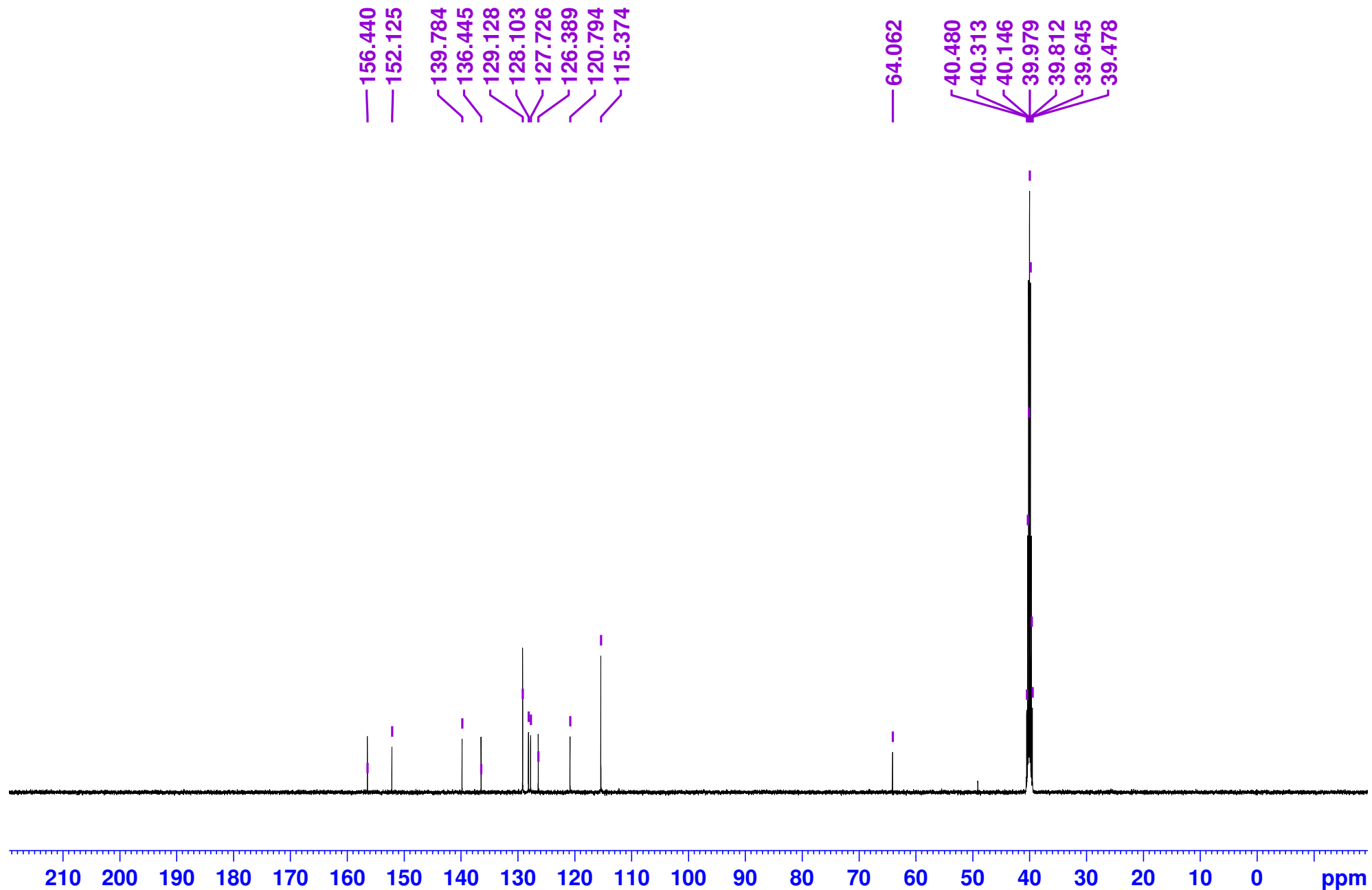
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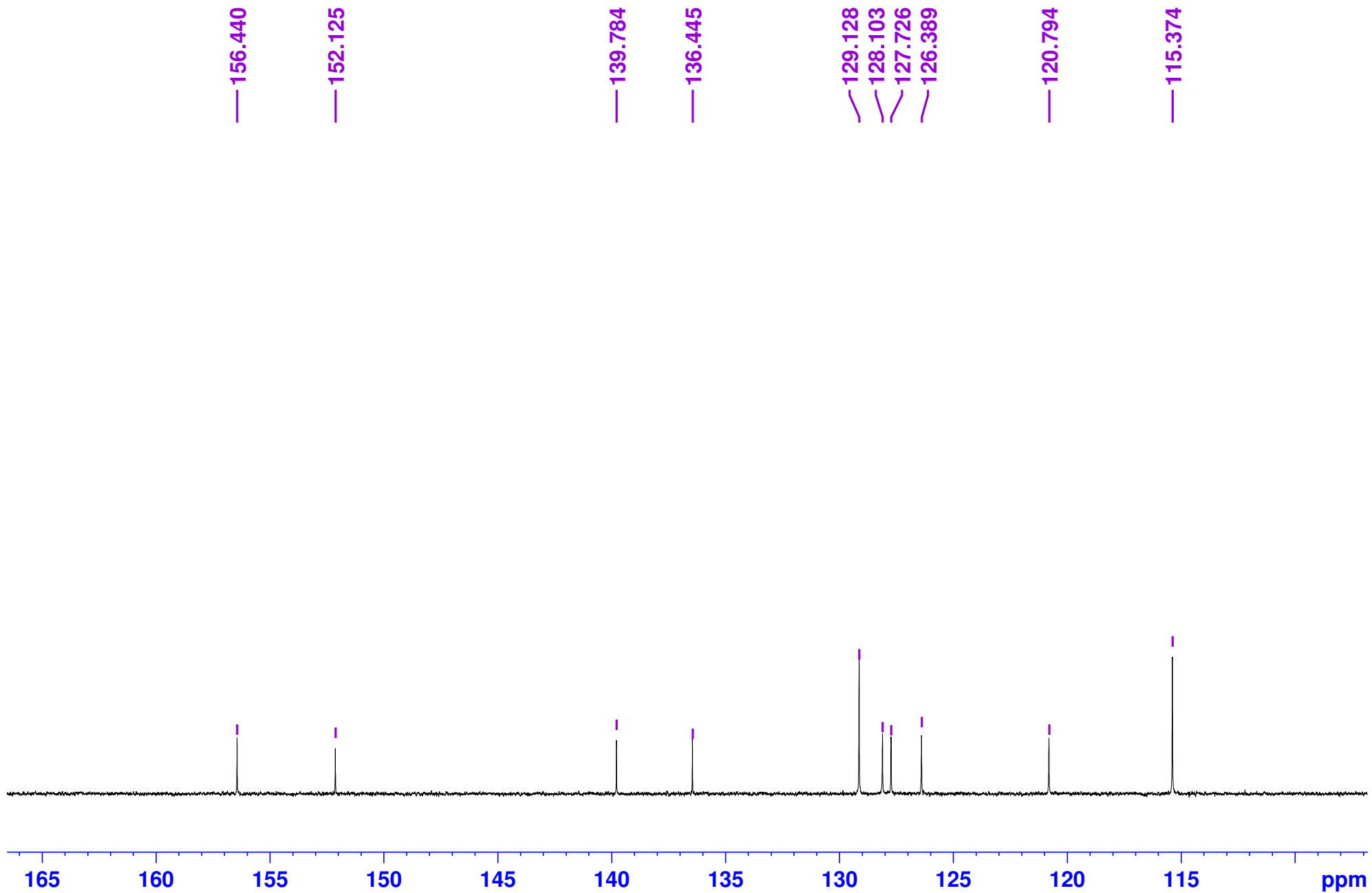
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