

Supporting information

**Ultrafast colorimetric detection of Cr(VI) using
Fe₃O₄@polydopamine/prussian blue composite as
highly-efficient peroxidase mimic**

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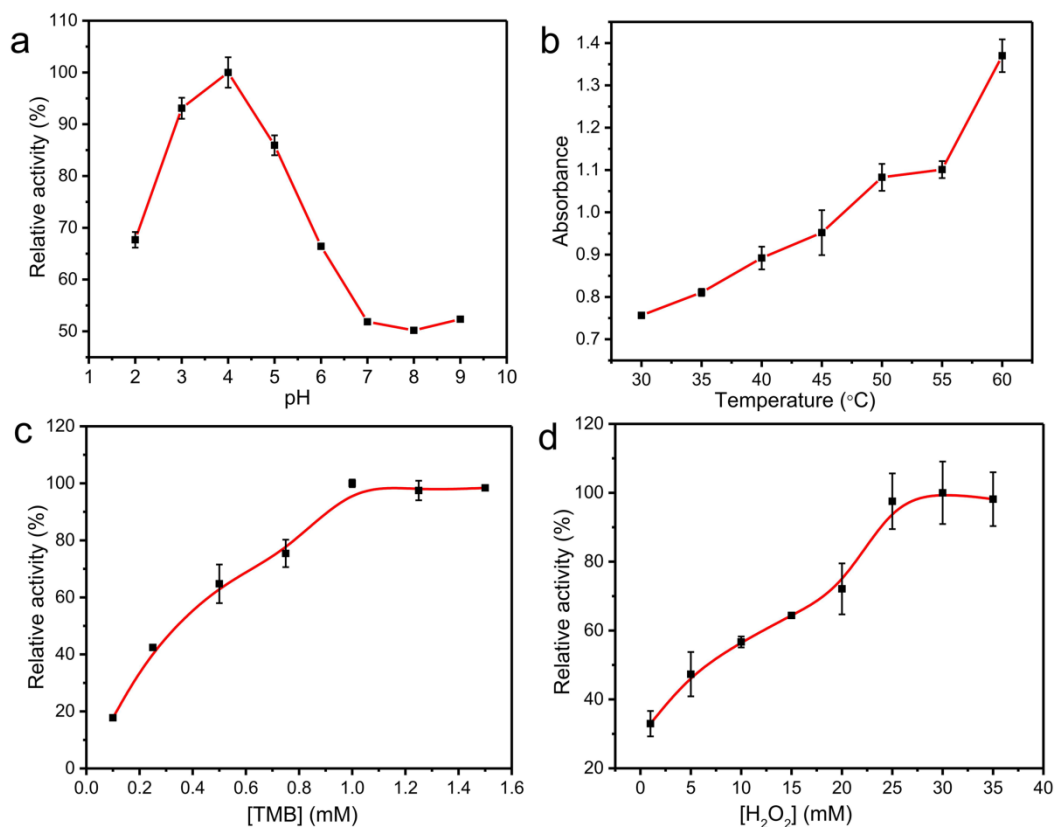


Fig. S1 The effects of experiment conditions on the peroxidase-like activity of $\text{Fe}_3\text{O}_4@PDA/PB$ composites: (a) pH, (b) temperature, (c) the concentration of TMB and (d) the concentration of H_2O_2 .

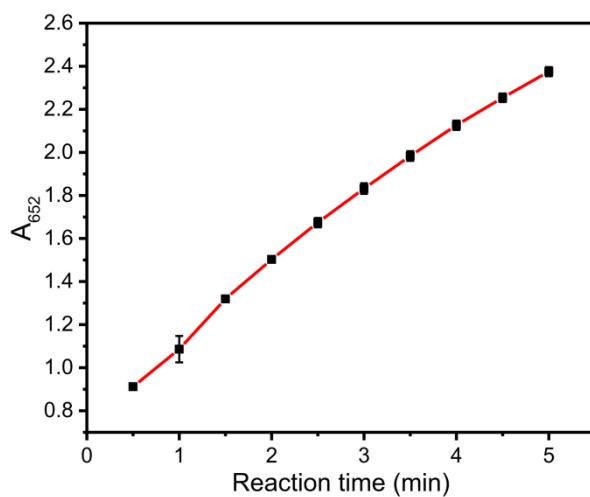


Fig. S2 A_{652} of the $\text{Fe}_3\text{O}_4@PDA/PB+TMB+H_2O_2$ reaction system as a function of reaction time.

