

SUPPORTING INFORMATION

Visible Light Promoted Synthesis of Ureas and Formamides from Amines and CO₂

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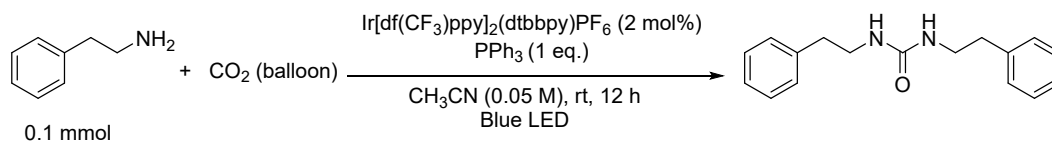
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1. General experimental methods

All reagents were purchased from commercial suppliers without further purification. Experiments were monitored by thin layer chromatography (TLC) and the TLC was performed on precoated silica gel plates. Visualization was accomplished with UV light (254 nm) and exposure to phosphomolybdic acid (PMA) or potassium permanganate solution followed by heating. Photocatalytic synthesis of formamides were carried out using a high pressure photoreactor from WATTCAS. The NMR spectra were recorded at room temperature on a Bruker Avance III 500 MHz. Chemical shifts were calibrated using residual undeuterated solvent as an internal reference (CDCl_3 : 7.26 ppm ^1H NMR, 77.16 ppm ^{13}C NMR and $\text{DMSO}-d_6$: 2.50 ppm ^1H NMR, 39.52 ppm ^{13}C NMR). Multiplicity was indicated as follows: s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet), dd (doublet of doublet), brs (broad singlet). The high resolution mass spectra (HRMS) were measured on AB Sciex Triple TOF 4600 mass spectrometer by ESI. Column chromatography was carried out over 200-300 mesh silica gel.

2. General procedures.

2.1 General procedure A: synthesis of ureas.



A 25-mL Schlenk sealing tube was charged with phenethylamine (0.1 mmol, 12.5 μ L), Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ (2 mol%, 2.3 mg), PPh₃ (1.0 eq., 26.2 mg). The atmosphere was exchanged by applying vacuum and backfilling with CO₂ (this process was conducted a total of three times). Under CO₂ atmosphere, CH₃CN (0.05 M) was added via syringe. Then the reaction was placed under a blue LED (wavelength 440 nm, 40 W) with an CO₂ balloon and irradiated for 12 h at room temperature. After the reaction was completed, 10 mL water was added and extracted with ethyl acetate for 3 times. The organic layer was washed with water, brine, dried with anhydrous Na₂SO₄, and concentrated under reduced pressure. The crude reaction mixture was purified by column chromatography on silica gel (eluent: PE: EA = 1:1) to give the product as a white solid in 81% yield (10.8 mg).

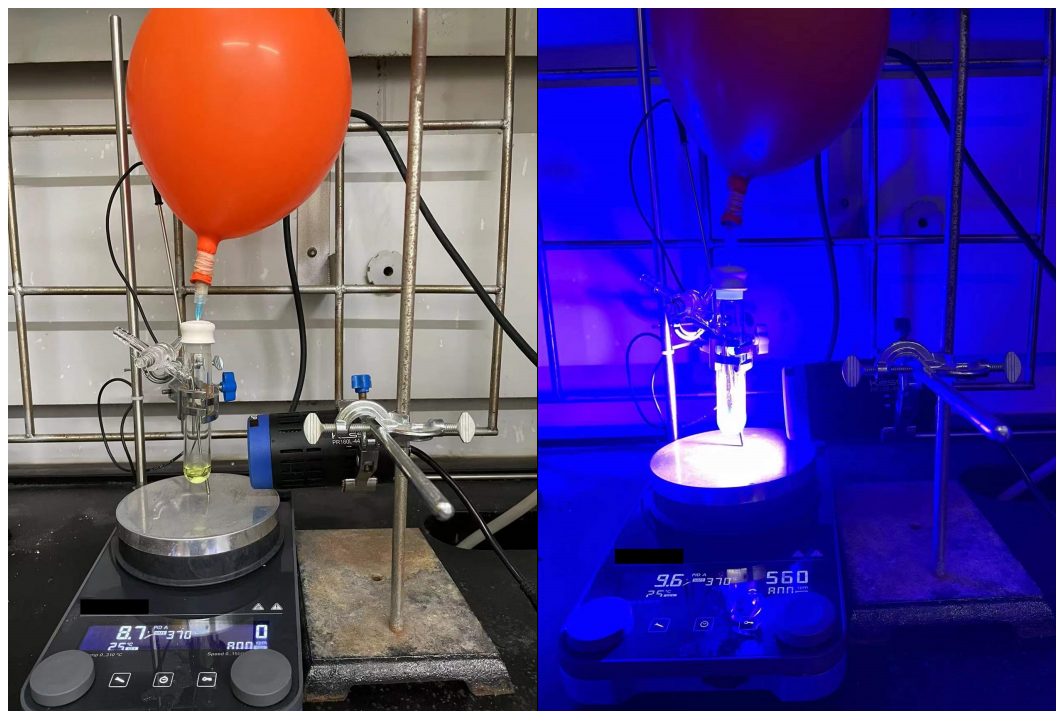
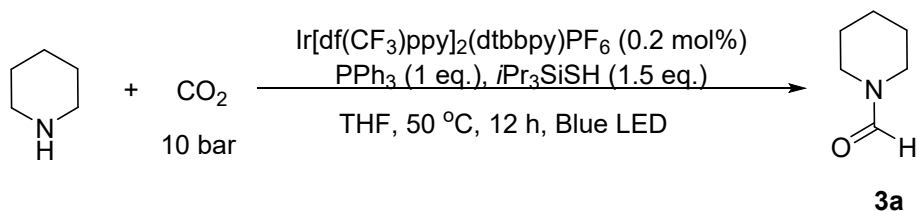


Figure S1. Devices for the photocatalytic reactions for synthesis ureas.

2.2 General procedure B: synthesis of formamides.



A 25 mL vial tube equipped with a magnetic stir was charged with piperidine (0.2 mmol), Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ (0.2 mol%), triisopropylsilanethiol (1.5 eq., 0.3 mmol), triphenylphosphine (1 eq., 0.2 mmol), anhydrous THF (2 mL). After evacuated and backfilled CO₂ three times, it was filled with 10 bar of CO₂. The reaction mixture was then irradiated with a 10 W blue LED lamp with heating from circulating liquid (ethylene glycol: H₂O = 1:1) for 12 h at 50 °C. After the reaction was completed, 10 mL water was added and extracted with ethyl acetate for 3 times. The organic layer was washed with water, brine, dried with anhydrous Na₂SO₄, and concentrated under reduced pressure. The crude reaction mixture was purified by column chromatography on silica gel (eluent: PE: EA = 1:1) to give the product as a yellow oil in 80% yield (18.1 mg).



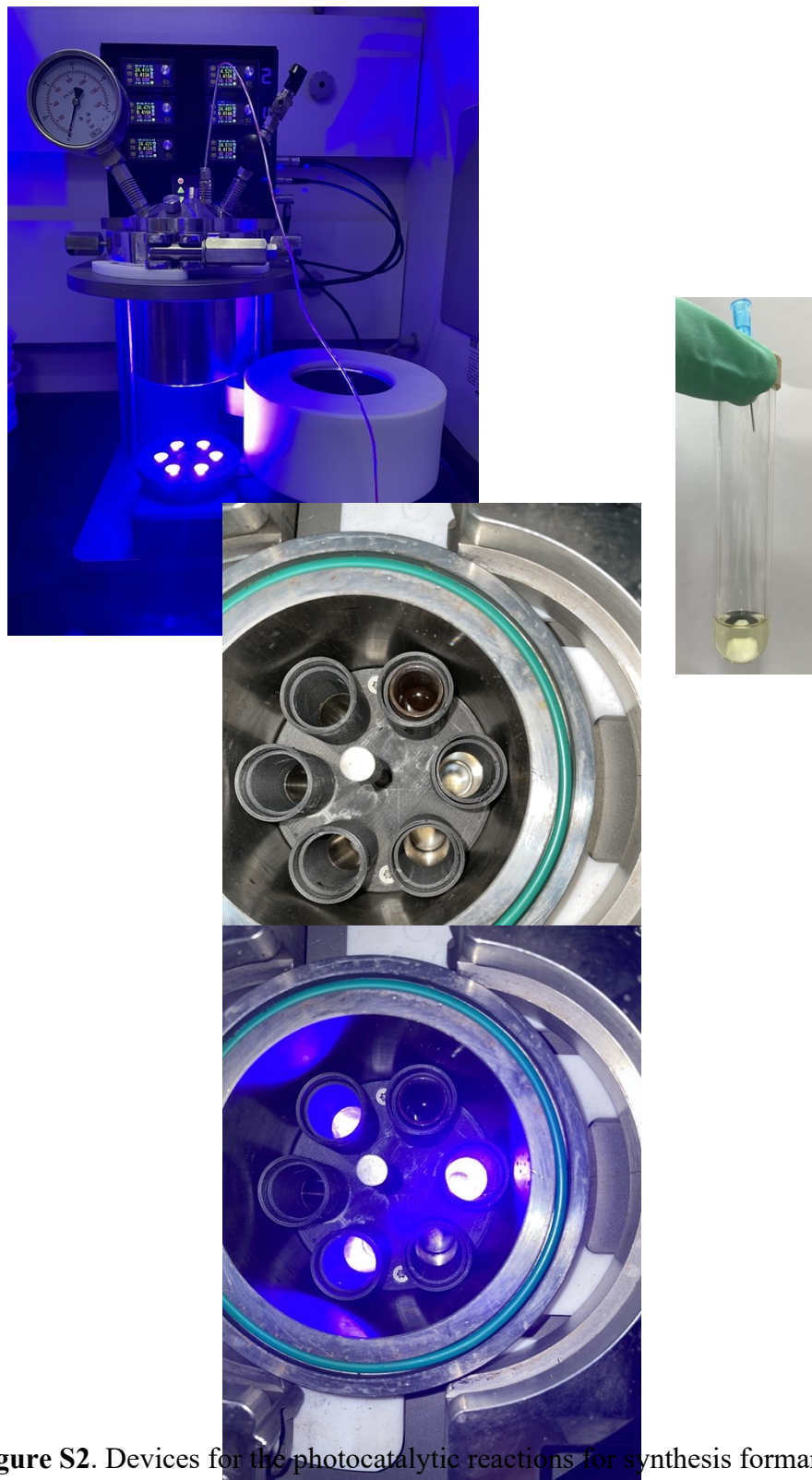
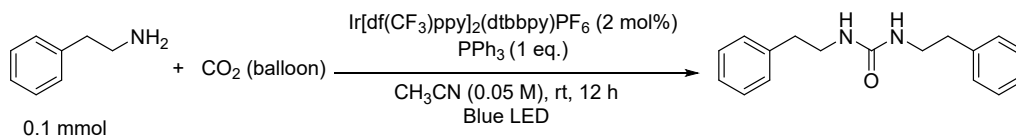


Figure S2. Devices for the photocatalytic reactions for synthesis formamides.

3. Investigation of the key reaction parameters.

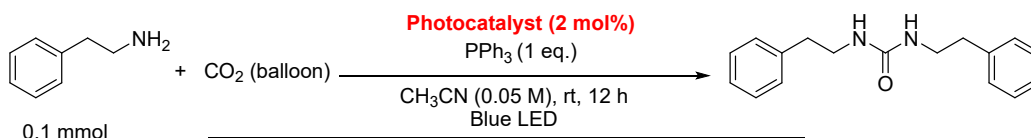
Table S1. Control reactions for synthesis of ureas ^a



entry	deviation	Yield ^b (%)
1	Without Ir[df(CF ₃)ppy] ₂ (dtbbpy)PF ₆	10
2	Without PPh ₃	<5
3	Without light	ND

^aGeneral conditions, unless otherwise noted: amine (0.1 mmol), Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ (2 mol%), PPh₃ (0.1 mmol), CH₃CN (2 mL), blue LED, CO₂ (balloon), 12 h. ^bYields determined by ¹H NMR spectroscopy using trimethoxybenzene as an internal standard. ^cIsolated yield.

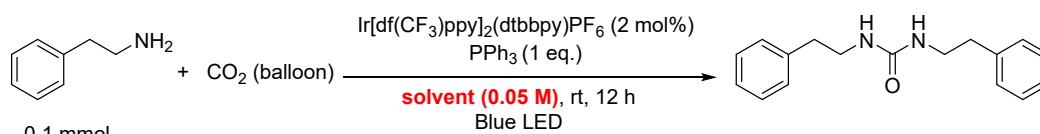
Table S2. Photocatalysts optimization for synthesis of ureas ^a



Entry	Photocatalyst	yield ^b (%)
1	Ir[df(CF₃)ppy]₂(dtbbpy)PF₆	82 (81^c)
2	4CzIPN	30
3	[Ir(ppy) ₂ (dtbbpy)][PF ₆]	42
4	Acridinium dye	33
5	[Ru(bpy) ₃][PF ₆] ₂	9
6	[Ru(bpy) ₃]Cl ₂	8
7	Eosin Y	25
8	Ir[df(CF ₃)ppy] ₂ (dtbbpy)PF ₆ (0.2 mol%)	30

^aGeneral conditions, unless otherwise noted: amine (0.1 mmol), PPh₃ (0.1 mmol), CH₃CN (2 mL), blue LED, CO₂ (balloon), 12 h. ^bYields determined by ¹H NMR spectroscopy using trimethoxybenzene as an internal standard. ^cIsolated yield.

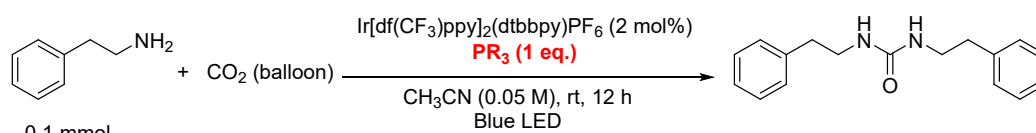
Table S3. Solvents optimization for synthesis of ureas ^a



Entry	Solvent	yield ^b (%)
1	CH₃CN	82 (81^c)
2	toluene	44
3	Acetone	ND
4	CH ₂ Cl ₂	ND
5	THF	41
6	DMSO	ND
7	1,4-dioxane	46
8	CH ₃ CN (0.1 M)	65
9	CH ₃ CN (0.025 M)	57
10	CH ₃ CN (use freeze pump thaw to remove oxygen)	36

^aGeneral conditions, unless otherwise noted: amine (0.1 mmol), Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ (2 mol%), PPh₃ (0.1 mmol), blue LED, CO₂ (balloon), 12 h. ^bYields determined by ¹H NMR spectroscopy using trimethoxybenzene as an internal standard. ^cIsolated yield.

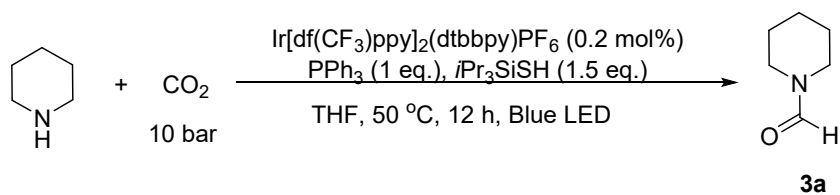
Table S4. Phosphine optimization for synthesis of ureas ^a



Entry	Phosphine	yield ^b (%)
1	PPh₃	82 (81^c)
2	P(Cy) ₃	42
3	Ph ₂ PCH ₂ CH ₂ PPh ₂	72
4	Ph ₂ POEt	69
5	P(OEt) ₃	ND
6	PPh ₃ (0.5 eq.)	45
7	PPh ₃ (1.5 eq.)	82

^aGeneral conditions, unless otherwise noted: amine (0.1 mmol), Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ (2 mol%), PR₃ (1 eq.), CH₃CN (2 mL), blue LED, CO₂ (balloon), 12 h. ^bYields determined by ¹H NMR spectroscopy using trimethoxybenzene as an internal standard. ^cIsolated yield.

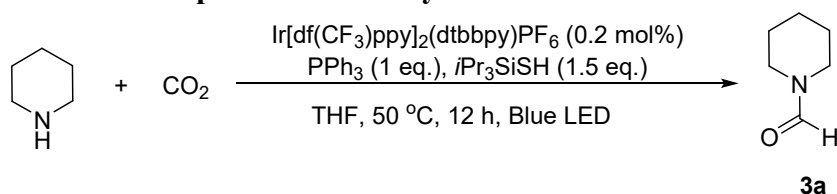
Table S5. Control reactions for synthesis of formamides ^a



entry	deviation	Yield ^b (%)
1	Without Ir[df(CF ₃)ppy] ₂ (dtbbpy)PF ₆	trace
2	Without PPh ₃	trace
3	Without <i>i</i> Pr ₃ SiSH	trace
4	Without light	trace

^aGeneral conditions, unless otherwise noted: amine (0.2 mmol), triisopropylsilanethiol (0.3 mmol), PPh₃ (0.2 mmol), Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ (0.2 mol%), THF (2 mL), 50 °C, blue LED, CO₂ (10 bar), 12 h. ^bYields determined by ¹H NMR spectroscopy using trimethoxybenzene as an internal standard.

Table S6. Pressure optimization for synthesis of formamides ^a



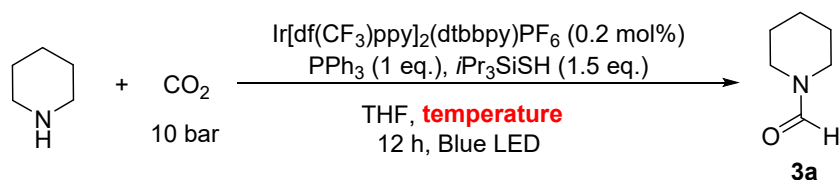
Entry	pressure	yield ^b (%)
1	CO ₂ (balloon)	44
2	10 bar	84 (80^c)

^aGeneral conditions, unless otherwise noted: amine (0.2 mmol), triisopropylsilanethiol (0.3 mmol), PPh₃ (0.2 mmol), Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ (0.2 mol%), THF (2 mL), 50 °C, blue LED, 12 h.

^bYields determined by ¹H NMR spectroscopy using trimethoxybenzene as an internal standard.

^cIsolated yield.

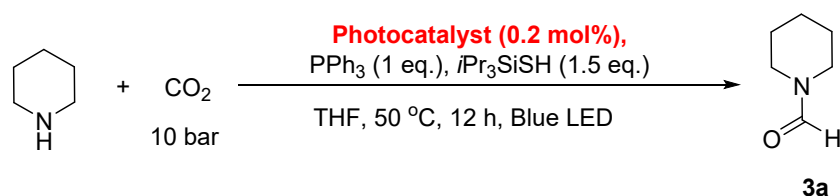
Table S7. Temperature optimization for synthesis of formamides ^a



Entry	temperature (°C)	yield ^b (%)
1	30	83
2	50	84 (80^c)
3	70	63

^aGeneral conditions, unless otherwise noted: amine (0.2 mmol), triisopropylsilanethiol (0.3 mmol), PPh₃ (0.2 mmol), Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ (0.2 mol%), THF (2 mL), blue LED, CO₂ (10 bar), 12 h. ^bYields determined by ¹H NMR spectroscopy using trimethoxybenzene as an internal standard. ^cIsolated yield.

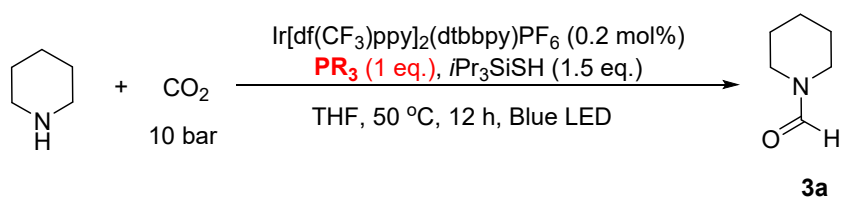
Table S8. Photocatalysts optimization for synthesis of formamides ^a



Entry	Photocatalyst	yield ^b (%)
1	Ir[df(CF₃)ppy]₂(dtbbpy)PF₆	84 (80^c)
2	4CzIPN	trace
3	[Ir(ppy) ₂ (dtbbpy)][PF ₆]	84
4	Acridinium dye	trace
5	[Ru(bpy) ₃][PF ₆] ₂	32
6	[Ru(bpy) ₃]Cl ₂	trace
7	Eosin Y	trace
8	Ir[df(CF ₃)ppy] ₂ (dtbbpy)PF ₆ (2 mol%)	64

^aGeneral conditions, unless otherwise noted: amine (0.2 mmol), triisopropylsilanethiol (0.3 mmol), PPh₃ (0.2 mmol), THF (2 mL), 50 °C, blue LED, CO₂ (10 bar), 12 h. ^bYields determined by ¹H NMR spectroscopy using trimethoxybenzene as an internal standard. ^cIsolated yield.

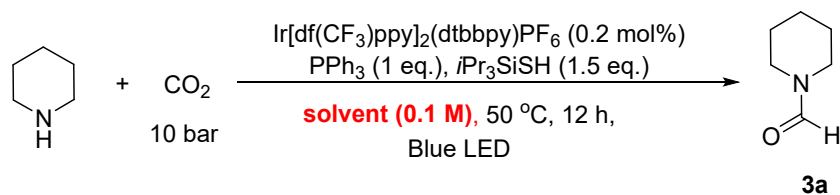
Table S9. Phosphine optimization for synthesis of formamides ^a



Entry	PR ₃	yield ^b (%)
1	Ph₃P	84 (80^c)
2	PhP(OEt) ₂	62
3	Ph ₂ POEt	84
4	Cy ₃ P	48
5	Ph ₂ PCH ₂ CH ₂ PPh ₂	75
6	Ph ₃ P (1.5 eq)	84
7	Ph ₃ P (0.5 eq)	45

^aGeneral conditions, unless otherwise noted: amine (0.2 mmol), triisopropylsilanethiol (0.3 mmol), Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ (0.2 mol%), THF (2 mL), 50 °C, blue LED, CO₂ (10 bar), 12 h. ^bYields determined by ¹H NMR spectroscopy using trimethoxybenzene as an internal standard. ^cIsolated yield.

Table S10. Solvent optimization for synthesis of formamides ^a



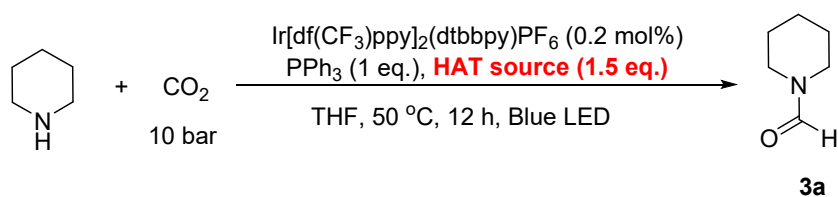
Entry	solvent	yield ^b (%)
1	THF	84 (80^c)
2	CH ₃ CN	64
3	DMSO	trace
5	1,4-dioxane	trace
6	DCE	58
7	EA	42
8	THF (0.2 M)	81
9	THF (use freeze pump thaw to remove oxygen)	80 ^c

^aGeneral conditions, unless otherwise noted: amine (0.2 mmol), triisopropylsilanethiol (0.3 mmol), PPh₃ (0.2 mmol), Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ (0.2 mol%), 50 °C, blue LED, CO₂ (10 bar), 12 h.

^bYields determined by ¹H NMR spectroscopy using trimethoxybenzene as an internal standard.

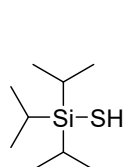
^cIsolated yield.

Table S11. HAT source optimization for synthesis of formamides ^a

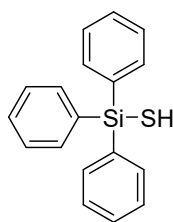


Entry	HAT source	yield ^b (%)
1	A	84 (80^c)
2	B	18
3	C	ND
4	D	ND
5	E	ND
6	F	ND
7	G	ND
8	H	ND
9	1,4-Cyclohexadiene	ND
10	Hantzsch ester	ND
11	(<i>p</i> -OMeC ₆ H ₄) ₂ S ₂	ND
12	A (1 eq)	77
13	A (0.8 eq)	67
14	A (0.5 eq)	42
15	A (0.3 eq)	30

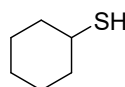
^aGeneral conditions, unless otherwise noted: amine (0.2 mmol), PPh₃ (0.2 mmol), Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ (0.2 mol%), THF (2 mL), 50 °C, blue LED, CO₂ (10 bar), 12 h. ^bYields determined by ¹H NMR spectroscopy using trimethoxybenzene as an internal standard. ^cIsolated yield.