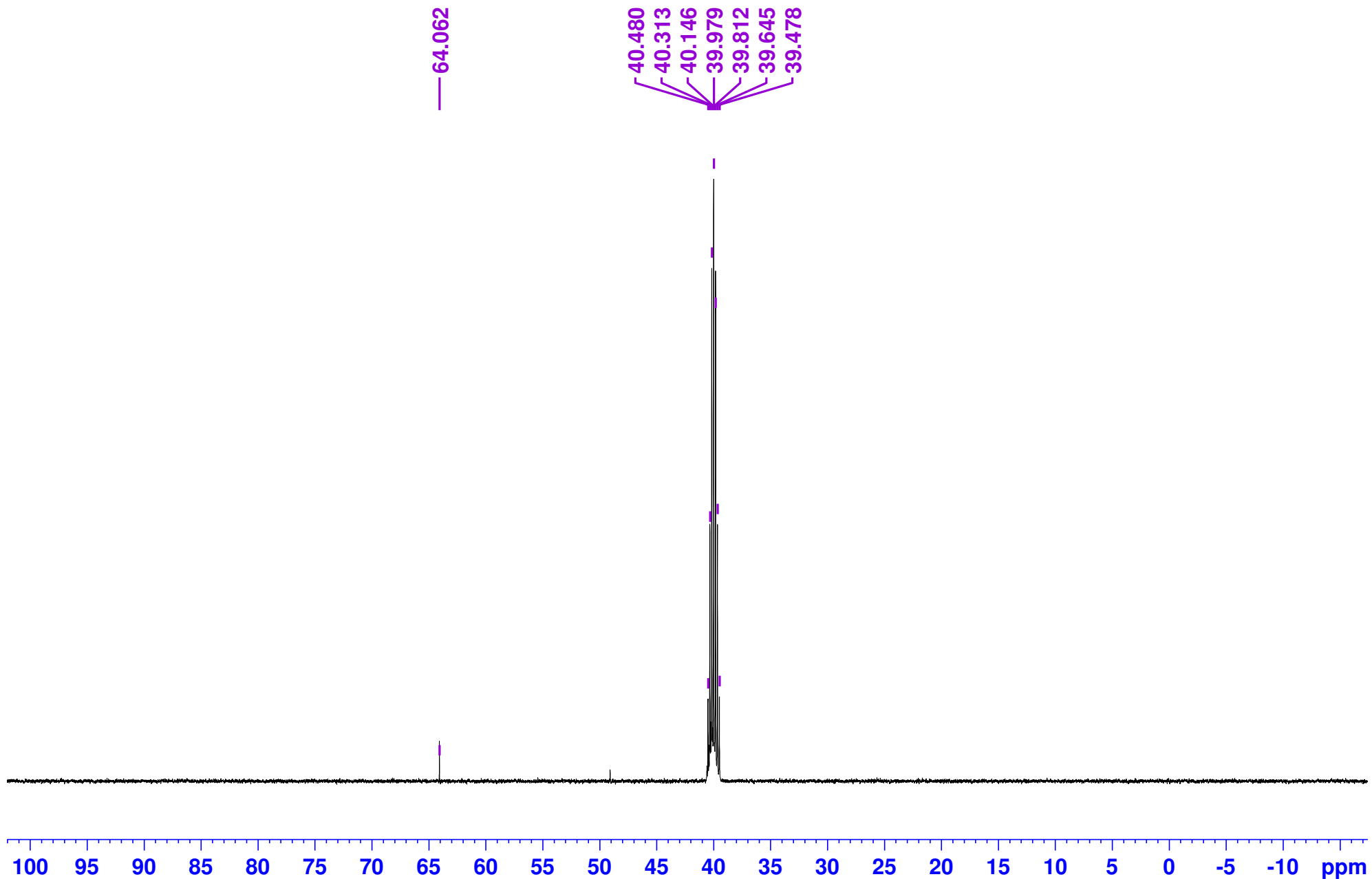
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Compound 1 (13C NMR, 125 MHz, DMSO-d6)



Compound 1 (13C NMR, 125 MHz, DMSO-d6)



