Supplementary Information

Electricity Mediated [3+2]-cycloaddition of N-sulfonylcyclopropanes with Olefins via N-centered Radical Intermediates: Access to Cyclopentane Analogs

Debarshi Saha,^[a] Irshad Maajid Taily,^[a] Nakshatra Banerjee,^[a] and Prabal Banerjee^{*[a]}

Email: prabal@iitrpr.ac.in

Department of Chemistry

Indian Institute of Technology

Ropar, Punjab

INDEX

1.	General procedure for preparation of <i>N</i> -sulfonylcyclopropylamines. (GP1)	S4
2.	General procedure for preparation of Ethyl/methyl cinnamates. (GP2)	.S4
3.	Representative procedure for electrochemical [3+2] cycloaddition reaction. (RP1)	.S4
4.	Optimization studies	-S6
5.	Mechanistic Studies	-S7
	a. Radical scavenging experiment	.S6
	b. Unsuccessful experiment	S6
	c. Divided cell experiment	S7
	d. Time based electrolysis	S7
	e. Intermediate trap experiment	S 8
	f. Cyclic Voltammetry	S9
6.	Characterization data of Starting materials	11
7.	Characterization data of Products, derivatives and Michael adduct	17
8.	Mechanism for formation of 3ak S	18
9.	Alternate mechanism for the protocol	18
10	NOESY Experiment	\$19
11	. Single Crystal data of 3ag and 3al	\$26
12	. ReferencesS	26
13	. ¹ H NMR of the Starting materials	31
14	. NMR and HRMS of the Products	65
15	. NMR and HRMS of the derivatives	69
16	NMR and HRMS of the Michael adduct	72
17	. NMR of the IVS	73
18	. HRMS of intermediate IIIS	73
19	. HRMS of intermediate FS	574

General Remarks

Electrochemical reactions were performed under air at room temperature using IKA Electrasyn 2.0 and DC power supply procured from Keysight technologies limited (Model: E36312A) and GW Instek limited (Model: GPP 3323). Acetonitrile and other solvents were obtained from Merck Life science Private Limited and were used directly without further purification and drying. Reagents like acrylates (spectrochem), acrylonitrile (spectrochem), methylvinyl ketone (TCI), styrene (sigma), diphenyl acrylamides, cyclopropylamine (spectrochem), sulfonyl chlorides (GLR innovations) were procured from various commercially available sources. Electrodes were commercially available from IKA. Cyclic voltammetry experiments were also performed on the CHI1110C instrument. The control of the measurement instrument, the acquisition, and processing of the cyclic voltammetric data was processed with the CHinstruments electrochemical analyzer. Yields refer to isolated compounds, estimated to be >95% pure as determined by ¹H NMR. Thin-layer chromatography was performed on Merck precoated silica gel 60 F254 aluminum sheets with detection under UV light at 254 nm and charring with the *p*-anisaldehyde solution. Chromatographic purifications were performed with silica gel (230-400 mesh) and melting points were taken on Stuart digital melting point apparatus. Nuclear magnetic resonance (NMR) spectroscopy was performed using JEOL 400 MHz and HRMS was recorded on Waters Xevo G2-XS (Q-TOF) and Agilent Technologies 6530 Accurate- Mass Q-TOF LC/MS. If not otherwise specified, chemical shifts (δ) are provided in ppm and coupling constants are absolute values and are expressed in Hertz. Unless otherwise specified, all the ¹H NMR and ¹³C NMR spectra were recorded in CDCl₃. Chemical shifts of ¹H and ¹³C NMR spectra are expressed in parts per million (ppm) and coupling constant values were given in absolute. The description of the signals includes the following: s = singlet, d = doublet, dd = doubletof doublet, t = triplet, dt = doublet of triplet, q = quartet, br = broad, and m = multiplet.

A) General procedure for preparation of N-sulfonylcyclopropylamines. (GP1)¹



To an oven-dried round bottom flask, cyclopropylamine (1.1 equiv.) was added dichloromethane and the reaction was cooled to 0°C under a nitrogen atmosphere. Further, triethylamine (Et₃N) (1.1 equiv.) was added dropwise, followed by the corresponding arylsulfonyl chloride (1.0 equiv.). The reaction was then brought to room temperature and stirred for 16 h. The reaction solution was quenched with H_2O (10 mL) and further extracted with ethyl acetate (2 X 20 mL). The combined organic layers were washed with brine solution (10 mL) dried over Na₂SO₄ and were further concentrated in vacuo. The crude obtained was purified using silica gel (230-400) column chromatography, where the desired compound was eluted in 20% ethyl acetate in hexane in 80-90% yield.

B) General procedure for preparation of Ethyl/methyl cinnamates. (GP2)²



Triethyl phosphonoacetate (1.1 equiv), DBU (0.035 equiv), and finely powdered anhydrous K_2CO_3 (2.0 equiv) were taken in an oven-dried round-bottom flask, and ArCHO (1.0 equiv) was added. The resulting mixture was stirred at room temperature for 4 h. Ethyl acetate was added to the crude mixture, and the solid was filtered off. The solid was washed with ethyl acetate, and the combined filtrate was concentrated in vacuo. The resulting oil was purified using silica gel (mesh size 230-400) column chromatography to give the corresponding alkene (84% yield) (E/Z = 99:1).

C) Representative procedure for electrochemical [3+2] cycloaddition reaction. (RP1)



An Electrasyn vial/ test tube was equipped with a magnetic stir bar and was added corresponding *N*-sulfonyl cyclopropane (1.0 equiv.), alkene (3.0 equiv.), ferrocene (0.5 equiv.), tetrabutylammonium tetrafluoroborate (Bu_4NBF_4) (2.0 equiv.), potassium carbonate (2.0 equiv.) in dichloroethane solvent (0.04 M). Further, the vial/test tube was purged with an argon balloon for 1-2 minutes and the solution was electrolyzed with carbon anode and nickel cathode at a constant current of 1.5 mA (voltage range of 2.8 V-3.2 V) for 2-6 h at room temperature (25° - 30° C). Upon completion (as monitored by TLC), evaporation of solvent gave the crude product. The crude obtained was then purified by silica gel column chromatography (230-400 mesh) and the desired product was eluted at 30% ethyl acetate in hexane.

D) Optimization studies



In an oven-dried undivided reaction flask (10 mL) equipped with a stir bar was added N-cyclopropyl-4methylbenzenesulfonamide (1.0 equiv.), methyl acrylate (3.0 equiv.), base (2.0 equiv.), mediator (0.5 equiv.), electrolyte (2.0 equiv.) and solvent (5 mL). The solution was electrolyzed at a constant current at room temperature under an argon atmosphere for 2-12 h. The solvent was evaporated in vacuo and the crude mixture was purified by silica gel column using chromatography 20-30% ethyl acetate in hexane methyl (1*R*,2*R*)-2-((4to get methylphenyl)sulfonamido)cyclopentane-1-carboxylate (3aa) in 59-85% isolated yields.

S.No	Electrode	Supporting	Mediator	Base	Current	solvent	Time (h)	% Yield
		Electrolyte						of 3aa
1.	C(+)/C(-)	Bu ₄ NBF ₄	-	-	1.5	CH ₃ CN	3	N.R.
2.	C(+)/Ni(-)	Bu ₄ NBF ₄	-	K ₂ CO ₃	1.5	CH ₃ CN	3.5	55 (3aa')
3.	C(+)/Ni(-)	Bu ₄ NBF ₄	-	K ₂ CO ₃	1.5	HFIP	3	N.R.
4.	C(+)/Ni(-)	Bu ₄ NPF ₆	-	K ₂ CO ₃	1.5	CH ₃ CN	3	c.m.
5.	C(+)/C(-)	Bu ₄ NBF ₄	-	K ₂ CO ₃	1.5	CH ₃ CN	3	N.R.
6.	C(+)/Ni(-)	Bu ₄ NOAc	-	K ₂ CO ₃	1.5	CH ₃ CN	3	N.R.
7.	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	1.5	CH ₃ CN: H ₂ O	4	34
8.	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	1.5	CH ₃ CN: H ₂ O	8	28
9.	C(+)/Ni(-)	Bu ₄ NPF ₆	Cp ₂ Fe	K ₂ CO ₃	1.5	CH ₃ CN: H ₂ O	4	12
10.	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	Na ₂ CO ₃	1.5	CH ₃ CN: H ₂ O	4	20
11.	C(+)/Cu(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	1.5	CH ₃ CN: H ₂ O	5	N.R.
12.ª	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	1.5	CH ₃ CN: H ₂ O	4	N.R.
13.	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	Et ₃ N	1.5	CH ₃ CN	5	N.R.
14.	C(+)/Ni(-)	Bu ₄ NBF ₄	-	DBU	1.5	CH ₃ CN	5	N.R.
15.	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	1.5	DCE	5	62
16.	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	Na ₂ CO ₃	1.5	DCE	5	57
17. ^b	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	1.5	DCE	5	43
18.	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	0.5	DCE	5	38
19.	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	2.5-3.0	DCE	5	47
20.	C(+)/Ni(-)	Et ₄ NOTs	Cp ₂ Fe	K ₂ CO ₃	1.5	DCE	5	N.R
21.	C(+)/Ni(-)	LiOTf	Cp ₂ Fe	K ₂ CO ₃	1.5	DCE	5	N.R
22.	C(+)/Ni(-)	Bu ₄ NPF ₆	Cp ₂ Fe	K ₂ CO ₃	1.5	DCE	5	N.R
23.	C(+)/Pt(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	1.5	DCE	5	N.R
24.	C(+)/Cu(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	1.5	DCE	5	N.R

25.	Pt(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	1.5	DCE	5	N.R
26.	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	1.5	DMSO	5	N.R
27.	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	1.5	DMF	5	N.R
28.	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	1.5	HFIP	5	N.R
29.	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	K ₂ CO ₃	1.5	DCE: H ₂ O	4	42
30.	C(+)/Ni(-)	Bu ₄ NBF ₄	Cp ₂ Fe	-	1.5	DCE	5	22

All the reactions are performed at room temperature of 25 °C-30 °C aReaction performed without electricity, ^b 0.1M instead of 0.04M.