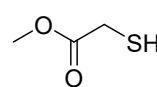
A



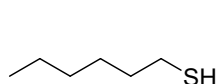
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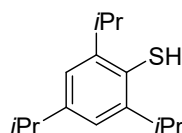
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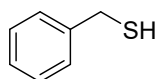
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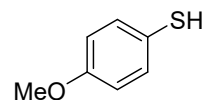
E



F



G



H

4. Mechanistic studies

4.1 Radical trapping experiment



A 25 mL vial tube equipped with a magnetic stir was charged with piperidine (**1a**) (0.2 mmol), Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ (0.2 mol%), triisopropylsilanethiol (1.5 eq., 0.3 mmol), triphenylphosphine (1 eq., 0.2 mmol), TEMPO (3 eq., 0.6 mmol), anhydrous THF (2 mL). After evacuated and backfilled CO₂ three times, it was filled with 10 bar of CO₂. The reaction mixture was then irradiated with a 10 W blue LED lamp with heating from circulating liquid (ethylene glycol: H₂O = 1:1) for 12 h at 50 °C. After the reaction was completed, the mixture was detected by LC-MS, the formation of **3a** was completely inhibited and 2,2,6,6-tetramethylpiperidin-1-yl piperidine-1-carboxylate (**4**) was detected by LC-MS.

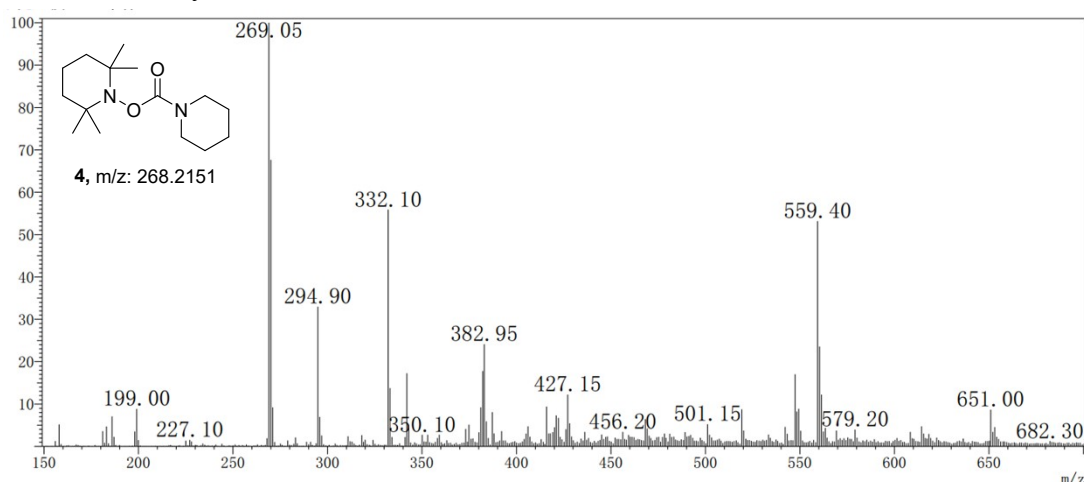


Figure S3. LC-MS spectrum of 2,2,6,6-tetramethylpiperidin-1-yl piperidine-1-carboxylate (**4**).

4.2 Stern–Volmer quenching studies

A solution of Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ in anhydrous CH₃CN (0.01 mM) was added with an appropriate amount of PPh₃ in a quartz cuvette. Then the emission of the sample was collected. The emission intensity at 380 nm was collected with excited wavelength of 474 nm.

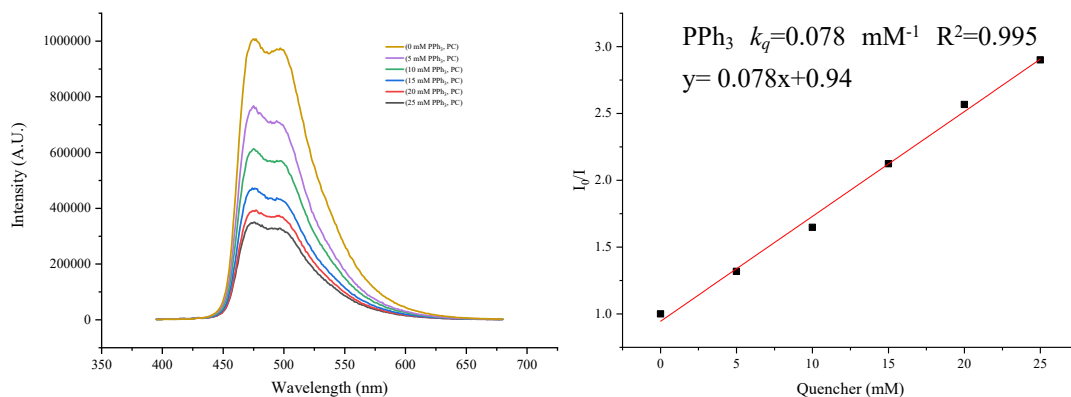


Figure S4. Stern-Volmer quenching by PPh₃.

A solution of Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ in anhydrous CH₃CN (0.01 mM) was added with an appropriate amount of phenethylamine in a quartz cuvette. Then the emission of the sample was collected. The emission intensity at 380 nm was collected with excited wavelength of 474 nm.

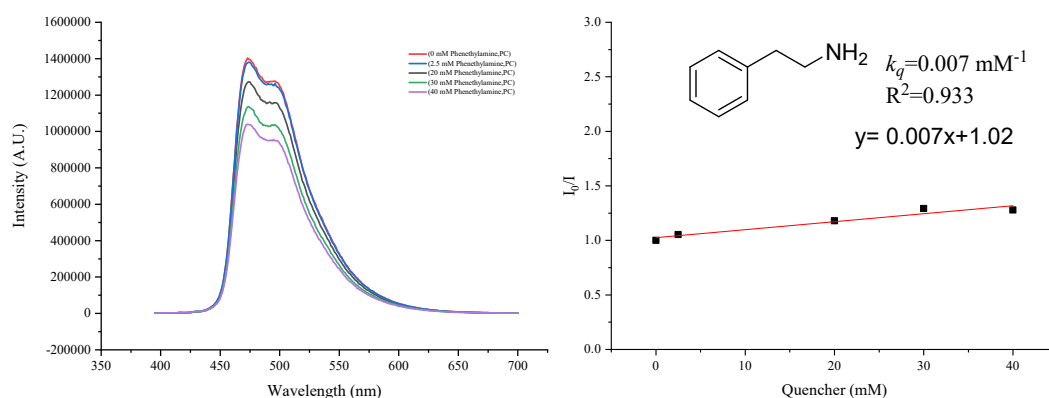


Figure S5 Stern-Volmer quenching by phenethylamine

A solution of Ir[df(CF₃)ppy]₂(dtbbpy)PF₆ in anhydrous THF (0.01 mM) was added with an appropriate amount of piperidine in a quartz cuvette. Then the emission of the sample was collected. The emission intensity at 378 nm was collected with excited wavelength of 479 nm.

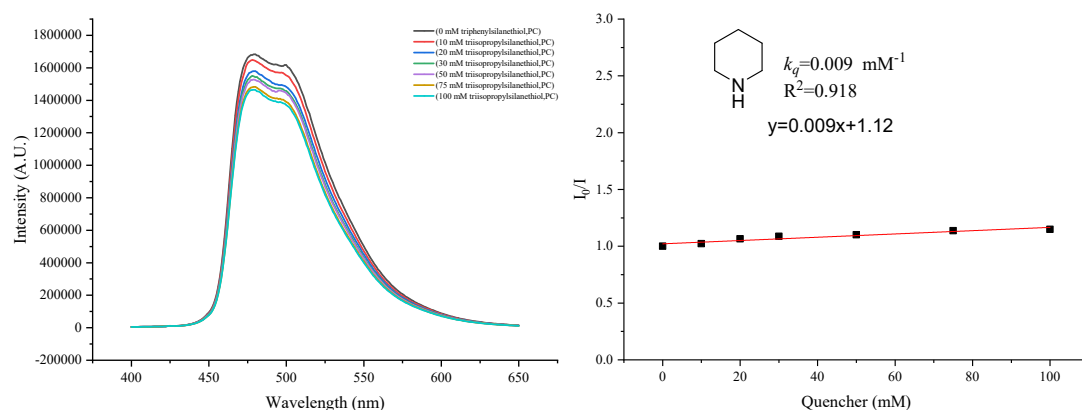


Figure S6 Stern-Volmer quenching by piperidine.

4.3 Proposed mechanism for synthesis of ureas via energy transfer

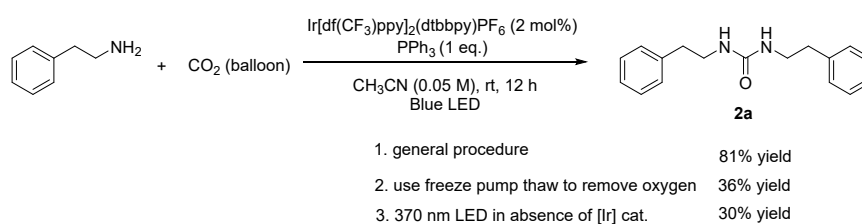


Figure S7 Control experiments

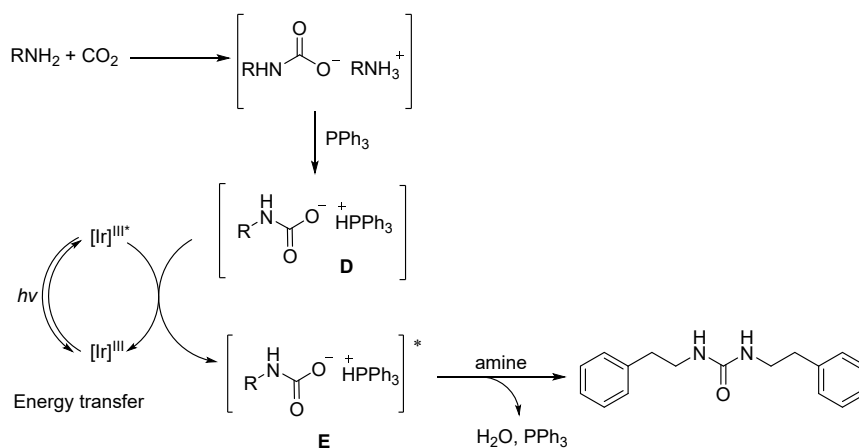
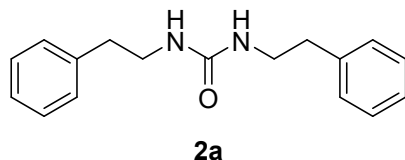


Figure S8 Proposed mechanism for synthesis of ureas via energy transfer

We found that **2a** could also be obtained under this condition indicating that another way may exist for the generation of ureas. When we used 370 nm LED instead of 440 nm LED, **2a** could be obtained in 30% yield in absence of photocatalyst. According to these results, we think that the generation of **2a** may also be promoted by energy transfer process (Figure S8). Since PPh_3 was essential for the reaction, we proposed that excited carbamate anion (**E**) may react with amine to produce the corresponding urea.

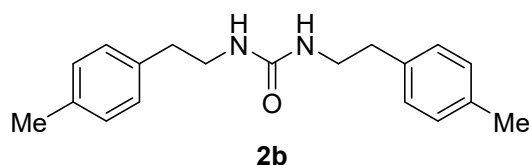
5. Characterization data

1,3-diphenethylurea (2a)¹



The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (10.9 mg, 81%); ¹H NMR (500 MHz, CDCl₃) δ 7.29 (t, *J* = 7.5 Hz, 4H), 7.21 (t, *J* = 7.4 Hz, 2H), 7.18 – 7.14 (m, 4H), 4.51 (t, *J* = 5.9 Hz, 2H), 3.37 (q, *J* = 6.6 Hz, 4H), 2.76 (t, *J* = 6.9 Hz, 4H).

1,3-bis(4-methylphenethyl)urea (2b)



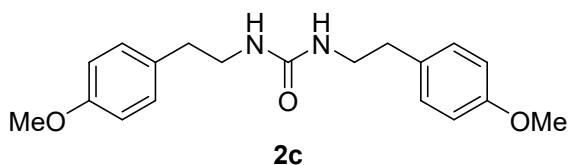
The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (11.4 mg, 77%); Mp: 183 - 185 °C

¹H NMR (500 MHz, CDCl₃) δ 7.10 (d, *J* = 7.8 Hz, 4H), 7.05 (d, *J* = 7.8 Hz, 4H), 4.44 (t, *J* = 5.9 Hz, 2H), 3.36 (q, *J* = 6.6 Hz, 4H), 2.72 (t, *J* = 6.9 Hz, 4H), 2.32 (s, 6H).

¹³C NMR (126 MHz, CDCl₃) δ 158.2, 136.1, 136.0, 129.4, 128.8, 41.9, 36.0, 21.1.

HRMS : *m/z* calcd. for C₁₉H₂₅N₂O [M+H]⁺: 297.1961; found: 297.1957.

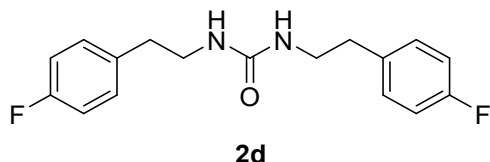
1,3-bis(4-methoxyphenethyl)urea (2c)²



The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (13.1 mg, 80%);

¹H NMR (500 MHz, CDCl₃) δ 7.07 (d, *J* = 8.6 Hz, 4H), 6.82 (d, *J* = 8.6 Hz, 4H), 4.44 (t, *J* = 5.9 Hz, 2H), 3.77 (s, 6H), 3.34 (q, *J* = 6.7 Hz, 4H), 2.70 (t, *J* = 6.9 Hz, 4H).

1,3-bis(4-fluorophenethyl)urea (2d)



The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (7.3 mg, 48%);

Mp: 127 - 129 °C

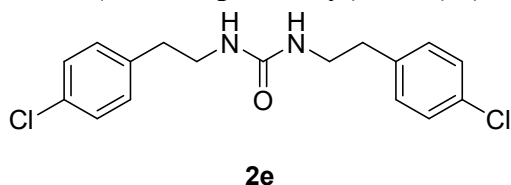
¹H NMR (500 MHz, CDCl₃) δ 7.12 (dd, *J* = 8.4, 5.5 Hz, 4H), 6.97 (t, *J* = 8.7 Hz, 4H), 4.30 (t, *J* = 6.0 Hz, 2H), 3.37 (q, *J* = 6.6 Hz, 4H), 2.75 (t, *J* = 6.9 Hz, 4H).

¹³C NMR (126 MHz, CDCl₃) δ 161.7 (d, *J* = 244.7 Hz), 158.0, 134.8 (d, *J* = 3.2 Hz), 130.3 (d, *J* = 7.9 Hz), 115.5 (d, *J* = 21.2 Hz), 41.8, 35.7.

¹⁹F NMR (470 MHz, CDCl₃) δ -116.64

HRMS: *m/z* calcd. for C₁₇H₁₉F₂N₂O [M+H]⁺: 305.1460; found: 305.1455.

1,3-bis(4-chlorophenethyl)urea (2e)



The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (8.9 mg, 53%);

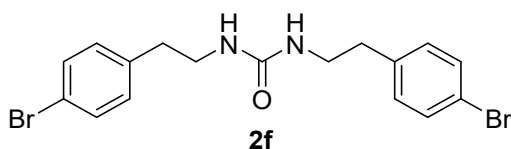
Mp: 141 - 143 °C

¹H NMR (500 MHz, CDCl₃) δ 7.25 (d, *J* = 8.7 Hz, 4H), 7.09 (d, *J* = 8.4 Hz, 4H), 4.32 (t, *J* = 5.9 Hz, 2H), 3.37 (q, *J* = 6.6 Hz, 4H), 2.74 (t, *J* = 6.9 Hz, 4H).

¹³C NMR (126 MHz, CDCl₃) δ 157.8, 137.6, 132.3, 130.2, 128.7, 41.5, 35.8.

HRMS: *m/z* calcd. for C₁₇H₁₉Cl₂N₂O [M+H]⁺: 337.0869; found: 337.0861.

1,3-bis(4-bromophenethyl)urea (2f)³

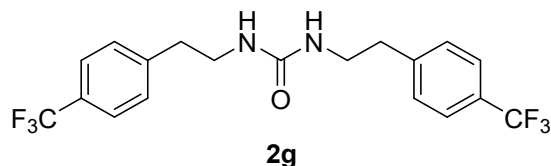


The title compound was prepared according to the General procedure A and yields determined by ¹H NMR spectroscopy using trimethoxybenzene as an internal standard (71%);

¹H NMR (500 MHz, DMSO-*d*₆) δ 7.46 (d, *J* = 8.3 Hz, 4H), 7.14 (d, *J* = 8.4 Hz, 4H), 5.83 (t, *J* = 5.8 Hz, 2H), 3.19 (q, *J* = 6.8 Hz, 5H), 2.63 (t, *J* = 7.0 Hz, 5H).

HRMS: *m/z* calcd. for C₁₇H₁₉Br₂O [M+H]⁺: 424.9859; found: 424.9859.

1,3-bis(4-(trifluoromethyl)phenethyl)urea (2g)



The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (12.5 mg, 62%); Mp: 124 - 126 °C

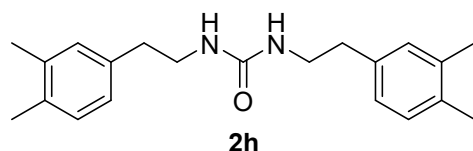
^1H NMR (500 MHz, $\text{DMSO}-d_6$) δ 7.63 (d, J = 8.0 Hz, 4H), 7.40 (d, J = 7.9 Hz, 4H), 5.87 (t, J = 5.8 Hz, 2H), 3.25 (q, J = 6.7 Hz, 4H), 2.76 (t, J = 7.0 Hz, 4H).

^{13}C NMR (126 MHz, $\text{DMSO}-d_6$) δ 157.8, 144.8, 129.5, 126.9, 125.0 (q, J = 3.9, 3.5 Hz), 123.4, 40.3, 35.8.

^{19}F NMR (470 MHz, $\text{DMSO}-d_6$) δ -60.8.

HRMS: m/z calcd. for $\text{C}_{19}\text{H}_{19}\text{F}_6\text{N}_2\text{O}$ $[\text{M}+\text{H}]^+$: 405.1396; found: 405.1389.

1,3-bis(3,4-dimethylphenethyl)urea (2h)



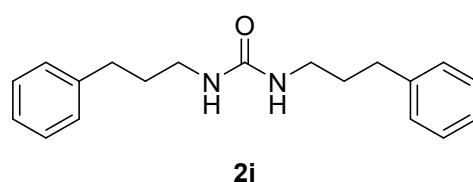
The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (12.3 mg, 76%); Mp: 149 - 151 °C

^1H NMR (500 MHz, CDCl_3) δ 7.05 (d, J = 7.6 Hz, 2H), 6.95 (s, 2H), 6.90 (d, J = 7.5 Hz, 2H), 4.24 (t, J = 5.8 Hz, 2H), 3.37 (q, J = 6.6 Hz, 4H), 2.71 (t, J = 6.9 Hz, 4H), 2.23 (s, 12H).

^{13}C NMR (126 MHz, CDCl_3) δ 158.1, 136.9, 136.6, 134.7, 130.3, 129.9, 126.3, 41.9, 36.0, 19.9, 19.4.

HRMS: m/z calcd. for $\text{C}_{21}\text{H}_{29}\text{N}_2\text{O}$ $[\text{M}+\text{H}]^+$: 325.2274; found: 325.2267.

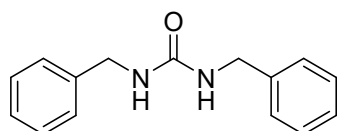
1,3-bis(3-phenylpropyl)urea (2i) ²



The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (11.4 mg, 77%);

^1H NMR (500 MHz, CDCl_3) δ 7.31 – 7.24 (m, 4H), 7.17 (dd, J = 7.3, 5.7 Hz, 6H), 4.42 (s, 2H), 3.15 (q, J = 6.4 Hz, 4H), 2.63 (t, J = 7.7 Hz, 4H), 1.80 (m, 4H).

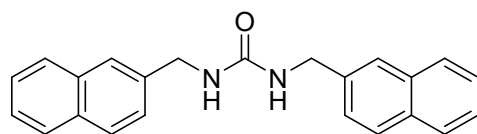
1,3-dibenzylurea (2j) ⁴



2j

The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (9.0 mg, 75%); ^1H NMR (500 MHz, CDCl_3) δ 7.32 – 7.27 (m, 4H), 7.25 – 7.21 (m, 6H), 4.88 (brs, 2H), 4.32 (d, J = 5.7 Hz, 4H).

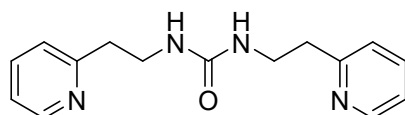
1,3-bis(naphthalen-2-ylmethyl)urea (2k) ⁵



2k

The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (4.4 mg, 26%); ^1H NMR (500 MHz, $\text{DMSO}-d_6$) δ 7.88 (t, J = 7.8 Hz, 6H), 7.81 (d, J = 7.6 Hz, 2H), 7.73 (s, 2H), 7.52 – 7.41 (m, 6H), 6.63 (t, J = 6.1 Hz, 2H), 4.42 (d, J = 6.0 Hz, 4H).

1,3-bis(2-(pyridin-2-yl)ethyl)urea (2l)



2l

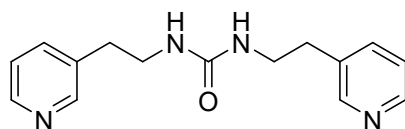
The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (8.5 mg, 63%); Mp: 265 - 267 °C

^1H NMR (500 MHz, CDCl_3) δ 8.44 (d, J = 4.9 Hz, 2H), 7.64 – 7.50 (m, 2H), 7.15 (d, J = 7.8 Hz, 2H), 7.13 – 7.06 (m, 2H), 5.25 (s, 2H), 3.55 (q, J = 6.2 Hz, 4H), 2.95 (t, J = 6.5 Hz, 4H).

^{13}C NMR (126 MHz, CDCl_3) δ 159.9, 158.4, 149.2, 136.7, 123.7, 121.6, 39.9, 38.1.

HRMS: m/z calcd. For $\text{C}_{15}\text{H}_{19}\text{N}_4\text{O}$ $[\text{M}+\text{H}]^+$: 271.1553; found: 271.1556.

1,3-bis(2-(pyridin-3-yl)ethyl)urea (2m)



2m

The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (8.9 mg, 66%);

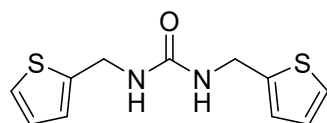
Mp: 261-263 °C

^1H NMR (500 MHz, CDCl_3) δ 8.35 (d, $J = 4.4$ Hz, 2H), 8.34 – 8.30 (m, 2H), 7.50 (dd, $J = 7.9, 2.1$ Hz, 2H), 7.19 (dd, $J = 7.8, 4.8$ Hz, 2H), 4.91 (t, $J = 5.9$ Hz, 2H), 3.43 (q, $J = 6.4$ Hz, 4H), 2.78 (t, $J = 6.8$ Hz, 4H).

^{13}C NMR (126 MHz, CDCl_3) δ 158.2, 150.1, 147.8, 136.6, 135.0, 123.7, 41.2, 33.8.

HRMS: m/z calcd. For $\text{C}_{15}\text{H}_{19}\text{N}_4\text{O}$ $[\text{M}+\text{H}]^+$: 271.1553; found:271.1560.

1,3-bis(thiophen-2-ylmethyl)urea (2n) ⁴

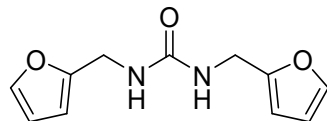


2n

The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a yellow solid (11.3 mg, 90%);

^1H NMR (500 MHz, $\text{DMSO}-d_6$) δ 7.30 (dd, $J = 4.5, 1.8$ Hz, 2H), 6.88 (d, $J = 4.6$ Hz, 4H), 6.47 (t, $J = 6.1$ Hz, 2H), 4.33 (d, $J = 6.0$ Hz, 4H).

1,3-bis(furan-2-ylmethyl)urea (2o) ⁵

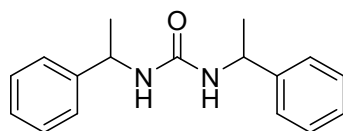


2o

The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a yellow solid (9.8 mg, 89%);

^1H NMR (500 MHz, $\text{DMSO}-d_6$) δ 7.55 (s, 2H), 6.39 – 6.31 (m, 4H), 6.18 (d, $J = 3.2$ Hz, 2H), 4.20 (d, $J = 5.8$ Hz, 4H).

1,3-bis(1-phenylethyl)urea(2p) ⁴



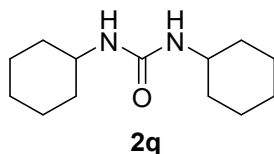
2p

The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (5.5 mg, 41%, rotamers ratio 52:48);;

The presence of two rotamers (ratio 52:48) were observed in the NMR spectra.

^1H NMR (500 MHz, CDCl_3) δ 7.35 – 7.20 (m, 16.57H), 7.15 – 7.09 (m, 4.15H), 4.88 – 4.67 (m, 8H), 1.37 (d, J = 6.7 Hz, 6H), 1.33 (d, J = 6.45 Hz, 7H).