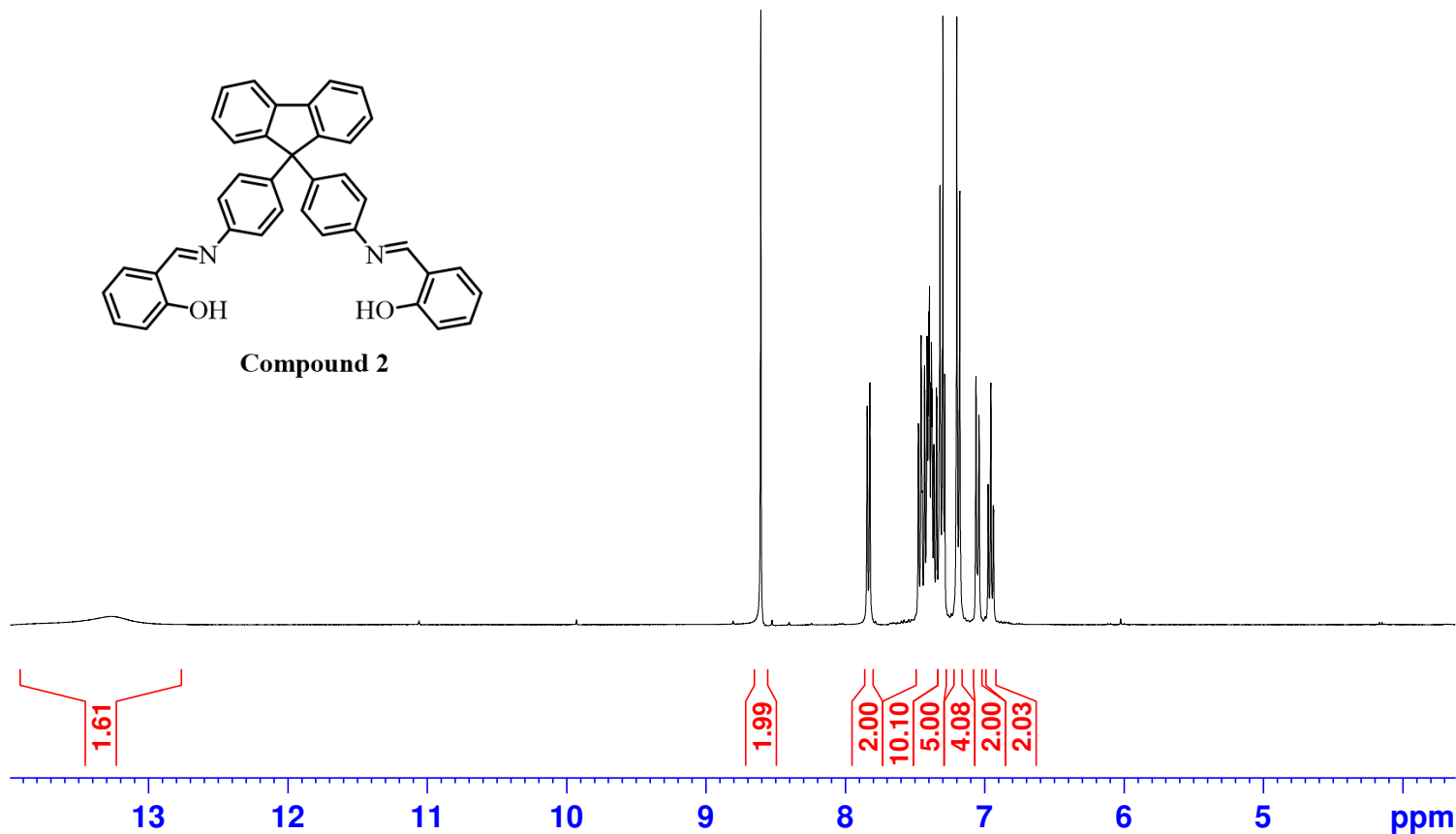
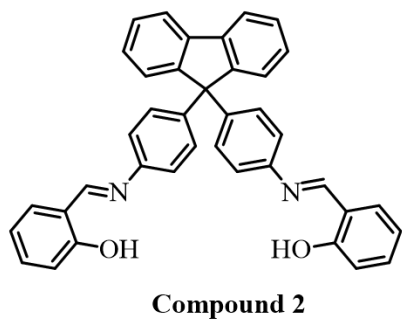
S21

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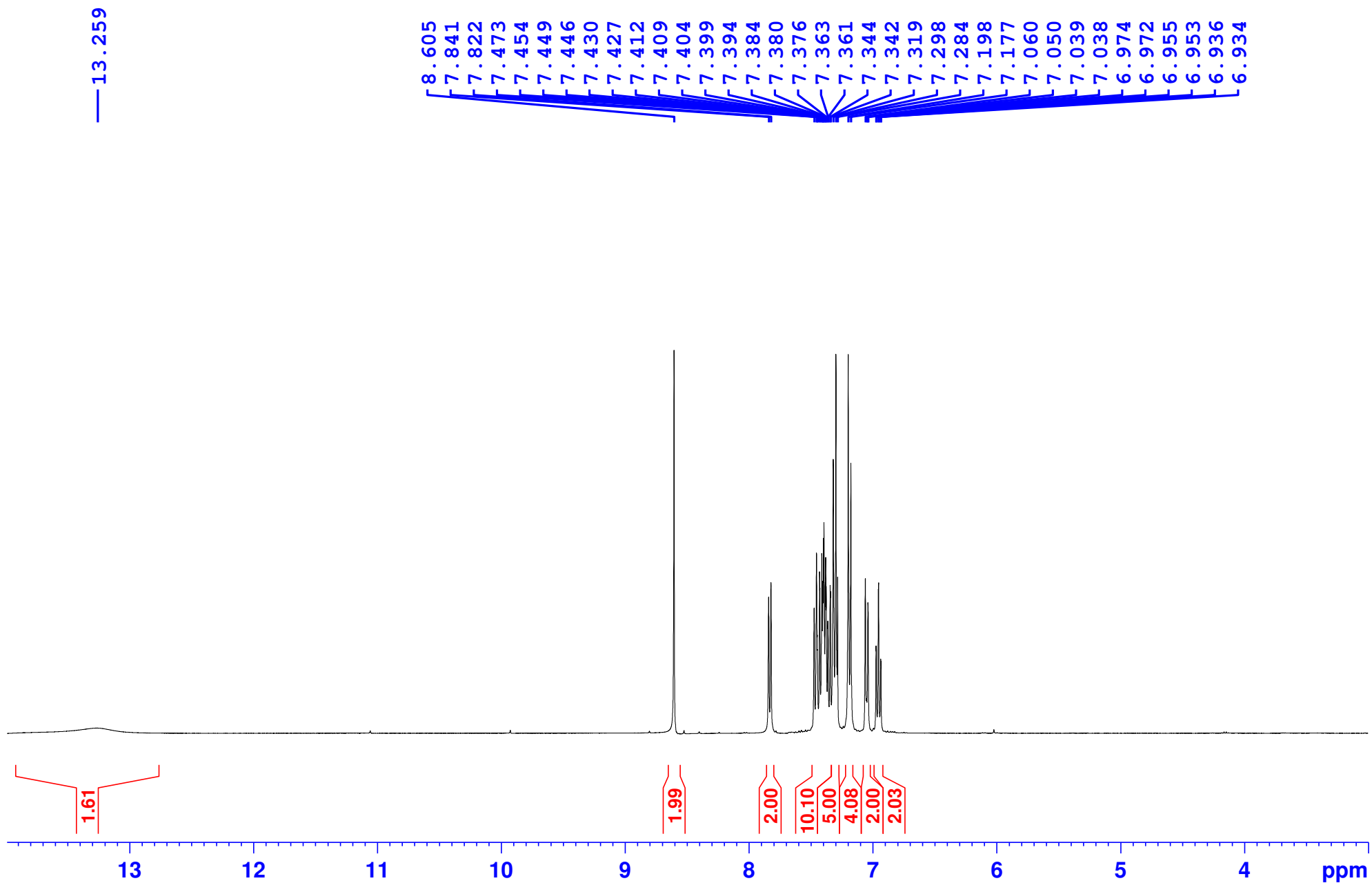