E) Mechanistic Studies

a) Radical scavenging experiment



An Electrasyn vial/ test tube was equipped with a magnetic stir bar and was added corresponding *N*-sulfonyl cyclopropane (1.0 equiv.), alkene (3.0 equiv.), ferrocene (0.5 equiv.), tetrabutylammonium tetrafluoroborate (Bu_4NBF_4) (2.0 equiv.), potassium carbonate (2.0 equiv.) in dichloroethane solvent (0.04M). Further, the vial/test tube was purged with an argon balloon for 1-2 minutes and the solution was electrolyzed with carbon anode and nickel cathode at a constant current of 1.5 mA (voltage range of 2.8 V-3.2 V) for 2-6 h at room temperature (25° - 30° C). The progress of the reaction was monitored by TLC, which shows that starting material remain unconsumed. This infers that employing TEMPO completely shut down the reaction indicating the possibility of a radical path for the reaction.

b) Unsuccessful experiment



An Electrasyn vial/ test tube was equipped with a magnetic stir bar and was added corresponding *N*-acetyl cyclopropane (1.0 equiv.), alkene (3.0 equiv.), ferrocene (0.5 equiv.), tetrabutylammonium tetrafluoroborate (Bu_4NBF_4) (2.0 equiv.), potassium carbonate (2.0 equiv.) in dichloroethane solvent. Further, the vial/test tube was purged with an argon balloon for 1-2 minutes and the solution was electrolyzed with carbon anode and nickel cathode at a constant current of 1.5 mA (voltage range of 3.0 V-4.0 V) for 2-6 h at room temperature (25° - 30° C). The progress of the reaction was monitored by TLC, which shows that starting material remain unconsumed. This infers that replacement of tosyl group with acetyl group proved detrimental for the reaction.

c) Divided cell experiment



Case 1: In anodic chamber: A divided cell was equipped with two magnetic stir bars in anodic and cathodic chamber respectively. Further, the anodic chamber was filled with corresponding *N*-tosyl cyclopropane (1.0 equiv.), alkene (3.0 equiv.), ferrocene (0.5 equiv.), tetrabutylammonium tetrafluoroborate (Bu_4NBF_4) (2.0 equiv.), potassium carbonate (2.0 equiv.) in dichloroethane solvent. The cathodic chamber was filled with supporting electrolyte solution and further both the chambers were purged with an argon for 1-2 minutes and the solution was electrolyzed with carbon anode (in anodic chamber) and nickel cathode (in cathodic chamber) at a constant current of +1.5 mA (voltage range of +2.8 V to +3.4 V) for 2-6 h at room temperature (25°-30°C). The progress of the reaction was monitored by TLC, which shows that starting material remain unconsumed. This infers that the cathodic reduction is a key factor in the reaction.

Case 2: In cathodic chamber: A divided cell was equipped with two magnetic stir bars in anodic and cathodic chamber respectively. Further, the cathodic chamber was filled with corresponding *N*-tosyl cyclopropane (1.0 equiv.), alkene (3.0 equiv.), ferrocene (0.5 equiv.), tetrabutylammonium tetrafluoroborate (Bu_4NBF_4) (2.0 equiv.), potassium carbonate (2.0 equiv.) in dichloroethane solvent. The anodic chamber was filled with supporting electrolyte solution and further both the chambers were purged with an argon for 1-2 minutes and the solution was electrolyzed with carbon anode (in anodic chamber) and nickel cathode (in cathodic chamber) at a constant current of -1.5 mA (voltage range of -2.8 V to -3.4 V) for 2-6 h at room temperature (25°-30°C). The progress of the reaction was monitored by TLC, which shows that starting material remain unconsumed. This infers that the reaction gets initiated by the oxidation of cyclopropyl amines.

d) Time based electrolysis



An Electrasyn vial/ test tube was equipped with a magnetic stir bar and was added corresponding *N*-sulfonyl cyclopropane (1.0 equiv.), alkene (3.0 equiv.), ferrocene (0.5 equiv.), tetrabutylammonium tetrafluoroborate (Bu_4NBF_4) (2.0 equiv.), potassium carbonate (2.0 equiv.) in dichloroethane solvent (0.04M). Further, the vial/test tube was purged with an argon for 1-2 minutes and the solution was electrolyzed with carbon anode and nickel cathode at a constant current of 1.5 mA (voltage range of 2.8 V-3.2 V) for 1 h at room temperature (25° - $30^{\circ}C$). The progress of the reaction was monitored by TLC, which shows that starting material remain unconsumed with trace amount of product formation After, that the reaction was stirred overnight without electricity. Again, the progress of the reaction was monitored by TLC, which

indicated that a similar TLC was observed. Further, the reaction mixture was purified through column chromatography and approx. 70% of the starting material was recovered. This indicates that the chain propagation step is absent in this case.

e) Intermediate detection experiment



An Electrasyn vial/ test tube was equipped with a magnetic stir bar and was added corresponding *N*-sulfonyl cyclopropane (1.0 equiv.), (3-bromoprop-1-en-2-yl)benzene (3.0 equiv.) (**2o**), ferrocene (0.5 equiv.), tetrabutylammonium tetrafluoroborate (Bu₄NBF₄) (2.0 equiv.), potassium carbonate (2.0 equiv.) in dichloroethane solvent (0.04M). Further, the vial/test tube was purged with an argon balloon for 1-2 minutes and the solution was electrolyzed with carbon anode and nickel cathode at a constant current of 1.5 mA (voltage range of 2.8 V-3.2 V) for 5 h at room temperature (25°-30°C). The progress of the reaction was monitored by TLC, which shows that starting material gets consumed. Further, the reaction mixture was passed through the silica gel column (230-400) at 40% ethylacetate in hexane to remove the supporting electrolyte. Next, the reaction mixture obtained after passing through column was submitted for HRMS. HRMS studies demonstrate the presence of intermediate III and further the purification of reaction mixture delivered the IV (NMR attached at the end). These studies indicate that the reaction follows an anionic pathway.

c) Cyclic Voltammetry

Cyclic voltammograms were collected with the three-electrode CHI1110C instrument. The control of the measurement instrument, the acquisition, and processing of the cyclic voltammetric data were performed with the CH instruments electrochemical analyzer. Measurements employed a glassy carbon working electrode, platinum wire counter electrode, and a 3 M KCl silver-silver chloride reference electrode. Samples were prepared with 0.05 mmol of substrate **1a**, 0.05 mmol of substrate **2a**, 0.1 mmol of K₂CO₃, 0.01 mmol of Cp₂Fe dissolved in 5 mL acetonitrile containing 0.1 M Bu₄NBF₄. The scan rate applied was 0.1 V/s. the maximum current (Ip) and potential (Ep) of each substrate was obtained using Origin. All the CV experiments were carried out in nitrogen atmosphere and demonstrated as follows: (a) Blank (black) (b) **2a** (Red) (c) **1a** (Blue) (d) Cp₂Fe (Brown) (e) **1a** + Cp₂Fe (Dark blue) (f) **1a** + Cp₂Fe + K₂CO₃ (Purple) (g) **1a** + **2a** (Pink) (h) **1a** + **2a** + K₂CO₃ (Green) (i) **1a** + **2a** + K₂CO₃ (Violet)



Figure S1: Cyclic Voltammetry experiment

Conclusion: Cyclic Voltammetry experiments indicate the importance of ferrocene and base for the cycloaddition reaction. As demonstrated by the above experiments the introduction of ferrocene enhances the catalytic current making the process more feasible to take place. Although as expected the potential does not shift to the lower region but the electrocatalytic effect led to a huge improvement in the reaction profile. A similar result was observed when the base was added to the same reaction mixture. In conclusion, ferrocene plays a significant role in the generation of *N*-center radical which results in the cycloaddition reaction with acrylate.

F) Characterization data of Starting materials

a) *N*-cyclopropyl-4-methylbenzenesulfonamide $(1a)^{1}$



Prepared according to **GP1**. ¹**H NMR** (400 MHz, **CDCl**₃): δ 7.77 (d, J = 8.3 Hz, 1H), 7.30 (d, J = 8.2 Hz, 1H), 5.08 (s, 1H), 2.41 (s, 1H), 2.24 – 2.17 (m, 1H), 0.60 – 0.52 (m, 1H).

b) N-cyclopropylbenzenesulfonamide (1b)¹



Prepared according to **GP1**. ¹**H NMR (400 MHz, CDCl₃):** δ 7.93 – 7.87 (m, 1H), 7.59 (ddd, *J* = 6.4, 4.2, 1.3 Hz, 1H), 7.55 – 7.49 (m, 1H), 4.94 (s, 1H), 2.27 – 2.20 (m, 1H), 0.62 – 0.56 (m, 1H).

c) *N*-cyclopropylnaphthalene-2-sulfonamide $(1c)^{1}$



Prepared according to **GP1**. ¹**H NMR (400 MHz, CDCl₃):** δ 8.49 (s, 1H), 7.97 (d, *J* = 8.5 Hz, 1H), 7.91 (d, *J* = 8.0 Hz, 1H), 7.86 (dd, *J* = 8.7, 1.7 Hz, 1H), 7.68 – 7.58 (m, 1H), 5.03 (s, 1H), 2.29 – 2.22 (m, 1H), 0.64 – 0.55 (m, 1H).

d) N-cyclopropyl-4-(trifluoromethyl)benzenesulfonamide (1d)



Prepared according to **GP1**. ¹**H NMR** (**400 MHz, CDCl**₃): δ 8.03 (d, J = 8.4 Hz, 1H), 7.79 (d, J = 8.3 Hz, 1H), 5.12 (s, 1H), 2.30 – 2.23 (m, 1H), 0.64 – 0.59 (m, 1H).

e) N-cyclopropyl-4-methoxybenzenesulfonamide (1e)¹



Prepared according to **GP1**. ¹**H NMR (400 MHz, CDCl₃):** δ 7.84 – 7.80 (m, 1H), 7.00 – 6.95 (m, 1H), 4.82 (s, 1H), 3.86 (s, 1H), 2.26 – 2.19 (m, 1H), 0.59 – 0.54 (m, 1H).

f) 4-bromo-*N*-cyclopropylbenzenesulfonamide $(1f)^{1}$



Prepared according to **GP1**. ¹**H NMR (400 MHz, CDCl₃)**: δ 7.78 – 7.74 (m, 1H), 7.68 – 7.64 (m, 1H), 4.86 (s, 1H), 2.28 – 2.22 (m, 1H), 0.60 (d, *J* = 5.0 Hz, 1H).

g) Ethyl cinnamate (21)²



Prepared according to **GP2**. ¹**H NMR** (**400 MHz, CDCl₃**): δ 7.66 (t, *J* = 11.3 Hz, 1H), 7.54 – 7.48 (m, 1H), 7.40 – 7.34 (m, 1H), 6.43 (d, *J* = 16.0 Hz, 1H), 4.25 (q, *J* = 7.1 Hz, 1H), 1.33 (dd, *J* = 9.5, 4.8 Hz, 1H).

h) (E)-ethyl 3-(4-fluorophenyl)acrylate (2m)²



Prepared according to **GP2**. ¹**H NMR** (**400 MHz, CDCl₃**): δ 7.62 (d, *J* = 16.0 Hz, 1H), 7.52 – 7.45 (m, 1H), 7.05 (t, *J* = 8.6 Hz, 1H), 6.34 (d, *J* = 16.0 Hz, 1H), 4.24 (q, *J* = 7.2 Hz, 1H), 1.31 (t, *J* = 7.1 Hz, 1H).

i) (E)-ethyl 3-(4-methoxyphenyl)acrylate (2n)²



Prepared according to **GP2**. ¹**H NMR** (**400 MHz, CDCl**₃): δ 7.63 (d, *J* = 16.0 Hz, 1H), 7.49 – 7.43 (m, 1H), 6.91 – 6.86 (m, 1H), 6.29 (d, *J* = 16.0 Hz, 1H), 4.23 (q, *J* = 7.1 Hz, 1H), 3.82 (s, 1H), 1.34 – 1.29 (m, 1H).

