

Au nanoclusters-decorated WO₃ nanorods for ultrasensitive photoelectrochemical sensing of Hg²⁺

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Instrumentation

Transmission electron microscopy (TEM) studies were performed on a TECNAI F-30 high-resolution transmission electron microscope. Scanning electron microscopy (SEM) studies were performed on a Hitachi S4800 scanning electron microscope. X-ray diffraction (XRD) pattern was obtained on a PANalytical X'pert Pro X-ray diffractometer using Cu K α radiation, operating at 40 kV and 30 mA, with a scan rate of 4° min⁻¹. X-ray photoelectron spectroscopy (XPS) measurement was investigated using a PHI QUANTUM 2000 X-ray photoelectron spectroscopic instrument. Ultraviolet-Visible (UV-Vis) spectra were obtained on U3310 UV-Vis spectrometer (Hitachi, Japan). Fourier transform infrared (FTIR) spectrum was collected on Nicolet Nexus 670 Fourier transform infrared spectrometer. All electrochemical experiments were carried out on the CHI760E electrochemical workstation. All photoelectrochemical experiments were carried out on electrochemical workstation (ZENNIUM, ZAHNER-elektrok GmbH & Co. KG, Germany) equipped with a controlled-intensity modulated-photospectroscopy setup (CIMPS, PP211, ZAHNERelektrok GmbH & Co. KG, Germany). A white light lamp (WLC02, ZAHNER-elektrok GmbH & Co. KG, Germany) with a visible-light filter (400-700 nm) was used as the light source.

Chemicals

Sodium tungstate, sodium chloride, mercury chloride, copper chloride, iron chloride, zinc chloride, cobalt chloride, potassium chloride, chromium chloride, calcium chloride, cadmium chloride, nickel chloride, hydrochloric acid and ascorbic acid (AA) were purchased from Sinophem Chemical Reagents Co. LTD. (Shanghai, China). Glutathione was purchased from Aladdin. H₂AuCl₄·3H₂O, 4-chloro-1-naphthol were purchased from Sigma Aldrich. All

reagents are analytical pure and used without further purification. Milli-Q ultrapure water (Millipore, $\geq 18 \text{ M}\Omega \text{ cm}$) was used throughout.

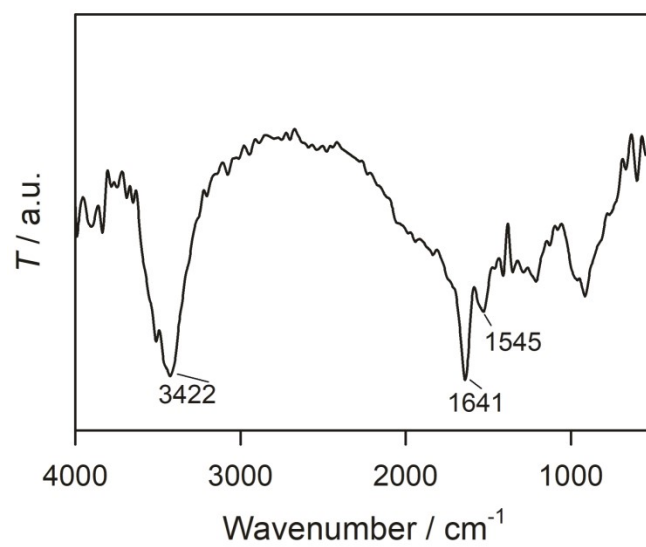


Fig. S1 FT-IR spectrum of AuNCs.