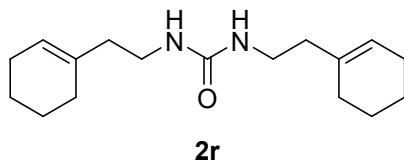
1,3-dicyclohexylurea (2q)¹



The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (9.0 mg, 80%);

^1H NMR (500 MHz, CDCl_3) δ 3.47 – 3.39 (m, 2H), 1.95 – 1.88 (m, 4H), 1.70 (m, 4H), 1.80 – 1.50 (m, 2H), 1.41 – 1.29 (m, 4H), 1.19 – 1.05 (m, 6H).

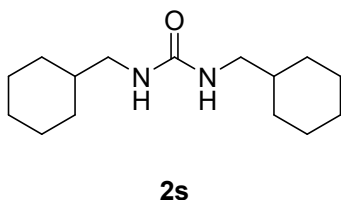
1,3-bis(2-(cyclohex-1-en-1-yl)ethyl)urea (2r)²



The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a yellow solid (8.8 mg, 64%);

^1H NMR (500 MHz, CDCl_3) δ 5.46 (s, 2H), 4.33 (t, J = 5.4 Hz, 2H), 3.21 (td, J = 6.8, 5.2 Hz, 4H), 2.12 (t, J = 6.8 Hz, 4H), 2.01 – 1.96 (m, 4H), 1.92 – 1.88 (m, 4H), 1.63 – 1.58 (m, 4H), 1.57 – 1.52 (m, 4H).

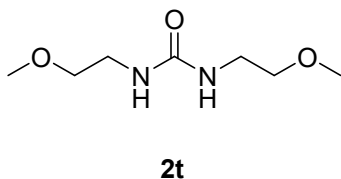
1,3-bis(cyclohexylmethyl)urea (2s)⁵



The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (8.1 mg, 64%);

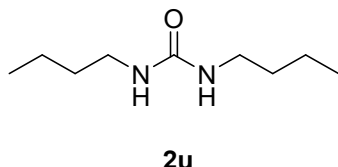
^1H NMR (500 MHz, CDCl_3) δ 4.61 (s, 2H), 2.98 (t, J = 6.3 Hz, 4H), 1.77 – 1.56 (m, 10H), 1.43 – 1.39 (m, 2H), 1.31 – 1.06 (m, 6H), 0.94 – 0.82 (m, 4H).

1,3-bis(2-methoxyethyl)urea (2t)⁶



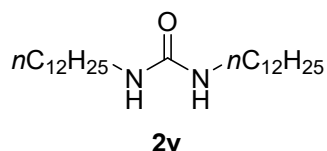
The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (6.4 mg, 73%); ^1H NMR (500 MHz, CDCl_3) δ 4.87 (s, 2H), 3.46 (t, J = 4.98 Hz, 4H), 3.38 – 3.35 (m, 10H).

1,3-dibutylurea (2u)⁵



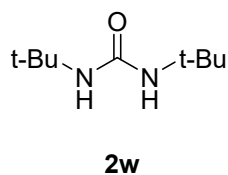
The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (5.9 mg, 69%); ^1H NMR (500 MHz, CDCl_3) δ 4.62 (s, 2H), 3.16 – 3.12 (m, 4H), 1.49 – 1.43 (m, 4H), 1.37 – 1.31 (m, 4H), 0.91 (t, J = 7.3 Hz, 6H).

1,3-didodecylurea (2v)²



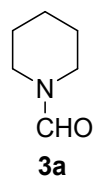
The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (5.3 mg, 28%); ^1H NMR (500 MHz, CDCl_3) δ 4.17 (brs, 2H), 3.14 (q, J = 6.6 Hz, 4H), 1.48 (t, J = 7.1 Hz, 4H), 1.25 (m, 36H), 0.88 (t, J = 6.8 Hz, 6H).

1,3-di-tert-butylurea (2w)¹



The title compound was prepared according to the General procedure A and purified by flash column chromatography to give the product as a white solid (1.3 mg, 15%); ^1H NMR (500 MHz, CDCl_3) δ 3.29 (s, 2H), 1.38 (s, 18H).

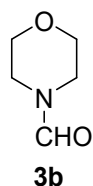
piperidine-1-carbaldehyde (3a)⁷



The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a yellow oil (18.1 mg, 80%);

^1H NMR (500 MHz, CDCl_3) δ 7.99 (s, 1H), 3.46 (t, J = 5.71 Hz, 2H), 3.29 (t, J = 5.60 Hz, 2H), 1.66 (q, J = 6.19 Hz, 2H), 1.54 (m, 4H).

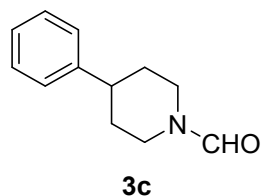
morpholine-4-carbaldehyde (3b)⁷



The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a yellow oil (12.7 mg, 55%);

^1H NMR (500 MHz, CDCl_3) δ 8.06 (s, 1H), 3.70 (t, J = 4.84 Hz, 2H), 3.66 (t, J = 4.92 Hz, 2H), 3.57 (t, J = 4.88 Hz, 2H), 3.39 (t, J = 4.83 Hz, 2H).

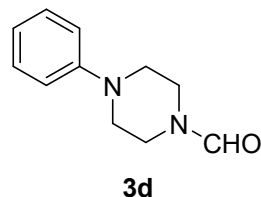
4-phenylpiperidine-1-carbaldehyde (3c)⁸



The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a yellow solid (30.3 mg, 80%);

^1H NMR (500 MHz, CDCl_3) δ 8.07 (s, 1H), 7.32 (t, J = 7.52 Hz, 2H), 7.21 (dd, J = 21.94, 7.42 Hz, 3H), 4.57 (d, J = 13.29 Hz, 0H), 3.73 (d, J = 13.26 Hz, 0H), 3.20 (t, J = 11.46 Hz, 1H), 2.85 – 2.66 (m, 2H), 2.01 – 1.86 (m, 1H), 1.69 – 1.51 (m, 2H).

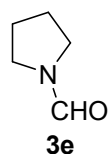
4-phenylpiperazine-1-carbaldehyde (3d)⁸



The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a yellow solid (22.1 mg, 58%);

^1H NMR (500 MHz, CDCl_3) δ 8.11 (s, 1H), 7.29 (m, 2H), 6.96 – 6.92 (m, 3H), 3.73 – 3.70 (m, 2H), 3.60 – 3.50 (m, 2H), 3.23 – 3.17 (m, 2H), 3.17 – 3.13 (m, 2H).

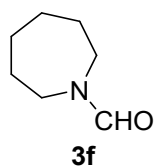
pyrrolidine-1-carbaldehyde (3e)⁹



The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a colorless oil (15.9 mg, 80%);

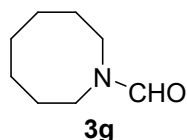
^1H NMR (500 MHz, CDCl_3) δ 8.25 (s, 1H), 3.49 (t, $J = 6.30$ Hz, 2H), 3.42 (t, $J = 6.70$ Hz, 2H), 1.91 (p, $J = 7.19$ Hz, 4H).

azepane-1-carbaldehyde (3f)⁹



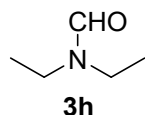
The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a colorless oil (13.7 mg, 54%); ^1H NMR (500 MHz, CDCl_3) δ 8.08 (s, 1H), 3.47 – 3.42 (m, 2H), 3.41 – 3.35 (m, 2H), 1.73 (t, $J = 6.24$ Hz, 4H), 1.61 – 1.51 (m, 4H).

azocane-1-carbaldehyde (3g)¹⁰



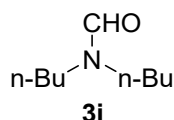
The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a colorless oil (7.6 mg, 27%); ^1H NMR (500 MHz, CDCl_3) δ 8.09 (s, 1H), 3.41 (t, $J = 6.18$ Hz, 2H), 3.32 (t, $J = 5.94$ Hz, 2H), 1.80 – 1.67 (m, 4H), 1.60 – 1.56 (m, 2H), 1.55 – 1.49 (m, 4H).

***N,N*-diethylformamide (3h)⁹**



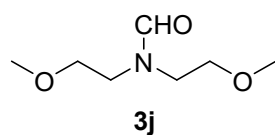
The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a yellow oil (10.1 mg, 50%); ^1H NMR (500 MHz, CDCl_3) δ 8.03 (s, 1H), 3.35 (q, $J = 7.21$ Hz, 2H), 3.26 (q, $J = 7.18$ Hz, 2H), 1.18 (t, $J = 7.19$ Hz, 3H), 1.12 (t, $J = 7.20$ Hz, 3H).

***N,N*-dibutylformamide (3i)⁹**



The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a brown oil (21.4 mg, 68%); ^1H NMR (500 MHz, CDCl_3) δ 8.02 (s, 1H), 3.30 – 3.24 (m, 2H), 3.18 (t, $J = 7.17$ Hz, 2H), 1.54 – 1.47 (m, 4H), 1.35 – 1.25 (m, 4H), 0.94 – 0.91 (m, 6H).

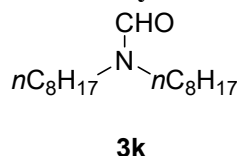
***N,N*-bis(2-methoxyethyl)formamide (3j)⁹**



The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a brown oil (12.2 mg, 38%);

¹H NMR (500 MHz, CDCl₃) δ 8.06 (s, 1H), 3.54 – 3.50 (m, 4H), 3.46 – 3.45 (m, 4H), 3.32 (d, *J* = 2.73 Hz, 6H).

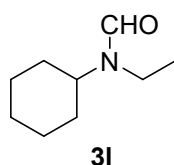
***N,N*-dioctylformamide (3k)⁹**



The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a brown oil (32.3 mg, 60%);

¹H NMR (500 MHz, CDCl₃) δ 8.03 (s, 1H), 3.26 (t, *J* = 7.65 Hz, 2H), 3.17 (t, *J* = 7.16 Hz, 2H), 1.51 (q, *J* = 6.72 Hz, 4H), 1.27 (m, 20H), 0.87 (td, *J* = 6.82, 2.85 Hz, 6H).

***N*-cyclohexyl-*N*-ethylformamide (3l)⁹**

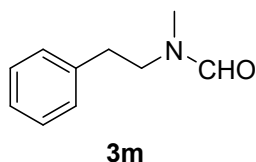


The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a yellow oil (9.3 mg, 30%);

The presence of two rotamers (ratio 73:27) were observed in the NMR spectra.

¹H NMR (500 MHz, CDCl₃) δ 8.12 (s, 0.73H), 8.08 (s, 0.27H), 3.29 (q, *J* = 7.15 Hz, 2H), 3.22 (m, 1H), 1.90 – 1.63 (m, 6H), 1.50 (dd, *J* = 12.49, 3.37 Hz, 2H), 1.29 – 1.18 (m, 2H), 1.15 (t, *J* = 7.13 Hz, 3H).

***N*-methyl-*N*-phenethylformamide (3m)¹¹**



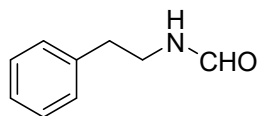
The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a yellow oil (27.1 mg, 83%);

The presence of two rotamers (ratio 63:37) were observed in the NMR spectra.

¹H NMR (500 MHz, CDCl₃) δ 8.04 (s, 0.37H), 7.83 (s, 0.63H), 7.37 – 7.31 (m, 2H), 7.29 – 7.24 (m, 2H), 7.21 – 7.15 (m, 1H), 3.65 – 3.56 (m, 0.74H), 3.50 (t, *J* = 7.02 Hz,

1.26H), 2.96 – 2.81 (m, 5H).

***N*-phenethylformamide (3n)¹²**



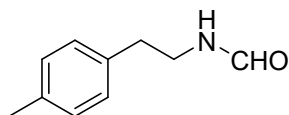
3n

The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a yellow oil (18.8 mg, 63%);

The presence of two rotamers (ratio 82:18) were observed in the NMR spectra.

¹H NMR (500 MHz, CDCl₃) δ 8.14 (d, *J* = 1.76 Hz, 0.83H), 7.93 (d, *J* = 12.02 Hz, 0.17H), 7.41 – 7.18 (m, 5H), 5.99 – 5.74 (brs, 1H), 3.61 (q, *J* = 6.68 Hz, 1.66H), 3.51 (q, *J* = 6.72 Hz, 0.34H), 2.87 (dt, *J* = 13.70, 6.87 Hz, 2H).

***N*-(4-methylphenethyl)formamide (3o)¹³**



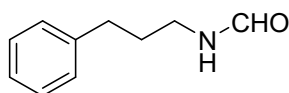
3o

The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a yellow oil (17.6 mg, 54%);

The presence of two rotamers (ratio 83:17) were observed in the NMR spectra.

¹H NMR (500 MHz, CDCl₃) δ 8.08 (d, *J* = 1.79 Hz, 0.82H), 7.86 (d, *J* = 11.91 Hz, 0.18H), 7.17 – 6.97 (m, 5H), 5.92 (brs, 1H), 3.52 (q, *J* = 6.71 Hz, 1.64H), 3.42 (q, *J* = 6.72 Hz, 0.36H), 2.77 (m, 3H), 2.32 (s, 3H).

***N*-(3-phenylpropyl)formamide (3p)⁸**



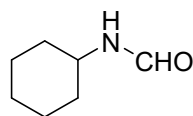
3p

The title compound was prepared according to the General procedure B and yields determined by ¹H NMR spectroscopy using trimethoxybenzene as an internal standard (78%);

The presence of two rotamers (ratio 82:18) were observed in the NMR spectra.

¹H NMR (500 MHz, CDCl₃) δ 8.17 (s, 0.82H), 8.04 (d, *J* = 11.94 Hz, 0.18H), 7.32 (t, *J* = 7.59 Hz, 2H), 7.22 (m, 3H), 6.04 – 5.84 (m, 1H), 3.35 (q, *J* = 6.76 Hz, 1.64H), 3.24 (q, *J* = 6.76 Hz, 0.36H), 2.69 (t, *J* = 7.69 Hz, 2H), 1.92 – 1.86 (m, 3H).

***N*-cyclohexylformamide (3q)¹⁴**



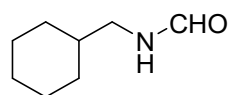
3q

The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a brown oil (24.5 mg, 75%);

The presence of two rotamers (ratio 86:14) were observed in the NMR spectra.

^1H NMR (500 MHz, CDCl_3) δ 8.12 (s, 0.14H), 8.09 (s, 0.86H), 5.73 (brs, 0.2H), 5.50 (brs, 0.8H), 3.85 (m, 0.84H), 3.71 (q, $J = 7.02$ Hz, 0.15H), 2.16 – 1.53 (m, 6H), 1.41 – 1.00 (m, 5H).

***N*-(cyclohexylmethyl)formamide (3r)¹⁵**



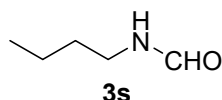
3r

The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a white solid (20.9 mg, 74%);

The presence of two rotamers (ratio 77:23) were observed in the NMR spectra.

^1H NMR (500 MHz, CDCl_3) δ 8.16 (s, 0.77H), 7.97 (d, $J = 11.91$ Hz, 0.23H), 5.94 (brs, 0.23H), 5.82 (brs, 0.77H), 3.12 (t, $J = 6.52$ Hz, 1.54), 3.02 (t, $J = 6.61$ Hz, 0.46H), 1.89 – 1.61 (m, 6H), 1.43 (m, 1H), 1.37 – 1.07 (m, 3H), 1.01 – 0.80 (m, 2H).

***N*-butylformamide (3s)¹⁶**



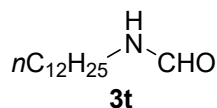
3s

The title compound was prepared according to the General procedure B and yields determined by ^1H NMR spectroscopy using trimethoxybenzene as an internal standard (71%);

The presence of two rotamers (ratio 77:23) were observed in the NMR spectra.

^1H NMR (500 MHz, CDCl_3) δ 8.20 (s, 0.77H), 8.07 (d, $J = 12.09$ Hz, 0.23H), 5.98 (brs, 0.23H), 5.76 (brs, 0.77H), 3.53 – 3.19 (m, 2H), 1.54 (m 2H), 1.48 – 1.24 (m, 2H), 0.96 (q, $J = 11.73, 9.55$ Hz, 3H)

***N*-(dodecyl)formamide (3t)¹²**



3t

The title compound was prepared according to the General procedure B and purified by flash column chromatography to give the product as a gray solid (29.8mg, 70%);

The presence of two rotamers (ratio 77:23) were observed in the NMR spectra.

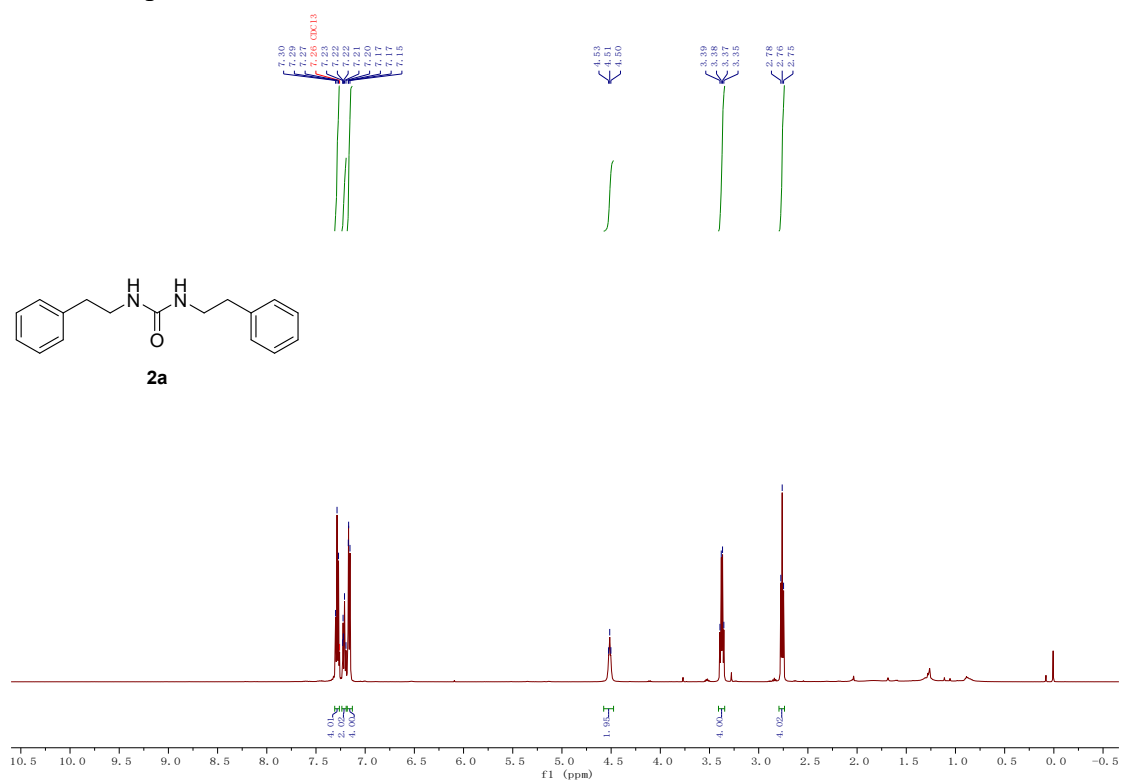
^1H NMR (500 MHz, CDCl_3) δ 8.13 (s, 0.81), 8.00 (d, $J = 11.96$ Hz, 0.19H), 5.95 (brs, 0.19H), 5.89 (brs, 0.81H), 3.26 (q, $J = 6.83$ Hz, 1.6H), 3.17 (t, $J = 6.83$ Hz, 0.4H), 1.49 (q, $J = 7.19$ Hz, 3H), 1.25 (m, 18H), 0.85 (t, $J = 6.80$ Hz, 3H).

6. Reference.

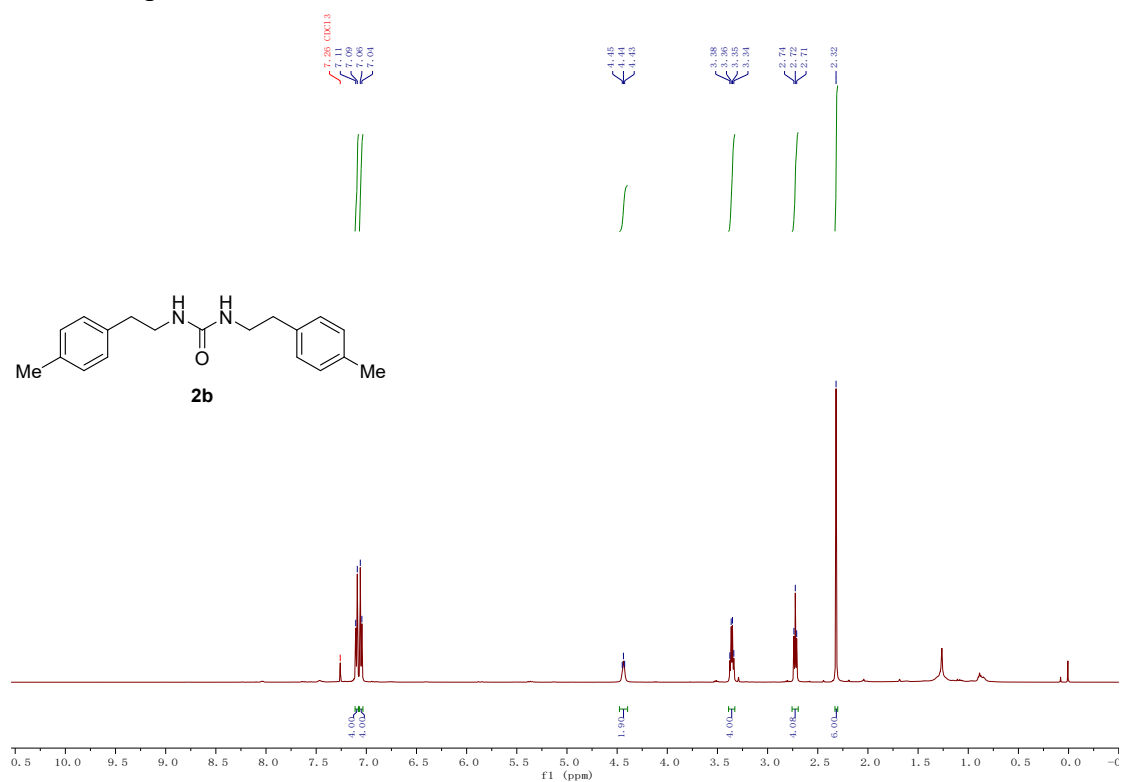
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7. NMR spectra

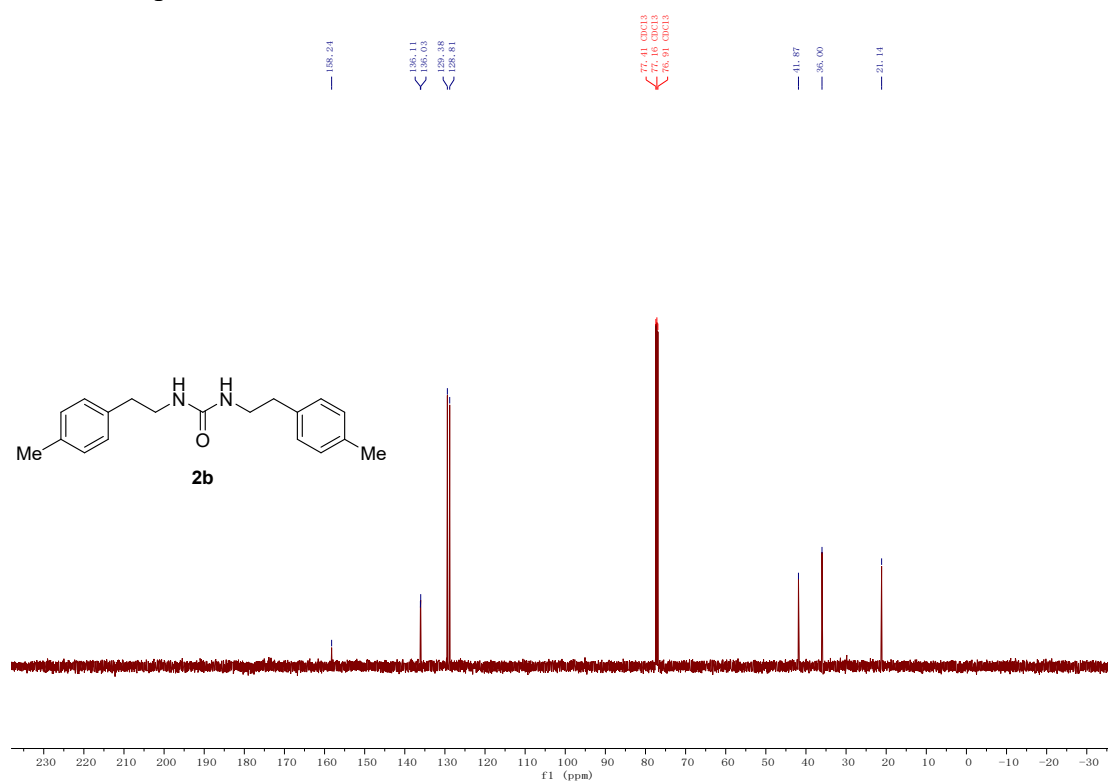
^1H NMR spectrum of **2a** in CDCl_3 at 500 MHz