G) Characterization data of Products

a) Ethyl (1*R*,2*R*)-2-((4-methylphenyl)sulfonamido)cyclopentane-1-carboxylate (3aa)



Prepared according to **RP1**. **1a** (0.050 g, 0.236 mmol), **2a** (0.071 g, 0.708 mmol), **3aa** (0.051 g, 72%). **Reaction time:** 5 h. **Yield:** 72%. **Nature:** oily liquid. ¹**H NMR (400 MHz, CDCl₃):** δ 7.74 (d, *J* = 8.3 Hz, 2H), 7.29 (d, *J* = 8.5 Hz, 2H), 4.80 (d, *J* = 6.0 Hz, 1H), 4.02 (q, *J* = 7.1 Hz, 2H), 3.72-3.64 (m, 1H), 2.61 (dd, *J* = 16.7, 7.7 Hz, 1H), 2.41 (s, 3H), 2.03-1.91 (m, 2H), 1.80-1.63 (m, 2H), 1.50-1.40 (m, 2H), 1.18 (t, *J* = 7.2 Hz, 3H).¹³**C NMR (101 MHz, CDCl**₃): δ 174.2, 143.5, 137.2, 129.7, 127.4, 60.9, 57.8, 50.9, 33.6, 28.0, 22.9, 21.6, 14.2. **HRMS:** [M + H]⁺ Calculated for C₁₅H₂₂NO₄S 312.1270, found 312.1338.



Prepared according to **RP1**. **1a** (0.050 g, 0.236 mmol), **2a** (0.061 g, 0.708 mmol) **3ad** (0.048 g, 67%). **Reaction time:** 5 h. **Yield:** 72%. **Nature:** oily liquid. ¹**H NMR (400 MHz, CDCl₃):** δ 7.73 (d, *J* = 8.3 Hz, 2H), 7.29 (d, *J* = 8.1 Hz, 2H), 4.75 (d, *J* = 5.9 Hz, 1H), 3.73 – 3.65 (m, 1H), 3.57 (s, 3H), 2.63 (dd, *J* = 16.7, 7.8 Hz, 1H), 2.41 (s, 3H), 2.03-1.92 (m, 3H), 1.81 – 1.68 (m, 2H), 1.51 – 1.41 (m, 1H).¹³**C NMR (101 MHz, CDCl₃)**: δ 174.6, 143.6, 137.0, 130.0, 129.8, 127.4, 57.9, 52.0, 50.7, 45.1, 34.4, 33.5, 28.0, 22.8, 21.6. **HRMS:** [M + H]⁺ Calculated for C₁₄H₂₀NO₄S 298.1111, found 298.1113.

c) *tert*-butyl (1*R*,2*R*)-2-((4-methylphenyl)sulfonamido)cyclopentane-1-carboxylate (3ae)



Prepared according to **RP1**. **1a** (0.040 g, 0.236 mmol), **2a** (0.090 g, 0.708 mmol), **3ae** (0.041 g, 64%). **Reaction time:** 5 h. **Yield:** 64%. **Nature:** oily liquid. ¹**H NMR (400 MHz, CDCl₃)**: δ 7.74 (d, *J* = 8.3 Hz, 2H), 7.28 (d, *J* = 8.5 Hz, 2H), 4.97 (d, *J* = 5.7 Hz, 1H), 3.66-3.58 (m, 1H), 2.52 (dd, *J* = 16.5, 7.5 Hz, 1H), 2.40 (s, 3H), 1.98-1.84 (m, 2H), 1.68-1.55 (m, 3H), 1.49-1.39 (m, 1H), 1.36 (s, 9H).¹³**C NMR (101 MHz, CDCl₃)**: δ 173.4, 143.5, 137.2, 129.8, 127.4, 81.0, 57.8, 51.7, 33.6, 28.0, 27.8, 22.9, 21.6. **HRMS:** [M + Na]⁺ Calculated for C₁₇H₂₅NO₄SNa 362.1402, found 362.1402.

d) Benzyl (1*R*,2*R*)-2-((4-methylphenyl)sulfonamido)cyclopentane-1-carboxylate (3af)



Prepared according to **RP1**. **1a** (0.050 g, 0.236 mmol), **2a** (0.115 g, 0.708 mmol), **3af** (0.055 g, 63%). **Reaction time:** 5 h. **Yield:** 63%. **Nature:** light yellow liquid. ¹**H NMR (400 MHz, CDCl₃)**: δ 7.70 (d, *J* = 8.3 Hz, 2H), 7.35-7.28 (m, 5H), 7.22 (d, *J* = 8.6 Hz, 2H), 5.08-4.96 (m, 3H), 3.80-3.70 (m, 1H), 2.71 (dd, *J* = 16.5, 7.4 Hz, 1H), 2.38 (s, 3H), 2.00-1.91 (m, 2H), 1.81-1.70 (m, 1H), 1.67-1.55 (m, 2H), 1.50-1.40 (m, 1H).¹³**C NMR (101 MHz, CDCl₃)**: δ 174.1, 143.5, 137.1, 135.8, 129.7, 128.7, 128.3, 128.2, 127.4, 66.7, 57.8, 50.9, 33.6, 28.2, 22.9, 21.6. **HRMS:** [M + H]⁺ Calculated for C₂₀H₂₄NO₄S 374.1426, found 374.1426.

e) Benzyl (1*R*,2*R*)-1-methyl-2-((4-methylphenyl)sulfonamido)cyclopentane-1-carboxylate (3ag)



Prepared according to **RP1**. **1a** (0.050 g, 0.236 mmol), **2a** (0.124 g, 0.708 mmol), **3ag** (0.061 g, 67%). **Reaction time:** 5.5 h. **Yield:** 67%. **Nature:** white solid. ¹**H NMR**: δ 7.70 (d, *J* = 8.3 Hz, 2H), 7.37-7.27 (m, 4H), 7.23 (d, *J* = 8.6 Hz, 2H), 5.05-4.91 (m, 3H), 3.8-3.83 (m, 1H), 2.38 (s, 3H), 2.09-1.97 (m, 1H), 1.95-1.84 (m, 1H), 1.65-1.52 (m, 3H), 1.47-1.37 (m, 1H), 1.17 (s, 3H). ¹³**C NMR (101 MHz, CDCl₃):** δ 176.9, 143.4, 137.4, 129.7, 127.3, 60.0, 52.1, 51.2, 36.1, 31.0, 21.6, 20.2, 17.6. **HRMS:** [M + H]⁺ Calculated for C₂₁H₂₆NO₄S 388.1583, found 388.1583.



found 312.1270.

Prepared according to **RP1**. **1a** (0.050 g, 0.236 mmol), **2a** (0.070 g, 0.708 mmol), **3ah** (0.048 g, 67%). **Reaction time:** 6 h. **Yield:** 67%. **Nature:** Transparent oily liquid. ¹**H NMR (400 MHz, CDCl₃)**: δ 7.73 (d, *J* = 8.3 Hz, 2H), 7.28 (d, *J* = 8.4 Hz, 2H), 4.79 (t, *J* = 5.7 Hz, 1H), 3.84-3.78 (m, 1H), 3.54 (s, 3H), 2.41 (s, 3H), 2.06-1.96 (m, 1H), 1.96-1.88 (m, 1H), 1.62-1.50 (m, 3H), 1.47-1.37 (m, 1H), 1.14 (s, 3H).¹³C **NMR (101 MHz, CDCl₃)**: δ 176.2, 143.4, 137.3, 136.0, 129.7, 128.6, 128.2, 128.0, 127.3, 66.7, 60.0, 51.4, 36.2, 31.0, 21.6, 20.3, 17.7. **HRMS:** [M + H]⁺ Calculated for C₁₅H₂₂NO₄S 312.1270,

g) 4-methyl-*N*-((1*R*,2*S*)-2-phenylcyclopentyl)benzenesulfonamide (3*ai*)



Prepared according to **RP1**. **1a** (0.050 g, 0.236 mmol), **2a** (0.124 g, 0.708 mmol), **3ai** (0.048 g, 66%). **Reaction time:** 5.5 h. **Yield:** 66%. **Nature:** Oily Liquid. ¹**H NMR (400 MHz, CDCl₃)**: δ 7.46 (d, *J* = 8.3 Hz, 2H), 7.17-7.12 (m, 3H), 7.10 (d, *J* = 8.6 Hz, 2H), 7.00-6.94 (m, 2H), 4.71 (d, *J* = 6.3 Hz, 1H), 3.50-3.40 (m, 1H), 2.73 (dd, *J* = 18.0, 9.8 Hz, 1H), 2.37 (s, 3H), 2.19-2.08 (m, 1H), 2.07-1.78 (m, 2H), 1.74 (d, *J* = 31.8 Hz, 2H), 1.60-1.52 (dd, *J* = 23.3, 8.5 Hz, 2H).¹³**C NMR (101 MHz, CDCl₃)**: δ 143.0, 141.6, 137.1, 129.5, 128.6, 127.3, 127.1, 126.7, 61.4, 52.5, 33.3, 32.3, 22.2, 21.6. **HRMS:** [M + H]⁺ Calculated for C₁₈H₂₂NO₂S 316.1371, found 316.1371.

h) (1R,2R)-2-(4-methylphenylsulfonamido)-N,N-diphenylcyclopentanecarboxamide (3aj)



Prepared according to **RP1**. **1a** (0.050 g, 0.236 mmol), **2a** (0.158 g, 0.708 mmol), **3aj** (0.069 g, 58%). **Reaction time:** 3 h. **Yield:** 58%. **Nature:** Transparent oily liquid. ¹**H NMR (400 MHz, CDCl**₃): δ 7.79 (d, J = 8.3 Hz, 2H), 7.44 – 7.29 (m, 1H), 7.20 – 7.14 (m, 1H), 4.28 (d, J = 6.8 Hz, 1H), 4.06 – 3.97 (m, 1H), 2.80 (dd, J = 15.7, 8.4 Hz, 1H), 2.41 (s, 3H), 1.95 – 1.85 (m, 1H), 1.80-1.76 (m, 2H), 1.69 – 1.58 (m, 1H), 1.30 – 1.20 (m, 2H).¹³**C NMR (101 MHz, CDCl**₃): δ 174.7, 143.7, 143.5, 142.9, 137.1, 129.8, 129.8, 129.1, 128.9, 128.0, 127.4, 126.8, 126.5, 126.2, 58.4, 50.3, 34.0, 30.7, 23.6, 21.7. **HRMS:** [M + H]⁺ Calculated for

 $C_{25}H_{27}N_2O_3S$ 435.1742, found 435.1742.