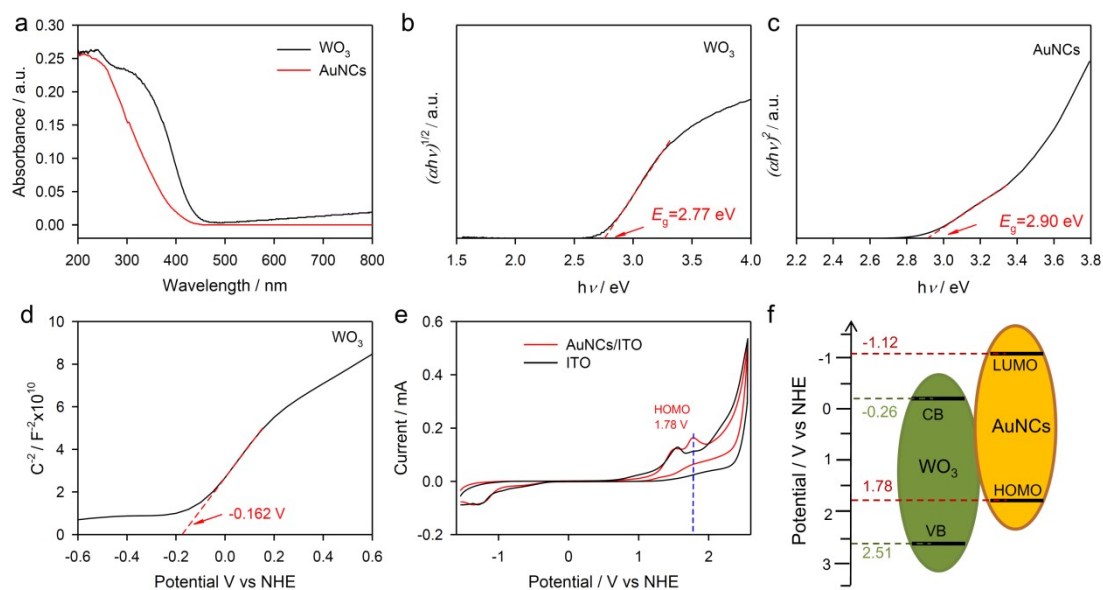


Fig. S2 (a) UV-vis diffuse reflectance spectra of WO₃ and AuNCs. (b) Plot of $(\alpha h\nu)^{1/2}$ versus $h\nu$ for WO₃. (c) $(\alpha h\nu)^2$ versus $h\nu$ for AuNCs. (d) Mott-Schottky plot of WO₃. (e) Cyclic voltammogram of AuNCs in degassed acetonitrile solution containing 0.1 M tetrabutyl ammonium perchlorate. (f) Illustration of energy-levels of AuNCs and WO₃.

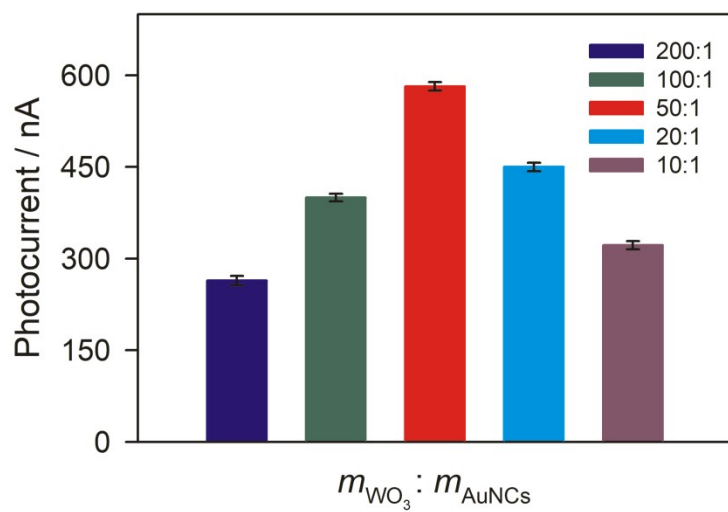
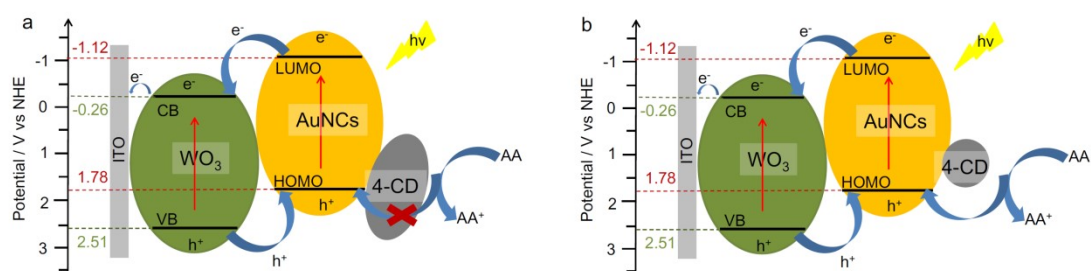