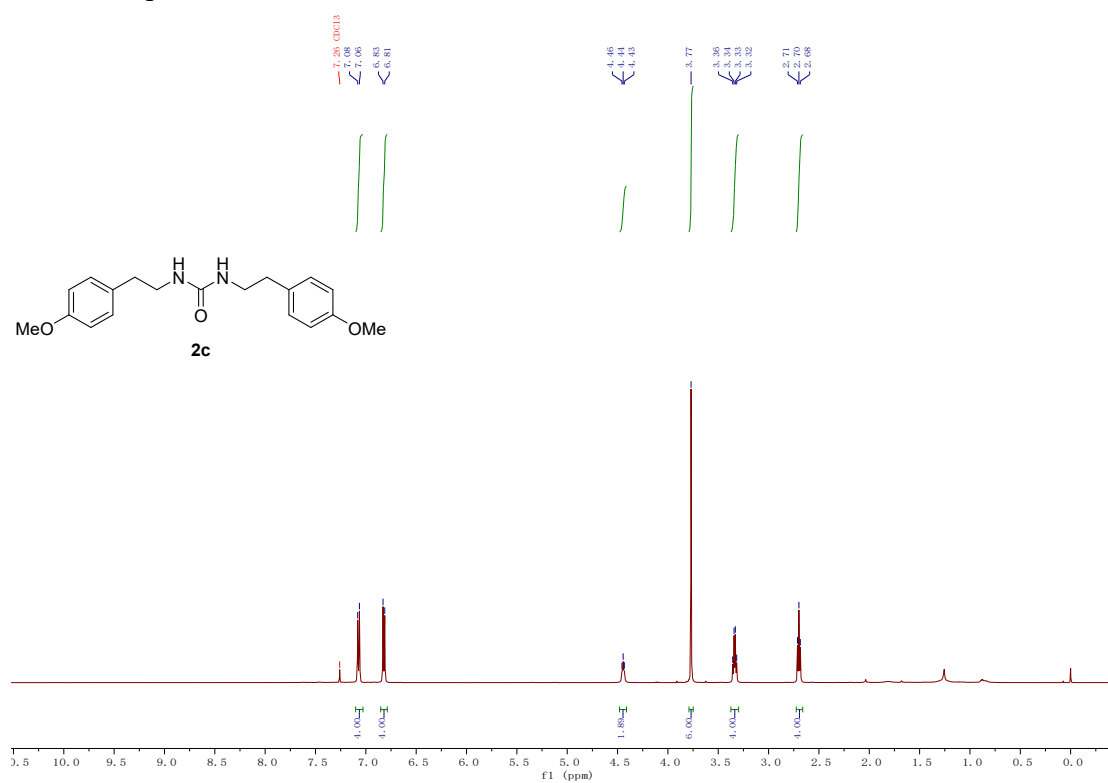
^1H NMR spectrum of **2b** in CDCl_3 at 500 MHz



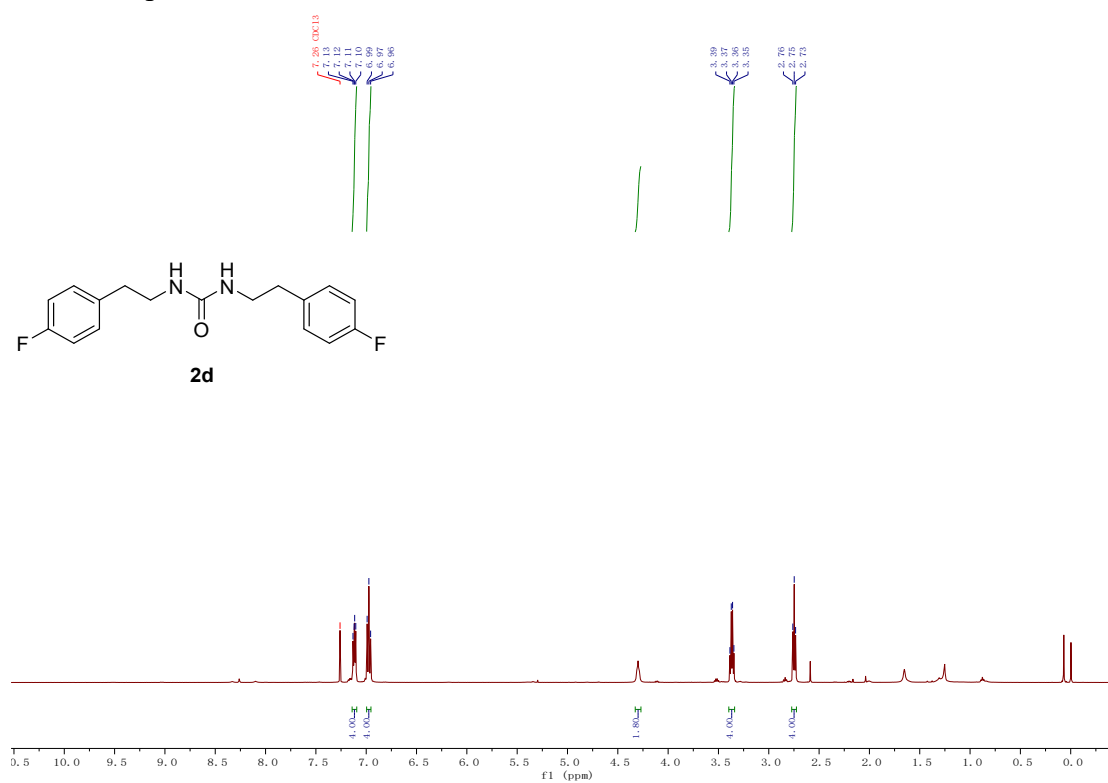
^{13}C NMR spectrum of **2b** in CDCl_3 at 126 MHz



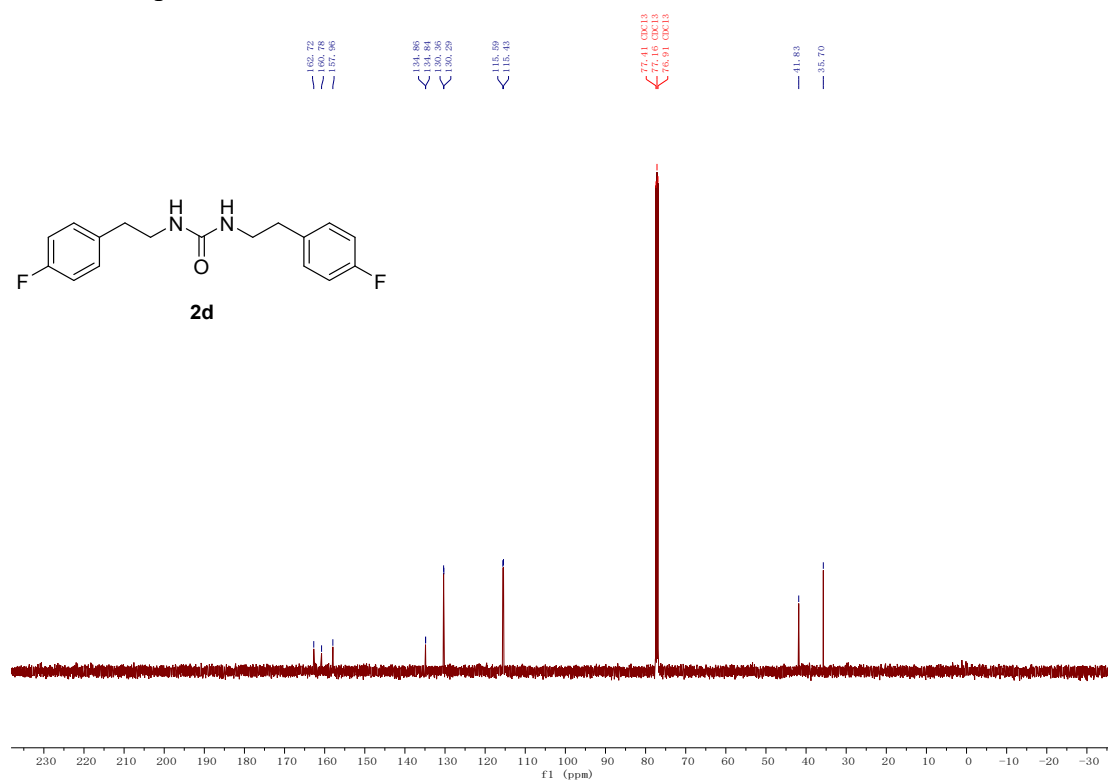
^1H NMR spectrum of **2c** in CDCl_3 at 500 MHz



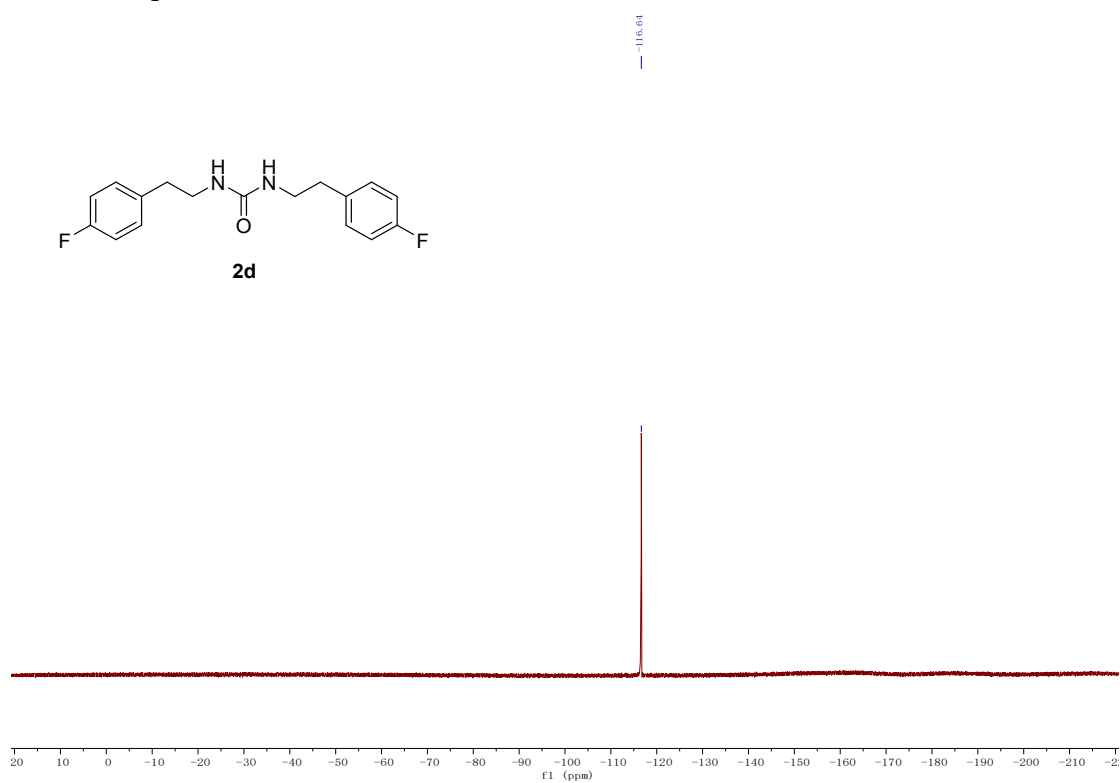
^1H NMR spectrum of **2d** in CDCl_3 at 500 MHz



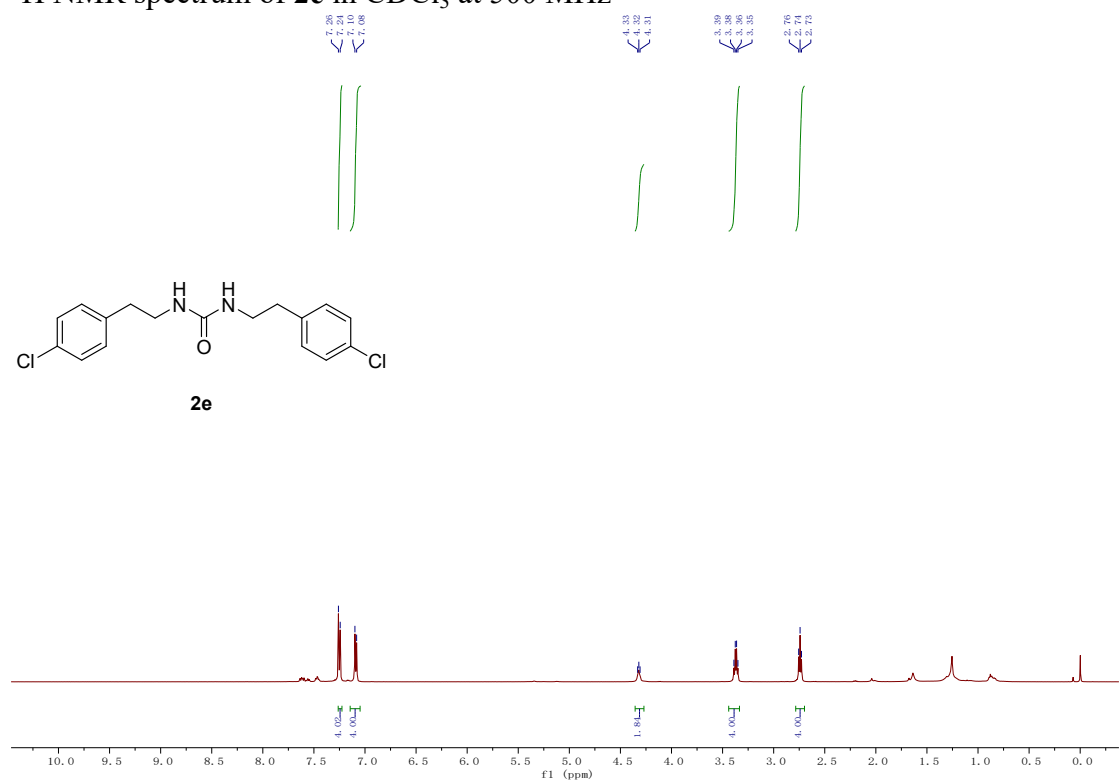
^{13}C NMR spectrum of **2d** in CDCl_3 at 126 MHz



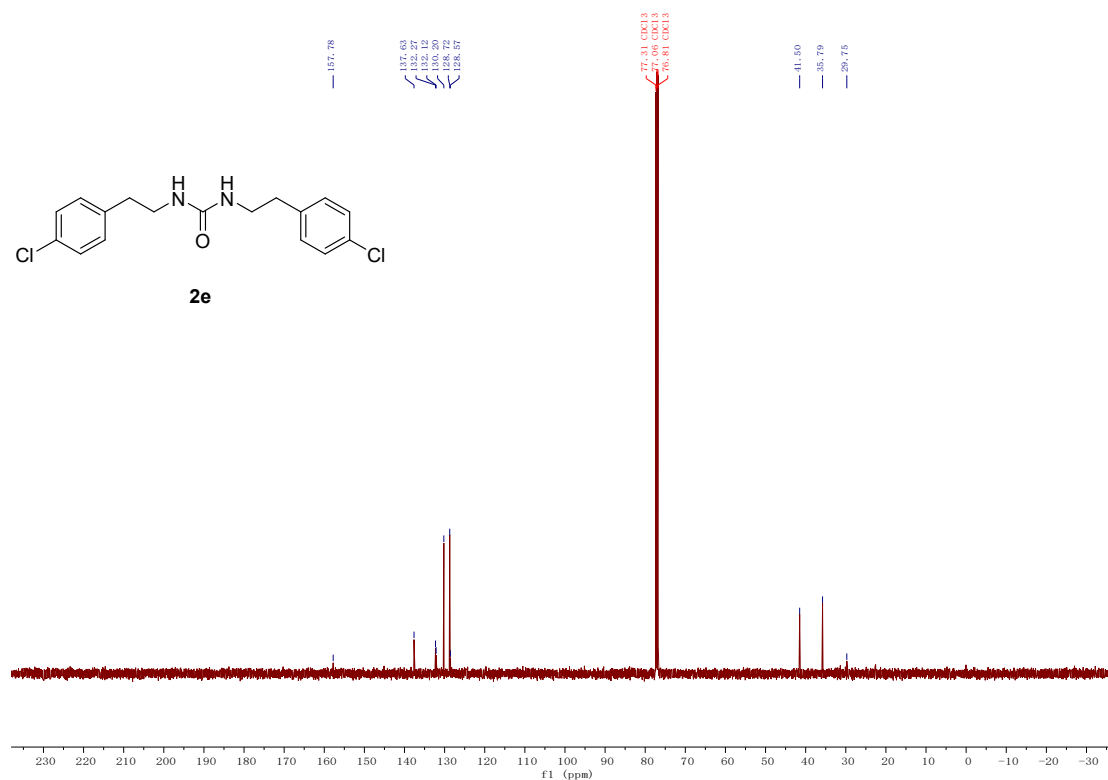
^{19}F NMR spectrum of **2d** in CDCl_3 at 475 MHz



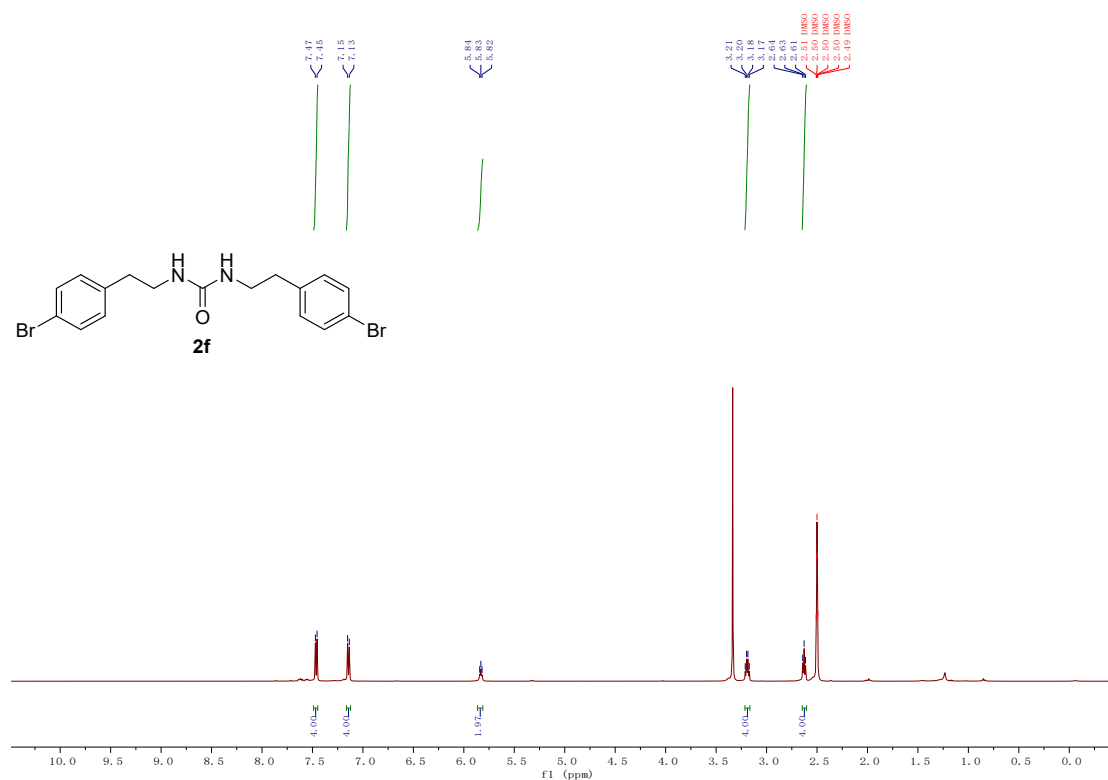
^1H NMR spectrum of **2e** in CDCl_3 at 500 MHz



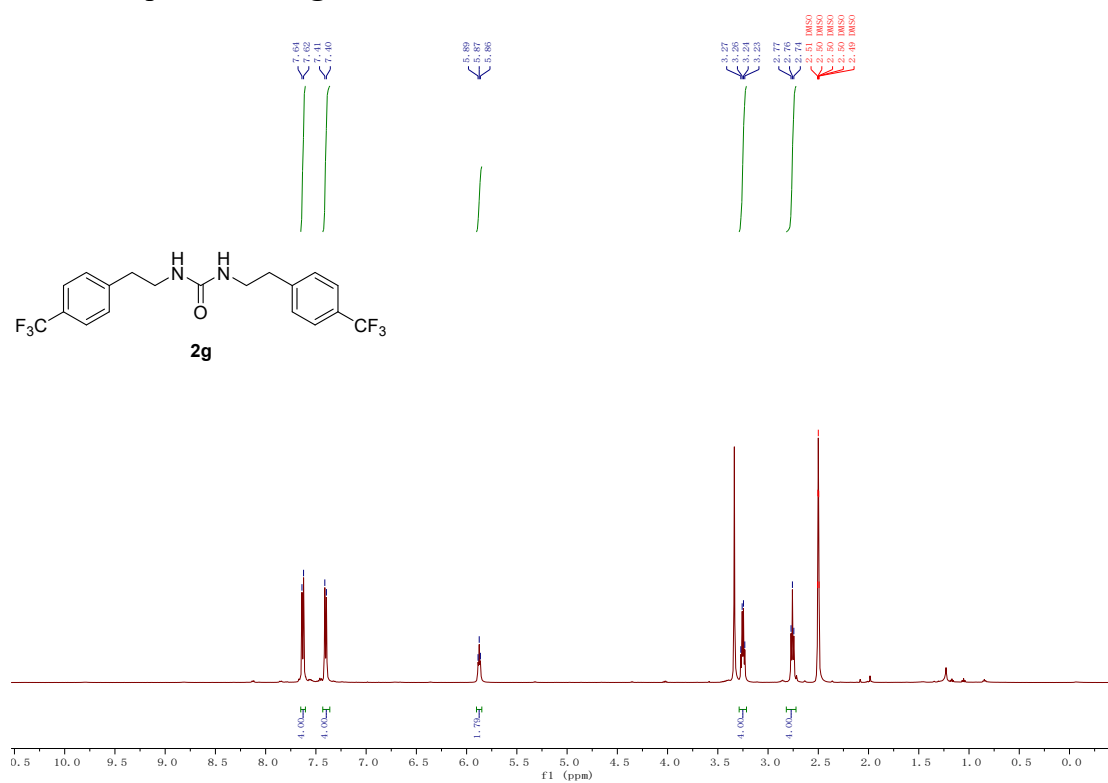
^{13}C NMR spectrum of **2e** in CDCl_3 at 126 MHz



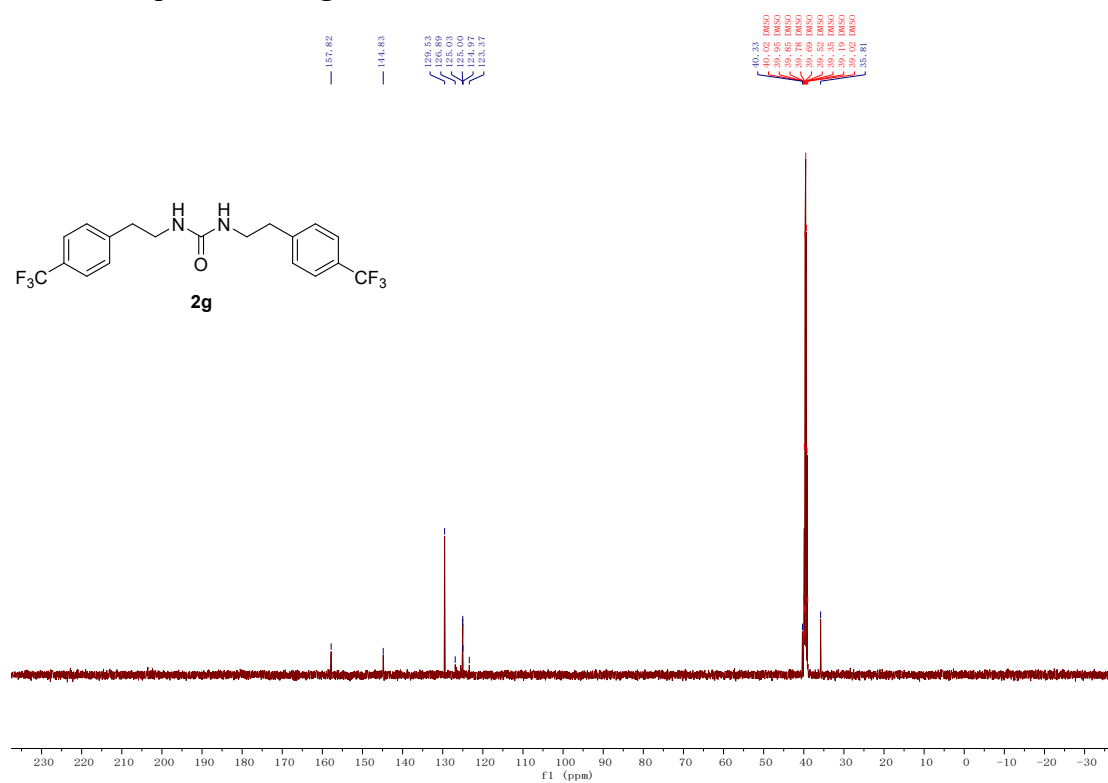
^1H NMR spectrum of **2f** in $\text{DMSO}-d_6$ at 500 MHz