Prepared according to **RP1**. **1a** (0.050 g, 0.236 mmol), **2a** (0.124 g, 0.708 mmol), **3ak** (0.055 g, 58%). **Reaction time:** 3 h. **Yield:** 58%. **Nature:** Transparent oily liquid. ¹**H NMR (400 MHz, CDCl**₃): δ 7.82 (d, *J* = 14.4 Hz, 1H), 7.73 (d, *J* = 8.4 Hz, 2H), 7.30 (d, *J* = 8.5 Hz, 2H), 7.00 (q, *J* = 2.4 Hz, 1H), 5.50 (s, 1H), 5.24 (d, *J* = 14.5 Hz, 1H), 4.13 (q, *J* = 7.1 Hz, 2H), 4.04-3.93 (m, 2H), 2.61-2.55 (m, 1H), 2.50-2.45 (m, 1H), 2.41 (s, 3H), 2.23-2.21 (m, 1H), 1.94-1.88 (m, 1H), 1.24 (t, *J* = 7.2 Hz, 3H), 1.16 (t, *J* = 7.1 Hz, 3H).¹³**C NMR (101 MHz, CDCl**₃): δ 167.4, 163.0, 148.5, 144.6, 136.1, 134.0, 129.9, 127.7, 99.3, 62.8, 60.6, 60.2, 31.2, 28.0, 21.7, 14.4, 14.1. **HRMS:** [M + H]⁺ Calculated for C₂₀H₂₆NO₆S 408.1481, found

408.1481.

j) Ethyl (1*S*,2*S*,3*R*)-3-((4-methylphenyl)sulfonamido)-2-phenylcyclopentane-1-carboxylate (3al)



Prepared according to **RP1**. **1a** (0.050 g, 0.236 mmol), **2a** (0.124 g, 0.708 mmol), **3al** (0.054 g, 60%). **Reaction time:** 4 h. **Yield:** 60%. **Nature:** Transparent liquid. ¹H **NMR (400 MHz, CDCl₃):** δ 7.45 (d, *J* = 8.3 Hz, 2H), 7.17-7.09 (m, 5H), 6.94 (dd, *J* = 7.5, 2.0 Hz, 2H), 4.77 (d, *J* = 6.9 Hz, 1H), 4.02 (q, *J* = 7.1 Hz, 2H), 3.63-3.59 (m, 1H), 3.07-3.05 (m, 1H), 2.80-2.78 (m, 1H), 2.37 (s, 3H), 2.20-2.18 (m, 1H), 2.07-1.99 (m, 2H), 1.72-1.63 (m, 1H), 1.11 (t, *J* = 7.2 Hz, 3H).¹³**C NMR (101 MHz, CDCl₃)**: δ 175.1, 143.1, 139.5, 137.2, 129.7, 129.6, 129.1, 128.7, 128.4, 127.5, 127.1, 127.0, 61.0, 60.9, 58.2, 55.2, 51.7, 49.5, 46.4, 32.6, 27.1, 21.6, 14.2. **HRMS:** [M + H]⁺ Calculated for C₂₁H₂₆NO₄S 388.1582, found 388.1582.

k) (1S,2S,3R)-ethyl 2-(4-fluorophenyl)-3-(4-methylphenylsulfonamido)cyclopentanecarboxylate (3am)



Prepared according to **RP1**. **1a** (0.050 g, 0.236 mmol), **2a** (0.137 g, 0.708 mmol), **3am** (0.055 g, 58%). **Reaction time:** 4.5 h. **Yield:** 58%. **Nature:** Transparent oily liquid. ¹**H NMR (400 MHz, CDCl**₃): δ 7.45 (d, *J* = 8.3 Hz, 2H), 7.11 (d, *J* = 8.1 Hz, 2H), 6.91-6.86 (m, 2H), 6.80 (t, *J* = 8.7 Hz, 2H), 4.57 (d, *J* = 7.4 Hz, 1H), 4.02 (q, *J* = 7.1 Hz, 2H), 3.61-3.56 (m, 1H), 3.03 (t, *J* = 9.6 Hz, 1H), 2.74-2.73 (m, 1H), 2.39 (s, 3H), 2.21-2.20 (m, 1H), 2.18-2.07 (m, 1H), 2.04-1.90 (m, 1H), 1.70-1.67 (m, 1H), 1.11 (t, *J* = 7.1 Hz, 3H).¹³**C NMR (101 MHz, CDCl**₃): δ 174.8, 143.4, 137.1, 135.1, 129.5, 129.0, 128.9, 127.0, 115.6, 115.4, 61.1, 60.9, 54.6, 49.5, 32.5, 26.8, 21.6, 14.2. **HRMS:** [M + H]⁺ Calculated for C₂₁H₂₅NO₄FS 406.1488, found

406.1488.



Prepared according to **RP1**. **1a** (0.050 g, 0.236 mmol), **2a** (0.146 g, 0.708 mmol), **3an** (0.056 g, 58%). **Reaction time:** 4 h. **Yield:** 58%. **Nature:** Transparent oily liquid. ¹**H NMR (400 MHz, CDCl₃)**: δ 7.44 (d, *J* = 8.3 Hz, 2H), 7.08 (dd, *J* = 7.9, 5.9 Hz, 3H), 6.71-6.70 (m, 1H), 6.55 (d, *J* = 7.6 Hz, 1H), 6.44-6.42 (m, 1H), 4.90 (d, *J* = 6.9 Hz, 1H), 4.02 (q, *J* = 7.2 Hz, 2H), 3.68 (s, 3H), 3.58-3.56 (m, 1H), 3.05 (t, *J* = 9.3 Hz, 1H), 2.79-2.77 (m, 1H), 2.36 (s, 3H), 2.19-2.18 (m, 1H), 2.06-2.04 (m, 3H), 1.72-1.66 (m, 1H), 1.11 (t, *J* = 7.1 Hz, 3H).¹³**C NMR (101 MHz, CDCl₃)**: δ 175.1, 159.8, 143.1, 141.2, 137.1, 129.7, 129.5, 127.0, 119.9, 112.9, 112.7, 61.1, 60.9, 55.2, 55.1, 49.3, 32.6, 27.1, 21.6, 14.2. **HRMS:** [M + H]⁺ Calculated for C₂₂H₂₈NO₅S 418.1688, found 418.1688.

m) Methyl (1R,2R)-2-(phenylsulfonamido)cyclopentane-1-carboxylate (3ba)



Prepared according to **RP1**. **1a** (0.040 g, 0.205 mmol), **2a** (0.053 g, 0.615 mmol), **3ba** (0.026 g, 46%). **Reaction time:** 4 h. **Yield:** 46%. **Nature:** transparent oily liquid. ¹**H NMR (400 MHz, CDCl**₃): δ 7.88-7.84 (m, 2H), 7.59-7.55 (m, 1H), 7.54-7.47 (m, 2H), 5.06 (d, *J* = 6.2 Hz, 1H), 3.77-3.70 (m, 1H), 3.54 (s, 3H), 2.63 (dd, *J* = 16.7, 7.7 Hz, 1H), 2.00-1.93 (m, 2H), 1.78-1.71 (m, 1H), 1.65-1.58 (m, 2H), 1.50-1.42 (m, 1H).¹³**C NMR (101 MHz, CDCl**₃): δ 174.6, 140.1, 132.8, 129.2, 127.3, 57.9, 52.0, 50.7, 33.7, 28.1, 22.9. **HRMS:** [M + H]⁺ Calculated for C₁₃H₁₇NO₄SNa 306.0776, found 306.0776.

n) Methyl (1*R*,2*R*)-2-(naphthalene-2-sulfonamido)cyclopentane-1-carboxylate (3ca)



Prepared according to **RP1**. **1a** (0.050 g, 0.205 mmol), **2a** (0.053 g, 0.615 mmol), **3ac** (0.042 g, 62%). **Reaction time:** 4 h. **Yield:** 62%. **Nature:** oily liquid. ¹**H NMR (400 MHz, CDCl₃):** δ 8.45 (s, 1H), 7.96 (dd, J = 8.1, 5.3 Hz, 2H), 7.89 (d, J = 8.9 Hz, 1H), 7.83 (dd, J = 8.7, 1.9 Hz, 1H), 7.65-7.56 (m, 2H), 5.16 (d, J =6.6 Hz, 1H), 3.83-3.76 (m, 1H), 3.42 (s, 3H), 2.64 (dd, J = 16.6, 7.6 Hz, 1H), 1.96 (d, J = 53.5 Hz, 2H), 1.76-1.69 (m, 1H), 1.66-1.51 (s, 2H), 1.53-1.42 (m, 1H). ¹³C NMR (101 MHz, CDCl₃): δ 174.6, 137.0, 134.9, 132.2, 129.5, 129.4, 128.9, 128.8, 128.0, 127.6, 122.5, 57.9, 51.9, 50.8, 33.7, 28.2. HRMS: [M + H]⁺

Calculated for $C_{17}H_{20}NO_4S$ 334.1118, found 334.1118.



Prepared according to **RP1**. **1a** (0.054 g, 0.205 mmol), **2a** (0.053 g, 0.615 mmol), **3ad** (0.043 g, 61%). **Reaction time:** 8 h. **Yield:** 61%. **Nature:** Oily liquid. ¹**H NMR (400 MHz, CDCl₃):** δ 8.00 (d, *J* = 8.3 Hz, 2H), 7.77 (d, *J* = 8.3 Hz, 2H), 5.15 (d, *J* = 6.6 Hz, 1H), 3.82-3.74 (m, 1H), 3.54 (s, 3H), 2.67-2.59 (m, 1H), 2.07-1.93 (m, 2H), 1.79-1.72 (m, 3H), 1.48-1.44 (m, 1H).¹³**C NMR (101 MHz, CDCl₃):** δ 174.5, 143.9, 127.9, 126.3, 126.3, 126.3, 126.2, 124.7, 122.0, 57.9, 52.0, 50.7, 33.7, 29.8, 28.3, 22.8. **HRMS:** [M + H]⁺ Calculated for C₁₄H₁₇NO₄SF₃ 352.0830, found 352.0830.

p) Ethyl (1R,2R)-2-((4-methylphenyl)sulfonamido)cyclopentane-1-carboxylate (3ea)



Prepared according to **RP1**. **1a** (0.047 g, 0.205 mmol), **2a** (0.053 g, 0.615 mmol), **3ea** (0.043 g, 68%). **Reaction time:** 5 h. **Yield:** 68%. **Nature:** Yellow oily liquid. ¹**H NMR (400 MHz, CDCl₃)**: δ 7.79 (d, *J* = 9.0 Hz, 2H), 6.95 (d, *J* = 9.0 Hz, 2H), 5.08 (d, *J* = 6.3 Hz, 1H), 3.85 (s, 3H), 3.73-3.65 (m, 1H), 3.57 (s, 3H), 2.64 (dd, *J* = 16.6, 7.6 Hz, 1H), 2.02-1.90 (m, 2H), 1.75-1.52 (m, 2H), 1.50-1.38 (m, 1H).¹³**C NMR (101 MHz, CDCl₃)**: δ 174.8, 163.0, 131.7, 129.5, 114.3, 57.8, 55.7, 52.0, 50.7, 33.6, 28.2, 22.9. **HRMS:** [M + H]⁺ Calculated for C₁₄H₂₀NO₅S 314.1062, found 314.1062.

q) Methyl (1R,2R)-2-((4-bromophenyl)sulfonamido)cyclopentane-1-carboxylate (3fa)



Prepared according to **RP1**. **1a** (0.056 g, 0.205 mmol), **2a** (0.053 g, 0.615 mmol), **3fa** (0.058 g, 62%). **Reaction time:** 6 h. **Yield:** 62%. **Nature:** Light yellow oily liquid. ¹**H NMR (400 MHz, CDCl₃)**: δ 7.75 – 7.70 (m, 2H), 7.66-7.61 (m, 2H), 5.15 (d, *J* = 6.6 Hz, 1H), 3.78-3.69 (m, 1H), 3.57 (s, 3H), 2.66-2.60 (m, 1H), 2.00-1.98 (m, 2H), 1.97-1.71 (m, 1H), 1.69-1.48 (m, 2H), 1.46-1.42 (m, 1H).¹³**C NMR** (**101 MHz, CDCl₃**): δ 174.6, 139.3, 132.4, 128.9, 127.7, 57.9, 52.1, 50.7, 33.6, 28.3, 22.9. **HRMS:** [M + H]⁺ Calculated for C₁₃H₁₇NO₄SBr 362.0062, found 362.0062.

r) N-((1R,2R)-2-(hydroxymethyl)cyclopentyl)-4-methylbenzenesulfonamide (3ada)



Prepared according to **RP1**. **3ad** (0.030 g, 0.100 mmol), **3ada** (0.025 g, 86%). **Reaction time:** 3 h. **Yield:** 86%. **Nature:** Transparent oily liquid. ¹H **NMR (400 MHz, CDCl₃)**: δ 7.75 (d, *J* = 8.3 Hz, 2H), 7.28 (d, *J* = 8.1 Hz, 2H), 5.13 (d, *J* = 5.7 Hz, 1H), 3.80 – 3.73 (m, 1H), 2.74-2.68 (m, 1H), 2.40 (s, 3H), 2.05-2.00 (m, 2H), 1.98-1.92 (m, 1H), 1.82-1.62 (m, 2H), 1.49-1.42 (m, 1H).¹³C **NMR (101 MHz, CDCl₃**): δ 179.5, 143.7, 136.9, 129.8, 127.4, 57.6, 50.7, 33.5, 28.3, 23.0, 21.7. **HRMS:** [M + H]⁺ Calculated for C₁₃H₁₈NO₄S 284.0957, found 284.0957.