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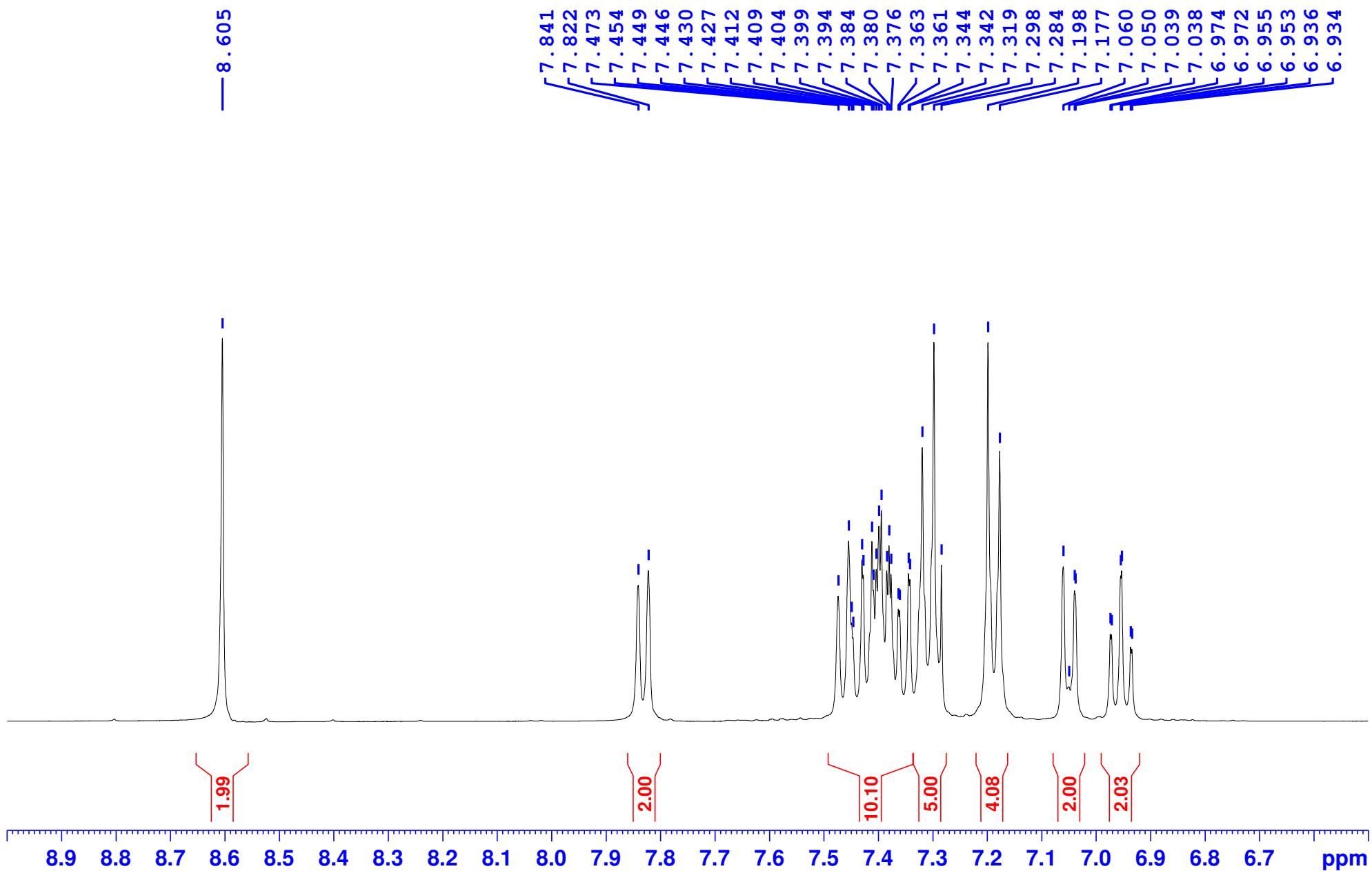
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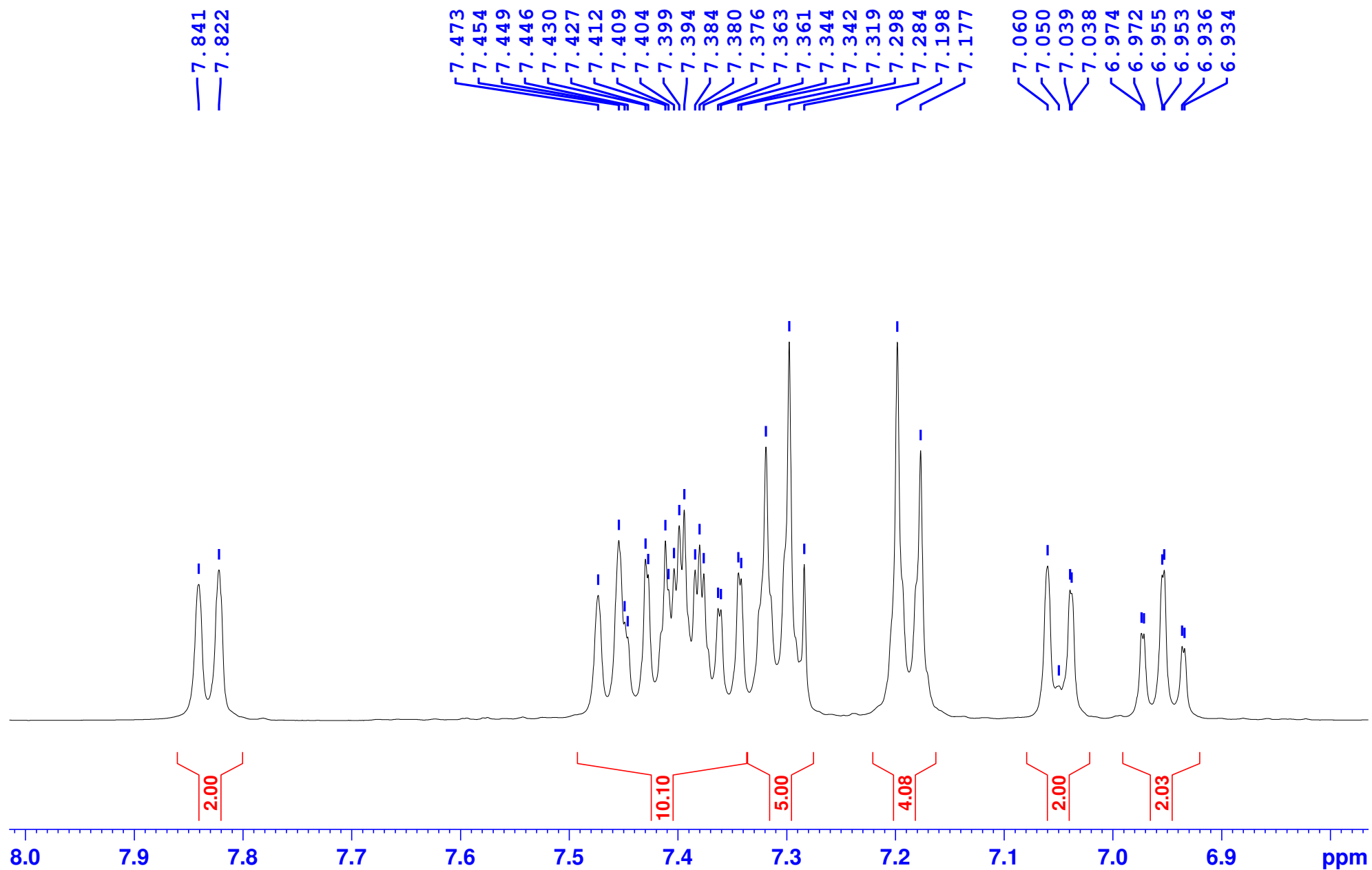