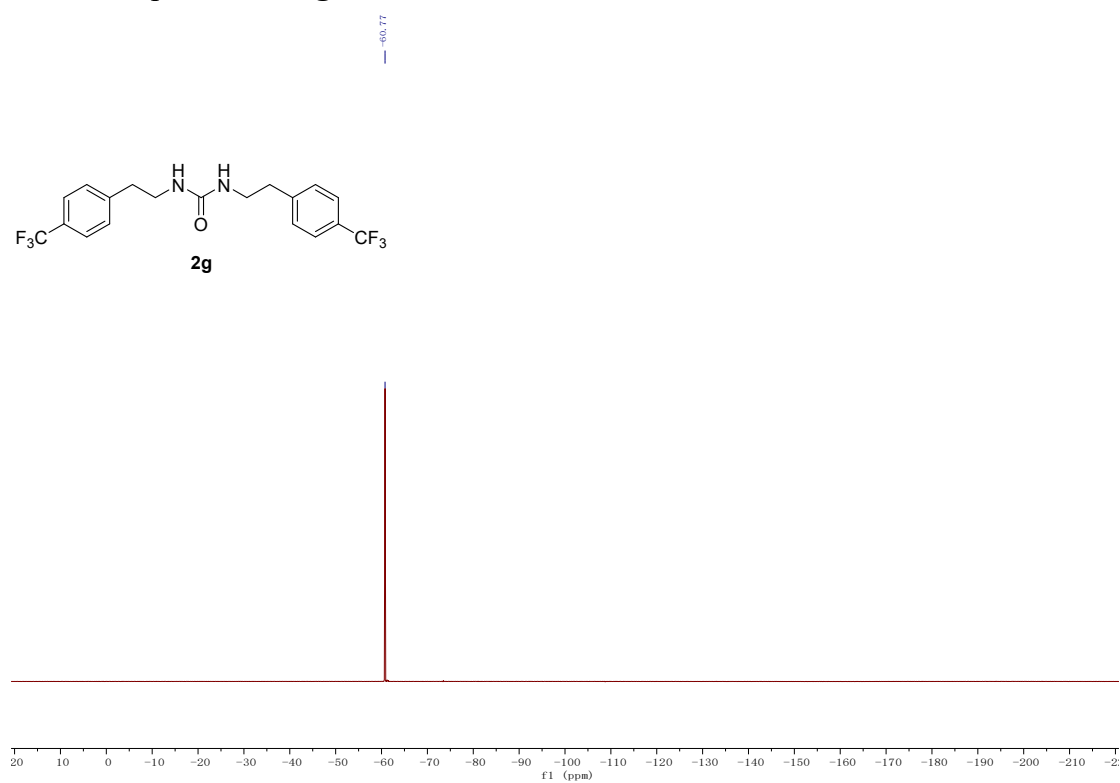
^1H NMR spectrum of **2g** in $\text{DMSO}-d_6$ at 500 MHz



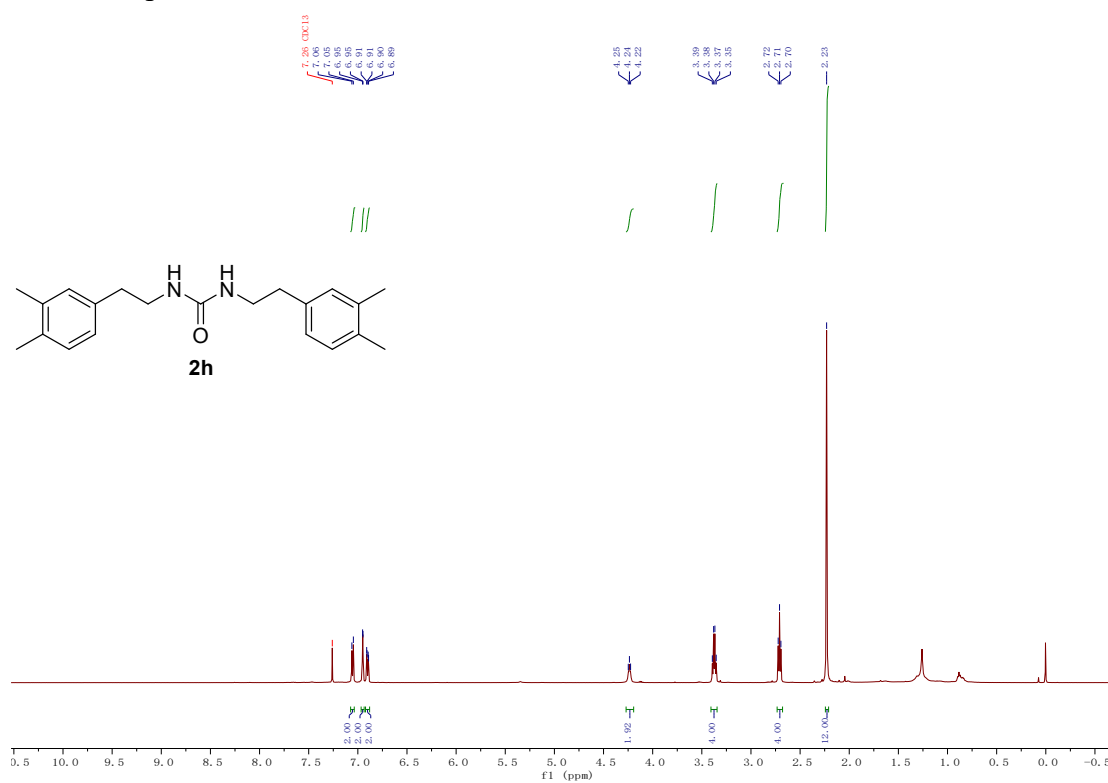
^{13}C NMR spectrum of **2g** in $\text{DMSO}-d_6$ at 126 MHz



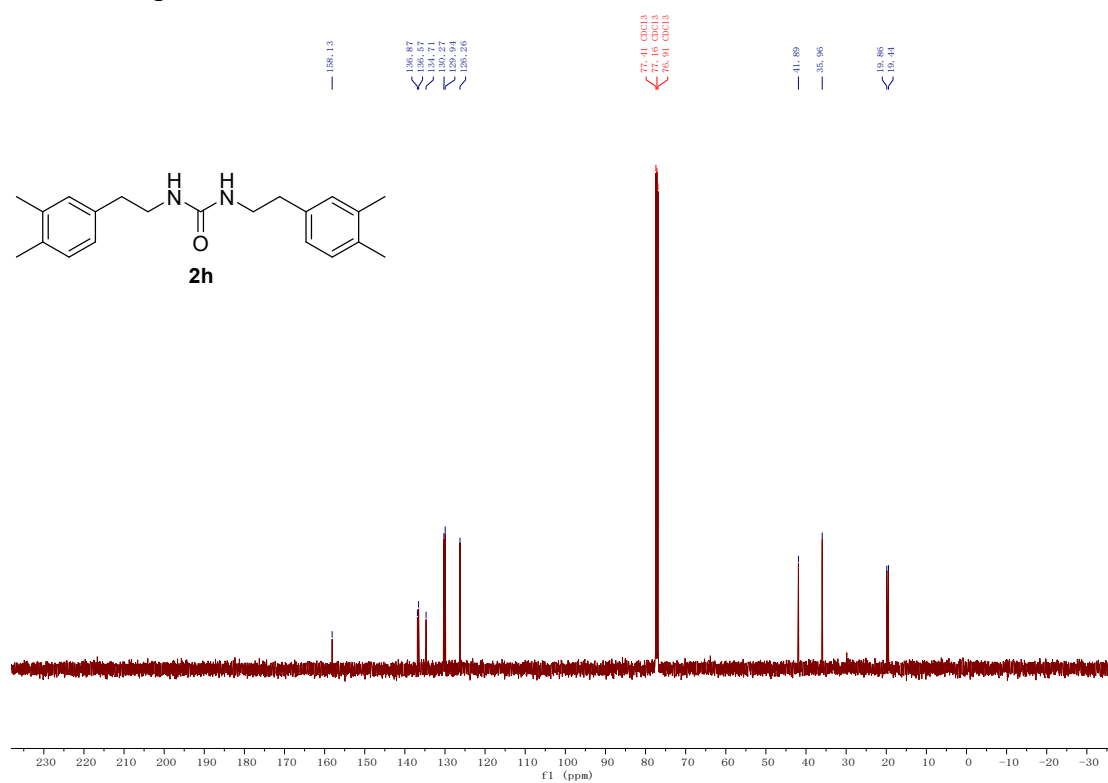
^{19}F NMR spectrum of **2g** in $\text{DMSO}-d_6$ at 470 MHz



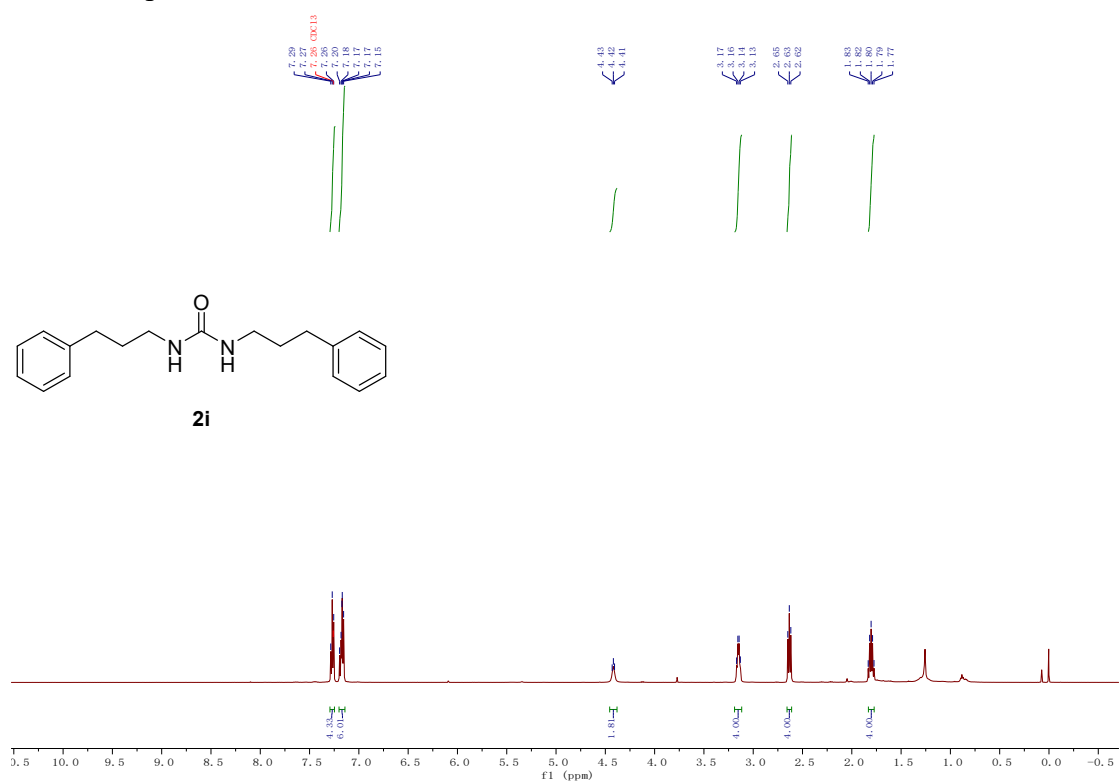
^1H NMR spectrum of **2h** in CDCl_3 at 500 MHz



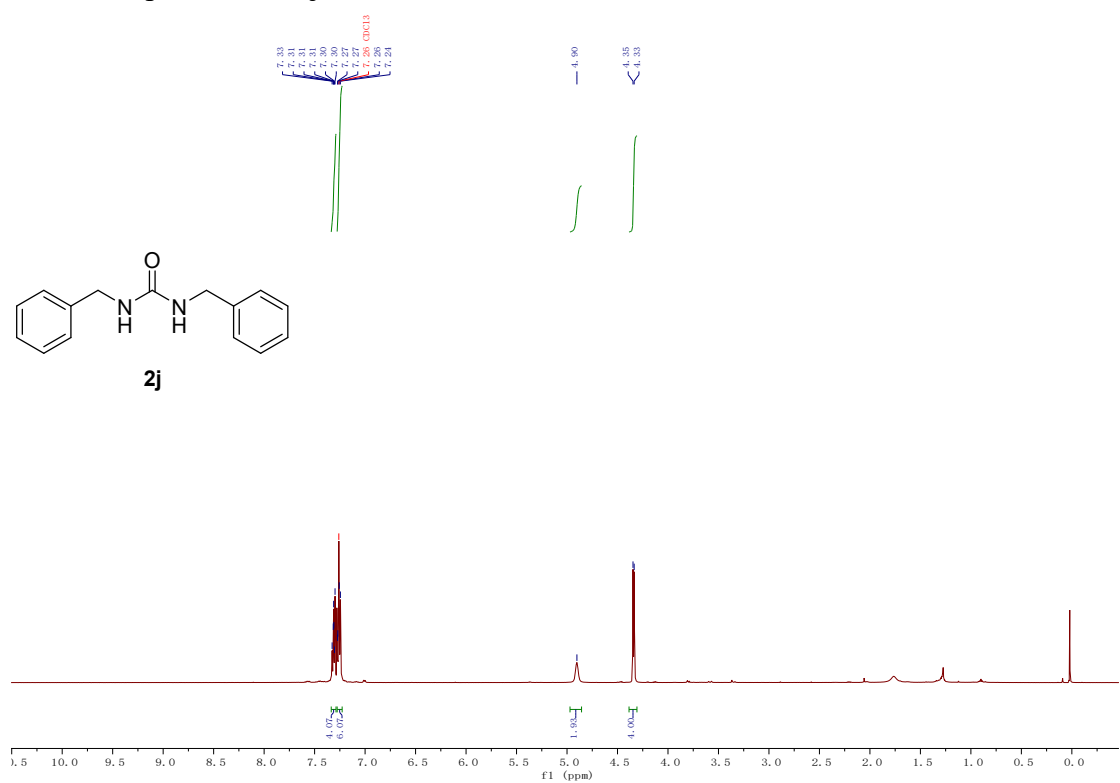
^{13}C NMR spectrum of **2h** in CDCl_3 at 126 MHz



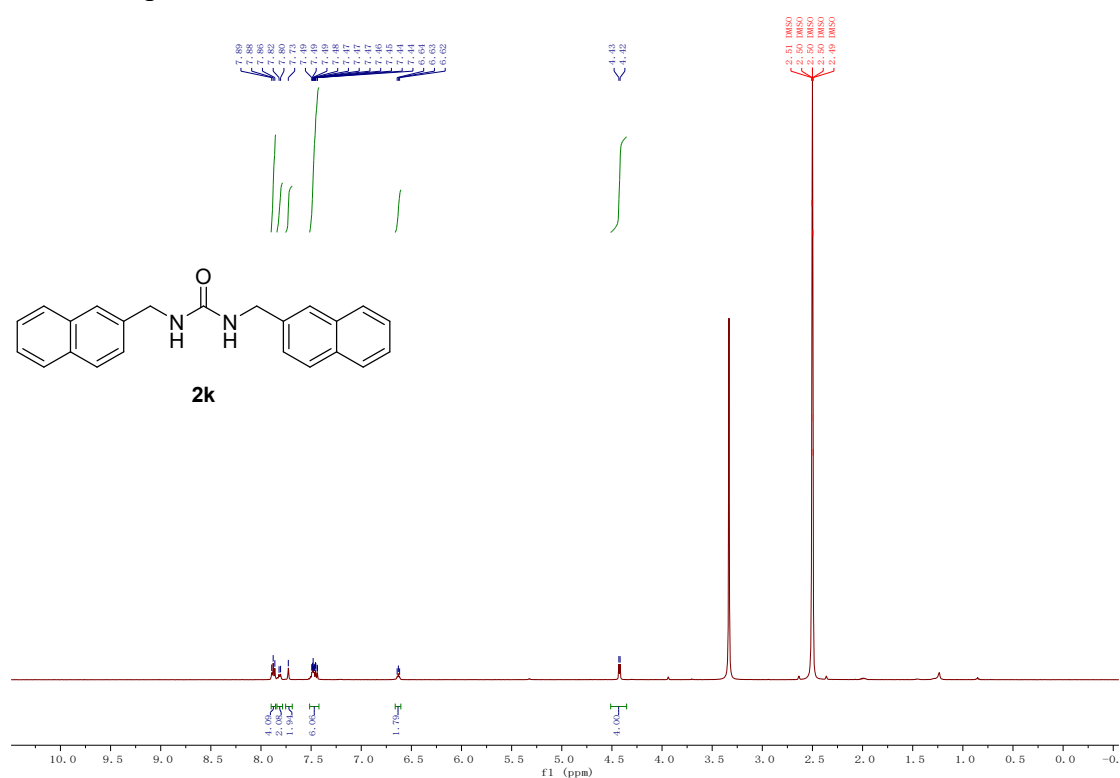
^1H NMR spectrum of **2i** in CDCl_3 at 500 MHz



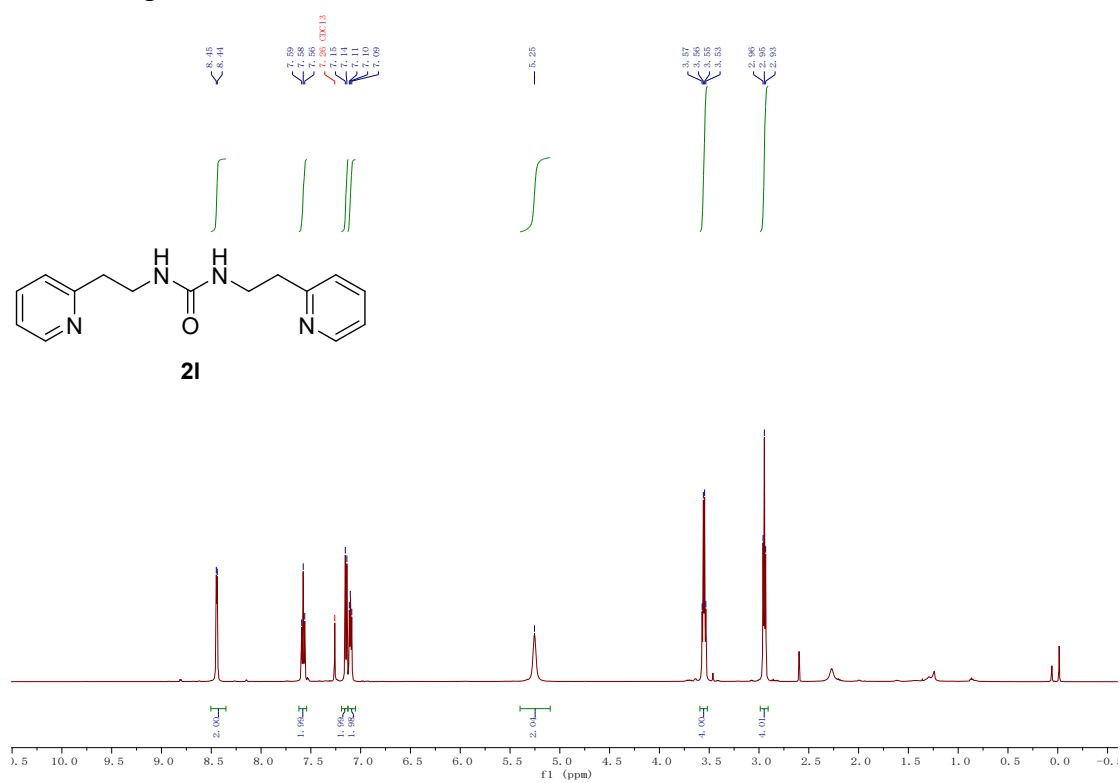
^1H NMR spectrum of **2j** in CDCl_3 at 500 MHz



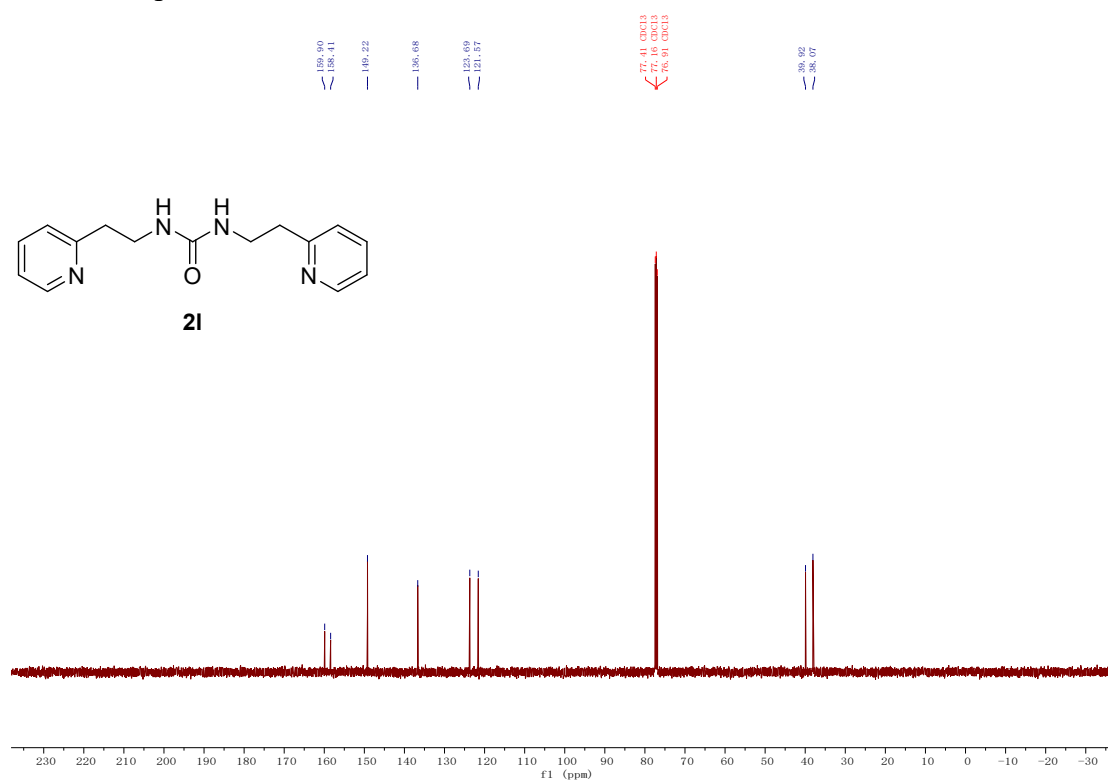
^1H NMR spectrum of **2k** in $\text{DMSO}-d_6$ at 500 MHz



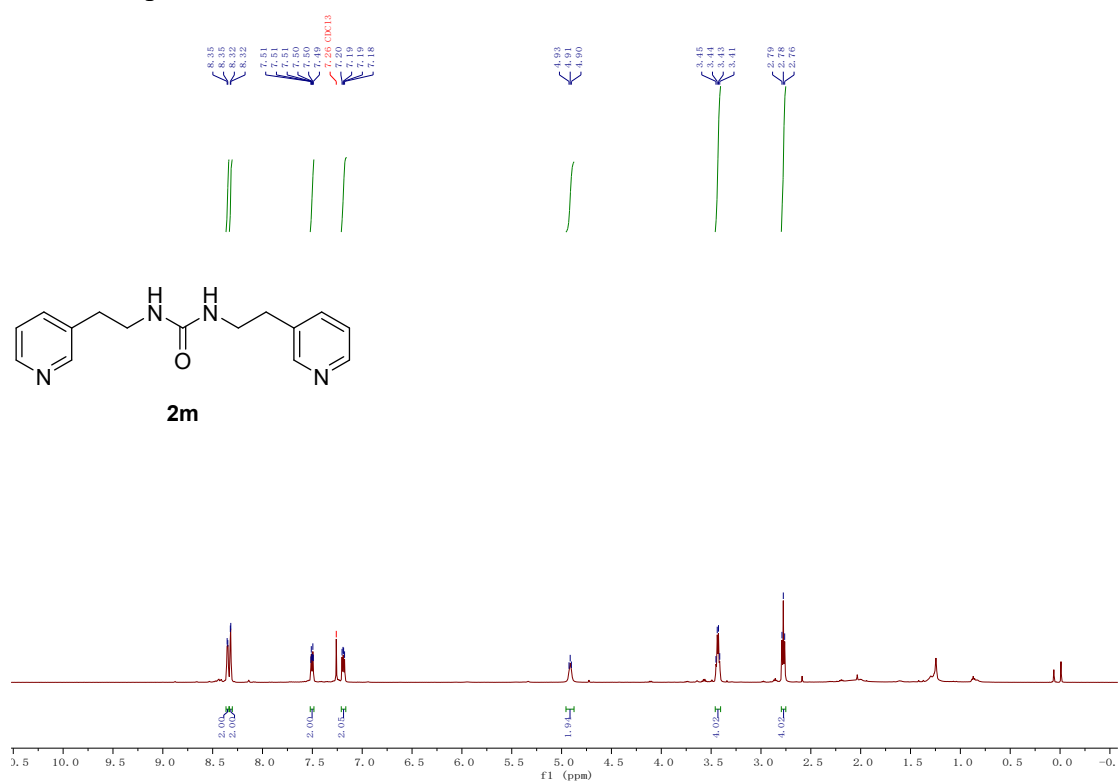
^1H NMR spectrum of **2l** in CDCl_3 at 500 MHz