270.1164.

Prepared according to **RP1**. **3ad** (0.030 g, 0.100 mmol), **3adb** (0.022 g, 81%). **Reaction time:** 4 h. **Yield:** 81%. **Nature:** Transparent oily liquid. ¹**H NMR (400 MHz, CDCl₃)**: δ 7.76 (d, J = 8.3 Hz, 2H), 7.29 (d, J = 8.2 Hz, 2H), 5.18 (d, J = 6.5 Hz, 1H), 3.67 (dd, J = 11.0, 5.0 Hz, 1H), 3.48 (dd, J = 11.0, 6.8 Hz, 1H), 3.29-3.20 (m, 1H), 2.41 (s, 3H), 2.12 (s, 1H), 1.90-1.78 (m, 1H), 1.77-1.68 (m, 2H), 1.66-1.54 (m, 1H), 1.54-1.41 (m, 1H), 1.40-1.30 (m, 1H), 1.29-1.18 (m, 1H).¹³C **NMR (101 MHz, CDCl₃)**: δ 143.5, 137.4, 129.8, 127.2, 64.5, 57.9, 48.4, 33.1, 26.1, 22.0, 21.6. **HRMS:** [M + H]⁺ Calculated for C₁₃H₂₀NO₃S 270.1164, found

t) N-(2-cyanoethyl)-N-cyclopropyl-4-methylbenzenesulfonamide (3ab')



Prepared according to **RP1**. **1a** (0.050 g, 0.236 mmol), **2a** (0.037 g, 0.708 mmol), **3an** (0.056 g, 77%). **Reaction time:** 4 h. **Yield:** 58%. **Nature:** Transparent oily liquid. ¹**H NMR (400 MHz, CDCl₃)**: δ 7.72 (d, *J* = 8.3 Hz, 2H), 7.33 (d, *J* = 7.9 Hz, 2H), 3.42 (t, *J* = 7.0 Hz, 2H), 2.72 (t, *J* = 6.9 Hz, 2H), 2.09 (tt, *J* = 6.9, 3.6 Hz, 1H), 0.88 (d, *J* = 16.1 Hz, 2H), 0.72 (d, *J* = 19.7 Hz, 2H).¹³**C NMR (101 MHz, CDCl₃)**: δ 144.3, 134.4, 129.9, 127.9, 118.0, 47.5, 31.6, 21.7, 18.9, 7.9. **HRMS**: [M + H]⁺ Calculated for C₁₃H₂₀NO₃S 265.0972, found 265.1011.

u) N-(2-cyanoethyl)-N-cyclopropyl-4-methylbenzenesulfonamide (3ac')



Prepared according to **RP1**. **1a** (0.050 g, 0.236 mmol), **2a** (0.049 g, 0.708 mmol), **3an** (0.055 g, 83%). **Reaction time:** 4 h. **Yield:** 58%. **Nature:** Transparent oily liquid. ¹**H NMR (400 MHz, CDCl₃)**: δ 7.69 (d, J = 8.3 Hz, 2H), 7.29 (d, J = 8.1Hz, 2H), 3.39-3.33 (m, 2H), 2.85-2.78 (m, 2H), 2.40 (s, 3H), 2.12 (s, 3H), 1.97-1.92 (m, 1H), 0.82-0.76 (m, 2H), 0.67-0.61 (m, 2H).¹³C **NMR (101 MHz, CDCl₃)**: δ 207.1, 143.7, 134.8, 129.7, 127.8, 46.3, 43.3, 31.4, 30.4, 21.6, 7.6.

v) 4-methylbenzenesulfonamide (IV)



Nature: white solid. ¹**H NMR (400 MHz, CDCl₃)**: δ 7.80 (d, *J* = 8.4 Hz, 2H), 7.30 (d, *J* = 7.9 Hz, 2H), 4.87 (s, 2H), 2.42 (s, 3H).

H) Mechanism for formation of 3ak



I) Alternate mechanism for the protocol





Zoomed Version



Conclusion: NOESY experiments reveals that there is no interaction Ha and Hb. This confirms the *trans* product



As described by Aggarwal et al, the reason for the diastereoselective formation of products could be the steric repulsion between the N-sulfonyl imine and the ester group which disfavors the cis product in comparison to the trans where the molecule do not suffer any steric repulsion.

K) Single Crystal data of 3ag and 3al

For the determination of X-ray crystal structures of **3ag** and **3al** a single crystal was selected and mounted with paratone oil on a glass fiber using gum. The data was collected at 293K on a CMOS based Bruker D8 Venture PHOTON 100 diffractometer equipped with an INCOATEC micro-focus source with graphite monochromatic Mo K α radiation ($\lambda = 0.71073$ Å) operation at 50 kV and 30 mA. For the integration of diffraction profiles SAINT program³ was used. Absorption correction was done applying SADABS program.⁴ The crystal structure was solved by SIR 92⁵ and refined by full matrix least square method using SHELXL-97⁶ WinGX system, Ver 1.70.01.⁷ All the non-hydrogen atoms in the structure were located the Fourier map and refined anisotropically. The hydrogen atoms were fixed by HFIX in their ideal positions and refined using riding model with isotropic thermal parameters. The crystal structure (excluding structure factor) has been deposited to Cambridge Crystallographic Data Centre and allocated deposition number: **3ag: CCDC 2115609, 3al: CCDC 2145541.**



Figure S1: X-ray crystal structure of 3ag

CCDC No.	CCDC 2115609
Formula	C21 H25 N O4 S
Formula weight	387.48
Crystal System	Monoclinic
Space group	P21/n
a, b, c (Å)	13.4901(9) 9.8844(7) 15.6641(11)
α, β, γ (°)	90 104.766(2) 90
V (Å ³)	2019.7(2)
Ζ	4
Calculated Density (g/cm ³)	1.274
Absorption coefficient (mm ⁻¹)	0.186
F(000)	824
Crystal Size (mm ³)	0.05 x 0.23 x 0.36
Theta range for data collection:	2.3° to 26.4°
Data set	-16: 16 ; -12: 12 ; -19: 19

Reflection	70148
Independent refl.	[R(int) = 0.066]
data $[I > 2\sigma(I)]$	3433
R indices (all data)	$R = 0.0461, WR_2 = 0.1201$
S	1.05
Min. and Max. Resd. Dens. (e/Å ³)	-0.27 and 0.32

Table S1: Selected bond lengths [Å] of 3ag

Atoms	Bond lengths [Å]	Atoms	Bond lengths [Å]
S1 -O1	1.4301(16)	C19 -C20	1.373(4)
S1 -O2	1.4301(17)	C1 -H1A	0.9300
S1 -N1	1.6103(15)	C2 -H2	0.9300
S1 -C7	1.7690(17)	C4 -H4A	0.9600
O3 -C13	1.193(2)	C4 -H4B	0.9600
O4 -C13	1.328(2)	C4 -H4C	0.9600
O4 -C14	1.456(2)	C5 -H5	0.9300
N1 -C8	1.459(2)	Сб -Нб	0.9300
C1 -C2	1.377(3)	C8 -H8	0.9800
C1 -C7	1.375(3)	С9 -Н9А	0.9700
N1 -H1	0.8600	C9 -H9B	0.9700
C00A -C8	1.552(2)	C10 -H10A	0.9700
C00A -C12	1.522(3)	C10 -H10B	0.9700
C00A -C13	1.514(2)	C11 -H11A	0.9700
C00A -C11	1.539(3)	C11 -H11B	0.9700
C2 -C3	1.376(3)	C12 -H12A	0.9600
C3 -C5	1.386(3)	C12 -H12B	0.9600
C3 -C4	1.512(3)	C12 -H12C	0.9600
C5 -C6	1.378(3)	C14 -H14A	0.9700
C6 -C7	1.389(3)	C14 -H14B	0.9700
C8 -C9	1.534(3)	C16 -H16	0.9300
C9 -C10	1.516(3)	C17 -H17	0.9300
C10 -C11	1.527(3)	C18 -H18	0.9300
C14 -C15	1.496(3)	C19 -H19	0.9300
C15 -C20	1.384(3)	C20 -H20	0.9300
C15 -C16	1.377(3)		
C16 -C17	1.378(4)		
C17 -C18	1.370(4)		
C18 -C19	1.366(4)		

Atoms	Bond angles[°]	Atoms	Bond angles[°]
O1 -S1 -O2	120.44(10)	N1 -C8 -C9	114.76(15)
O1 -S1 -N1	107.68(8)	C8 -C9 -C10	105.53(16)
O1 -S1 -C7	107.11(9)	C9 -C10 -C11	106.52(17)
O2 -S1 -N1	105.93(9)	C00A -C11 -C10	105.97(17)
O2 -S1 -C7	107.39(9)	O3 -C13 -C00A	124.38(17)
N1 -S1 -C7	107.74(8)	03 -C13 -O4	122.49(16)
C13 -O4 -C14	115.95(14)	O4 -C13 -C00A	113.10(15)
S1 -N1 -C8	121.57(12)	O4 -C14 -C15	107.32(15)
C2 -C1 -C7	120.04(18)	C14 -C15 -C16	120.51(17)
C8 -N1 -H1	119.00	C16 -C15 -C20	118.77(19)
S1 -N1 -H1	119.00	C14 -C15 -C20	120.69(18)
C8 -C00A -C12	112.01(15)	C15 -C16 -C17	120.7(2)
C8 -C00A -C11	100.43(14)	C16 -C17 -C18	119.7(2)
C11 -C00A -C13	114.20(15)	C17 -C18 -C19	120.4(3)
C12 -C00A -C13	109.40(15)	C18 -C19 -C20	120.1(3)
C11 -C00A -C12	112.02(15)	C15 -C20 -C19	120.4(2)
C8 -C00A -C13	108.51(13)	C2 -C1 -H1A	120.00
C1 -C2 -C3	121.5(2)	C7 -C1 -H1A	120.00
C2 -C3 -C4	121.6(2)	C1 -C2 -H2	119.00
C2 -C3 -C5	118.12(19)	С3 -С2 -Н2	119.00
C4 -C3 -C5	120.27(19)	C3 -C4 -H4A	109.00
C3 -C5 -C6	121.19(19)	C3 -C4 -H4B	109.00
C5 -C6 -C7	119.69(19)	C3 -C4 -H4C	109.00
S1 -C7 -C6	119.66(14)	H4A -C4 -H4B	110.00
C1 -C7 -C6	119.48(16)	H4A -C4 -H4C	109.00
S1 -C7 -C1	120.86(13)	H4B -C4 -H4C	109.00
C00A -C8 -C9	103.29(14)	С3 -С5 -Н5	119.00
N1 -C8 -C00A	113.74(14)	С6 -С5 -Н5	119.00
С5 -С6 -Н6	120.00	C00A -C12 -H12B	109.00
С7 -С6 -Н6	120.00	C00A -C12 -H12C	109.00
N1 -C8 -H8	108.00	H12A -C12 -H12B	109.00
C00A -C8 -H8	108.00	H12A -C12 -H12C	109.00
С9 -С8 -Н8	108.00	H12B -C12 -H12C	109.00
С8 -С9 -Н9А	111.00	O4 -C14 -H14A	110.00
С8 -С9 -Н9В	111.00	O4 -C14 -H14B	110.00
С10 -С9 -Н9А	111.00	C15 -C14 -H14A	110.00
С10 -С9 -Н9В	111.00	C15 -C14 -H14B	110.00
H9A -C9 -H9B	109.00	H14A -C14 -H14B	109.00
C9 -C10 -H10A	110.00	C15 -C16 -H16	120.00
C9 -C10 -H10B	110.00	C17 -C16 -H16	120.00
C11 -C10 -H10A	110.00	С16 -С17 -Н17	120.00
С11 -С10 -Н10В	110.00	С18 -С17 -Н17	120.00
H10A -C10 -H10B	109.00	С17 -С18 -Н18	120.00
C00A -C11 -H11A	111.00	C19 -C18 -H18	120.00
C00A -C11 -H11B	111.00	C18 -C19 -H19	120.00
C10 -C11 -H11A	111.00	С20 -С19 -Н19	120.00
C10 -C11 -H11B	111.00	С15 -С20 -Н20	120.00
H11A -C11 -H11B	109.00	С19 -С20 -Н20	120.00
C00A -C12 -H12A	109.00		