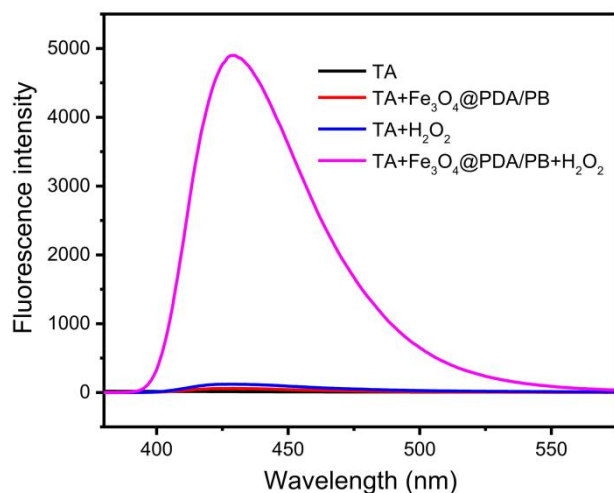


Fig. S3 Fluorescence emission spectra of TAOH from different reaction systems for inferring the catalytic mechanism of the $\text{Fe}_3\text{O}_4@PDA/PB$ composites.

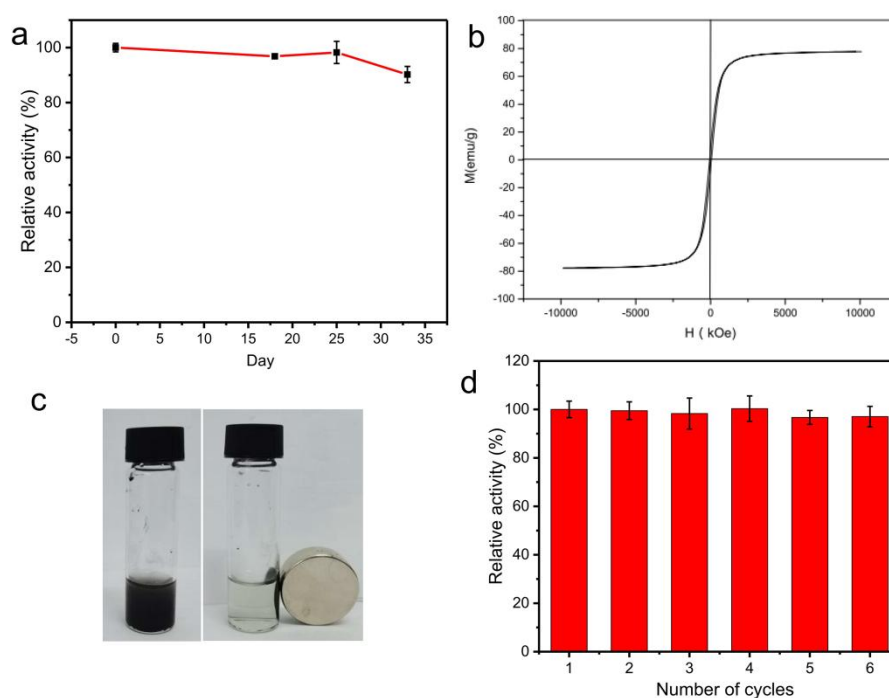


Fig. S4 (a) The relative activity of $\text{Fe}_3\text{O}_4@PDA/PB$ composites during the storage for 33 days at 4 °C. (b) Magnetic hysteresis curve of Fe_3O_4 NPs (provided by the supplier). (c) Photograph of $\text{Fe}_3\text{O}_4@PDA/PB$ composites before and after magnetic treatment for 1 min. (d) The relative activity of $\text{Fe}_3\text{O}_4@PDA/PB$ composites for six consecutive cycles.