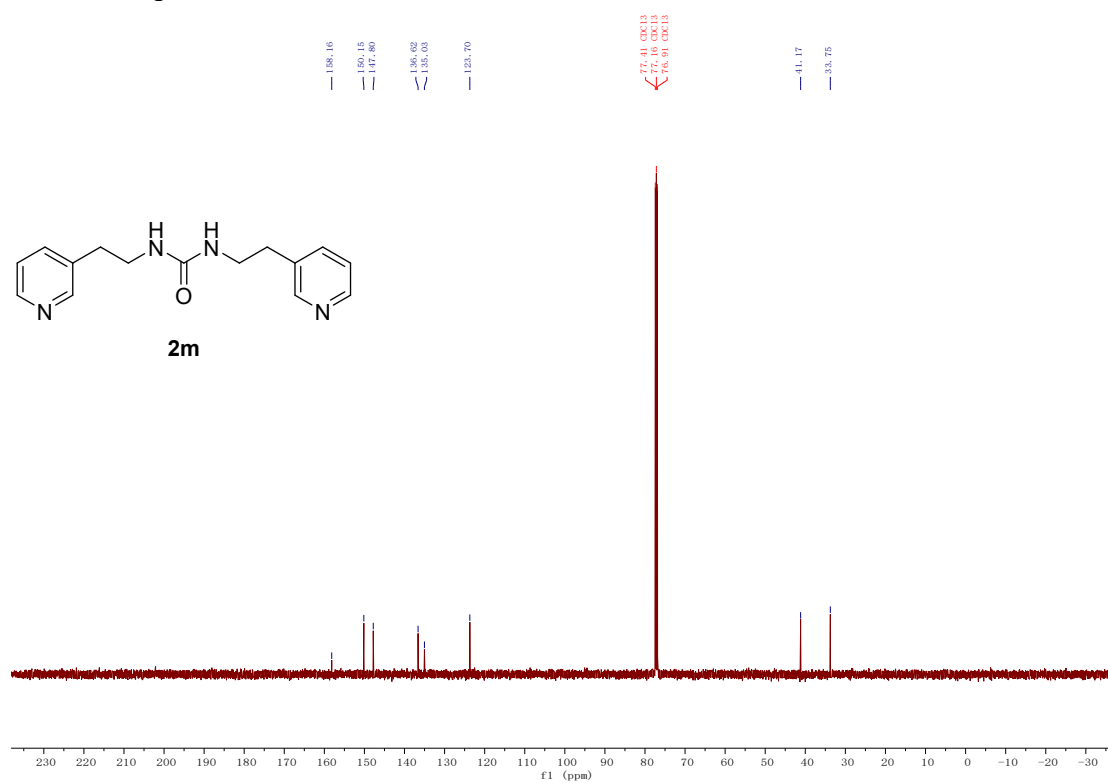
^{13}C NMR spectrum of **2l** in CDCl_3 at 126 MHz



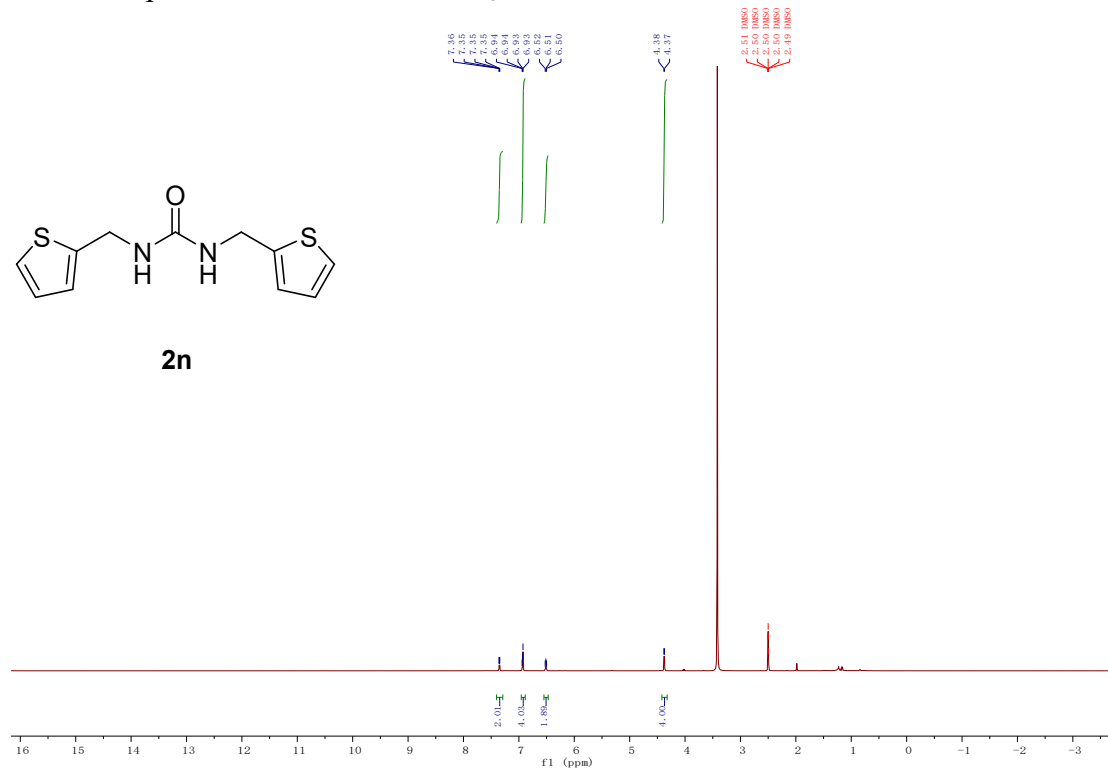
^1H NMR spectrum of **2m** in CDCl_3 at 500 MHz



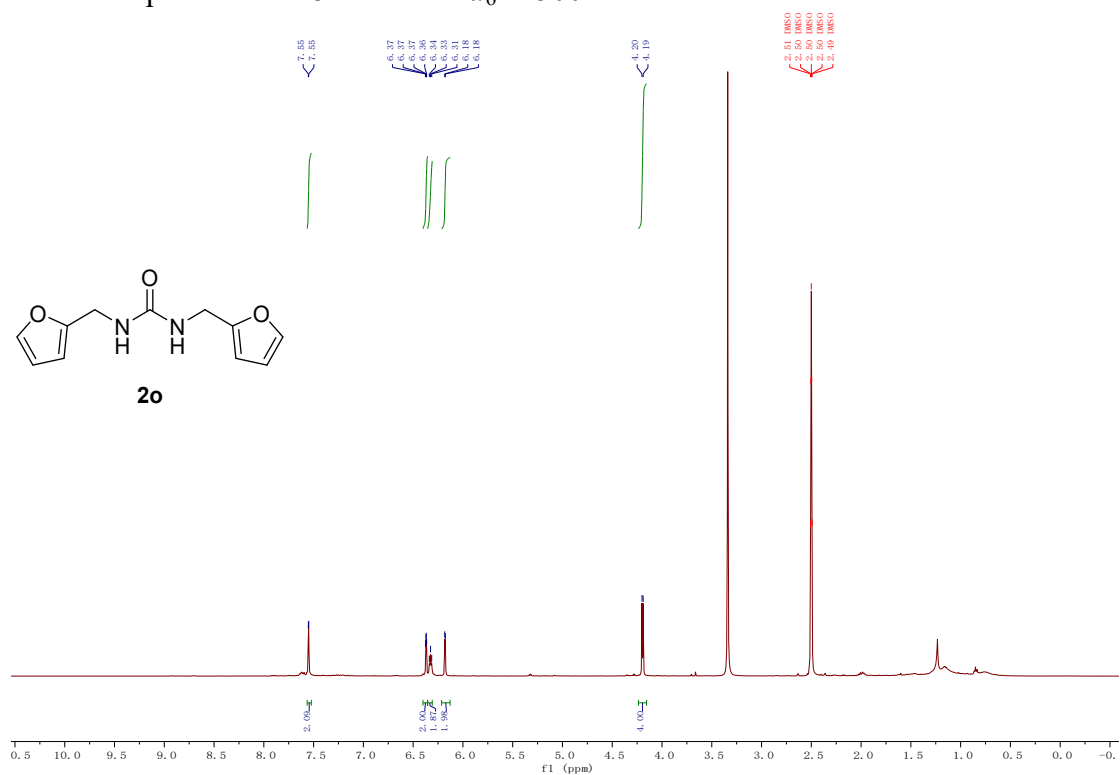
^{13}C NMR spectrum of **2m** in CDCl_3 at 126 MHz



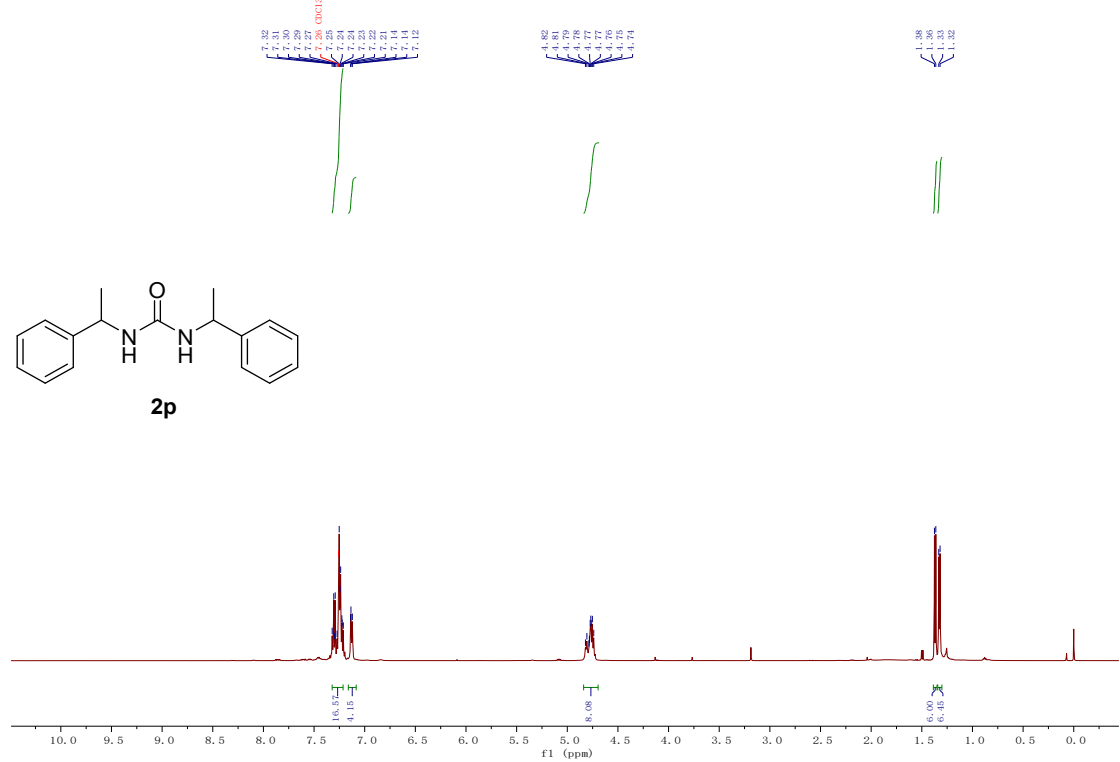
^1H NMR spectrum of **2n** in $\text{DMSO}-d_6$ at 500 MHz



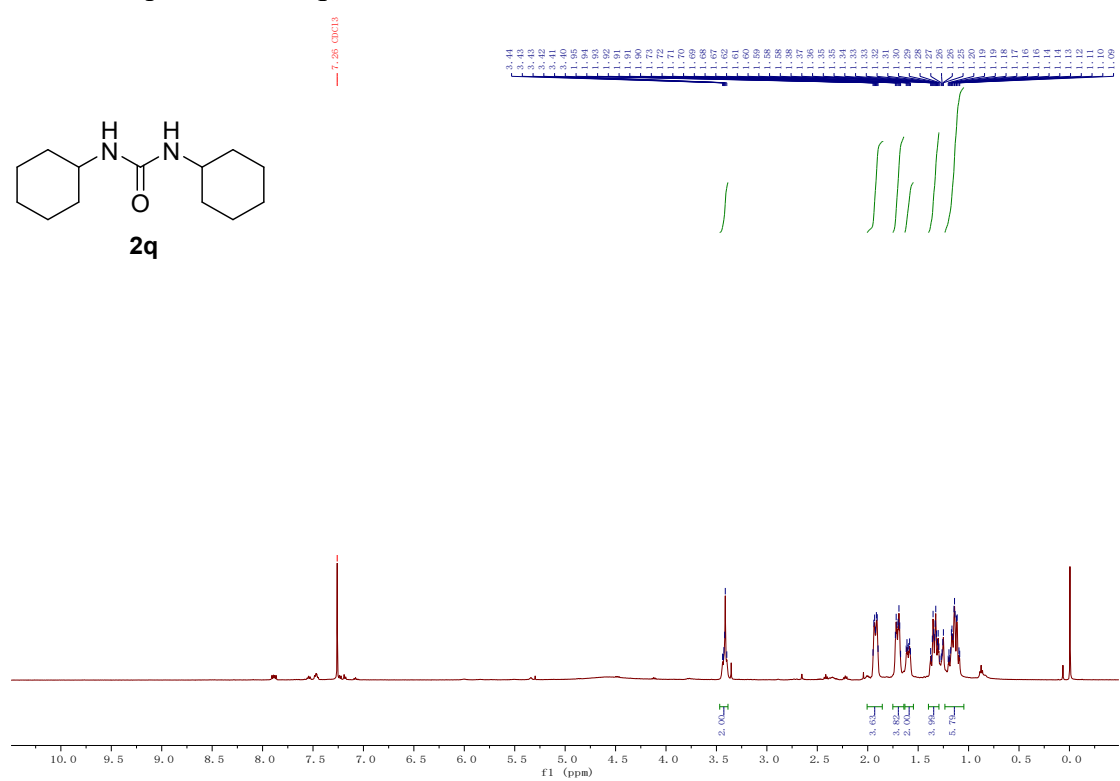
^1H NMR spectrum of **2o** in $\text{DMSO}-d_6$ at 500 MHz



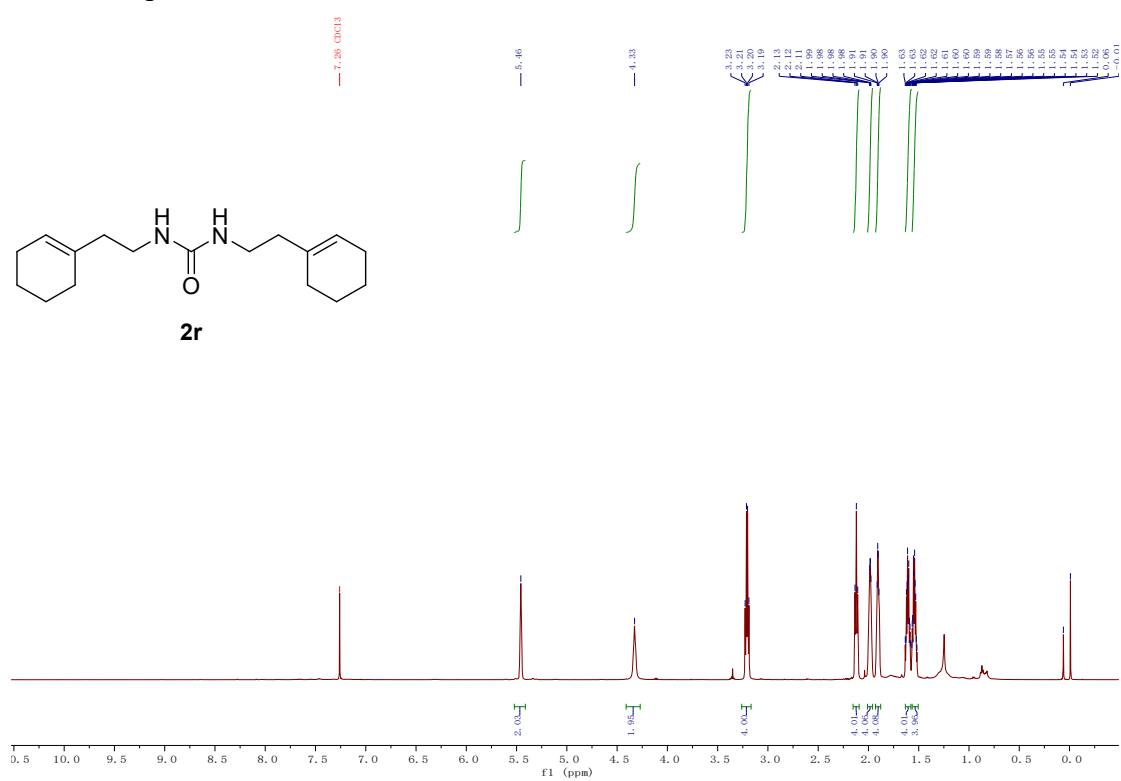
^1H NMR spectrum of **2p** in CDCl_3 at 500 MHz



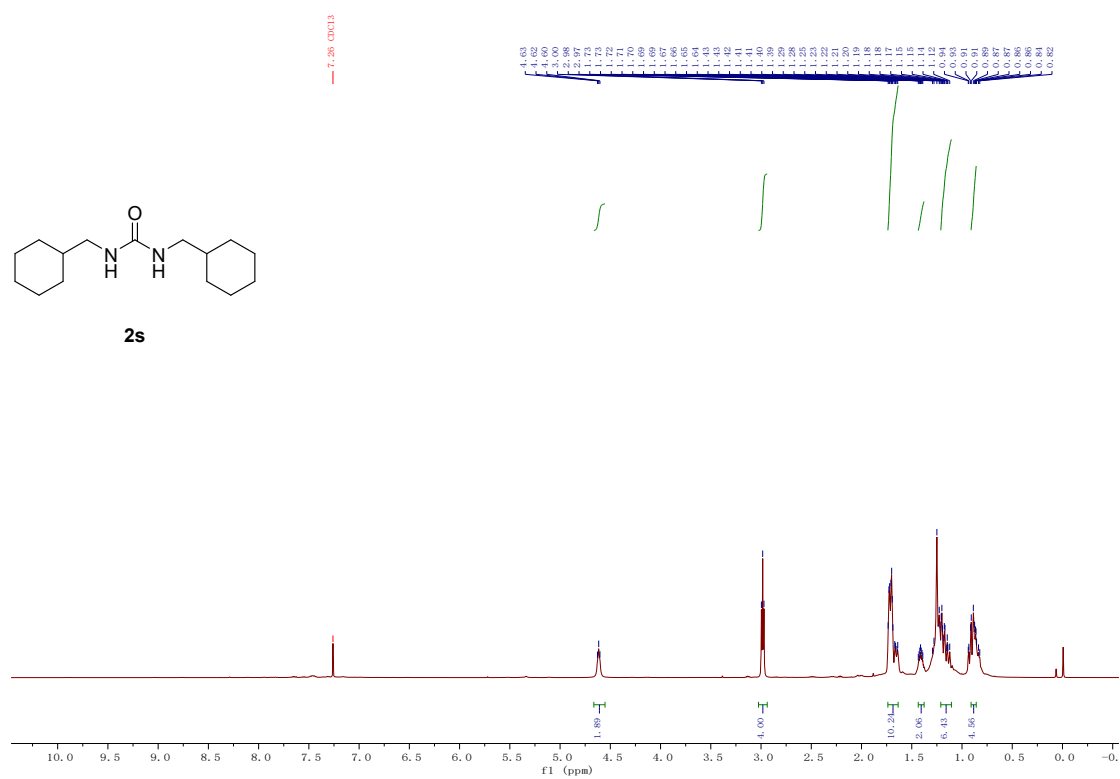
^1H NMR spectrum of **2q** in CDCl_3 at 500 MHz



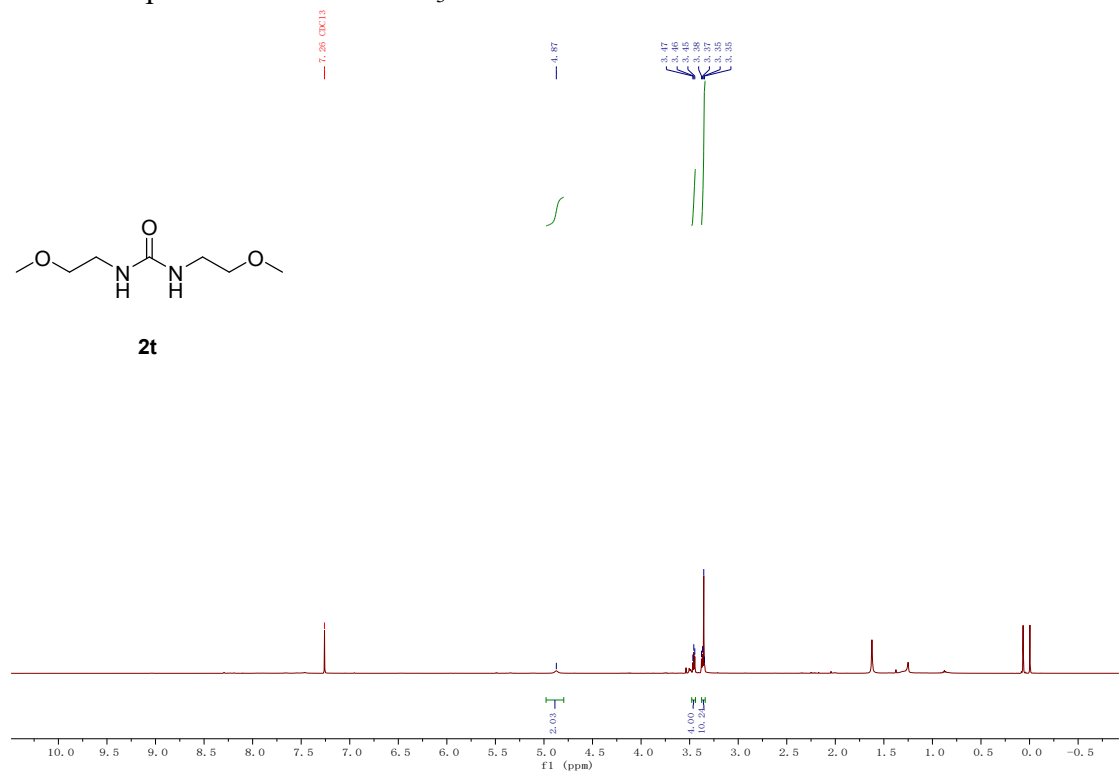
^1H NMR spectrum of **2r** in CDCl_3 at 500 MHz