Table S2: Selected bond angles [°] of 3ag

Table S3: Selected hydrogen bonding geometry [Å, º] for a compound 3ag						
DH A	DH	DH HA		DHA		
N1 H1 O3	0.8600	2.3000	2.880(2)	125.00		
C1 H1A O1	0.9300	2.5300	2.880(2)	105.00		
C8 H8 O1	0.9800	2.5600	2.964(2)	105.00		
C11 H11A O4	0.9700	2.3800	2.811(2)	106.00		

Ŷ



Figure S1: X-ray crystal structure of **3al**

CCDC No.	CCDC 2145541
Formula	C21 H25 N O4 S
Formula weight	387.48
Crystal System	Monoclinic
Space group	P21/n
a, b, c (Å)	15.4139(12) 8.4945(6) 17.2042(15)
α, β, γ (°)	90 112.732(2) 90
$V(Å^3)$	2077.6(3)
Ζ	4
Calculated Density (g/cm ³)	1.239
Absorption coefficient (mm ⁻¹)	0.181
F(000)	824
Crystal Size (mm ³)	0.28 x 0.29 x 0.31
Theta range for data collection:	2.6° to 25.7°

Data set	-18: 18 ; -10: 10 ; -20: 21
Reflection	28113
Independent refl.	[R(int) = 0.063]
data $[I > 2\sigma(I)]$	2816
R indices (all data)	$R = 0.0503, WR_2 = 0.1393$
S	1.03
Min. and Max. Resd. Dens. (e/Å ³)	-0.25 and 0.28

Table S1: Selected bond lengths [Å] of 3al

Atoms	Bond lengths [Å]	Atoms	Bond lengths [Å]
S1 -O3	1.4331(18)	C18 -C19	1.377(5)
S1 -O4	1.426(2)	C1 -H1	0.9300
S1 -N101	1.600(2)	C025 -H02A	0.9600
S1 -C20	1.763(2)	С025 -Н02В	0.9600
O1 -C12	1.205(4)	С025 -Н02С	0.9600
O2 -C12	1.314(3)	C2 -H2	0.9300
O2 -C13	1.443(4)	С3 -Н3	0.9300
N101 -C8	1.453(3)	C4 -H4	0.9300
N101 -H101	0.8600	С5 -Н5	0.9300
C1 -C3	1.347(6)	С7 -Н7	0.9800
C1 -C2	1.356(6)	C8 -H8	0.9800
C025 -C18	1.508(4)	C9 -H9A	0.9700
C2 -C4	1.399(5)	C9 -H9B	0.9700
C3 -C5	1.379(4)	C10 -H10A	0.9700
C4 -C6	1.370(4)	C10 -H10B	0.9700
C5 -C6	1.375(4)	C11 -H11	0.9800
C6 -C7	1.509(3)	C13 -H13A	0.9700
C7 -C11	1.545(4)	C13 -H13B	0.9700
C7 -C8	1.534(4)	C14 -H14A	0.9600
C8 -C9	1.534(4)	C14 -H14B	0.9600
C9 -C10	1.508(4)	C14 -H14C	0.9600
C10 -C11	1.534(4)	C15 -H15	0.9300
C11 -C12	1.491(4)	C16 -H16	0.9300
C13 -C14	1.375(6)	C17 -H17	0.9300
C15 -C20	1.372(4)	C19 -H19	0.9300
C15 -C19	1.383(4)		
C16 -C17	1.376(4)		
C16 -C20	1.378(4)		
C17 -C18	1.376(4)		

Table S2:	Selected	bond	angles	[°]	of 3al
-----------	----------	------	--------	-----	--------

Atoms	Bond angles[°]	Atoms	Bond angles[°]		
O3 -S1 -O4	119.83(12)	C10 -C11 -C12	114.9(2)		
O3 -S1 -N101	105.54(11)	O2 -C12 -C11	111.7(2)		
O3 -S1 -C20	108.31(11)	O1 -C12 -C11	124.6(3)		
O4 -S1 -N101	108.51(11)	O1 -C12 -O2	123.7(3)		
O4 -S1 -C20	107.14(12)	O2 -C13 -C14	110.5(3)		
N101 -S1 -C20	106.86(12)	C19 -C15 -C20	119.6(3)		
C12 -O2 -C13	117.5(2)	C17 -C16 -C20	119.6(3)		
S1 -N101 -C8	123.03(18)	C16 -C17 -C18	121.6(3)		
C8 -N101 -H101	118.00	C17 -C18 -C19	117.8(3)		
S1 -N101 -H101	119.00	C025 -C18 -C17	120.4(3)		
C2 -C1 -C3	119.4(3)	C025 -C18 -C19	121.8(3)		
C1 -C2 -C4	120.6(4)	C15 -C19 -C18	121.5(3)		
C1 -C3 -C5	120.3(4)	C15 -C20 -C16	119.9(2)		
C2 -C4 -C6	120.6(3)	S1 -C20 -C15	119.9(2)		
C3 -C5 -C6	122.0(3)	S1 -C20 -C16	120.2(2)		
C4 -C6 -C7	120.9(2)	C2 -C1 -H1	120.00		
C5 -C6 -C7	122.1(2)	C3 -C1 -H1	120.00		
C4 -C6 -C5	117.1(2)	C18 -C025 -H02A	109.00		
C8 -C7 -C11	101.96(19)	C18 -C025 -H02B	109.00		
C6 -C7 -C8	115.7(2)	C18 -C025 -H02C	109.00		
C6 -C7 -C11	115.9(2)	H02A -C025 -H02B	109.00		
C7 -C8 -C9	103.9(2)	H02A -C025 -H02C	109.00		
N101 -C8 -C7	113.0(2)	H02B -C025 -H02C	109.00		
N101 -C8 -C9	112.9(2)	C1 -C2 -H2	120.00		
C8 -C9 -C10	106.2(2)	C4 -C2 -H2	120.00		
C9 -C10 -C11	107.1(2)	С1 -С3 -Н3	120.00		
C7 -C11 -C12	109.9(2)	С5 -С3 -Н3	120.00		
C7 -C11 -C10	104.5(2)	C2 -C4 -H4	120.00		
<u>C6 -C4 -H4</u>	120.00	C10 -C11 -H11	109.00		
<u>C3 -C5 -H5</u>	119.00	C12 -C11 -H11	109.00		
<u>C6 -C5 -H5</u>	119.00	O2 -C13 -H13A	110.00		
<u>C6 -C7 -H7</u>	108.00	O2 -C13 -H13B	110.00		
<u>C8 -C7 -H7</u>	108.00	<u>C14</u> -C13 -H13A	110.00		
<u>CII -C/ -H/</u>	108.00	C14 -C13 -H13B	110.00		
N101 -C8 -H8	109.00	H13A -C13 -H13B	108.00		
<u>C/ -C8 -H8</u>	109.00	C13 -C14 -H14A	109.00		
<u>C9 -C8 -H8</u>	109.00	C13 -C14 -H14B	109.00		
C8 - C9 - H9A	110.00	<u>U13 -C14 -H14C</u>	109.00		
$\begin{array}{c} C8 - C9 - H9B \\ \hline C10 - C0 - H0A \\ \hline \end{array}$	111.00	H14A -C14 -H14B	109.00		
$\begin{array}{ccc} C10 & -C9 & -H9A \\ \hline C10 & C0 & H0B \\ \hline \end{array}$	111.00	H14A -C14 -H14C	110.00		
	111.00	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	109.00		
ПЭА -СЭ -ПЭВ СО С10 Ц10A	109.00	$C17 - C13 - \Pi13$ $C20 - C15 - \Pi15$	120.00		
$\begin{array}{c} C_{7} - C_{10} - \Pi I 0 A \\ \hline C_{9} - C_{10} - \Pi I 0 B \\ \hline \end{array}$	110.00	С20 -С13 -П13	120.00		
C11 _C10 _H10A	110.00	C17 - C10 - 1110 C20 - C16 - H16	120.00		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	110.00	C16 -C17 -H17	110.00		
H10A _C10 _H10R	10.00	C18 -C17 -H17	119.00		
C7 -C11 -H11	109.00	C15 - C19 - H19	119.00		
	107.00		117.00		

Table S3: Selected hydrogen bonding geometry [A, ^o] for a compound 3al								
$\mathbf{D}_{\cdot \cdot}\mathbf{H}$	HA	D A	DHA					
0.8600	2.1200	2.950(3)	162.00					
0.9700	2.5100	2.924(4)	105.00					
0.9800	2.5300	3.218(3)	128.00					
0.9300	2.5300	2.900(3)	104.00					
	d hydrogen bon DH 0.8600 0.9700 0.9800 0.9300	d hydrogen bonding geometry DH HA 0.8600 2.1200 0.9700 2.5100 0.9800 2.5300 0.9300 2.5300	d hydrogen bonding geometry [A, *] for a comp DH HA DA 0.8600 2.1200 2.950(3) 0.9700 2.5100 2.924(4) 0.9800 2.5300 3.218(3) 0.9300 2.5300 2.900(3)					

- ?

References

- 1. D. H. White, A. Noble, K. L. Booker-Milburn, and V. K. Aggarwal, Org. Lett. 2021, 23, 3038.
- 2. R. Kumar and P. Banerjee, J. Org. Chem., 2021, 86, 16104.
- Bruker, SAINT V7.68A, Bruker AXS Inc., Madison (WI, USA) 2005. 3.
- 4. G. M. Sheldrick, SADABS 2008/2, Göttingen 2008.
- 5. A. Altomare, G. Cascarano, C. Giacovazzo, A. Guagliardi, J. Appl. Cryst., 1993, 26, 343.
- Sheldrick, G. M. SHELXL-97, Program for Crystal Structure Solution and Refinement, University of Göttingen, Göttingen, 6. Germany 1997.
- 7. L. J. Farrugia, J. Appl. Cryst. 1999, 32, 837.
- 8. CCDC 2115609 and CCDC 2145541 contains supplementary crystallographic data for the compounds 3ag and 3al respectively.

