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ARTICLE

## Fluorescence digital image-based method using carbon quantum dots to evaluate the compliance of a biocidal agent

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### Electronic Supplementary Information

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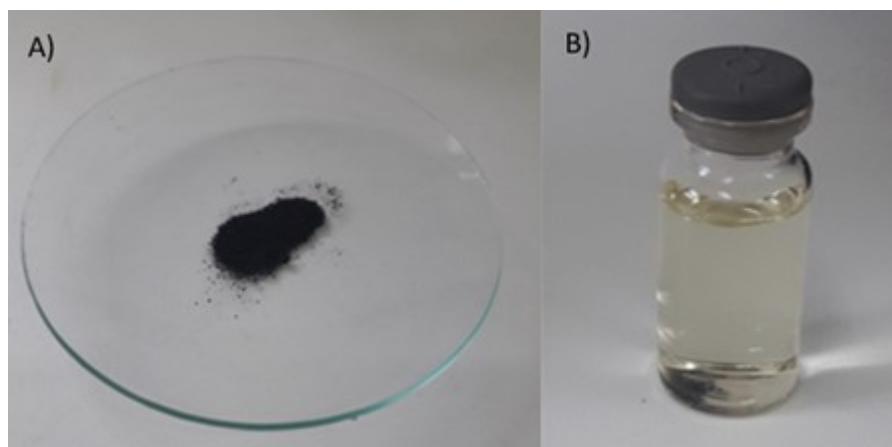
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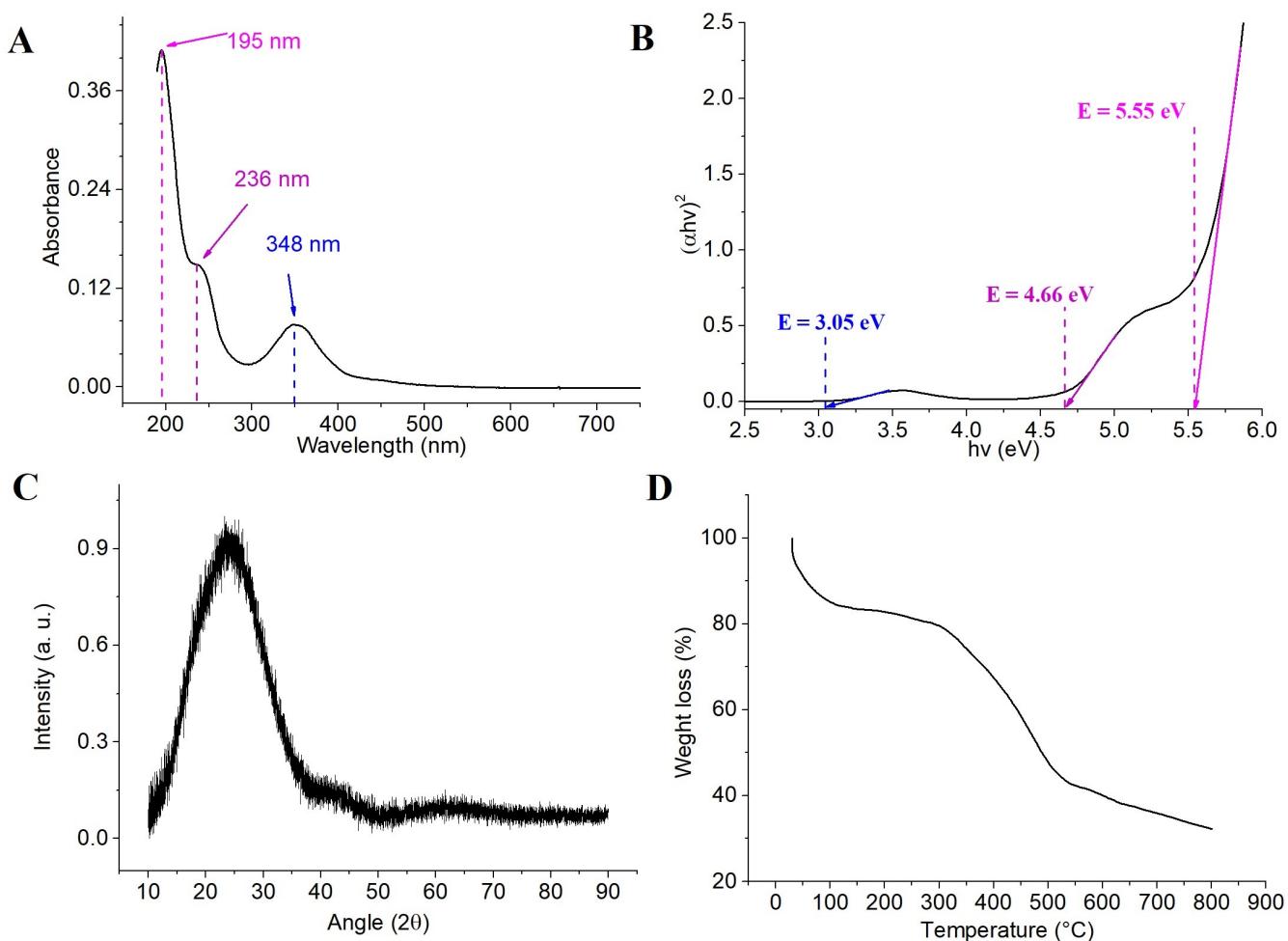
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Electronic Supplementary Information (ESI) available: See DOI: 10.1039/x0xx00000x