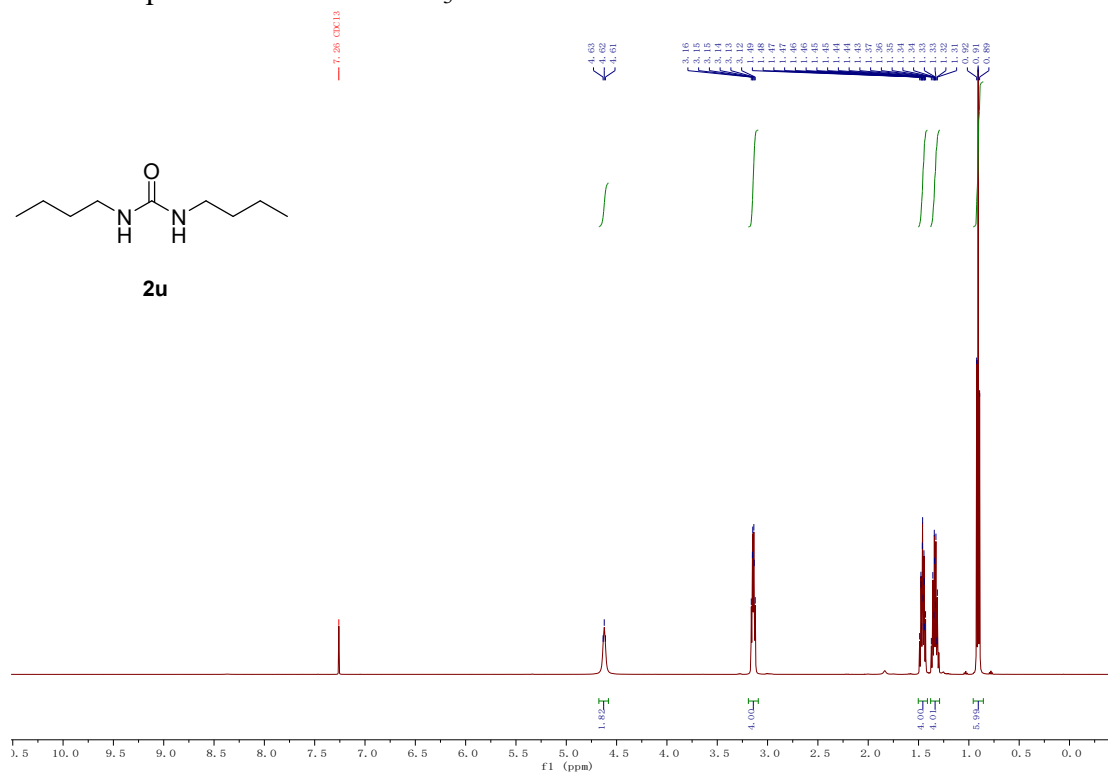
^1H NMR spectrum of **2s** in CDCl_3 at 126 MHz



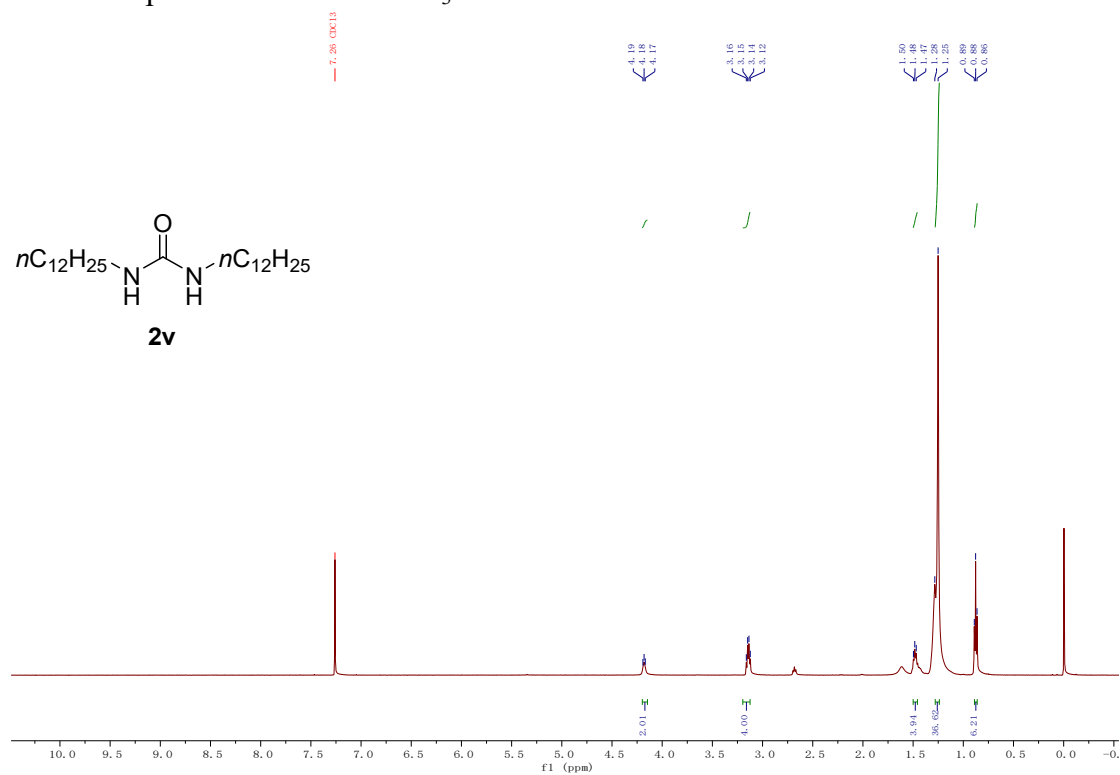
^1H NMR spectrum of **2t** in CDCl_3 at 500 MHz



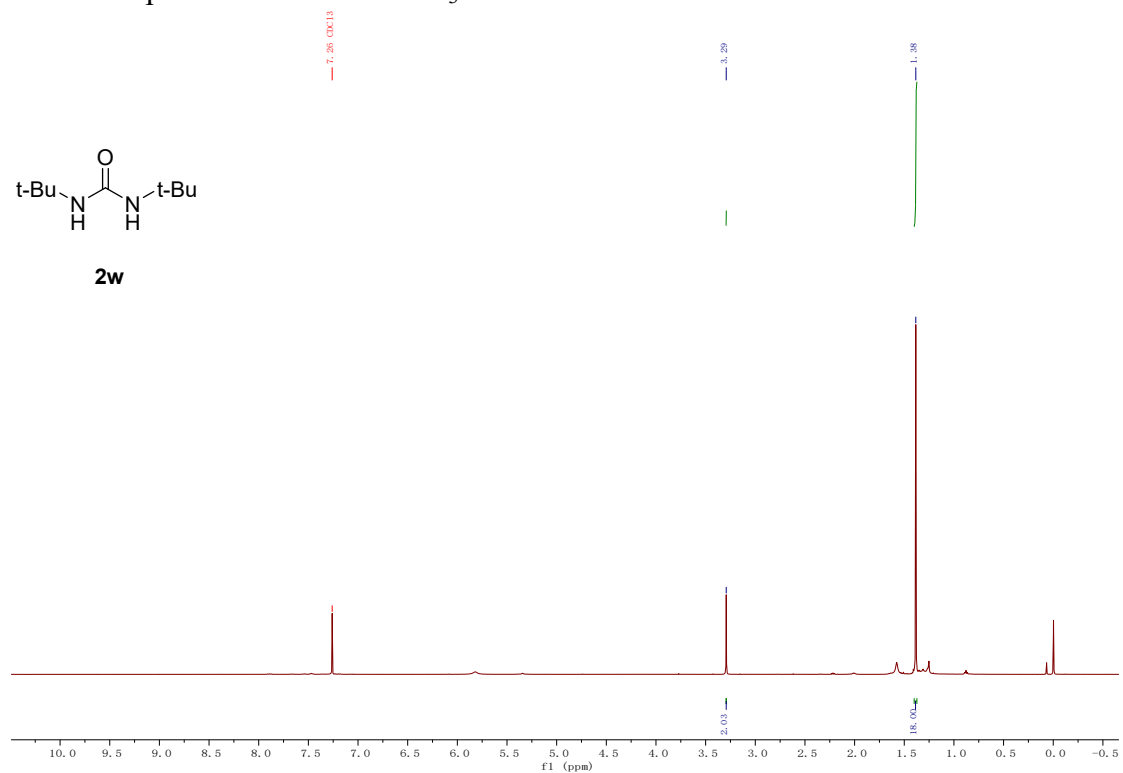
^1H NMR spectrum of **2u** in CDCl_3 at 500 MHz



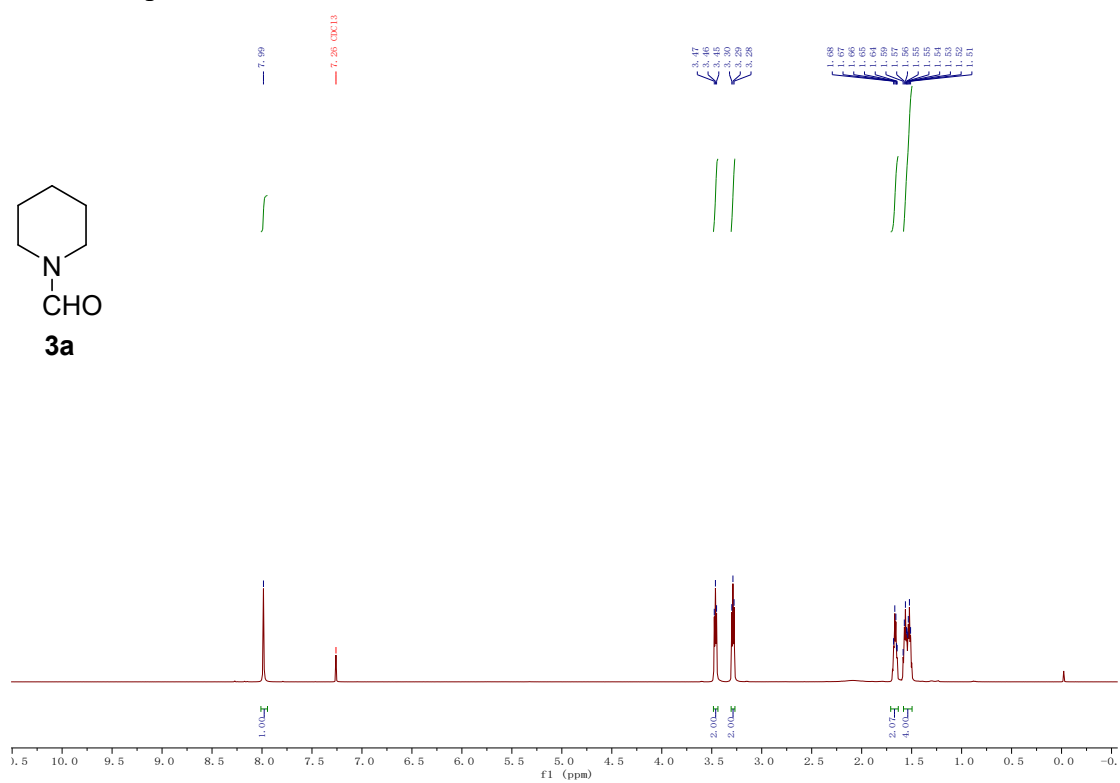
^1H NMR spectrum of **2v** in CDCl_3 at 500 MHz



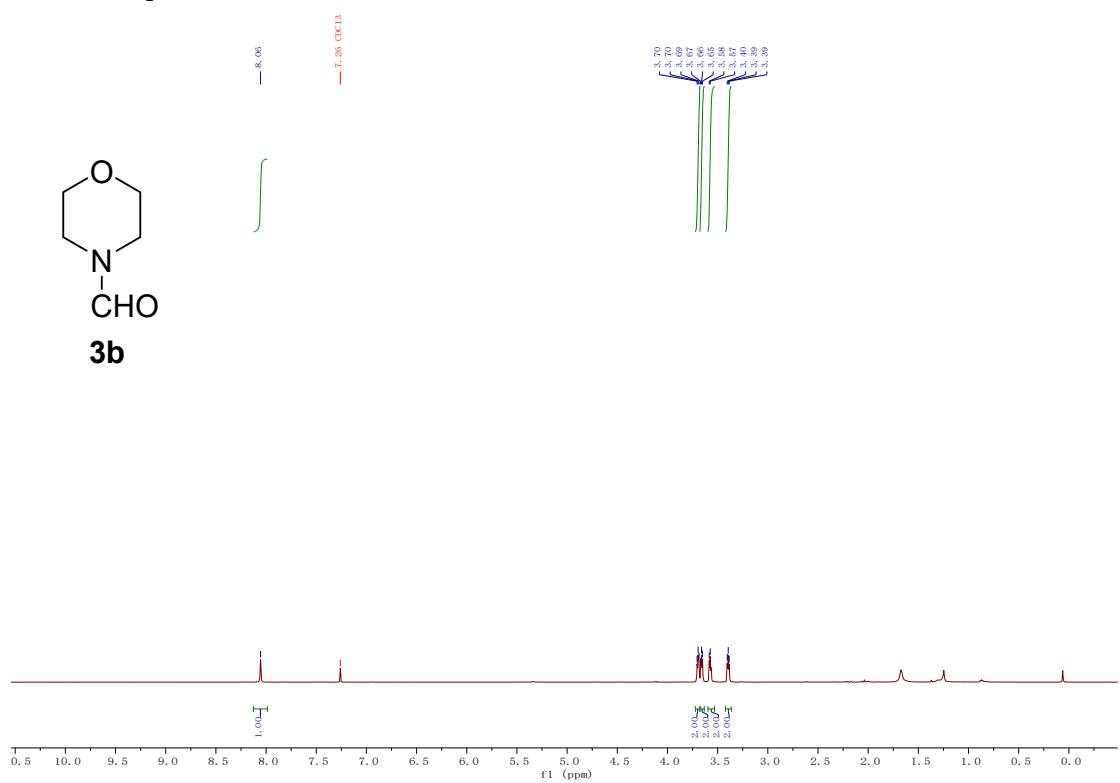
^1H NMR spectrum of **2w** in CDCl_3 at 500 MHz



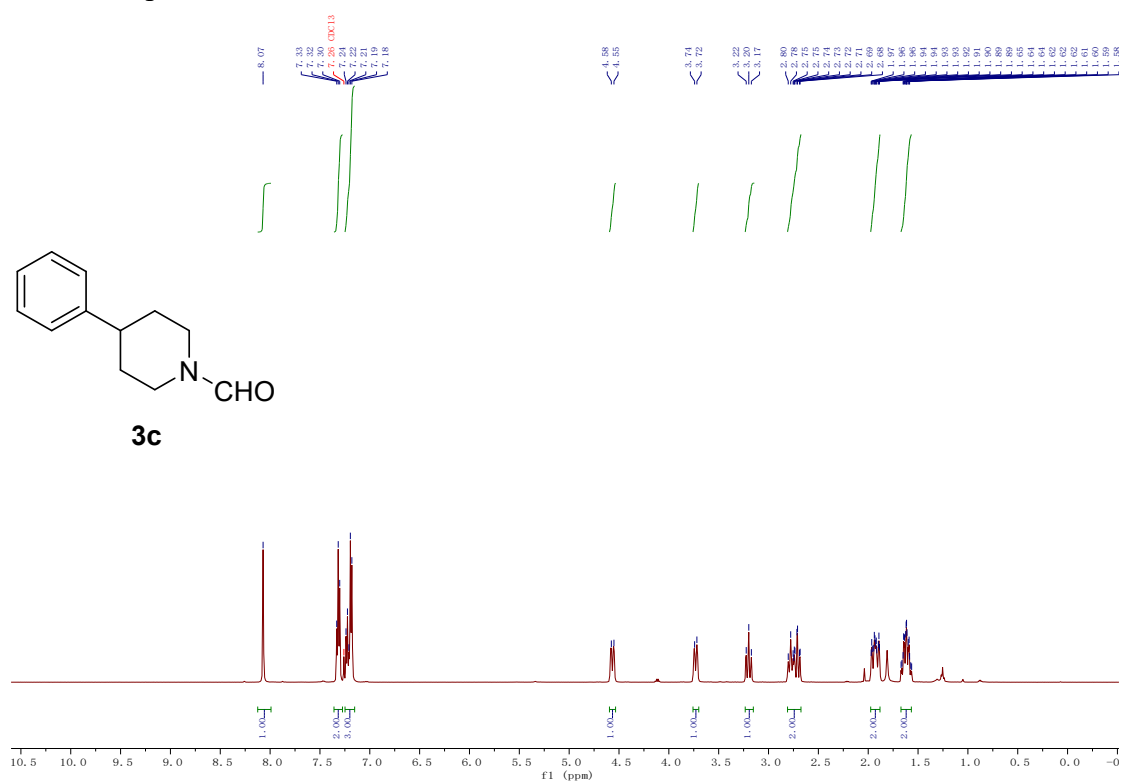
^1H NMR spectrum of **3a** in CDCl_3 at 500 MHz



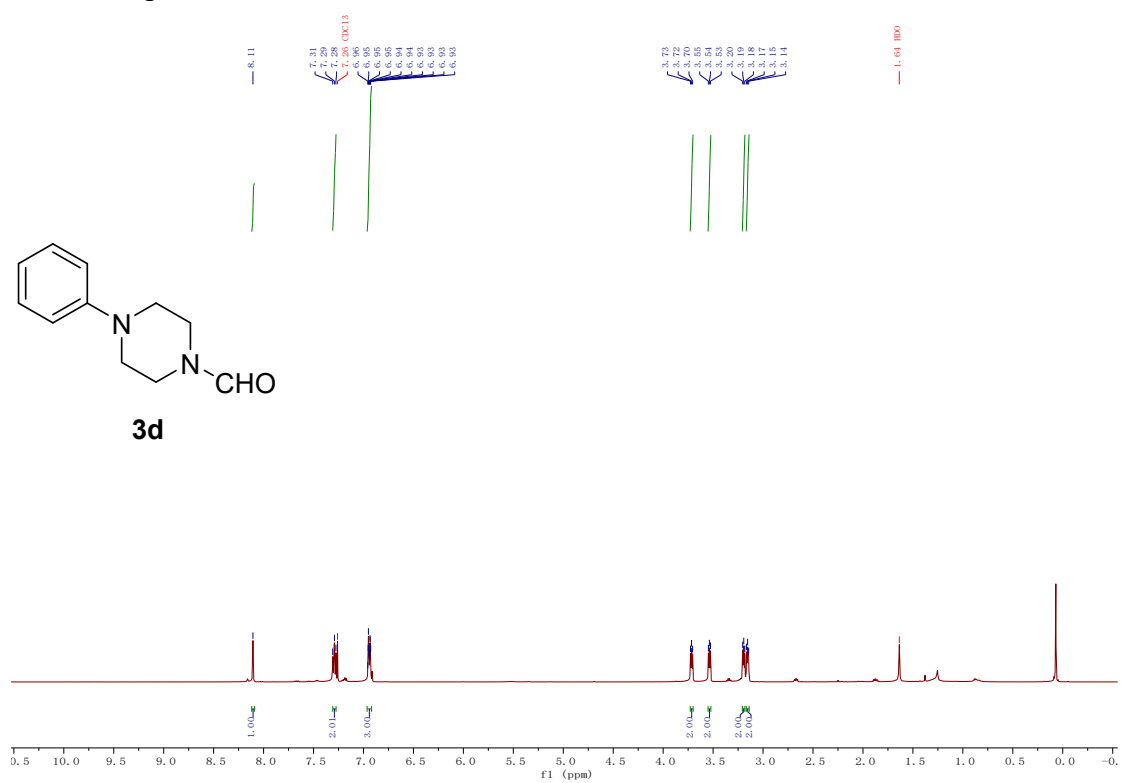
^1H NMR spectrum of **3b** in CDCl_3 at 500 MHz



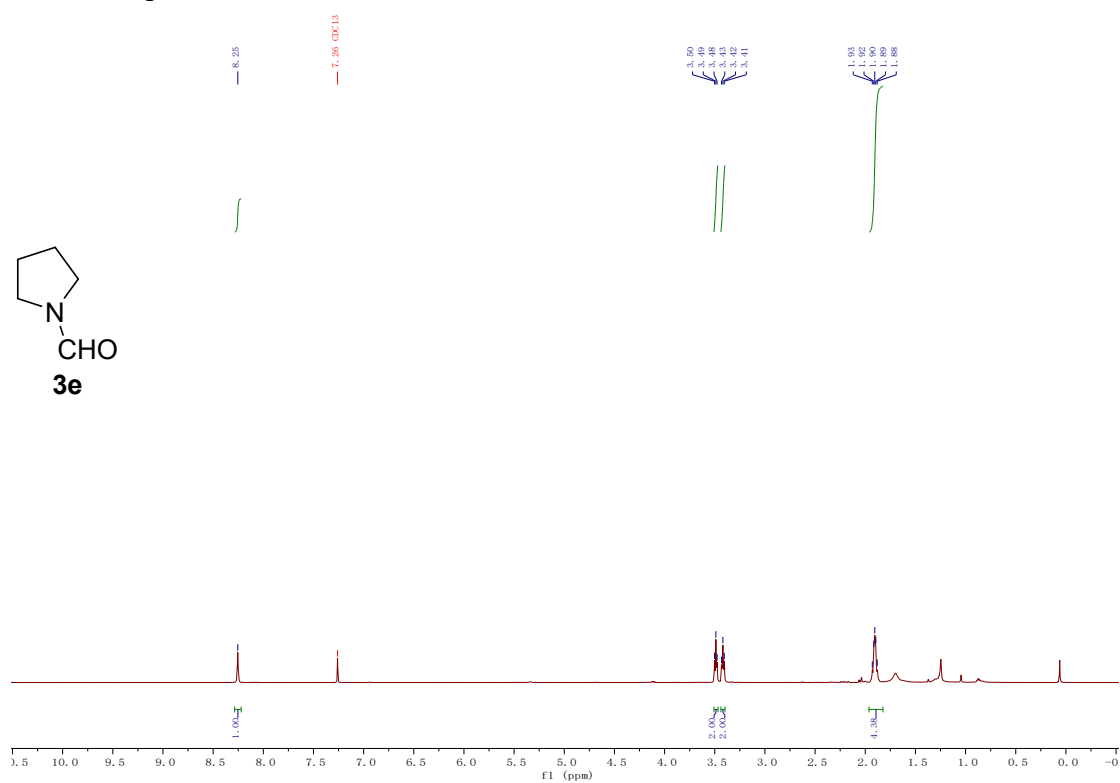
^1H NMR spectrum of **3c** in CDCl_3 at 500 MHz



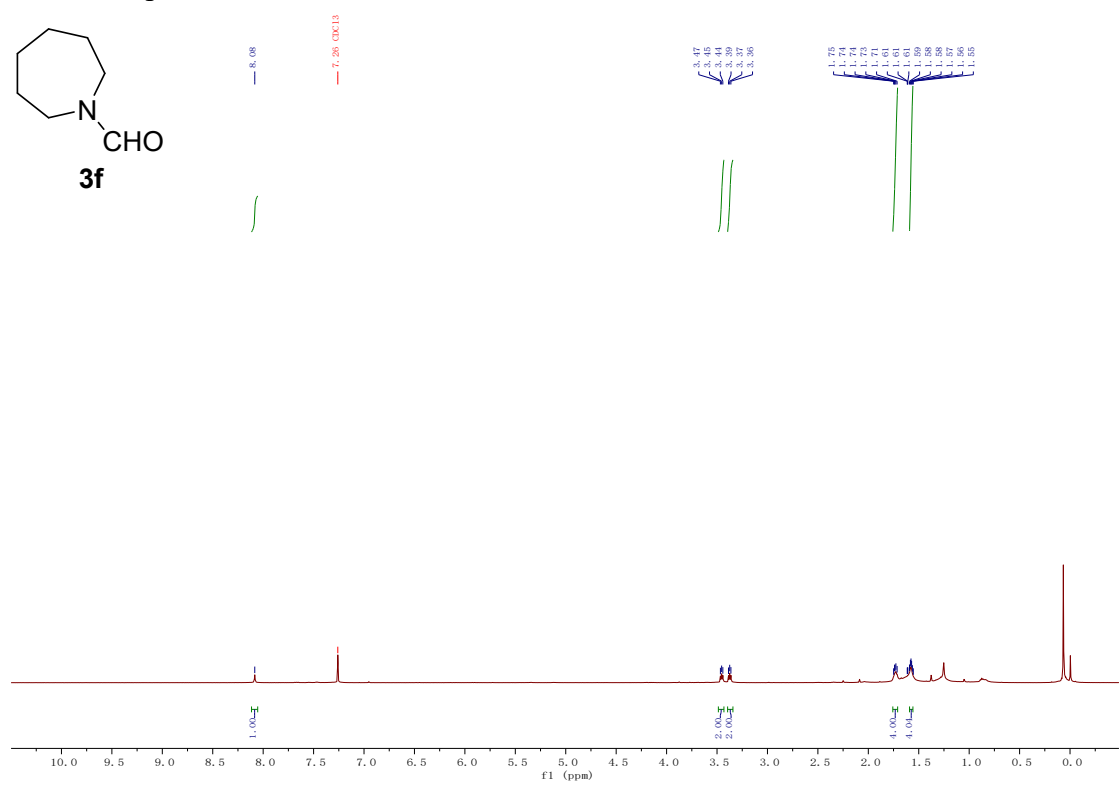
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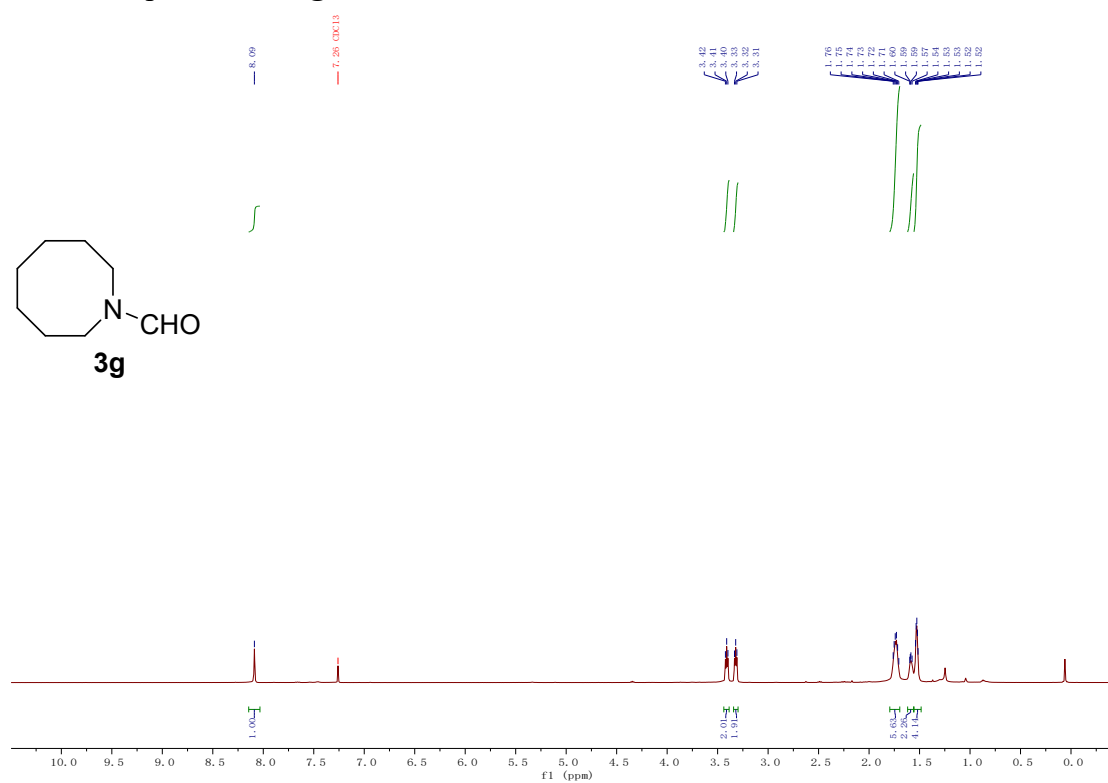
^1H NMR spectrum of **3e** in CDCl_3 at 500 MHz