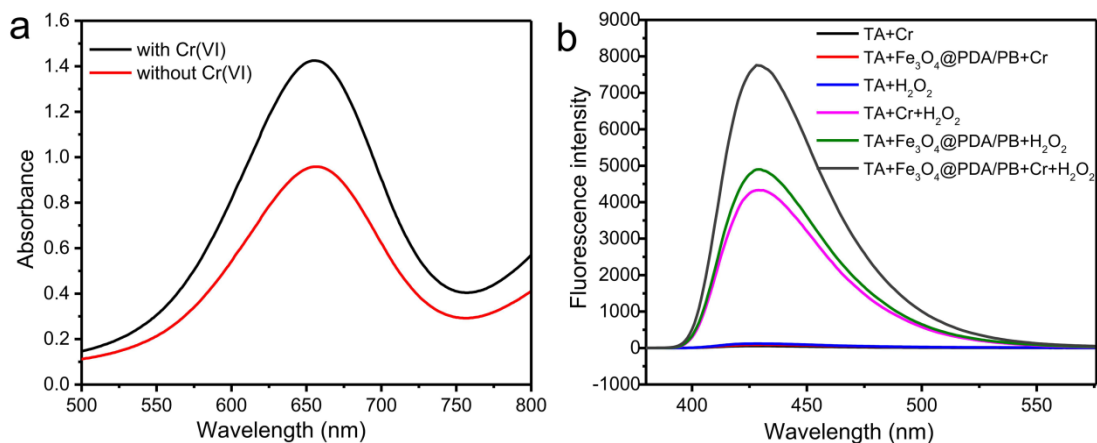


Fig. S5 (a) UV-vis absorption spectra of the Fe₃O₄@PDA/PB+TMB+H₂O₂ reaction system in the absence and presence of Cr(VI). (b) Fluorescence emission spectra of TAOH from different reaction systems to explain the detection principle of Cr(VI).

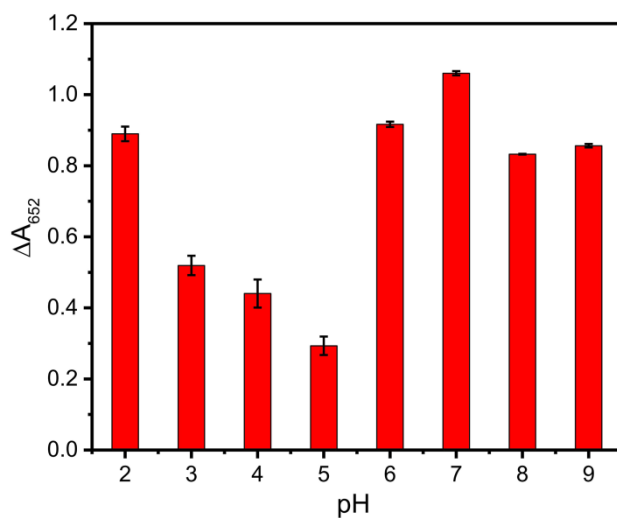


Fig. S6 ΔA_{652} response of the Fe₃O₄@PDA/PB+TMB+H₂O₂ reaction system in the presence of Cr(VI) at different pH.

2. Additional Table S1

Table S1 Comparison of the peroxidase-like activity (kinetic parameters and conditions) of Fe₃O₄@PDA/PB with others.