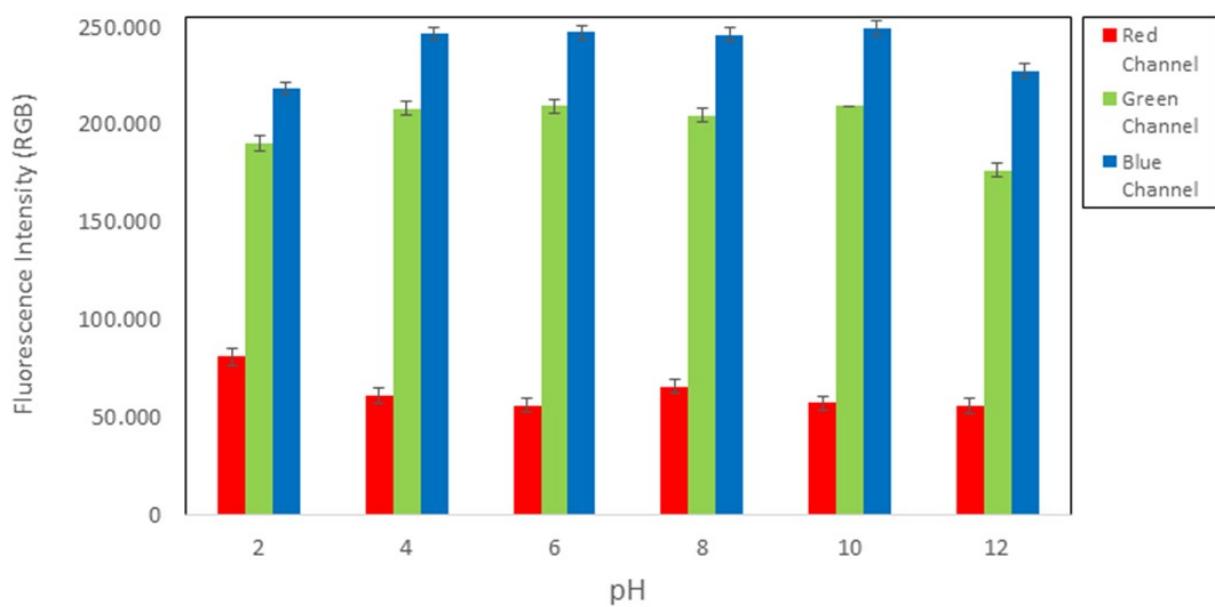
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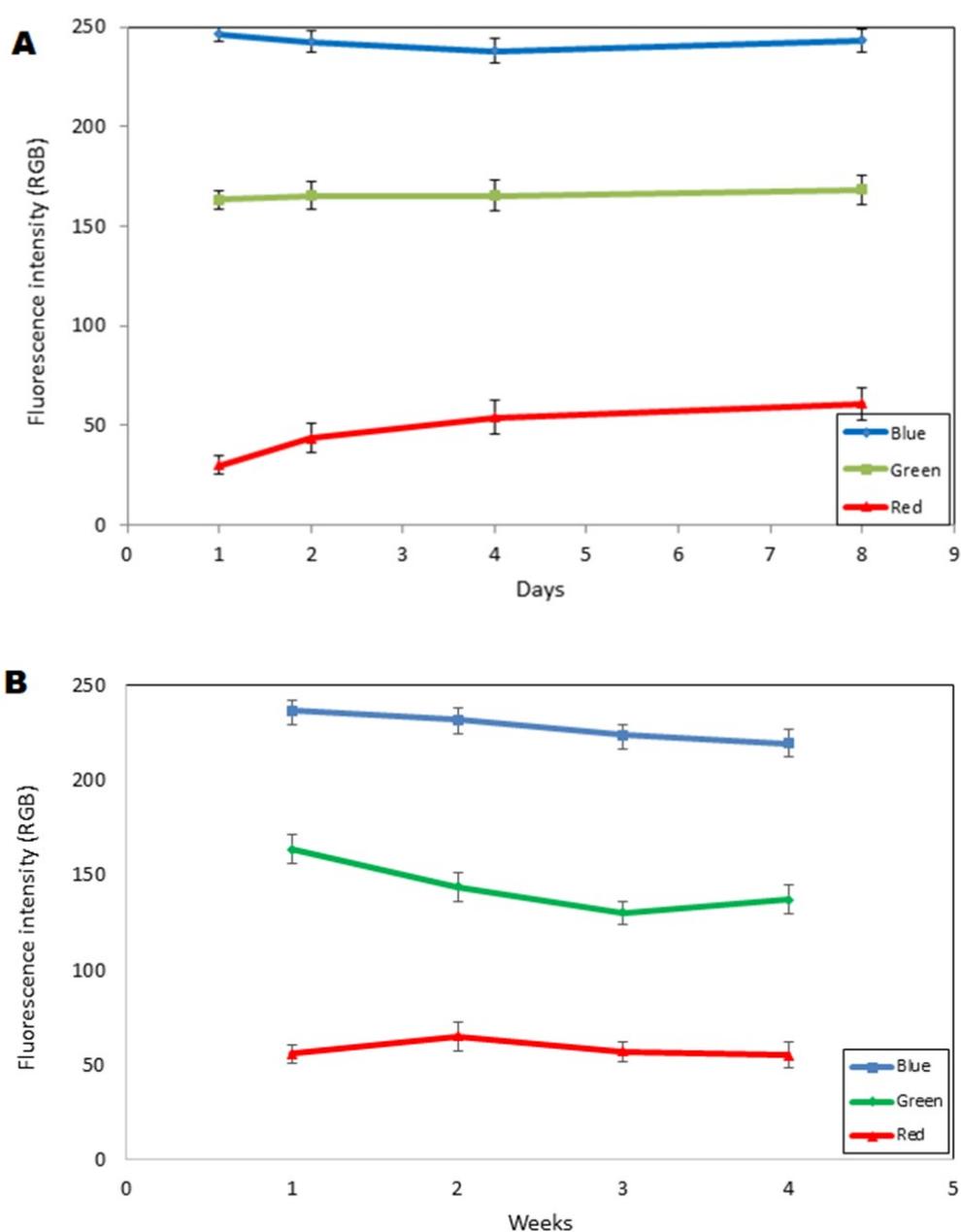
**Fig. S1.** A) Carbon nanoparticles after extraction, B) CQD solution at a concentration of 1.6 g/L in water.



