Supporting Information

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Scheme S1. The chemical structures of some common MALDI matrices.



Scheme S2. Structures of N-heterocyclic molecules, investigated as initiators for ROP of lactide. Reprinted with permission from ref. 130.



Scheme S3. Terpolymerizations of TAz, PA, and PO. The synthesis of diblock dialternating terpolymers with a sharp boundary instead of a tapered sequence at the conjunction between the two alternating blocks. Reprinted with permission from ref. 133.

No.	polymer	Symbol	Structure
1	polypropylene glycol	PPG	$\left[\begin{array}{c} CH_3 \\ 0 \end{array} \right]_n$
2	poly(ethylene glycol)	(PEG)	$ \begin{bmatrix} -CH_2 - CH_2 - O \end{bmatrix}_n$
3	polyisoprene	PIP	$- \left[- CH_2C(CH_3) = CHCH_2 \right]_n$
4	polystyrene	PS	-CH ₂ -CH-n
5	polybutadiene	PB	
6	polyvinylpyrrolidone	PVP	$ CH - CH_2 + n$
7	poly(ethylene oxide)	PEO	$-\left(-\text{OCH}_2\text{CH}_2\right)_{n}$
8	poly(dimethyl siloxane)	PDMS	$- \begin{bmatrix} -O & -Si \end{bmatrix}_{n}^{CH_3}$
9	poly(methylmethacrylate)	РММА	$- \underbrace{\begin{bmatrix} H_2 \\ - \\ - \end{bmatrix}}_{C} \underbrace{\begin{bmatrix} CH_3 \\ - \\ - \\ - \end{bmatrix}}_{n}_{COOCH_3}$
10	poly (methyl acrylate)	РМА	$-\left[-CH_2-CH_{n}\right]_n$ COOCH ₃
11	poly (δ-valerolactone)	PVL	$\left\{ \begin{array}{c} O \\ O \\ O \\ O \end{array} \right\}_n$
12	Poly(ε-caprolactone)	PCL	√o∱n o
13	Poly(lactic acid)	PLA	
14	Nylon-6	Ny6	- $ -$

Table S1. The structures and types of the synthetic homopolymers commonly used

for the MALDI-TOF-MS analyses

Table S2. The structures and types of the synthetic copolymers commonly used for the

MALDI-TOF-MS analyses

No.	polymer	Symbol	Structure
1	poly(styrene-co-4-vinylpyridine)	P(S-4VP)	$ \begin{array}{c} $
2	polymethyl methacrylate-co- methacrylic acid	(PMMA-co- MAA)	O O O O O O O O H
3	poly (2-oxazoline)	P(EtOx-b-EPOx)	
4	hybrid poly (butylene adipate) copolymers	НРВА	$CH_{2}-CH_{2}-CO-(CH_{2})_{4}-CO-(-PBA)^{**}$ $CH_{2}-CC+CH_{2}-CO-(CH_{2})_{4}-CO-(-PBA)^{**}$ $CH_{2}-CO-(-CH_{2})_{4}-CO-(-PBA)^{**}$ $*PBA= +CO(CH_{2})_{4}CO-O-(CH_{2})_{4}O + \frac{1}{X_{1}}$
6	poly (styrene oxide- <i>alt</i> - succinic anhydride)	P(SO-alt-SA)	
7	poly (phthalic anhydride- <i>alt</i> - cyclohexane oxide)	P(PA-alt-CHO)	to in the second
8	poly(1,4-butadiene)-graft-poly(L- lactide)	PBun-g-PLLAm	

Kendrick analysis

The Kendrick analysis adds a second dimension to the mass spectrum through the evaluation of the "fractional parts" of the accurately mass measured m/z values. The fractional part of m/z contains information about the elemental composition of the associated ions. Plotting the fractional parts of m/z as a function of m/z, while at the same time retaining the intensity of the associated peaks via colour coding or point size, produces a two-dimensional map of the mass spectrum¹. It can help us to explore the complex mass spectral data or filtering the unwanted signal.



Fig. S1 Generic representation of the change in viewpoint from a one-dimensional mass spectrum to a two-dimensional Kendrick plot. Reprinted with permission from ref. 79.