Catalysts	Substrate	K _m	V _{max} (M s ⁻¹)	pH	temperature	Ref
HRP	TMB	0.434	1×10 ⁻⁷	4.0	40 °C	[1]
	H ₂ O ₂	3.70	8.71×10 ⁻⁸			
CuS-BSA	TMB	0.2	2.2×10 ⁻⁸	3.9	25 °C	[2]
	H ₂ O ₂	0.014	2×10 ⁻⁸			
PNPG-PEG	TMB	0.2828	3.0276×10 ⁻⁹	4.0	35 °C	[3]
	H ₂ O ₂	0.0799	2.467×10 ⁻⁹			
CoFe ₂ O ₄ /H ₂ PPOP	TMB	0.075	6.54×10 ⁻⁵	3.35	35 °C	[4]
	H ₂ O ₂	2.68	7.44×10 ⁻⁵			
MoS ₂	TMB	0.387	7.23×10 ⁻⁸	4.0	25 °C	[5]
	H ₂ O ₂	0.116	2.42×10 ⁻⁸			
CS-MoSe ₂ NS	TMB	1.317	46.85×10 ⁻⁸	4.0	25 °C	[6]
	H ₂ O ₂	12.89	23.26×10 ⁻⁸			
PBMNPs	TMB	0.307	1.06×10 ⁻⁶	4.6	25 °C	[7]
	H ₂ O ₂	323.6	1.17×10 ⁻⁶			
Au-NPFe ₂ O ₃ NC	TMB	0.0429	5.882×10 ⁻⁸	3.5	25 °C	[8]
	H ₂ O ₂	138.5	4.770×10 ⁻⁸			
Mesoporous γ-Fe ₂ O ₃	TMB	0.0997	5.20×10 ⁻⁷	3.5	25 °C	[9]
	H ₂ O ₂	144.30	1.84×10 ⁻⁸			
Mesoporous α-Fe ₂ O ₃	TMB	0.5304	5.43×10 ⁻⁸	3.5	25 °C	[9]
	H ₂ O ₂	127.92	3.77×10 ⁻⁸			
IONF_250	TMB	0.24	3.07×10 ⁻⁸	3.5	25 °C	[10]
	H ₂ O ₂	150.47	3.07×10 ⁻⁸			
IO-MC	TMB	0.242	1.93×10 ⁻⁸	4.0	25 °C	[11]
	H ₂ O ₂	14.9	1.64×10 ⁻⁸			
Fe ₃ O ₄ @PDA/PB	TMB	0.1182	16.17×10 ⁻⁸	4.0	30 °C	This
	H ₂ O ₂	10.70	16.19×10 ⁻⁸			work

3. Additional References

- 1 L. Gao, J. Zhuang, L. Nie, J. Zhang, Y. Zhang, N. Gu, T. Wang, J. Feng, D. Yang and S. Perrett, *Nat. Nanotechnol.*, 2007, **2**, 577–583.
- 2 A. Swaidan, P. Borthakur, P. K. Boruah, M. R. Das, A. Barras, S. Hamieh, J. Toufaily, T. Hamieh, S. Szunerits and R. Boukherroub, *Sens. Actuators, B*, 2019, **294**, 253–262.
- 3 S. Ghayyem, A. Swaidan, A. Barras, M. Dolci, F. Faridbod, S. Szunerits and R. Boukherroub, *Talanta*, 2021, **226**, 122082.
- 4 X. Guo, F. Yang, L. Jing, J. Li, Y. Li, R. Ding, B. Duan and X. Zhang, *J. Hazard. Mater.*, 2022, **431**, 128621.
- 5 P. Borthakur, P. K. Boruah, M. R. Das, S. B. Artemkina, P. A. Poltarak and V. E. Fedorov, *New J. Chem.*, 2018, **42**, 16919–16929.
- 6 L. Huang, Q. Zhu, J. Zhu, L. Luo, S. Pu, W. Zhang, W. Zhu, J. Sun and J. Wang, *Inorg. Chem.*, 2019, **58**, 1638–1646.
- 7 X. Zhang, S. Gong, Y. Zhang, T. Yang, C. Wang and N. Gu, *J. Mater. Chem.*, 2010, **20**, 5110–5116.
- 8 M. K. Masud, S. Yadav, M. N. Islam, N.-T. Nguyen, C. Salomon, R. Kline, H. R. Alamri, Z. A. Allothman, Y. Yamauchi and M. S. A. Hossain, *Anal. Chem.*, 2017, **89**, 11005–11013.
- 9 M. K. Masud, J. Kim, M. M. Billah, K. Wood, M. J. Shiddiky, N. T. Nguyen, R. K. Parsapur, Y. V. Kaneti, A. A. Alshehri and Y. G. Alghamidi, *J. Mater. Chem. B*, 2019, **7**, 5412–5422.
- 10 S. Tanaka, M. K. Masud, Y. V. Kaneti, M. J. Shiddiky, A. Fatehmulla, A. M. Aldhafiri, W. A. Farooq, Y. Bando, M. S. A. Hossain and Y. Yamauchi, *ChemNanoMat*, 2019, **5**, 506–513.
- 11 M. A. Wahab, S. A. Hossain, M. K. Masud, H. Park, A. Ashok, M. Mustapić, M. Kim, D. Patel, M. Shahbazi and M. S. A. Hossain, *Sens. Actuators, B*, 2022, **366**, 131980.