Fig. S3 Effects of weight ratios of WO₃ to AuNCs in WO₃/AuNCs on transient photocurrents of the prepared WO₃/AuNCs.



Scheme S1 (a) Charge transfer pathway of WO₃/AuNCs/ITO after 4-CN + H₂O₂ incubation.

(b) Charge transfer pathway of WO₃/AuNCs/ITO after Hg²⁺ incubation and then 4-CN + H₂O₂ incubation.

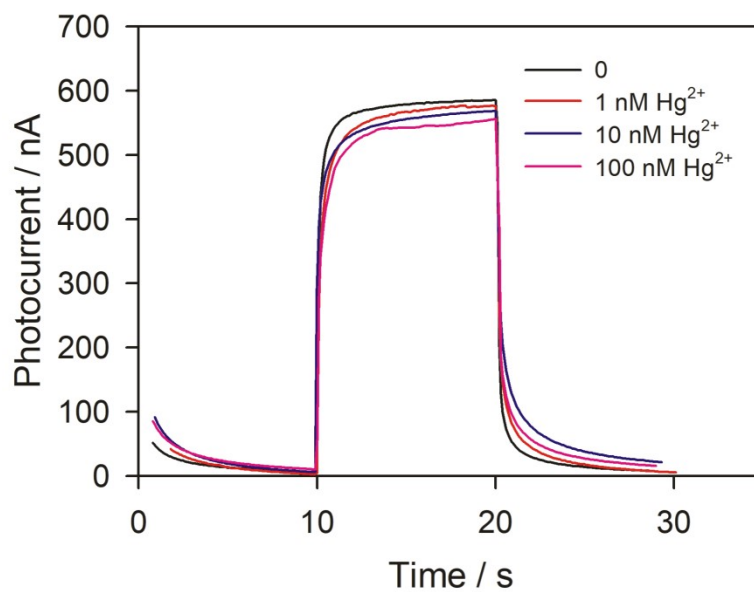


Fig. S4 Transient photocurrents of WO₃/AuNCs/ITO after the incubation with different concentrations of Hg²⁺.