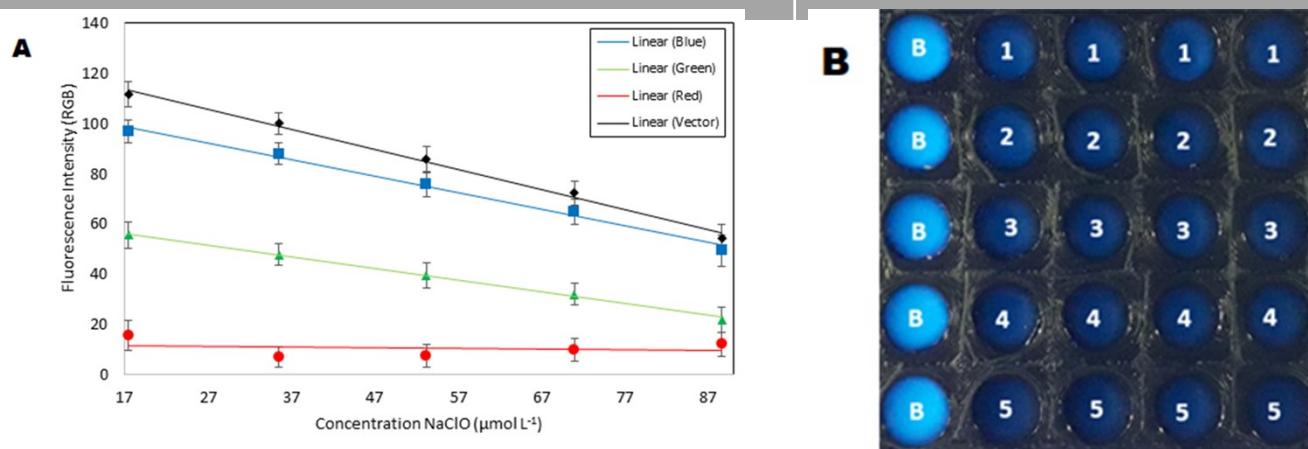
**Fig. S2.** A) Absorption spectrum of CQD in water using a suspension with 1.6 g/L. B) Tauc plot from the UV-Vis spectrum. C X-ray diffractogram for synthesized CQD and D) Thermogravimetric analysis of CQD nanoparticles.