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Compound 2 (1H NMR, 400 MHz, CDCl3)



Compound 2 (1H NMR, 400 MHz, CDCl3)



Compound 2 (13C NMR, 400 MHz, CDCl3)

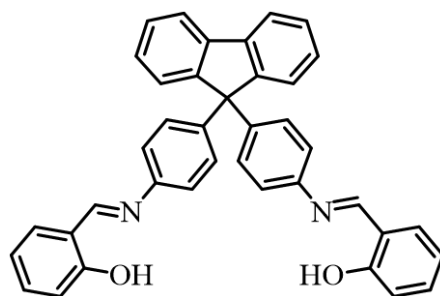


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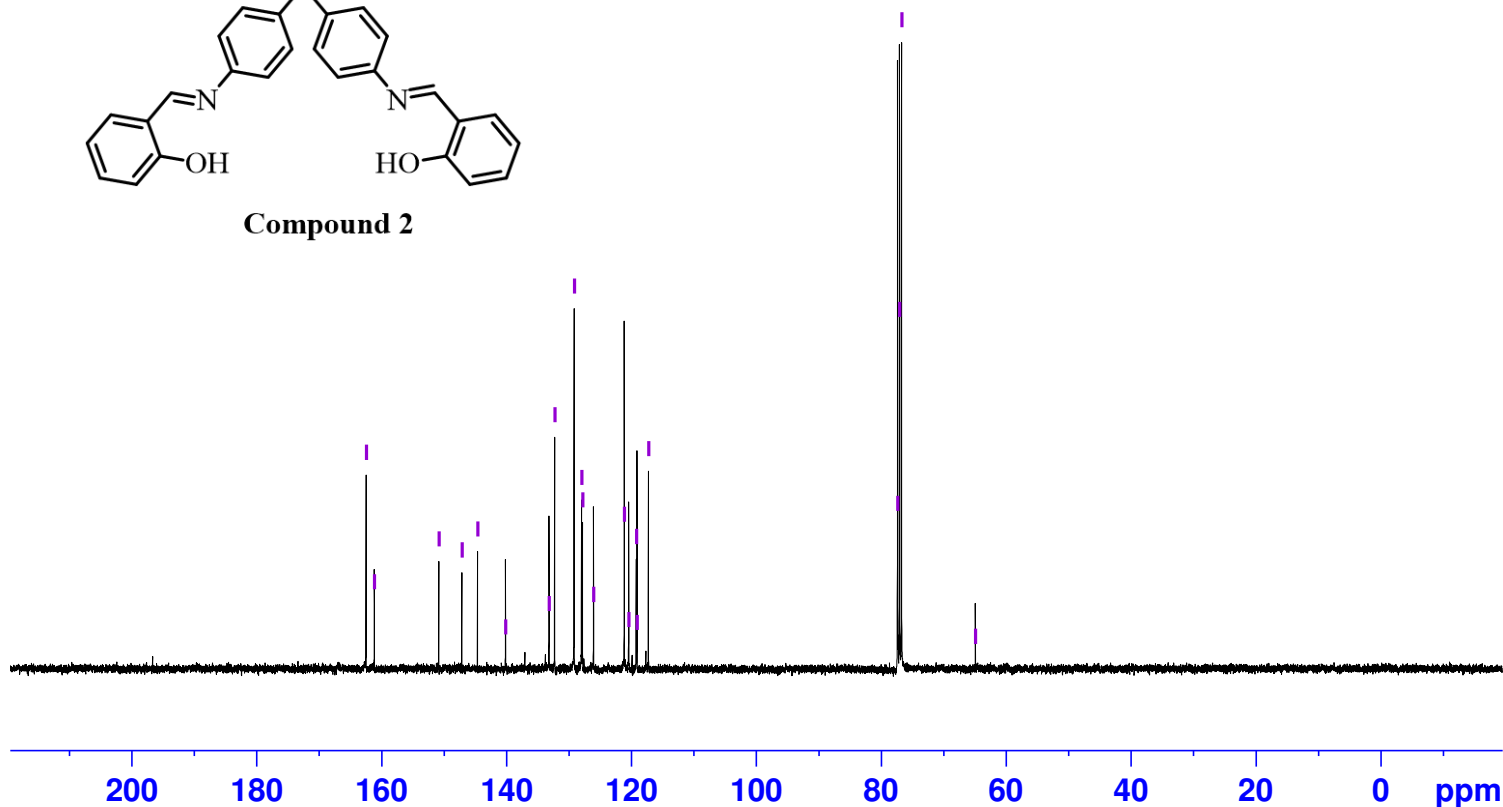
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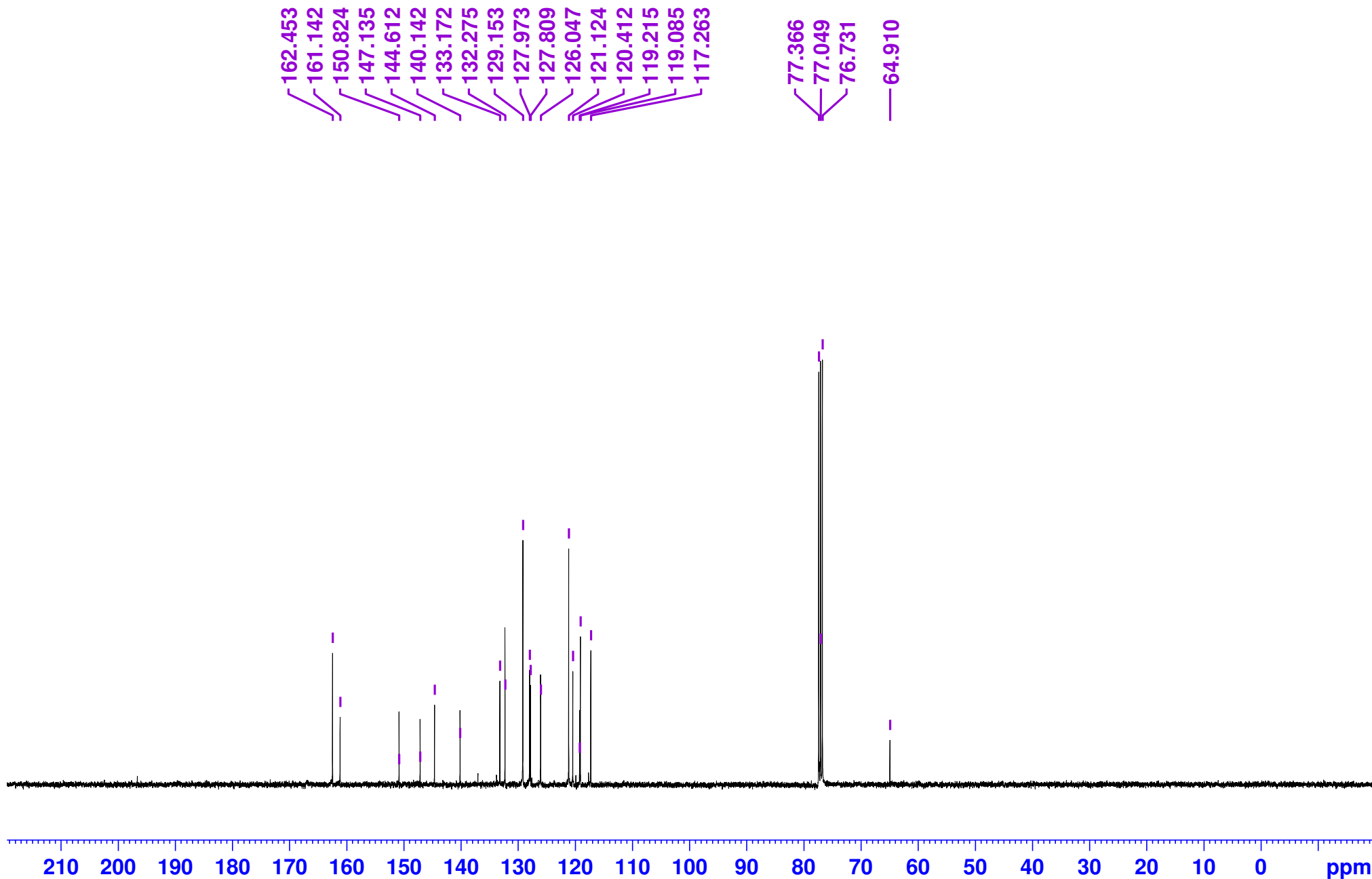