L) ¹H NMR of the Starting materials













M) NMR and HRMS of the Products



Elemental Composition Report

Single Mas Tolerance = Element pre Number of is	S Analysis 300.0 PPM / diction: Off sotope peaks u	DBE: m sed for i-	in = -1.5 FIT = 5	, max = 5	50.0					
Monoisotopic 20 formula(e) Elements Use C: 15-15 H	Mass, Even Ele evaluated with f d: l: 0-100 N: 0-	ctron lons results v 1 O: 0-	vithin limit	s (all resul 1	lts (up to 1	1000) for	each mass)			
Sample Name	: 230821_18-02-	248			IITRPR				XEVO G2-XS C	TOF
Test Name 230821_18-02-2	: 248 44 (0.933)								1: TOF MS	ES+
¹⁰⁰ %- 0- 230	38.1034 245.0889 1011 1011 1011 1011 1011 1011 1011 101	266.0967 267 267 260 270	.0992 	301.153 290 30	4 312.13	38 111111111111111111111111111111111111	334.1155 336.11 336.11 330 340	50.0886 61 363.1512 350 360 3	2 385.1352 4444444 370 380 390	, m/z
Minimum: Maximum:		2.0	300.0	-1.5 50.0						
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula		
312.1338	312.1270	6.8	21.8	5.5	839.2	n/a	n/a	C15 H22 N C	04 S	

Page 1



Elemental Composition Report

Single Mas	s Analysis										
Tolerance = 3	300.0 PPM /	DBE: mi	n = -1.5,	max = 5	0.0						
Element pred	liction: Off										
Number of is	otope peaks us	ed for i-F	IT = 5								
Monoisotopic I 4 formula(e) e Elements Use C: 14-14 H	Mass, Even Elec valuated with 1 n d: : 0-100 N: 0-1	tron lons esults with 0: 4-4	hin limits (4 S: 0-7	(all result: 1	s (up to 10	00) for ea	ch mass)			VEV0.023	/0 OTOF
Sample Name	: 160921_18-02-2	60-IR			IIIRPR					XEVO G2-)	SQIOF
160921_18-02-2	60-IR 10 (0.232)									1: TOF 4	MS ES+ .78e+007
100- 	266	0848		320.0928							
230.2494	238.0901	267.0882	298.1111	321.	.0961	368.075	4 394.347	6 408.3279	439.3793	467.410247	7.2048
220	240 260	280	300	320	340	360	380 40	0 420	440	460	480
Minimum: Maximum:		2.0	300.0	-1.5 50.0							
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula			
298.1111	298.1113	-0.2	-0.7	5.5	1771.3	n/a	n/a	C14 H20 1	N 04 S		

Page 1


Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

Monoisotopic Mass, Even Electron Ions

645 formula(e) evaluated with 9 results within limits (up to 1 closest results for each mass) Elements Used: C: 10-25 H: 1-30 N: 1-3 O: 1-5 S: 0-2 CI: 0-2 Na: 0-1

Sample Name : 18_02_300 Test Name : 12012022_18_02_300 22 (0.480) IITRPR

XEVO G2-XS QTOF

1: TOF MS ES+ 5.17e+006

100 					362.1	402	37	8.1143		
326.66	67 331.1024	341.30	70	354.0616 3	57.1857	363.1439	513 371.1072	379.1177	385.2169	92.1189 398.1393
320.0	330.0	340.0		350.0	360.0	. 3	370.0	380.0	390.0	400.0
Minimum: Maximum:		2.0	10.0	-1.5 50.0						
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula		
362.1402	362.1402	0.0	0.0	5.5	1468.2	n/a	n/a	C17 H25	N O4 S Na	L



S38

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

Monoisotopic Mass, Even Electron Ions 2833 formula(e) evaluated with 23 results within limits (up to 1 closest results for each mass) Elements Used: C: 1-30 H: 1-30 N: 1-3 O: 1-6 F: 0-3 S: 0-2 CI: 0-2 Br: 0-1

Sample Name : 18_02_292 Test Name : 12012022_18_02_292 13 (0.294) IITRPR

XEVO G2-XS QTOF 1: TOF MS ES+

1.74e+007 356.1322 100-374.1426 396.1245 % 357.1356 375.1461 412.0985 419.2012 432.7202 ,358.1327 338.1216 342.3123 447.2380 313.2743 ,388.1594 0 450 m/z 420 430 440 360 300 310 330 340 380 320 370 1111 350 390 400 410 Minimum: -1.5 2.0 10.0 50.0 Maximum: Calc. Mass PPM DBE i-FIT Conf(%) Formula Mass mDa Norm 374.1426 374.1426 0.0 0.0 9.5 1551.7 n/a n/a C20 H24 N 04 S

¹H NMR (CDCl₃, 400 MHz) of 3ag





Page 1

XEVO G2-XS QTOF

1: TOF MS ES+

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

Monoisotopic Mass, Even Electron lons 641 formula(e) evaluated with 6 results within limits (up to 1 closest results for each mass) Elements Used: C: 10-25 H: 1-30 N: 1-3 O: 1-5 Na: 0-1 S: 0-2 CI: 0-2 Sample Name : 18_02_289 IITRPR Test Name : 12012022_18_02_289 46 (0.978)

100 % 355.0	1750 368.081 360.0	370.1480 5 371. 370.0	1538 382.0 380.0	388. 1685	1583 389.1622 390.0	405. 5.1025 400.0	410.140	11.1439 413.1452 420.0	426.1163 427.1222 430.0	439.3830 440.0
Minimum: Meximum:		2.0	10.0	-1.5 50.0						
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(3) Formula		
388.1583	388.1583	0.0	0.0	9.5	1112.1	n/a	n/a	C21 H26	N 04 S	





110 100 f1 (ppm) -10 -20 J-12 %

Single Mass Analysis Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5 Monoisotopic Mass, Even Electron Ions 280 formula(e) evaluated with 2 results within limits (up to 1 closest results for each mass) Elements Used: C: 10-25 H: 1-30 N: 1-3 O: 1-5 S: 0-2 CI: 0-2 Sample Name : 18_02_284 IITRPR XEVO G2-XS QTOF Test Name 12012022_18_02_284 30 (0.639) 1: TOF MS ES+ 5.30e+006 252.1057 100-334.1088 280.1008 475.3259 253.1089 453.3439 312.1270 476.3289 350.0827 155.0179 186.0556 196.0540 227.1763 396.0813 435.3398 որիսրություրուդ m/z 460 480 500 0-ափափոփ

100	120	140	160	100	200	220	240	260	200	300	320	340	300	300	400	420	44	0 400	400	ວ
Minimun Maximun	n: n:			2	.0	10.	0	-1.5 50.0												
Mass		Calc.	Mas	s m	Da	PPM	I	DBE	i-	FIT	Nor	m	Conf (%) F	'ormu!	la				
312.127	70	312.1	270	0	.0	0.0		5.5	13	94.0	n/a		n/a	С	15 H2	22 N	04	S		





Single Mas Tolerance = Element pre Number of is	ss Analysis 10.0 PPM / I diction: Off sotope peaks us	DBE: mir sed for i-l	i = -1.5, r FIT = 5	nax = 50	0.0					
Monoisotopic 666 formula(e Elements Use C: 1-30 H:	Mass, Even Elec e) evaluated with ed: 1-30 N: 1-3 C	ctron lons 3 results): 1-6 S:	within limi 0-2 CI:	ts (up to 1 0-2 Br:	1 closest re 0-1	esults for e	ach mass)			
Sample Name	: 18_02_319				IITRPR				XEVO 0	32-XS QTOF
12012022_18_0	02_319 16 (0.356)								1: T	OF MS ES+ 5.34e+006
100				316.1	371	338.1189	1			
254.0876 250 2	275.0538 60 270 28	<u>279.1434</u> ³ 0 290	01.1434 ³¹	3.2743 310	335. 320 33	1104 30 340	354.09 40.1193 4 	62 355.0975 370.10 360 370)45 386.0874 380 390	400.0897
Minimum: Maximum:		2.0	10.0	-1.5 50.0						
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula		
316.1371	316.1371	0.0	0.0	8.5	1493.1	n/a	n/a	C18 H22 N 03	2 S	







Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

Monoisotopic Mass, Even Electron Ions 3110 formula(e) evaluated with 10 results within limits (up to 1 closest results for each mass) Elements Used: C: 1-30 H: 1-30 N: 1-3 O: 1-6 F: 0-3 S: 0-2 CI: 0-2 Br: 0-1

IITRPR XEVO G2-XS QTOF Sample Name : 18_02_341 Test Name 12012022_18_02_341 10 (0.232) 1: TOF MS ES+ 1.16e+007 435.1742 100₇ %-457.1562 475.3263 508.2687 519.1322 313.2745 341.3058 359.3169 402.2035 415.1455 567.3444 581.3604 100 320 340 360 380 400 420 $\frac{1}{1}$ 0 m/z _____ 520 440 580 300 460 480 500 540 560 600 Minimum: -1.5 10.0 Maximum: 2.0 50.0 Mass Calc. Mass mDa PPM DBE i-FIT Norm Conf(%) Formula 435.1742 C25 H27 N2 O3 S 435.1742 0.0 0.0 13.5 1368.9 n/a n/a



HRMS of 3ak

Elemental Composition Report

Single Mass Analysis Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

Monoisotopic Mass, Even Electron Ions 793 formula(e) evaluated with 8 results within limits (up to 1 closest results for each mass) Elements Used: C: 1-30 H: 1-30 N: 1-3 O: 1-6 S: 0-2 CI: 0-2 Br: 0-1

Sample Name : 18_02_294 Test Name :

12012022_18_0	02_294 12 (0.265)						1: TOF MS ES+ 6.38e+006			
384.088	9388.6906 394.1	362 402	408	1481	1526 419	1358 42	5.1747	431.1334	439.3820	46.1040
385.0	390.0 395.0	400.0	405.0	410.0	415.0 4	420.0 4	430	.0 435.0	440.0 4	45.0 450.0
Minimum: Maximum:		2.0	10.0	-1.5 50.0						
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula		
408.1481	408.1481	0.0	0.0	8.5	1268.7	n/a	n/a	C20 H26 N	1 06 S	

IITRPR

XEVO G2-XS QTOF

¹H NMR (CDCl₃, 400 MHz) of 3al





HRMS of 3al

Elemental Composition Report

Single Mass Analysis Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5 Monoisotopic Mass, Even Electron Ions 317 formula(e) evaluated with 4 results within limits (up to 1 closest results for each mass) Elements Used: C: 10-25 H: 1-30 N: 1-3 O: 1-5 S: 0-2 CI: 0-2 Sample Name : 18_02_266 IITRPR XEVO G2-XS QTOF Test Name 12012022_18_02_266 25 (0.543) 1: TOF MS ES+ 9.39e+006 388.1582 100₇ 342.1164 % 410.1401 475.3261 343.1198 453.3442 217.1229 234.0704 128.0632 143.0849 ,476.3293 301.1453_313.2746 0մոր հասևասեստյ munum m/z 180 200 280 300 140 160 260 320 340 360 380 100 120 220 240 400 420 440 460 480 500 -1.5 Minimum: Maximum: 2.0 10.0 50.0 i-FIT Conf(%) Formula Mass Calc. Mass mDa PPM DBE Norm 388.1582 388.1583 n/a -0.1 -0.3 9.5 1466.6 n/a C21 H26 N O4 S