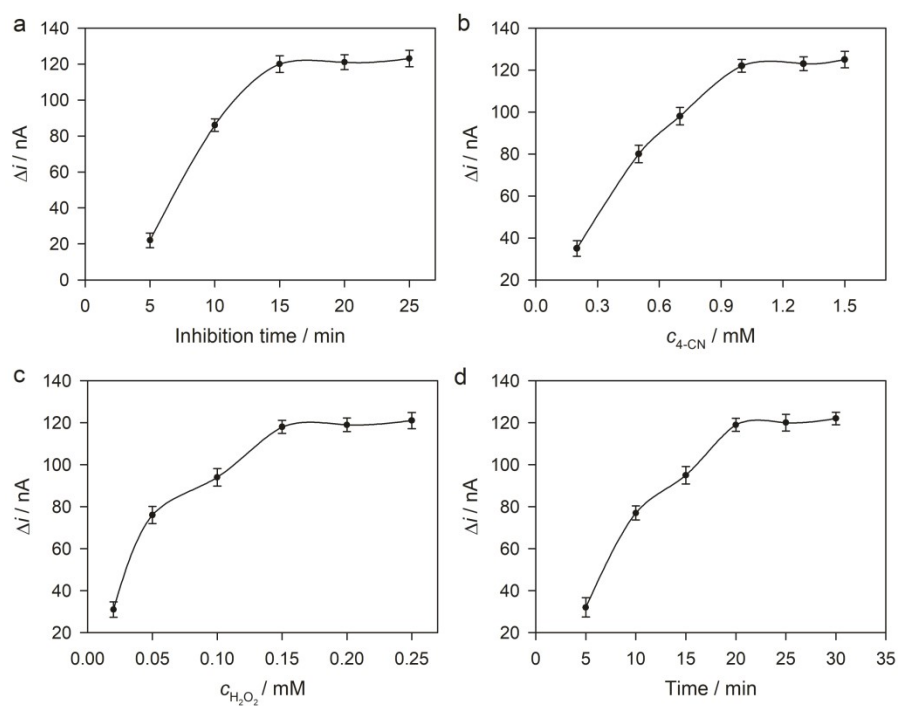


Fig. S5 Effects of (a) Hg^{2+} inhibition time, (b) 4-CN concentration, (c) H_2O_2 concentration, and (d) precipitation reaction time on photocurrent responses for detecting 10 nM Hg^{2+} .

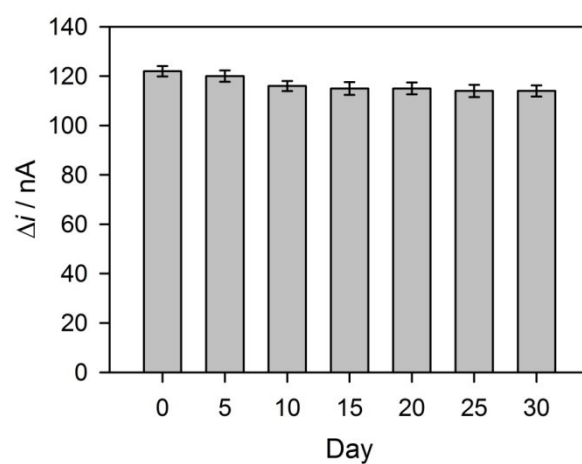


Fig. S6 Storage stability of the PEC sensor for detection of 10 nM Hg^{2+} .

Table S1. The comparison of the determination of Hg²⁺ in the literature.

Technique	Linear range (pM)	Detection limit (pM)	Ref.
Electrochemistry	$1.0 \times 10^4 - 1.0 \times 10^6$	5.8×10^3	1
Electrochemistry	$1.0 \times 10^2 - 1.3 \times 10^5$	30	2
Electrochemistry	$5.0 \times 10^2 - 5.0 \times 10^6$	2.3×10^2	3
Colorimetry	$3.0 \times 10^3 - 5.0 \times 10^5$	3.0×10^3	4
Colorimetry	$2.0 \times 10^4 - 2.0 \times 10^7$	5.0×10^3	5
Fluorescence	10- 1.0×10^5	4.5	6
Fluorescence	100- 1.0×10^5	65	7
Fluorescence	$1.0 \times 10^3 - 5.0 \times 10^6$	3.5×10^2	8
Fluorescence	$1.0 \times 10^5 - 6.0 \times 10^6$	3.8×10^3	9
Fluorescence	0- 1.6×10^3	1.1×10^2	10
Surface-enhanced Raman scattering	1- 10^3	0.4	11
Surface-enhanced Raman scattering	100- 10^4	30	12
Photoelectrochemistry	1.0- 2.0×10^3	1.0	13
Photoelectrochemistry	1.0- 5.0×10^3	0.5	14
Photoelectrochemistry	1.0- 5.0×10^4	0.2	This work

Table S2. Results of determination of Hg²⁺ in a river water sample.

Original (pM)	Added (pM)	Detected (pM)	Recovery (%)	RSD
	10.00	58.30	94.0	3.3
52.00	100.0	146.0	96.0	4.7
	1000	1092	104	1.1

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