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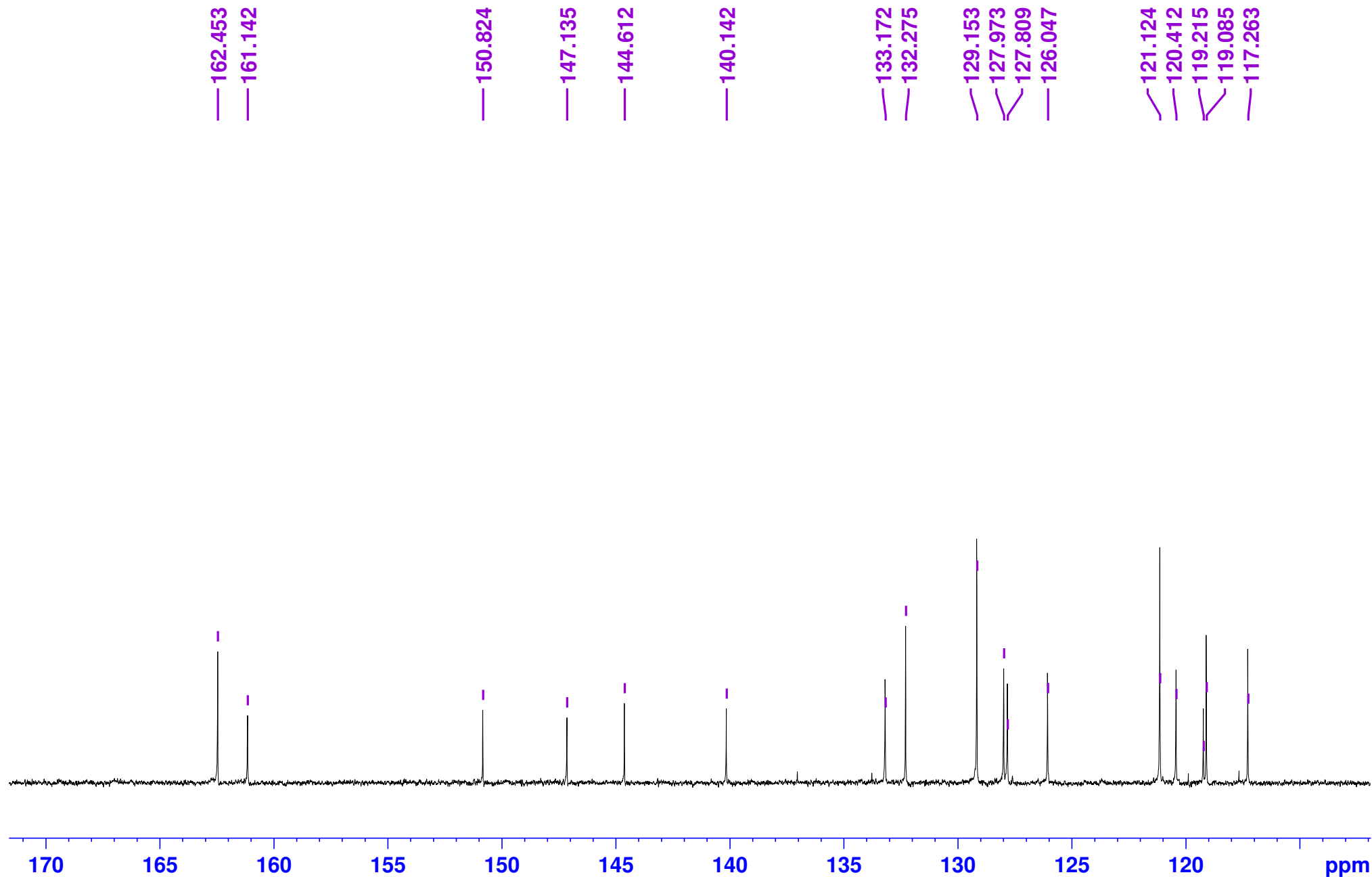
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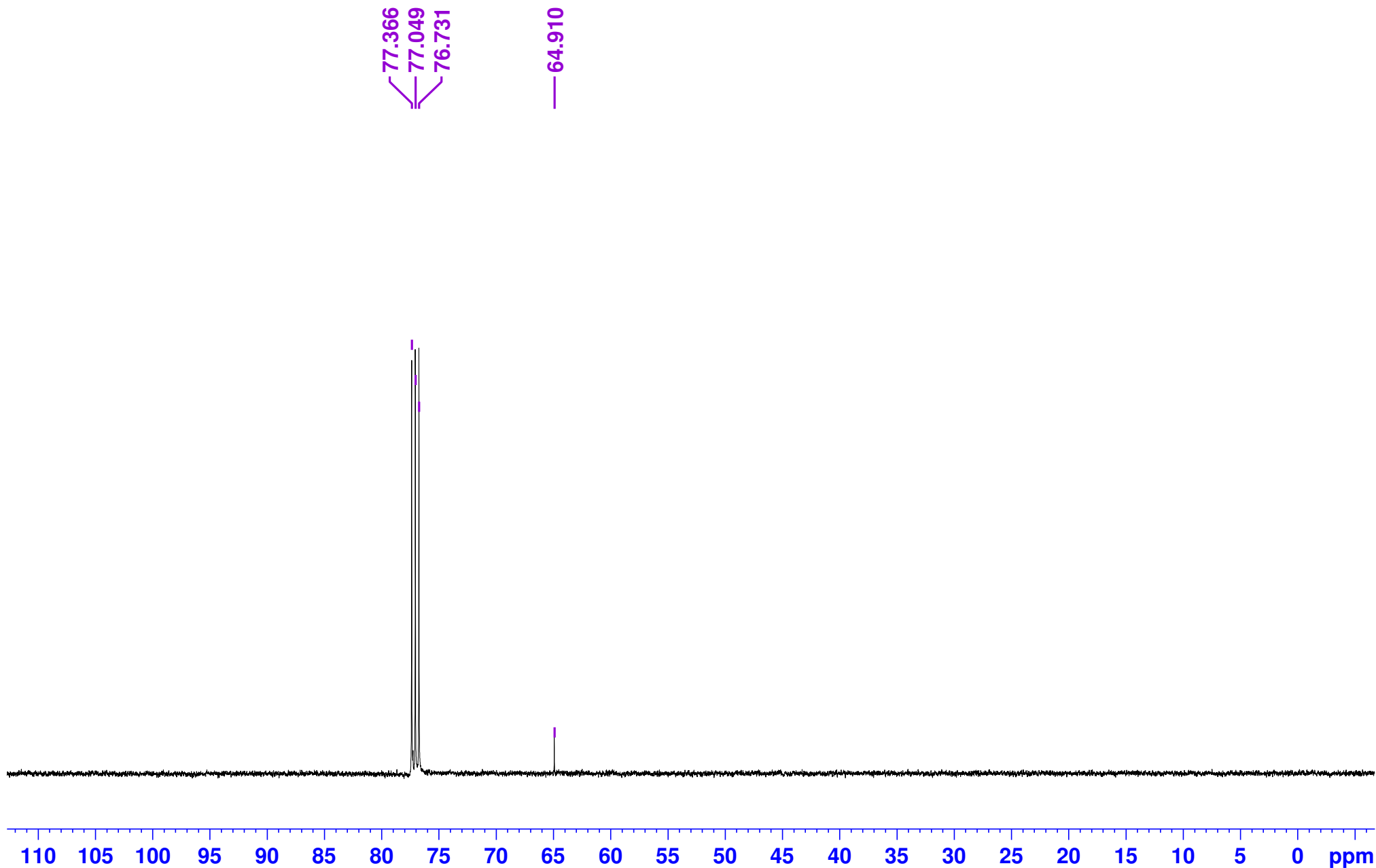
Compound 2 (¹³C NMR, 400 MHz, CDCl₃)



Compound 2 (¹³C NMR, 400 MHz, CDCl₃)



Compound 2 (¹³C NMR, 400 MHz, CDCl₃)



Compound 2 (Mass spectrum)

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