**Fig. S3.** Study of the influence of pH on the fluorescence of CQDs.



**Fig. S4.** A) Study of the stability of CQDs regarding fluorescence in one week and B) in one month.



**Fig. S5.** A) Analytical curves to detection of hypochlorite using the FDIB from a range of  $17.44 \mu\text{mol L}^{-1}$  to  $90 \mu\text{mol L}^{-1}$  for R, G and B channels and vectorial calculation. B is the solution of  $0.16 \text{ g / L}$  of CQD in water. B) From 1 to 5 the concentration of hypochlorite was  $17.44 \mu\text{mol L}^{-1}$  (1),  $35.44 \mu\text{mol L}^{-1}$  (2),  $53.16 \mu\text{mol L}^{-1}$  (3),  $70.88 \mu\text{mol L}^{-1}$  (4) and  $90 \mu\text{mol L}^{-1}$  (5), respectively with  $n = 5$ .

**Table S1.** Results of the linear regression for the blue, red, green and vector channels, as well as the coefficient of determination ( $R^2$ ), LD and LQ values.  $[\text{NaClO}] / \text{mmol L}^{-1}$

Channel	Linear equation	$R^2$	LD/ $\mu\text{mol L}^{-1}$	LQ/ $\mu\text{mol L}^{-1}$
Vector (G, B)	$y = (-0.803 \pm 0.086) [\text{NaClO}] + (127.450 \pm 4.872)$	0.9935	3.30	10.96
Blue	$y = (-0.662 \pm 0.092) [\text{NaClO}] + (110.180 \pm 5.015)$	0.9908	4.00	13.29
Green	$y = (-0.467 \pm 0.083) [\text{NaClO}] + (64.253 \pm 4.854)$	0.9973	5.67	18.84

**Table S2.** Monitoring of the NaClO concentrations in the mat using the FDIB method and a spectrofluorometry used as a reference method. The F-test (tabulated = 19.00) and t-test (tabulated = 12.706) data for a confidence level of 95% with  $n = 2$  is presented.

Time/min	FDIB ([NaClO]/mmol L <sup>-1</sup> )	Spectrofluorometry ([NaClO]/mmol L <sup>-1</sup> )	F-Test	t-Test
5	10.53 ± 0.47	9.84 ± 1.61	11.73	0.469
30	9.39 ± 0.46	10.43 ± 0.30	2.351	1.935
60	9.24 ± 0.49	10.28 ± 0.25	3.842	1.988
90	9.13 ± 0.51	9.97 ± 0.51	1.000	1.165
120	9.12 ± 0.49	8.54 ± 0.42	1.361	0.901
150	8.29 ± 0.40	8.60 ± 0.71	3.151	0.395
180	7.63 ± 0.47	8.24 ± 0.84	3.194	0.659
210	7.24 ± 0.49	8.15 ± 0.85	3.009	0.960
240	7.18 ± 0.47	7.70 ± 0.32	2.157	0.931
270	0.17 ± 0.59	0.43 ± 0.87	2.174	0.252
300	0.02 ± 0.32	0.06 ± 0.52	2.641	0.067