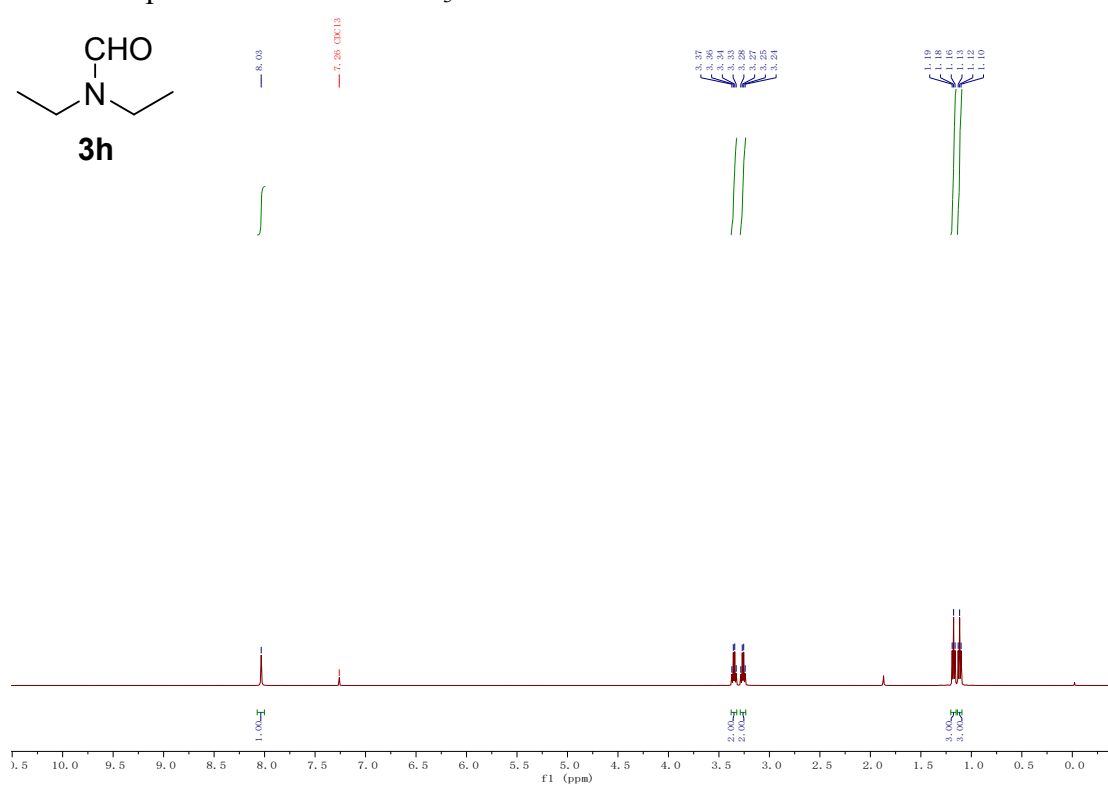
^1H NMR spectrum of **3f** in CDCl_3 at 500 MHz



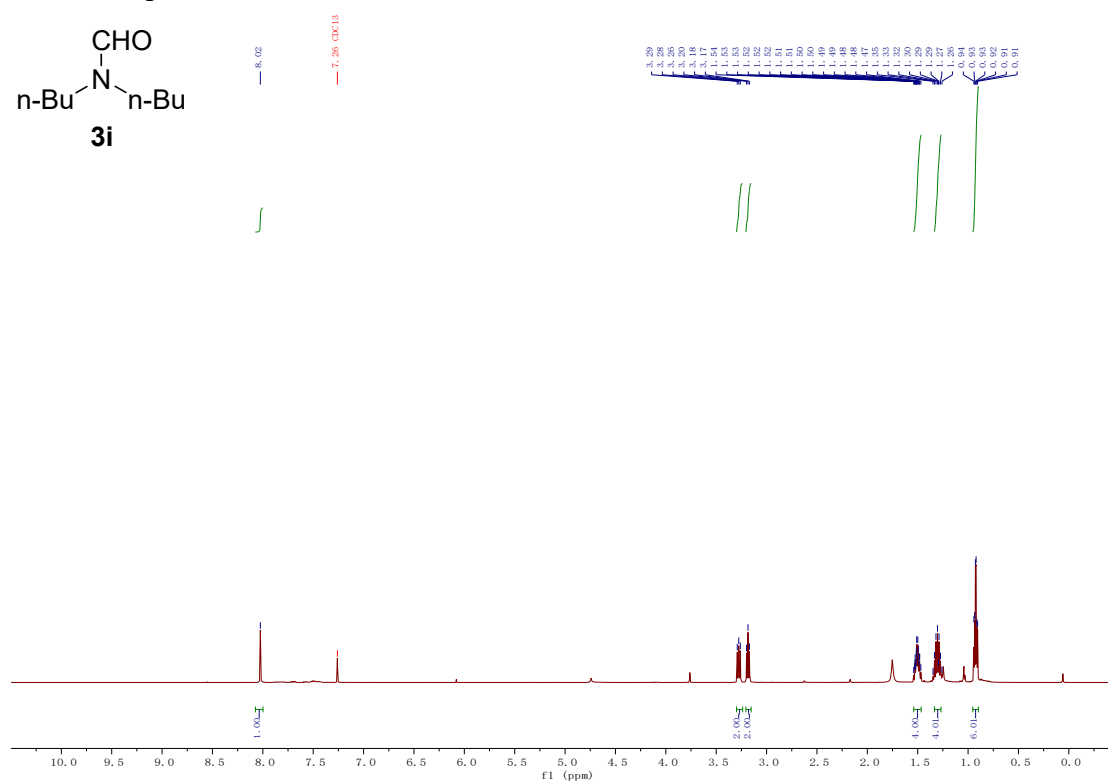
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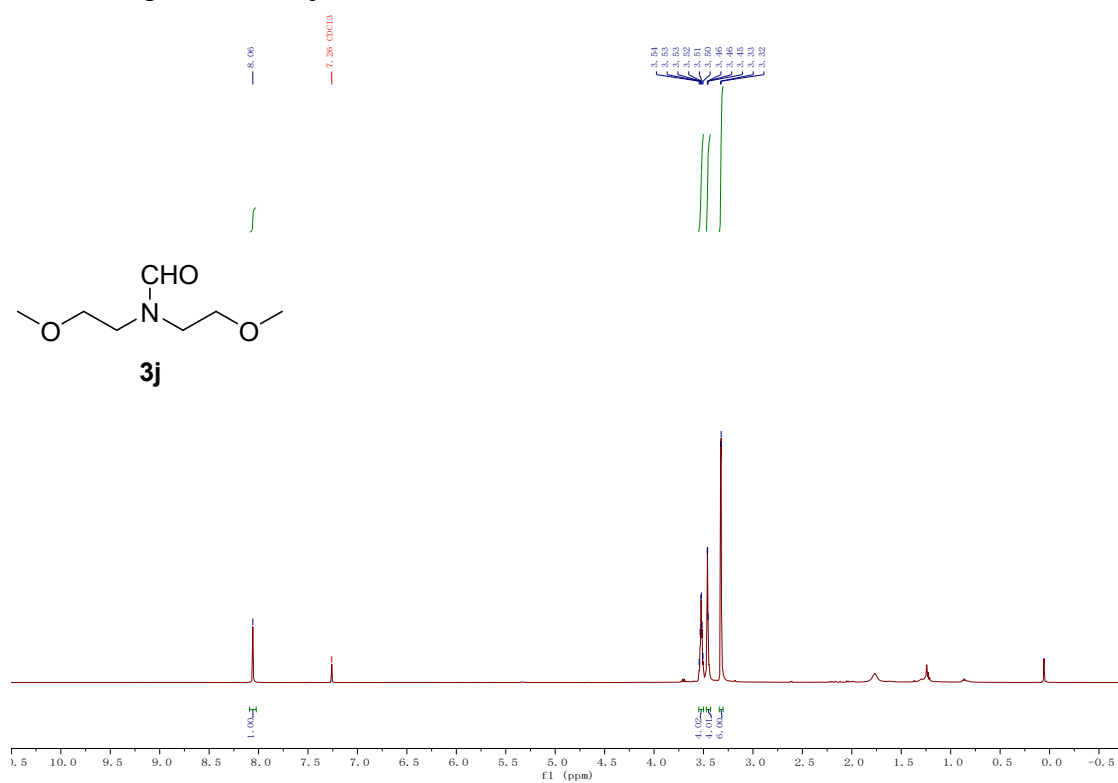
^1H NMR spectrum of **3h** in CDCl_3 at 500 MHz



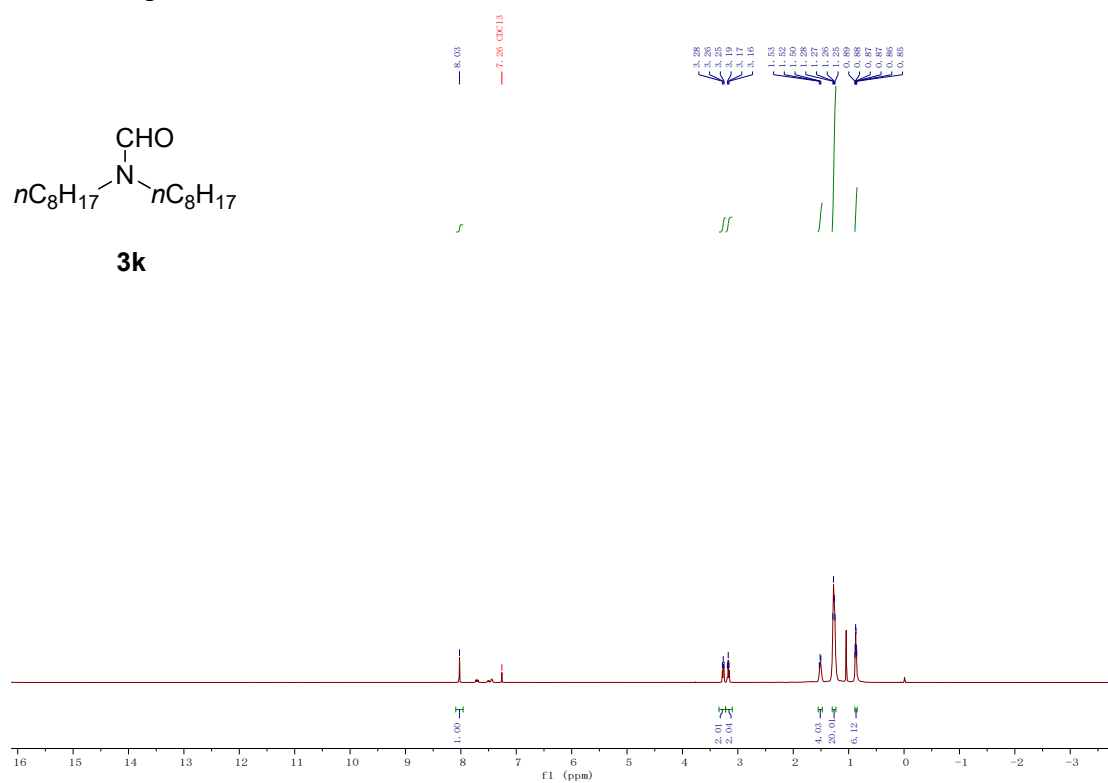
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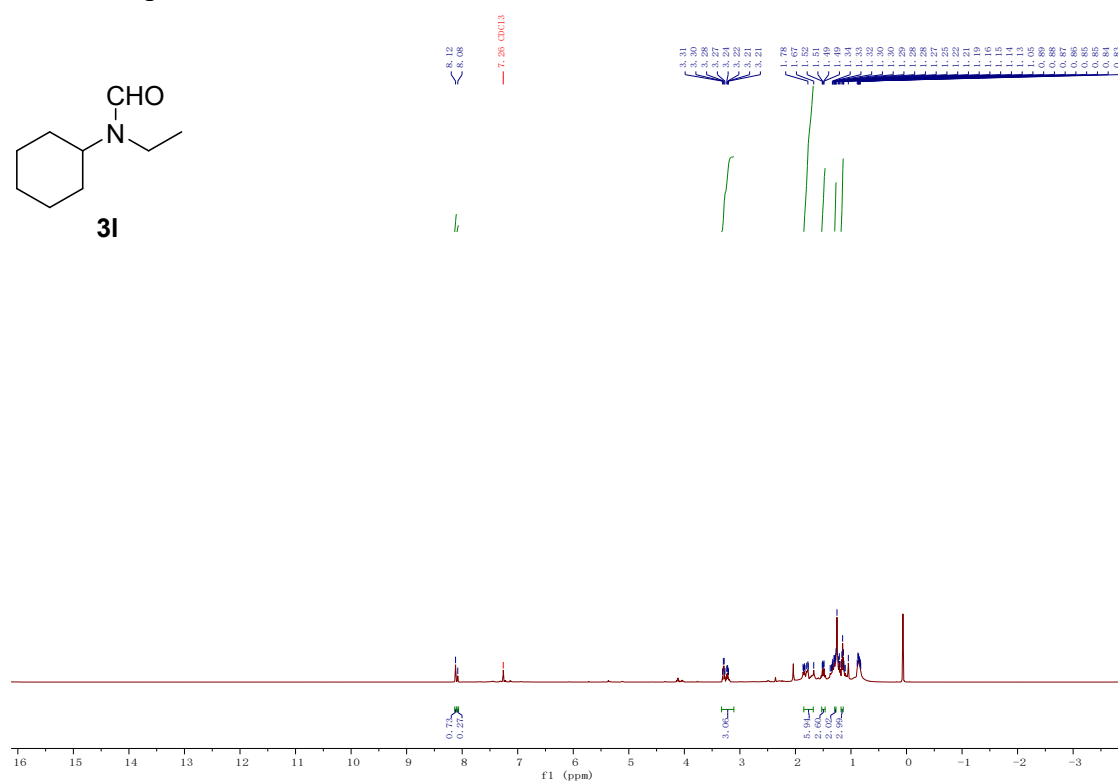
^1H NMR spectrum of **3j** in CDCl_3 at 500 MHz



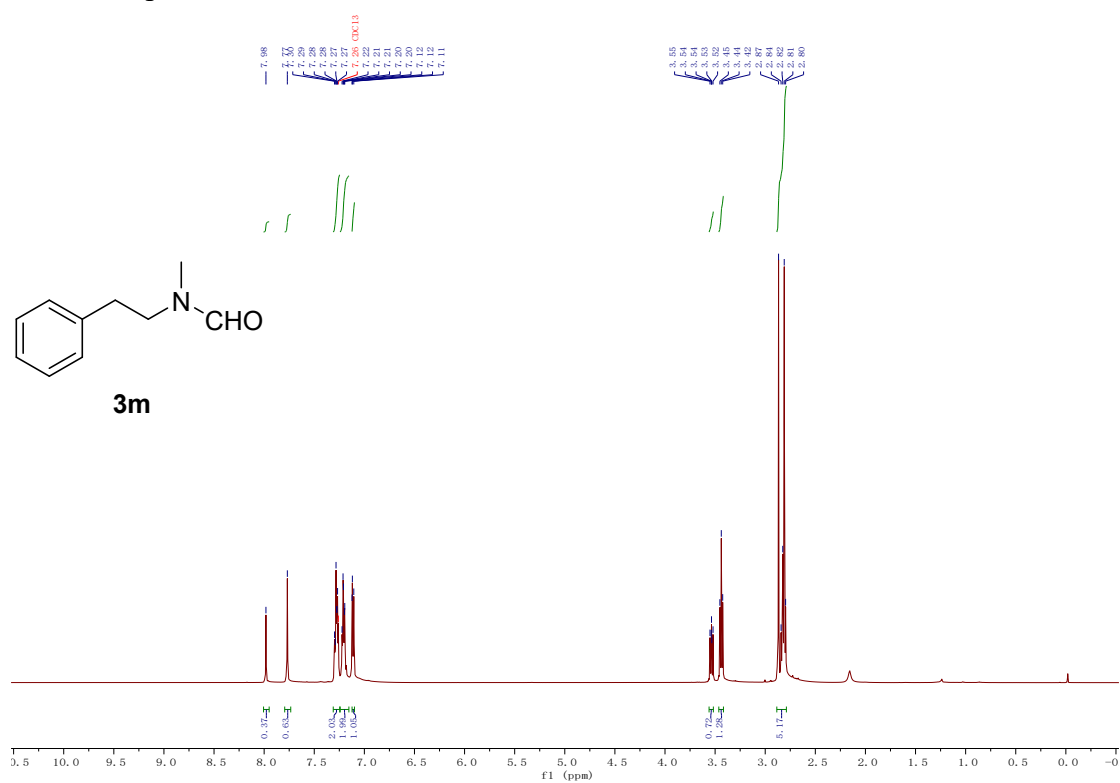
^1H NMR spectrum of **3k** in CDCl_3 at 500 MHz



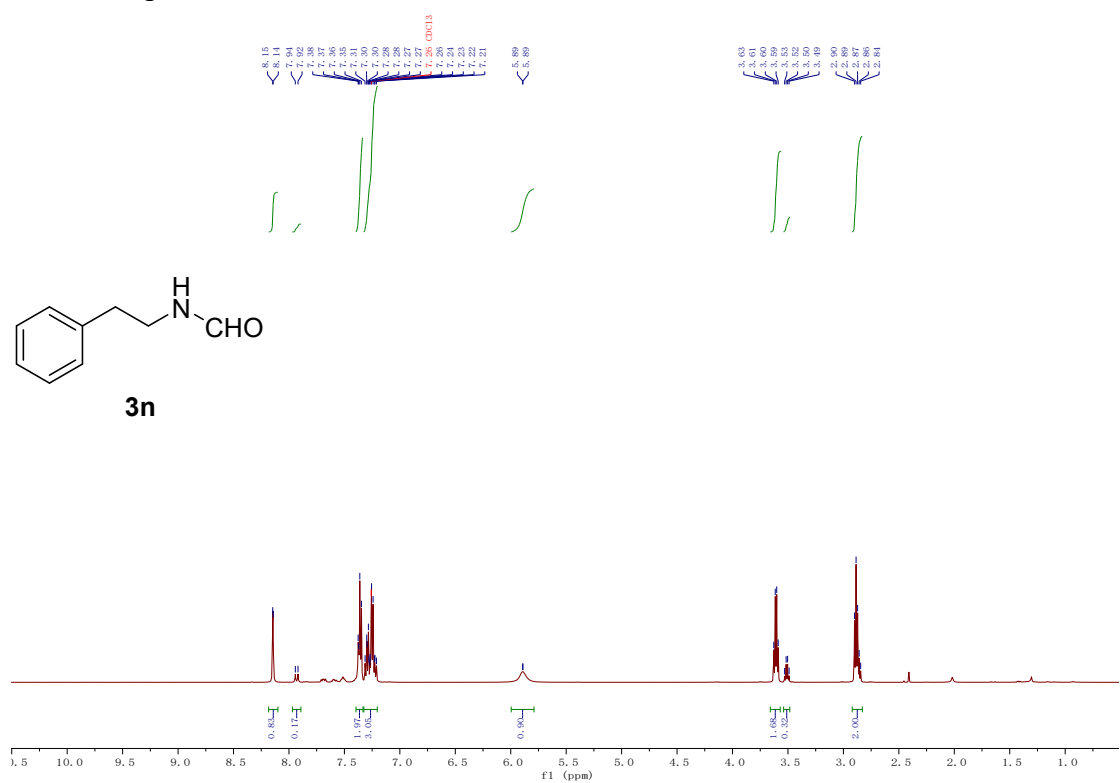
^1H NMR spectrum of **3l** in CDCl_3 at 500 MHz



^1H NMR spectrum of **3m** in CDCl_3 at 500 MHz



^1H NMR spectrum of **3n** in CDCl_3 at 500 MHz



Cc1ccc(cc1)CCNC=O

3o

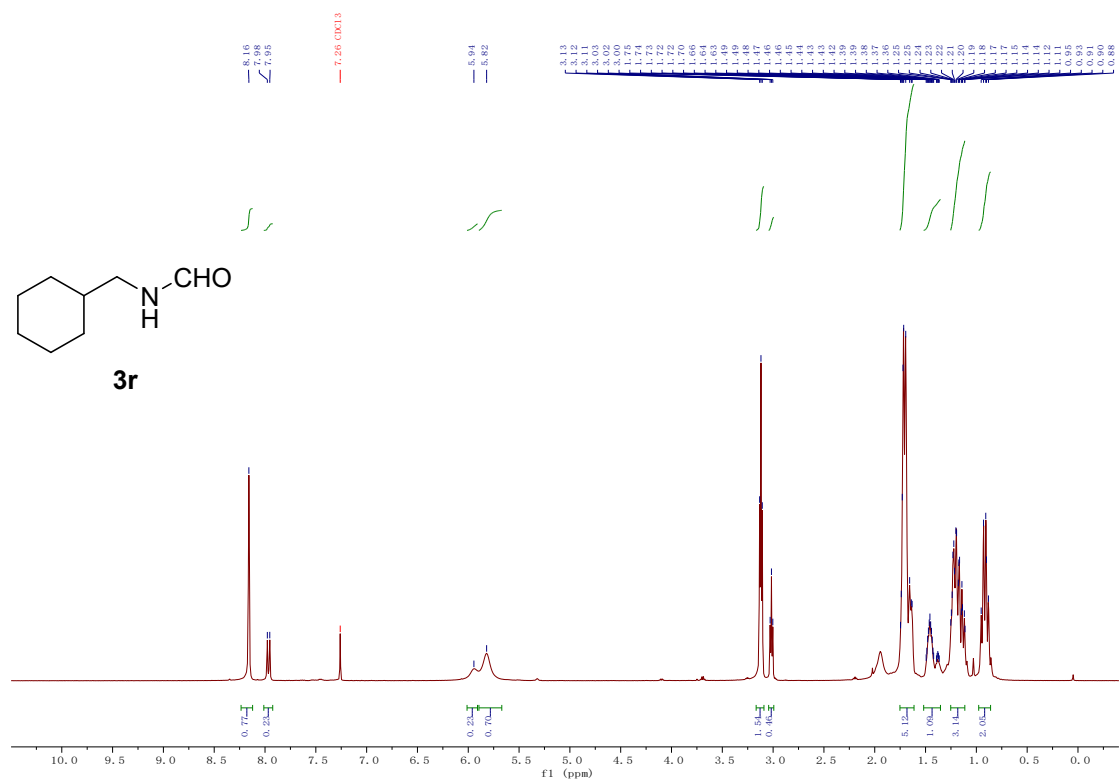
¹H NMR spectrum (CDCl₃) of compound **3o**. The spectrum shows peaks at 7.75 (d, 2H), 7.25 (d, 2H), 6.00 (s, 1H), 3.55 (q, 2H), 2.95 (q, 2H), and 2.35 (s, 3H). Integration values are shown below the peaks: 0.82, 0.18, 4.01, 0.96, 1.06, 0.34, 2.01, and 3.00. A chemical structure of **3o** is shown in the top left.

C1CCCCC1NC=O

3q

8.12
7.26 CDCl₃
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7.12
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6.99
6.98
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6.14
6.13
6.12

53



¹H NMR spectrum of **3t** in CDCl₃ at 500 MHz