¹H NMR (CDCl₃, 400 MHz) of 3am



Page 1

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

Monoisotopic Mass, Even Electron Ions 3025 formula(e) evaluated with 27 results within limits (up to 1 closest results for each mass) Elements Used: C: 1-30 H: 1-30 N: 1-3 O: 1-6 F: 0-3 S: 0-2 CI: 0-2 Br: 0-1

XEVO G2-XS QTOF Sample Name : 18_02_357 IITRPR Test Name 12012022_18_02_357 11 (0.249) 1: TOF MS ES+ 1.81e+006 406.1488 100₋ % 428.1307 407.1520 364.2750 378.2913_381.3002_387.6921 408.1514 415.2389 429.1348 444.1067447.1142 .0 450.0 401.7082 380.0 390.0 400.0 410.0 370.0 420.0 430.0 440.0 -1.5 Minimum: 2.0 10.0 50.0 Maximum: Mass Calc. Mass mDa PPM DBE i-FIT Norm Conf(%) Formula 406.1488 406.1488 0.0 0.0 9.5 1260.5 n/a n/a C21 H25 N 04 F S





Page 1

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

Monoisotopic Mass, Even Electron Ions 3058 formula(e) evaluated with 15 results within limits (up to 1 closest results for each mass) Elements Used: C: 1-30 H: 1-30 N: 1-3 O: 1-6 F: 0-3 S: 0-2 CI: 0-2 Br: 0-1

Sample Name : 18_02_356 Test Name : 12012022_18_02_356 19 (0.418)

IITRPR

XEVO G2-XS QTOF

1: TOF MS ES+ 1.55e+007

100 %- 0	15 400	407.7181 410	418.	1688 419.1719 420.1697 420	44 437.1420 430	40.1507 441.1544 440 45	456.125	i1 463.2270	6 475.3262 70 480	491.3015 490	502.1241 500 m/z
Minimum: Maximum:			2.0	10.0	-1.5 50.0						
Mass	Calc.	Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula		
418.1688	418.1	688	0.0	0.0	9.5	1451.6	n/a	n/a	C22 H28 N	05 S	



Page 1

XEVO G2-XS QTOF

1: TOF MS ES+ 1.28e+007

Single Mass Analysis

Tolerance = 100.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

 Monoisotopic Mass, Even Electron Ions

 117 formula(e) evaluated with 4 results within limits (up to 1 closest results for each mass)

 Elements Used:

 C: 13-13
 H: 1-30
 N: 1-3
 O: 1-5
 Na: 1-1
 S: 0-2
 CI: 0-2

 Sample Name
 : 18_02_291
 IITRPR

 Test Name
 :
 12012022_18_02_291 12 (0.265)

 100 306.0776

%-	270.0802 274.27	72 284.09	59 _286.0962 25	96.0772	03.0692	7.0810	322.051	7 326.6666	338.3430	,341.3062 m/z
011111	270.0 2	80.0	290.0	300	.0 3	10.0	320.0	330.0	340.0	0 350.0
Minimum: Maximum:		2.0	100.0	-1.5 50.0						
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula		
306.0776	306.0776	0.0	0.0	5.5	1730.4	n/a	n/a	C13 H17 1	N O4 Na	s



XEVO G2-XS QTOF

1: TOF MS ES+

Single Mass Analysis

Tolerance = 300.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

Monoisotopic Mass, Even Electron Ions 14 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass) Elements Used: C: 17-17 H: 0-300 N: 0-1 O: 0-6 S: 1-1 Sample Name : 121035_18-02-297 IITRPR Test Name : 121035_18-02-297 34 (0.729)

3.14e+006 356.0939 100₇ 302.0858 291.1352 9% 357.0971 334.1118 425.2155 442.1304 149.0241 180.0820 194.1188 230.2489 246.2440 42.1304 477.2254 0-hynny 220 minn 280 120 140 160 180 200 240 260 300 320 340 360 380 400 420 440 460 480 500 -1.5 Minimum: 300.0 50.0 2.0 Maximum: Mass Calc. Mass mDa PPM DBE i-FIT Norm Conf(%) Formula 334.1118 334.1113 0.5 1.5 8.5 1365.l n/a n/a C17 H20 N 04 S



-10 0 ò f1 (ppm)

Page 1

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

Monoisotopic Mass, Even Electron Ions 1184 formula(e) evaluated with 4 results within limits (up to 1 closest results for each mass) Elements Used: C: 14-14 H: 1-30 N: 1-3 O: 1-6 S: 0-2 CI: 0-2 Br: 0-1 F: 0-3

Sample Name : 18_02_337 Test Name

IITRPR

XEVO G2-XS QTOF

12012022_18_02_337 13 (0.294) 100₀ 352.0830 1: TOF MS ES+ 5.12e+006





Elemental Composition Report	HRMS of 3ea	Page 1
Single Mass Analysis Tolerance = 10.0 PPM / DBE: min = -1. Element prediction: Off Number of isotope peaks used for i-FIT = 5	5, max = 50.0 5	
Monoisotopic Mass, Even Electron Ions 332 formula(e) evaluated with 4 results within I Elements Used: C: 1-30 H: 1-30 N: 1-3 O: 1-5 S: 0-	imits (up to 1 closest results for each mass)	
Sample Name : 18_02_305	IITRPR	XEVO G2-XS QTOF
12012022_18_02_305 16 (0.356)		1: TOF MS ES+
100-		1.110.001
%-	336.0880 314.1062	
227.1765 254.0866 271.9939	333.0795 352.0621359.1659	384.0745398.0607 422.1290 m/z
230 240 250 260 270 280 290	300 310 320 330 340 350 360 370	380 390 400 410 420 430
Minimum: Maximum: 2.0 10.0	-1.5 50.0	
Mass Calc. Mass mDa PPM	DBE i-FIT Norm Conf(%)	Formula
314.1062 314.1062 0.0 0.0	5.5 1600.2 n/a n/a	C14 H20 N 05 S





Page 1

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

Monoisotopic Mass, Even Electron Ions 650 formula(e) evaluated with 10 results within limits (up to 1 closest results for each mass) Elements Used: C: 1-30 H: 1-30 N: 1-3 O: 1-5 S: 0-2 CI: 0-2 Br: 0-1

Sample Name : 18_02_287 Test Name : 12012022_18_02_287 4 (0.107)



1: TOF MS ES+ 1.31e+007

XEVO G2-XS QTOF



L) NMR and HRMS of derivatives



Page 1

Single Mass Analysis Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

Monoisotopic Mass, Even Electron Ions 578 formula(e) evaluated with 3 results within limits (up to 1 closest results for each mass) Elements Used: C: 1-30 H: 1-30 N: 1-3 O: 1-6 S: 0-2 CI: 0-2 Br: 0-1

Sample Name Test Name 12012022_18	02	18_02_ _318 21	318 (0.452)				IITRPF	2			XEVO G2-XS QTOF 1: TOF MS ES+
100 56 0 200	22	0.0204	227.176	0 238.0910 0 240	249.1604	266.0 256.0005 260	267.0884	284.0957	306. 303.0696 	0777 313.2743 322.0517 11.11.11.11.11.11.11.11.11.11.11.11.11.	341.3056
Minimum: Maximum:				2.0	10.0	-1.5 50.0					
Mass	0	alc.	Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula	
284.0957	2	284.09	57	0.0	0.0	5.5	1313.8	n/a	n/a	C13 H18 N 04 S	



Page 1

XEVO G2-XS QTOF

1: TOF MS ES+

Single Mass Analysis Tolerance = 10.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

Monoisotopic Mass, Even Electron lons 1586 formula(e) evaluated with 8 results within limits (up to 1 closest results for each mass) Elements Used: C: 1-30 H: 1-30 N: 1-3 O: 1-6 F: 0-3 S: 0-2 CI: 0-2 Br: 0-1

Sample Name : 18_02_355 Test Name : 12012022_18_02_355 5 (0.124)

100 %- 0- 200	212.0211 ^{227.1797} 210 220 2	238.0910 250 230 240	252.10	253.1087 27 266.0854 260	70.1164 274.3 270 28	292.1 2747 0 290	293.1019	3.0744 ^{313.2744} 310 320	338.3425 330 340	2.77e+007 350.1401 350 350
Minimum: Maximum:		2.0	10.0	-1.5 50.0						
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula		
270.1164	270.1164	0.0	0.0	4.5	1930.0	n/a	n/a	C13 H20 N (03 5	

IITRPR

N) NMR and HRMS of Michael adducts



Single Mas Tolerance = Element pred Number of is	s Analysis 300.0 PPM / diction: Off otope peaks us	DBE: mi ed for i-f	n = -1.5, -IT = 5	max =	50.0					
Monoisotopic 4 formula(e) e Elements Use C: 13-13 H	Mass, Even Elect valuated with 1 r d: : 0-300 N: 1-2	tron lons esults wit 2 O: 1-2	hin limits 2 S: 1-	(all resul	ts (up to 100	00) for e	ach mass)			
Sample Name	: 201021_18-02-2	86			IITRPR				XEVC	G2-XS QTOF
201021_18-02-2	286 7 (0.169)								1:	TOF MS ES+ 2.63e+007
100 % 0 139.013 100 150	265.097 0 264.1432 0 200 250	2287.0786 313.26 313.26 300 35	85 416.10 416.10 400	029 149.2101 149.21001 149.21000000000000000000000000000000000000	71.2128 551. 11.2128 550	.1767 (600	663.4580685.4 	406 809.212 	4.835.1365 94 1991 - 1995 - 1995 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 19	19.5447 11111111111111111111111111111111111
Minimum: Maximum:		2.0	300.0	-1.5 50.0						
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula		
265.0972	265.1011	-3.9	-14.7	6.5	2367.7	n/a	n/a	C13 H17 N2	2 02 S	






S73

Elemental Composition Report

Page 1

Single Mass Analysis

Tolerance = 5.0 PPM / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 5

Monoisotopic Mass, Even Electron Ions 370 formula(e) evaluated with 2 results within limits (up to 1 closest results for each mass) Elements Used: Ts. C: 0-50 H: 0-50 N: 0-5 O: 0-5 S: 0-2 Sample Name : 18_02_CRUDE IITRPR XEVO G2-XS QTOF Test Name 07032022_18_02_CRUDE 4 (0.107) 1: TOF MS ES+ 3.27e+007 320.0948 100 CO₂Me 298.1115 [M+H]⁺=298.1113 266.0843 % 321.0951 186.0115 453.3433^{465.6358}475.3252 224.0730 184.0162 187.0144 ,322.0909 382.0836406.1291 238.0889 mununin m/z ակահիստրուրուրեսրեկություրուրուրությո 0 200 22 thhhhhhhhhh 0 240 ritte 160 180 220 260 280 300 320 340 360 380 400 420 440 460 480 500 Minimum: -1.5 2.0 5.0 50.0 Maximum: Calc. Mass PPM DBE i-FIT Conf(%) Formula Mass mDa Norm 298.1115 298.1113 0.2 0.7 2029.7 C14 H20 N O4 S 5.5 n/a n/a